

## Chirality transfer through sulfur or selenium to chiral propellers

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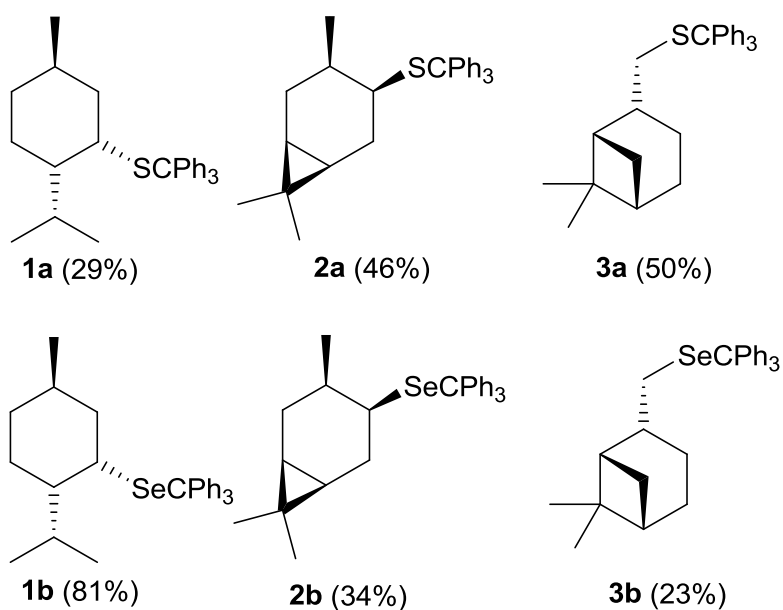


Figure 1. trityl sulfides **1a**, **2a**, **3a** and trityl selenides **1b**, **2b**, **3b**

### EXPERIMENTAL

Melting points were measured with a Büchi Tottoli SPM-20 heating unit and are uncorrected. NMR spectra were recorded on Bruker Avance III/400 or Bruker Avance III/700. Chemical shifts are expressed in parts per million (ppm) relative to TMS. Optical rotations were measured in 10-mm cells with a polAAR 3000 polarimeter. TLC was conducted on precoated silica gel plates (Merck 60F254) and the spots were visualized under UV light. Column chromatography was carried out on column using Silica Gel 60 Merck (70–230 mesh). All reactions requiring anhydrous conditions were conducted in flame-dried apparatus.

## 1. General procedure for the synthesis of sulfur trityl derivatives

Trityl chloride (0.85g, 3.0mmol) was slowly added to the solution of thiol (3.0mmol) and pyridine (0.26g, 3.3mmol) in dry DCM (6ml). The mixture was stirred at room temperature for 24h, poured into water (100ml), and extracted with DCM (2 x 50ml). The combined organic layers were washed with water (2 x 50ml), dried over magnesium sulphate, and evaporated. The crude product was purified by column chromatography (silica gel, hexane/ethyl acetate 98/2).

### 1.1. (+)-Neomenthyl trityl sulfide (1a)

Yield = 29%, oil;  $[\alpha]_{\text{D}}^{20} = +99.83$  (c 3.0,  $\text{CHCl}_3$ );  $^1\text{H NMR}$  (700 MHz,  $\text{CDCl}_3$ )  $\sigma = 0.36$  (d,  $J=6.3$  Hz, 3H,  $\text{CH}_3$ ), 0.70-0.74 (m, 1H), 0.75 (d,  $J=7.0$  Hz, 3H,  $\text{CH}_3$ ), 0.81 (d,  $J=7.0$  Hz, 3H,  $\text{CH}_3$ ), 0.82-0.92 (m, 2H), 1.23-1.28 (m, 1H), 1.28-1.37 (m, 1H), 1.67-1.75 (m, 2H), 1.77-1.82 (m, 1H), 2.19-2.28 (m, 1H), 2.57 (d,  $J=2.8$  Hz, 1H), 7.19-7.22 (m, 3H, arom.), 7.28 (t,  $J=7.7$  Hz, 6H, arom.), 7.59-7.63 (m, 6H, arom.) ppm;  $^{13}\text{C NMR}$  (100 MHz,  $\text{CDCl}_3$ )  $\sigma = 20.39$  ( $\text{CH}_3$ ), 21.00 ( $\text{CH}_3$ ), 21.62 ( $\text{CH}_3$ ), 26.79 ( $\text{CH}_2$ ), 26.88 (CH), 28.97 (CH), 35.36 ( $\text{CH}_2$ ), 39.88 ( $\text{CH}_2$ ), 45.64 (CH), 50.13 (CH), 66.50 (C), 126.17 (3 x CH), 127.43 (6 x CH), 129.51 (6 x CH), 145.42 (3 x C) ppm.

