

ELECTRONIC SUPPORTING INFORMATION (ESI)

The Non-Innocent Nature of Graphene Oxide as Theranostic Platform for Biomedical Applications and its Reactivity towards Metal-Based Anticancer Drugs

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Theoretical calculations (geometry optimization Fe-Pz₂Py)

Input File (Spartan 14) for the calculation of the geometry optimized structure of Fe-Pz₂Py (charge = +2, multiplicity = 5)

```
MEM_TOTAL 2000 # MB [ 4 processes ]
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(Site specific preferences)
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... SCF_ALGORITHM DIIS_GDM
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... THRESHDIIS -1 (i.e. don't switch on delta-E)
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... GUI GUI_SPARTAN
... TERSE_OUTPUT TRUE
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... EXCHANGE M06
... CORRELATION none (built-in)
... INCDFE FALSE (N/A in MetaGGA)
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... MAXSCF 1000
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... VARTHRESH 2 (default DFT)
... BASIS LACVP*
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Coordinate (xyz) of the Fe-Pz₂Py complex from optimized geometry

		Coordinates (Angstroms)		
ATOM		X	Y	Z
1	H	0.000000	0.000000	4.752645
2	C	0.000000	0.000000	3.665240
3	N	0.000000	0.000000	0.929633
4	C	-1.213099	0.001082	2.982360
5	C	1.213099	-0.001082	2.982360
6	C	1.155745	-0.001370	1.596865
7	C	-1.155745	0.001370	1.596865
8	H	-2.154863	0.001366	3.522046
9	H	2.154863	-0.001366	3.522046
10	N	-2.284786	0.002812	0.763129
11	N	-2.102030	-0.002947	-0.589849
12	C	-3.324671	0.001178	-1.109418
13	C	-4.310481	0.009849	-0.106597
14	C	-3.613793	0.010761	1.073677
15	H	-3.453156	-0.001888	-2.186249
16	H	-5.384785	0.015066	-0.232088
17	H	-3.961376	0.016654	2.099078
18	N	2.284786	-0.002812	0.763129
19	C	3.613793	-0.010761	1.073677
20	C	4.310481	-0.009849	-0.106597
21	C	3.324671	-0.001178	-1.109418
22	N	2.102030	0.002947	-0.589849

23	H	3.961376	-0.016654	2.099078
24	H	5.384785	-0.015066	-0.232088
25	H	3.453156	0.001888	-2.186249
26	Fe	0.000000	0.000000	-1.208249
27	O	-0.013334	2.152433	-1.526782
28	H	0.759561	2.711368	-1.352350
29	H	-0.795144	2.698706	-1.351748
30	O	0.000000	0.000000	-3.291293
31	H	0.007136	-0.789511	-3.854483
32	H	-0.007136	0.789511	-3.854483
33	O	0.013334	-2.152433	-1.526782
34	H	0.795144	-2.698706	-1.351748
35	H	-0.759561	-2.711368	-1.352350

Point Group: c2 Number of degrees of freedom: 48
Energy = -1050.057740208 au

Figure S1. The optimized structure of Fe-Pz₂Py with the relevant Fe-N/O distances, spin density isosurface and Mulliken atomic charges as calculated by DFT UM06/6-31G*/LanL2DZ.

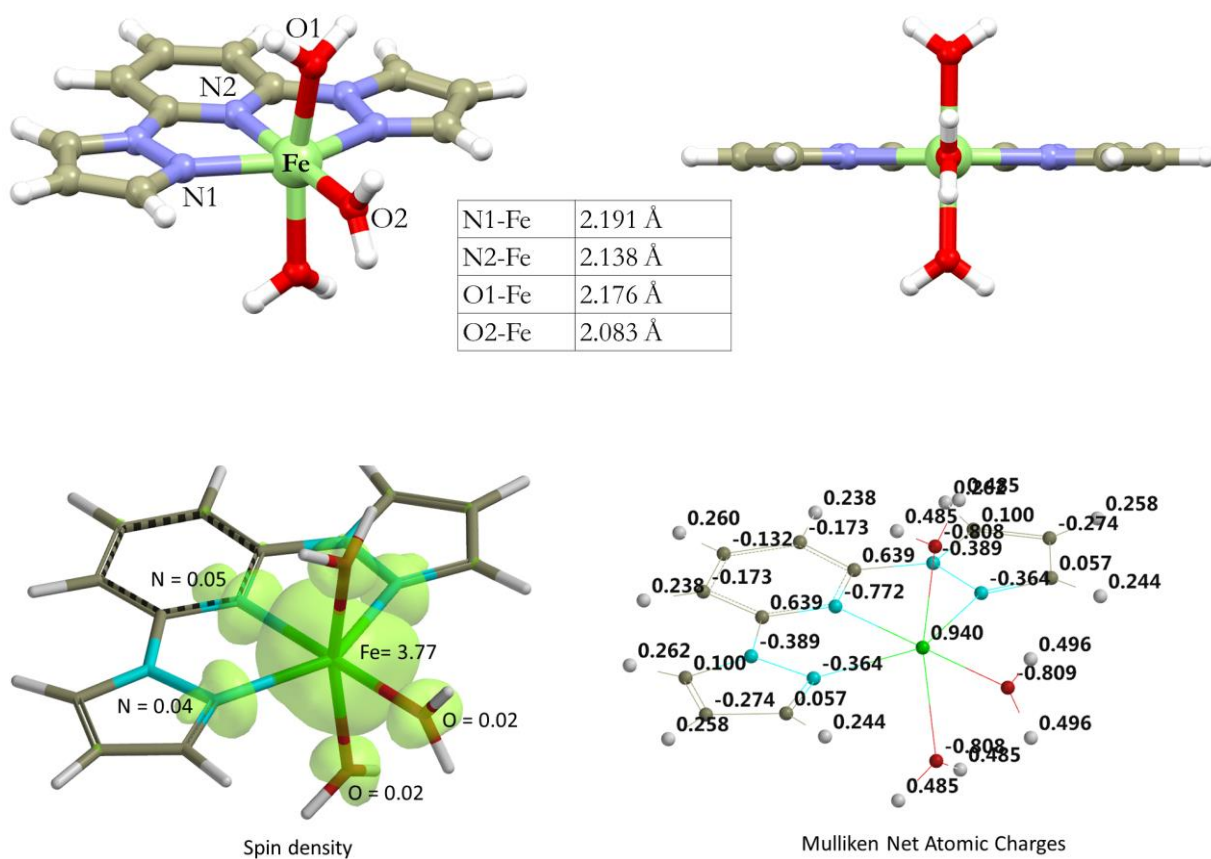


Figure S2. Mossbauer spectra of Fe-Pz₂Py (powder) recorded at $T = 300$ K (a) and $T = 5$ K (b) validating the presence in the complex of only Fe(II) cation in high spin ($S = 2$) state.

