

A novel bisoxazoline/Pd composite microsphere: a highly active catalyst for the Heck reactions

Junke Wang,^a Yingxiao Zong,^{a,b} Guoren Yue,^b Yulai Hu^{*b} and Xicun Wang^{*b}

^aKey Laboratory of Hexi corridor resources utilization of Gansu Universities, College of Chemistry and Chemical Engineering, Hexi University, Zhangye 734000, China

^bGansu Key Laboratory of Polymer Materials, College of Chemistry and Chemical Engineering, Northwest Normal University, Lanzhou 730070, China

Table of Contents

1. Materials and methods	2
2. Optimization of the catalytic conditions	3
3. Preparation and analytical data of catalyst C	4
4. General Experimental Procedures for Heck Couplings.	5
5. NMR spectra of the materials and products.....	8

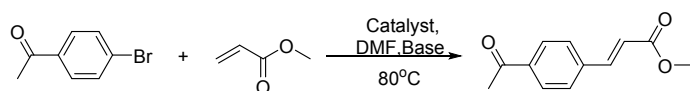
1. Materials and methods

Melting points were determined on a Perkin-Elmer differential scanning calorimeter and were uncorrected. The IR spectra were run on a Nicolette spectrometer (KBr). NMR spectra were recorded at 400 (^1H) and 100 (^{13}C) MHz, respectively, on a Varian Mercury plus-400 instrument using CDCl_3 as solvent and TMS as the internal standard. Scanning electron microscopy (SEM) was performed on a FEI Quanta 450 FEG FESEM instrument. High resolution mass spectra (HRMS) were obtained on an Agilent LC-MSD-Trap-XCT spectrometer with micromass MS software using electrospray ionisation (ESI). All the solvents used were strictly dried according to standard operation and stored on 4 Å molecular sieves.

All other chemicals (AR grade) were commercially available and used without further purification.

2. Optimization of the catalytic conditions

Table 1 The effect of solvents and bases^a



Entry	Catalyst (Pd mol%)	Base	Time (h)	Yield(%) ^b
1	0.05	Na_2CO_3	2	56
2	0.05	K_2CO_3	2	72
3	0.05	KOAc	2	40
4	0.05	NaOAc	2	30
5	0.05	TEA	2	21
6	0.05	Pyridine	2	14
7	0	K_2CO_3	2	0
8	0.075	K_2CO_3	2	80
9	0.10	K_2CO_3	2	90
10	0.15	K_2CO_3	2	90
11	0.10	K_2CO_3	3	94
12	0.10	K_2CO_3	4	98
13	0.10	K_2CO_3	5	98

^aReaction conditions: bisoxazoline/Pd microsphere, 1 mmol of p-bromoacetophenone, 1 mmol of methyl acrylate, 2 mmol of base, 5 ml of solvent, 80 °C in air.

^b Isolated yield.

3. Preparation and analytical data of catalyst C

Synthesis of bisacylthiourea B

To a solution of 4,4'-Oxybisbenzoyl chloride **A** (2 mmol) in CH₂Cl₂ (10 mL) was added ammonium thiocyanate (2.6 mmol) and PEG-400 (0.2 mmol). The mixture was then stirred at room temperature for 60 min and cooled to 0°C, and the solution of 2-aminoethanol (1.8 mmol) in CH₂Cl₂ (2 mL) was added. The mixture was continuously stirred for 60 min. After the completion of the reaction, the solvent was removed by distillation, and water (10 mL) was added to obtain a white solid. The analytical sample was produced by flash chromatography (acetone and petroleum ether) to give a white solid **B**. Yield: 85%. Melting point: 209-211°C. Spectral data: IR (KBr) (cm⁻¹): ν 3337, 3225, 2944, 1670, 1531. ¹H NMR (400 MHz, DMSO) δ 11.35 (s, 2H), 11.05 (s, 2H), 8.02 (d, J = 8.8 Hz, 4H), 7.17 (d, J = 8.8 Hz, 4H), 4.98 (s, 2H), 3.83-3.44 (m, 8H). ¹³C NMR (100 MHz, DMSO) δ 180.71, 167.65, 159.87, 131.68, 128.22, 118.95, 58.75, 47.97, 40.38, 40.17, 39.96, 39.75, 39.54. HR-MS: m/z calcd for C₂₀H₂₁N₂O₅S₂ [M+H]⁺: 433.0892; found: 433.0889.

Synthesis of bisoxazoline C

To a solution of compound **B** (1 mmol) in DMF (5 mL) was added dicyclohexylcarbodiimide (DCC) (1 mmol) and TEA (1 mmol). The mixture was stirred for 2 h at 80°C, and cooled to room temperature. After the addition of water (5 mL), the white solid was obtained by the filtration. This solid was added into CH₃CN (5 mL) to be dissolved, followed by the filtration and concentration to afford the target compound **C**. Yield: 98%. Melting point: 195-196°C. Spectral data: IR (KBr) (cm⁻¹): ν 3310, 2921, 1638, 1548. ¹H NMR (400 MHz, DMSO) δ 9.61 (s, 2H), 8.28 – 7.99 (m, 4H), 7.20 – 6.98 (m, 4H), 4.47 (t, J = 8.6 Hz, 4H), 3.78 (t, J = 8.6 Hz, 4H). ¹³C NMR (100 MHz, DMSO) δ 180.71, 167.65, 159.87, 131.68, 128.22, 118.95, 58.75, 47.97, 40.59, 40.38, 40.17, 39.96, 39.75, 39.54, 39.33. HR-MS: m/z calcd for C₂₀H₁₉N₄O₅ [M+H]⁺: 395.1355; found: 395.1395.

Synthesis of catalyst D

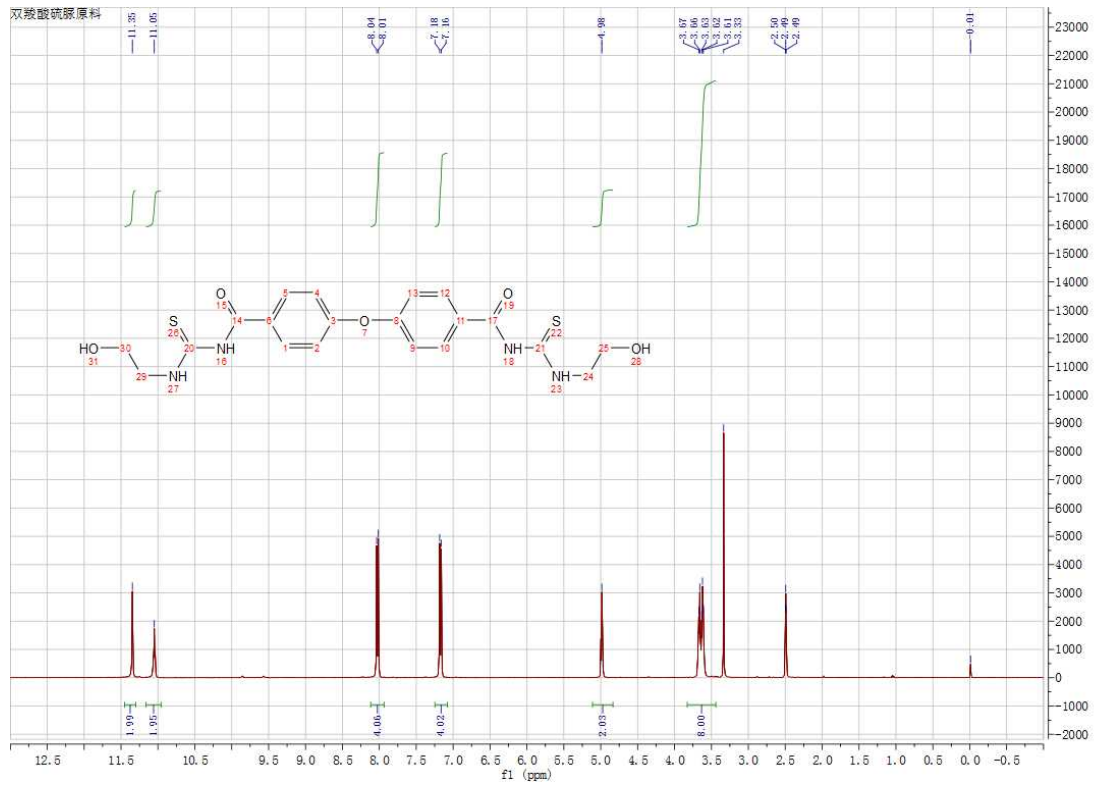
To the solution of Pd(AcO)₂ (2 mmol) in CH₃CN (5 mL) was added dropwise into the obtained compound **C** (1.36 g, 6 mmol) in CH₃CN (2 mL), followed by the stirring for 10 h. On completion, the filtration was conducted to a yellow solid. Washing with commercial anhydrous CH₃CN (3 × 5 mL) and drying at 50 °C overnight gave bisoxazoline/Pd microsphere as a pale yellow powder (compound **D**). IR (KBr) (cm⁻¹): ν 3443, 2907, 1592. The Pd content of the bisoxazoline/Pd microsphere catalyst is 20.01 wt% (1.8 mmol/g) measured by atomic absorption spectroscopy (AAS).