### 1.2. (-)-Caranyl trityl sulfide (2a)

Yield = 46%, oil;  $[\alpha]_{\text{D}}^{20} = -49.13$  (c 2.5,  $\text{CHCl}_3$ );  $^1\text{H NMR}$  (700 MHz,  $\text{CDCl}_3$ )  $\sigma = 0.25$ -0.32 (m, 2H), 0.79-0.82 (m, 2H), 0.89 (s, 3H,  $\text{CH}_3$ ), 0.91 (s, 3H,  $\text{CH}_3$ ), 0.98 (d,  $J=7.0$  Hz, 3H,  $\text{CH}_3$ ), 1.02-1.05 (m, 1H), 1.82-1.89 (m, 1H), 1.92-1.98 (m, 1H), 2.40-2.44 (m, 1H), 7.22-7.25 (m, 3H, arom.), 7.29-7.33 (m, 6H, arom.), 7.57-7.59 (m, 6H, arom.) ppm;  $^{13}\text{C NMR}$  (176 MHz,  $\text{CDCl}_3$ )  $\sigma = 15.52$  ( $\text{CH}_3$ ), 17.45 (C), 18.19 (CH), 18.81 ( $\text{CH}_3$ ), 24.06 ( $\text{CH}_2$ ), 24.67 (CH), 26.87( $\text{CH}_2$ ), 28.39 ( $\text{CH}_3$ ), 29.16 (CH), 45.19 (CH), 67.27 (C), 126.44 (3 x CH), 127.78 (6 x CH), 129.77 (6 x CH), 145.52 (3 x C) ppm.

### 1.3. (-)-Myrtanyl trityl sulfide (3a)

Yield = 50%, oil;  $[\alpha]_{\text{D}}^{20} = -40.25$  (c 2.8,  $\text{CHCl}_3$ );  $^1\text{H NMR}$  (700 MHz,  $\text{CDCl}_3$ )  $\sigma = 0.75$  (s, 3H,  $\text{CH}_3$ ), 0.82 (d,  $J=9.8$  Hz, 1H), 0.87-0.92 (m, 1H), 1.14 (s, 3H,  $\text{CH}_3$ ), 1.30-1.38 (m, 1H), 1.78-1.81 (m, 1H), 1.82-1.90 (m, 2H,  $\text{CH}_2$ ), 1.96-1.98 (m, 1H), 2.04-2.10 (m, 1H), 2.18 (dd,  $J=7.7$  Hz,  $J=11.2$  Hz, 1H), 2.29 (dd,  $J=7.7$  Hz,  $J=11.2$  Hz, 1H) 2.29-2.32 (m, 1H), 7.22-7.24 (m, 3H, arom.), 7.29-7.33 (m, 6H, arom.), 7.44-7.46 (m, 6H, arom.) ppm;  $^{13}\text{C NMR}$  (176 MHz,  $\text{CDCl}_3$ )  $\sigma = 22.21$  ( $\text{CH}_2$ ), 22.78 ( $\text{CH}_3$ ), 26.19 ( $\text{CH}_2$ ), 27.94 ( $\text{CH}_3$ ), 33.36 ( $\text{CH}_2$ ), 38.94 ( $\text{CH}_2$ ), 40.33 (CH), 41.23 (CH), 45.51 (CH), 66.30 (C), 126.48 (3 x CH), 127.78 (6 x CH), 129.65 (6 x CH), 145.12 (C) ppm.

## 2. General procedure for the synthesis of selenyl trityl derivatives

Selenol (2.3 mmol) in dry DCM (10ml) was carefully added to the suspension of sodium hydride (0.06g, 2.5mmol) in dry DCM (5ml) under argon atmosphere. After stirring for 10 min the mixture was cooled to 0 °C and trityl chloride (0.64g, 2.3mmol) in dry DCM (10ml) was added. The mixture was stirred at room temperature for 1h, poured into water (100ml) and extracted with DCM (3 x 50ml). The combined organic layers were dried over magnesium sulphate, and evaporated. The crude product was purified by column chromatography (silica gel, hexane/ethyl acetate 99/1).

### 2.1. (+)-Neomenthyl trityl selenide (1b)

Yield 81%, mp = 137-139 °C (from methanol);  $[\alpha]_D^{20} = -199.00$  (c 3.01, CHCl<sub>3</sub>); <sup>1</sup>H NMR (700 MHz, CDCl<sub>3</sub>)  $\sigma = 0.54$  (d,  $J=6.3$  Hz, 3H, CH<sub>3</sub>), 0.79-1.02 (m, 9H), 1.16-1.21 (m, 1H), 1.36 (d,  $J=13.3$  Hz, 1H), 1.64-1.69 (m, 1H), 1.79 (t,  $J=14.7$  Hz, 2H, CH<sub>2</sub>), 2.25-2.33 (m, 1H), 2.59 (s, 1H), 7.19 (t,  $J=7.0$  Hz, 3H, arom.), 7.25-7.29 (m, 6H, arom.), 7.57 (d,  $J=7.7$  Hz, 6H, arom.) ppm; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\sigma = 20.63$  (CH<sub>3</sub>), 21.09 (CH<sub>3</sub>), 21.69 (CH<sub>3</sub>), 27.88 (CH), 28.20 (CH<sub>2</sub>), 30.34 (CH), 35.28 (CH<sub>2</sub>), 40.41 (CH<sub>2</sub>), 47.62 (CH), 50.57 (CH), 64.50 (C), 126.15 (3 x CH), 127.56 (6 x CH), 129.89 (6 x CH), 145.76 (3 x C) ppm; <sup>77</sup>Se NMR (76 MHz, CDCl<sub>3</sub>)  $\sigma = 474.04$  ppm.

