Supporting Information

For

A turn-on fluorescent probe for hydrogen sulfide and its application

in living cells

Miao Zhao, "Hua Li," Hui Li, "Qinglong Qiao," Cheng Cao, "Zhaochao Xu".*

^a Key Laboratory of Separation Science for Analytical Chemistry of CAS, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, Dalian 116023, China.

^b Hebei Vocational & Technical College of Building Materials, Hebei, China.

E-mail: zcxu@dicp.ac.cn

Supplementary figures and text:

Supplementary Figure S1-S3. NMR spectra of synthesized compounds.

Supplementary Figure S4. IR spectra of probe 1.

Supplementary Figure S5. Fluorescence spectra of 1 in the presence of H_2S and NBD-NH₂.

Supplementary Figure S6. Cell viability of probe $1 (5 \mu M)$ at different times in HT-29 cells.

Supplementary Figure S7. Images of HT-29 cells incubated with DAPI.







Figure S2.¹³C-NMR spectra of compound 1 in CDCl₃.







Figure S4.IR spectra of probe 1



Figure S5. Fluorescent emission spectra of 10 μ M compound **1** with 20 eq H₂S and 10 μ M DCDHF-NH₂ in aqueous solution (CH₃CN: PBS = 9:1, pH = 7.4). Excitation at 470 nm.



Figure S6. Cell viability of probe 1 (5 μ M) at different times in HT-29 cells.



Fig. S7. Images of HT-29 cells incubated with DAPI (2 μ M) at 37 °C. a) bright field. b) 2.0 μ M DAPI. c) merged images of b) and probe 1 incubated with 20 μ M H₂S.