

Two Luminescent Metal-Organic Frameworks with Multifunctional Properties for Nitroaromatic Compounds Sensing and Photocatalysis

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Supplementary Materials

Table S1 Crystal data and structure refinement for **1** and **2**

Compound	1	2
Chemical formula	C ₂₄ H ₂₀ N ₃ O ₈ Zn	C ₄₈ H ₄₀ Cd ₂ N ₆ O ₁₆
Formula weight	543.80	1181.66
Temperature [K]	113(2)	113(2)
Wavelength[Å]	0.71073	0.71073
Crystal system	Monoclinic	Monoclinic
Space group	P2(1)/c	P2(1)/c
a[Å]	10.170(2)	10.275(2)
b[Å]	16.233(3)	16.704(3)
c[Å]	12.934(3)	13.021(3)
α [°]	90	90
β [°]	100.72(3)	102.80(3)
γ [°]	90	90
Volume [Å ³]	2098.1(7)	2179.2(8)
Z, Calculated density [Mg/m ³]	4, 1.722	2, 1.801
Absorption coefficient[mm ⁻¹]	1.233	1.062
Theta range for data collection (deg)	2.04 to 28.02	2.01 to 27.95
	-13<=h<=13, -17<=k<=21,	-13<=h<=13, -21<=k<=21, -
Limiting indices	-16<=l<=17	17<=l<=16
<i>F</i> (000)	1116	1188
Reflections collected/unique	21070/5032[R(int) = 0.0448]	21709/5203 [R(int) = 0.0356]
Goodness-of-fit on <i>F</i> ²	0.999	1.085
Final R indices [<i>I</i> >2 σ (<i>I</i>)]	R ₁ = 0.0373, wR ₂ = 0.0946	R ₁ = 0.0313, wR ₂ = 0.0638
R indices (all data)	R ₁ = 0.0463, wR ₂ = 0.0993	R ₁ = 0.0381, wR ₂ = 0.0667
Largest diff. peak and hole[e.Å ⁻³]	0.764 and -0.578	0.785 and -0.882

Table S2 Selected bond lengths and angles for **1-2** (Å, °)

1			
Zn(1)-O(1)	2.0408(15)	Zn(1)-N(3)	2.1097(17)
Zn(1)-O(6)	2.1148(15)	Zn(1)-N(2)	2.1597(17)
Zn(1)-O(4)#1	2.1831(16)	Zn(1)-O(3)#1	2.3191(16)
O(1)-Zn(1)-N(3)	117.04(6)	O(1)-Zn(1)-O(6)	93.39(6)
N(3)-Zn(1)-O(6)	92.84(7)	N(3)-Zn(1)-N(2)	77.92(7)
O(6)-Zn(1)-N(2)	168.09(6)	O(4)#1-Zn(1)-O(3)#1	58.13(6)
O(1)-Zn(1)-O(4)#1	92.59(6)	N(3)-Zn(1)-O(4)#1	147.52(6)
O(6)-Zn(1)-O(4)#1	98.67(6)	N(2)-Zn(1)-O(4)#1	85.68(6)
O(1)-Zn(1)-O(3)#1	148.99(6)	N(3)-Zn(1)-O(3)#1	93.90(6)
O(6)-Zn(1)-O(3)#1	82.34(6)	N(2)-Zn(1)-O(3)#1	90.71(6)
O(4)#1-Zn(1)-O(3)#1	58.13(6)	O(1)-Zn(1)-C(8)#1	121.75(7)
2			

Cd(1)-O(4)#1	2.240(4)	Cd(1)-O(7)	2.289(4)
Cd(1)-N(3)	2.298(5)	Cd(1)-N(2)	2.335(4)
Cd(1)-O(1)	2.391(4)	Cd(1)-O(2)	2.441(4)
O(4)#1-Cd(1)-O(7)	98.66(14)	O(4)#1-Cd(1)-N(3)	123.85(16)
O(7)-Cd(1)-N(3)	91.28(15)	O(4)#1-Cd(1)-N(2)	97.25(15)
O(7)-Cd(1)-N(2)	161.70(14)	N(3)-Cd(1)-N(2)	72.29(16)
O(4)#1-Cd(1)-O(1)	85.77(14)	O(7)-Cd(1)-O(1)	106.79(14)
N(3)-Cd(1)-O(1)	143.02(14)	N(2)-Cd(1)-O(1)	83.32(15)
O(4)#1-Cd(1)-O(2)	137.04(14)	O(7)-Cd(1)-O(2)	80.74(14)
N(3)-Cd(1)-O(2)	99.06(15)	N(2)-Cd(1)-O(2)	93.68(15)
O(1)-Cd(1)-O(2)	54.41(13)	O(4)#1-Cd(1)-C(1)	112.40(15)