4. General Experimental Procedures for Suzuki-Miyaura Couplings

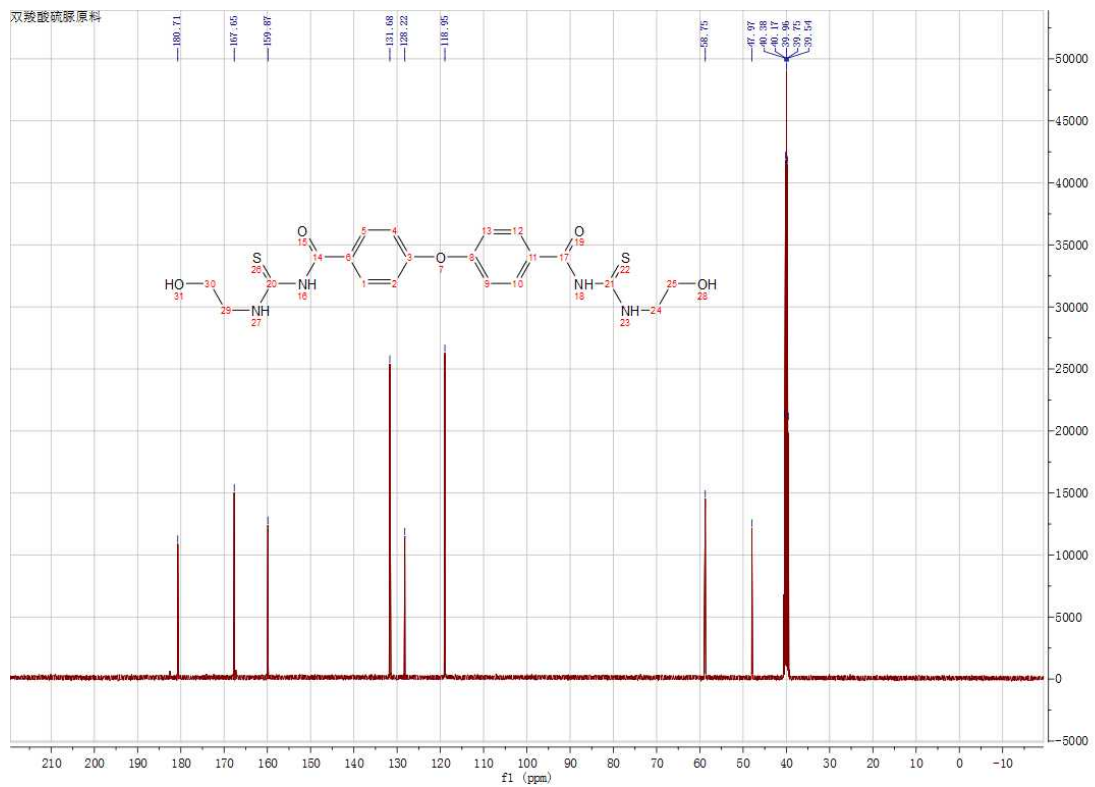
In a typical experiment, the bisoxazoline/Pd microsphere catalyst (0.10 mmol of Pd) was added to a mixture of aryl halide (1.0 mmol), olefins (1.2 mmol), and K₂CO₃ (1.0 mmol) in DMF (5.0 mL), and the reaction mixture was stirred at 80°C. After the reaction was monitored to be complete by TLC analysis, the catalyst was removed by filtration, washed with ethanol (3 × 3 mL), and dried under vacuum for the next run. The organic fractions were then concentrated on a rotary evaporator to afford the desired compound in excellent yield. The crude products were purified by column chromatography on silica gel using hexane/ethyl acetate.