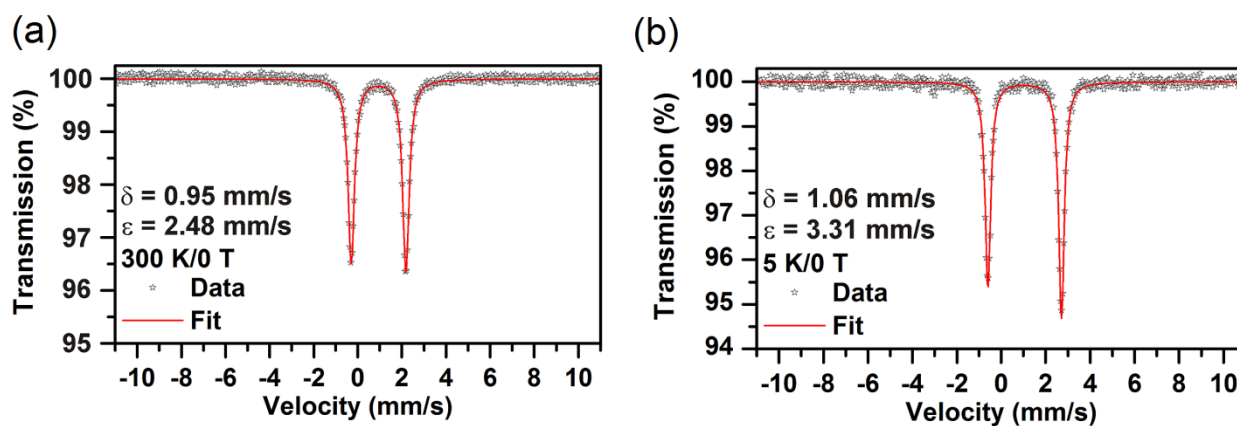


Figure S3. FT-IR spectra of GO (a), Fe-Pz₂Py/GO (b) and Fe-Pz₂Py (c).

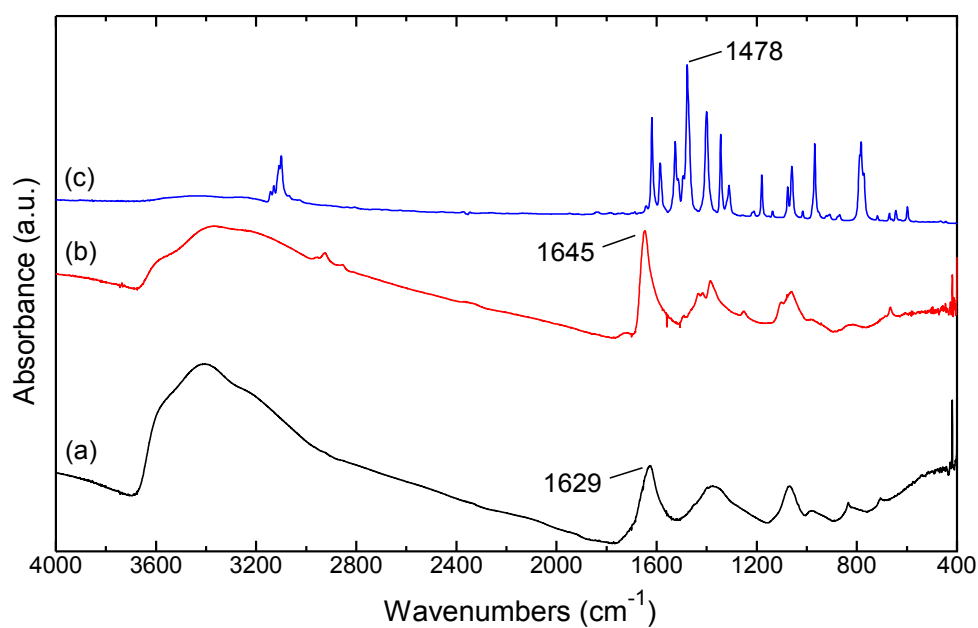


Figure S4. Raman spectra of GO (a), and Fe-Pz₂Py/GO (b)

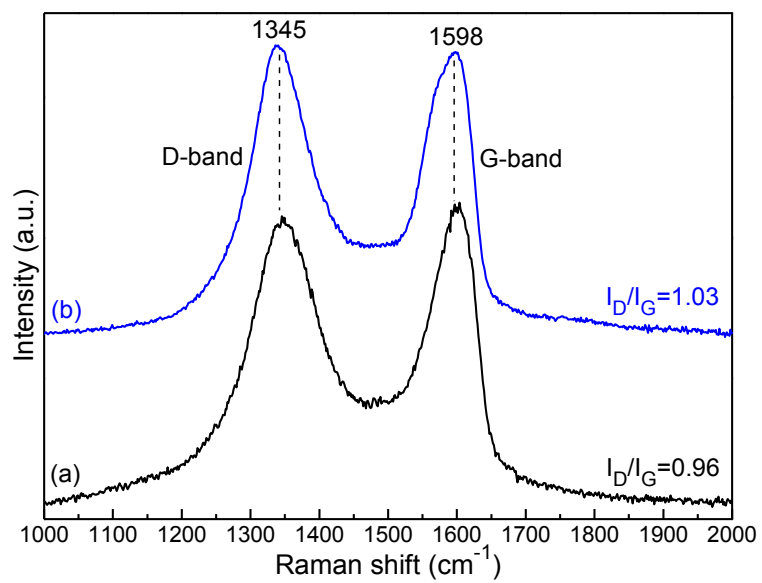
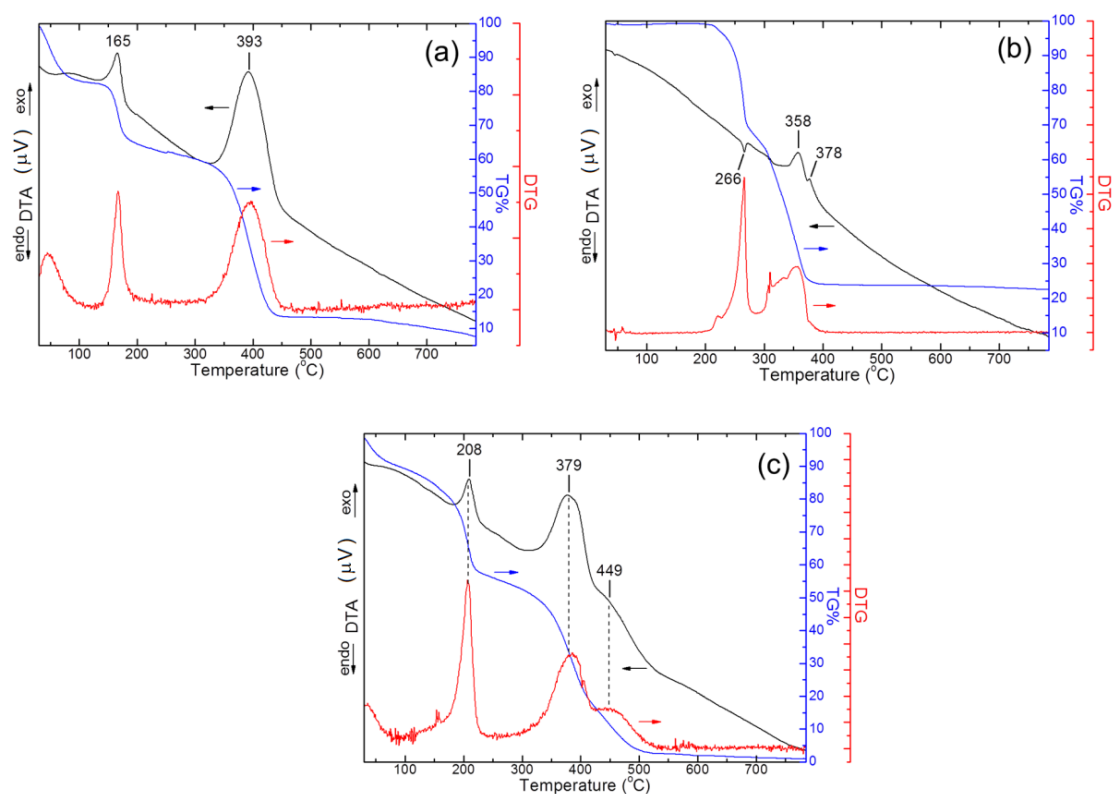


Figure S5. DTA/TG%/DTG curves of GO (a), Fe-Pz₂Py (b) and Fe-Pz₂Py/GO (c)



Comparison of all DTA/TG%/DTG curves of GO, Fe-Pz₂Py and Fe-Pz₂Py/GO clearly confirms the formation of the hybrids system (Fe-Pz₂Py/GO). GO materials exhibit two exothermic peaks; one at around 160-210 °C and one at ~400 °C (Panel a), that are ascribed to their decomposition and combustion respectively. The hybrid material demonstrates additional exothermic peaks which are associated with the Fe-Pz₂Py content and its combustion (Panel c). All exothermic peaks are accompanied by corresponding weight losses which are better displayed by the DTG signals. Moreover, in the temperature region ~130-550 °C which includes all decomposing and combustion of organic/inorganic carbon-based content of the materials, Fe-Pz₂Py/GO losses 85% of its weight, and GO just 69%. This