

Insights into the Morita-Baylis-Hillman reaction of Isomeric Dibenzofuran Carbaldehydes: A Theoretical and Mass spectral Study

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Supporting Information

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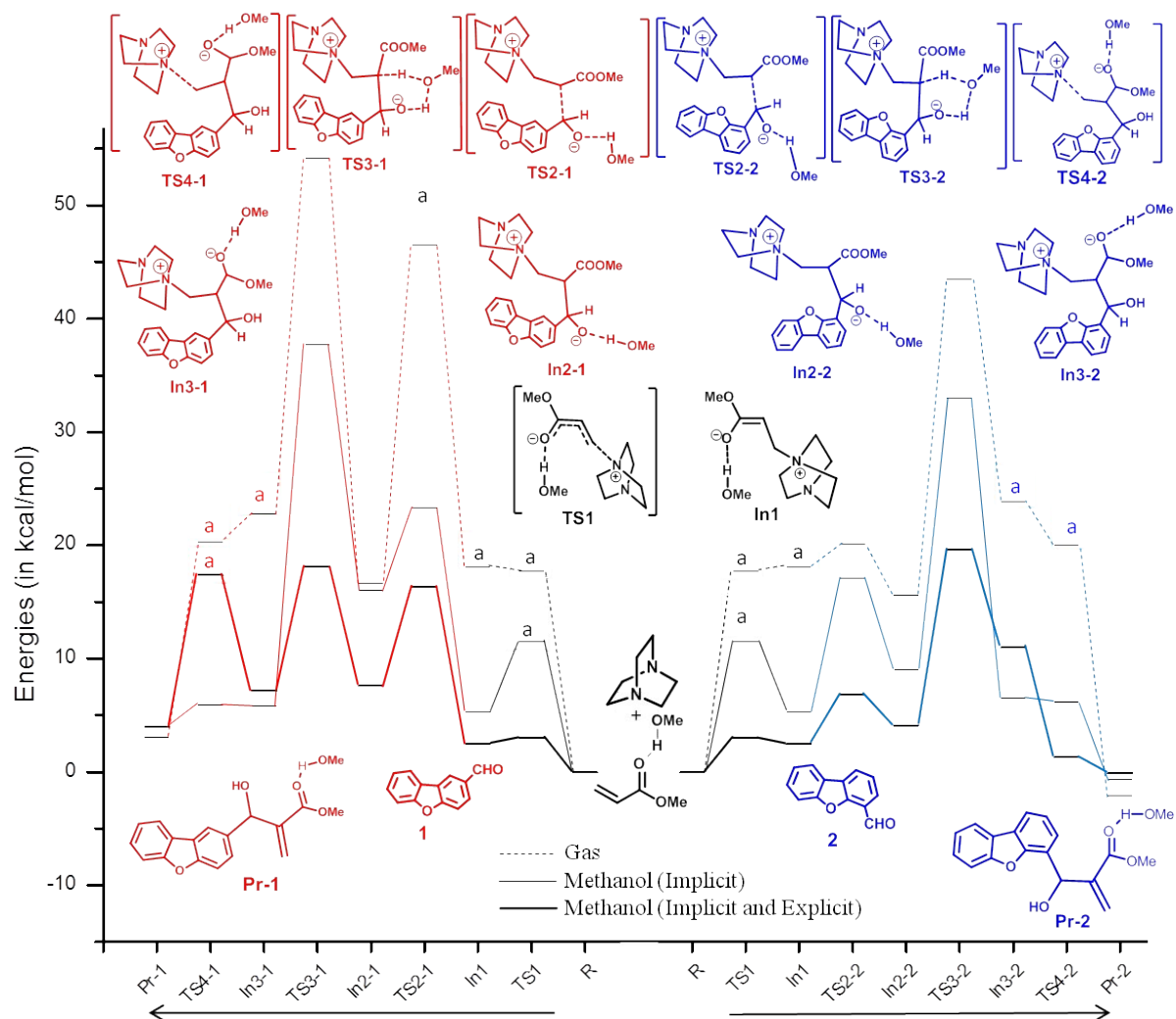


Figure S1. Gas phase as well as solvent (implicit and explicit methanol) phase reaction energy profiles for the considered reaction obtained using the aldehydes **1** (red) and **2** (blue) at mPW1K/6-31+G(d) level of theory. a: represents the structures obtained at mPW1K/6-31+g(d)//HF/3-21G level of theory.

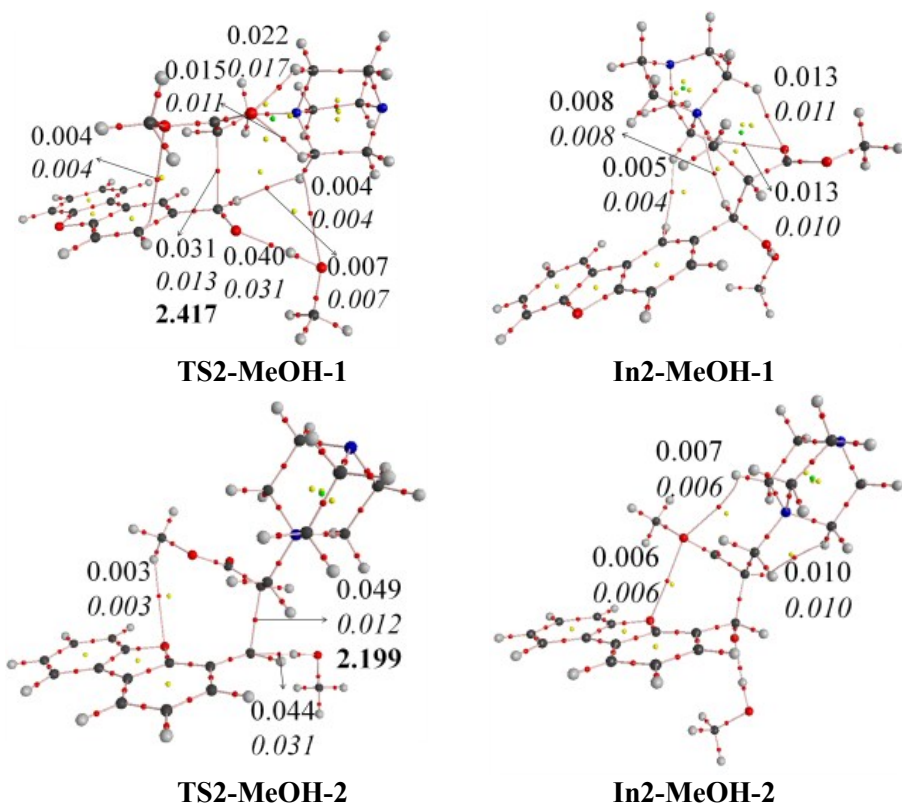
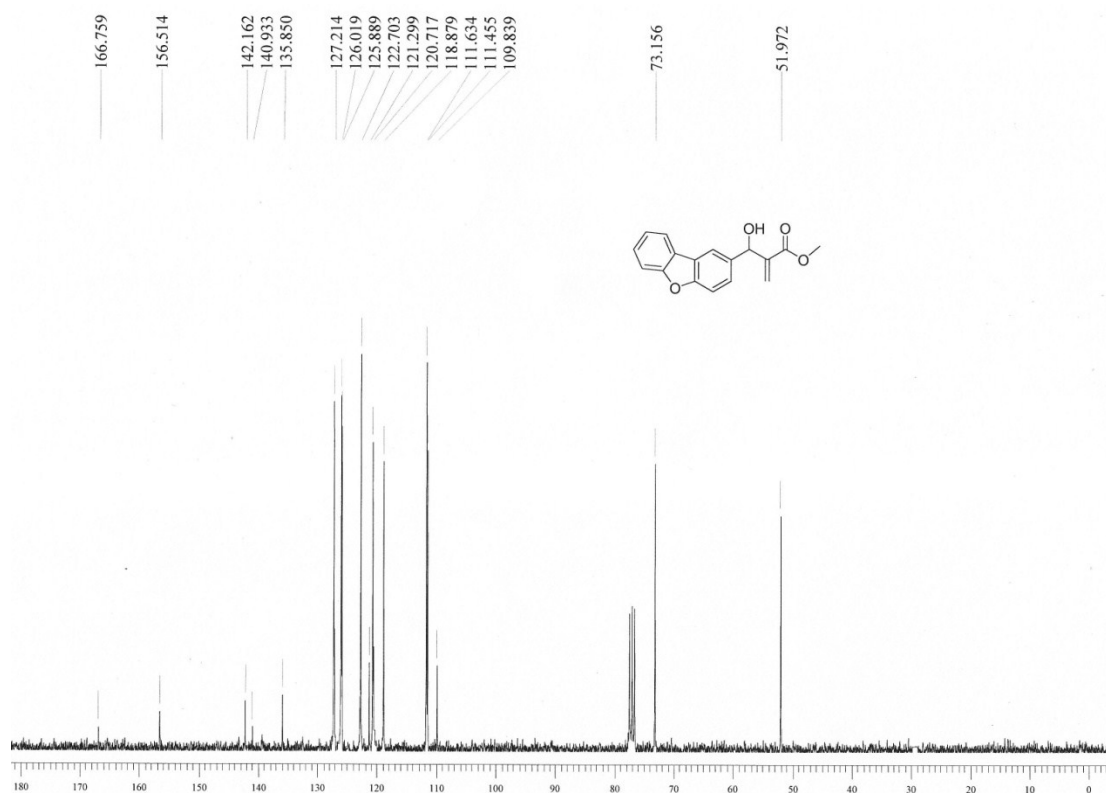
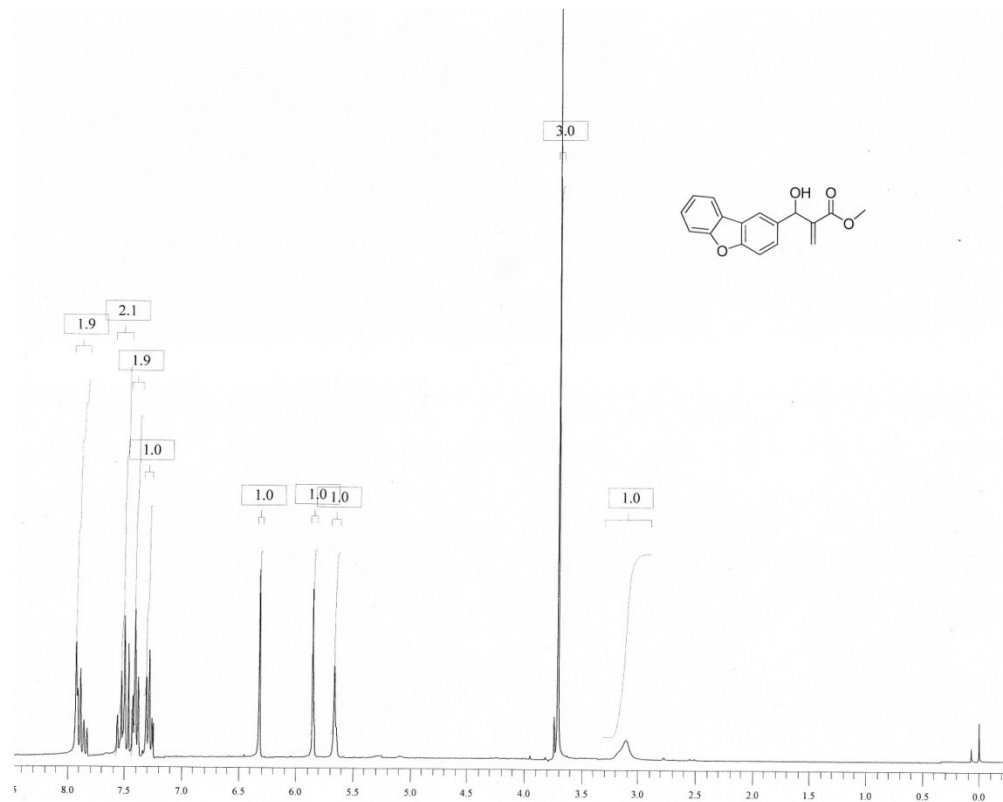


