

Supporting Information

Superior-Enhanced Antibacterial Activity of New “Composite” Biocides with both *N*-chloramine and Quaternary Ammonium Moieties

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1. ¹H and ¹³C NMR of compounds # 1-3, 5-9 and 10-12

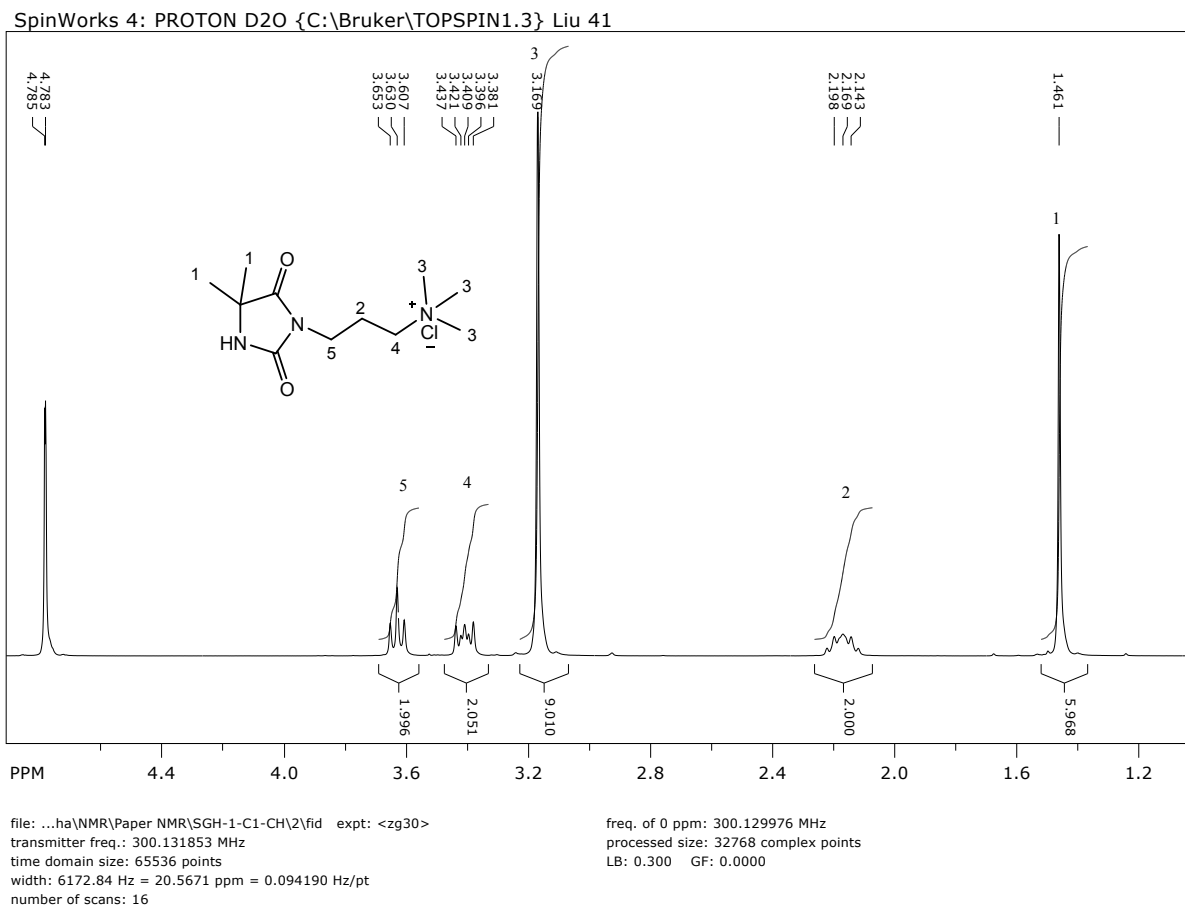
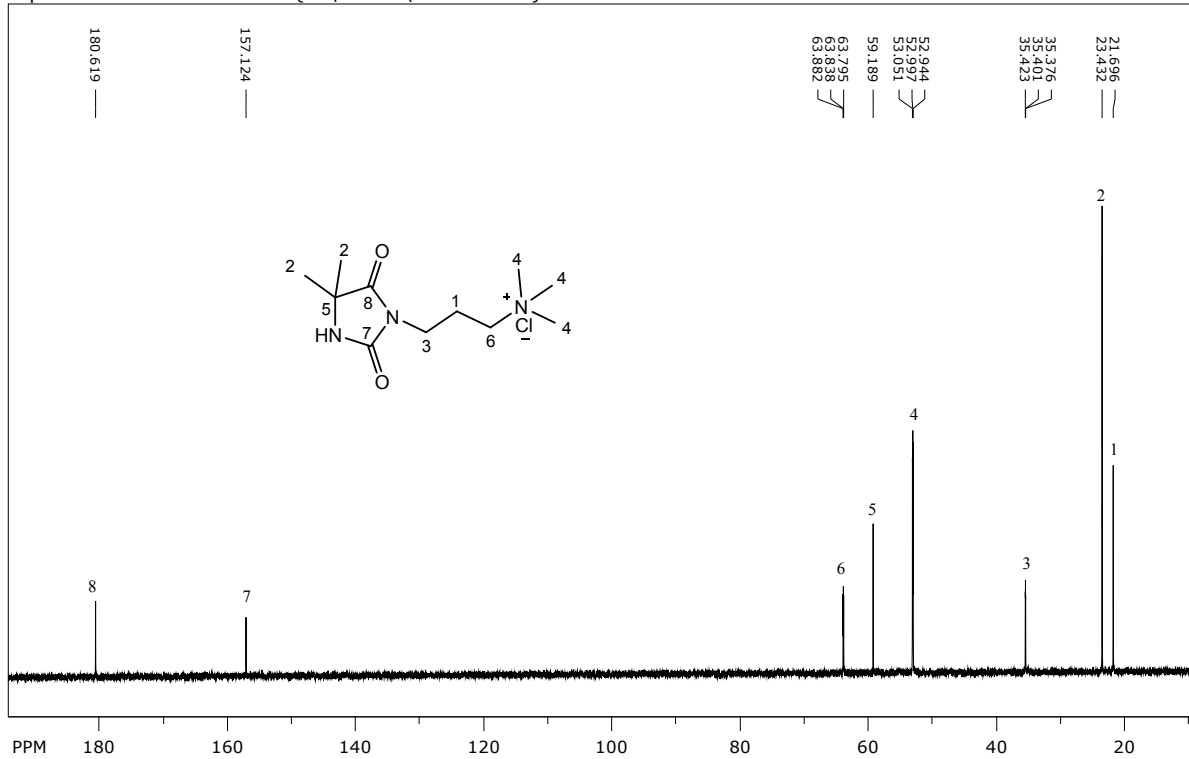


Figure S1. ¹H-NMR spectrum of compound#1 (D₂O).

SpinWorks 4: C13CPD D2O {C:\Bruker\TOPSPIN1.3} Liu 41



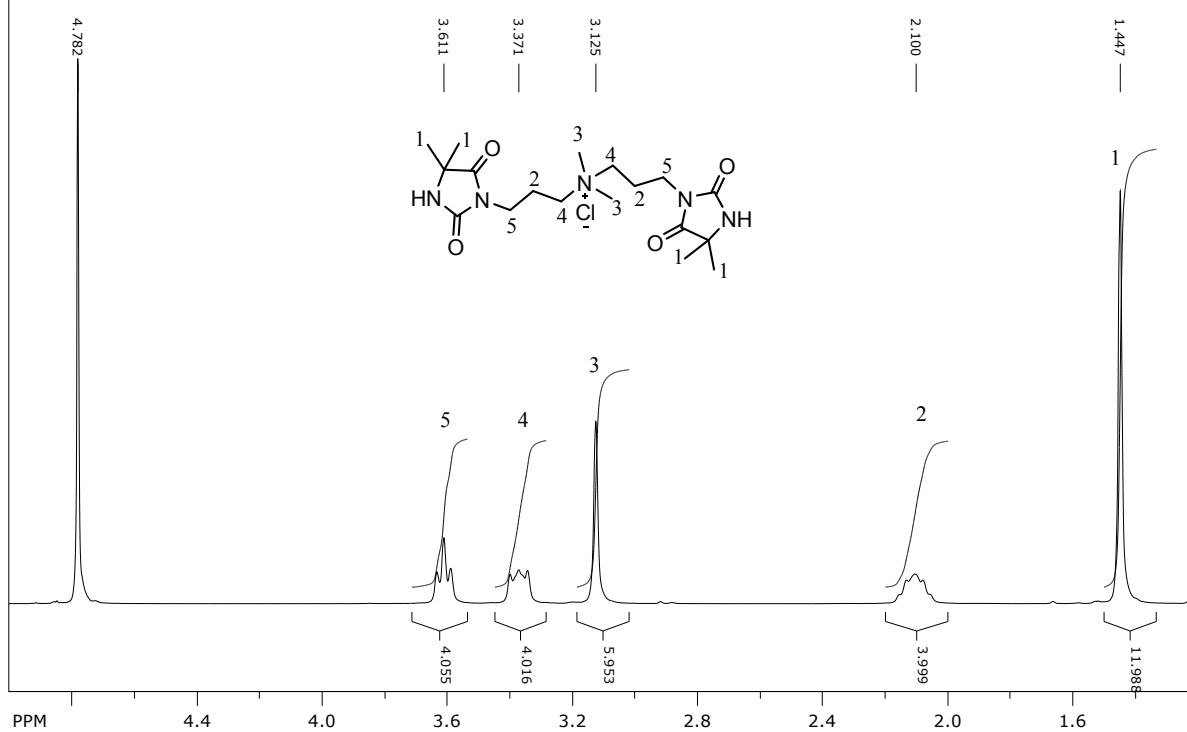
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number of scans: 1024

freq. of 0 ppm: 75.467749 MHz
processed size: 32768 complex points
LB: 1.000 GF: 0.0000

Figure S2. ¹³C NMR spectrum of **compound#1** (D₂O).

SpinWorks 4: SL-48-1

PROTON D2O {C:\Bruker\TOPSPIN1.3} student 58

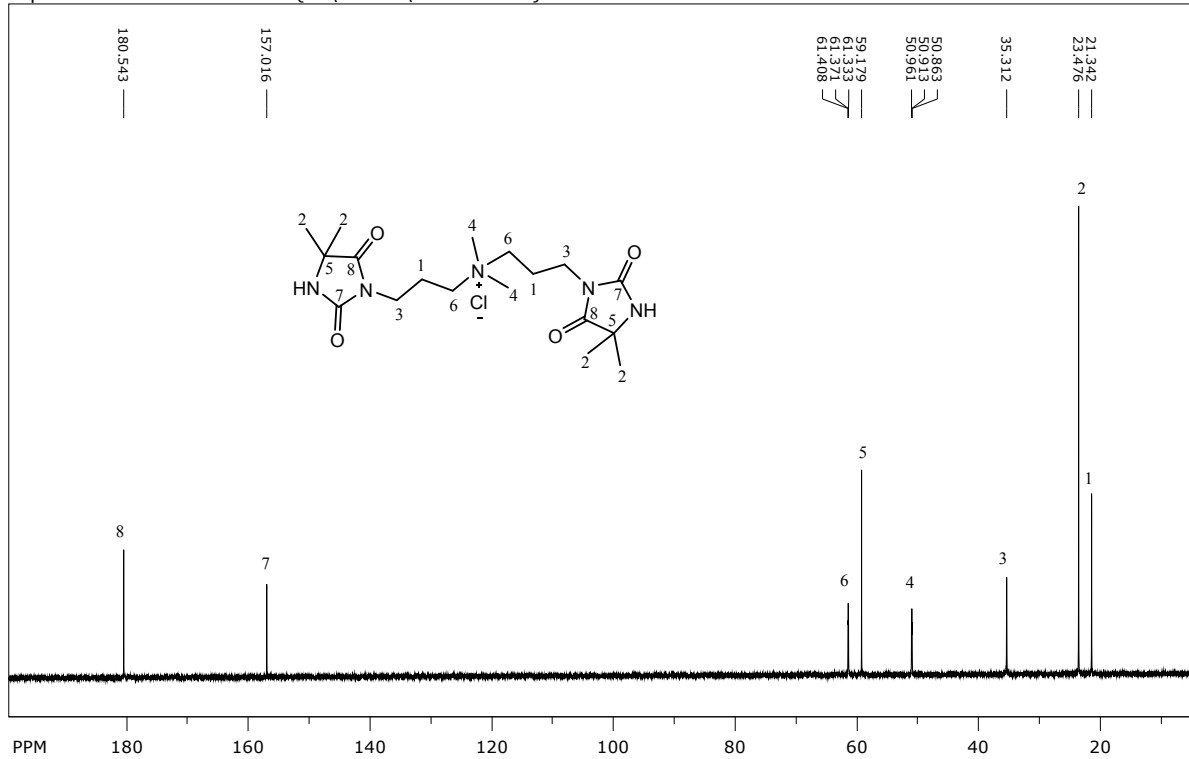


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number of scans: 16

freq. of 0 ppm: 300.129976 MHz
processed size: 32768 complex points
LB: 0.300 GF: 0.0000

Figure S3. ¹H-NMR spectrum of **compound#2** (D₂O).

SpinWorks 4: C13CPD D2O {C:\Bruker\TOPSPIN1.3} Liu 42



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number of scans: 1024

freq. of 0 ppm: 75.467749 MHz
processed size: 32768 complex points
LB: 1.000 GF: 0.0000

Figure S4. ¹³C NMR spectrum of **compound#2** (D₂O).

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 43

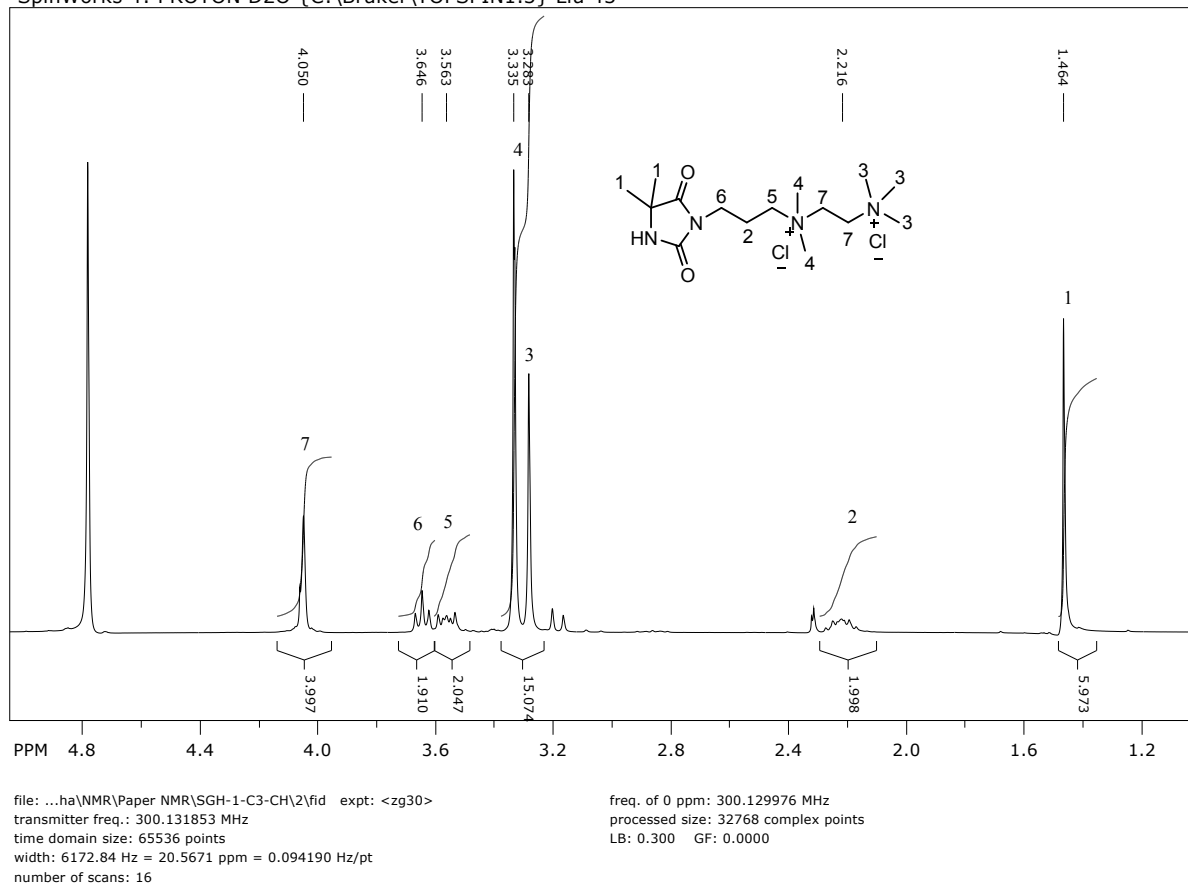
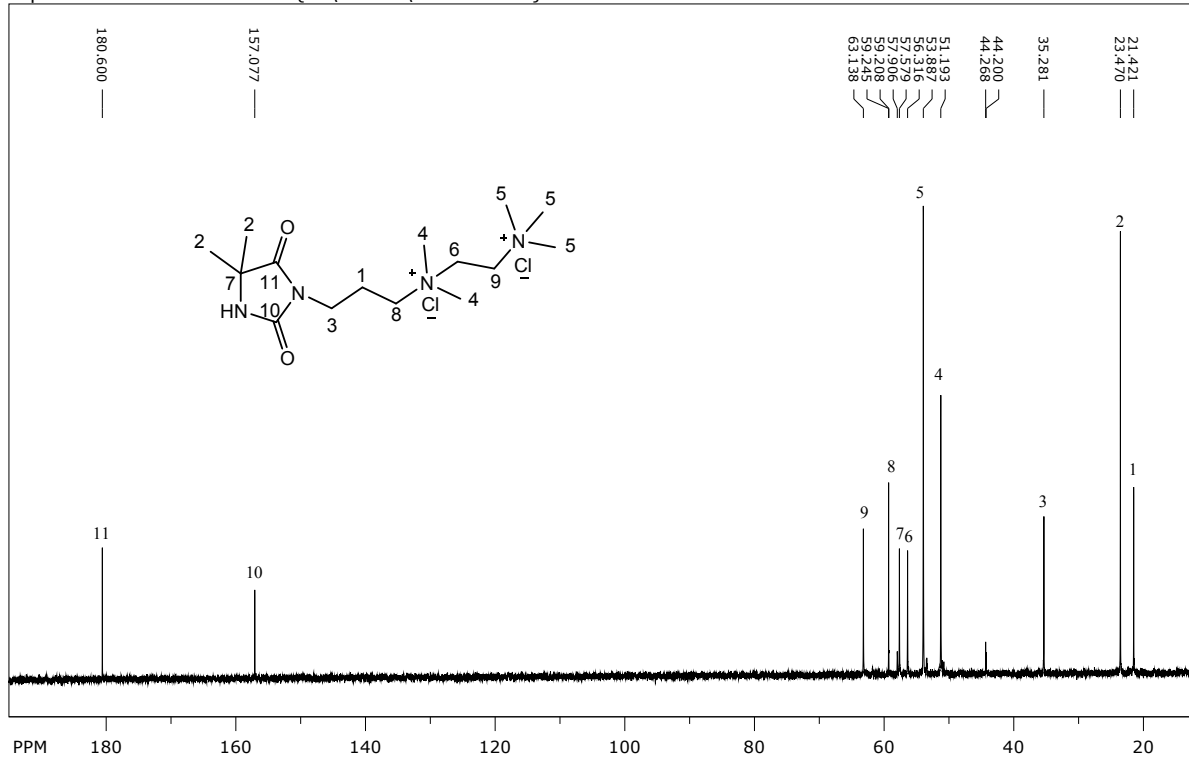