Symmetry transformations used to generate equivalent atoms:

For 1: #1 $x+1, y, z$ #2 $x-1, y, z$ #3 $-x+1, -y+1, -z+1$

For 2: #1 $x-1, y, z$ #2 $x+1, y, z$ #3 $-x+1, -y+1, -z+1$

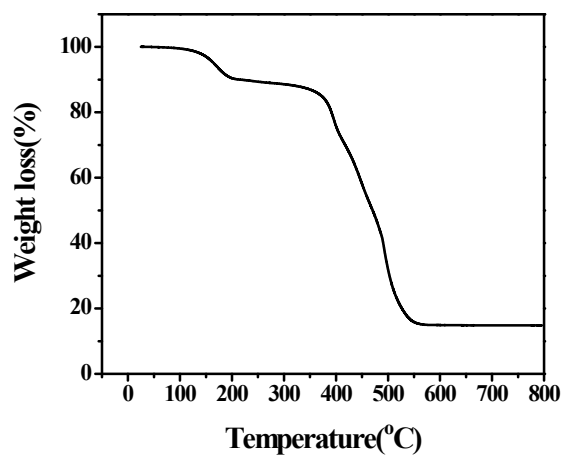


Fig. S1 TG curve of 1.

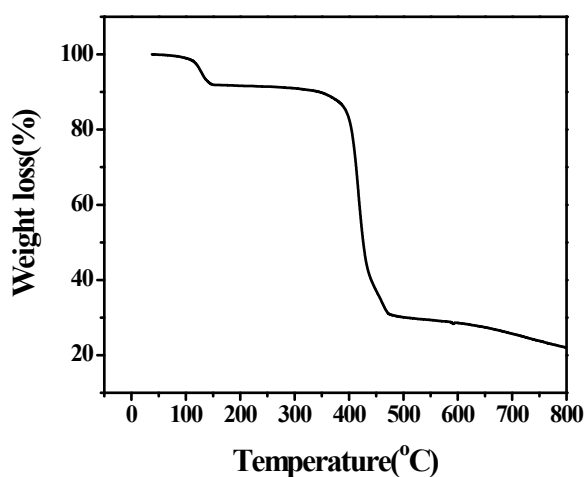


Fig. S2 TG curve of 2.