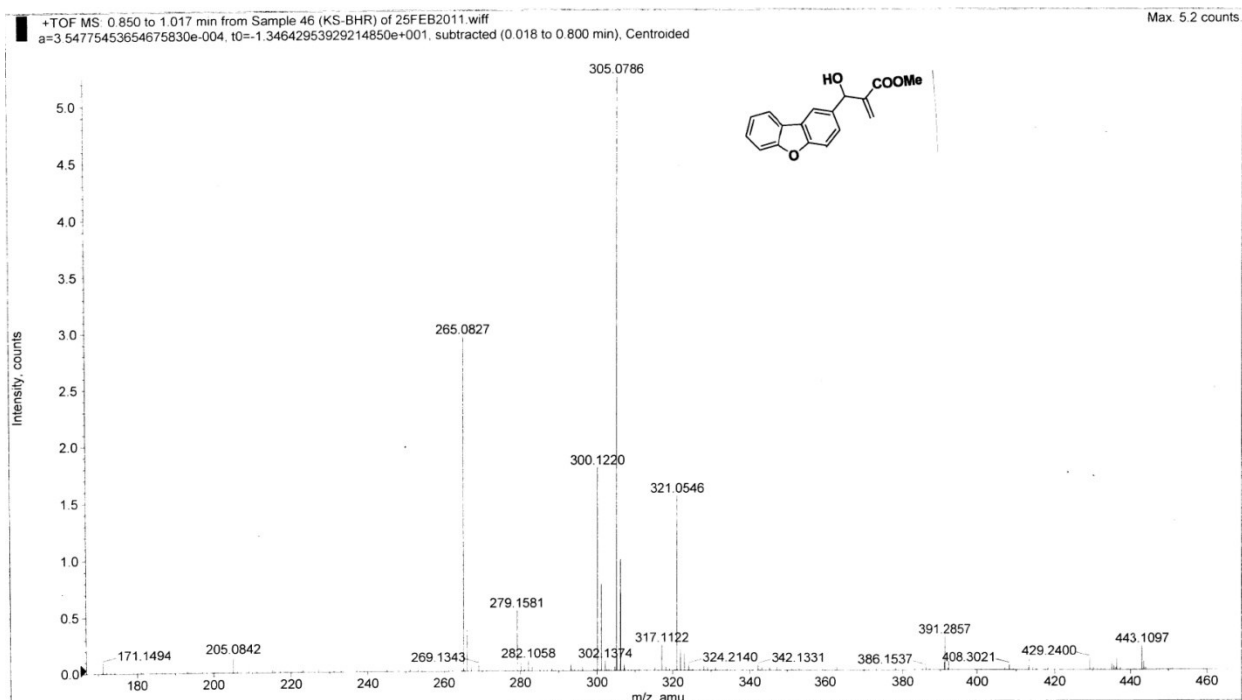
Figure S2. Bond critical points for **TS2-MeOH-1**, **TS2-MeOH-2**, **In2-MeOH-1** and **In2-MeOH-2**. Electron density values: normal, Laplacian of electron density: italics, Bond lengths: bold.

Table S1: Activation (for transition states) and reaction energies (for intermediates) obtained at mPW1K/6-31+G(d,p) and B97-D/6-31+G(d,p) levels on HF/6-31+G(d,p) optimized geometries, and HF/6-31+G(d,p) level. ΔE_1 is for aldehyde **1** and ΔE_2 is for aldehyde **2**.

		mPW1K	HF	B97-D
TS1-G	ΔE	17.4	23.7	12.8
In1-G	ΔE	19.7	22.2	13.0
In3-G	ΔE_1	20.5	14.1	12.6
	ΔE_2	16.7	9.9	7.9
TS4-G	ΔE_1	19.6	15.4	10.7
	ΔE_2	15.0	10.2	6.2
In4-G-A	ΔE_1	21.4	11.9	11.2
	ΔE_2	17.9	8.1	7.2
TS5-G-A	ΔE_1	18.2	12.0	8.0
	ΔE_2	15.7	9.0	5.1

These results suggest the compatibility in trends while comparing the energetic of mPW1K/6-31+G(d,p)//HF/6-31+(d,p) level with those of HF/6-31+G(d,p) and B97-D/6-31+G(d,p)//HF/6-31+(d,p) levels.





