

Electronic Supporting Information

PSS-GN nanocomposites as highly-efficient peroxidase mimic and their application to colorimetric detection of glucose in serum

Jing Chen,^a Jia Ge,^{ab} Lin Zhang,^a Zhaohui Li,^{*ab} Saisai Zhou^a and
Lingbo Qu^{*a}

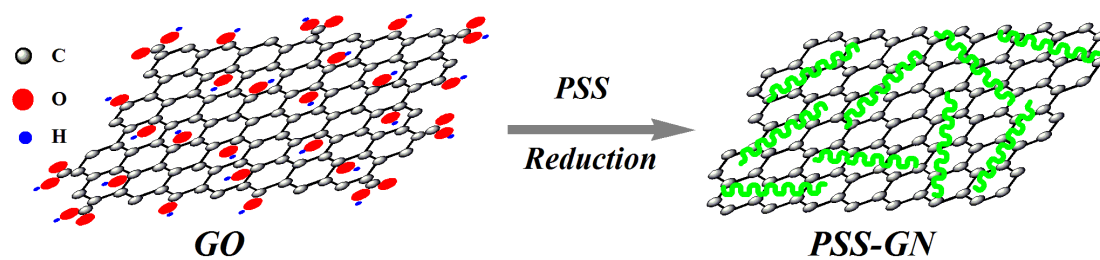
^a College of Chemistry and Molecular Engineering, Zhengzhou University,
Zhengzhou 450001, P. R. China

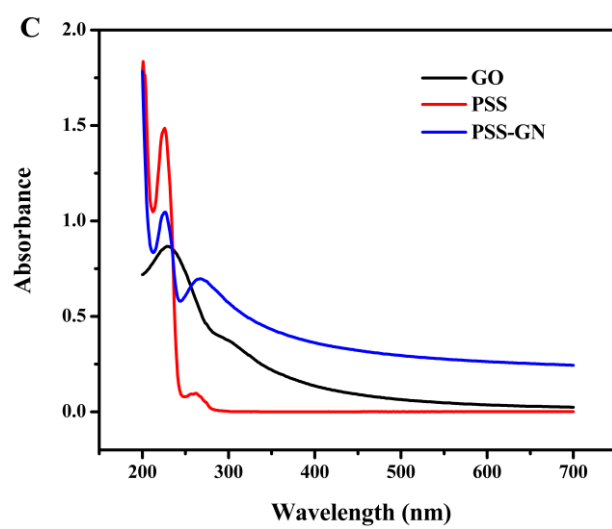
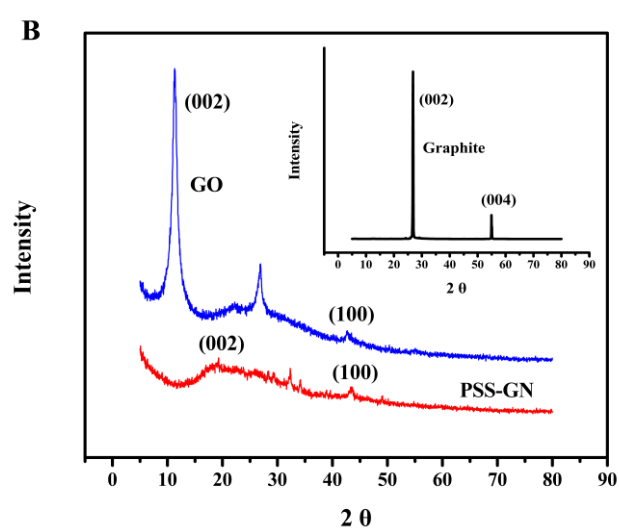
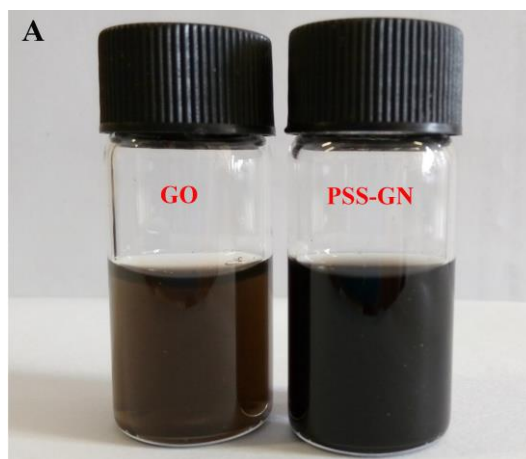
^b State Key Laboratory of Chemo/Biosensing and Chemometrics, Hunan University,
Changsha 410082, P.R. China

*Corresponding author

E-mail: zhaohui.li@zzu.edu.cn; qulingbo@zzu.edu.cn

Scheme S1 Schematic illustration of the formation of PSS-GN nanocomposites.