### 2.2. (-)-Caranyl trityl selenide (2b)

Yield = 34%, oil;  $[\alpha]_D^{20} = -24.95$  (c 3.00, CHCl<sub>3</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\sigma = 0.27$ -0.32 (m, 2H), 0.77-0.88 (m, 1H), 0.90 (s, 3H, CH<sub>3</sub>), 0.93 (s, 3H, CH<sub>3</sub>), 0.97 (d,  $J=6.8$  Hz, 2H), 1.00-1.09 (m, 2H), 1.35-1.43 (m, 1H), 1.92-1.97 (m, 2H), 2.49-2.51 (m, 1H), 7.19-7.24 (m, 3H, arom.), 7.28-7.35 (m, 6H, arom.), 7.51-7.53 (m, 6H, arom.) ppm; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\sigma = 15.51$  (CH<sub>3</sub>), 17.31 (C), 18.37 (CH), 20.29 (CH<sub>3</sub>), 24.67 (CH), 24.69 (CH<sub>2</sub>), 26.69 (CH<sub>2</sub>), 28.29 (CH<sub>3</sub>), 29.49 (CH), 45.26 (CH), 64.76 (C), 126.28 (3 x CH), 127.67 (6 x CH), 130.09 (6 x CH), 145.83 (3 x C) ppm; <sup>77</sup>Se NMR (76 MHz, CDCl<sub>3</sub>)  $\sigma = 551.30$  ppm.

### 2.3. (-)-Myrtanyl trityl selenide (3b)

Yield = 23%, oil;  $[\alpha]_D^{20} = -28.60$  (c 0.93, CHCl<sub>3</sub>); <sup>1</sup>H NMR (700 MHz, CDCl<sub>3</sub>)  $\sigma = 0.80$  (d,  $J=9.6$  Hz, 1H), 0.81 (s, CH<sub>3</sub>, 3H), 1.14 (s, 3H, CH<sub>3</sub>), 1.32-1.35 (m, 1H), 1.75-1.89 (m, 5H), 2.08-2.11 (m, 1H), 2.25-2.31 (m, 1H), 2.34 (dd,  $J=8.0$  Hz,  $J=10.8$  Hz, 1H), 2.45-2.50 (dd,  $J=7.6$  Hz,  $J=10.8$  Hz, 1H), 7.19-7.24 (m, 3H, arom.), 7.27-7.30 (m, 6H, arom.), 7.35-7.36 (m, 6H, arom.) ppm; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\sigma = 22.76$  (CH<sub>2</sub>), 22.80 (CH<sub>3</sub>), 26.08 (CH<sub>2</sub>), 27.87 (CH<sub>3</sub>), 33.33 (CH<sub>2</sub>), 35.76 (CH<sub>2</sub>), 40.84 (CH), 41.13 (CH), 46.32 (CH), 64.03 (C), 126.37 (3 x CH), 127.69 (6 x CH), 129.97 (6 x CH), 145.79 (3 x C) ppm; <sup>77</sup>Se NMR (76 MHz, CDCl<sub>3</sub>)  $\sigma = 492.63$  ppm.