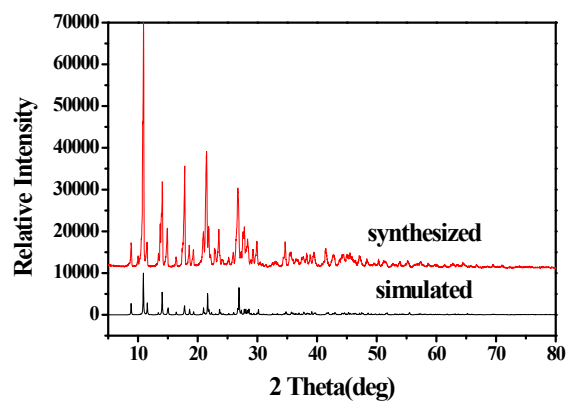


Fig. S3 PXRD pattern of 1

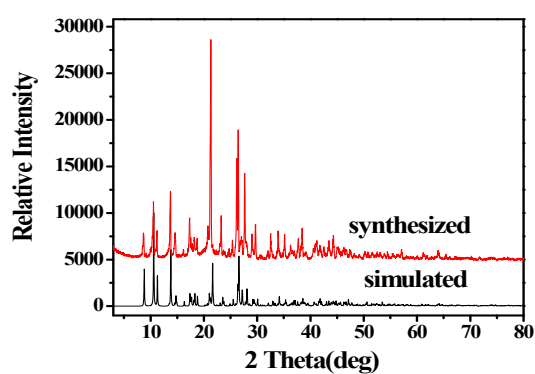


Fig. S4 PXRD pattern of 2

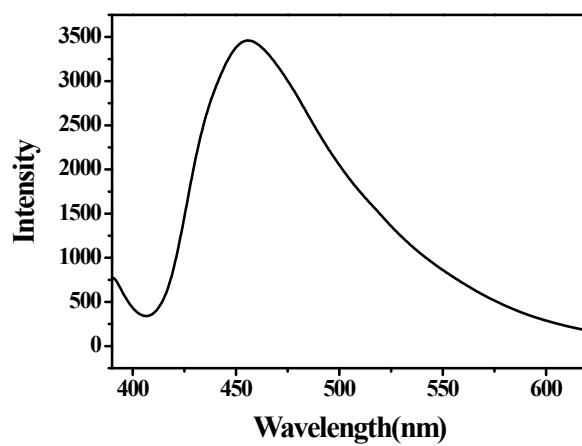


Fig. S5 The fluorescence spectra of 1 ($\lambda_{\text{ex}} = 330 \text{ nm}$) at the solid state at room temperature

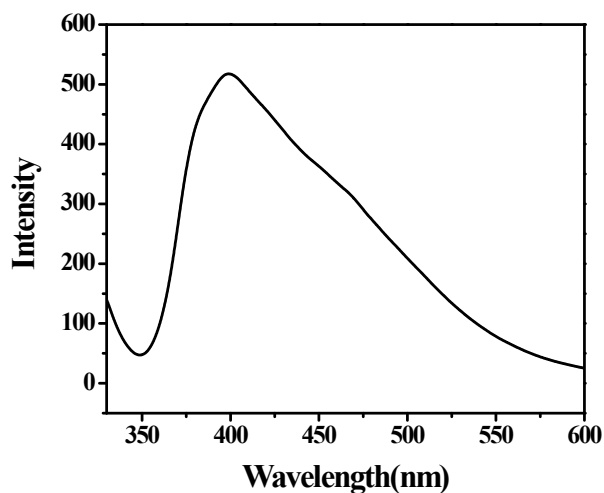


Fig. S6 The fluorescence spectra of **2** ($\lambda_{\text{ex}} = 318 \text{ nm}$) at the solid state at room temperature

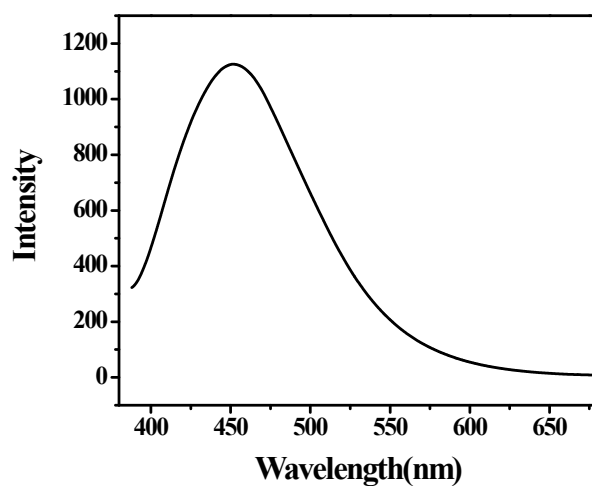


Fig. S7 The fluorescence spectra of **L** ligand ($\lambda_{\text{ex}} = 357 \text{ nm}$) at the solid state at room temperature

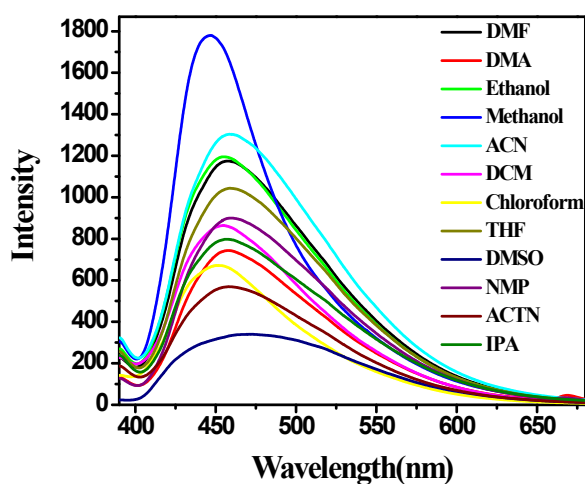


Fig. S8 Fluorescence emission spectra of **1** in different organic solvent at room temperature.