5. NMR spectra of the materials and products

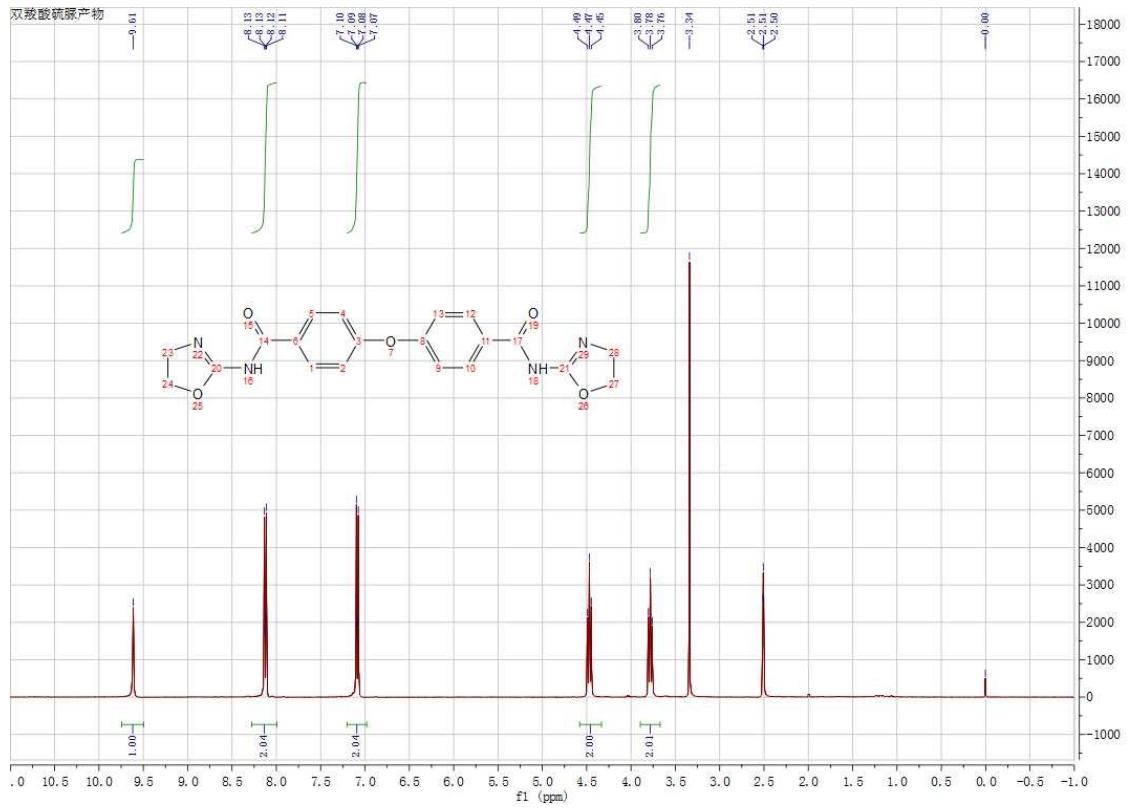
¹H NMR of bisacylthiourea B



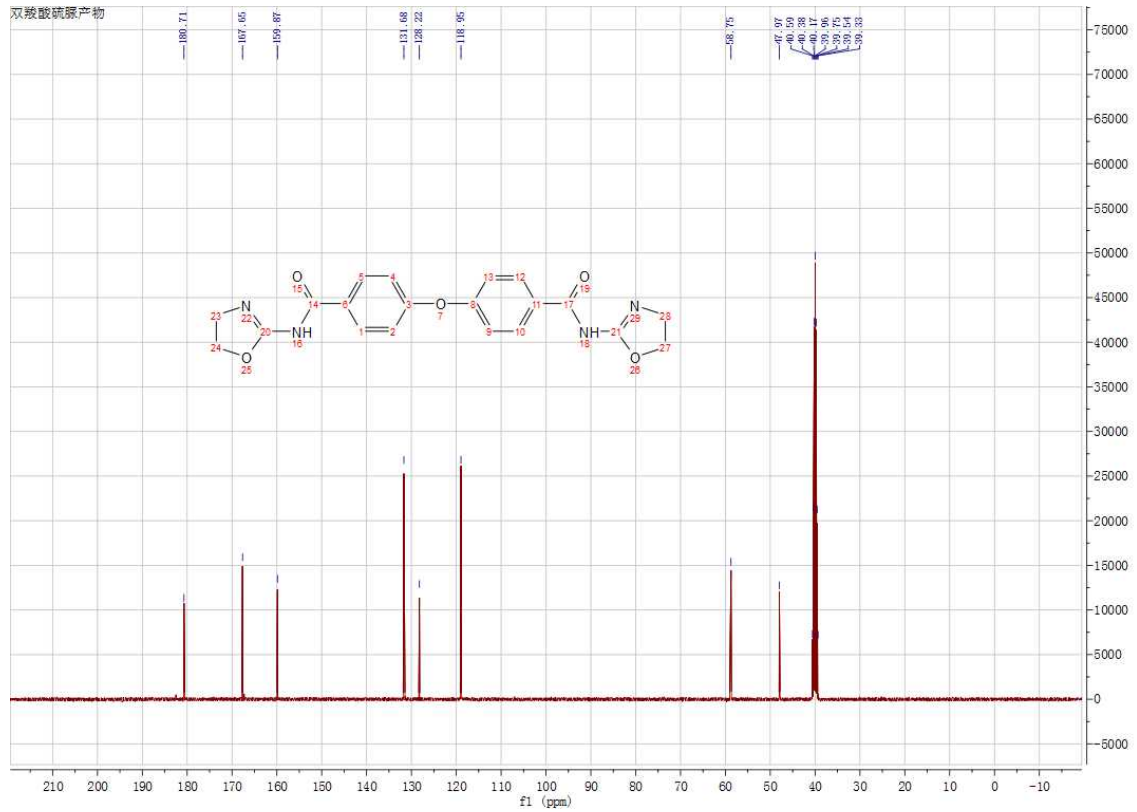
¹³C NMR of bisacylthiourea B



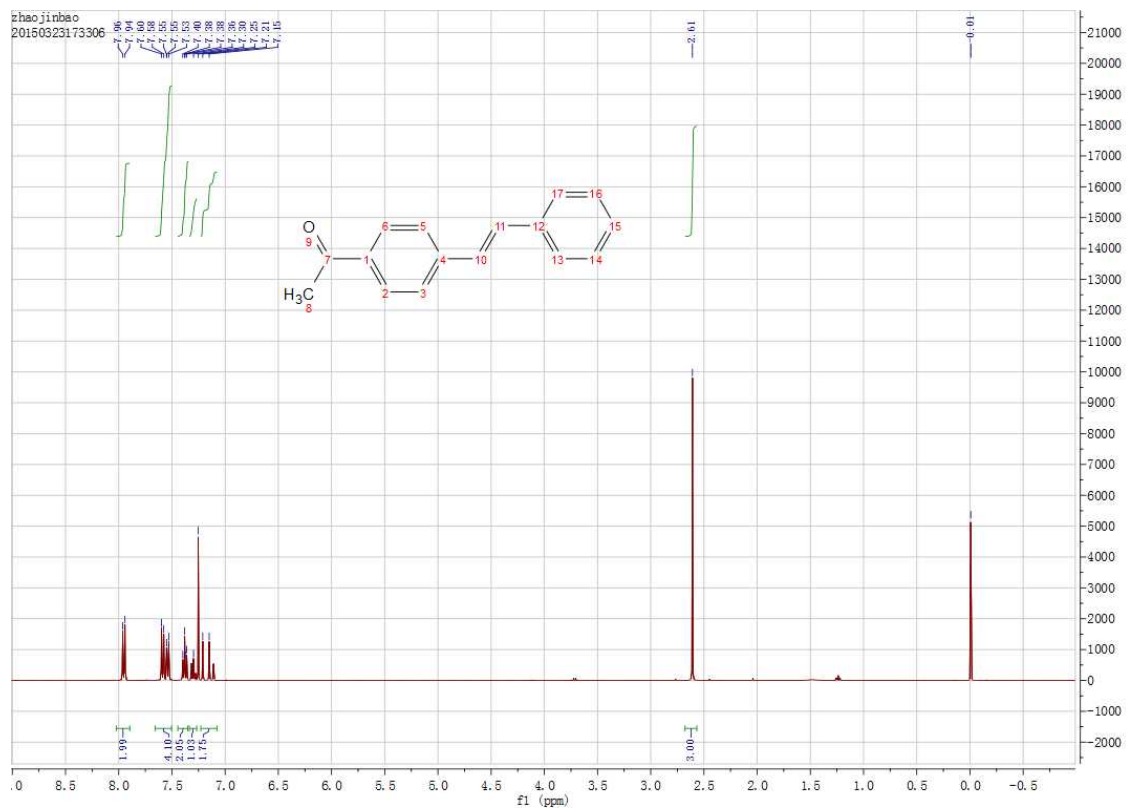
¹H NMR of bisoxazoline C



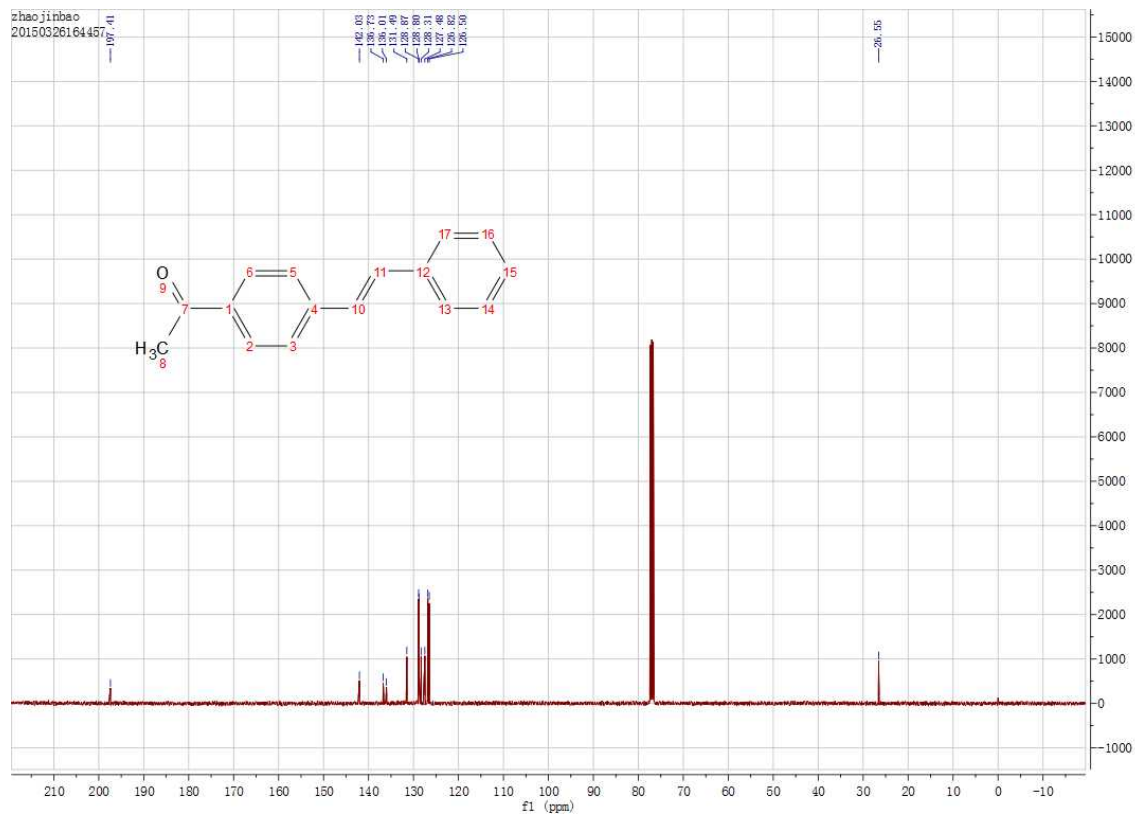
¹³C NMR of bisoxazoline C



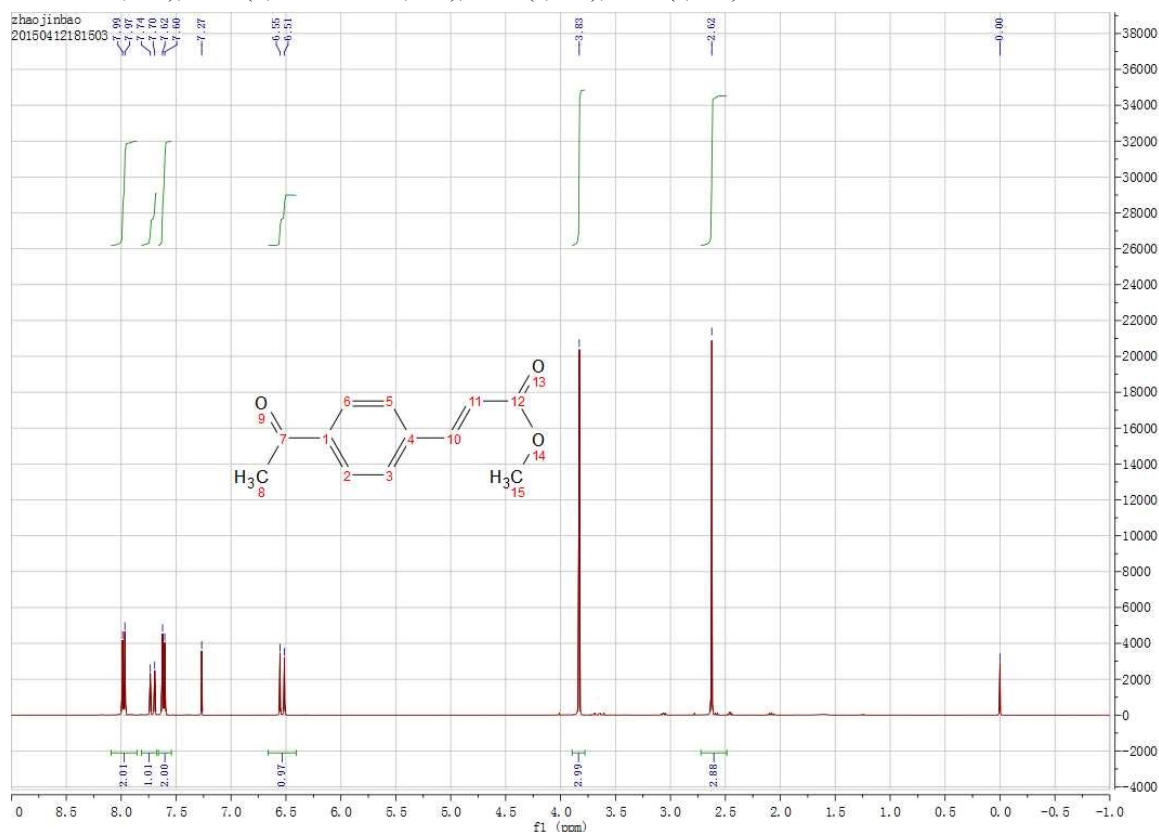
