# Selective and Sensitive Detection of Hydrogen Sulfide in Live Cells

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#### Supplementary data

### 1. General experimental methods

All reagents were purchased from commercially available suppliers, and used without further purification unless otherwise stated. BODIPY fluorophore **1** was synthesized according to literature procedures. Reactions were monitored by thin-layer chromatography (TLC) carried out on Silica gel 60 F254 plates supplied by Qingdao Puke Separation Material Corporation using UV light as the visualizing agen. Flash column chromatography was performed using 200-300 mesh silica gel supplied by Qingdao Marine Chemical Factory, Qingdao, P. R. <sup>1</sup>H NMR spectrum was recorded on a Bruker Fourier transform spectrometer (500 MHz) at 25°C. <sup>13</sup>C NMR spectrum was recorded on a Bruker Fourier transform spectrometer (125 MHz) spectrometer and were calibrated using residual undeuterated CDCl<sub>3</sub> as an internal reference (<sup>1</sup>H NMR = 7.26, <sup>13</sup>C NMR = 77.16). All chemical shifts were given in ppm and coupling constants (*J*) in Hz. The following abbreviations or combinations thereof were used to explain the multiplicities: s = singlet, d = doublet, m = multiplet. High resolution mass spectra (HRMS) were measured on an Agilent 6224 TOF LC/MS spectrometer using ESI-TOF (electrospray ionization-time of flight). Fluorescence measurements were carried out on a SHIMADZU RF-5301pc spectrofluorimeter.

#### 2. Synthesis and characterization of PS1



BODIPY 1

BODIPY 1 was synthesized according to literature procedures in 42% yield as a red powder.<sup>1</sup>

m. p. >250°C

<sup>1</sup>H NMR (500 MHz, DMSO)  $\delta$  10.18 (s, 1H), 7.75 (d, J = 16.2 Hz, 1H), 7.71 (s, 1H), 7.64 (s, 1H), 7.51 (d, J = 8.6 Hz, 2H), 7.26 (d, J = 16.2 Hz, 1H), 7.13 (s, 1H), 7.04 (d, J = 3.7 Hz, 1H), 6.88 (d, J = 8.6 Hz, 2H), 6.48 (dd, J = 3.7, 2.1 Hz, 1H), 2.33 (s, 3H).

<sup>1</sup> J. S. Lee, N. Y. Kang, Y. K. Kim, A. Samanta, S. Feng, H. K. Kim, M. Vendrell, J. H. Park, Y. T. Chang, *J. Am. Chem. Soc.* 2009, **131**, 10077.

### Probe PS1

To a stirred solution of BODIPY 1 (50 mg, 0.15 mmol) in DMF (2.0 mL) was added  $K_2CO_3$  (21 mg, 0.15 mmol) and 2, 4-dinitrofluorobenzene (34 mg, 0.18 mmol). After being stirred at room temperature for 5 min, the reaction was quenched by the addition of  $H_2O$  (5 mL). The mixture was then extracted with EtOAc (10 mL x 3) and the combined organic phases was washed sequentially with  $H_2O$  (5 mL x 3) and brine (5 mL x 3). After being dried over anhydrous  $Na_2SO_4$ , the mixture was concentrated under reduced pressure and the resulting crude residue was purified by column chromatography on silica gel (petroleum:EtOAc = 6:1) to yield the product as a dark solid (50 mg, 67% yield).

## m. p. >250 °C

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  8.87 (d, J = 2.7 Hz, 1H), 8.35 (dd, J = 9.2, 2.7 Hz, 1H), 7.70 (m, 3H), 7.61 (d, J = 16.3 Hz, 1H), 7.35 (d, J = 16.3 Hz, 1H), 7.22 (s, 1H), 7.16 (d, J = 8.6 Hz, 2H), 7.10 (d, J = 9.2 Hz, 1H), 6.98 (d, J = 3.7 Hz, 1H), 6.76 (s, 1H), 6.49 (m, 1H), 2.34 (s, 3H).

<sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 157.80, 155.71, 154.69, 144.70, 141.97, 140.09, 139.95, 137.81, 137.49, 134.48, 133.59, 130.13, 129.01, 127.07, 124.00, 122.31, 120.94, 119.73, 119.13, 117.13, 117.08, 11.66. Mass data: ESI-HRMS (m/z): [M+Na]<sup>+</sup> calc'd. for  $C_{24}H_{17}BF_2N_4NaO_5 = 513.1158$ ; found 513.1153

3. <sup>1</sup>H NMR spectra of BODIPY 1



4. <sup>1</sup>H NMR <sup>13</sup>C NMR and HRMS spectra of PS1











**Qualitative Analysis Report** 

5. Supplementary figures



**Fig. S1** Plot of the fluorescent intensity of **PS1** (5  $\mu$ M) at 570 nm after treated with various concentrations of NaHS for 20 min *versus* NaHS concentration. Data were collected in PBS (10 mM, PH=7.4, 5% ethanol, 37°C) with CTAB (100  $\mu$ M) as a cationic surfactant.  $\lambda_{ex}$ =535 nm. Slit widths: 10 nm for excitation and 5 nm for emission.



Fig. S2 Detection limit determination. The detection limit of PS1 was determined to be the concentration of NaHS that induced three-fold fluorescence intensity enhancement. a) Fluorescent spectra of PS1 (5  $\mu$ M) in the presence of various concentrations of NaHS. b) Fluorescent intensity of PS1 (5  $\mu$ M) at 570 nm after treated with various concentrations of NaHS for 20 min. Measurements were conducted in PBS (10 mM, PH=7.4, 5% ethanol, 37°C) with CTAB (100  $\mu$ M).  $\lambda_{ex}$ =535 nm. Slit widths: 10 nm for excitation and 10 nm for emission.