Table X1. NATURAL POPULATIONS: Natural atomic orbital occupancies on sulfur atom

NAO	Atom	No	lang	Type(AO)	Occupancy	Energy
295	S	30	S	Cor( 1s)	2.00000	-87.68626
296	S	30	S	Cor( 2s)	1.99946	-8.79986
297	S	30	S	Val( 3s)	1.68161	-0.70676
298	S	30	S	Ryd( 4s)	0.00461	1.00685
299	S	30	S	Ryd( 5s)	0.00001	14.06031
300	S	30	S	Ryd( 6s)	0.00000	182.68265
301	S	30	px	Cor( 2p)	1.99988	-5.87156
302	S	30	px	Val( 3p)	0.89619	-0.13957
303	S	30	px	Ryd( 4p)	0.00465	0.83838
304	S	30	px	Ryd( 5p)	0.00015	2.56111
305	S	30	px	Ryd( 6p)	0.00000	16.74413
306	S	30	py	Cor( 2p)	1.99997	-5.86387
307	S	30	py	Val( 3p)	1.89676	-0.22085
308	S	30	py	Ryd( 4p)	0.00461	0.64485
309	S	30	py	Ryd( 5p)	0.00011	2.32687
310	S	30	py	Ryd( 6p)	0.00000	16.82928
311	S	30	pz	Cor( 2p)	1.99991	-5.86902
312	S	30	pz	Val( 3p)	1.32865	-0.18850
313	S	30	pz	Ryd( 4p)	0.00191	0.39130
314	S	30	pz	Ryd( 5p)	0.00014	2.49061
315	S	30	pz	Ryd( 6p)	0.00000	16.73898
316	S	30	dxy	Ryd( 3d)	0.00118	1.12932
317	S	30	dxz	Ryd( 3d)	0.00625	1.33749
318	S	30	dyz	Ryd( 3d)	0.00130	1.04503
319	S	30	dx2y2	Ryd( 3d)	0.00290	1.22194
320	S	30	dz2	Ryd( 3d)	0.00175	1.09093

Natural Electron Configuration of sulfur atom: [core]3S( 1.68)3p( 4.12)3d( 0.01)4p( 0.01)

Table X2. Occupancy of natural orbitals (NBOs) located on sulfur atom (except core NBOs)

(Occupancy)	Bond orbital/	Coefficients/	Hybrids
13. (1.97232)	BD ( 1) C 10 - S 30		
( 51.28%)	0.7161* C 10 s( 18.71%)p 4.34( 81.16%)d 0.01( 0.14%)		
	0.0002 0.4322 -0.0176 -0.0007 0.6794		
	-0.0078 -0.0071 -0.0706 -0.0053 0.0019		
	-0.5872 0.0054 0.0057 -0.0050 -0.0312		
	0.0047 0.0176 0.0050		
( 48.72%)	0.6980* S 30 s( 16.90%)p 4.89( 82.63%)d 0.03( 0.47%)		
	0.0000 0.0003 0.4095 0.0365 0.0006		
	0.0000 -0.0001 -0.6328 0.0436 0.0048		
	0.0005 0.0000 0.1185 0.0011 -0.0015		
	-0.0001 0.0002 0.6401 -0.0131 0.0049		
	0.0000 -0.0110 -0.0582 0.0127 0.0247		
	0.0213		
31. (1.94403)	BD ( 1) S 30 - C 31		
( 46.76%)	0.6838* S 30 s( 14.79%)p 5.73( 84.78%)d 0.03( 0.43%)		
	0.0000 0.0005 0.3829 0.0353 -0.0009		
	0.0000 0.0003 0.7663 -0.0405 -0.0054		
	-0.0005 0.0000 0.0706 0.0010 -0.0002		
	-0.0002 0.0002 0.5039 -0.0001 0.0048		
	-0.0007 0.0086 0.0545 0.0048 0.0349		
	-0.0006		
( 53.24%)	0.7297* C 31 s( 15.88%)p 5.29( 84.02%)d 0.01( 0.10%)		
	0.0000 0.3972 -0.0322 -0.0003 -0.8308		
	0.0300 0.0044 -0.0275 0.0003 -0.0003		
	-0.3850 0.0067 0.0027 0.0032 0.0217		
	0.0016 0.0212 -0.0059		
111. (1.97559)	LP ( 1) S 30		
	s( 68.54%)p 0.46( 31.45%)d 0.00( 0.01%)		
	0.0000 -0.0003 0.8277 -0.0162 0.0000		
	0.0000 -0.0001 -0.0414 -0.0020 0.0000		
	-0.0001 0.0000 -0.0910 -0.0011 0.0018		
	0.0000 -0.0001 -0.5518 0.0015 -0.0015		
	0.0003 0.0002 0.0012 -0.0037 -0.0074		
	-0.0071		
112. (1.91543)	LP ( 2) S 30		
	s( 0.00%)p 1.00( 99.98%)d 0.00( 0.02%)		
	0.0000 -0.0001 0.0005 0.0027 -0.0005		
	0.0000 0.0001 -0.0178 -0.0005 0.0004		
	-0.0003 0.0000 -0.9856 -0.0298 0.0002		
	-0.0006 0.0000 0.1647 0.0070 0.0021		
	-0.0002 -0.0011 0.0007 -0.0122 0.0004		
	0.0038		

Table X3. Second Order Perturbation Theory Analysis of Fock Matrix in NBO Basis – selected data for interactions of NBOs located on sulfur atom