Figure S5. ¹H-NMR spectrum of **compound#3** (D₂O).

SpinWorks 4: C13CPD D2O {C:\Bruker\TOPSPIN1.3} Liu 43



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time domain size: 65536 points
width: 17985.61 Hz = 238.2980 ppm = 0.274439 Hz/pt
number of scans: 1024

freq. of 0 ppm: 75.467749 MHz
processed size: 32768 complex points
LB: 1.000 GF: 0.0000

Figure S6. ^{13}C NMR spectrum of **compound#3** (D_2O).

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 23

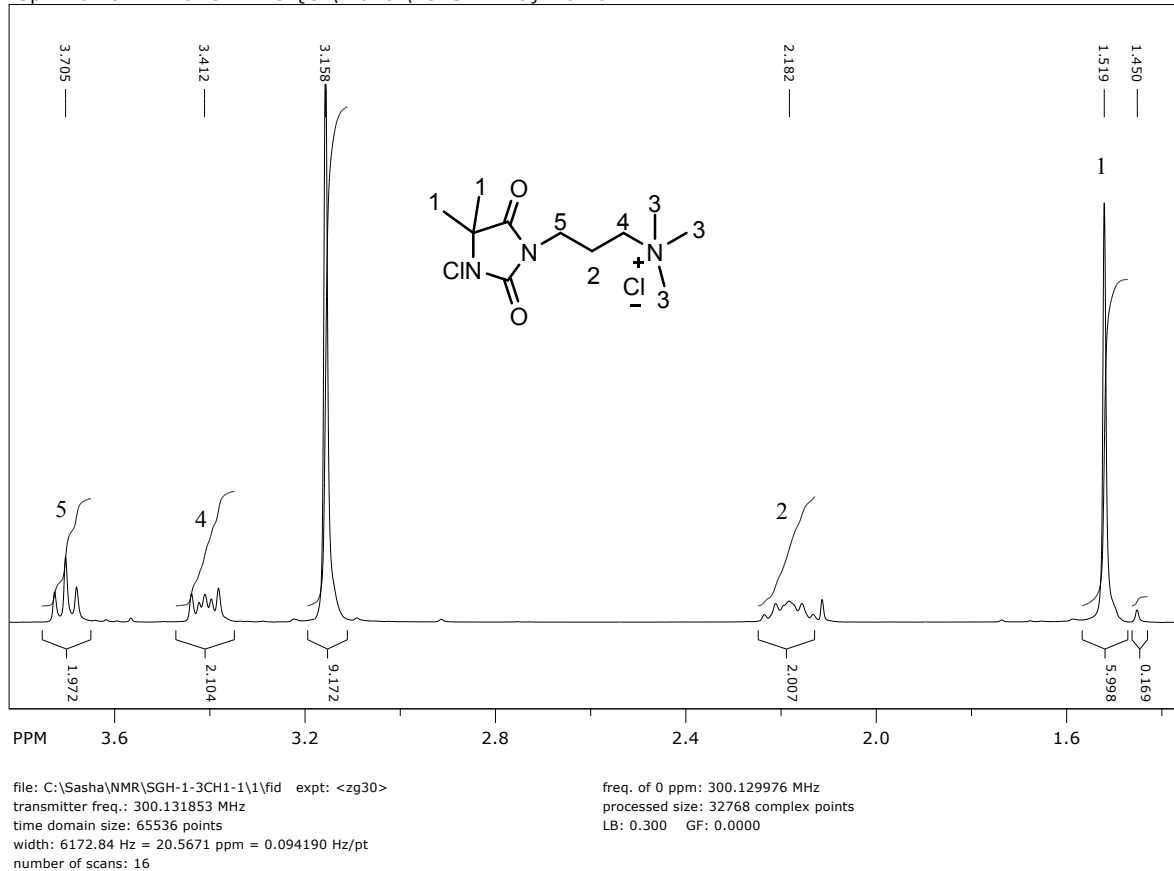


Figure S7. ^1H -NMR spectrum of **compound#4** (D_2O).

SpinWorks 4: C13CPD D2O {C:\Bruker\TOPSPIN1.3} Liu 44

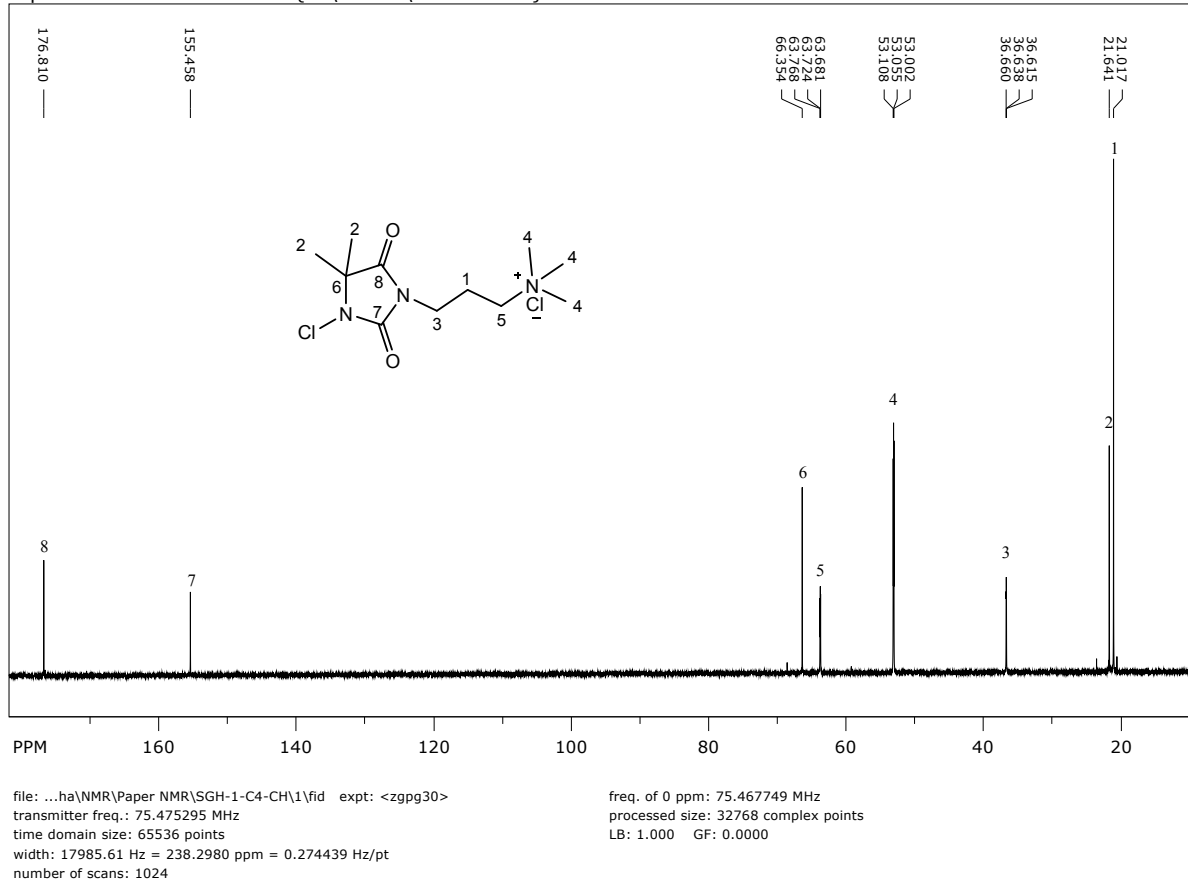


Figure S8. ¹³C NMR spectrum of **compound#4** (D₂O).

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 24

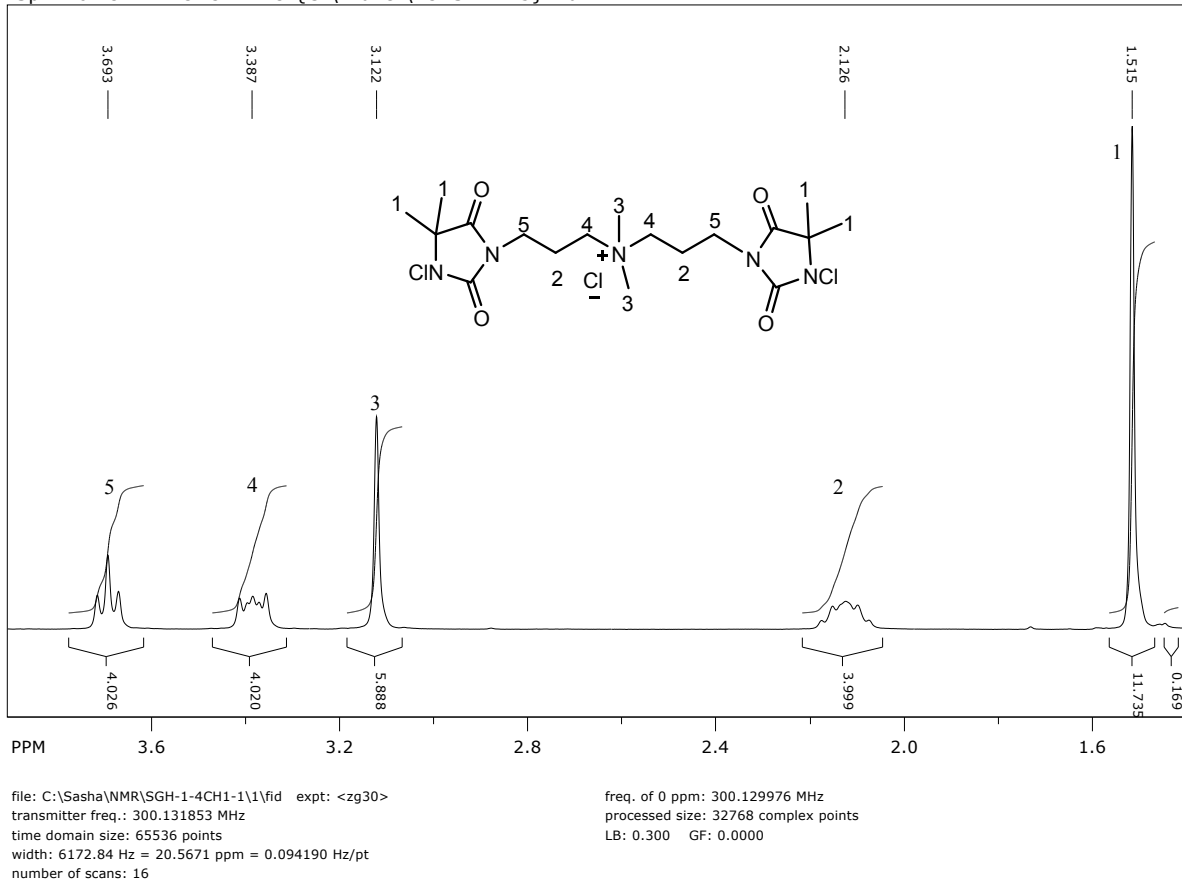
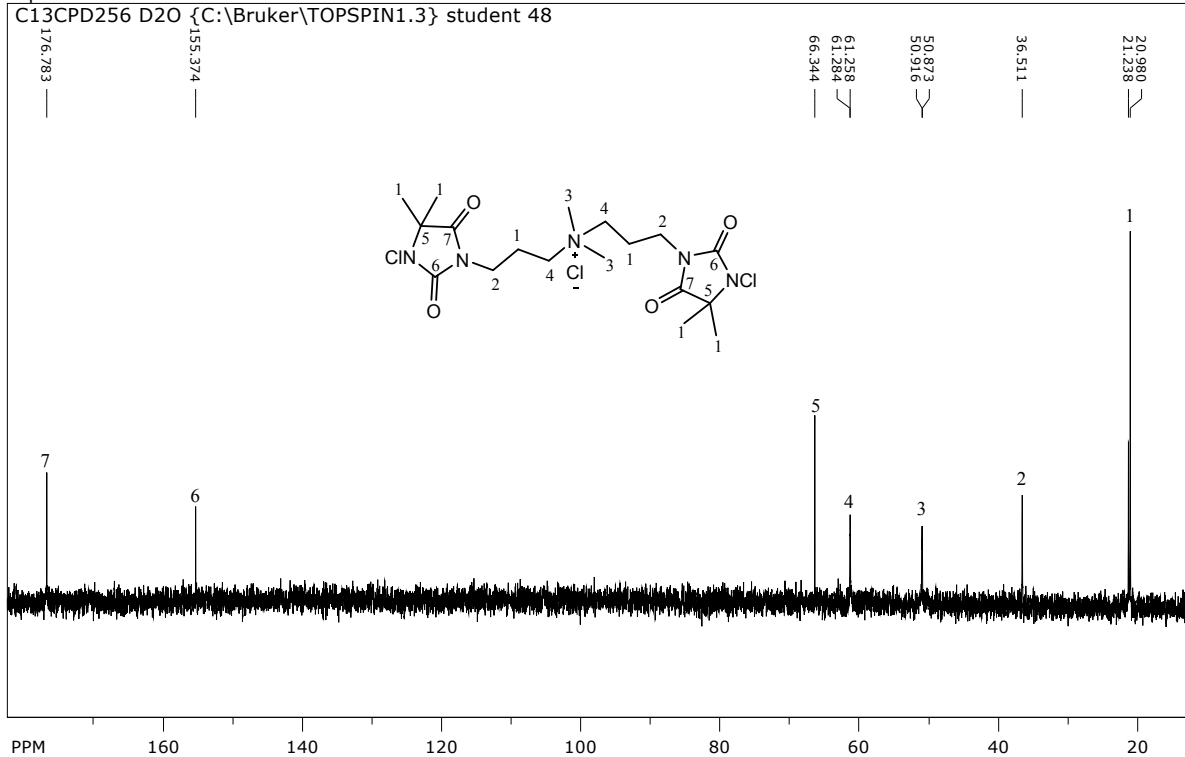


Figure S9. ¹H-NMR spectrum of compound#5 (D₂O).

