## **Supporting Information**

## Influence of Interface Properties on Charge Density, Band Edge Shifts, and

kinetics of Photoelectrochemical Process in p-Type NiO Photocathode

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**Fig. S1.** XRD film diffraction patterns of the (a) NiO and (b) NiO-Ac films. The NiO peaks are indexed according to JCPDF file 65-2901. The FTO peaks are indexed according to JCPDF file 01-077-0447.



Fig. S2. SEM of (I) surface and (II) cross-section for (A) NiO and (B) NiO-Ac films.



Fig. S3. Results of the XPS analysis: (A) Ni 2p and (B) O 1s spectra for NiO-Ac. Raw

data is given by  $(\bullet)$  and with fitted components (orange lines).



**Fig. S4.** Current-voltage characteristics of NiO and NiO-Ac devices under standardized AM 1.5 illumination of 100 mW/cm<sup>2</sup> (active area 0.25 cm<sup>2</sup>).

Table S1. Comparison of the EIS fitting data for the bare and post-treated NiO

DSSCs.										
sample	$R_{s}\left(\Omega\right)$	$R_{pt}\left(\Omega\right)$	$R_t\left(\Omega\right)$	$R_{rec}\left(\Omega\right)$	N	$C_u$	T <sub>th</sub>	T <sub>h</sub>	L <sub>n</sub>	$\eta_{cc}$
						[mF/cm <sup>2</sup> ]	[ms]	[s]	[um]	
NiO	10.5	0.75	10.7	280	0.93	0.87	9.3	0.24	5.1	0.96
NiO-S	11.2	1.43	11.9	452	0.93	0.86	10.3	0.39	6.1	0.97
NiO-ozone	10.1	0.46	9.2	200	0.94	0.81	7.4	0.16	4.6	0.95

For a more exact fitting, the capacitance element is replaced by a constant phase element (CPE), in which  $Y_0(Q)$  is a constant with the dimension of Siemens  $\sec^{\alpha}$ , and  $\alpha(N)$  is an empirical constant. All the parameters above could achieve by transmission line equivalent circuit. When CPE is in parallel with a resistance, the capacitance can be calculated by the following Equation<sup>1</sup>:

$$C_{\mu} = (QR)^{1/N} / R \tag{S1}$$

After calculated the parameter  $C_{\mu}$ , then the other parameters shown in Table S1 were calculated by using the following Equations<sup>2-4</sup>:

$$\tau_h = R_{rec} C_\mu \tag{S2}$$

$$\tau_{th} = R_t C_\mu \tag{S3}$$

$$L_n = d\sqrt{R_{rec}/R_t} \tag{S4}$$

$$\eta_{cc} = 1 - \tau_{th} / \tau_h \tag{S5}$$

 $C_{\mu}\!\!:$  Chemical capacitance calculated by fitting data

R<sub>s</sub>: Series resistance measured by impedance

R<sub>pt</sub>: Charge-transfer resistance at counter electrode measured by impedance

Rt: Transport resistance of the film measured by impedance

R<sub>rec</sub>: Recombination resistance of the film measured by impedance

L<sub>n</sub>: hole diffusion length

 $\eta_{cc}:$  charge collection efficiency

All the bias in Fig. 5 have been corrected by Equation:

$$V_{corr} = V_{app} - I_{dark} \mathbf{g} \mathbf{R}_{s} + (1/3) \mathbf{R}_{t} + \mathbf{R}_{pt}$$
(S6)

sample	a
NiO	0.024
NiO-S	0.023
NiO-ozone	0.053

Table S2. Comparison of a parameters calculated from  $C_{\mu}$  that obtained by EIS.

The parameter a could be achieved by fitting the plots in Fig. 5C using the exponent Equation (3), which represents for the depth of the trap energy distribution, and the a is an absolute value.

Table S3. Comparison of short-circuit photocurrent density  $(J_{sc})$ , open-circuit photovoltage  $(V_{oc})$ , fill factor (FF), measured under AM 1.5, of p-type DSSCs

	V <sub>oc</sub>	J <sub>sc</sub>	FF	η
sample	[mV]	[mA/cm <sup>2</sup> ]	[%]	[%]
NiO	74	1.18	37.7	$0.032 \pm 0.002$
NiO-S	81	1.13	39.2	$0.036\pm0.002$
NiO-ozone	66	0.72	42.2	$0.020 \pm 0.001$
NiO-Ac	80	1.45	36.2	$0.042 \pm 0.002$

employing bare NiO and post-treated NiO film.

	Ols				Vi2p <sub>3/2</sub>		
sample	NiO	Ni(OH) <sub>2</sub>	NiOOH	NiO	Ni(OH) <sub>2</sub>	NiOOH	
	[eV]	[eV]	[eV]	[eV]	[eV]	[eV]	(NiO:Ni(OH) <sub>2</sub> :NiOOH)
NiO-Ac	529.28	530.94	-	853.71	855.20	856.05	1:2.49:0.90

Table S4. Surface analysis by XPS spectra for the Ni(CH<sub>3</sub>COOH)<sub>2</sub> treated NiO.

## **Notes and References**

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