Threshold for printing: 0.50 kcal/mol

Donor NBO (i)		Acceptor NBO (j)		E(2) kcal/mol	E(j)-E(i) a.u.	F(i,j) a.u.
<b>9. BD ( 1) C 7 - H 9</b>	<b>/689. BD*( 1) C 10 - S 30</b>	<b>6.33</b>	<b>0.63</b>	<b>0.057</b>		
12. BD ( 1) C 10 - C 12	/338. RY*( 1) S 30	0.56	1.39	0.025		
12. BD ( 1) C 10 - C 12	/707. BD*( 1) S 30 - C 31	0.97	0.70	0.023		
34. BD ( 1) C 31 - C 34	/689. BD*( 1) C 10 - S 30	0.80	0.77	0.022		
35. BD ( 1) C 32 - C 35	/707. BD*( 1) S 30 - C 31	0.70	0.80	0.021		
<b>36. BD ( 2) C 32 - C 35</b>	<b>/707. BD*( 1) S 30 - C 31</b>	<b>5.07</b>	<b>0.36</b>	<b>0.041</b>		
<b>38. BD ( 1) C 33 - C 45</b>	<b>/707. BD*( 1) S 30 - C 31</b>	<b>2.03</b>	<b>0.80</b>	<b>0.036</b>		
<b>39. BD ( 2) C 33 - C 45</b>	<b>/707. BD*( 1) S 30 - C 31</b>	<b>2.21</b>	<b>0.36</b>	<b>0.027</b>		
<b>43. BD ( 2) C 34 - C 44</b>	<b>/707. BD*( 1) S 30 - C 31</b>	<b>2.13</b>	<b>0.36</b>	<b>0.027</b>		
92. CR ( 1) C 31	/707. BD*( 1) S 30 - C 31	0.76	10.22	0.080		
712. BD*( 2) C 32 - C 35	/707. BD*( 1) S 30 - C 31	7.18	0.08	0.046		
715. BD*( 2) C 33 - C 45	/707. BD*( 1) S 30 - C 31	2.19	0.08	0.026		
719. BD*( 2) C 34 - C 44	/707. BD*( 1) S 30 - C 31	1.78	0.07	0.023		
13. BD ( 1) C 10 - S 30	/159. RY*( 1) C 7	0.58	1.49	0.026		
13. BD ( 1) C 10 - S 30	/161. RY*( 3) C 7	0.71	1.49	0.029		
13. BD ( 1) C 10 - S 30	/200. RY*( 1) C 12	0.69	1.44	0.028		
13. BD ( 1) C 10 - S 30	/202. RY*( 3) C 12	0.54	1.43	0.025		
13. BD ( 1) C 10 - S 30	/357. RY*( 3) C 31	0.59	1.39	0.026		
<b>13. BD ( 1) C 10 - S 30</b>	<b>/685. BD*( 1) C 7 - H 9</b>	<b>2.17</b>	<b>0.98</b>	<b>0.041</b>		
<b>13. BD ( 1) C 10 - S 30</b>	<b>/690. BD*( 1) C 12 - H 13</b>	<b>2.32</b>	<b>0.99</b>	<b>0.043</b>		
13. BD ( 1) C 10 - S 30	/710. BD*( 1) C 31 - C 34	0.78	0.95	0.024		
<b>31. BD ( 1) S 30 - C 31</b>	<b>/370. RY*( 3) C 32</b>	<b>1.17</b>	<b>1.27</b>	<b>0.035</b>		
31. BD ( 1) S 30 - C 31	/382. RY*( 2) C 33	0.54	1.73	0.028		
31. BD ( 1) S 30 - C 31	/383. RY*( 3) C 33	0.59	1.32	0.025		
<b>31. BD ( 1) S 30 - C 31</b>	<b>/395. RY*( 2) C 34</b>	<b>1.27</b>	<b>1.64</b>	<b>0.041</b>		
<b>31. BD ( 1) S 30 - C 31</b>	<b>/688. BD*( 1) C 10 - C 12</b>	<b>1.15</b>	<b>0.91</b>	<b>0.029</b>		
31. BD ( 1) S 30 - C 31	/711. BD*( 1) C 32 - C 35	0.83	1.11	0.027		
<b>31. BD ( 1) S 30 - C 31</b>	<b>/712. BD*( 2) C 32 - C 35</b>	<b>4.04</b>	<b>0.57</b>	<b>0.046</b>		
<b>31. BD ( 1) S 30 - C 31</b>	<b>/714. BD*( 1) C 33 - C 45</b>	<b>2.79</b>	<b>1.11</b>	<b>0.050</b>		
<b>31. BD ( 1) S 30 - C 31</b>	<b>/715. BD*( 2) C 33 - C 45</b>	<b>1.19</b>	<b>0.57</b>	<b>0.025</b>		
<b>31. BD ( 1) S 30 - C 31</b>	<b>/718. BD*( 1) C 34 - C 44</b>	<b>3.10</b>	<b>1.11</b>	<b>0.053</b>		
<b>31. BD ( 1) S 30 - C 31</b>	<b>/719. BD*( 2) C 34 - C 44</b>	<b>1.55</b>	<b>0.57</b>	<b>0.029</b>		
88. CR ( 2) S 30	/183. RY*( 2) C 10	0.83	9.75	0.080		
88. CR ( 2) S 30	/355. RY*( 1) C 31	0.93	9.78	0.085		
<b>111. LP ( 1) S 30</b>	<b>/355. RY*( 1) C 31</b>	<b>1.52</b>	<b>1.60</b>	<b>0.044</b>		
111. LP ( 1) S 30	/686. BD*( 1) C 7 - C 10	0.90	1.01	0.027		
111. LP ( 1) S 30	/687. BD*( 1) C 10 - H 11	0.67	1.08	0.024		
<b>111. LP ( 1) S 30</b>	<b>/708. BD*( 1) C 31 - C 32</b>	<b>1.23</b>	<b>1.01</b>	<b>0.032</b>		
111. LP ( 1) S 30	/709. BD*( 1) C 31 - C 33	0.81	1.00	0.026		
112. LP ( 2) S 30	/685. BD*( 1) C 7 - H 9	0.75	0.63	0.020		
<b>112. LP ( 2) S 30</b>	<b>/686. BD*( 1) C 7 - C 10</b>	<b>4.69</b>	<b>0.61</b>	<b>0.049</b>		
<b>112. LP ( 2) S 30</b>	<b>/687. BD*( 1) C 10 - H 11</b>	<b>2.31</b>	<b>0.69</b>	<b>0.036</b>		
112. LP ( 2) S 30	/688. BD*( 1) C 10 - C 12	0.55	0.60	0.016		
<b>112. LP ( 2) S 30</b>	<b>/708. BD*( 1) C 31 - C 32</b>	<b>2.16</b>	<b>0.62</b>	<b>0.033</b>		
<b>112. LP ( 2) S 30</b>	<b>/709. BD*( 1) C 31 - C 33</b>	<b>3.86</b>	<b>0.61</b>	<b>0.044</b>		
112. LP ( 2) S 30	/712. BD*( 2) C 32 - C 35	0.75	0.26	0.013		
112. LP ( 2) S 30	/733. BD*( 1) C 40 - H 55	0.62	0.69	0.019		
112. LP ( 2) S 30	/752. BD*( 1) C 49 - H 64	0.53	0.69	0.017		