SpinWorks 4: SL-77-2

C13CPD256 D2O {C:\Bruker\TOPSPIN1.3} student 48



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number of scans: 256

freq. of 0 ppm: 75.467749 MHz
processed size: 32768 complex points
LB: 1.000 GF: 0.0000

Figure S10. ^{13}C NMR spectrum of compound#5 (D_2O).

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 21

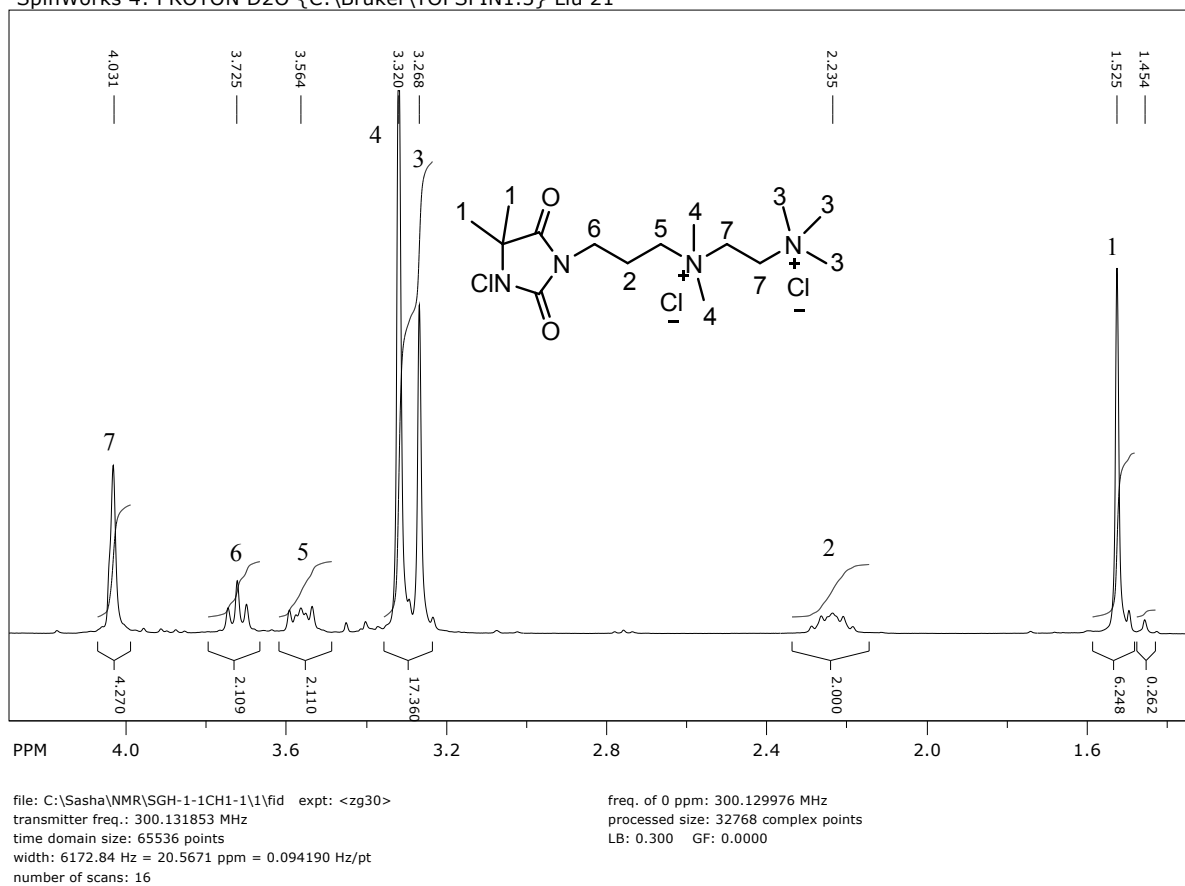
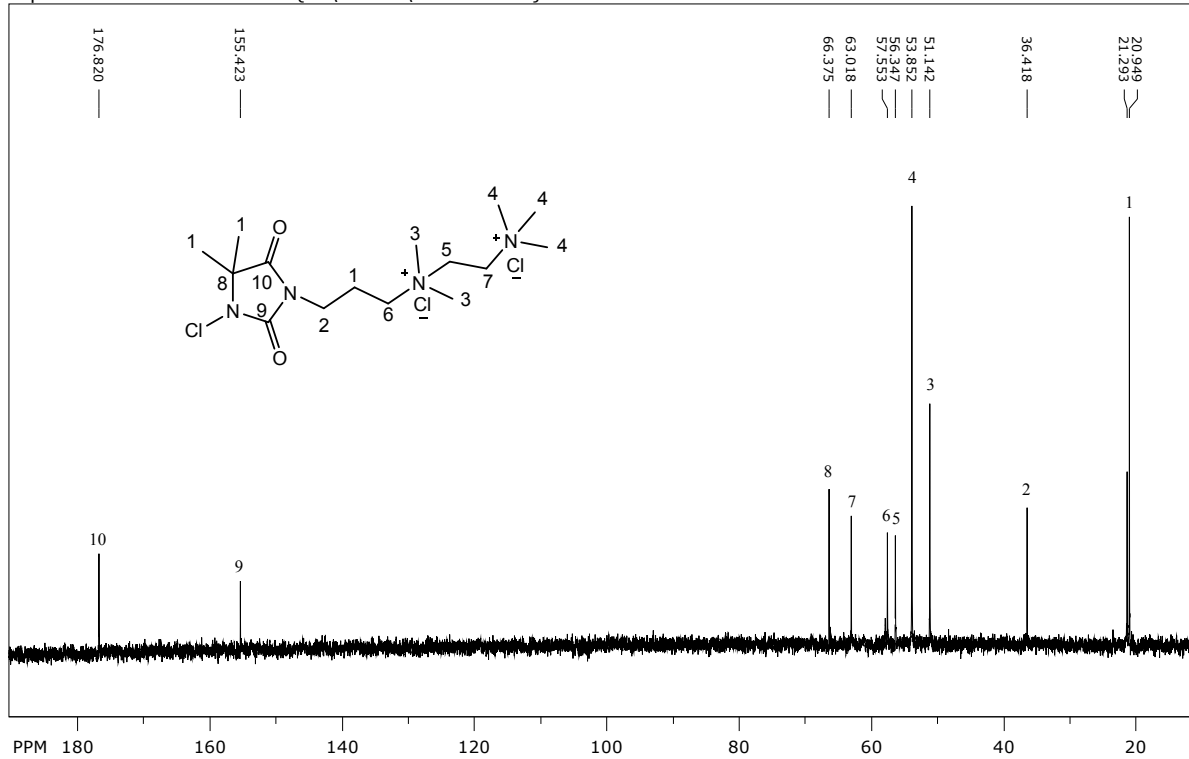


Figure S11. $^1\text{H-NMR}$ spectrum of **compound#6** (D_2O).

SpinWorks 4: C13CPD D2O {C:\Bruker\TOPSPIN1.3} Liu 46



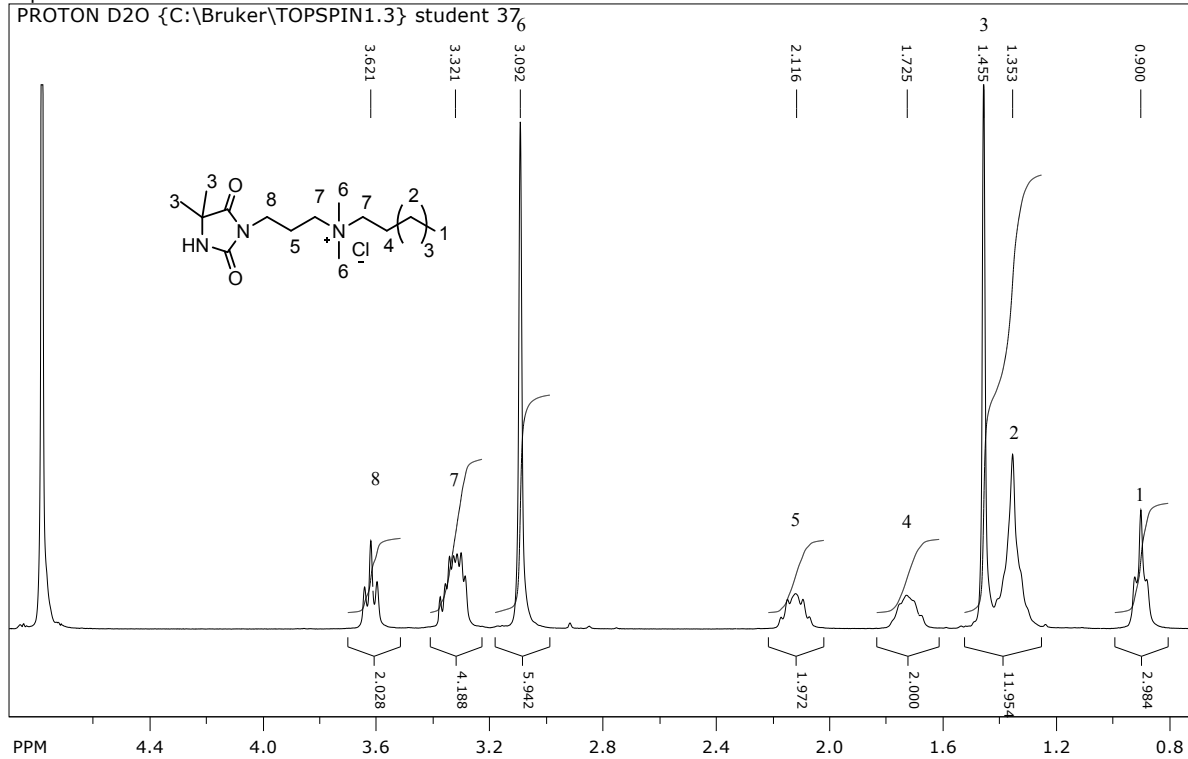
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number of scans: 1024

freq. of 0 ppm: 75.467749 MHz
processed size: 32768 complex points
LB: 1.000 GF: 0.0000

Figure S12. ¹³C NMR spectrum of compound#6 (D₂O).

SpinWorks 4: SL-135-1

PROTON D2O {C:\Bruker\TOPSPIN1.3} student 376



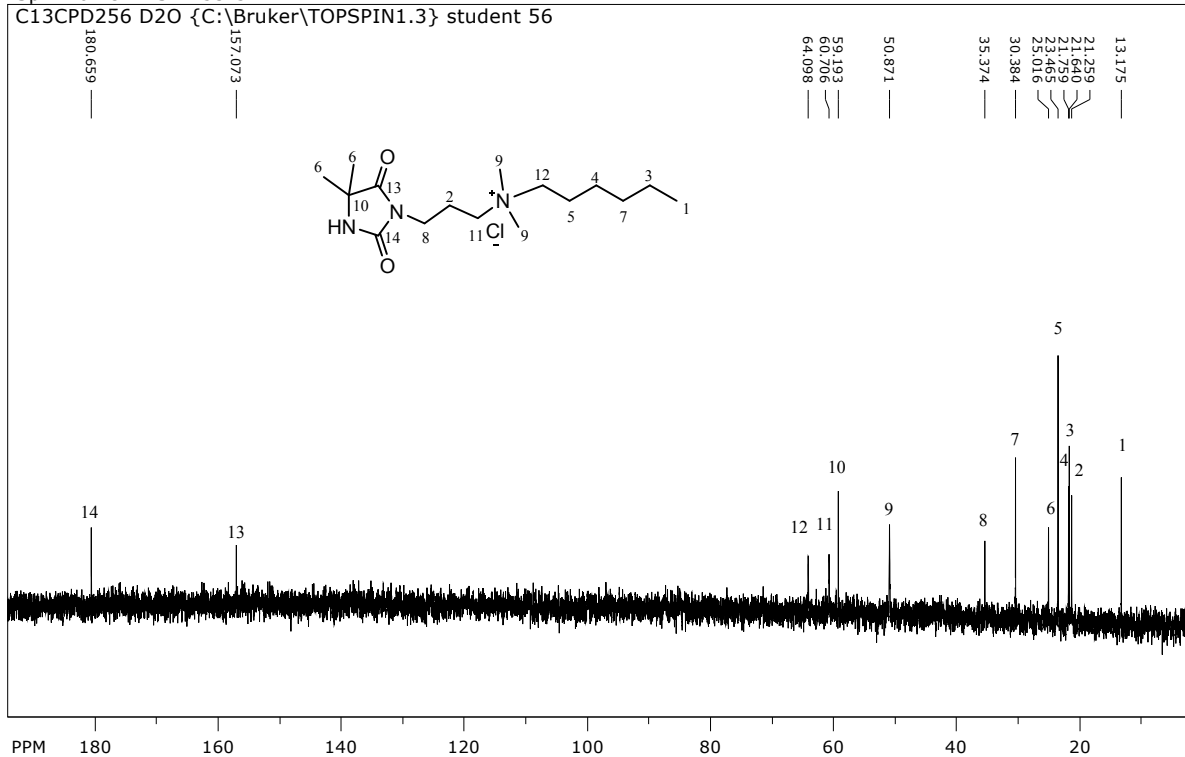
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number of scans: 16

freq. of 0 ppm: 300.129976 MHz
processed size: 32768 complex points
LB: 0.300 GF: 0.0000

Figure S13. ¹H-NMR spectrum of **compound#7** (D₂O).

SpinWorks 4: SL-135-3

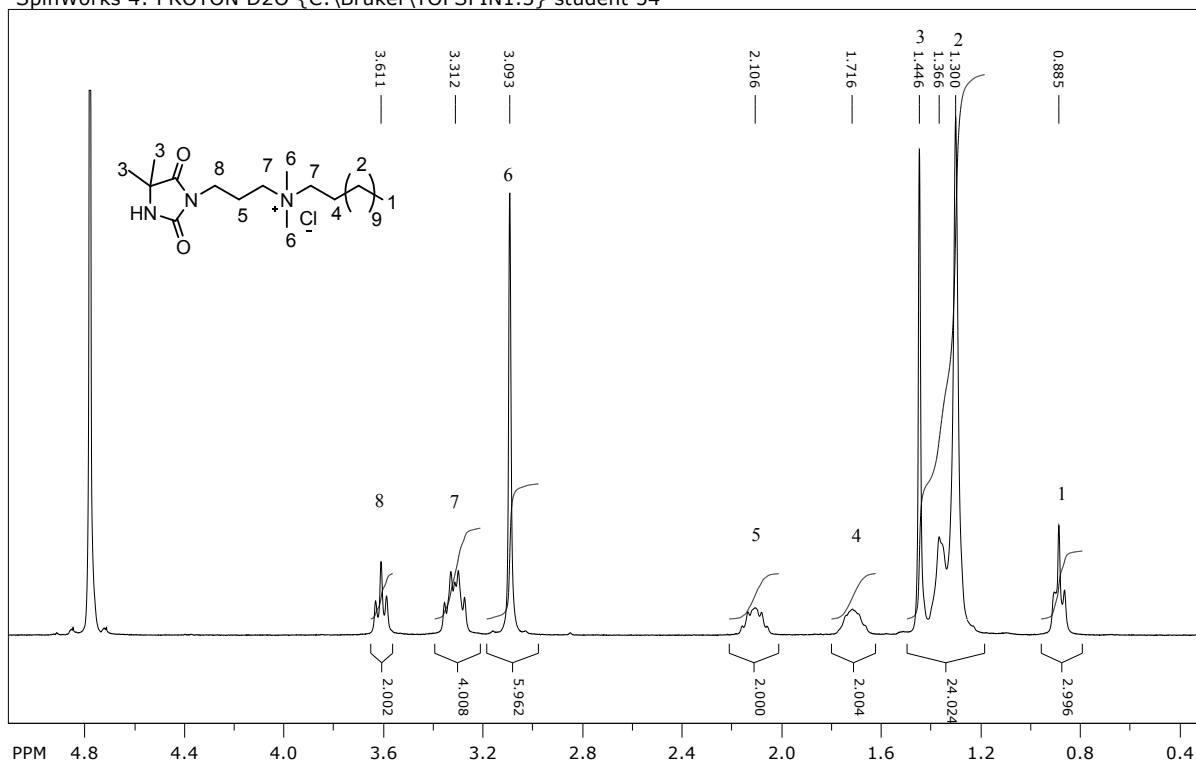
C13CPD256 D2O {C:\Bruker\TOPSPIN1.3} student 56



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number of scans: 256

freq. of 0 ppm: 75.467749 MHz
processed size: 32768 complex points
LB: 1.000 GF: 0.0000

Figure S14. ^{13}C NMR spectrum of compound#7 (D_2O).