Entry 1. ^1H NMR (400 MHz, CDCl_3) δ 7.95 (d, $J = 8.4$ Hz, 2H), 7.66 – 7.50 (m, 4H), 7.38 (dd, $J = 8.1, 6.8$ Hz, 2H), 7.30 (s, 1H), 7.18 (d, $J = 24.2$ Hz, 2H), 2.61 (s, 3H).



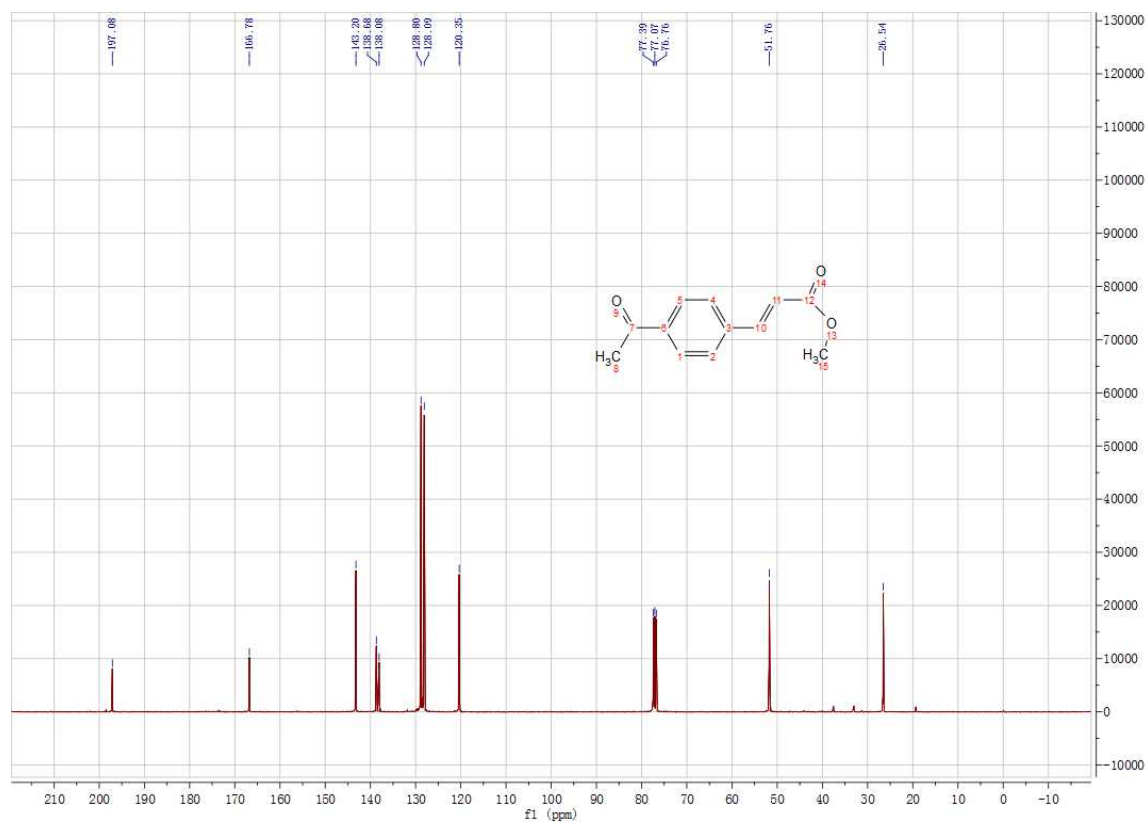
^{13}C NMR (101 MHz, CDCl_3) δ 197.41, 142.03, 136.73, 136.01, 131.49, 128.83, 128.31, 127.48, 126.82, 126.50, 26.55.



