

1 **Supporting Information**

2 **Performance of Wastewater Biological Phosphorus Removal under Long-term**

3 **Exposure to CuNPs: Adapting Toxicity via Microbial Community Structure**

4 **Adjustment**

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14 Analytical Methods

15 **Synthetic Wastewater.** The synthetic wastewater included stock SOP solution named “P-water”,
16 “concentration solution”, “trace-element solution” (adapted from Smolders et al.¹), a certain amount of carbon
17 source and tap water. The P-water contained (g/L): 23.5 KH_2PO_4 and 17.6 $\text{K}_2\text{HPO}_4 \cdot 3\text{H}_2\text{O}$. The concentrated
18 solution contained (g/L): 25.88 peptone, 4.24 yeast extract, 33.94 $\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$, 19.09 $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 8.91
19 $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$, 8.91 NH_4Cl and 0.11 allylthiourea (nitrification inhibitor). The trace-element solution contained
20 (g/L): 1.50 $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$, 0.03 $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$, 0.12 $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$, 0.06 $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$, 0.12 $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$, 0.15
21 $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$, 0.18 KI, 0.15 H_3BO_3 and 10 ethylenediamine tetraacetic acid. One liter of synthetic wastewater
22 contained 1.94mL of P-water, 2.14 mL of concentrated solution, 0.27 mL of trace-element solution, and 0.267 mL
23 of acetic acid. The synthetic wastewater final composition of the influent contained (mg/L): 55.38 peptone, 9.07
24 yeast extract, 72.63 $\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$, 40.85 $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, 19.07 $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$, 19.07 NH_4Cl , 0.24 allylthiourea, 0.41
25 $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$, 0.01 $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$, 0.03 $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$, 0.02 $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$, 0.03 $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$, 0.04 $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$,
26 0.05 KI, 0.04 H_3BO_3 , 2.70 ethylenediamine tetraacetic acid, 45.59 KH_2PO_4 , 34.14 $\text{K}_2\text{HPO}_4 \cdot 3\text{H}_2\text{O}$ and 280 acetic
27 acid.

28 **Set-up and Operation of Parent Sequencing Batch Reactor (SBR).** The EBPR activated sludge was cultured
29 with synthetic municipal wastewater in an anaerobic-aerobic SBR, which was covered by foil on the outside and
30 had a working volume of 4 L with a cycle of 8 h consisting of 2 h anaerobic and 3h aerobic periods, followed by
31 1h settling, 10 min decanting and 110 min idle periods. The reactor was maintained at 21 ± 1 °C and constantly
32 mixed with a magnetic stirrer except during the settling, decanting and idle periods. The influent pH was
33 adjusted to 7.5 by adding 4 M NaOH or 4M HCl. In the aerobic stage, air was provided by an aerator using an
34 on/off control system and the dissolved oxygen (DO) concentration maintained at around 6 mg/L. After the
35 settling period, 3 L of the supernatant was discharged, and replaced with 3 L fresh synthetic wastewater during the
36 next initial 10 min of the anaerobic time. Sludge was wasted at the end of aerobic periods to keep the solids

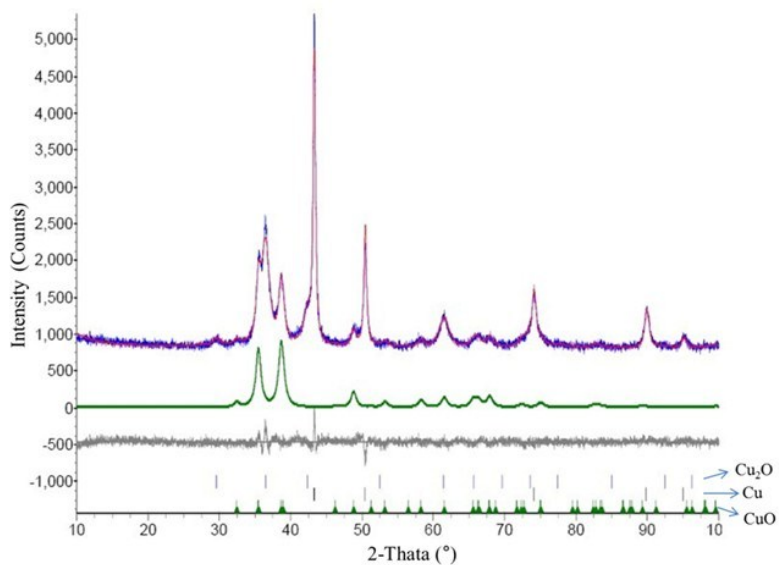
37 retention time (SRT) at approximately 10 d. The initial BOD concentration in the SBR was increased
38 progressively over a 30 d period from around 80 to approximately 300 mg/L. The beginning ratio of BOD/SOP
39 was 16, and it was increased to the final value of 20. After around 100 days' acclimatization, the phosphorus
40 anaerobic release and aerobic uptake as well as net removal in the SBR reached relatively stable. The mixed
41 liquid suspended solids (MLSS) and mixed liquid volatile suspended solids (MLVSS) were 3200 ± 190 and 2380
42 ± 126 mg/L in the parent SBR.