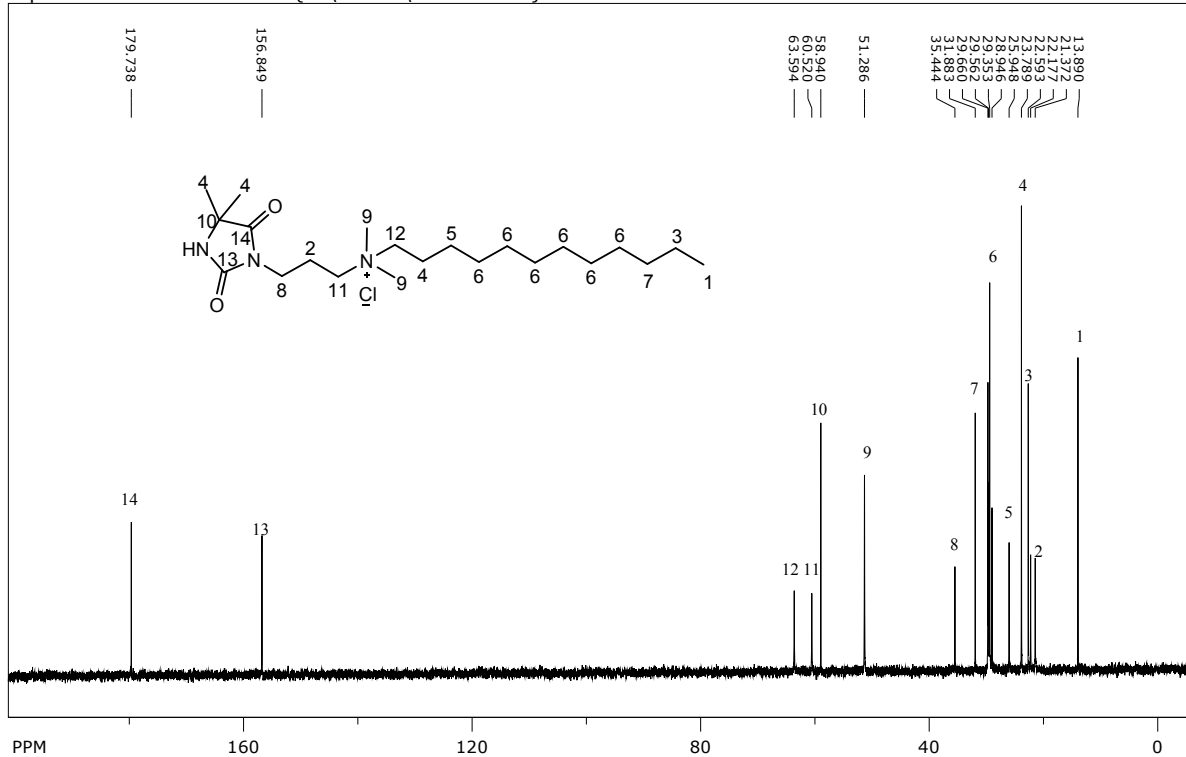


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number of scans: 16

freq. of 0 ppm: 300.129976 MHz
processed size: 32768 complex points
LB: 0.300 GF: 0.0000

Figure S15. ^1H -NMR spectrum of **compound#8** (D_2O).

SpinWorks 4: C13CPD D2O {C:\Bruker\TOPSPIN1.3} Liu 48



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number of scans: 1024

freq. of 0 ppm: 75.467749 MHz
processed size: 32768 complex points
LB: 1.000 GF: 0.0000

Figure S16. ^{13}C NMR spectrum of **compound#8** (D_2O).

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 47

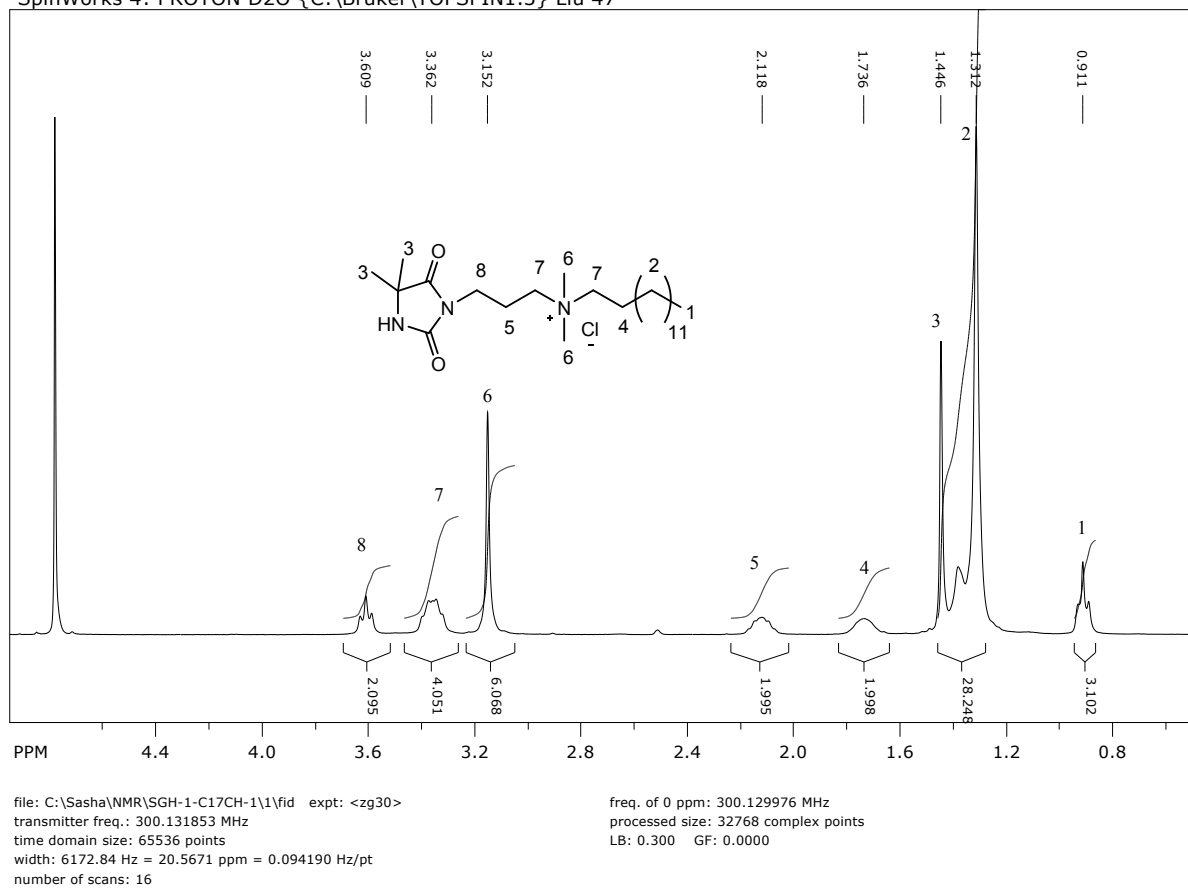


Figure S17. ^1H -NMR spectrum of **compound#9** (D_2O).

SpinWorks 4: C13CPD D2O {C:\Bruker\TOPSPIN1.3} Liu 49

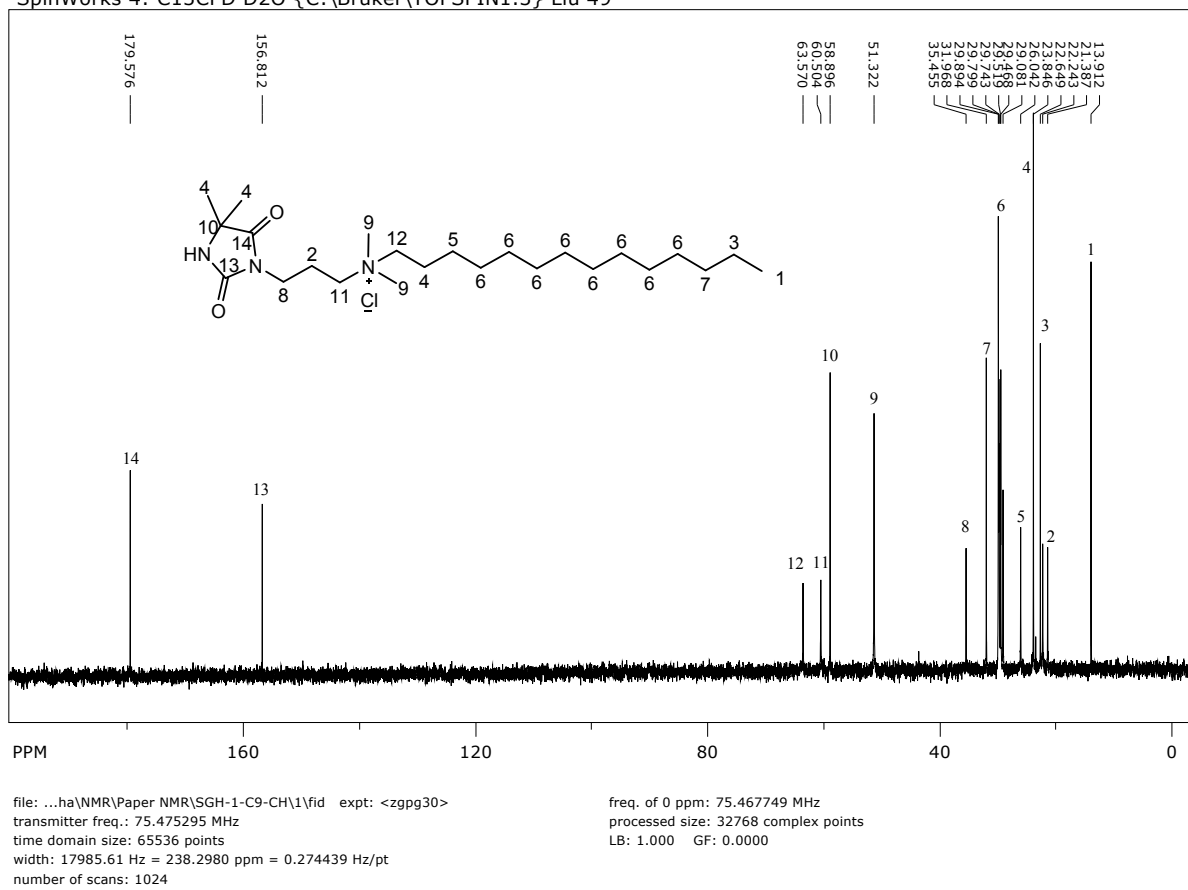
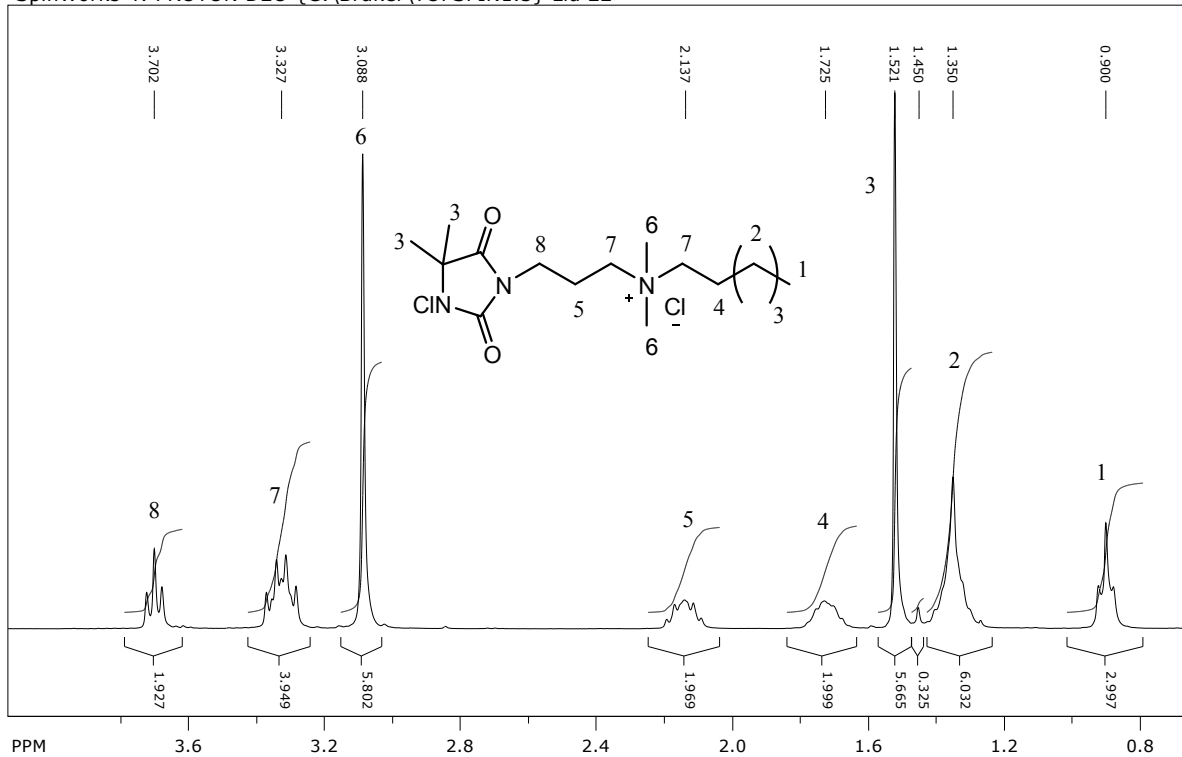


Figure S18. ^{13}C NMR spectrum of **compound#9** (D_2O).

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 22

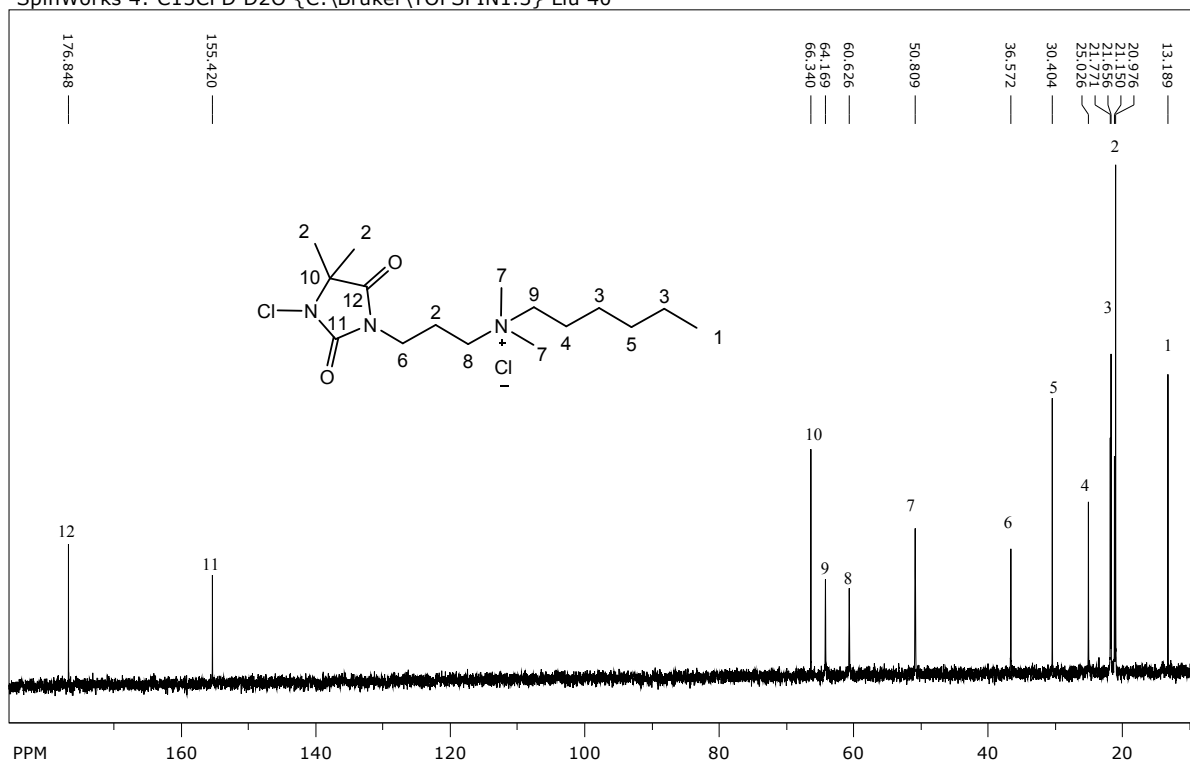


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number of scans: 16

freq. of 0 ppm: 300.129976 MHz
processed size: 32768 complex points
LB: 0.300 GF: 0.0000

Figure S19. ¹H-NMR spectrum of **compound#10** (D₂O).