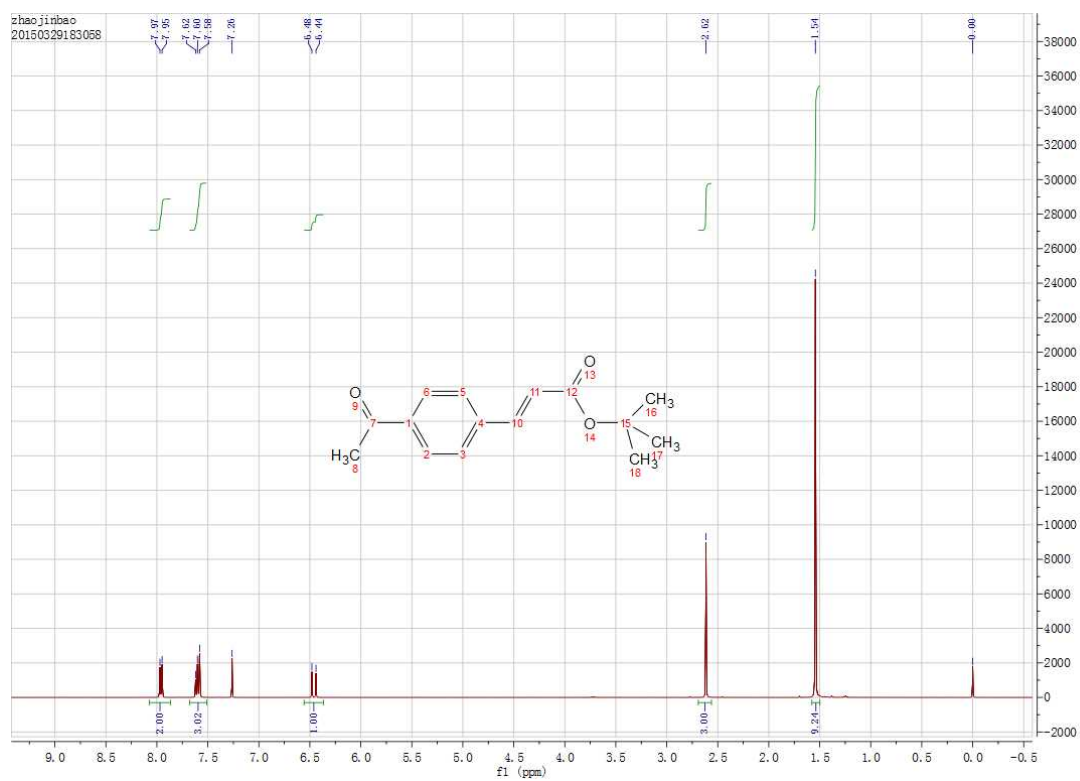
Entry 2. ^1H NMR (400 MHz, CDCl_3) δ 7.98 (d, $J = 8.3$ Hz, 2H), 7.72 (d, $J = 16.1$ Hz, 1H), 7.61 (d, $J = 8.2$ Hz, 2H), 6.53 (d, $J = 16.1$ Hz, 1H), 3.83 (s, 3H), 2.62 (s, 3H).



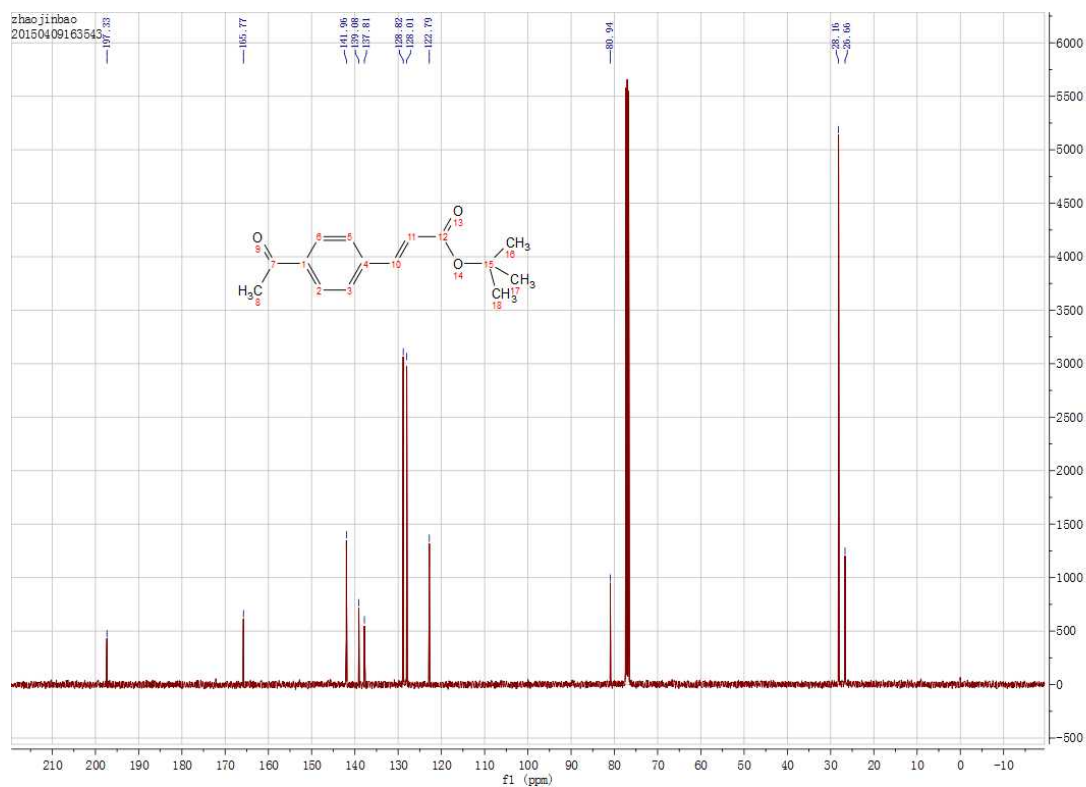
^{13}C NMR (101 MHz, CDCl_3) δ 197.08 , 166.78 , 143.20 , 138.68 , 138.08 , 128.80 , 128.09 , 120.35 , 77.39 , 77.07 , 76.76 , 51.76 , 26.54 .



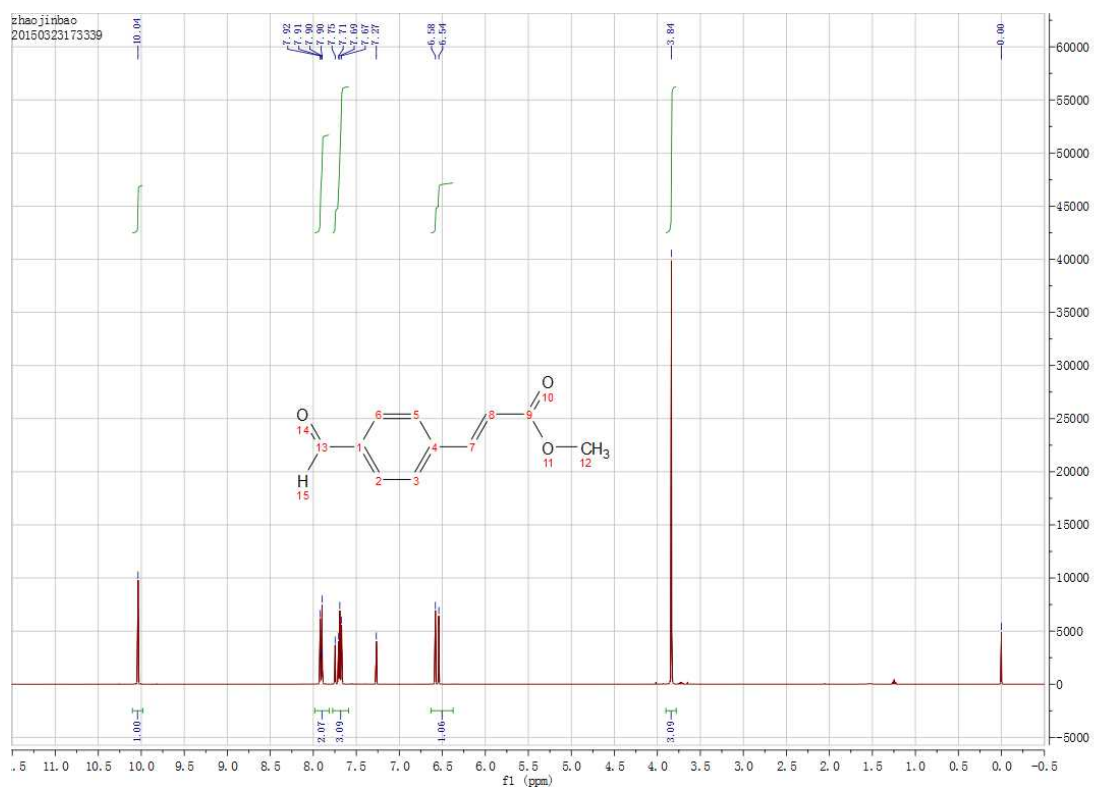
Entry 3. ^1H NMR (400 MHz, CDCl_3) δ 7.96 (d, $J = 8.3$ Hz, 2H), 7.68 – 7.51 (m, 3H), 6.46 (d, $J = 16.1$ Hz, 1H), 2.62 (s, 3H), 1.54 (s, 9H).