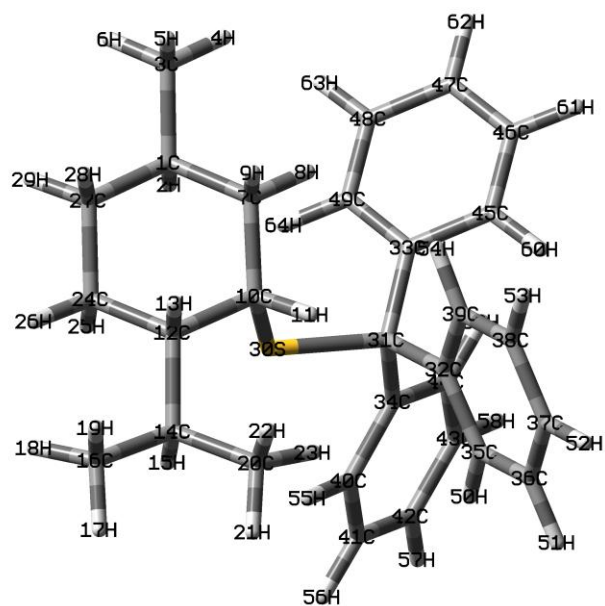
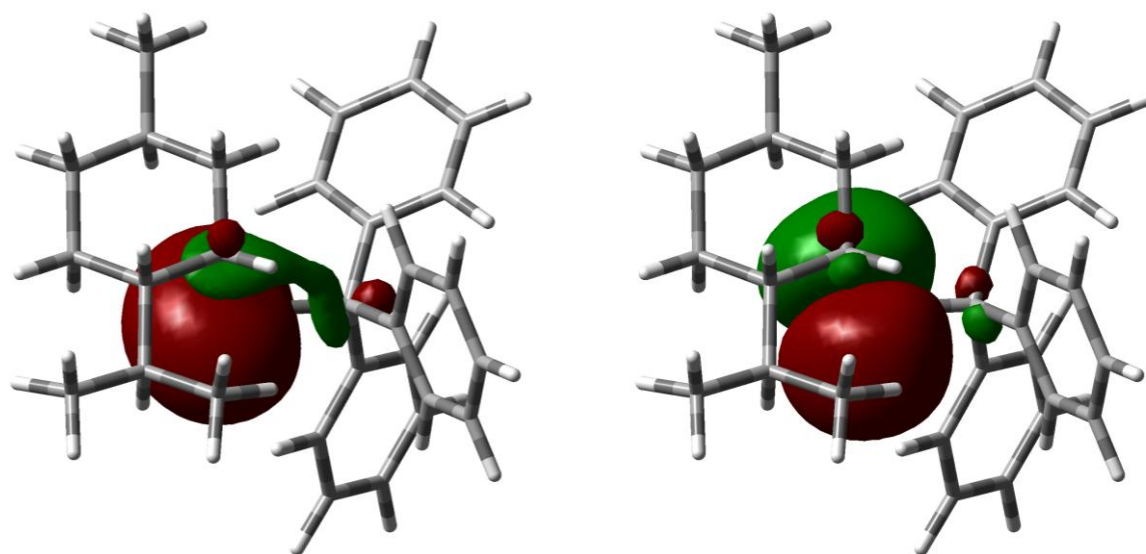