SpinWorks 4: C13CPD D2O {C:\Bruker\TOPSPIN1.3} Liu 40

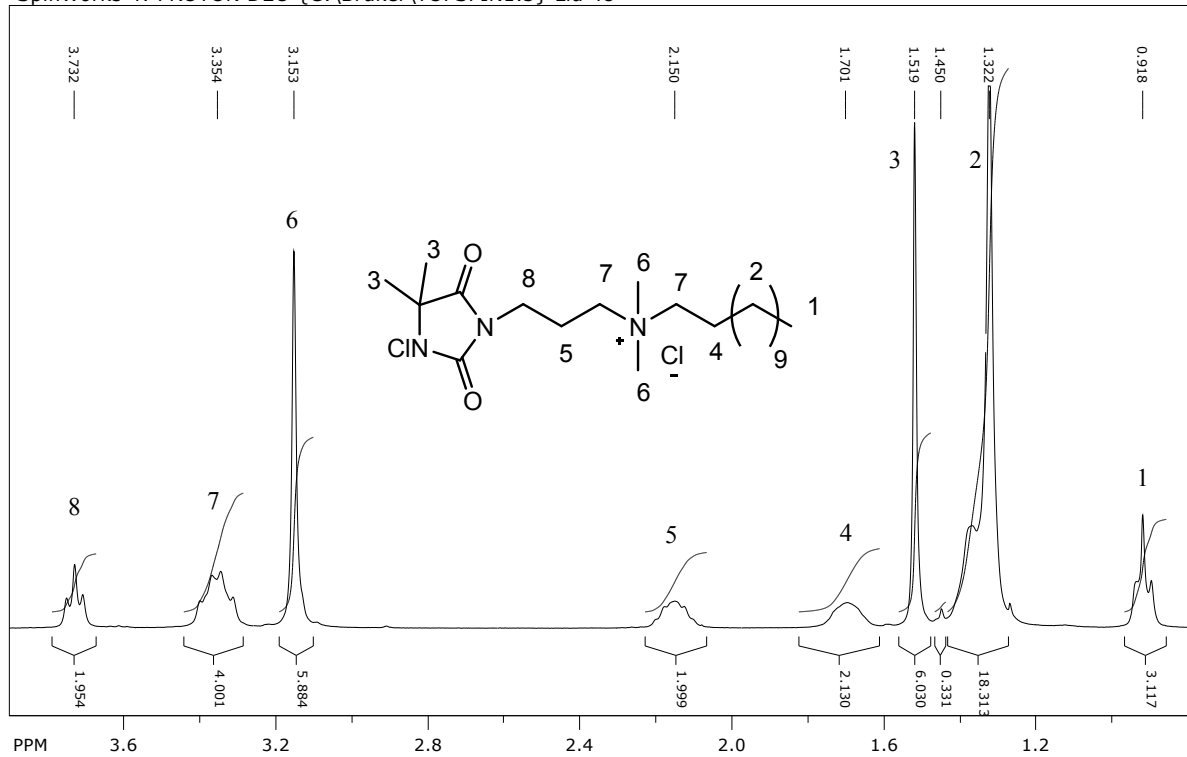


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width: 17985.61 Hz = 238.2980 ppm = 0.274439 Hz/pt
number of scans: 1024

freq. of 0 ppm: 75.467749 MHz
processed size: 32768 complex points
LB: 1.000 GF: 0.0000

Figure S20. ¹³C NMR spectrum of **compound#10** (D₂O).

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 48



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number of scans: 16

freq. of 0 ppm: 300.129976 MHz
processed size: 32768 complex points
LB: 0.300 GF: 0.0000

Figure S21. ¹H-NMR spectrum of **compound#11**(D₂O).

SpinWorks 4: C13CPD D2O {C:\Bruker\TOPSPIN1.3} Liu 47

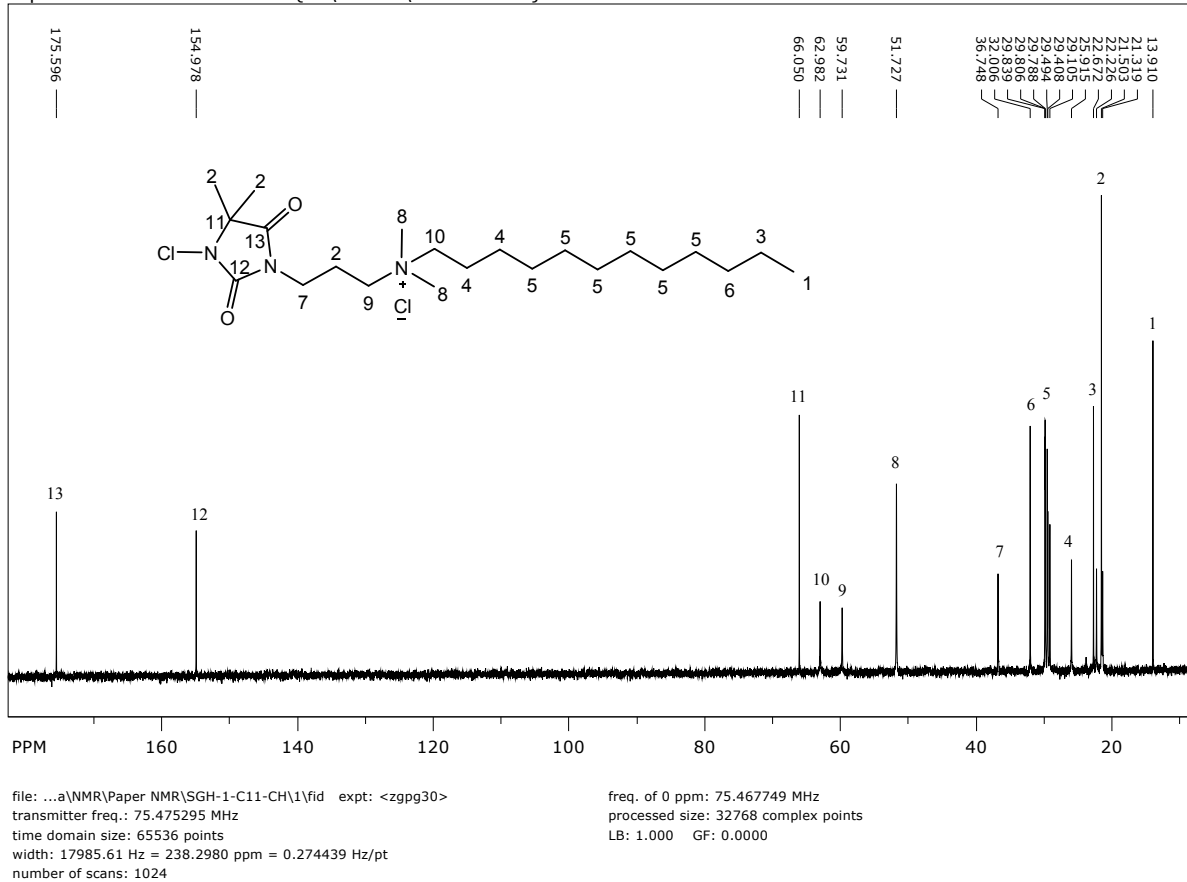
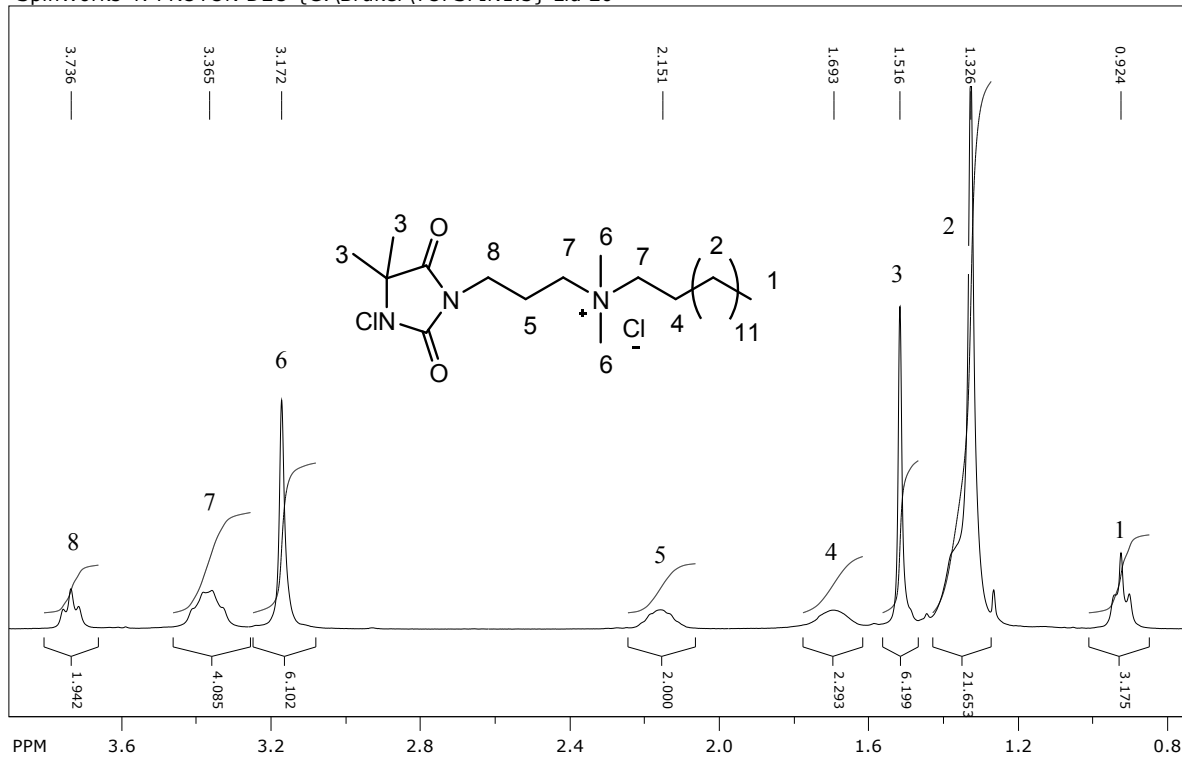


Figure S22. ^{13}C NMR spectrum of **compound#11**(D_2O).

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 20

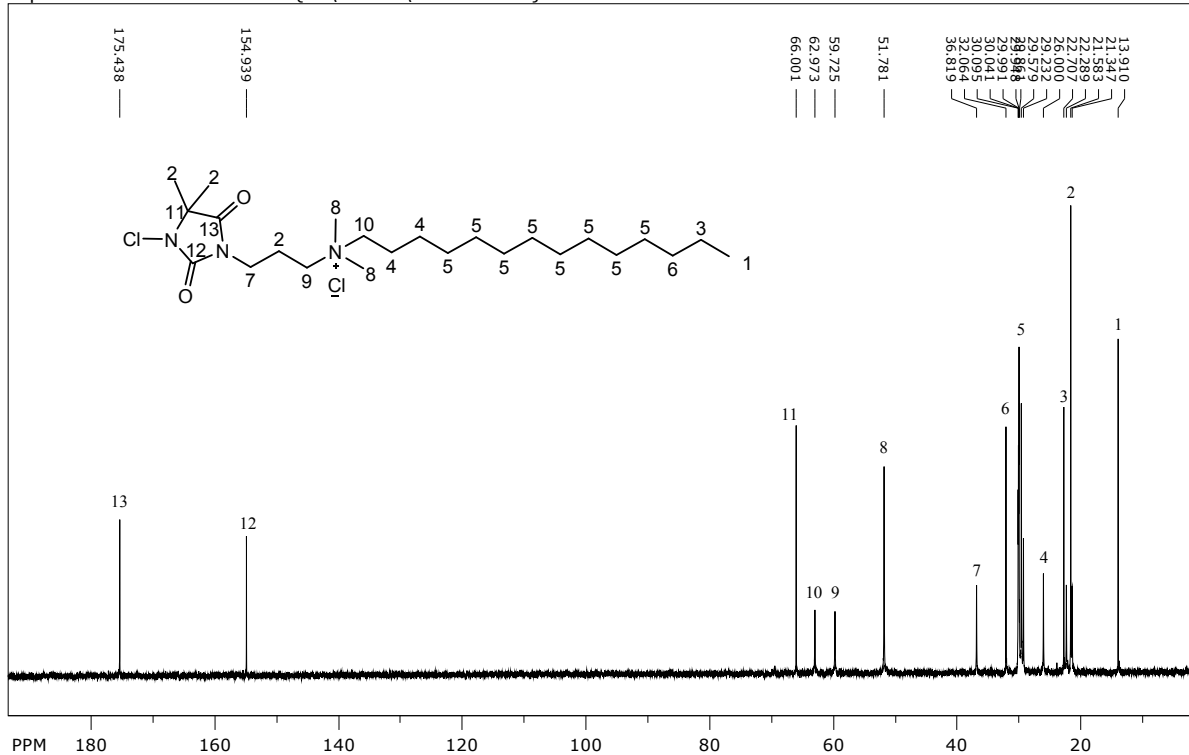


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number of scans: 16

freq. of 0 ppm: 300.129976 MHz
processed size: 32768 complex points
LB: 0.300 GF: 0.0000

Figure S23. ^1H -NMR spectrum of **compound#12** (D_2O).

SpinWorks 4: C13CPD D2O {C:\Bruker\TOPSPIN1.3} Liu 52



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number of scans: 1024

freq. of 0 ppm: 75.467749 MHz
processed size: 32768 complex points
LB: 1.000 GF: 0.0000

Figure S24. ^{13}C NMR spectrum of **compound#12** (D_2O).

2. HPLC-MS analysis of compounds # 1-3,7-9

Analytic HPLC was run on a Varian 212 HPLC instrument, equipped with X-Bridge™ BEH C18 2.5µm and Inertsil C8 3.3µm columns and interfaced with Varian 500 MS-Ion trap detector. Eluent system was gradient: acetonitrile: water (5:95, v/v, 0.1% of formic acid), a linear gradient was applied for 20 min, up to an acetonitrile: water ratio of 90:10 (v/v), after which elution with a acetonitrile: water (5:95, v/v) gradient was used. Flow: 0.4 ml min⁻¹. Compound #7-8 were analyzed using X-Bridge™ BEH C18 2.5µm and the rest were ran through Inertsil C8 3.3µm column.