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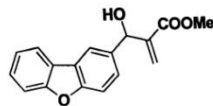
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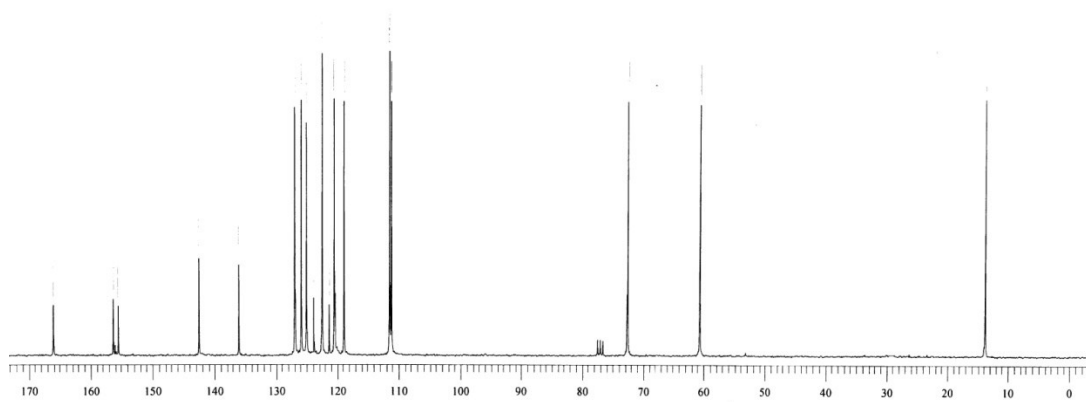
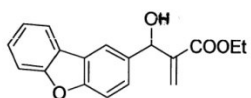
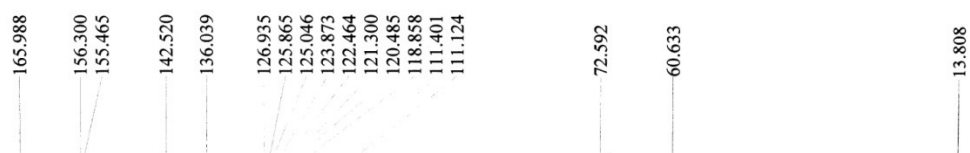
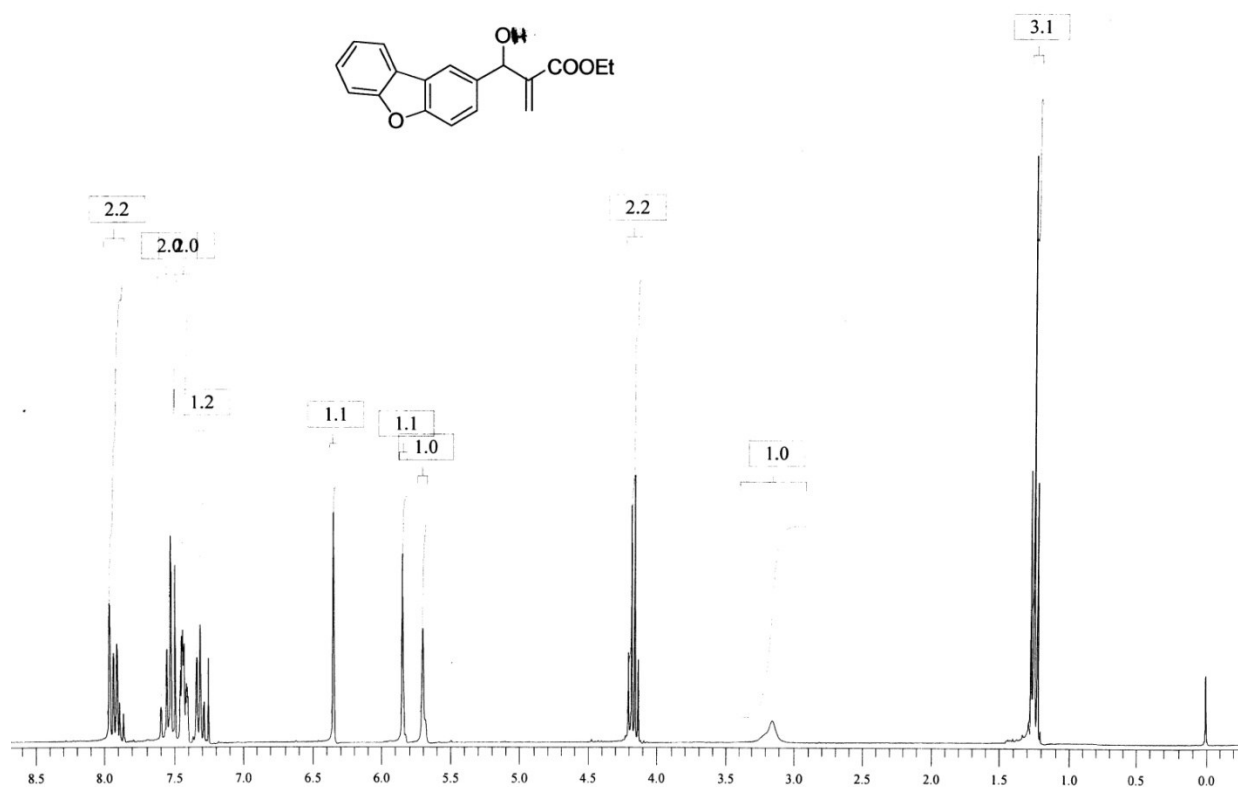
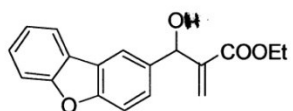
Elemental composition calculator



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Min DBE: -0.5000 Max DBE: +50.0000
Electron state: OddAndEven
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Add water: N/A
Add proton: N/A
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	Elements	Min Number	Max Number
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2	H	0	22
3	Na	0	1
4	O	0	5

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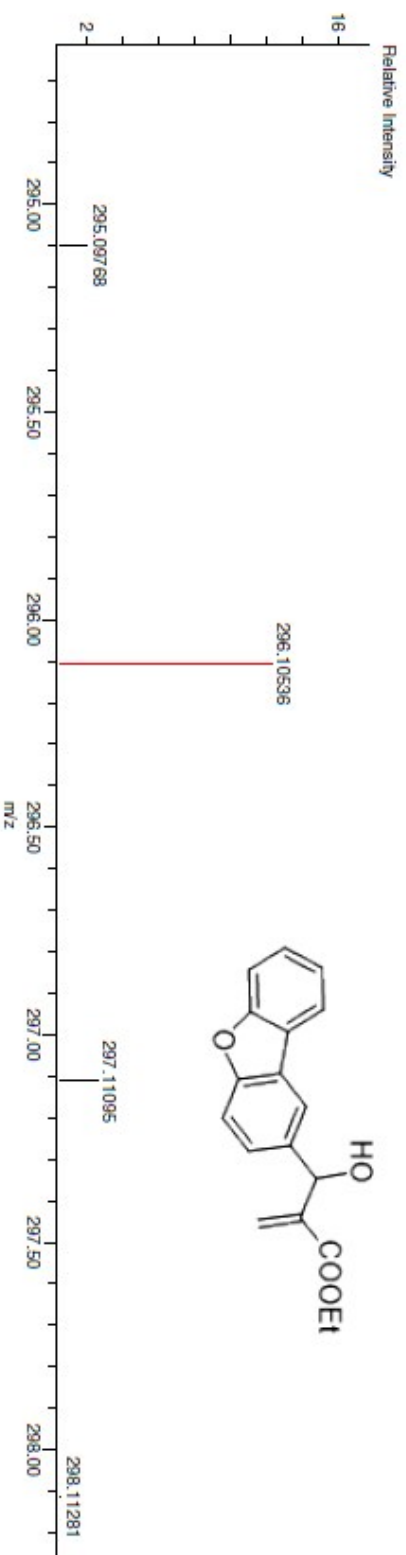
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NATIONAL CENTRE FOR MASS SPECTROMETRY

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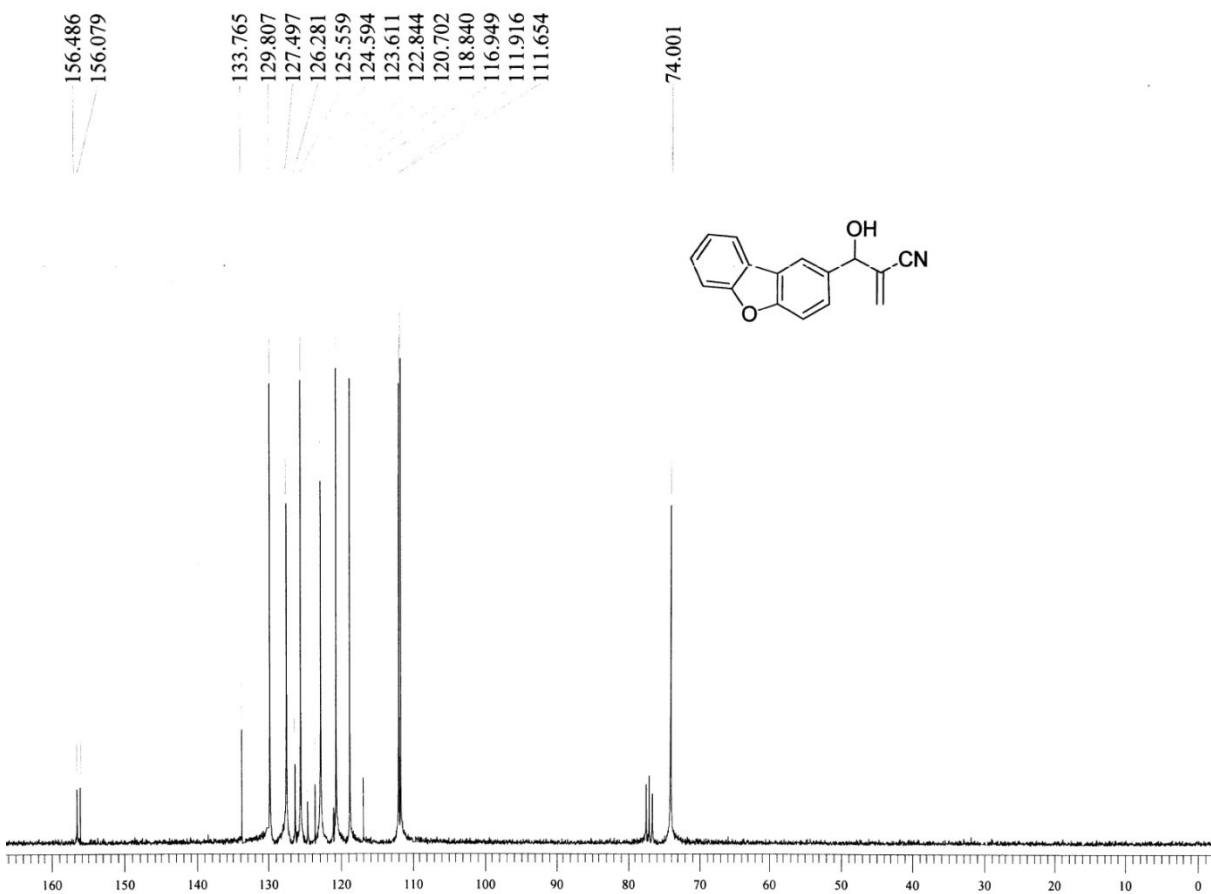
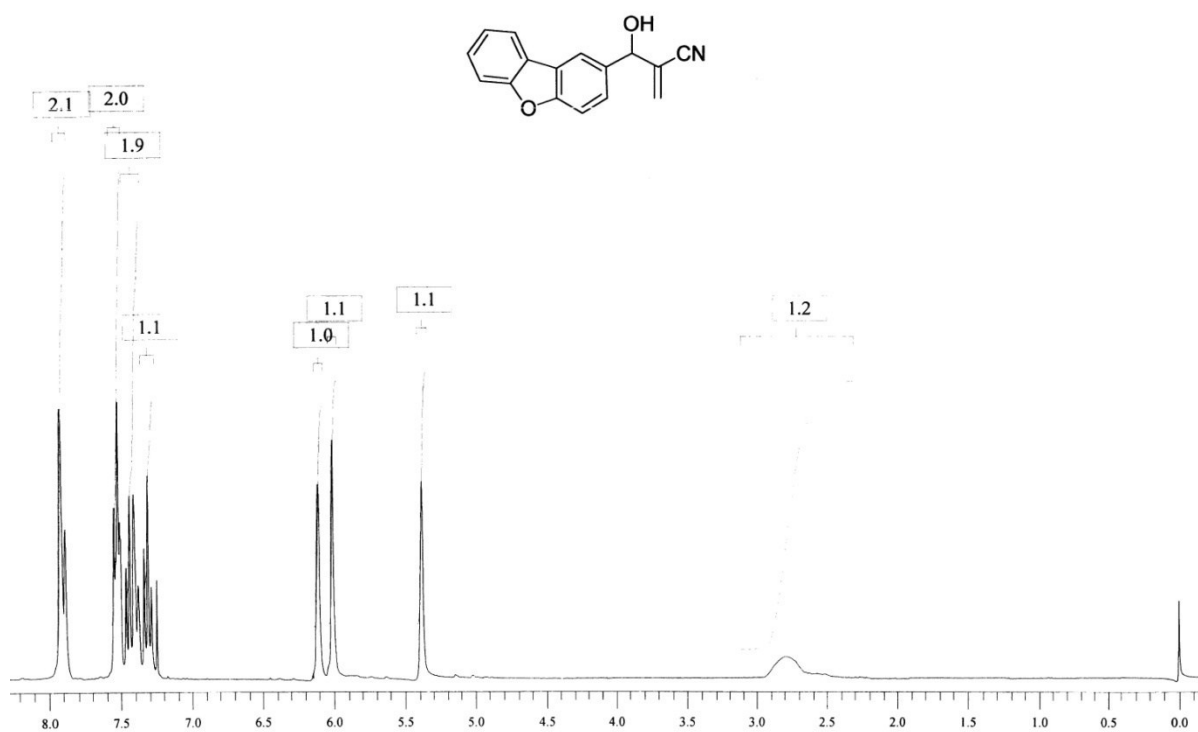
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Created by:JEOL