Figure X1. Atom numbering in the structure of conf 1 of **1**.



a/  
b/  
Figure X2. Lone pairs orbitals a/ lower in energy *s*-type, b/ highest occupied NBO of *p*-type

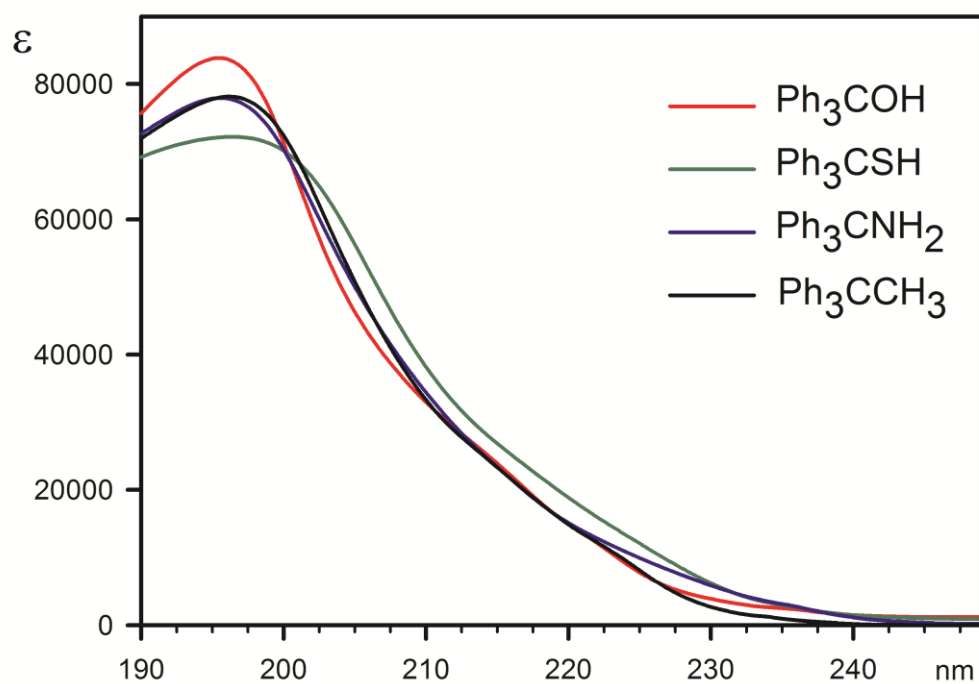
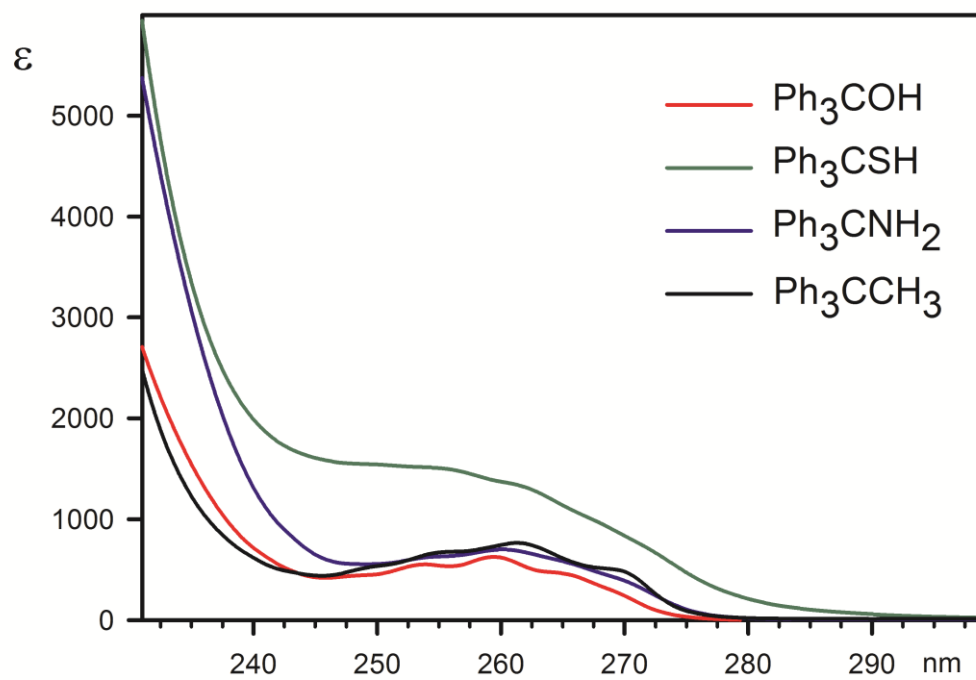