43 **Stock Synthetic Wastewater Used in Batch Experiment.** In the batch experiment of short-term effect of
44 CuNPs on phosphorus removal, 100 mL of synthetic wastewater was added to each reactor, and the final
45 volume of the mixture in each reactor was 400 mL. Thus, the concentration of synthetic wastewater
46 used in batch experiment would be 4 times higher than that used in the parent SBR, which meant
47 that one liter of synthetic wastewater contains 1.07 mL acetic acid, 7.76 mL P-water, 8.56 mL
48 concentrated solution, and 1.08 mL trace-element solution.

49 **Determination of CuNPs Removal Efficiency during EBPR.** At the beginning of the batch
50 experiment of short-term effect of CuNPs on phosphorus removal, the CuNPs concentration in
51 reactor A was 5 mg/L. At the end of aerobic stage, the sludge mixture was collected, and was acid
52 digested using concentrated nitric acid at 125 °C for 2 h; Then the digested solution was filtered
53 through 0.22 um mixed cellulose ester membrane, and determined by inductively coupled plasma
54 optical emission spectrometry (ICP-OES, PerkinElmer Optima 2100 DV, USA). The aim of
55 detecting the CuNPs concentration in the mixture was to confirm that the detected real
56 concentration was matched with the theory one, and the detected concentration was set as C_0 . In
57 addition, at the end of settling period, the supernatant was withdrawn and the CuNPs concentration
58 in the supernatant was detected following the same steps as the sludge mixture, and the
59 concentration in the supernatant was set as C_1 . Therefore, the removal efficiency was calculated

60 as: $(1-C_1/C_0)*100\%$.

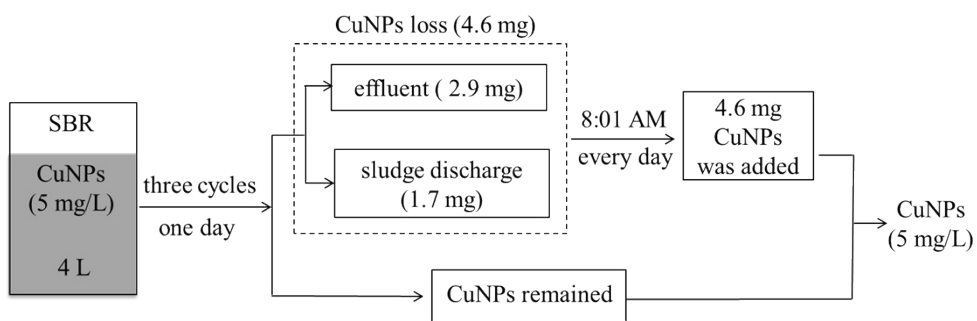
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Figure S1. The XRD pattern of the Cu-NPs used in this study.