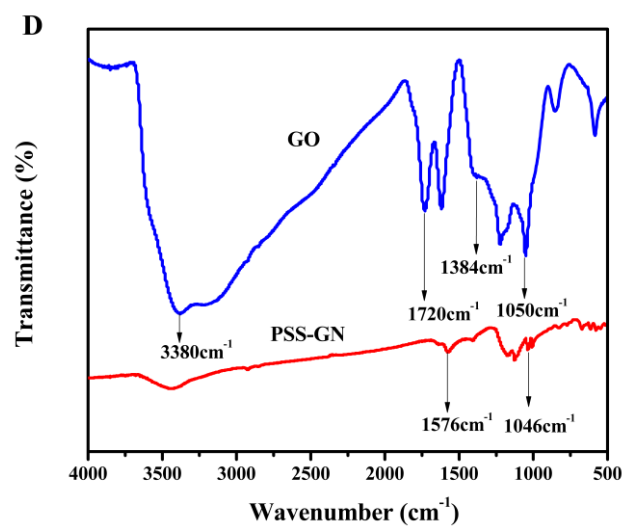


Fig. S1 (A) Photographs of water dispersion of 0.5 mg mL^{-1} GO and 0.5 mg mL^{-1} PSS-GN nanocomposites. (B) XRD patterns of GO and PSS-GN nanocomposites; inset: graphite. (C) UV-vis spectra of GO, PSS, and PSS-GN nanocomposites. (D) FT-IR spectra of the GO and PSS-GN nanocomposites.