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Unsaturation Number:1.5 .. 50.0 (Fraction:Both)



Mass	Intensity	Calc. Mass	Mass Difference (mmu)	Mass Difference (ppm)	Possible Formula	Unsaturation Number
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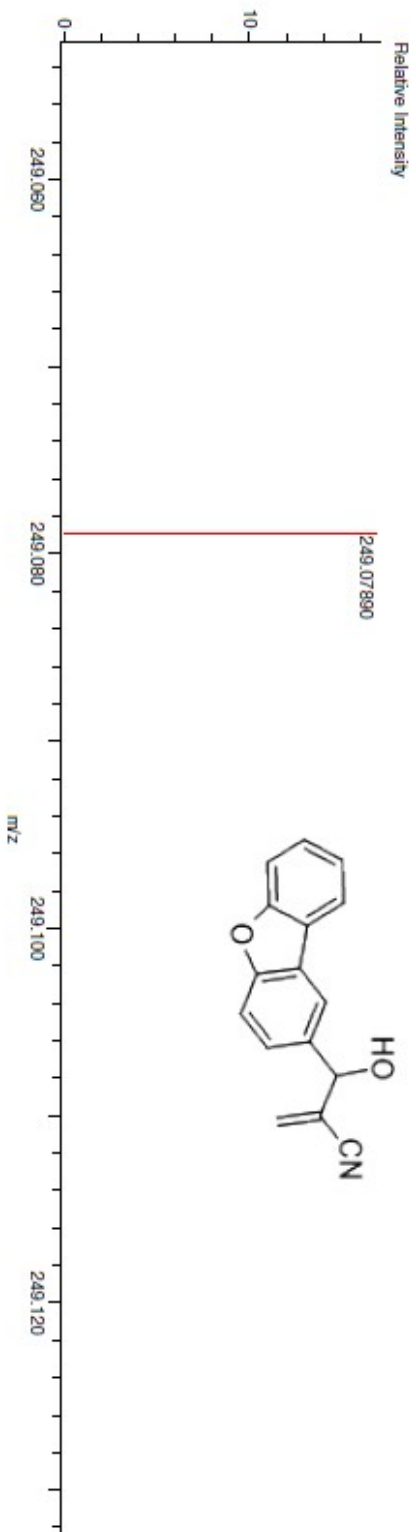
CSIR-INDIAN INSTITUTE OF CHEMICAL TECHNOLOGY
NATIONAL CENTRE FOR MASS SPECTROMETRY

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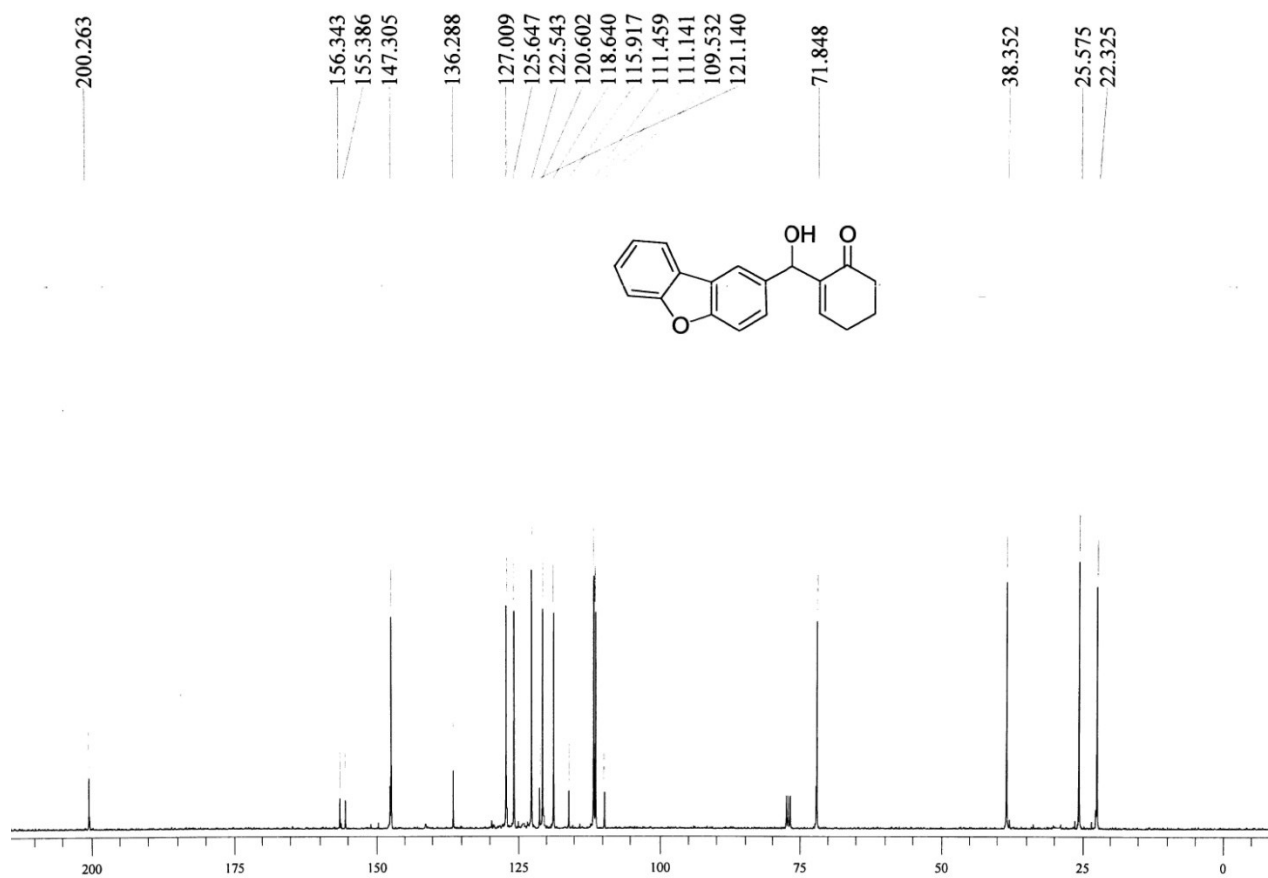
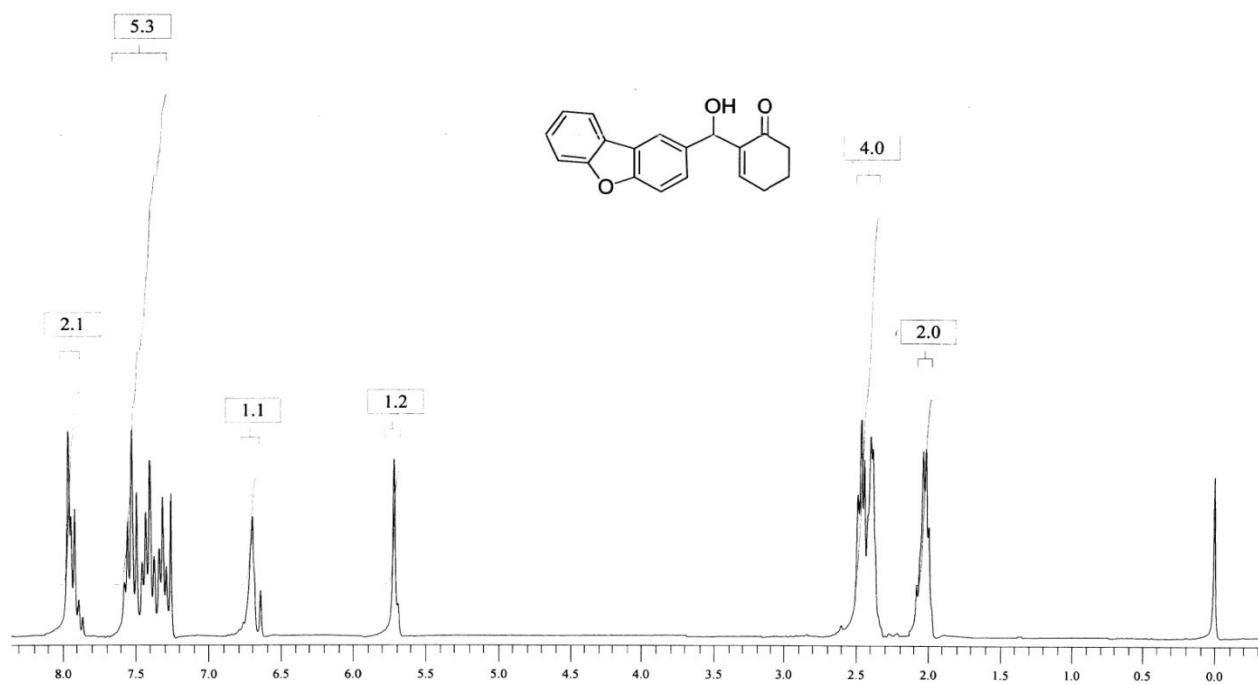
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Data:KS-BH2CYH-20-1-15