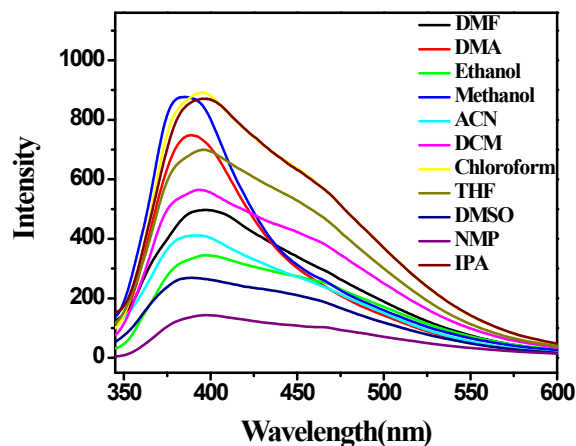


Fig. S9 Fluorescence emission spectra of **2** in different organic solvent at room temperature

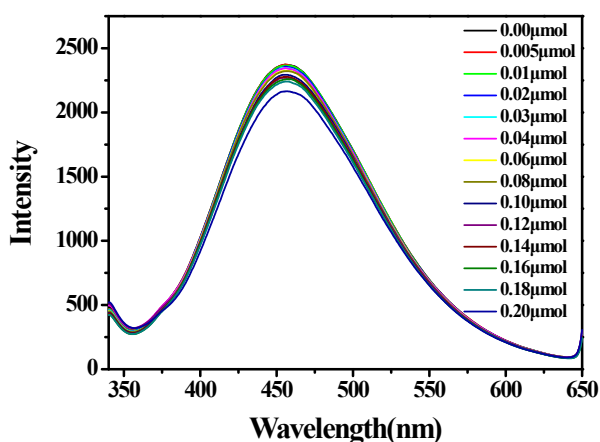


Fig. S10 Fluorescence titration of **1** dispersed in DMA with the addition of different volume of 10^{-3} M DMA solution of *p*-nitrotoluene. The excitation wavelength was 330 nm and fluorescence emission was monitored from 340 to 650 nm. The slit width for both excitation and emission were 5 nm.

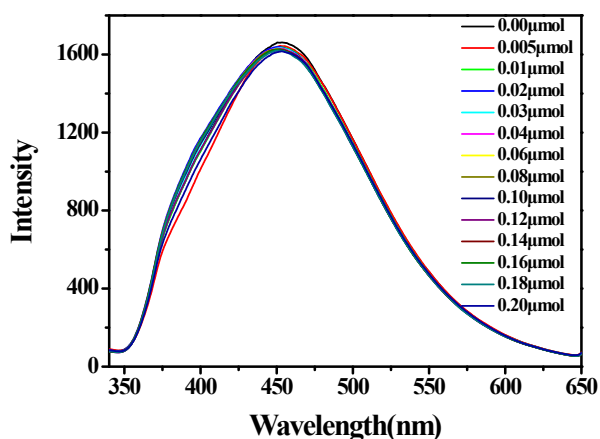


Fig. S11 Fluorescence titration of **1** dispersed in DMA with the addition of different volume of 10^{-3} M DMA solution of nitrobenzene. The excitation wavelength was 330 nm and fluorescence emission was monitored from 340 to 650 nm. The slit width for both excitation and emission were 5 nm.