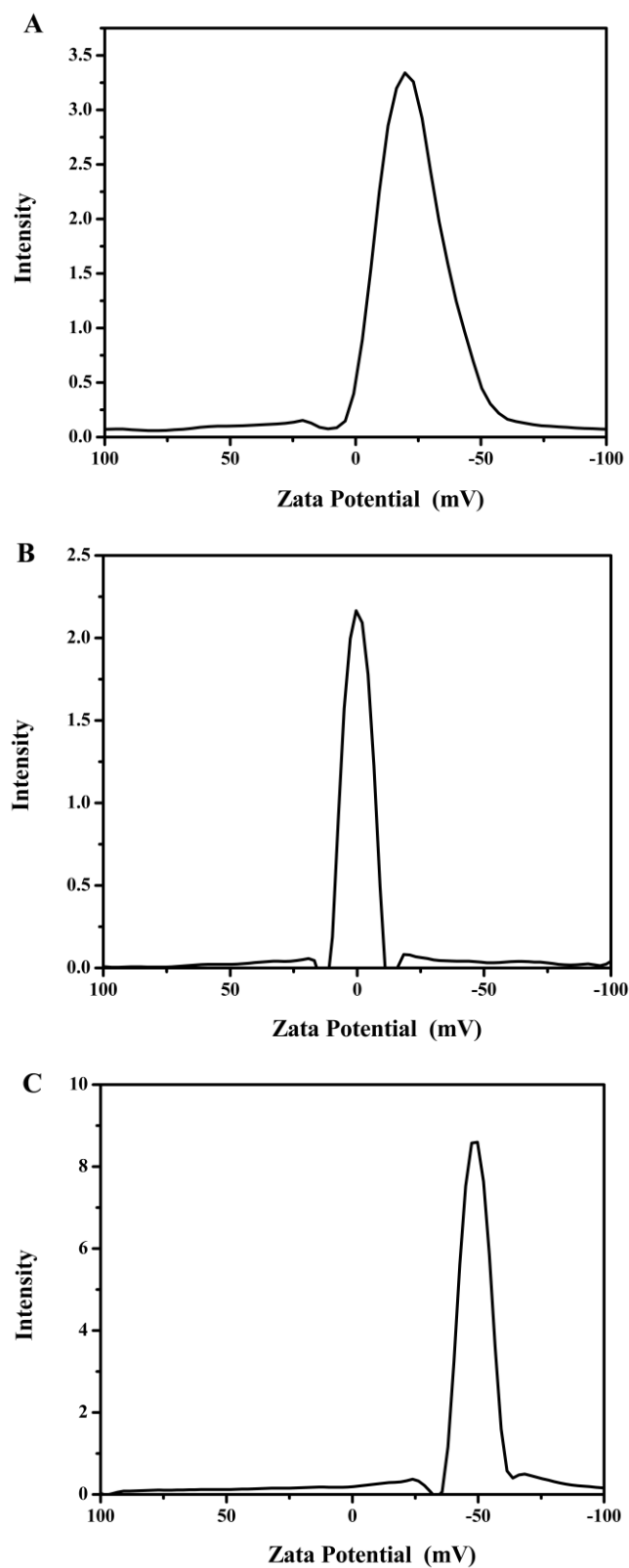


Fig. S2 Zeta potential analysis of GO (A), GN (B) and PSS-GN nanocomposites (C).