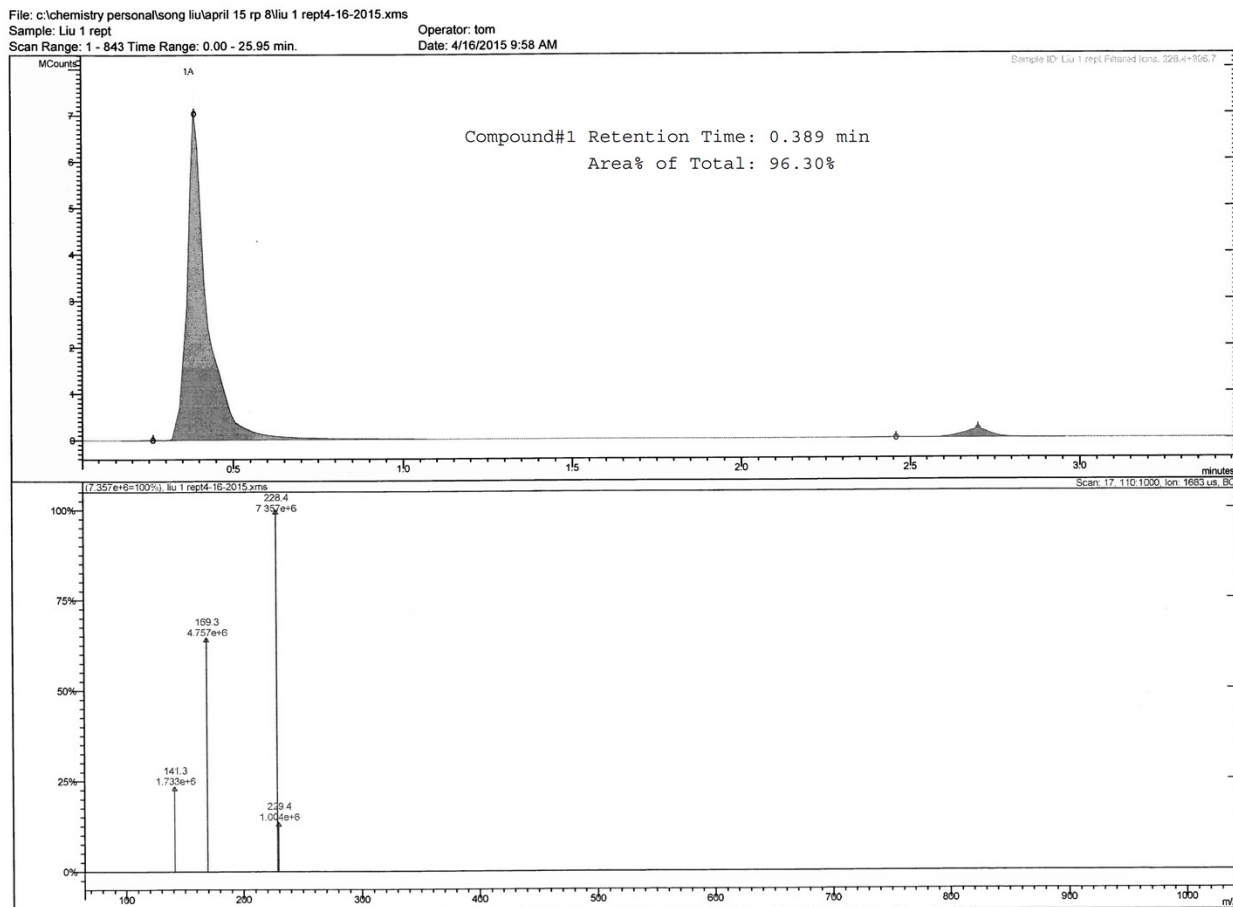


Figure S25. LCMS analysis of compound#1, purity: 96.30%

File: c:\chemistry\personallsong liu\april 15 rp 8\lul 2 .01 to 2 rept4-15-2015.xms
Sample: Lul 2 .01 to 2 rept Operator: tom
Scan Range: 1 - 925 Time Range: 0.00 - 25.95 min. Date: 4/15/2015 10:51 AM

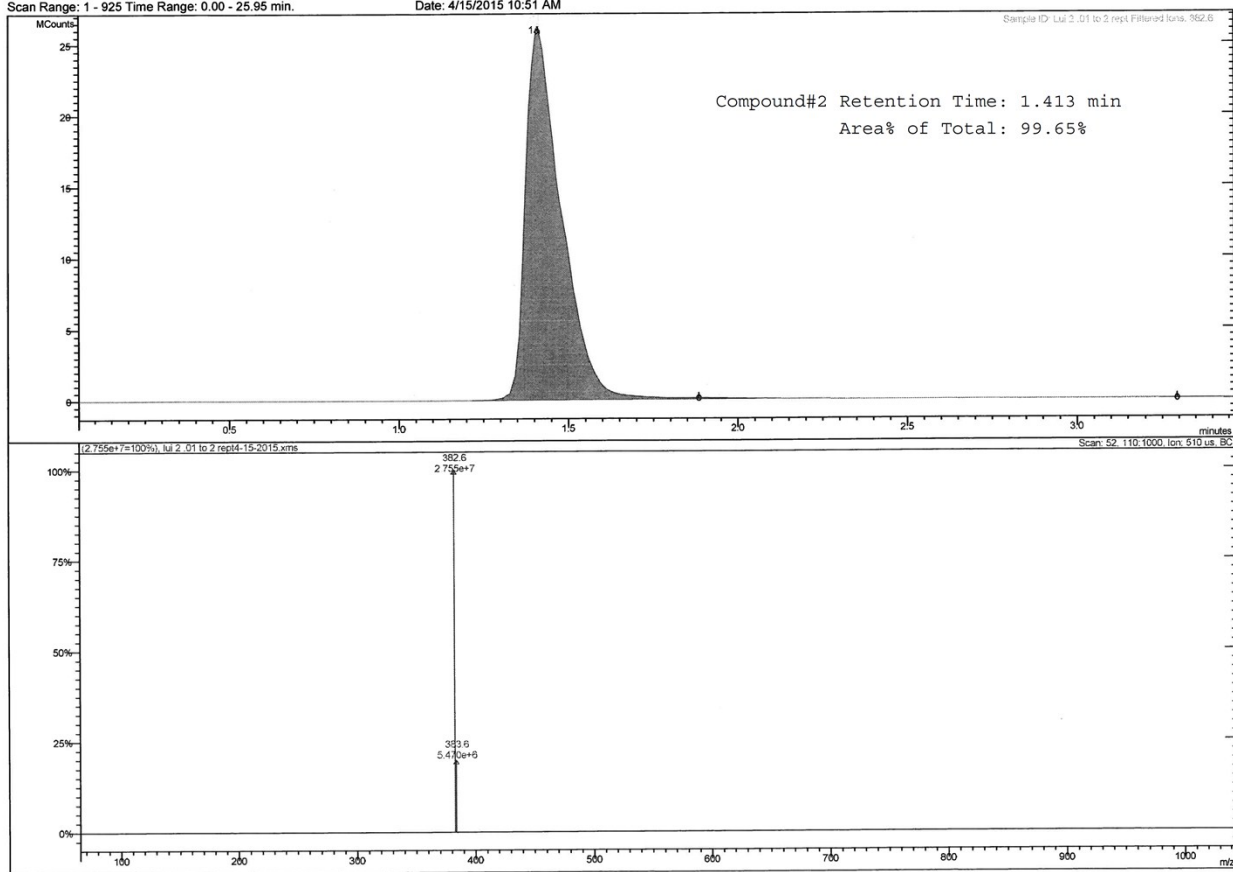


Figure S26. LCMS analysis of compound#2, purity: 99.65%.

File: c:\chemistry\personalsong liu\april 15 rp 8\liu 3 .01 to 24-15-2015001.xms
Sample: Liu 3 .01 to 2 Operator: tom
Scan Range: 1 - 836 Time Range: 0.00 - 25.96 min. Date: 4/15/2015 11:39 AM

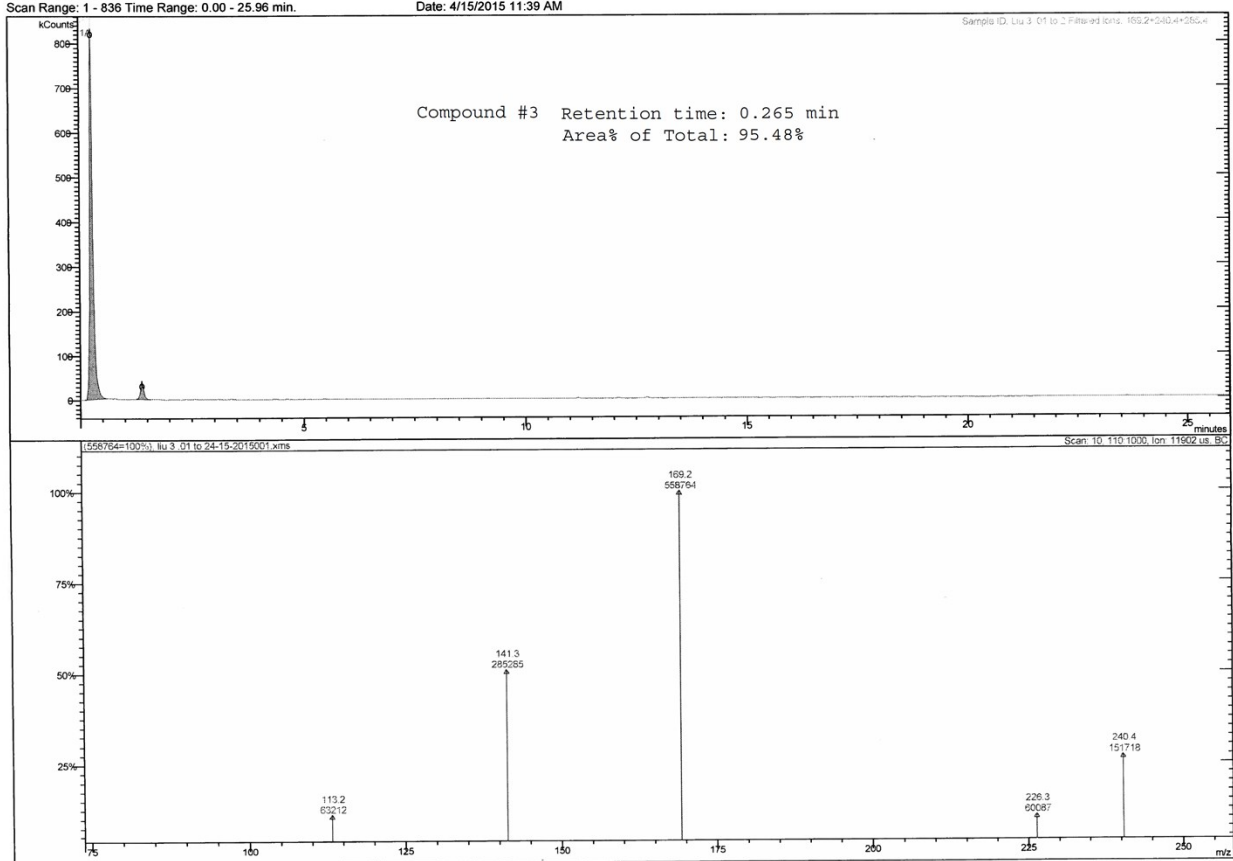


Figure S27. LCMS analysis of compound#3, purity: 95.48%.

File: c:\chemistry\personal\song liu\april 15 rp 8\lul 7 .01 to 24-15-2015.xms
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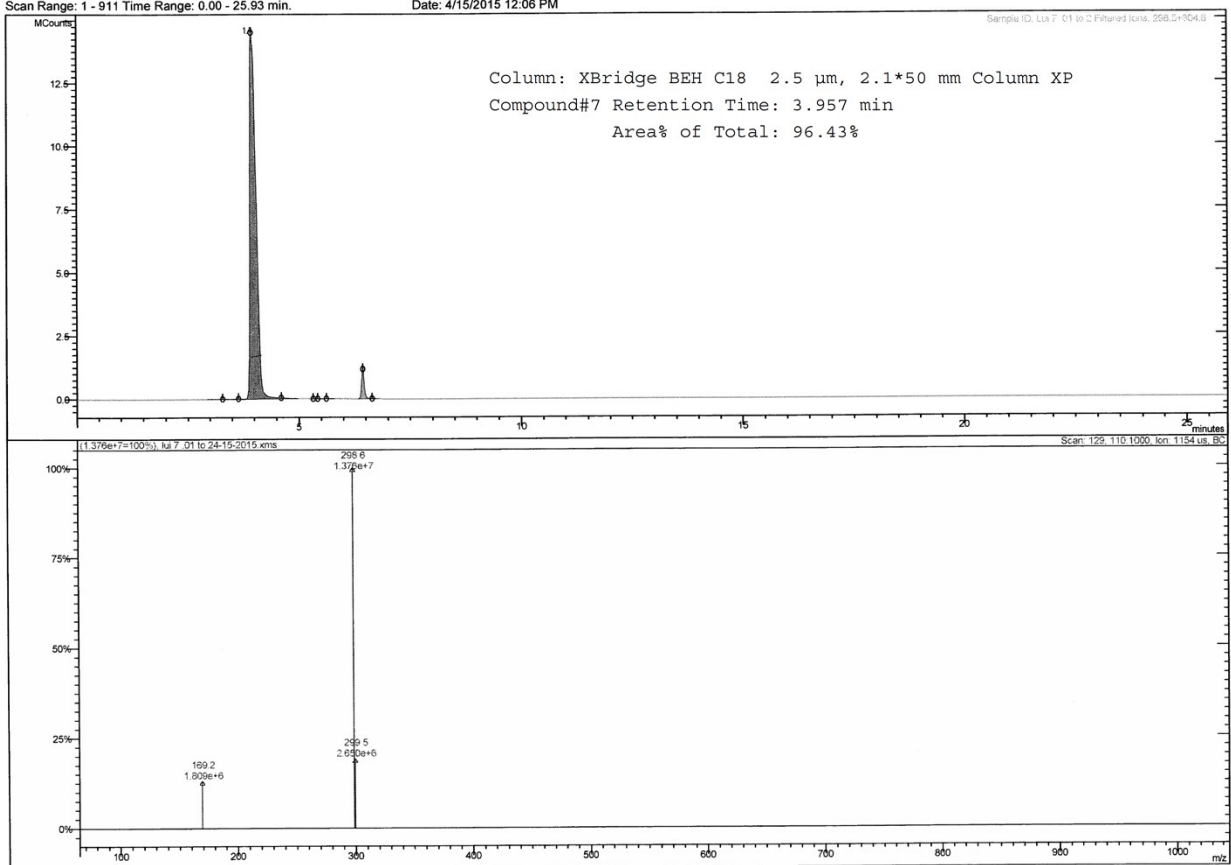


Figure S28. LCMS analysis of compound#7, purity: 96.43%.

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Date: 4/9/2015 11:56 AM

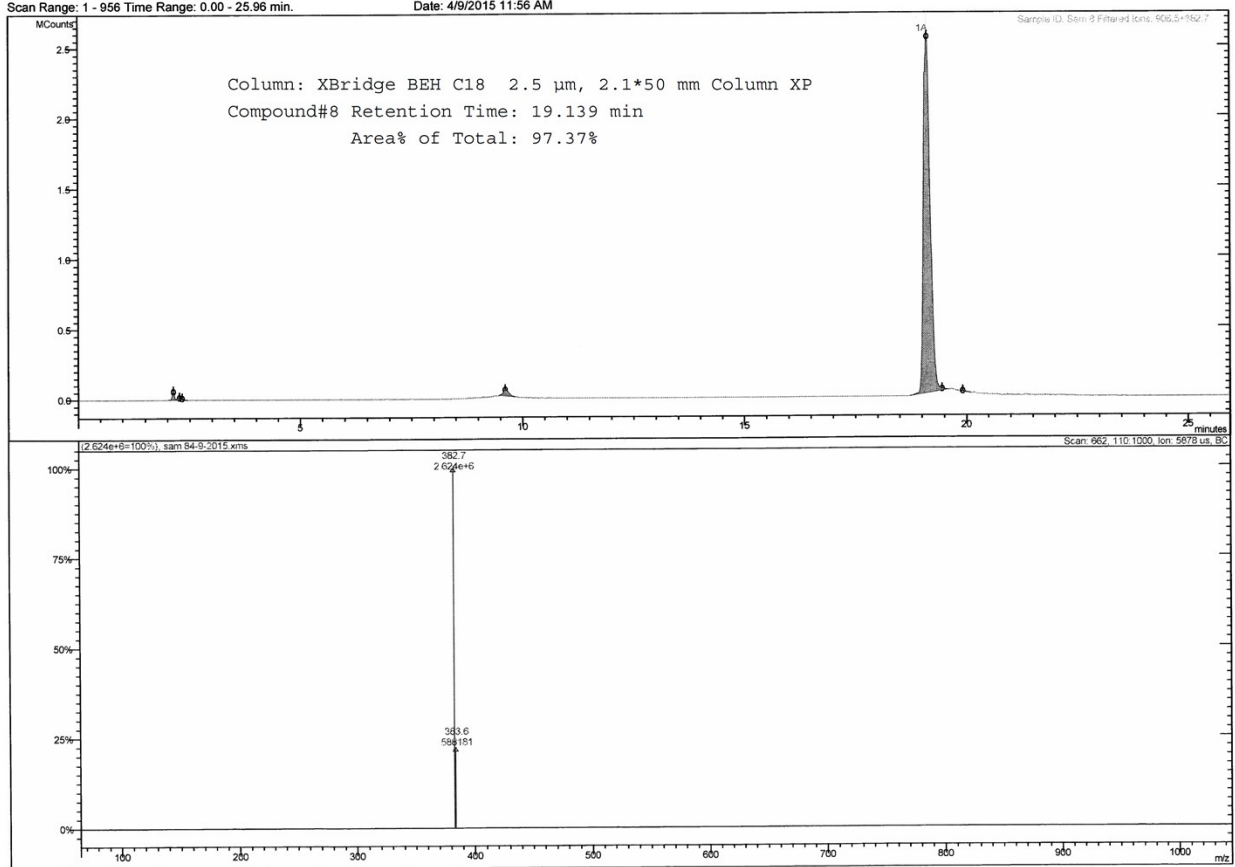


Figure S29. LCMS analysis of compound#8, purity: 97.37%.