difference is associated with the Fe-Pz₂Py presence (Panel b). Thus, while combustion peaks of pristine GO and Fe-Pz₂Py are overlapping and the exact Fe-Pz₂Py content cannot be exactly calculated, an estimation from the TG% signal leads to ~16 wt% Fe-Pz₂Py content for Fe-Pz₂Py/GO. This amount of loaded Fe-Pz₂Py has been further validated by atomic absorption spectroscopy following analysis of the Fe cation (16.3 ± 0.3 wt% of Fe-Pz₂Py in the hybrid). After drying the sample Fe-Pz₂Py/GO being kept in water for weeks no significant signal shifts or changes in the DTA/TG%/DTG relative intensity were observed, indicating a highly stable material, both as dry powder and in solution, even after prolonged storage.

Figure S6. The N1s core level X-ray photoemission spectra of sample Fe-Pz₂Py/GO.

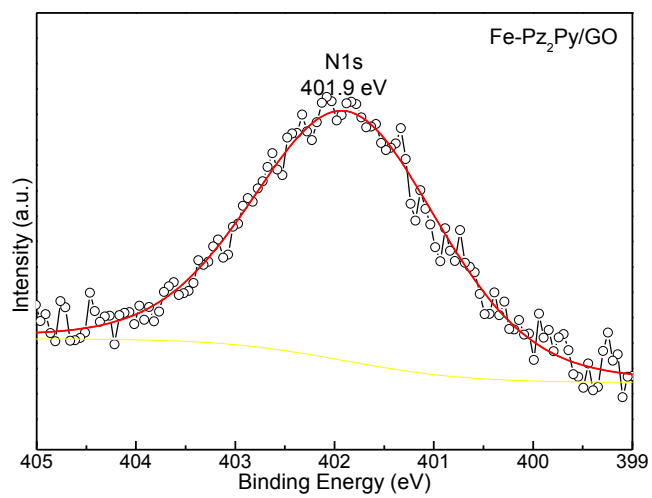


Figure S7. The SEM and TEM micrographs of the drug loaded GO composite (Fe-Pz₂Py/GO).

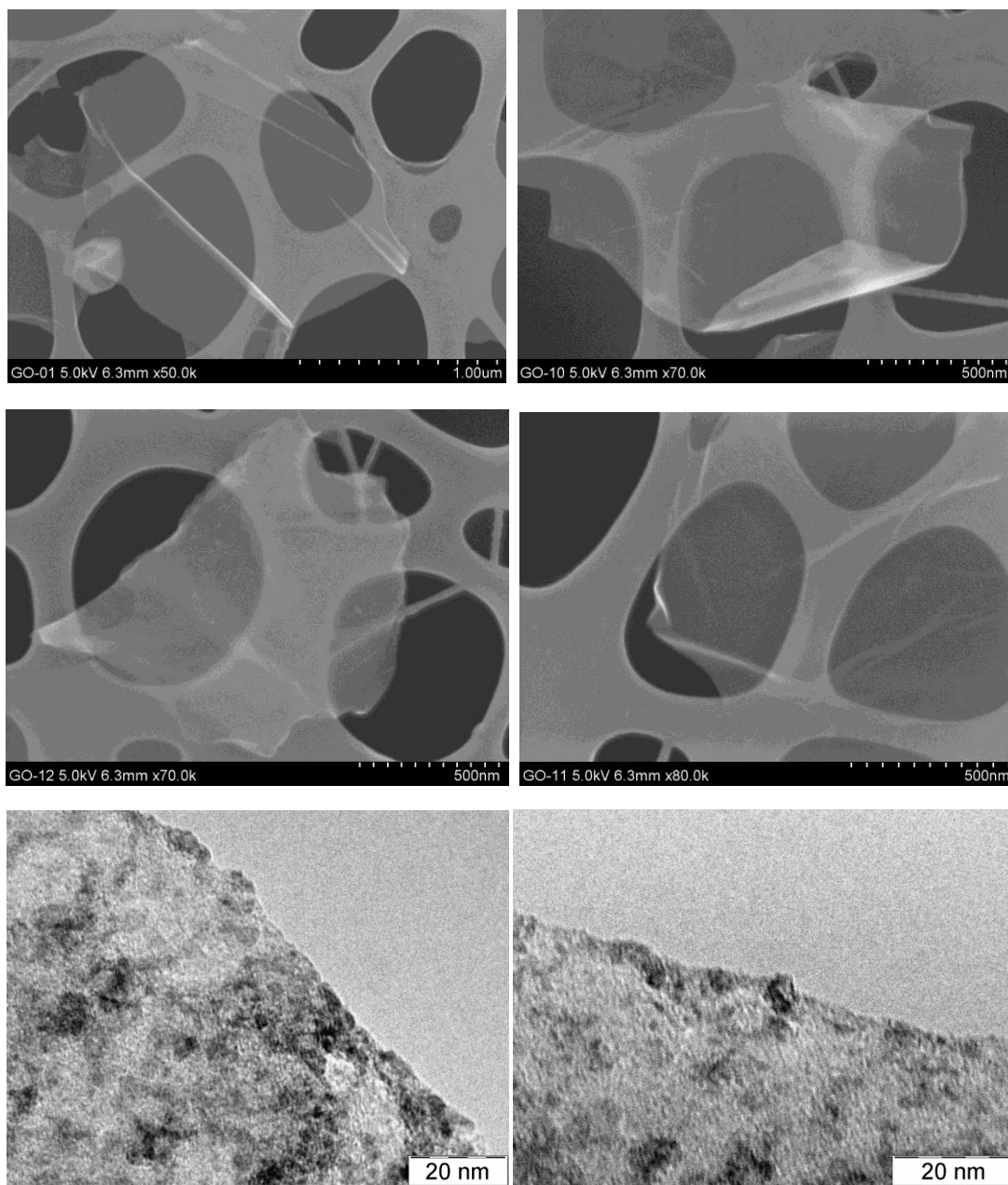


Figure S8. The UV-Vis spectrum of Fe-Pz₂Py in water (0.03 mM) (b) and the calculated absorption envelope (a) by TDDFT-UEDF2/6-31G** in vacuum (excited states = 30, frozen cores = 28).

