

## SUPPLEMENTARY INFORMATION

### **Adsorption behavior of gluten hydrolysate on mild steel in 1M HCl and its role as a green corrosion inhibitor**

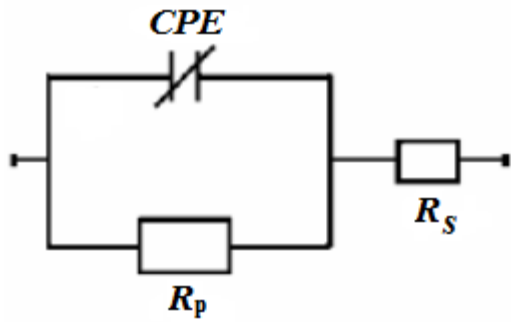
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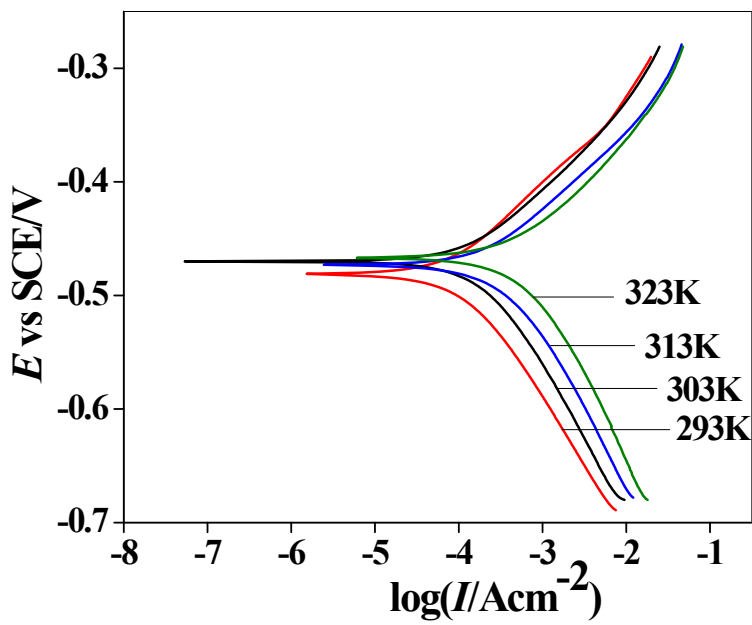
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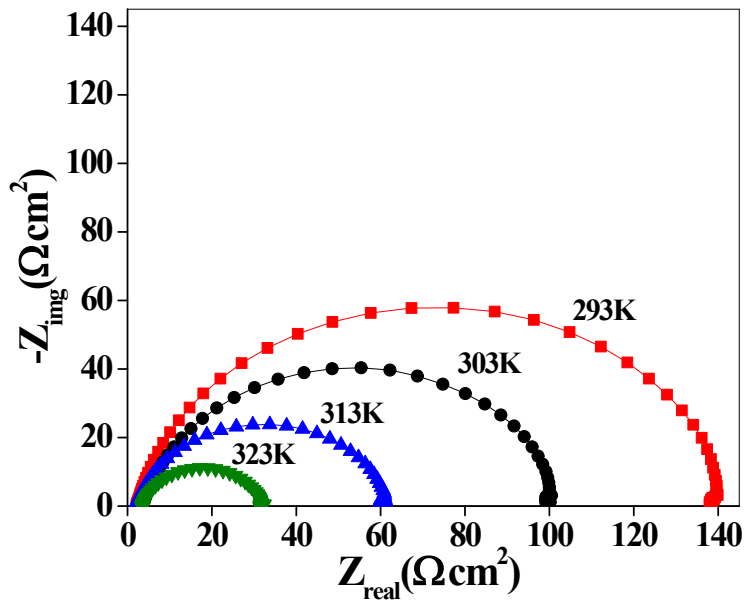
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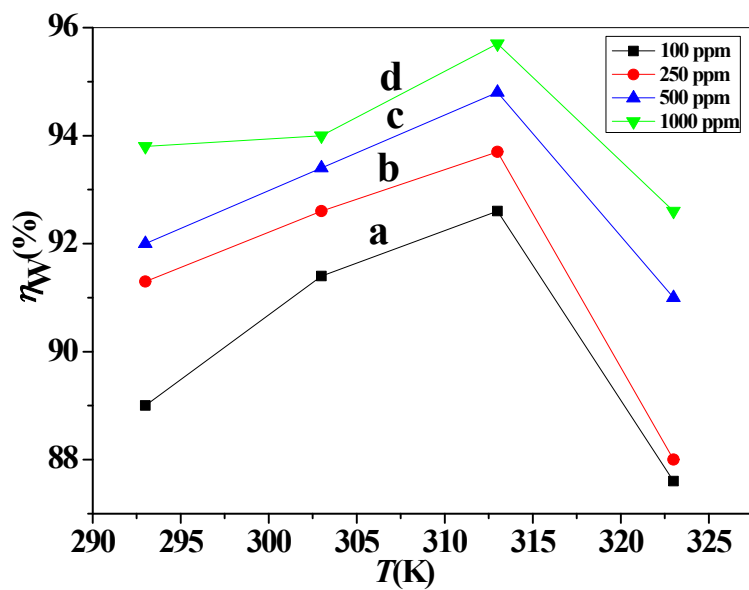
**Fig. S1** Equivalent circuit model used to fit the impedance spectra.