Figure X3. UV spectra of trityl derivatives in acetonitrile

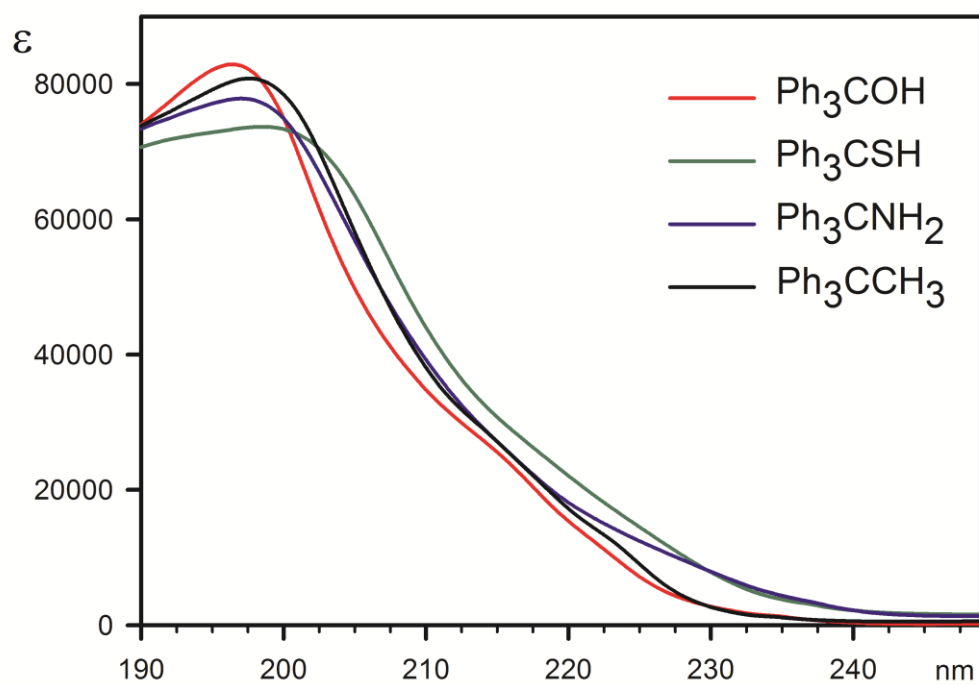
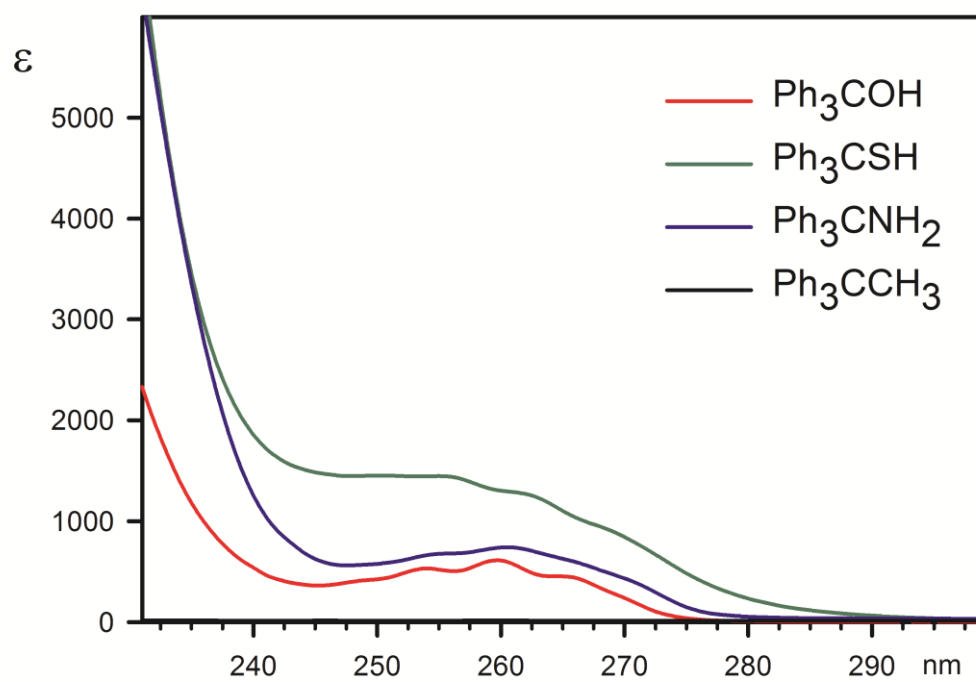


Figure X4. UV spectra of trityl derivatives in cyclohexane