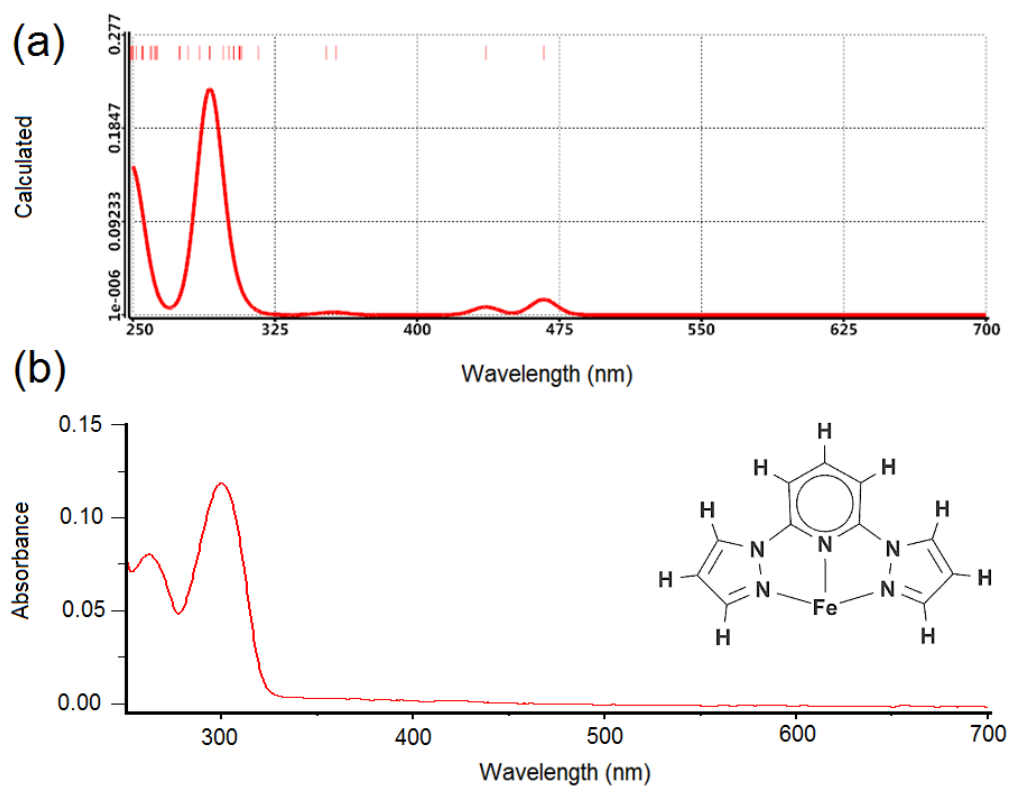
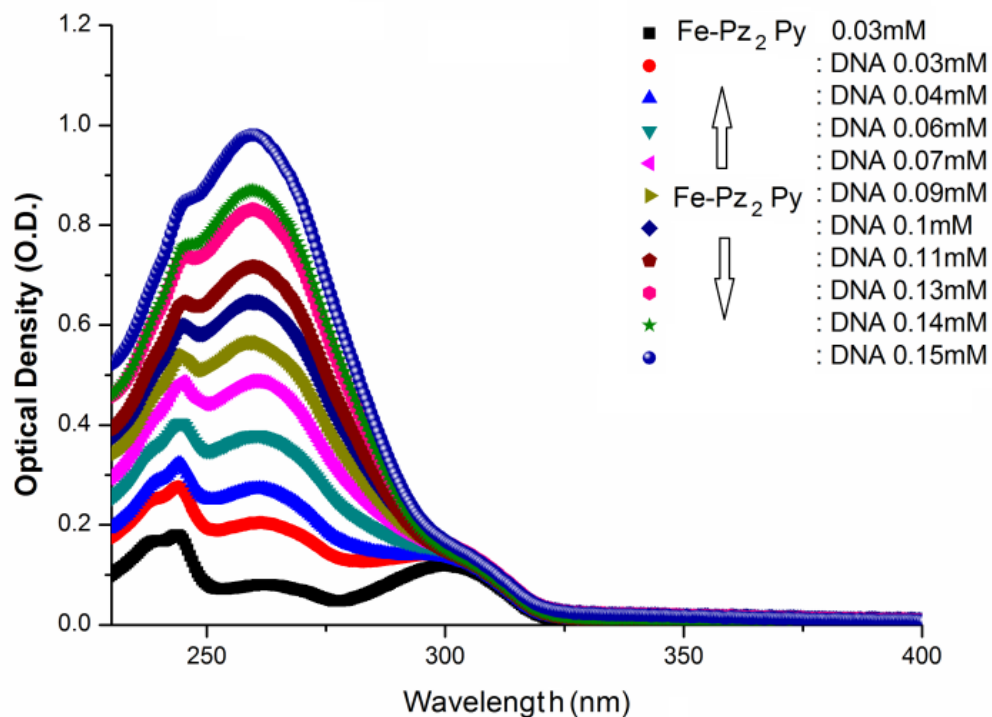
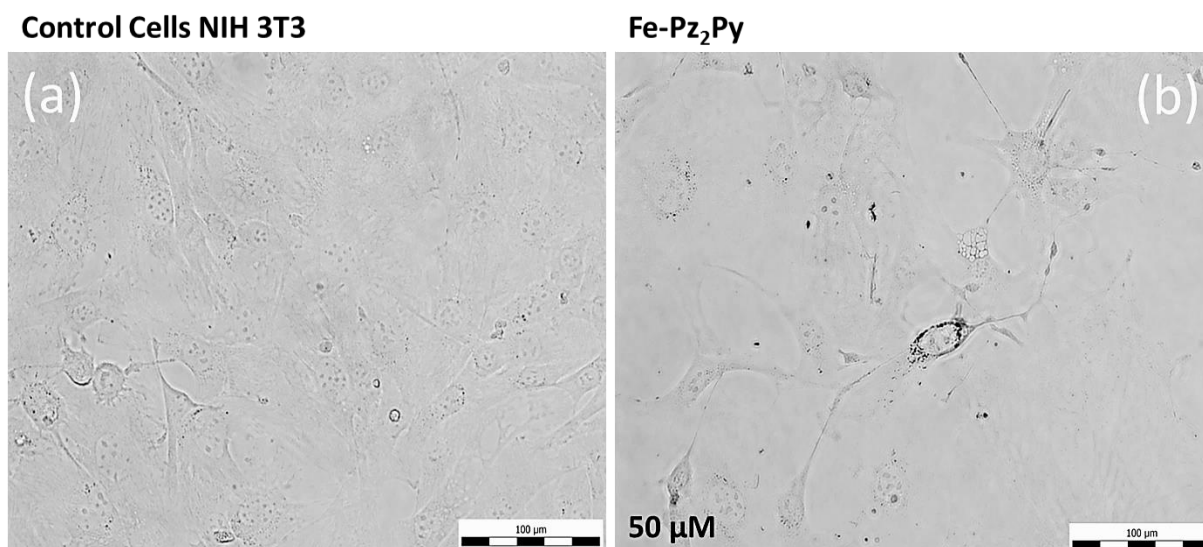


Figure S9. Steady-state optical absorption spectra of Fe-Pz₂Py in water in presence of increasing amounts of DNA showing the absorbance changes with the increase of DNA concentration.