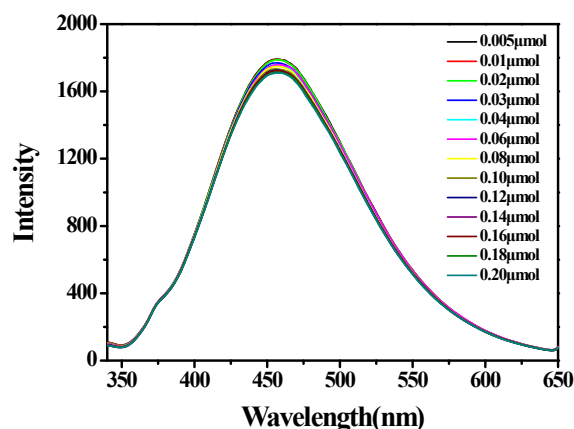


Fig. S12 Fluorescence titration of **1** dispersed in DMA with the addition of different volume of 10^{-3} M DMA solution of *m*-nitrotoluene. The excitation wavelength was 330 nm and fluorescence emission was monitored from 340 to 650 nm. The slit width for both excitation and emission were 5 nm.

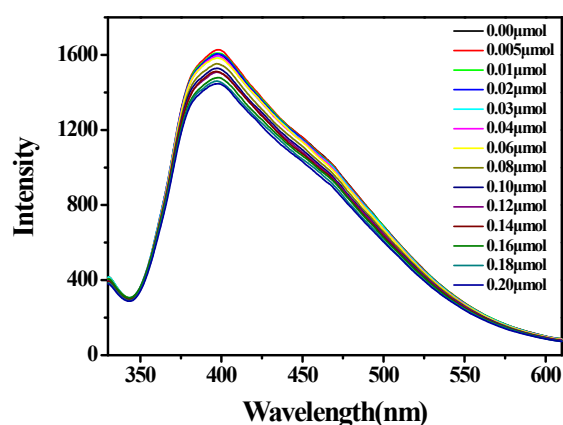


Fig. S13 Fluorescence titration of **2** dispersed in isopropanol with the addition of different volume of 10^{-3} M isopropanol solution of *p*-nitrotoluene. The excitation wavelength was 318 nm and fluorescence emission was monitored from 330 to 600 nm. The slit width for both excitation and emission were 5 nm.

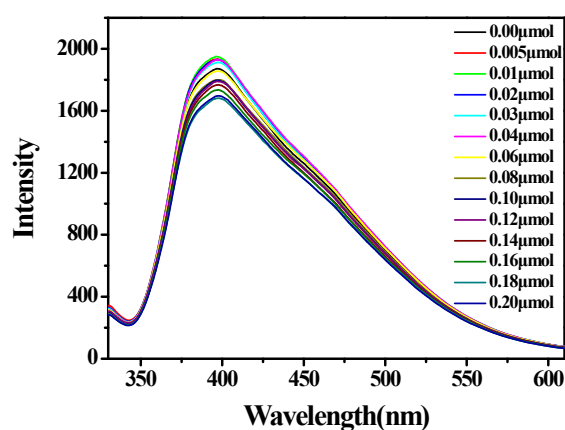


Fig. S14 Fluorescence titration of **2** dispersed in isopropanol with the addition of different volume of 10^{-3} M isopropanol solution of nitrobenzene. The excitation wavelength was 318 nm and fluorescence emission was monitored from 330 to 600 nm. The slit width for both excitation and

emission were 5 nm.

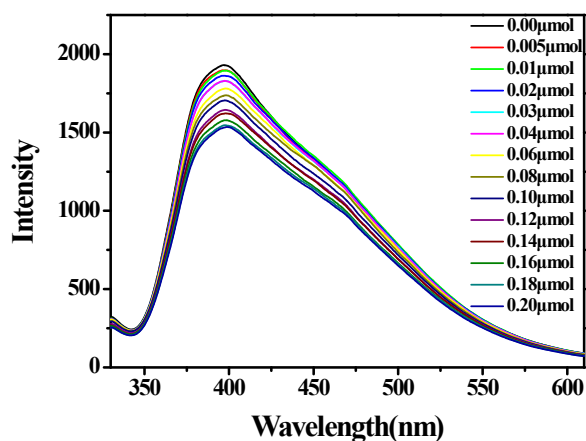


Fig. S15 Fluorescence titration of **2** dispersed in isopropanol with the addition of different volume of 10^{-3} M isopropanol solution of *m*-nitrotoluene. The excitation wavelength was 318 nm and fluorescence emission was monitored from 330 to 600 nm. The slit width for both excitation and emission were 5 nm.