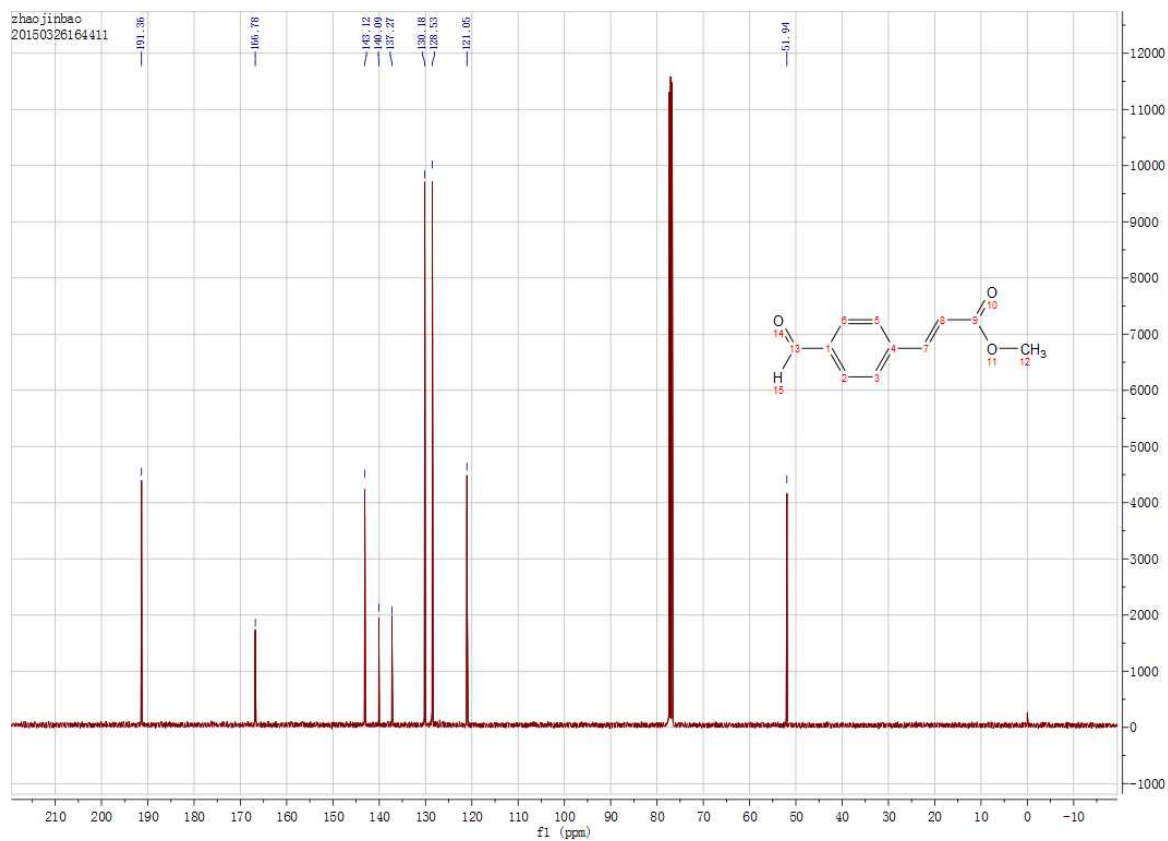
^{13}C NMR (101 MHz, CDCl_3) δ 197.33 , 165.77 , 141.96 , 139.08 , 137.81 , 128.82 , 128.01 , 122.79 , 80.94 , 77.34 , 77.02 , 76.70 , 28.16 , 26.66 .



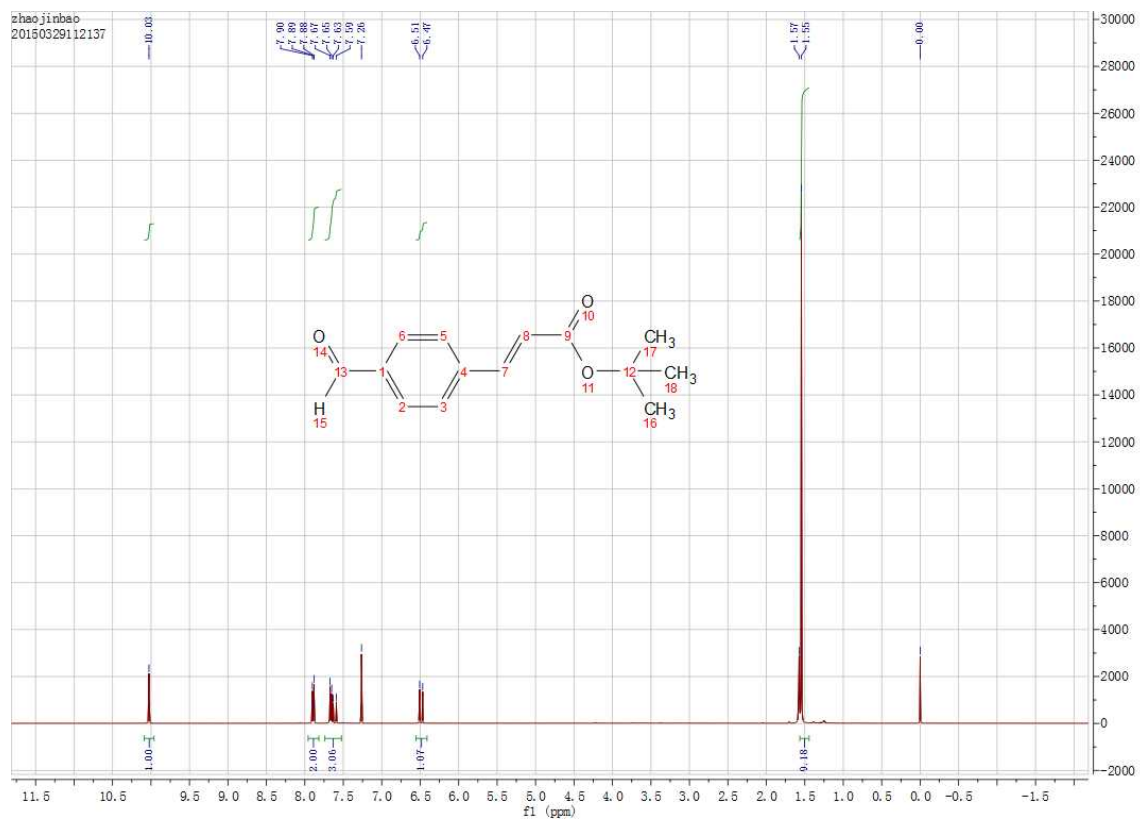
Entry 4. ^1H NMR (400 MHz, CDCl_3) δ 10.04 (s, 1H), 7.98 – 7.82 (m, 2H), 7.70 (dd, $J = 17.7, 12.2$ Hz, 3H), 6.56 (d, $J = 16.0$ Hz, 1H), 3.84 (s, 3H).