Sample Name:KS-BH2CYH-20-1-15, Dr K SRINIVAS,

EL-HR-MS, D/P-250

Description:

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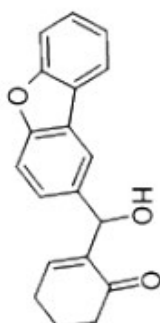
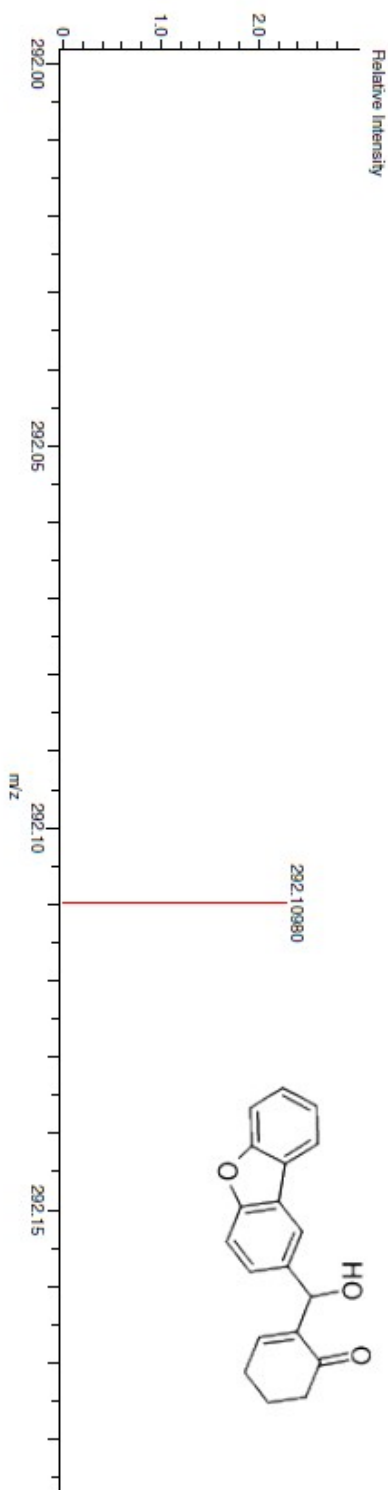
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Tolerance:1000.00(ppm)

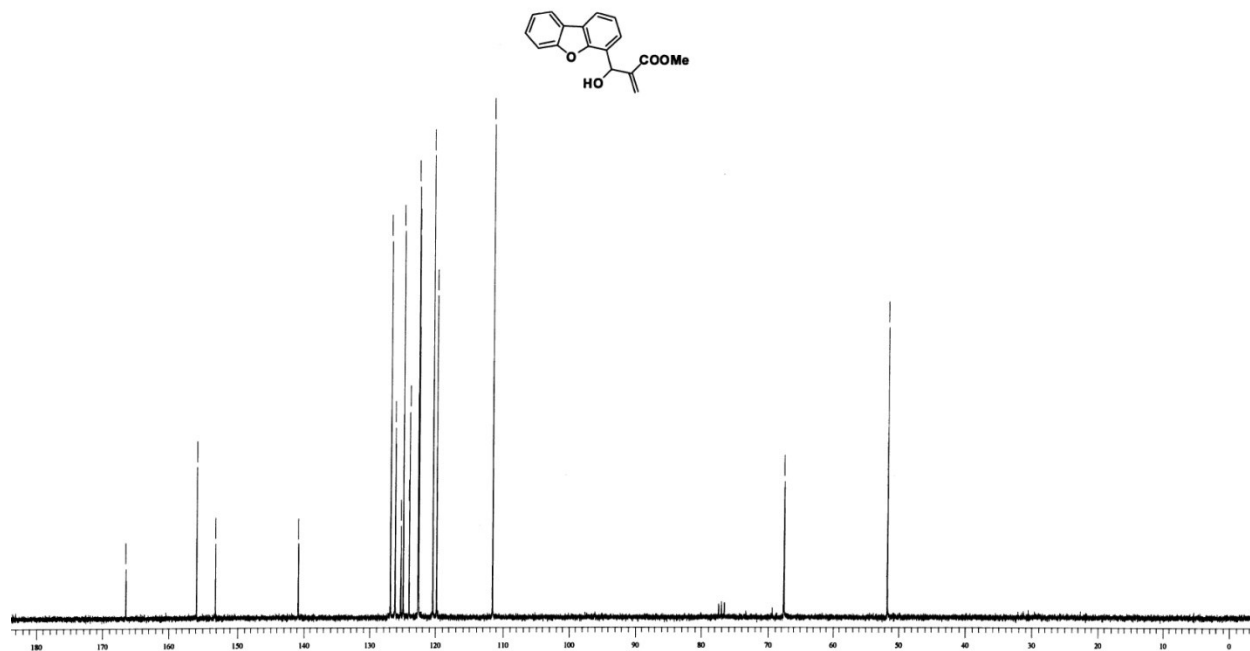
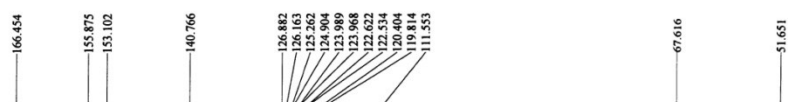
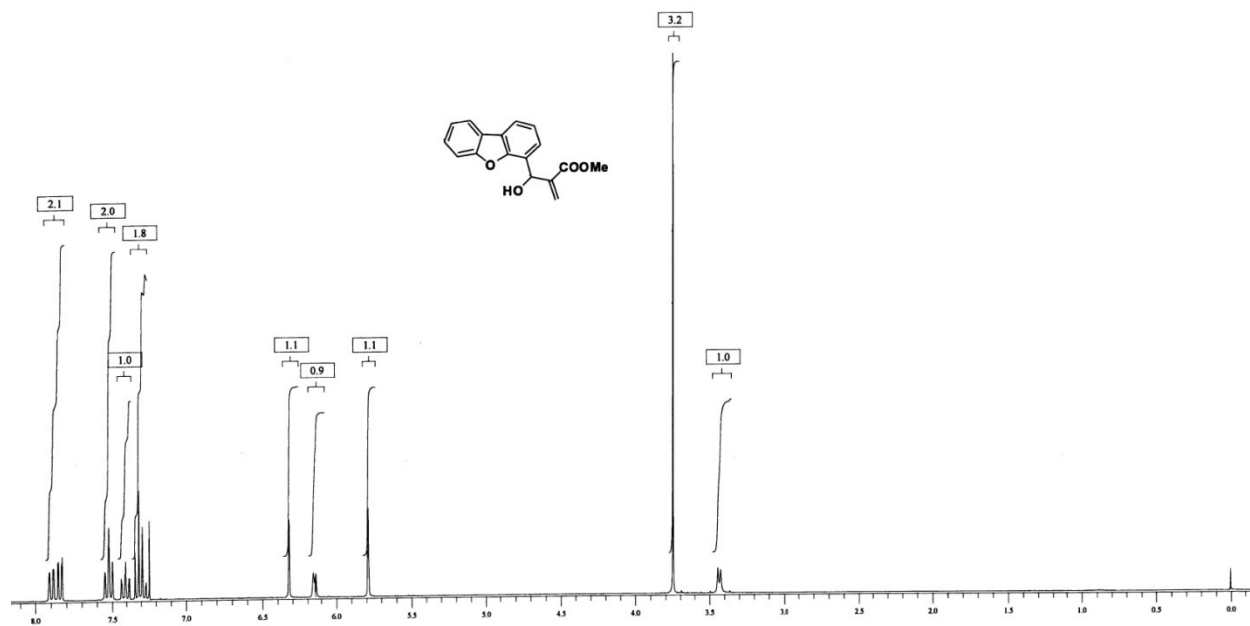
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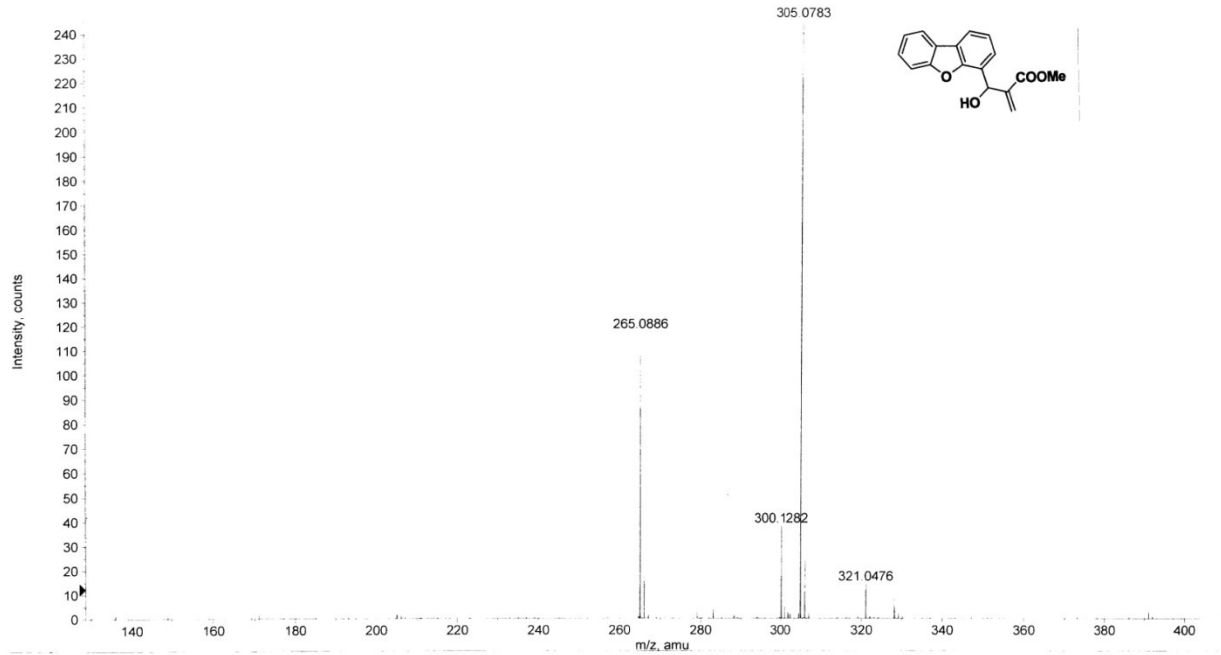
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292.10980	16043.12	292.10994	-0.14	-0.49	¹² C ₁₉ ¹ H ₁₆ ¹⁶ O ₃	12.0