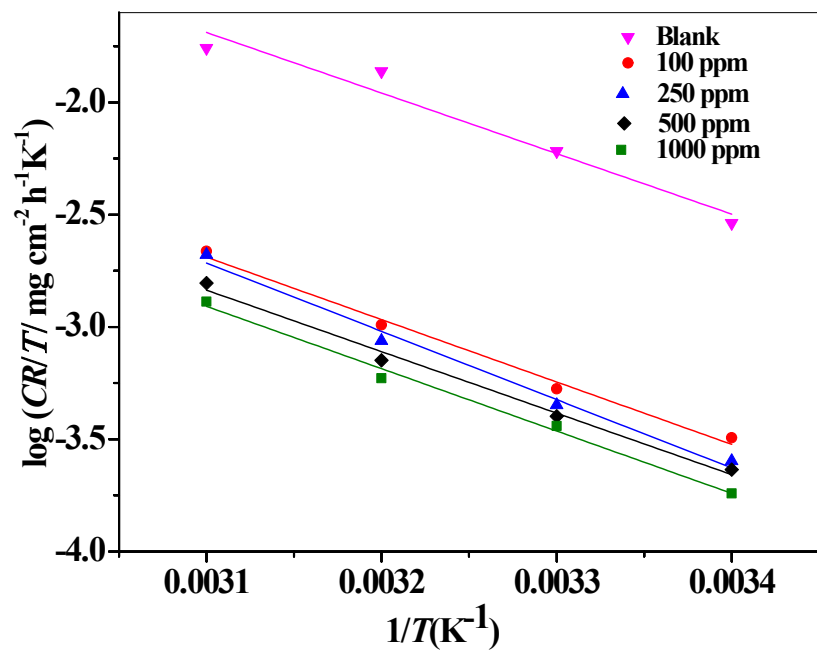
**Fig. S2** Potentiodynamic polarization curves for mild steel in 1 M HCl in presence of 500 ppm gluten hydrolysate at different temperatures.



**Fig. S3** Nyquist plots for mild steel in 1 M HCl in presence of 500 ppm gluten hydrolysate at different temperatures.



**Fig. S4** Variation of inhibition efficiency (from weight loss method) of gluten hydrolysate with temperature towards corrosion of mild steel in 1M HCl.



**Fig. S5** Arrhenius plots of  $\log(CR/T)$  vs.  $1/T$  for mild steel in 1 M HCl without and with different concentrations of gluten hydrolysate.

**Table S1** Corrosion parameters from weight loss measurement for mild steel at different immersion time in 1M HCl.

Temperature (K)	Inhibitor conc. (ppm)	Time (Hour)	CR ((mg cm <sup>-2</sup> h <sup>-1</sup> )	Inhibition efficiency $\eta_w$ (%)
303	BLANK	2	2.5	
		5	1.9	
		24	1.263	
		48	0.785	
		72	0.36	
	250	2	0.23	90.8
		5	0.19	90
		24	0.12	90.5
		48	0.11	85.9
		72	0.103	71.3
	1000	2	0.175	93
		5	0.148	92.2
		24	0.093	92.6
		48	0.073	90.6
		72	0.0727	79.8