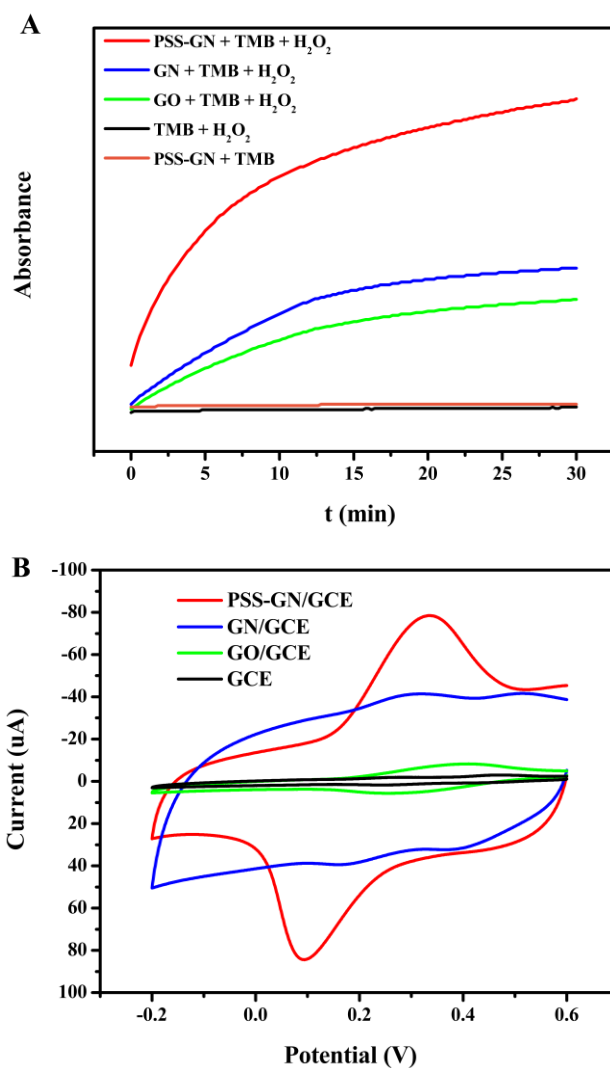
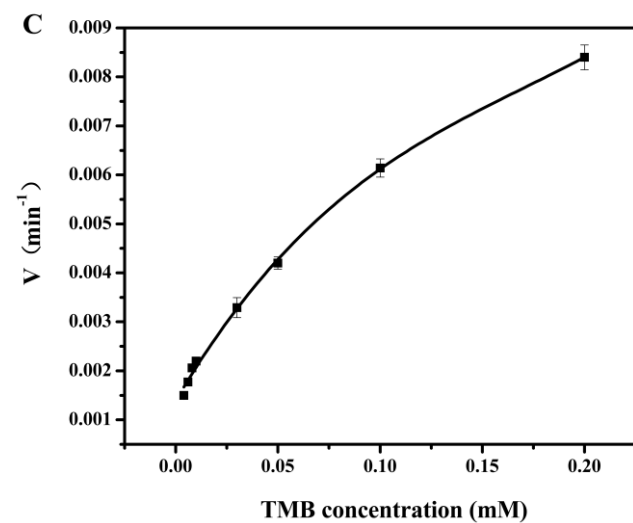
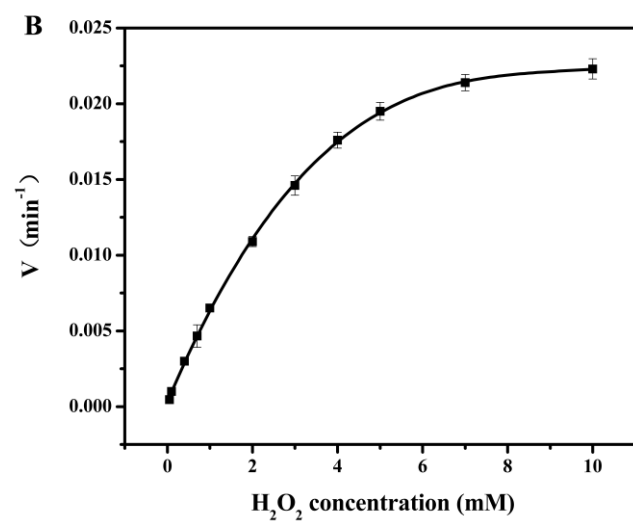
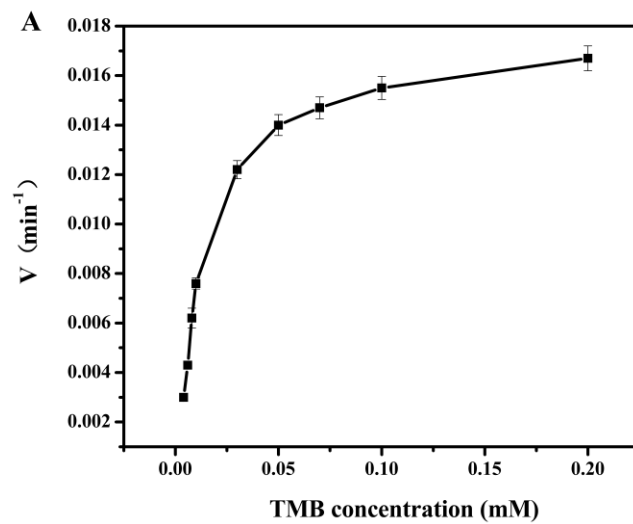


