Electronic Supplementary Material (ESI) for RSC Advances. This journal is © The Royal Society of Chemistry 2015

Supporting Information for Publication

A new fluorescent pH probes for acid conditions

Xuan Zhang^{a,b}, Guang-Jie Song^c, Xiang-Jian Cao^{a,b}, Jin-Ting Liu^a, Ming-Yu Chen^{a,b}, Xiao-Qun Cao^c, Bao-Xiang Zhao^{a,*}

^aSchool of Chemistry and Chemical Engineering, Shandong University, Jinan 250100, PR China ^bTaishan College, Shandong University, Jinan 250100, PR China ^cTaishan Medical University, Tai'an 271000, PR China

*corresponding author:Institute of Organic Chemistry School of Chemistry & Chemical Engineering Shandong University
27# ShandaNanlu, Jinan 250100,PR China Tel:0086-531-88366425
Fax: 0086-531-88564464
Email address: bxzhao@sdu.edu.cn

Contents

Table S1. Summary of crystal data and structure refinement for the probe······2
Table S2. Selected geometric parameters (Å, °)
Table S3. Table S3Hydrogen-bond geometry (Å, °).
Table S4 . Analysis of Y-XCg(π-Ring) Interactions(Å, °) ·······3
Fig. S1. (a) The ¹ H NMR spectroscopy of the probe in DMSO-d ₆ after adding a drop of DCl;
(b) The ¹ H NMR spectroscopy of the probe in DMSO- d_6 4
Fig. S2. Absorption spectra of the probe4
Fig. S3. ¹ H NMR spectra of the probe in DMSO- d_6
Fig. S4. ¹ H NMR spectra of the probe in DMSO- d_6 with a drop of DCl \cdots 5
Fig. S5. ¹³ C NMR spectra of the probe in DMSO- d_6
Fig. S5. HRMS spectra of the probe

Compound	probe
Empirical formula	$C_{12}H_{13}ClN_2O_2$
Formula weight	252.69
Crystal system	monoclinic
Space group	$P2_1/c$
Temperature (K)	173
<i>a, b, c</i> (Å)	4.7190(13), 22.606(6), 11.214(3)
α, β, χ(°)	90, 92.837(3), 90
V(Å ³)	1194.8(6)
Ζ	4
Calculated density [Mg/m ³]	1.405
μ (mm ⁻¹)	0.311
Radiation[Å]	ΜuKα, 0.71000
Crystal size [mm]	0.09×0.10×0.10
F(000)	528
No. of measured, independent and	5895, 2103
observed $[I > 2\sigma(I)]$ reflections	1993
Rint	0.015
θ Range for data collection [°]	2.6-25.0
Ranges of indices <i>h</i> , <i>k</i> , <i>l</i>	$-5 \le h \le 5, -26 \le k \le 26 - 13 \le l \le 13$
Final R indices $[I > 2\sigma(I)]$	$R_1 = 0.0298, wR_2 = 0.0781$
Goodness of fit on F^2	1.02
$\Delta \rho$ (max./min) [e Å ⁻³]	0.32; -0.24

Table S1 Summary of crystal data and structure refinement for the probe

Table	52	Selected	geometric	narameters	(Å	0	`
I able	54	Scielleu	geometric	parameters	ιA,		,

Tuble 52 Beleeted ge	ometrie parameters	(11,)	
Cl1—C5	1.7162(15)	N1—C5	1.3429(18)
O1—C1	1.2717(18)	N1—C6	1.3356(18)
O2—C1	1.2672(18)	N2—C4	1.4058(18)
C1—C2	1.476(2)	N2—C6	1.3620(18)
C2—C3	1.3677(19)	N2—C7	1.3933(17)
C2—C8	1.437(2)	C4—C5	1.3810(19)
C3—C4	1.4028(19)	С6—С9	1.489(2)
С7—С8	1.3408(19)	C9—C10	1.533(2)
C10—C11	1.522(2)	C11—C12	1.519(2)
C5—N1—C6	105.90(11)	C4—N2—C6	107.75(11)
C4—N2—C7	120.76(11)	C6—N2—C7	131.48(12)
O1—C1—O2	123.74(13)	O1—C1—C2	117.79(12)
O2—C1—C2	118.47(12)	Cl1—C5—N1	122.24(10)

Table S3 Hydrogen-bond geometry (Å, $^{\circ}$).

D—HA	D—H	НА	DA	D—HA
01—H1O2 ⁱ	0.84	1.78	2.6107(17)	167
C8—H8N1 ⁱⁱ	0.95	2.57	3.353(2)	140

Symmetry codes: (i) 1-x,1-y,-z; (ii)-1+x,1/2-y,-1/2+z

Table S4 Analysis of Y-X...Cg(π -Ring) Interactions(Å, °).

Y—XCg	XCg	YCg	Y—XCg
C5—Cl1Cg1 ⁱⁱⁱ	3.6890(12)	4.1240(19)	92.05(5)
C5—Cl1Cg2 ⁱⁱⁱⁱ	3.5087(12)	3.6057(18)	79.21(5)

Symmetry codes: (iii) 1+x, y, z



Fig. S1 (a) The ¹H NMR spectroscopy of the probe in DMSO-d₆ after adding a drop of DCl; (b) The ¹H NMR spectroscopy of the probe in DMSO-d₆



Fig. S2 Absorption spectra of the probe $(30 \,\mu M)$ in solution (1:1, B-R-EtOH, v/v) with different pH.



Fig. S3 ¹H NMR spectra of the probe in DMSO- d_6



Fig. S4 ¹H NMR spectra of the probe in DMSO-*d*₆with a drop of DCl



Fig. S5 ¹³C NMR spectra of the probe in DMSO- d_6



Fig. S6 HRMS spectra of the probe