1. 28FEB2011

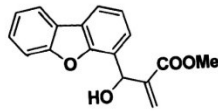
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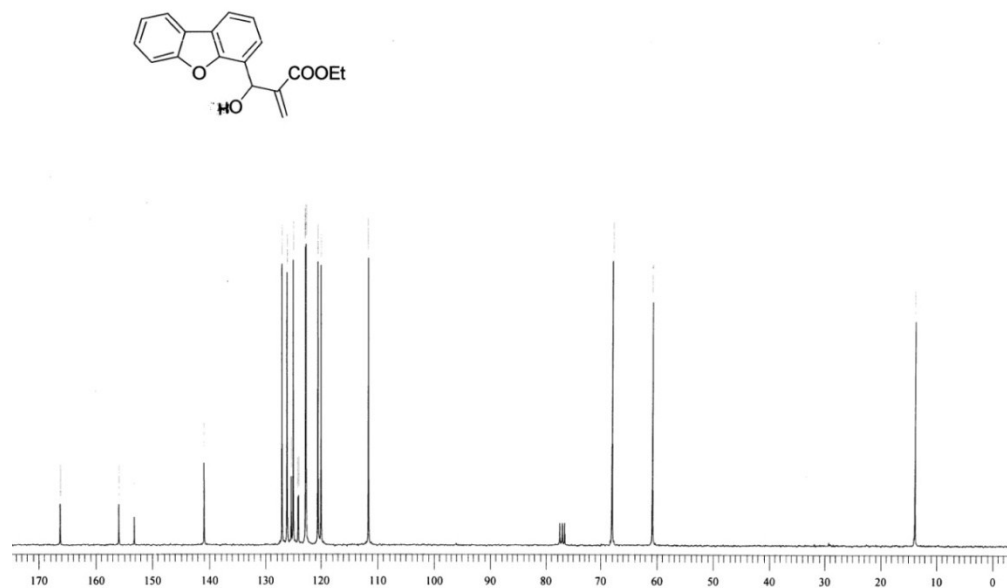
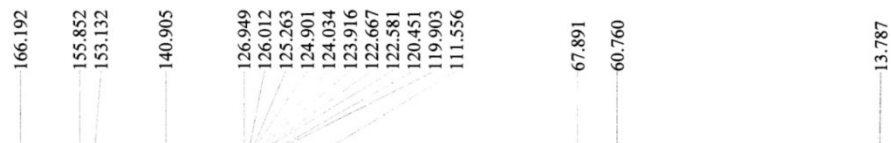
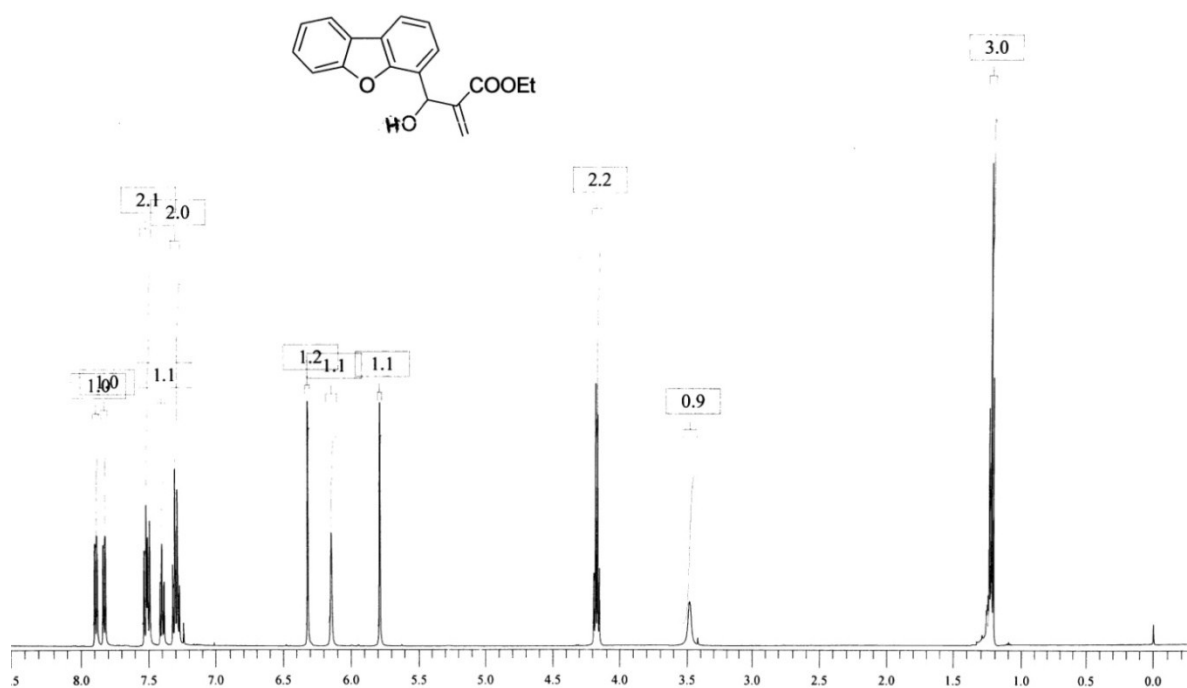
Elemental composition calculator



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Min DBE: -0.5000 Max DBE: +50.0000
Electron state: OddAndEven
Num of charges: 0
Add water: N/A
Add proton: N/A
File Name: 28FEB2011.wiff

	Elements	Min Number	Max Number
1	C	0	31
2	H	0	28
3	N	0	1
4	Na	0	1
5	O	0	5

	Formula	Calculated m/z (amu)	mDa Error	PPM Error	DBI
1	C17 H14 O4 Na	305.0789	-0.6788	-2.2250	10.