The optical spectrum of the Fe-Pz₂Py complex (black line) displays transitions centered at ~ 238nm, 243nm, 262nm, 300nm and between 350-400 nm. The 475 nm band likely arises from a charge transfer transition between the iron and the nitrogen-ligands of the Pz₂Py backbone.

Figure S10. Confocal microscopy images of the mouse fibroblast NIH 3T3 cells culture without (a) and with 50 μ M (IC₅₀) of Fe-Pz₂Py (b). Cells were maintained at 37°C with 5% humidified CO₂ in an incubation chamber for 24 hours. Images were collected using an optical microscope (Olympus IX 70) with an objective of 40 \times .



NOTE:

It is interesting to report also that when the cell line NIH 3T3 (mouse fibroblasts) was used in the cytotoxicity screening for Fe-Pz₂Py, the calculated IC₅₀ decreased considerably, down to 50 μ M, showing that the L929 cells are less affected to the drug exposure. Kamatchi et al. demonstrated that the IC₅₀ for NIH 3T3 cells equals to 177 μ M for cisplatin and 109 μ M to 248 μ M for different types or ruthenium complexes containing bidentate carboxylates ligands, hence Fe-Pz₂Py is more toxic than cis-Pt and Ru-based drugs against the NIH 3T3 cell's line.

Reference

T. S. Kamatchi, N. Chitrapriya, H. Lee, C. F. Fronczek, F. R. Fronczek, K. Natarajan, *Dalton Trans.*, 2012, **41**, 2066-2077.

Figure S11. Fluorescence quenching spectra of Fe-Pz₂Py at different concentrations of base pairs at 298 K; Fe-Pz₂Py = 0.03 mM, Base pairs from 0.015 mM to 0.15 mM (E_{ex} = 295 nm) with (a) Adenine, (b) Thymine, (c) Guanine and (d) Cytosine. The arrows in (b) and (d) indicate the fluorescence quenching process.

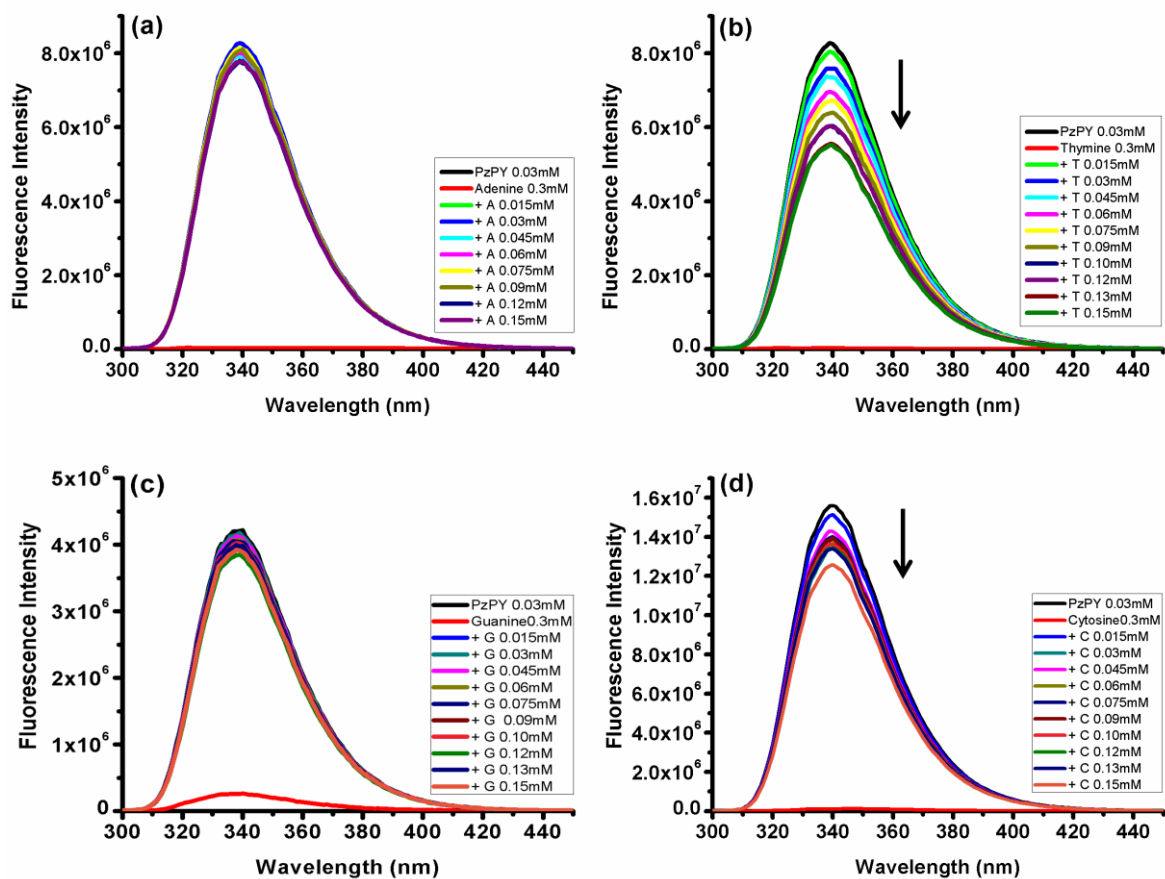


Figure S12. Stern-Volmer plot of the binding of Fe-Pz₂Py to DNA (a), thymine (b) and cytosine (c) recorded at room temperature. Plot of $\log[(F_0-F)/F]$ vs \log [Substrate] for the binding of Fe-Pz₂Py to DNA (d), thymine (e) and cytosine (f). The Table in the bottom shows a summary of the constant K_q , K_{sv} , K and n for DNA, thymine and cytosine.