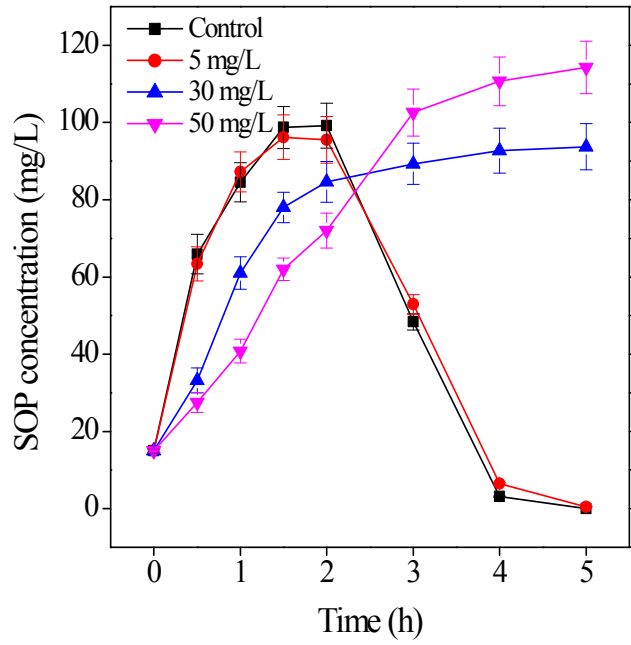
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Figure S2. Flow-process diagram of how to keep the CuNPs concentration in the reactor

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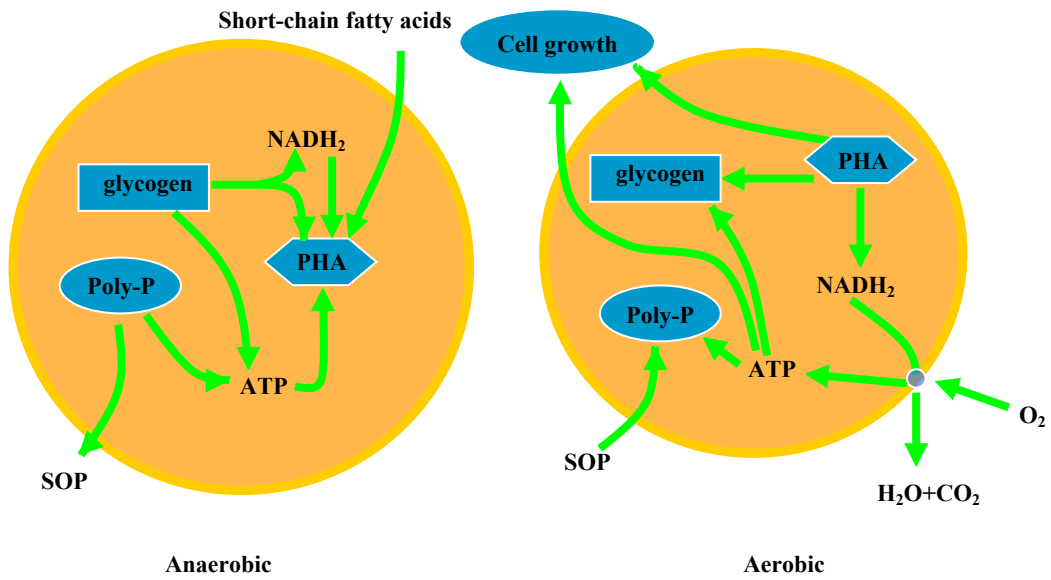
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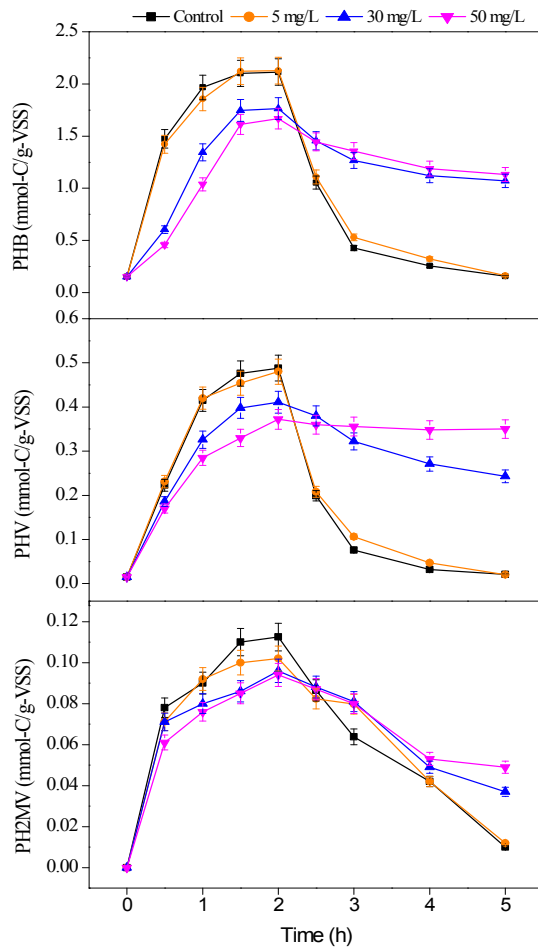


