

Electronic Supplementary Material

A naked-eye pH-modulated ratiometric photoluminescence sensor based on dual-emission quantum dot@silica nanoparticles for Zn^{2+} and IO_3^-

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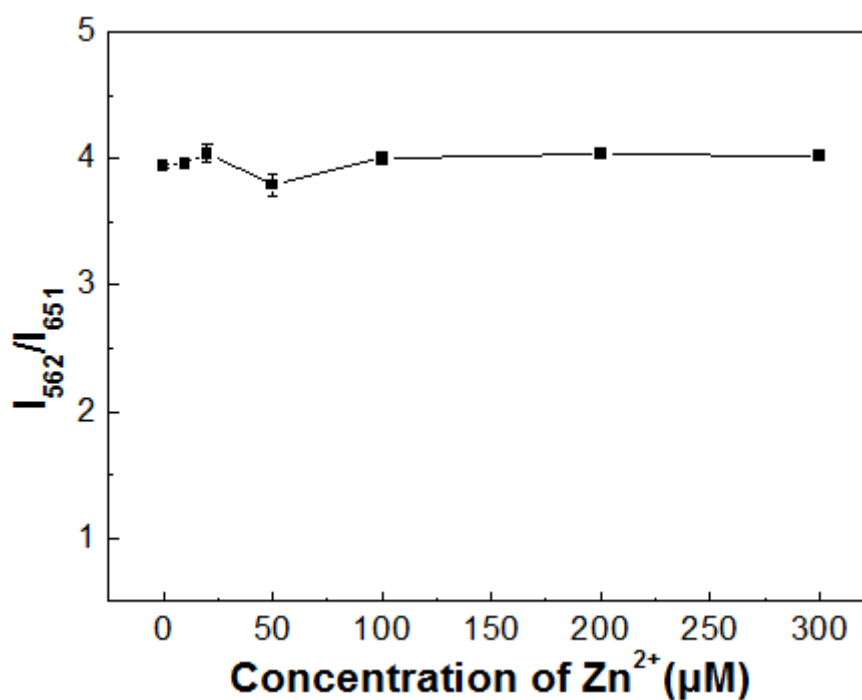


Fig. S1 Effect of Zn^{2+} concentration on the PL intensity ratios I_{562}/I_{651} of QSR.

Table S1 Comparison of different methods for the detection of Zn^{2+} and IO_3^-

Analytes	Method	Linear range (μM)	LOD (μM)	Ref.
Zn^{2+}	Ratiometric PL nanoprobe	1-20	0.5	23
	PL nanosensor based on CuInS_2	5-1000	4.5	35
	Carboxymethyl chitosan-QDs	5-150	4.5	36
	CdSe/ZnS core/shell QDs	5-500	2.4	20
	QDs-based “off-on” PL probe	0.9-16	0.7	37
	QDs-based Ratiometric PL nanoprobe	5-100	1.15	This work
IO_3^-	Electrochemical	20-2000	15	38
	Glassy carbon electrode with iron(III)-porphyrin film	10-4000	2.5	39
	Electrochemical copolymerization	50-500	0.8	40
	Amperometry	2.86-142.86	2.86	41
	QDs-based Ratiometric PL nanoprobe	5-150	1.76	This work

Table S2 The interference of coexisting substances on the detection of Zn^{2+} (20 μM)

Ion /Added as	Tolerable concentration	$\Delta I/I$ (%)
	(μM)	
Na^+ /NaCl	10000	0.5
K^+ /KCl	10000	1.2
Cl^- /NaCl	10000	0.7
NO_3^- / NaNO_3	10000	0.1
Mg^{2+} / $\text{Mg}(\text{NO}_3)_2$	4000	3.4
SO_4^{2-} / Na_2SO_4	4000	2.8
Mn^{2+} / MnCl_2	4000	4.3
Ni^{2+} / $\text{Ni}(\text{NO}_3)_2$	1000	-3.6
Ba^{2+} / BaCl_2	1000	-2.4
Fe^{3+} / FeCl_3	200	1.7
Cd^{2+} / CdCl_2	200	-5.3
Cu^{2+} / CuCl_2	200	2.2

$\Delta I = I_0 - I$, where I_0 and I are the fluorescence intensity of ratiometric PL probe /phen/ Zn^{2+} system in absence and presence of coexisting ions.

Table S3 The interference of coexisting substances on the detection of IO_3^- (20 μM)

Ion /Added as	Tolerable concentration	$\Delta I/I$ (%)
	(μM)	
$\text{Na}^+ / \text{NaCl}$	10000	0.7
K^+ / KCl	10000	0.3
$\text{Cl}^- / \text{NaCl}$	10000	1.1
$\text{Mg}^{2+} / \text{Mg}(\text{NO}_3)_2$	4000	2.5
$\text{NH}_4^+ / (\text{NH}_4)_2\text{SO}_4$	4000	0.2
$\text{NO}_3^- / \text{NaNO}_3$	4000	0.6
$\text{SO}_4^{2-} / \text{Na}_2\text{SO}_4$	4000	1.9
$\text{CO}_3^{2-} / \text{Na}_2\text{CO}_3$	2000	2.7
$\text{HCO}_3^- / \text{NaHCO}_3$	2000	-3.4
$\text{CH}_3\text{COO}^- / \text{CH}_3\text{COONa}$	2000	2.3
F^- / NaF	2000	-3.6
I^- / KI	2000	3.3
$\text{SO}_3^{2-} / \text{NaSO}_3$	100	2.9
$\text{NO}_2^- / \text{NaNO}_2$	100	-3.1
$\text{O}_2^{2-} / \text{H}_2\text{O}_2$	100	2.7

$\Delta I = I_0 - I$, where I_0 and I are the fluorescence intensity of ratiometric PL probe /phen/ IO_3^- system in absence and presence of coexisting ions.