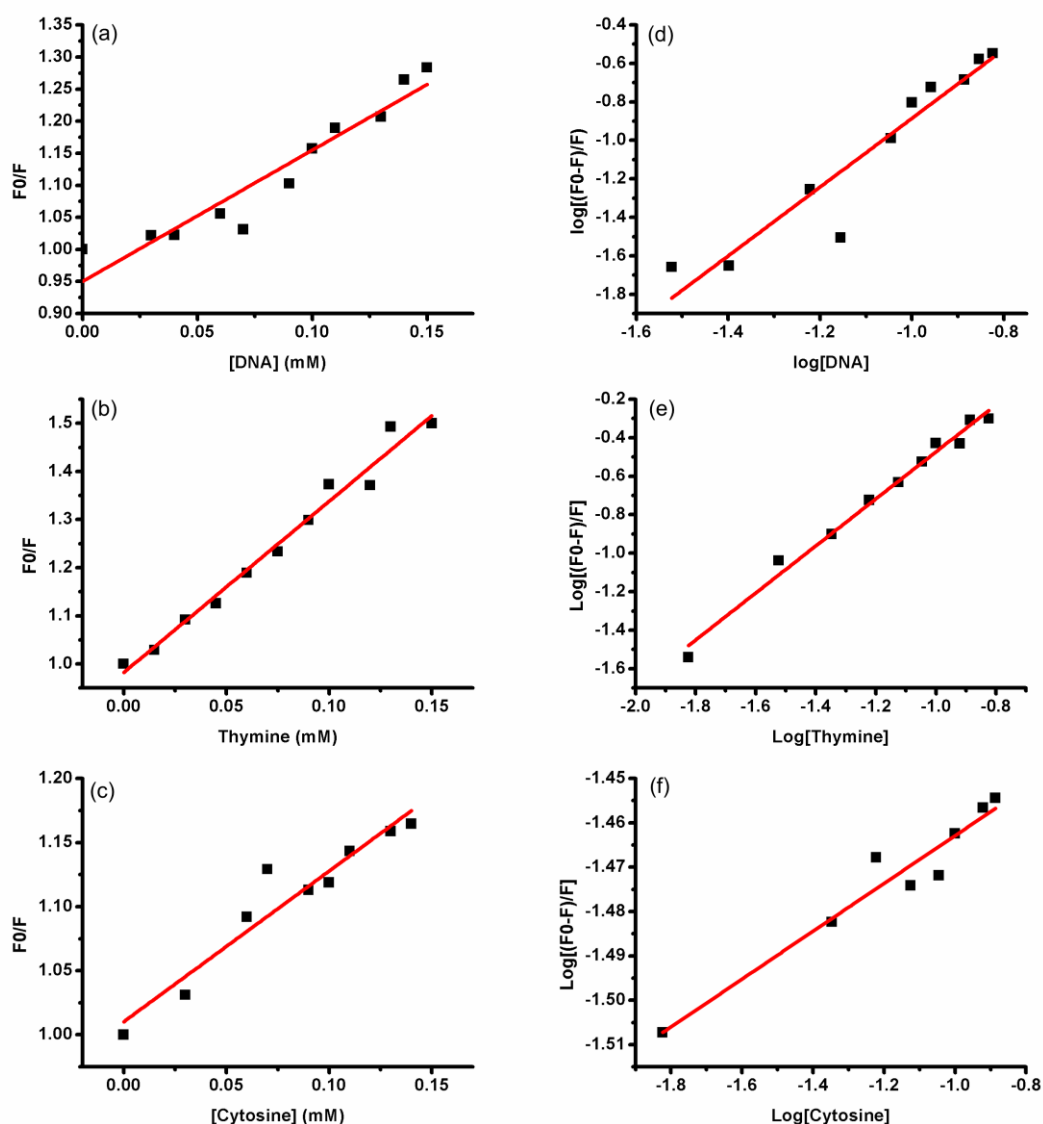


Table S1

	K_q ($M^{-1}s^{-1}$)	K_{sv} (M^{-1})	K (M^{-1})	n
DNA	2.0×10^{11}	2.0×10^3	7.9 ± 1.6	1.8
Thymine	7.2×10^{12}	3.6×10^3	5.0 ± 1.1	1.2
Cytosine	2.4×10^{12}	1.2×10^3	0.04 ± 1	0.05

Note: Using the following equation, the formation of complex between Fe-Pz₂Py and DNA, thymine or cytosine can be confirmed from the values of quenching rate constant K_q :

$$K_q = K_{sv}/\tau_0$$

where τ_0 is the average lifetime of the molecule without the quencher. For various quenchers, the maximum scatter collision quenching constant, K_q , with biopolymer is $2 \times 10^{10} M^{-1}s^{-1}$. If we assume the lifetime (τ_0) ($10^{-12} s^{-1}$) of the calf thymus, which is equal to 10 ps, is similar to the salmon sperm DNA, then we can extract K_q .

Theoretical calculations (UV-Vis of Fe-Pz₂Py)

Input File (Spartan 14) for the calculation of the UV-Vis spectrum of Fe-Pz₂Py (charge = +2, multiplicity = 5) as shown in Figure S7.

Job type: Single point.

Excited States: 30

Method: UEDF2(FC)

Basis set: 6-31G**

Number of shells: 128

Number of basis functions: 389

Charge : +2

Multiplicity: 5

Processing \$rem in C:/Program Files/Wavefunction/Spartan14v100/P4e/../../auxdir/config/preferences.