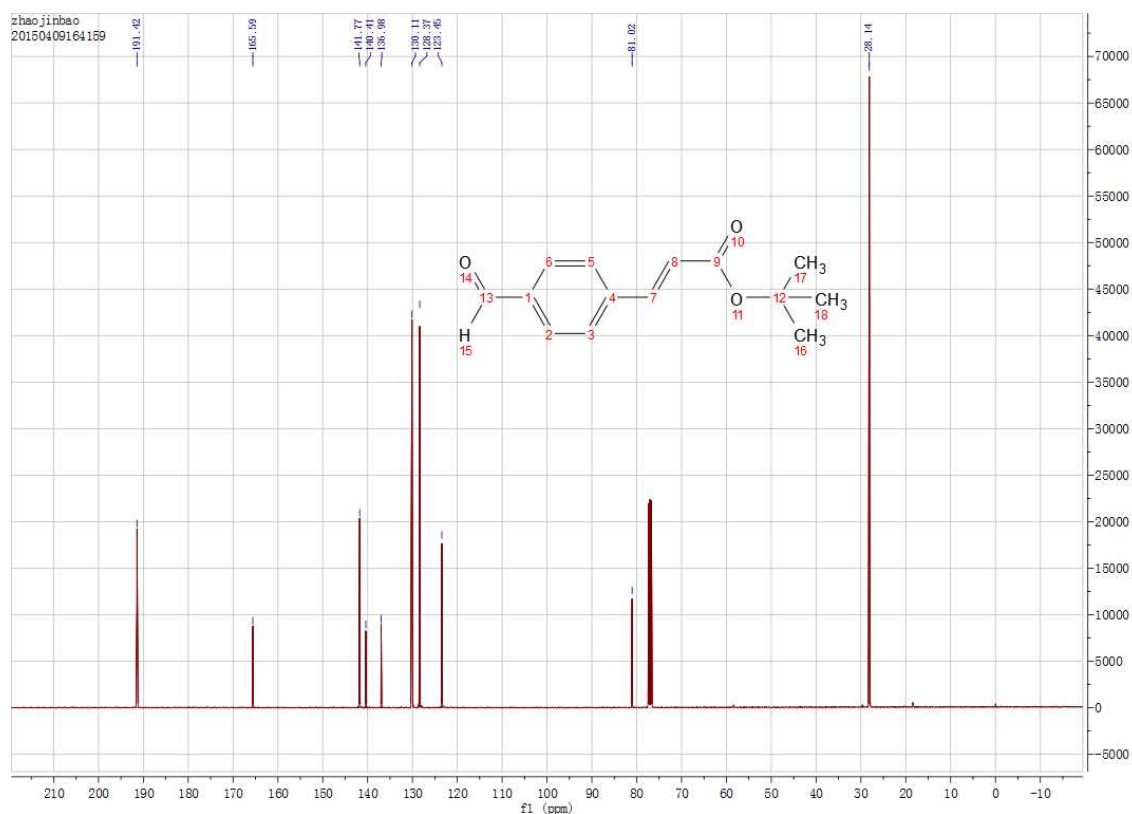
^{13}C NMR (101 MHz, CDCl_3) δ 191.36, 166.78, 143.12, 140.09, 137.27, 130.18, 128.53, 121.05, 51.94.



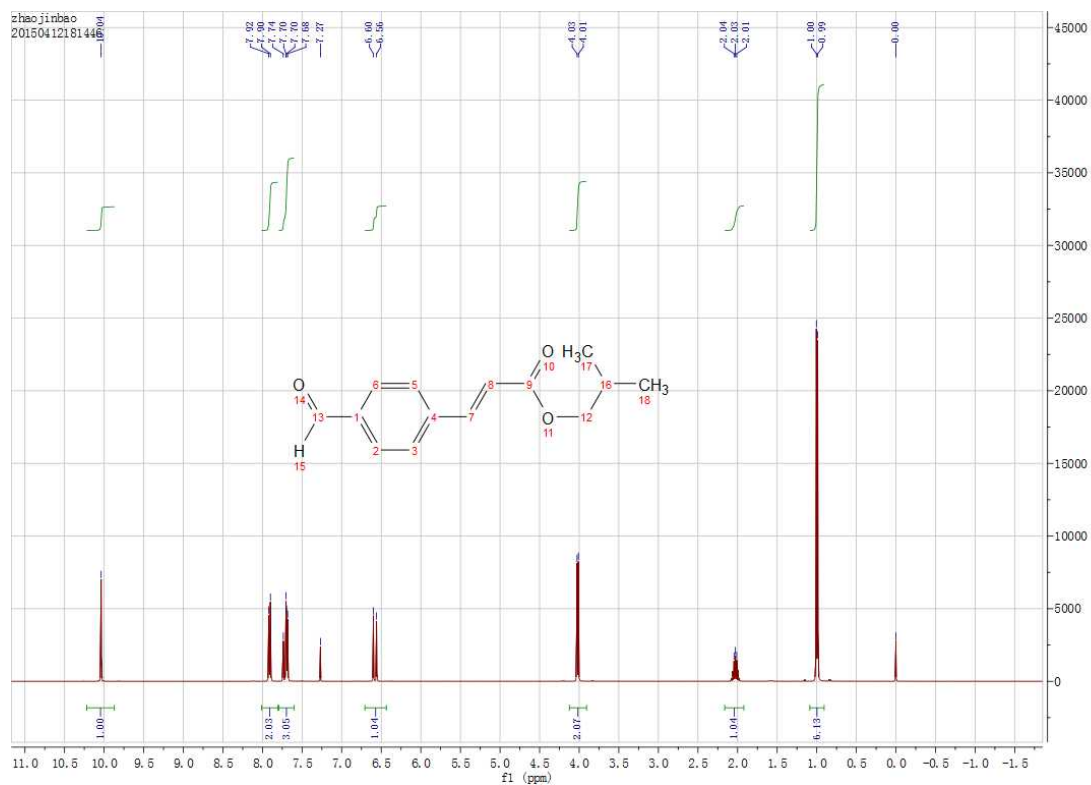
Entry 5. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 10.03 (s, 1H), 7.96 – 7.82 (m, 2H), 7.64 (dd, $J = 20.3, 12.2$ Hz, 3H), 6.49 (d, $J = 16.0$ Hz, 1H), 1.55 (s, 9H).



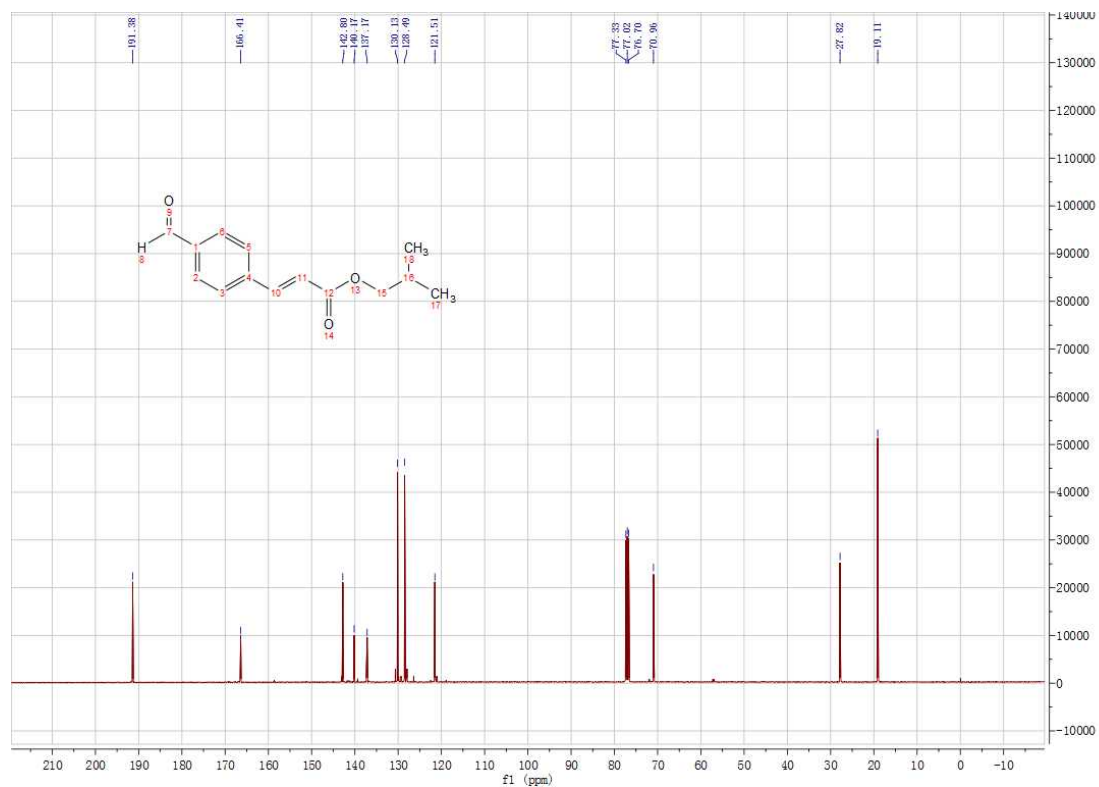
^{13}C NMR (101 MHz, CDCl_3) δ 191.42 , 165.59 , 141.77 , 140.41 , 136.98 , 130.11 , 128.37 , 123.45 , 81.02 , 28.14 .