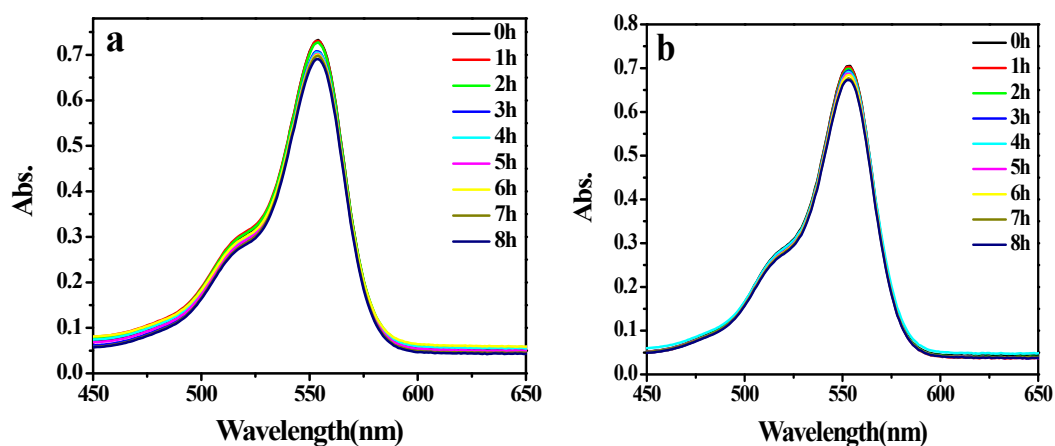


Fig. S16 Absorption spectra of RhB degraded by **1** (a) and **2** (b) under dark conditions.

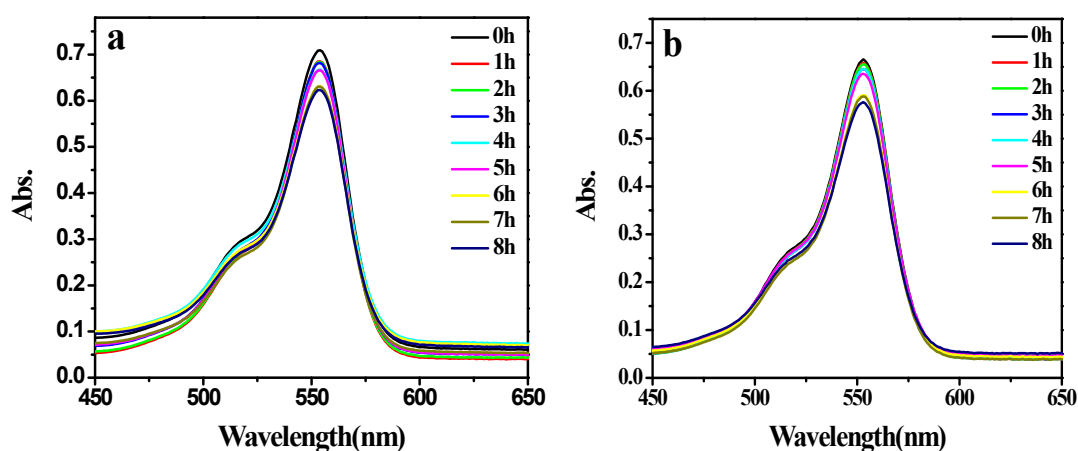


Fig. S17 Absorption spectra of RhB degraded by **1** (a) and **2** (b) under natural illumination.