Fig. S3 (A) Time-dependent absorbance changes at 652 nm in different reaction systems. Reaction condition: $15 \mu\text{g mL}^{-1}$ enzyme mimics (GO, GN or PSS-GN), 0.2 mM TMB and 2 mM H_2O_2 . (C) Voltammogram of $16 \mu\text{M}$ TMB and $20 \mu\text{M}$ H_2O_2 in 0.1 M HAc-NaAc (pH 4.0) on different electrodes: bare GCE, GO/GCE, GN/GCE and PSS-GN/GCE. Accumulation time: 2 min; scan rate: 100 mV s^{-1} .



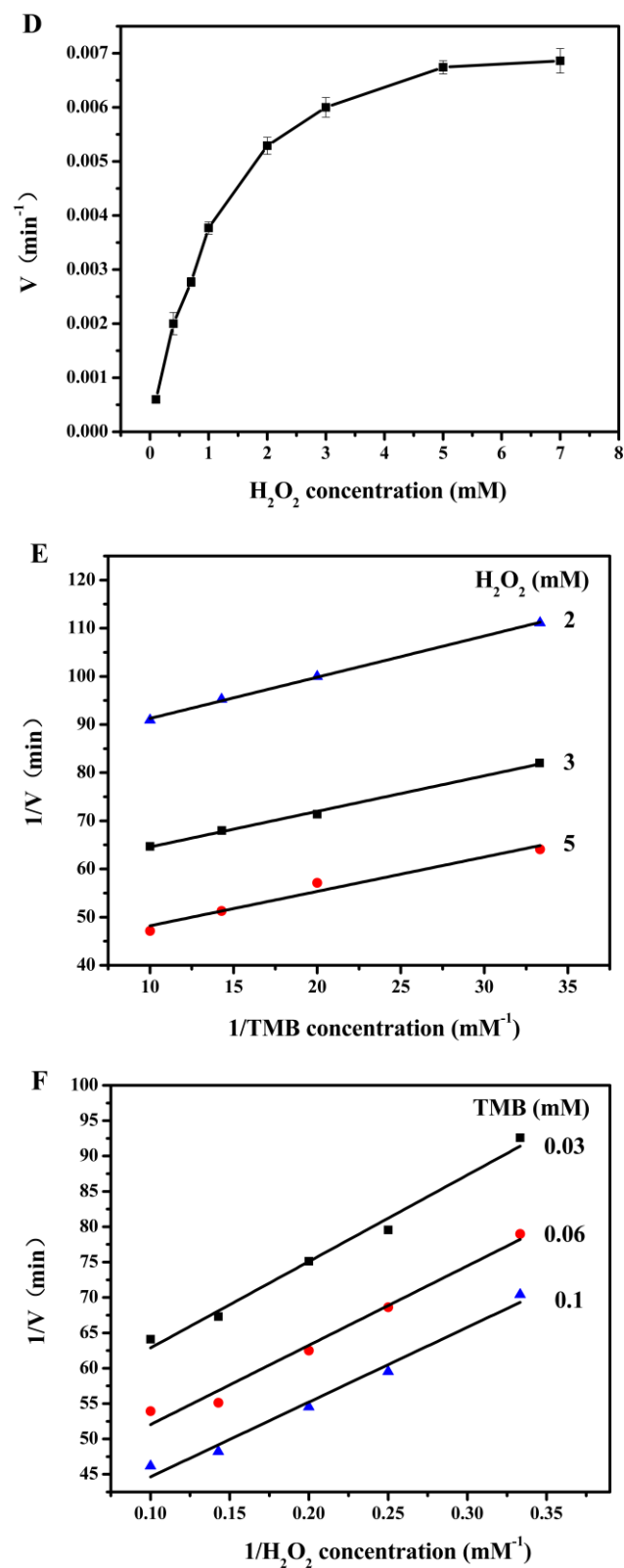


Fig. S4 Steady-state kinetic assay and catalytic mechanism of enzyme mimics

(PSS-GN nanocomposites and GO) (A-D): The velocity (v) of the reaction was measured using $15 \mu\text{g mL}^{-1}$ PSS-GN nanocomposites (A, B) or $15 \mu\text{g mL}^{-1}$ GO (C, D). (A, C) The concentration of H_2O_2 was 3 mM and the TMB concentration was varied. (B, D) The concentration of TMB was 0.2 mM and the H_2O_2 concentration was varied. (E and F) double reciprocal plots of activity of PSS-GN nanocomposites with the concentration of one substrate (H_2O_2 or TMB) fixed and the second substrate varied.

