

Electronic Supplementary Information (ESI) for Large-scale preparation of Mg doped $\text{LiFePO}_4@C$ for lithium ion batteries via carbon thermal reduction combined with aqueous rheological phase technology

Yuanchao Li,^a Jinghao Hao,^b Guangwei Geng,^a Yafang Wang,^a Xiaokun Shang,^a Changchun

Yang*^a and Baojun Li*^a

^a College of Chemistry and Molecular Engineering, Zhengzhou University, Zhengzhou 450001, P.

R. China

^b Department of Quality Examination and Management, Henan University of Animal Husbandry
and Economy, Zhengzhou 450011, P. R. China

*Corresponding author. E-mail: changchunyangzzu@126.com and lbjfc1@zzu.edu.cn

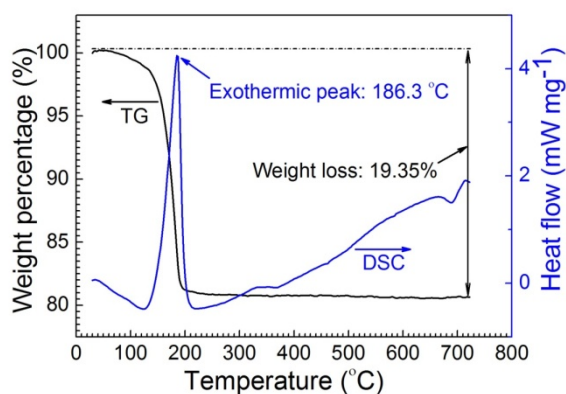


Figure S1. TG-DSC curves of $\text{FePO}_4 \cdot 2\text{H}_2\text{O}$.

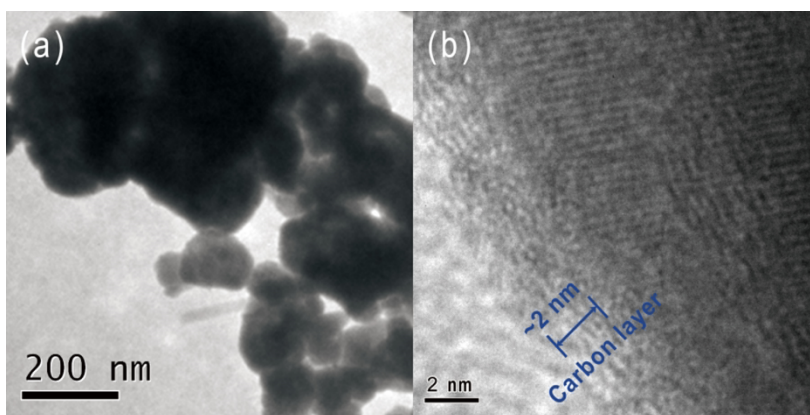


Figure S2. TEM images of the al-LFP.

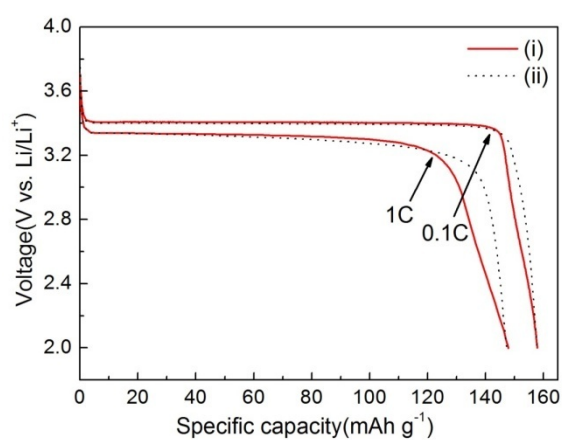


Figure S3. Discharge curves of the al-LFP (i) and the commercial LiFePO_4 product (ii).

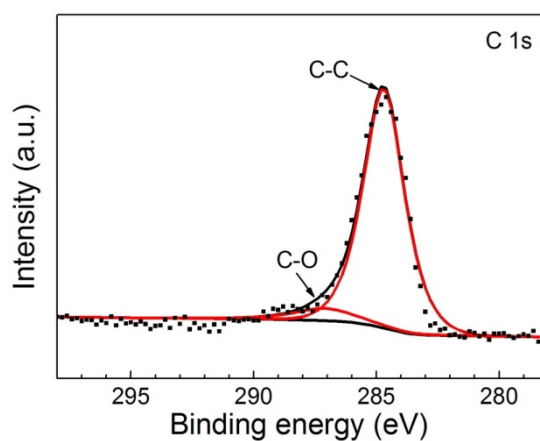


Figure S4. XPS spectra of C1s spectra of the al-LFP sample.

Table S1. Lattice parameters of the al-LFP and aq-LFP samples, and crystallite sizes calculated by the Scherrer's equation.

Samples	al-LFP	aq-LFP
Lattice constant/Å		
$a/\text{Å}$	10.314(4)	10.312(21)
$b/\text{Å}$	6.000(98)	5.999(55)
$c/\text{Å}$	4.689(82)	4.689(89)
$v/\text{Å}^3$	290.27(4)	290.15(7)
Reliability factors/%		
$R_p(\%)$	11.3	11.1
$R_{wp}(\%)$	7.20	7.02
$R_{exp}(\%)$	5.63	5.75
χ^2	1.63	1.49
$R_{Bragg}(\%)$	5.73	4.93
$R_f(\%)$	4.51	4.01

Table S2. Elemental composition of the al-LFP and aq-LFP samples determined using ICP (the other elements were not calculated*).

Sample	Li	Fe	Mg	P
al-LFP	0.97(9)	0.98(4)	0.0093	1
aq-LFP	0.98(7)	0.97(2)	0.0090	1
Recovery rates	1.01	0.987	----	0.985

*Molar ratio was based on element P as 1.