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Operator: tom
Date: 4/9/2015 4:41 PM

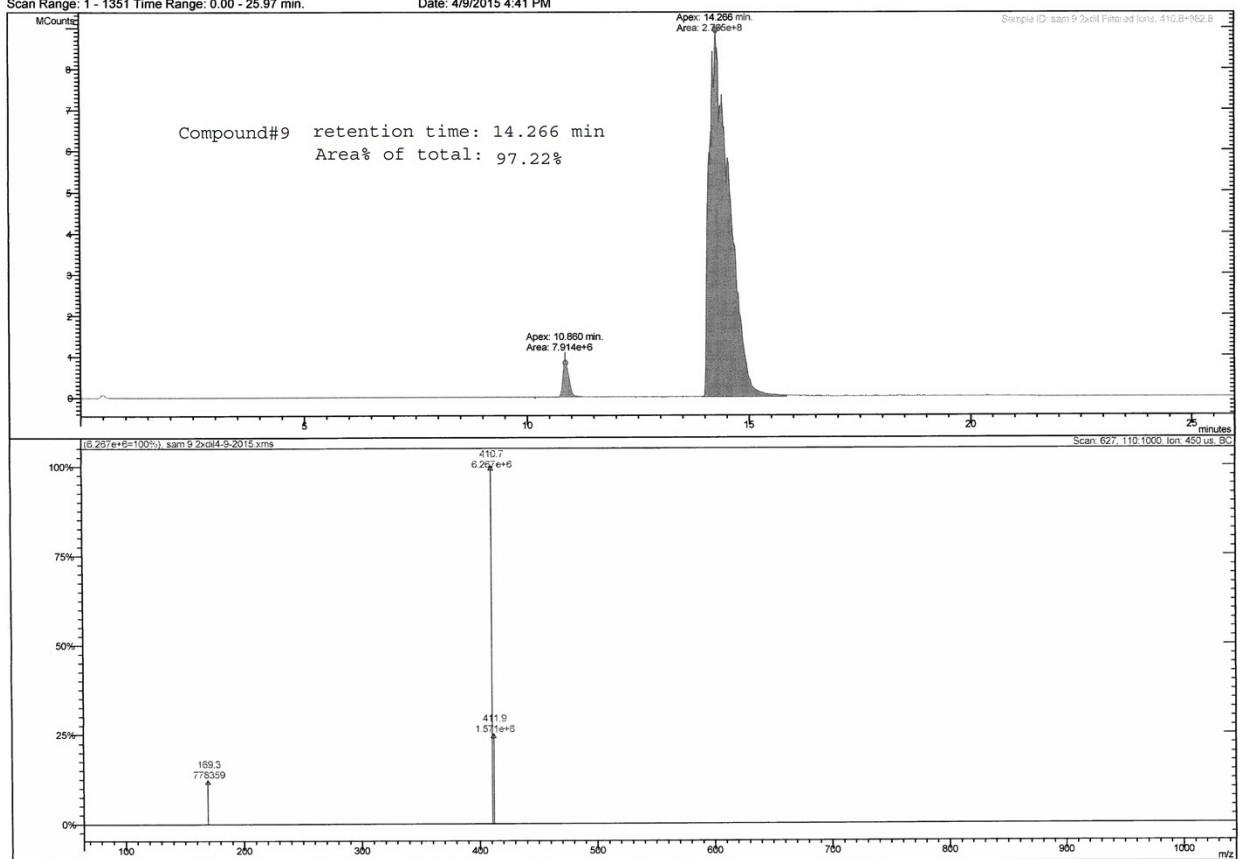


Figure S30. LCMS analysis of compound#9, purity: 97.22%.

3. Absolute quantitative ^1H NMR (qHNMR) of compounds# 4-6,10-12

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 12

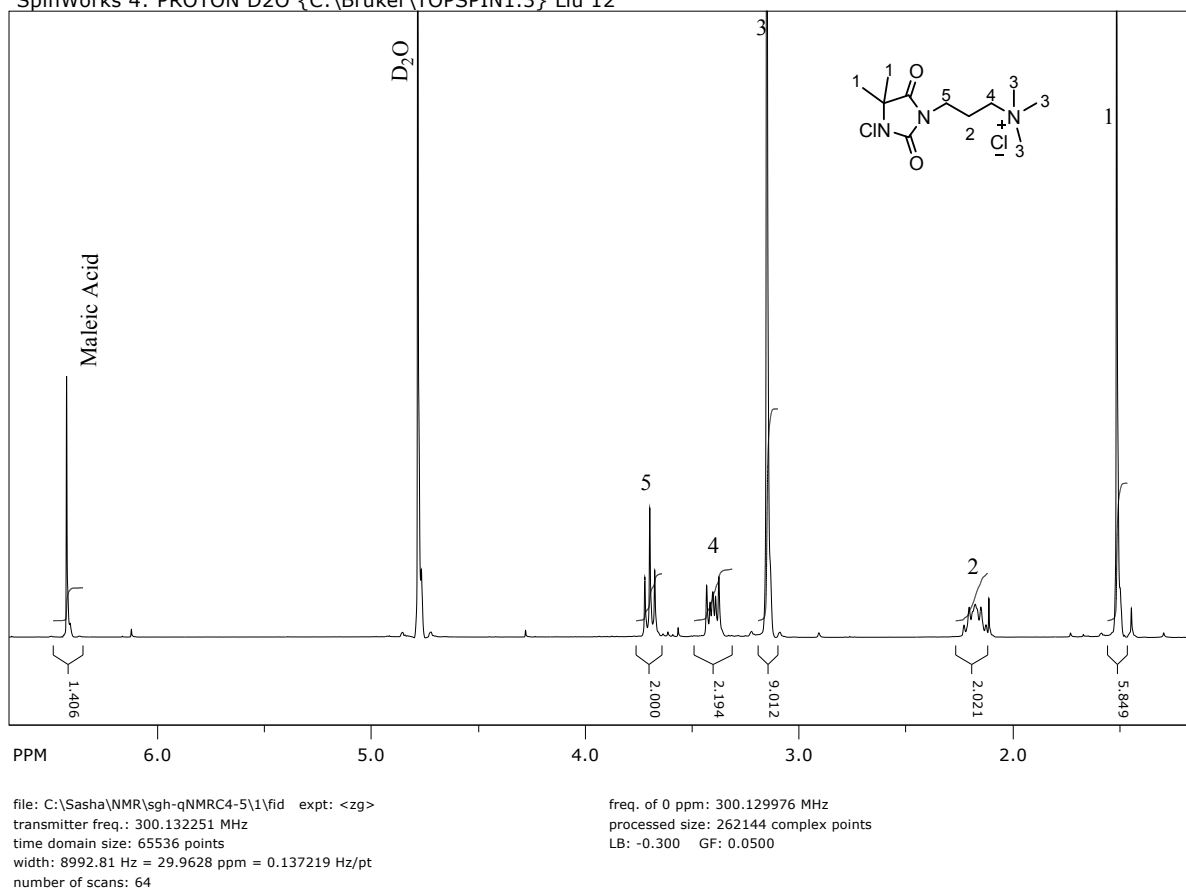


Figure S31. Quantitative ^1H NMR (qHNMR) spectrum of compound#4, purity: 99.30%.

$$P[\%] = \frac{n_{IC} \cdot Int_t \cdot MW_t \cdot m_{IC}}{n_t \cdot Int_{IC} \cdot MW_{IC} \cdot m_s} \cdot P_{IC}$$

$m_s = 4.995$ mg, $m_{IC} = 1.512$ mg, $P_{IC} = 99.94$,

$Int_t = 17.586$, $n_t = 19$, $Int_{IC} = 1.449$, $n_{IC} = 2$,

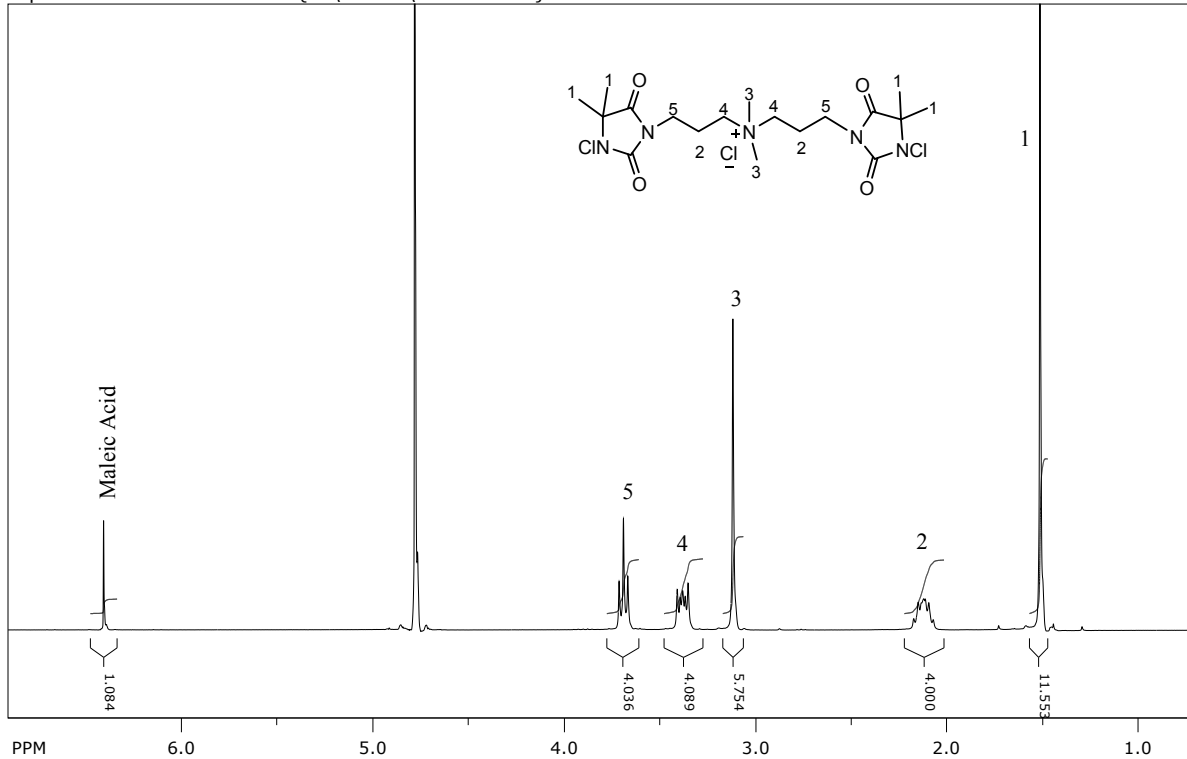
$MW_t = 298.21$ g/mol, $MW_{IC} = 116.07$ g/mol

Table S1. Absolute quantitative ^1H NMR (qHNMR) analysis of compound#4 (4.995 mg/0.6 mL) in D_2O with the addition of maleic acid (99.94% pure) as internal calibrant.

	Peaks (ppm)	Integration	Number of protons	Weight (mg)	MWt (g/mol)
Compound 4	1.471-1.549	5.388	6	4.995	298.21
	3.108-3.205	8.341	9		
	3.301-3.472	2	2		
	3.640-3.785	1.857	2		
	Sum	17.586	19		
Internal St	6.285-6.493	1.449	2	1.512	116.07

$$P[\%] = \frac{2 * 17.586 * 298.21 * 1.512}{19 * 1.449 * 116.07 * 4.995} * 99.94 = \frac{15858.82}{15961.64} * 99.94 = 99.30 \%$$

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 8



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number of scans: 64

freq. of 0 ppm: 300.129976 MHz
processed size: 262144 complex points
LB: -0.300 GF: 0.0500

Figure S32. Quantitative ^1H NMR (qHNMR) spectrum of compound#5, purity: 96.55%.

$$P[\%] = \frac{n_{IC} \cdot Int_t \cdot MW_t \cdot m_{IC}}{n_t \cdot Int_{IC} \cdot MW_{IC} \cdot m_s} \cdot P_{IC}$$

$m_s = 4.66$ mg, $m_{IC} = 0.593$ mg, $P_{IC} = 99.94$,

$Int_t = 29.433$, $n_t = 30$, $Int_{IC} = 1.084$, $n_{IC} = 2$,

$MW_t = 486.82$ g/mol, $MW_{IC} = 116.07$ g/mol

Table S2. Absolute quantitative ^1H NMR (qHNMR) analysis of compound#5 (4.66 mg/0.6 mL) in D_2O with the addition of maleic acid (99.94% pure) as internal calibrant.

	Peaks (ppm)	Integration	Number of protons	Weight (mg)	MWt (g/mol)
Compound 5	1.472-1.567	11.553	12	4.66	486.82
	2.014-2.221	4	4		
	3.063-3.171	5.755	6		
	3.275 -3.478	4.089	4		
	3.610-3.777	4.036	4		
	Sum	29.433	30		
Internal St	6.338-6.478	1.084	2	0.593	116.07

$$P[\%] = \frac{2 * 29.433 * 486.82 * 0.593}{30 * 1.084 * 116.07 * 4.66} * 99.94 = \frac{16993.69}{17589.62} * 99.94 = 96.55 \%$$

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 9

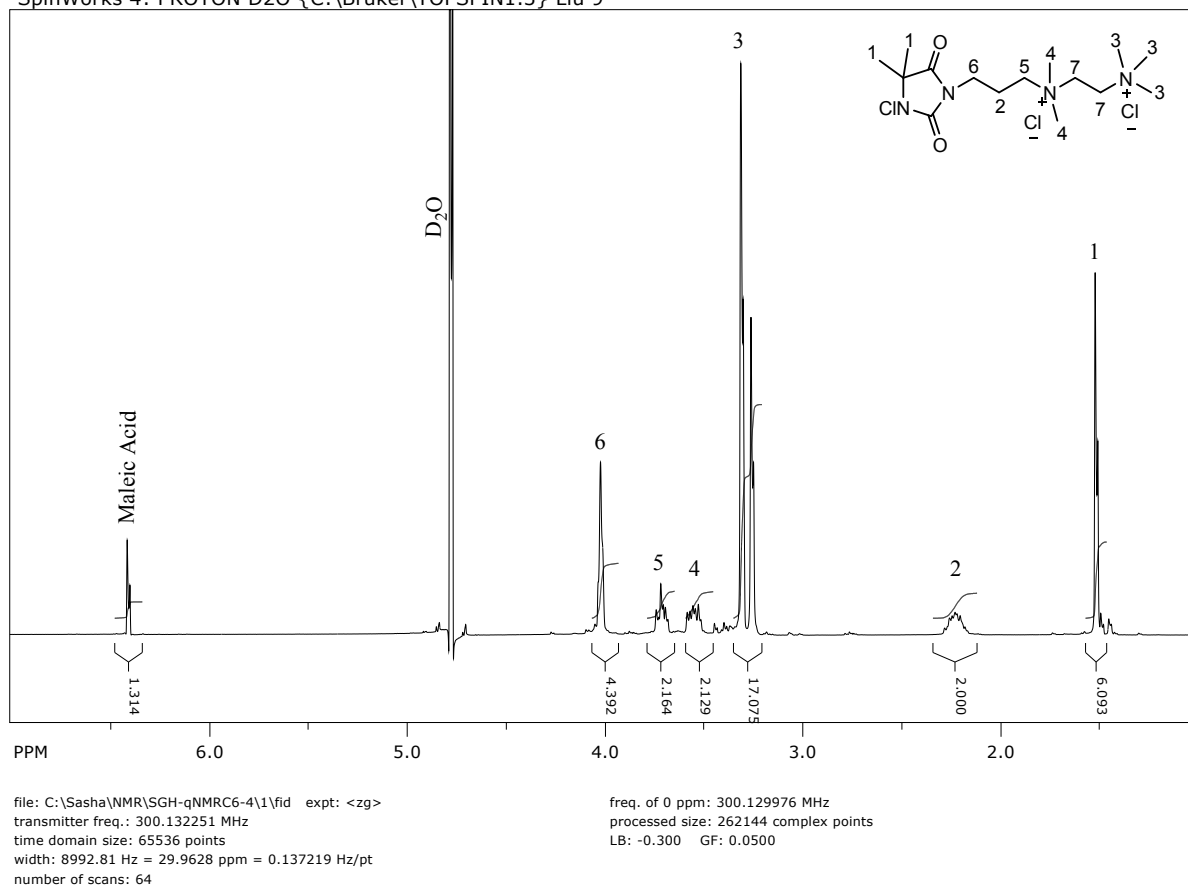


Figure S33. Quantitative ^1H NMR (qHNMR) spectrum of compound#6, purity: 97.02%.

$$P[\%] = \frac{n_{IC} \cdot Int_t \cdot MW_t \cdot m_{IC}}{n_t \cdot Int_{IC} \cdot MW_{IC} \cdot m_s} \cdot P_{IC}$$

$m_s = 6.123$ mg, $m_{IC} = 1.023$ mg, $P_{IC} = 99.94$,

$Int_t = 33.851$, $n_t = 31$, $Int_{IC} = 1.314$, $n_{IC} = 2$,

$MW_t = 405.79$ g/mol, $MW_{IC} = 116.07$ g/mol

Table S3. Absolute quantitative ^1H NMR (qHNMR) analysis of compound#6 (6.123 mg/0.6 mL) in D_2O with the addition of maleic acid (99.94% pure) as internal calibrant.