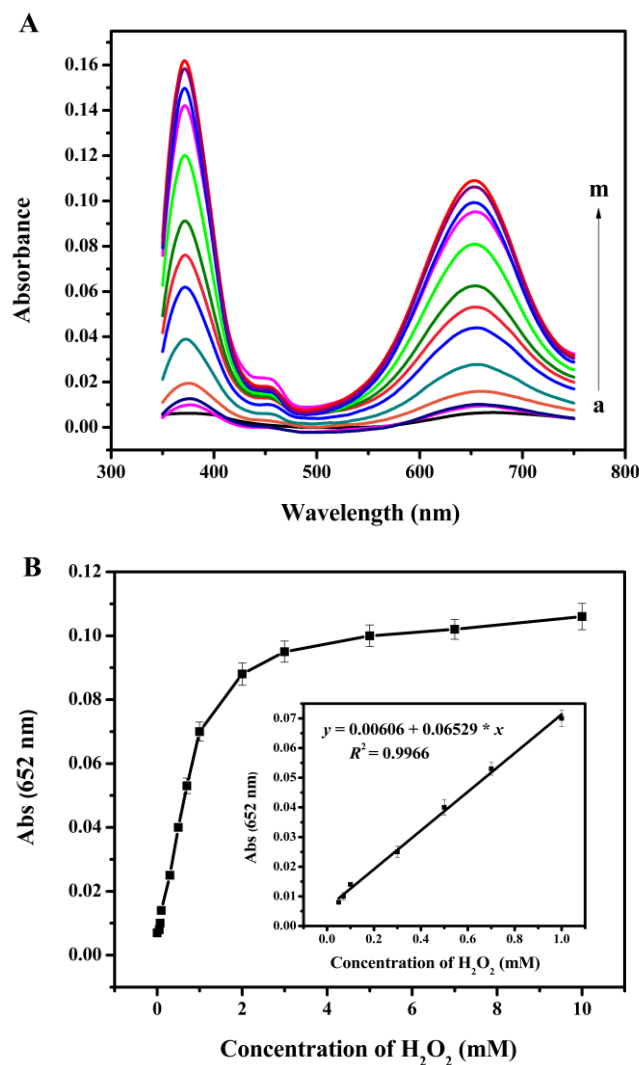


Fig. S5 (A) UV-vis absorption spectra of the TMB/GO solutions in the presence of H_2O_2 at various concentrations (a-m): 0, 0.05, 0.07, 0.1, 0.3, 0.5, 0.7, 1, 2, 3, 5, 7, and 10 mM using GO as an artificial mimetic. (B) The absorbance intensity at 652 nm plotted against the H_2O_2 concentration. The insert shows linear calibration curve between the absorbance at 652 nm and concentration of H_2O_2 .

Table 1S Comparison of the apparent Michaelis–Menten constant (K_m) and maximum reaction rate (V_{max}) of PSS-GN nanocomposites, GO and HRP.

Catalyst	K_m (mM)		V_{max} (10^{-8} M s $^{-1}$)		Ref.
	TMB	H ₂ O ₂	TMB	H ₂ O ₂	
PSS-GN	0.015	2.06	30.2	35.5	This study
GO	0.049	3.62	17.6	6.18	This study
HRP	0.43	3.7	10	8.71	1
C ₆₀ [C(COOH) ₂] ₂	0.23	24.58	34.73	40.11	2
C-Dots	0.039	26.77	3.61	30.61	3
CNPs	0.05	26.06	4.27	6.05	4
N-GQDs	11.19	0.10	0.38	0.14	5
HCNTs	0.020	0.4	-	-	6

Table S2 Comparison of nanomaterials-based methods for the determination of glucose

Nanomaterials	Method	Linear ranges (μM)	LODs (μM)	Refs
Fe ₃ O ₄ magnetic nanoparticles	Colorimetric	50-1000	30	7
TiO ₂ nanotubes	Electrochemical	400-3600	5	8
PEI/{GOD/PEI} ₃ /CNT	Electrochemical	0-300	50	9
RGO/HAp/GOx	Electrochemical	100-11500	30	10
WS ₂ nanosheets	Colorimetric	5-300	2.9	11
ZnFe ₂ O ₄ Nanoparticles	Colorimetric	1.25-18.75	0.3	12
Magnetic mesoporous carbon	Electrochemical	500-10000	200	13
PVA/GQD	Fluorescence	250-24000	10	14
Carbon nanodots	Colorimetric	1-500	0.4	3
Nitrogen-doped graphene	Colorimetric	25-375	16	5
Quantum dots				
Carbon nitride dots	Colorimetric	1-5	0.5	15
Graphene dots	Colorimetric	0.5-200	0.5	16
g-C ₃ N ₄	Electrochemical	1000-12000	11	17
PEDOT/GO nanocomposite	Electrochemical	0.1-1300	0.047	18
GCNT-Fe ₃ O ₄ nanocomposite	Electrochemical	50-5000	22	19
GO-Fe ₃ O ₄ nanocomposites	Colorimetric	2-200	0.74	20
Co ₃ O ₄ /rGO nanocomposites	Colorimetric	1-100	1	21
Fe-Phen-CFs composite	Fluorescence	0.5-200	0.19	22
Molecular beacons	Fluorescence	10-1000	2	23
PSS-GN nanocomposites	Colorimetric	6-400	0.28	this work

