Supplementary Information.

Poly(ω-BromoalkyInorbornenes-*co*-norbornene) by ROMP-Hydrogenation:

a Robust Support Amenable to Post-Polymerization Functionalization

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- 1- Additional experimental data.
- 2- Figures showing NMR spectra of polymers.
- 3- Figures showing IR spectra of polymers.
- 4- SEM Images for polymers 4a and 4b.

1- Additional experimental data.

Determination of reactivity ratios.

Copolymerizations of norbornene and 1 were carried out at the monomer feed ratios f = NB/1 shown in Tables S1 and S2, using 0.72 mmol of **1a** and 0.55 mmol of **1b** in each experiment and following the conditions of Table 1 in the text. After 60 min the reactions were quenched by pouring the mixture onto methanol (80-100% conversion). The composition of the copolymers (F) was determined by quantitative analysis of the bromo content in the material. The composition data of the copolymers (F) obtained for the copolymerization at different monomer feed ratios (f) were fitted to the Finemann-Ross equation $f(1-F)/F = r_2 - (f^2/F) r_1$ (Figures S1 and S2).¹

a) Copolymerization 1a and norbornene (NB):

Table S1

f (NB/1a)	F (NB/1a)	f(1-F)/F	f^2/F
0.94	1.36	-0.25	0.650
1.87	3.08	-1.26	1.135
3.8	6.03	-3.17	2.395
7.61	11.97	-6.97	4.838

 $r_{1a} = 0.65 r_{NB} = 1.58.$

¹ J. M. G. Cowie, *Polymers: Chemistry & Physics of Modern Materials*: Chapman & Hall: Cheltenham, 1991.



Figure S1.

b) Copolymerization 1b and norbornene (NB):

Table S2.			
f (NB/1b)	F (NB/1b)	f(1-F)/F	f²/F
1	1.60	-0.38	0.625
2	3.16	-1.37	1.266
4	6.50	-3.38	2.462
8	13.87	-7.42	4.614

 $r_{1b} = 0.84$ y $r_{NB} = 1.78$.



Figure S2.

Reaction of homopolymer 3a (x/y = 0) with 2nd generation Grubbs' catalyst 2.

Homopolymer ROMP-PNBCH₂Br (**3a**, 80 mg, 0.43 mmol of Br) was dissolved in CH₂Cl₂ (4.8 mL) under a nitrogen atmosphere. A solution of **2** in CH₂Cl₂ (0.49 mL, 8.84 x 10^{-3} M, 4.3 x 10^{-3} mmol) was added dropwise. The reaction mixture was stirred for 24 h at room temperature. The reaction mixture was poured onto MeOH (30 mL), the CH₂Cl₂ was evaporated and the remaining solid was filtered, washed with MeOH (3 x 10 mL) and air-dried. Off-white powder, 48.7 mg, 61 % yield.

The reaction with the homopolymer ROMP-PNB(CH₂)₄Br **3b** (x/y = 0) was carried out in the same way.

The polymer properties before and after exposure to Grubbs' catalyst are shown in Table S3. After 24 h the polymers recovered were shorter and more polydisperse, showing that secondary metathesis is taking place.²

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Polymer	M _n	M_w/M_n	Trans:cis ratio
3a, before	20291	2.04	2.31
3a , after exposure	3125	2.84	3.69
3b , before	39398	2.17	2.15
3b , after exposure	6591	3.13	2.26

Reaction of copolymer 3a (x/y = 1.1) with $[PdBr(C_6F_5)(NCMe)_2]$.

ROMP copolymer **3a** (x/y = 1.1, 0.006 g, 0.043 alkene equivalents) and $[PdBr(C_6F_5)(NCMe)_2]$,³ (0.002 g, 0.0046 mmol) were dissolved in CDCl₃ (0.6 mL) in a 5 mm NMR tube. The reaction was monitored at room temperature by ¹⁹F NMR. Figure 3 in the text shows the spectra obtained at 10 min, 2 h, and 24 h.

² H. D. Maynard, R. H. Grubbs, *Macromolecules*, 1999, **32**, 6917.

³ A. C. Albeniz, P. Espinet, C. Foces-Foces, F. H. Cano, *Organometallics* 1990, **9**, 1079.

Determination of the % of Br substitution in Table 4.