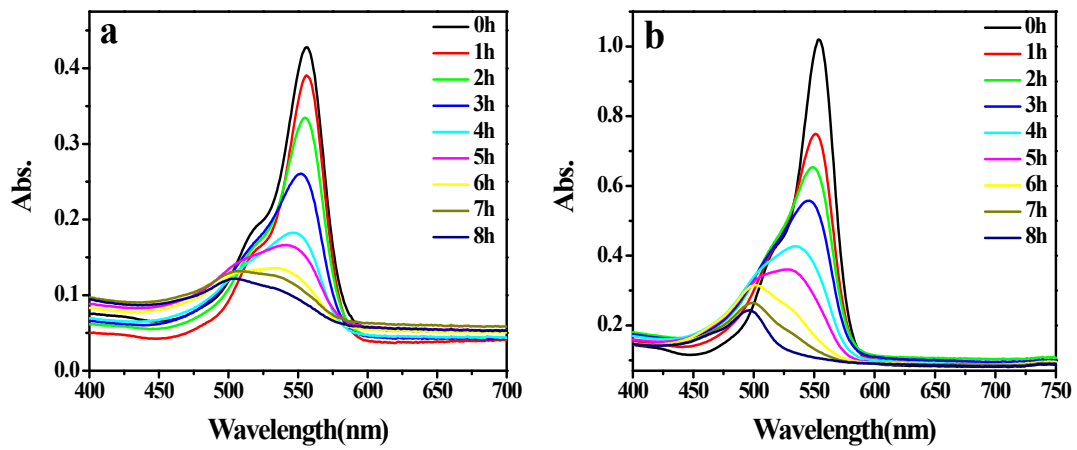


Fig. S18 Absorption spectra of RhB degraded by 1 (a) and 2 (b) under UV light.

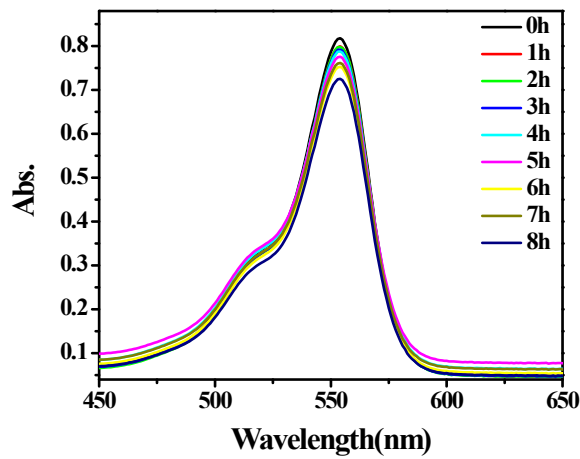


Fig. S19 Absorption spectra of RhB degraded without any catalyst under UV light.

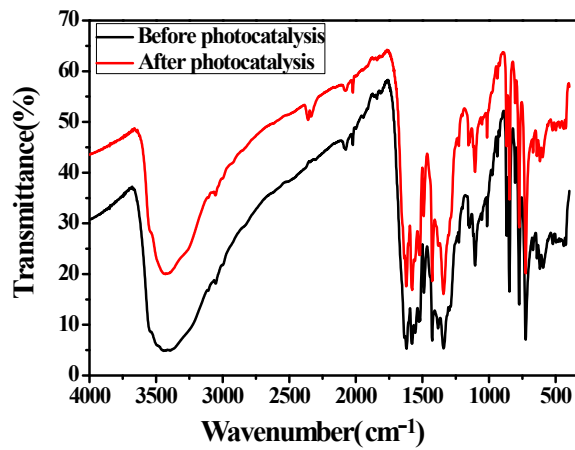


Fig. S20 The IR spectra of 1 before and after photocatalysis

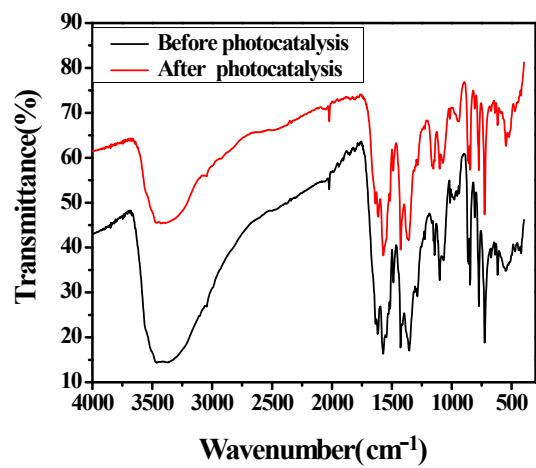


Fig. S21 The IR spectra of 2 before and after photocatalysis