References

- 1 L. Z. Gao, J. Zhuang, L. Nie, J. B. Zhang, Y. Zhang, N. Gu, T. H. Wang, J. Feng, D. L. Yang, S. Perrett and X. Y. Yan, *Nat. Nanotechnol.*, 2007, **2**, 577-583.
- 2 R. M. Li, M. M. Zhen, M. R. Guan, D. Q. Chen, G. Q. Zhang, J. C. Ge, P. Gong, C. R. Wang, C. Y. Shu, *Biosens. Bioelectron.*, 2013, **47**, 502-507.
- 3 W. B. Shi, Q. L. Wang, Y. J. Long, Z. L. Cheng, S. H. Chen, H. Z. Zheng and Y. M. Huang, *Chem. Commun.*, 2011, **47**, 6695-6697.
- 4 X. H. Wang, K. G. Qu, B. L. Xu, J. S. Ren and X. G. Qu, *Nano Res.*, 2011, **4**, 908-920.
- 5 L. P. Lin, X. H. Song, Y. Y. Chen, M. C. Rong, T. T. Zhao, Y. R. Wang, Y. Q. Jiang and X. Chen, *Anal. Chim. Acta.*, 2015, **869**, 89-95.
- 6 R. J. Cui, Z. D. Han and J. J. Zhu, *Chem. Eur. J.*, 2011, **17**, 9377-9384.
- 7 H. Wei and E. K. Wang, *Anal. Chem.*, 2008, **80**, 2250-2254.
- 8 L. L. Zhang, L. Han, P. Hu, L. Wang and S. J. Dong, *Chem. Commun.*, 2013, **49**, 10480-10482.
- 9 C. Y. Deng, J. H. Chen, Z. Nie and S. H. Si, *Biosens. Bioelectron.*, 2010, **26**, 213-219.
- 10 G. Bharath, R. Madhu, S. M. Chen, V. Veeramani, A. Balamurugan, D. Mangalaraj, C. Viswanathan and N. Ponpandian, *J. Mater. Chem. B*, 2015, **3**, 1360-1370.
- 11 T. R. Lin, L. S. Zhong, Z. P. Song, L. Q. Guo, H. Y. Wu, Q. Q. Guo, Y. Chen, F. F. Fu and G. N. Chen, *Biosens. Bioelectron.*, 2014, **62**, 302-307.
- 12 L. J. Feng, X. M. Zhou, C. L. Ren, H. H. Li and X. G. Chen, *Anal. Chem.*, 2012, **84**, 5753-5758.
- 13 M. Kim, Y. J. Ye, B. Y. Won, S. Shin, J. Lee and H. G. Park, *Adv. Funct. Mater.*, 2011, **21**, 2868-2875.
- 14 P. P. Zhang, X. N. Zhao, Y. C. Ji, Z. F. Ouyang, X. Wen, J. F. Li, Z. Q. Su and G. Wei, *J. Mater. Chem. B*, 2015, **3**, 2487-2496.
- 15 S. Liu, J. Q. Tian, L. Wang, Y. L. Luo and X. P. Sun, *RSC Adv.*, 2012, **2**, 411-413.
- 16 A. X. Zheng, Z. X. Cong, J. R. Wang, J. Li, H. H. Yang and G. N. Chen, *Biosens.*

Bioelectron., 2013, **49**, 519-524.

17 J. Q. Tian, Q. Liu, C. J. Ge, Z. C. Xing, A. M. Asiri, A. O. Al-Youbicd and X. P. Sun, *Nanoscale*, **2013**, *5*, 8921-8924.

18 N. Hui, W. T. Wang, G. Y. Xu and X. L. Luo, *J. Mater. Chem. B*, 2015, **3**, 556-561.

19 H. Wang, S. A. Li, Y. M. Si, Z. Z. Sun, S. Y. Li and Y. H. Lin, *J. Mater. Chem. B*, 2014, **2**, 4442-4448.

20 Y. L. Dong, H. G. Zhang, Z. U. Rahman, L. Su, X. J. Chen, J. Hu and X. G. Chen, *Nanoscale*, 2012, **4**, 3969-3976.

21 J. X. Xie, H. Y. Cao, H. Jiang, Y. J. Chen, W. B. Shi, H. Z. Zheng and Y. M. Huang, *Anal. Chim. Acta.*, 2013, **796**, 92-100.

22 R. Z. Zhang, S. J. He, C. M. Zhang and W. Chen, *J. Mater. Chem. B*, 2015, **3**, 4146-4154.

23 R. Hu, Y. R. Liu, X. B. Zhang, W. H. Tan, G. L. Shen and Yu, R. Q. *Biosens. Bioelectron.*, 2013, **41**, 442-445.