The maximum Nu functionalization in polymers **5-9** (mmol Nu/g) is given by the formula: Max mmolNu/g = $1000\alpha/\{1000 + \alpha (M_w(Nu)-79.904)\}$ where $\alpha = \text{mmol Br/g}$ of the starting polymer **4**.

Since the Br substitution is high in all cases, the % can be estimated by the ratio: {(residual mmol Br/g in the polymer)/ Max mmolNu/g}100.

2- NMR spectra of polymers.

¹³C NMR 500 MHz 3a : ROMP-PNB-NBCH₂Br (x/y = 4.29)











4.0 f1 (ppm)



¹H NMR 500 MHz 3b : ROMP-PNB-NB(CH₂)₄Br (x/y = 7.09)











¹³C NMR 400 MHz 4a : ROMPH-PNB-NBCH₂Br



¹H NMR 400 MHz 4a : ROMPH-PNB-NBCH₂Br









¹³C NMR 400 MHz 4b : ROMPH-PNB-NB(CH₂)₄Br



¹H NMR 400 MHz 4b : ROMPH-PNB-NB(CH₂)₄Br



¹³C NMR 400 MHz 4b : ROMPH-PNB(CH₂)₄Br





¹³C NMR 400 MHz 5a : ROMPH-PNB-NBCH₂CN



¹H NMR 500 MHz 5a : ROMPH-PNB-NBCH₂CN



¹³C CP-MAS NMR 400 MHz 6a : ROMPH-PNB-NBCH₂OCOMe









¹³C NMR 500 MHz 7a : ROMPH-PNB-NBCH₂SPh



¹H NMR 500 MHz 7a : ROMPH-PNB-NBCH₂SPh







¹³C CP-MAS NMR 400 MHz 8b : ROMPH-PNB-NB(CH_2)₄N₃









¹¹⁹Sn NMR 400 MHz 9b : ROMPH-PNB-NB(CH₂)₄SnBu₂(*p*-C₆H₄OMe)



¹³C CP-MAS NMR 400 MHz 10a : ROMPH-PNB-NBCH₂triazole











¹³C CP-MAS NMR 400 MHz 13a : ROMPH-PNB-NBCH₂triazole







¹⁹F CP-MAS NMR 400 MHz 14a : ROMPH-PNB-NBCH₂triazole



* Spinning sidebands are present.





¹³C CP-MAS NMR 400 MHz 15b: ROMPH-PNB-NB(CH₂)₄triazole



3- IR spectra of polymers



FT/IR $3a: ROMP-PNBCH_2Br (x/y = 0)$











FT/IR 4b : ROMPH-PNB-NB(CH₂)₄Br















FT/IR 8b : ROMPH-PNB-NB(CH₂)₄N₃





FT/IR 10a : ROMPH-PNB-NBCH₂triazole



FT/IR 11a : ROMPH-PNB-NBCH₂triazole



FT/IR 12a : ROMPH-PNB-NBCH₂triazole



FT/IR 13a : ROMPH-PNB-NBCH₂triazole



%T

FT/IR 14a : ROMPH-PNB-NBCH₂triazole



FT/IR 15a : ROMPH-PNB-NBCH₂triazole



FT/IR 15b : ROMPH-PNB-NB(CH₂)₄triazole



4- SEM Images for polymers 4a and 4b.

Scanning Electron Microscopy (SEM) measurements have been carried out at the *Unidad de Microscopía Avanzada* of the *Parque Científico UVa* by means of an Environmental Scanning Electron Microscope (ESEM), model FEI-Quanta 200FEG provided with a Schottky-Field Emission filament. The SEM analyses were performed at Low Vacuum Mode using water vapour as auxiliary gas. This imaging mode allows working with non-conductive samples without any specific preparation or metallic coatings. The measurements were performed at a working pressure in the chamber ranging between 0.6-1 Torr (80-133 Pa) and an accelerating voltage of 7kV. The Secondary Electron SEM images were acquired with a Large Field Detector (LFD) which is the suitable one for working at Low Vacuum Mode.

SEM images of the aliphatic polymers show a different surface topography for polymers **4a** and **4b**. Whereas a granular texture is observed for **4a** (Figure S3, a and b), **4b** shows a smoother surface with large pores (Figure S3, c and d).



Figure S3. SEM images with different magnification for 4a (a and b) and 4b (c and d).