(Site specific preferences)

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... SCF_ALGORITHM    DIIS_GDM
... MAXSCF           100
... MAXDIIS          50
... THRESHDIIS       -1 (i.e. don't switch on delta-E)
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... GUI              GUI_SPARTAN
... TERSE_OUTPUT     TRUE
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Processing \$rem in input file

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... VARTHRESH         2 (default DFT)
... INCDFT            TRUE (default DFT)
... GUI              GUI_SPARTAN
... TERSE_OUTPUT     TRUE
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TDDFT/TDA Excitation Energies

Excited state 1: excitation energy (eV) = 0.0391
Total energy for state 1: -2190.442481939248
<S**2> : 6.0110
Trans. Mom.: 0.1355 X 0.0013 Y 0.0000 Z
Strength : 0.0000
D(52) --> S(1) amplitude = 0.8220 beta
D(52) --> S(4) amplitude = 0.5426 beta

Excited state 2: excitation energy (eV) = 0.1262
Total energy for state 2: -2190.439281151341
<S**2> : 6.0072
Trans. Mom.: 0.0000 X 0.0000 Y 0.0035 Z
Strength : 0.0000
D(52) --> S(3) amplitude = 0.9883 beta

Excited state 3: excitation energy (eV) = 0.9847
Total energy for state 3: -2190.407733138254
<S**2> : 6.0061
Trans. Mom.: -0.0024 X 0.0104 Y 0.0000 Z
Strength : 0.0000
D(52) --> V(1) amplitude = 0.9631 beta
D(52) --> V(4) amplitude = 0.2099 beta

Excited state 4: excitation energy (eV) = 1.4444
Total energy for state 4: -2190.390838883862
<S**2> : 6.0077
Trans. Mom.: 0.0021 X 0.0059 Y 0.0000 Z
Strength : 0.0000
D(52) --> V(1) amplitude = -0.2189 beta
D(52) --> V(2) amplitude = -0.5845 beta
D(52) --> V(4) amplitude = 0.7228 beta
D(52) --> V(7) amplitude = 0.2655 beta

Excited state 5: excitation energy (eV) = 2.6563
Total energy for state 5: -2190.346301921167
<S**2> : 7.1826
Trans. Mom.: 0.0000 X 0.0000 Y 0.4812 Z
Strength : 0.0151
D(52) --> S(2) amplitude = 0.9733 beta

Excited state 6: excitation energy (eV) = 2.8418
Total energy for state 6: -2190.339484954966
<S**2> : 7.0280
Trans. Mom.: -0.3322 X -0.0006 Y 0.0000 Z
Strength : 0.0077
D(52) --> S(1) amplitude = -0.5546 beta
D(52) --> S(4) amplitude = 0.8213 beta

Excited state 7: excitation energy (eV) = 3.4719
Total energy for state 7: -2190.316328029332
<S**2> : 7.7851
Trans. Mom.: 0.0000 X 0.0000 Y -0.1419 Z
Strength : 0.0017
S(4) --> V(2) amplitude = -0.6091 alpha
D(51) --> S(1) amplitude = 0.6531 beta
D(51) --> S(4) amplitude = -0.3240 beta

Excited state 8: excitation energy (eV) = 3.5206

Total energy for state 8: -2190.314538389789
<S**2> : 7.9817
Trans. Mom.: -0.1036 X -0.0001 Y 0.0000 Z
Strength : 0.0009
S(4) --> V(1) amplitude = 0.7053 alpha
D(51) --> S(2) amplitude = 0.6670 beta

Excited state 9: excitation energy (eV) = 3.9212
Total energy for state 9: -2190.299818806492
<S**2> : 6.2670
Trans. Mom.: 0.0005 X -0.0689 Y 0.0000 Z
Strength : 0.0005
S(3) --> V(1) amplitude = 0.9925 alpha

Excited state 10: excitation energy (eV) = 4.0335
Total energy for state 10: -2190.295692020174
<S**2> : 7.7892
Trans. Mom.: 0.0000 X 0.0000 Y -0.0793 Z
Strength : 0.0006
S(1) --> V(2) amplitude = 0.3795 alpha
S(2) --> V(1) amplitude = -0.4473 alpha
D(49) --> S(1) amplitude = 0.3934 beta
D(49) --> S(4) amplitude = -0.1720 beta
D(50) --> S(2) amplitude = 0.4524 beta
D(52) --> V(3) amplitude = -0.2941 beta

Excited state 11: excitation energy (eV) = 4.0457
Total energy for state 11: -2190.295241064957
<S**2> : 7.7855
Trans. Mom.: 0.0372 X -0.0030 Y 0.0000 Z
Strength : 0.0001
D(52) --> V(2) amplitude = 0.1641 alpha
S(1) --> V(1) amplitude = -0.4798 alpha
S(1) --> V(3) amplitude = 0.1624 alpha
S(2) --> V(2) amplitude = 0.3586 alpha
D(48) --> S(1) amplitude = 0.1730 beta
D(49) --> S(2) amplitude = 0.3885 beta
D(49) --> V(3) amplitude = -0.1603 beta
D(50) --> S(1) amplitude = 0.5170 beta
D(50) --> S(4) amplitude = -0.1833 beta

Excited state 12: excitation energy (eV) = 4.0517
Total energy for state 12: -2190.295021136484
<S**2> : 7.0286
Trans. Mom.: -0.0009 X -0.0898 Y 0.0000 Z
Strength : 0.0008
D(52) --> V(2) amplitude = 0.7963 beta
D(52) --> V(4) amplitude = 0.5883 beta

Excited state 13: excitation energy (eV) = 4.0878
Total energy for state 13: -2190.293695368173
<S**2> : 6.3860
Trans. Mom.: 0.0000 X 0.0000 Y -0.3844 Z
Strength : 0.0148
S(4) --> V(2) amplitude = 0.3528 alpha
D(51) --> S(1) amplitude = 0.6725 beta
D(51) --> S(4) amplitude = 0.5922 beta

Excited state 14: excitation energy (eV) = 4.1230
Total energy for state 14: -2190.292400956618
<S**2> : 6.0463
Trans. Mom.: 0.0781 X 0.0056 Y 0.0000 Z

Strength : 0.0006
D(51) --> S(3) amplitude = 0.9925 beta

Excited state 15: excitation energy (eV) = 4.1639
Total energy for state 15: -2190.290898737517
<S**2> : 7.1130
Trans. Mom.: 0.0000 X 0.0000 Y 0.0623 Z
Strength : 0.0004
S(2) --> V(1) amplitude = -0.1513 alpha
D(52) --> V(3) amplitude = 0.9317 beta

Excited state 16: excitation energy (eV) = 4.2679
Total energy for state 16: -2190.287075471239
<S**2> : 6.0446
Trans. Mom.: 1.4522 X 0.0022 Y 0.0000 Z
Strength : 0.2205
S(4) --> V(1) amplitude = -0.6557 alpha
D(51) --> S(2) amplitude = 0.6731 beta

Excited state 17: excitation energy (eV) = 4.3489