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Sample Name:KS-BH1EARNI-20-1-15, Dr K SRINIVAS,

EL-HR-MS, DIP-250

Description:

Ionization Mode:EI+

History: Determine m/z[Peak Detect[Centroid,1,Aveq],Smooth[3],Correct Baseq],Smooth[3],Average(MS[1] 0.47,0.67)

Charge number:1

Tolerance:1000.00(ppm)

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Unsaturation Number:1.5 .. 50.0 (Fraction:Both)

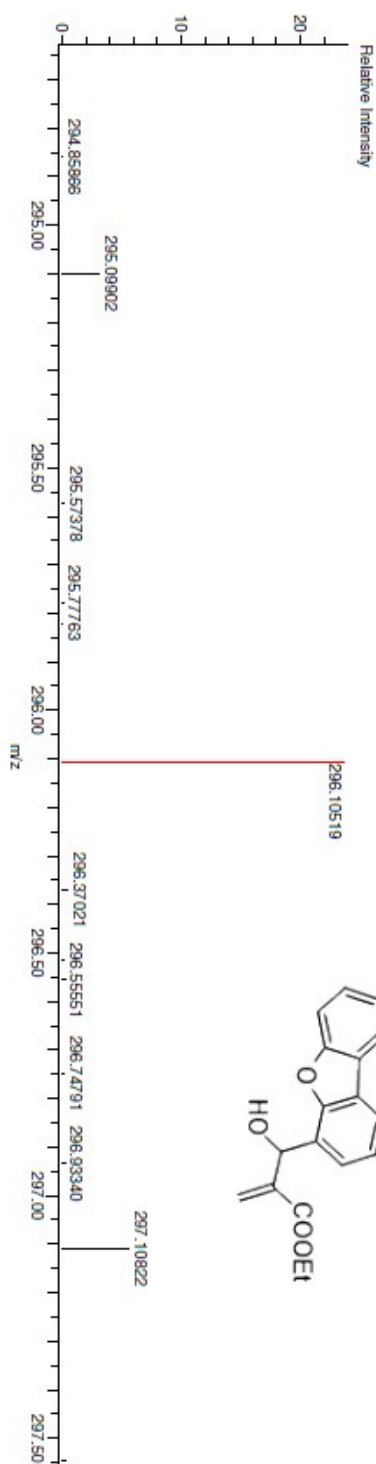
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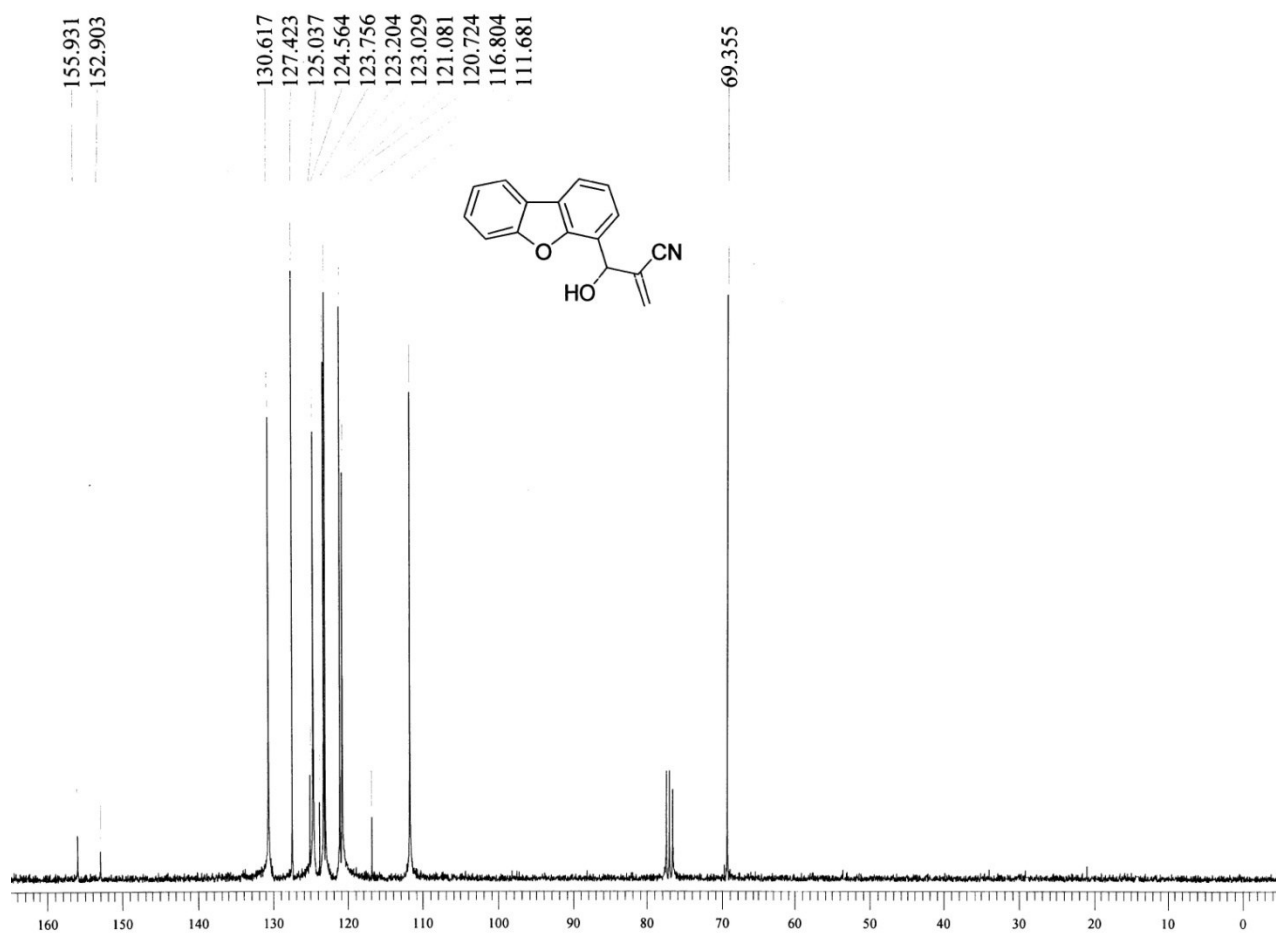
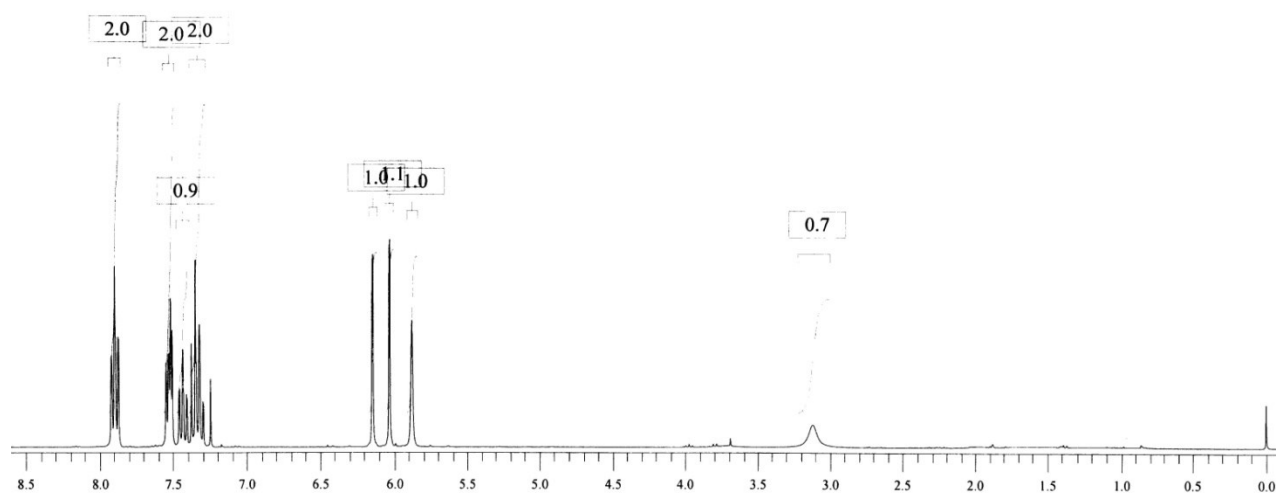
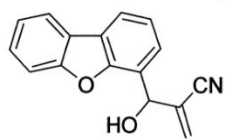
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Sample Name: KS-BH1ACRN-20-1-15, Dr K SRINIVAS,

EH-R-MS, DIP-260

Description:

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Unsaturation Number: 1.5 .. 50.0 (Fraction: both)