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 69 **Figure S3. Effect of shock load of CuNPs on biological phosphorus removal. Error bars represents**
 70 **standard deviations of triplicate tests.**
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 74 **Figure S4. The schematic diagram of main substrates metabolism involved in the anaerobic and aerobic**
 75 **stages of enhanced biological phosphorus removal.**



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77 **Figure S5. Effect of shock load of CuNPs on transformations of PHB, PHV and PH2MV during one cycle of**
 78 **EBPR. Error bars represents standard deviations of triplicate tests.**

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Table S1. Oligonucleotide Probes Used in This Study

Probe	Specificity	Sequence (5'-3')	FA (%)
PAO462	<i>Rhodocyclus</i> -related PAO in <i>Betaproteobacteria</i>	CCGTCATCTACWCAGGGTATTAAC	20-35
PAO651	<i>Rhodocyclus</i> -related PAO in <i>Betaproteobacteria</i>	CCCTCTGCCAAACTCCAG	35
PAO846	<i>Rhodocyclus</i> -related PAO in <i>Betaproteobacteria</i>	GTTAGCTACGGCACTAAAAGG	35
GAO Q431	<i>Gammaproteobacteria</i> <i>Competibacter</i> spp.	TCCCCGCCTAAAGGGCTT	35
GAO Q989	<i>Gammaproteobacteria</i> <i>Competibacter</i> spp.	TTCCCCGGATGTCAAGGC	35
GB_G2	<i>Gammaproteobacteria</i> <i>Competibacter</i> spp.	TTCCCCAGATGTCAAGGC	35
TFO_DF218	<i>Defluvicoccus</i> -related TFO in <i>Alphaproteobacteria</i>	GAAGCCTTTGCCCTCAG	20-35
TFO_DF618	<i>Defluvicoccus</i> -related TFO in <i>Alphaproteobacteria</i>	GCCTCACTTGTCTAACCG	20-35
DF988	<i>Defluvicoccus</i> -related DF in <i>Alphaproteobacteria</i>	GATACGACGCCCATGTCAAGGG	35
DF1020	<i>Defluvicoccus</i> -related DF in <i>Alphaproteobacteria</i>	CCGGCCGAACCGACTCCC	35
EUB338	Most Bacteria	GCTGCCTCCCGTAGGAGT	35
EUB338-II	<i>Planctomycetales</i> and other Bacteria not detected by EUB338	GCAGCCACCCGTAGGTGT	35
EUB338-III	<i>Verrucomicrobiales</i> and other Bacteria not detected by EUB338	GCTGCCACCCGTAGGTGT	35

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Table S2. Statistical Analysis Results of Effect of Shock Load of 5 mg/L CuNPs on Transformations of PHA and Glycogen during One EBPR Cycle (Compared with the Control)

	F_{observed}	$F_{\text{significance}}$	$P_{(0.05)}$
PHA	7.5×10^{-3}	4.5	0.98
Glycogen	6.2×10^{-3}	4.5	0.94

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Table S3. Statistical Analysis Results of Effect of Shock Load of 5 mg/L CuNPs on Transformations of PHA and Glycogen during One EBPR Cycle (Compared with the Control)

	F_{observed}	$F_{\text{significance}}$	$P_{(0.05)}$
PPX	0.74	7.71	0.44
PPK	0.29	7.1	0.62

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86 Reference

- 87 (1) Smolders, G. J. F.; Vandermeij, J.; Vanloosdrecht, M. C. M.; Heijnen, J. J. Model of the anaerobic metabolism
 88 of the biological phosphorus removal process -stoichiometry and pH influence. *Biotechnol. Bioeng.* **1994**, *43*,
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