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Supporting information

pH-modulated double LCST behaviors with diverse aggregation processes of random-copolymer grafted silica nanoparticles in aqueous solution

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The detailed formulations in the ATRP reaction are presented in Table S1.

	Table 51. Feed Compositions of ATAT Teaction						
East Composition	SiO ₂ -g-P(NIPAM-co-DMAEMA) Sample Code						
reed Composition	SiO ₂ -N50D50	SiO ₂ -N83D17	SiO ₂ -N33D67	SiO ₂ -N67D33	SiO ₂ -D100		
SiO ₂ -Br (mg)	300	300	300	300	300		
CuCl (mg)	25	25	25	25			
bpy (mg)	42.7	42.7	42.7	42.7			
MeOH/H ₂ O (1:1) (mL)	4	4	4	4			
CuBr (mg)					36		
PMDETA (µL)					52		
Anisole (mL)					4		
NIPAM (g)	0.312	0.531	0.192	0.384			
DMAEMA (mL)	0.464	0.155	0.572	0.286	0.928		

Table S1. Feed Compositions of ATRP reaction

Table S2. The atomic content analysis of XPS

Samplas	Atomic content (%)					
Samples	Br	Si	С	Ν	0	
SiO ₂ -Br	0.52	24.54	25.39	3.01	46.53	
SiO ₂ -N50D50	0	1.43	73.30	7.75	17.53	
(theoretical value)	(0)	(0)	(73.68)	(10.53)	(15.79)	
SiO ₂ -N83D17	0	0.79	71.95	8.24	19.02	
(theoretical value)	(0)	(0)	(74.51)	(11.76)	(13.73)	
SiO ₂ -D100	0	3.26	66.86	8.01	21.86	
(theoretical value)	(0)	(0)	(72.73)	(9.09)	(18.18)	

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The XPS data can also be used to roughly confirm the composition of grafted polymer layers. The elemental content analysis is shown in Table S2. As for SiO₂-Br, a high atomic content of O was observed due to the APTES and hydroxyl group on the silica surface. The mole ratio of C and N was $25.39:3.01\approx8:1$, which was approximately consistent with the theoretical value of 7:1. Furthermore, the content of Br was much less than that of N, indicating the quenching and decreasing of initiator in the post-processing. For PNIPAM and PDMAEMA grafted silica nanoparticles, the Si contents dramatically decreased, indicating the covering of the silica cores by a polymer layer. The experimental contents of different atoms in synthesized samples were similar to those theoretical ones, which suggested that the ratios of two grafted monomers were consistent with the corresponding feeding ones.



Fig. S1. pH-dependent hydrodynamic diameters of SiO₂-N67D33 at 25 °C.

_	LCSTs/°C				
_	Acidic	Neutral		Basic	
	PNIPAM/PDMAEMA	PNIPAM	PDMAEMA	PNIPAM	PDMAEMA
SiO ₂ -N50D50	65±1	42±2	52±1	38±1	59±1
SiO ₂ -N83D17	64±1	42±2	56±2	38±1	57±1

1 able 55. LC515 of SiO ₂ -g-P(NIPANI-co-DMAEMA) from DLS result	Table	e S3. L(CSTs of SiO	2-g-P(NIPAM-	-co-DMAEMA`) from D	LS result
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