	Peaks (ppm)	Integration	Number of protons	Weight (mg)	MWt (g/mol)
Compound 6	1.462-1.569	6.09	6	6.123	405.79
	2.118-2.342	2	2		
	3.207-3.351	17.076	15		
	3.453-3.593	2.129	2		
	3.649-3.788	2.164	2		
	3.933-4.067	4.392	4		
	Sum	33.851	31		
Internal St	6.342-6.482	1.314	2	1.023	116.07

$$P[\%] = \frac{2 * 33.851 * 405.79 * 1.023}{31 * 1.314 * 116.07 * 6.123} * 99.94 = \frac{28104.67}{28949.51} * 99.94 = 97.02 \%$$

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 30

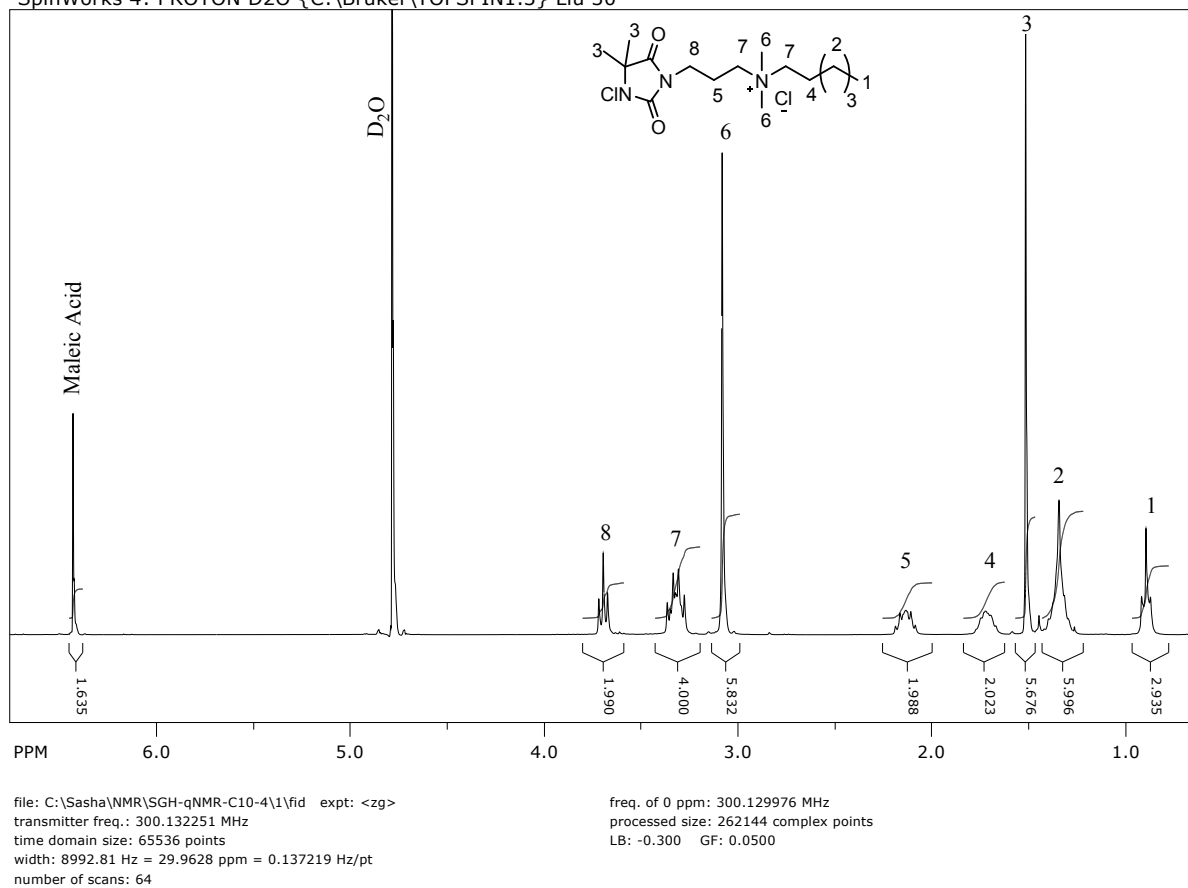


Figure S34. Quantitative ¹H NMR (qHNMR) spectrum of compound#10, purity: 95.54%.

$$P[\%] = \frac{n_{IC} \cdot Int_t \cdot MW_t \cdot m_{IC}}{n_t \cdot Int_{IC} \cdot MW_{IC} \cdot m_s} \cdot P_{IC}$$

$m_s = 7.794$ mg, $m_{IC} = 1.944$ mg, $P_{IC} = 99.94$,

$Int_t = 30.441$, $n_t = 31$, $Int_{IC} = 1.635$, $n_{IC} = 2$,

$MW_t = 370.36$ g/mol, $MW_{IC} = 116.07$ g/mol

Table S4. Absolute quantitative ^1H NMR (qHNMR) analysis of compound#**10** (7.794 mg/0.6 mL) in D_2O with the addition of maleic acid (99.94% pure) as internal calibrant.

	Peaks (ppm)	Integration	Number of protons	Weight (mg)	MWt (g/mol)
Compound 10	0.776-0.965	2.935	3	7.794	370.36
	1.217-1.430	5.997	6		
	1.465-1.568	5.676	6		
	1.623-1.836	2.023	2		
	1.997-2.253	1.988	2		
	2.990-3.136	5.832	6		
	3.195-3.427	4	4		
	3.589-3.801	1.99	2		
	Sum	30.441	31		
Internal St	6.382-6.453	1.635	2	1.944	116.07

$$P[\%] = \frac{2 * 30.441 * 370.36 * 1.944}{31 * 1.635 * 116.07 * 7.794} * 99.94 = \frac{43833.81}{45852.16} * 99.94 = 95.54 \%$$

SpinWorks 4: PROTON D2O {C:\Bruker\TOPSPIN1.3} Liu 31

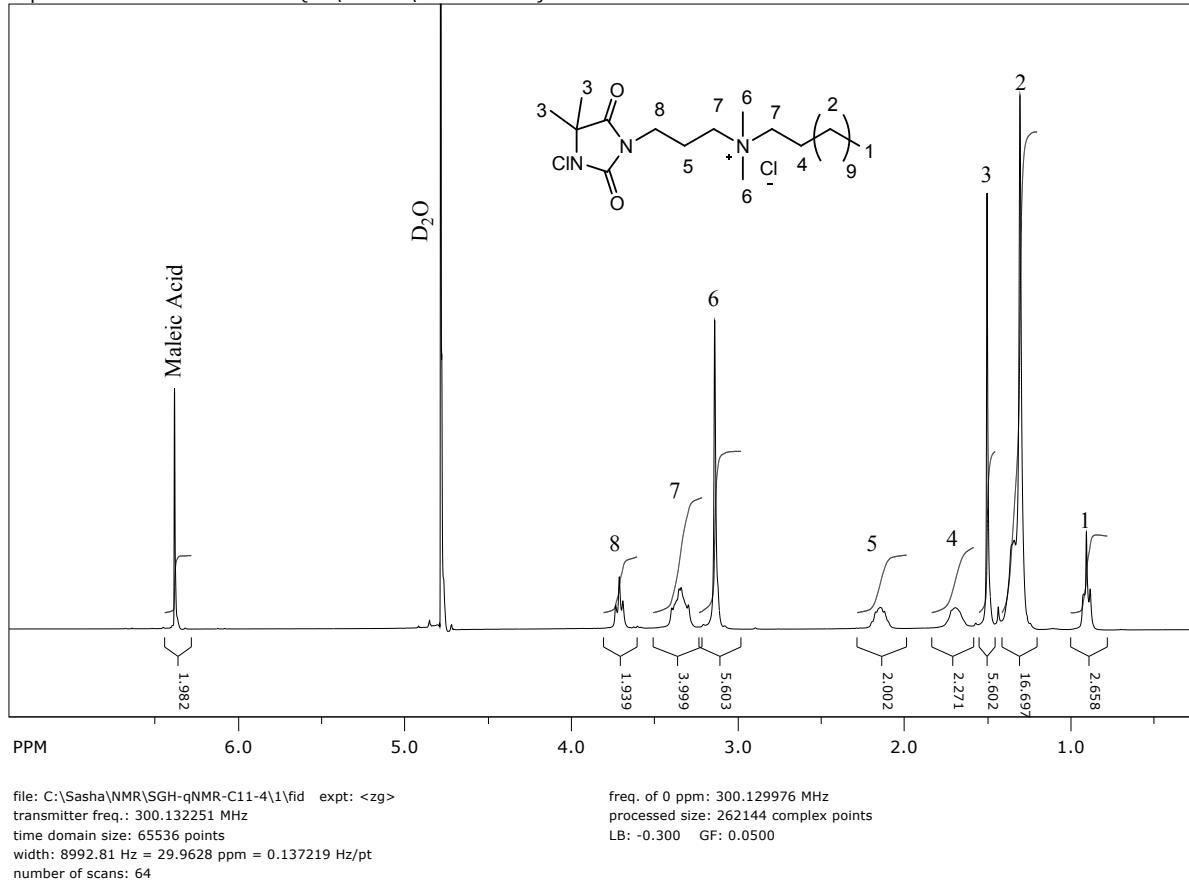


Figure S35. Quantitative ^1H NMR (qHNMR) spectrum of compound#11, purity: 95.76%.

$$P[\%] = \frac{n_{IC} \cdot Int_t \cdot MW_t \cdot m_{IC}}{n_t \cdot Int_{IC} \cdot MW_{IC} \cdot m_s} \cdot P_{IC}$$

$m_s = 7.498$ mg, $m_{IC} = 1.926$ mg, $P_{IC} = 99.94$,

$Int_t = 40.773$, $n_t = 43$, $Int_{IC} = 1.982$, $n_{IC} = 2$,

$MW_t = 452.5$ g/mol, $MW_{IC} = 116.07$ g/mol

Table S5. Absolute quantitative ^1H NMR (qHNMR) analysis of compound#11 (7.498 mg/0.6 mL) in D_2O with the addition of maleic acid (99.94% pure) as internal calibrant.

	Peaks (ppm)	Integration	Number of protons	Weight (mg)	MWt (g/mol)
Compound 11	0.779-0.999	2.658	3	7.498	452.5
	1.201-1.412	16.697	18		
	1.453-1.55	5.602	6		
	1.582-1.834	2.272	2		
	1.986-2.284	2.002	2		
	2.981-3.233	5.603	6		
	3.215-3.509	4	4		
	3.605-3.807	1.939	2		
	Sum	40.773	43		
Internal St	6.284-6.445	1.982	2	1.926	116.07

$$P[\%] = \frac{2 * 40.773 * 452.5 * 1.926}{43 * 1.982 * 116.07 * 7.498} * 99.94 = \frac{71068.56}{74171.58} * 99.94 = 95.76 \%$$

$$P[\%] = \frac{n_{IC} \cdot Int_t \cdot MW_t \cdot m_{IC}}{n_t \cdot Int_{IC} \cdot MW_{IC} \cdot m_s} \cdot P_{IC}$$

$m_s = 8.887$ mg, $m_{IC} = 2.803$ mg, $P_{IC} = 99.94$,

$Int_t = 46.572$, $n_t = 47$, $Int_{IC} = 2.704$, $n_{IC} = 2$,

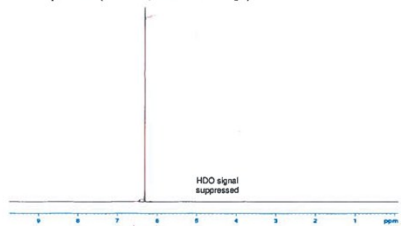
$MW_t = 480.55$ g/mol, $MW_{IC} = 116.07$ g/mol

Table S6. Absolute quantitative ^1H NMR (qHNMR) analysis of compound#12 (8.887 mg/0.6 mL) in D_2O with the addition of maleic acid (99.94% pure) as internal calibrant.

	Peaks (ppm)	Integration	Number of protons	Weigh (mg)	MWt (g/mol)
Compound 12	0.801-1.01	2.8	3	8.887	480.55
	1.17-1.407	21.65	22		
	1.44-1.549	6.051	6		
	1.586-1.818	2.393	2		
	2.035-2.286	2.059	2		
	3.052-3.236	5.696	6		
	3.255-3.515	4	4		
	3.629-3.856	1.923	2		
	Sum	46.572	47		
Internal St	6.319-6.437	2.704	2	2.803	116.07

$$P[\%] = \frac{2 * 46.572 * 480.55 * 2.803}{47 * 2.704 * 116.07 * 8.887} * 99.94 = \frac{125463.26}{131093.06} * 99.94 = 95.65 \%$$

1H-NMR Spectrum (600 MHz, Maleic acid in D₂O)



REFERENCES

ISO Guide 31, 2nd Ed. (2000), "Reference materials - Contents of certificates and labels"
 Eurachem/CITAC Guide, 1st Ed. (2003), "Traceability in chemical measurement"
 ISO Guide 35, 3rd Ed. (2005), "Reference materials - General and statistical principles for certification"
 Eurachem/CITAC Guide, 3rd Ed. (2012), "Quantifying uncertainty in analytical measurement"
 ISO/IEC 17025, 2nd Ed. (2005), "General requirements for the competence of testing and calibration laboratories"
 ISO Guide 34, 3rd Ed. (2009), "General requirements for the competence of reference material producers"

TraceCERT[®]
 Traceable Certified Reference Materials

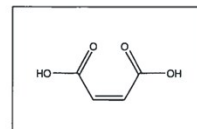


Certificate

Produced in double accredited laboratory fulfilling ISO/IEC 17025 and ISO Guide 34

This certificate is designed in accordance with ISO Guide 31^[1].

Product name: **Maleic acid**
 Product no.: **92816**
 Lot no.: **BCBM8127V**
 Formula: **C₄H₄O₄**
 Molecular mass: **116.07**
 Traceability^[2]: **NIST SRM 841 (KHP) and NIST SRM 350b (Benzoic acid)**
 Certificate issue date: **April 16, 2014**
 Expiry: **MAR 2018**



Certified value and uncertainty according to ISO Guide 35 ^[3] and Eurachem/CITAC Guide ^[4]		
Substance	Certified value as mass fraction (g/g)	Expanded uncertainty, $U = k \cdot u_c$ ($k = 2$) as mass fraction (g/g)
Maleic acid	99.94 %	0.16 %

Minimum sample: The sample is solid at room-temperature. 20 mg is recommended as the minimal sample amount. If less material is used, it is recommended to increase the certified uncertainty by a factor of two for half of sample and a factor of four for a quarter of sample.
 Drying instruction: This material does not require drying before use.
 Intended use: This CRM shall be used as internal standard for quantitative ¹H-NMR measurements.
 Storage and handling: The CRM should be stored in the original bottle at room-temperature (20-25 °C). After use the bottle should be tightly closed and protected from excessive moisture and light. Storage under Argon is recommended.

CRM operations: *A. Rück*
 A. Rück, Ph.D.
 Certification body: *K.-D. Schmidt, Ph.D.*
 K.-D. Schmidt, Ph.D.



Figure S37. Certificate of the internal calibrant maleic acid.

Table S7. Comparison of the antibacterial activity of tetradecyl # 9 and 12 with Benzylododecyldimethylammonium bromide (C12) and Benzyltetradecyldimethylammonium chloride(C14) against 10⁷ CFU/mL of MRSA and 10⁶ CFU/mL MDR *P. aeruginosa*.

Gram-positive MRSA (10⁷ CFU/mL)***								
Synthetic compounds*		Bacterial Reduction at Various Contact Time (min)						
		1	3	5	10	20	30	60
9	%	94.23 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00
	Log ₁₀	1.24 ± 0.00	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09
12	%	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00
	Log ₁₀	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09
C12	%	37.40 ± 52	58.96 ± 25	81.54 ± 16	99.64 ± 0.42	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00
	Log ₁₀	0.300 ± 0.42	0.430 ± 0.28	0.840 ± 0.45	2.69 ± 0.72	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09
C14	%	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00
	Log ₁₀	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09	7.21 ± 0.09
Gram-negative MDR <i>P. aeruginosa</i> (10⁶ CFU/mL)***								
		1	3	5	10	20	30	60
9	%	85.60 ± 18.2	99.98 ± 0.03	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00
	Log ₁₀	1.18 ± 0.88	4.73 ± 1.8	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45
12	%	98.94 ± 1.5	99.99 ± 0.01	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00
	Log ₁₀	3.85 ± 3.1	4.91 ± 1.6	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45
C12	%	97.09 ± 0.00	98.99 ± 1.4	99.44 ± 0.79	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00
	Log ₁₀	1.54 ± 0.00	2.8 ± 1.5	2.25 ± 0.04	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45
C14	%	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00	100.0 ± 0.00
	Log ₁₀	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45	6.36 ± 0.45

*. Concentration of the compounds was set at 0.141 mM