Total energy for state 17: -2190.284101578425
<S**2> : 6.2351
Trans. Mom.: 0.0000 X 0.0000 Y -0.0030 Z
Strength : 0.0000
S(3) --> V(2) amplitude = 0.9946 alpha

Excited state 18: excitation energy (eV) = 4.4406
Total energy for state 18: -2190.280728831889
<S**2> : 7.6551
Trans. Mom.: 0.0000 X 0.0000 Y 0.0213 Z
Strength : 0.0000
D(49) --> V(1) amplitude = 0.1803 alpha
D(50) --> V(1) amplitude = -0.1554 alpha
D(52) --> V(1) amplitude = 0.4713 alpha
S(1) --> V(2) amplitude = -0.1734 alpha
S(2) --> V(1) amplitude = -0.4502 alpha
D(47) --> S(2) amplitude = -0.2081 beta
D(48) --> S(2) amplitude = -0.3051 beta
D(49) --> S(1) amplitude = -0.3271 beta
D(50) --> S(2) amplitude = 0.3286 beta
D(51) --> S(4) amplitude = -0.1840 beta

Excited state 19: excitation energy (eV) = 4.5145
Total energy for state 19: -2190.278014781491
<S**2> : 7.0745
Trans. Mom.: 0.1570 X 0.0002 Y 0.0000 Z
Strength : 0.0027
D(50) --> V(2) amplitude = 0.1841 alpha
S(1) --> V(1) amplitude = 0.2508 alpha
S(4) --> V(3) amplitude = 0.2337 alpha
D(48) --> S(1) amplitude = -0.3506 beta
D(49) --> S(2) amplitude = -0.2277 beta
D(50) --> S(1) amplitude = 0.6822 beta
D(51) --> V(3) amplitude = 0.2072 beta

Excited state 20: excitation energy (eV) = 4.7168
Total energy for state 20: -2190.270581125494
<S**2> : 6.0450
Trans. Mom.: 0.0000 X 0.0000 Y -0.0005 Z
Strength : 0.0000
D(50) --> S(3) amplitude = 0.7568 beta
D(51) --> V(1) amplitude = -0.6430 beta

Excited state 21: excitation energy (eV) = 4.7272
Total energy for state 21: -2190.270198399535
<S**2> : 6.0453
Trans. Mom.: 0.0000 X 0.0000 Y 0.0266 Z
Strength : 0.0001
D(50) --> S(3) amplitude = 0.6373 beta
D(51) --> V(1) amplitude = 0.7637 beta

Excited state 22: excitation energy (eV) = 4.7402
Total energy for state 22: -2190.269721830728
<S**2> : 6.9799
Trans. Mom.: -0.0102 X -0.0018 Y 0.0000 Z
Strength : 0.0000
D(50) --> V(2) amplitude = -0.1784 alpha
S(2) --> V(2) amplitude = -0.4565 alpha
S(4) --> V(3) amplitude = -0.2975 alpha
D(48) --> S(1) amplitude = 0.3305 beta
D(49) --> S(2) amplitude = -0.1682 beta
D(50) --> S(1) amplitude = 0.4614 beta
D(50) --> S(4) amplitude = 0.3009 beta
D(51) --> V(3) amplitude = -0.2815 beta
D(52) --> V(6) amplitude = -0.2135 beta

Excited state 23: excitation energy (eV) = 4.7693
Total energy for state 23: -2190.268649819271
<S**2> : 7.0363
Trans. Mom.: 0.0000 X 0.0000 Y -0.0447 Z
Strength : 0.0002
D(52) --> V(5) amplitude = 0.9709 beta

Excited state 24: excitation energy (eV) = 4.7831
Total energy for state 24: -2190.268143813801
<S**2> : 6.1772
Trans. Mom.: 0.0032 X 0.1695 Y 0.0000 Z
Strength : 0.0034
D(51) --> V(1) amplitude = 0.9911 alpha

Excited state 25: excitation energy (eV) = 4.8565
Total energy for state 25: -2190.265447970550
<S**2> : 6.5812
Trans. Mom.: 0.0000 X 0.0000 Y 0.0645 Z
Strength : 0.0005
D(49) --> V(1) amplitude = 0.1518 alpha
D(50) --> V(1) amplitude = 0.2585 alpha
D(52) --> V(1) amplitude = 0.4164 alpha
S(1) --> V(2) amplitude = 0.2857 alpha
S(2) --> V(1) amplitude = 0.5313 alpha
S(4) --> V(2) amplitude = 0.1630 alpha
D(48) --> S(2) amplitude = 0.2358 beta
D(50) --> S(2) amplitude = 0.3401 beta
D(51) --> S(4) amplitude = -0.2163 beta
D(52) --> V(5) amplitude = 0.1563 beta

Excited state 26: excitation energy (eV) = 4.8664
Total energy for state 26: -2190.265083234634
<S**2> : 6.9060
Trans. Mom.: 0.0000 X 0.0000 Y -0.0542 Z
Strength : 0.0004
D(50) --> V(1) amplitude = -0.3212 alpha
D(52) --> V(1) amplitude = -0.3518 alpha
S(2) --> V(1) amplitude = 0.2945 alpha

S(4) --> V(6) amplitude = -0.1500 alpha
D(47) --> S(2) amplitude = 0.3149 beta
D(49) --> S(1) amplitude = -0.2279 beta
D(50) --> S(2) amplitude = 0.5930 beta

Excited state 27: excitation energy (eV) = 4.9245
Total energy for state 27: -2190.262949246336
<S**2> : 7.1325
Trans. Mom.: -0.2780 X 0.0001 Y 0.0000 Z
Strength : 0.0093
S(1) --> V(1) amplitude = -0.2203 alpha
S(2) --> V(2) amplitude = -0.2649 alpha
D(50) --> S(4) amplitude = 0.2323 beta
D(52) --> V(6) amplitude = 0.8696 beta

Excited state 28: excitation energy (eV) = 4.9592
Total energy for state 28: -2190.261672027625
<S**2> : 6.0473
Trans. Mom.: 0.0285 X 0.0339 Y 0.0000 Z
Strength : 0.0002
D(49) --> S(3) amplitude = 0.9805 beta

Excited state 29: excitation energy (eV) = 4.9651
Total energy for state 29: -2190.261457242253
<S**2> : 7.0578
Trans. Mom.: 0.0000 X 0.0000 Y -0.0568 Z
Strength : 0.0004
D(52) --> V(1) amplitude = 0.4462 alpha
S(2) --> V(1) amplitude = 0.1816 alpha
S(4) --> V(2) amplitude = 0.1838 alpha
S(4) --> V(6) amplitude = -0.1596 alpha
D(47) --> S(2) amplitude = 0.1853 beta
D(48) --> S(2) amplitude = -0.3610 beta
D(49) --> S(1) amplitude = 0.5864 beta

Excited state 30: excitation energy (eV) = 4.9857
Total energy for state 30: -2190.260697917812
<S**2> : 6.6327
Trans. Mom.: 0.0000 X 0.0000 Y 1.0615 Z
Strength : 0.1376
D(49) --> V(1) amplitude = 0.2710 alpha
D(50) --> V(1) amplitude = -0.1608 alpha
S(1) --> V(2) amplitude = -0.2231 alpha
S(2) --> V(1) amplitude = 0.2421 alpha
S(4) --> V(2) amplitude = -0.3683 alpha
D(47) --> S(2) amplitude = -0.3365 beta
D(49) --> S(1) amplitude = 0.4039 beta
D(49) --> S(4) amplitude = 0.3221 beta
D(50) --> S(2) amplitude = 0.2576 beta
D(51) --> S(4) amplitude = 0.2642 beta