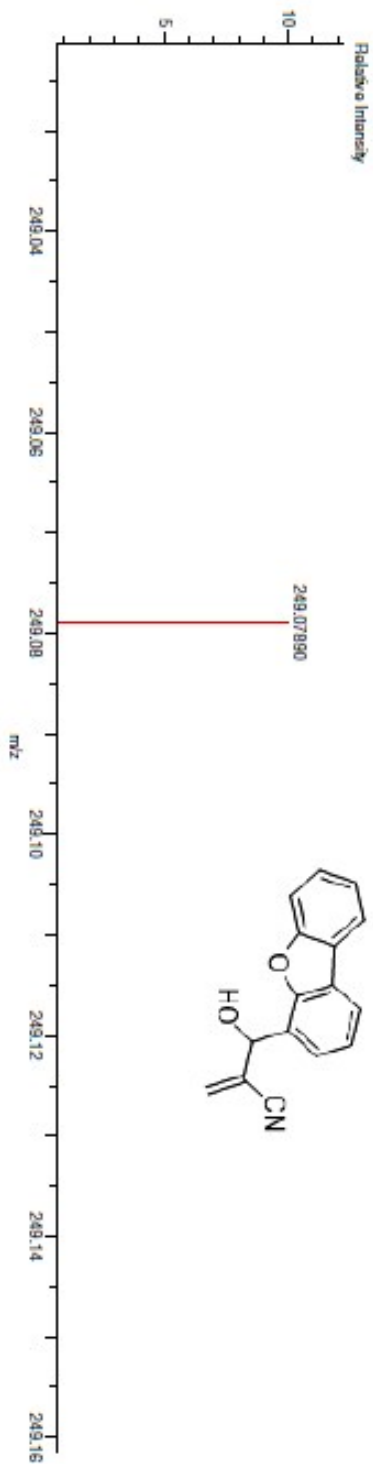
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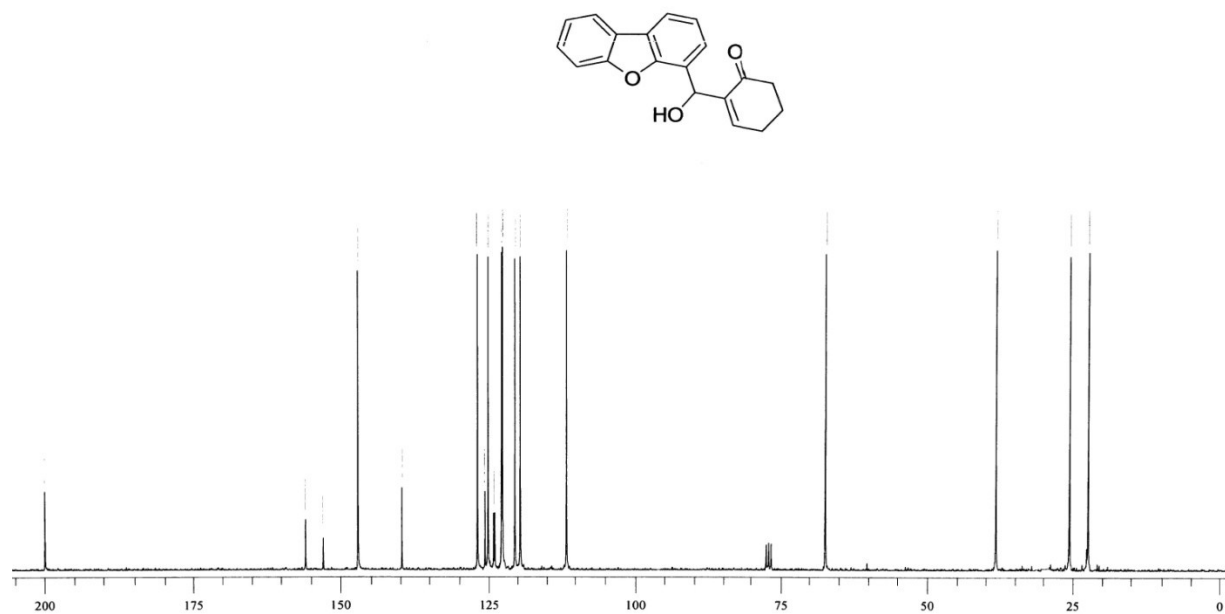
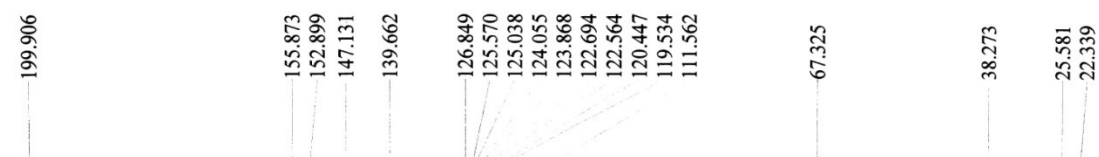
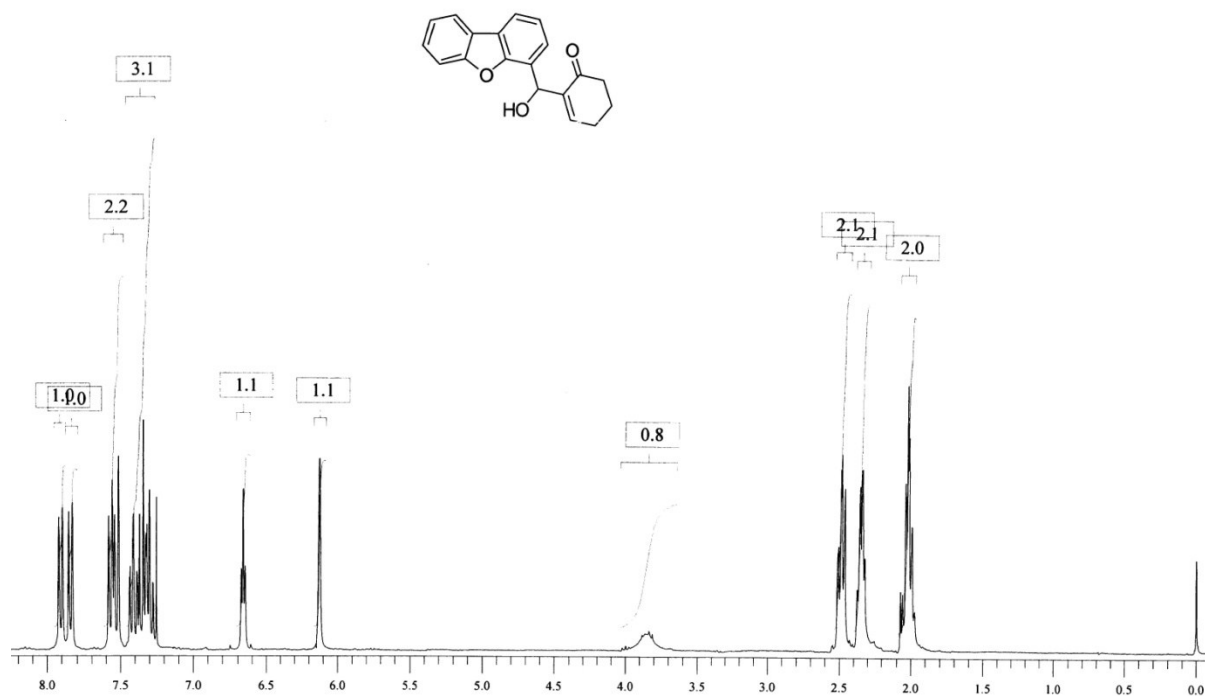
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249.07990	69786.912	249.07998	-0.008	-0.31	¹² C ₁₆ ¹ H ₁₁ ¹⁴ N ₁ ¹⁶ O ₂	12.0



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Data:KS-BH12CYH-20-1-15

Sample Name:KS-BH12CYH-20-1-15, Dr K SRINIVAS,

EL-HR-MS, DIP-250

Description:

Ionization Mode:EI+

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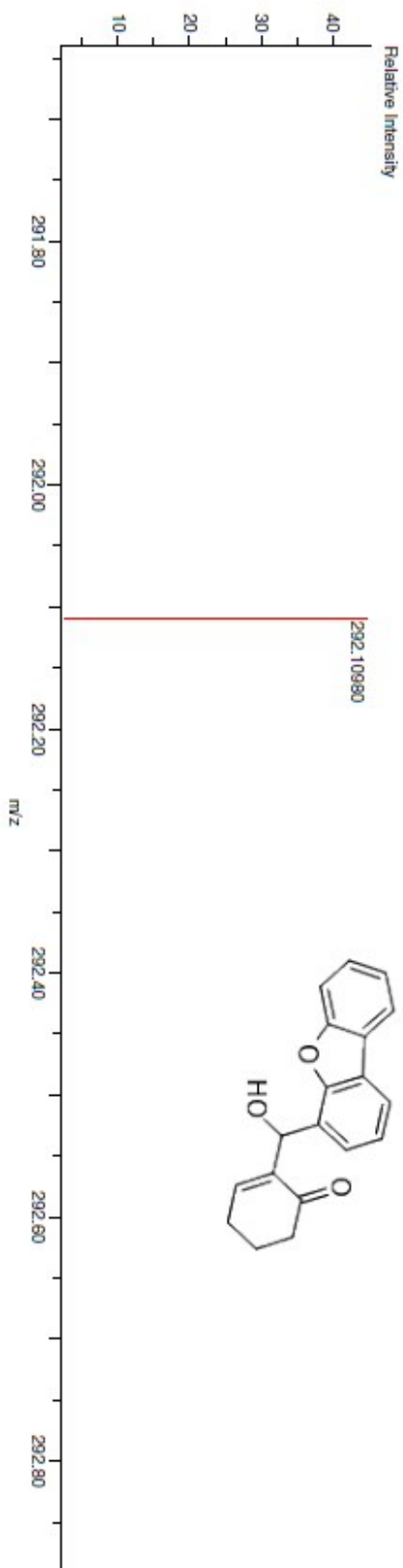
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