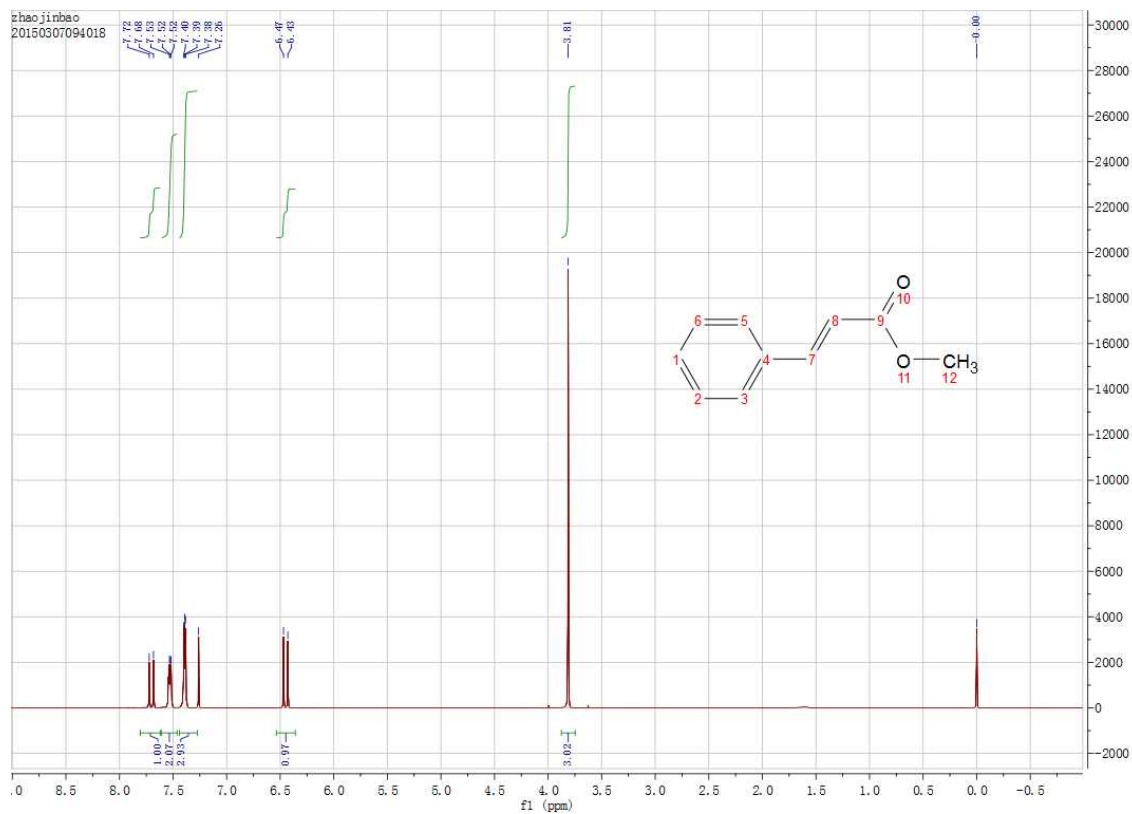
Entry 6. ^1H NMR (400 MHz, CDCl_3) δ 10.04 (s, 1H), 7.91 (d, $J = 8.3$ Hz, 2H), 7.70 (dd, $J = 12.0$, 10.0 Hz, 3H), 6.58 (d, $J = 16.1$ Hz, 1H), 4.02 (d, $J = 6.7$ Hz, 2H), 2.16 – 1.92 (m, 1H), 1.00 (d, $J = 6.7$ Hz, 6H).



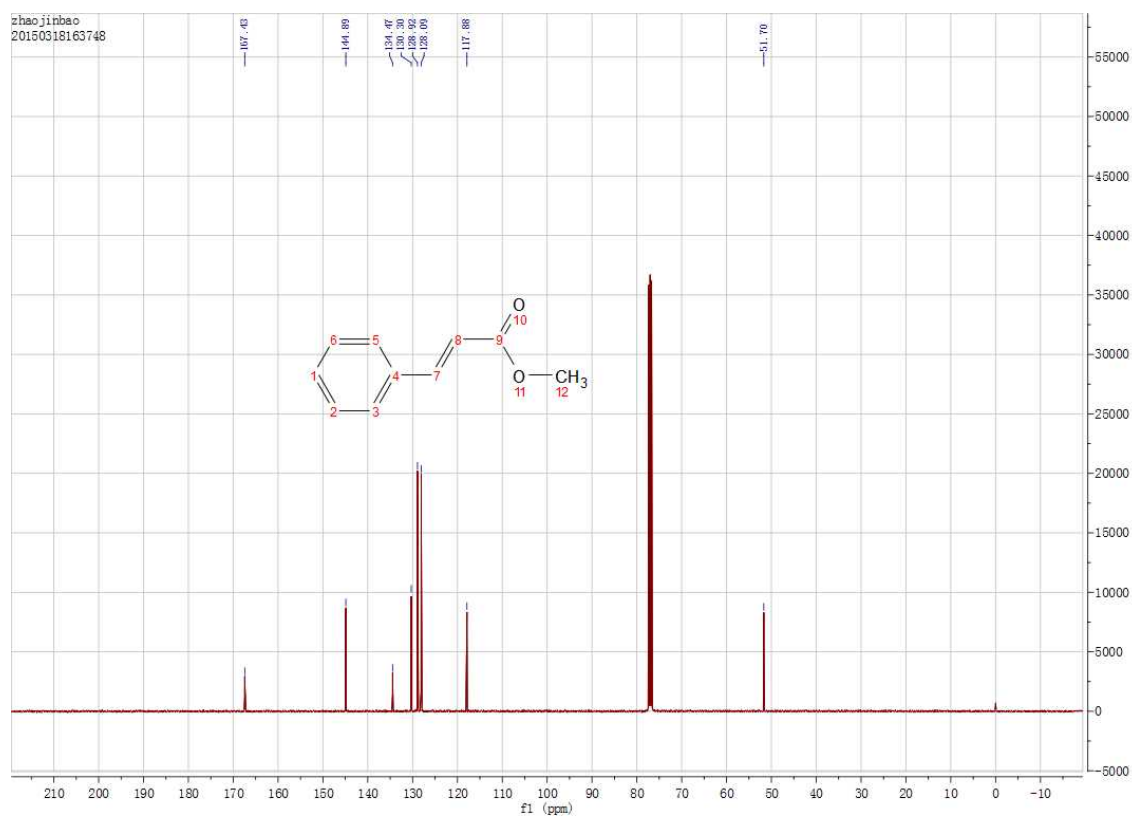
^{13}C NMR (101 MHz, CDCl_3) δ 191.38 , 166.41 , 142.80 , 140.17 , 137.17 , 130.13 , 128.49 , 121.51 , 77.33 , 77.02 , 76.70 , 70.96 , 27.82 , 19.11 .



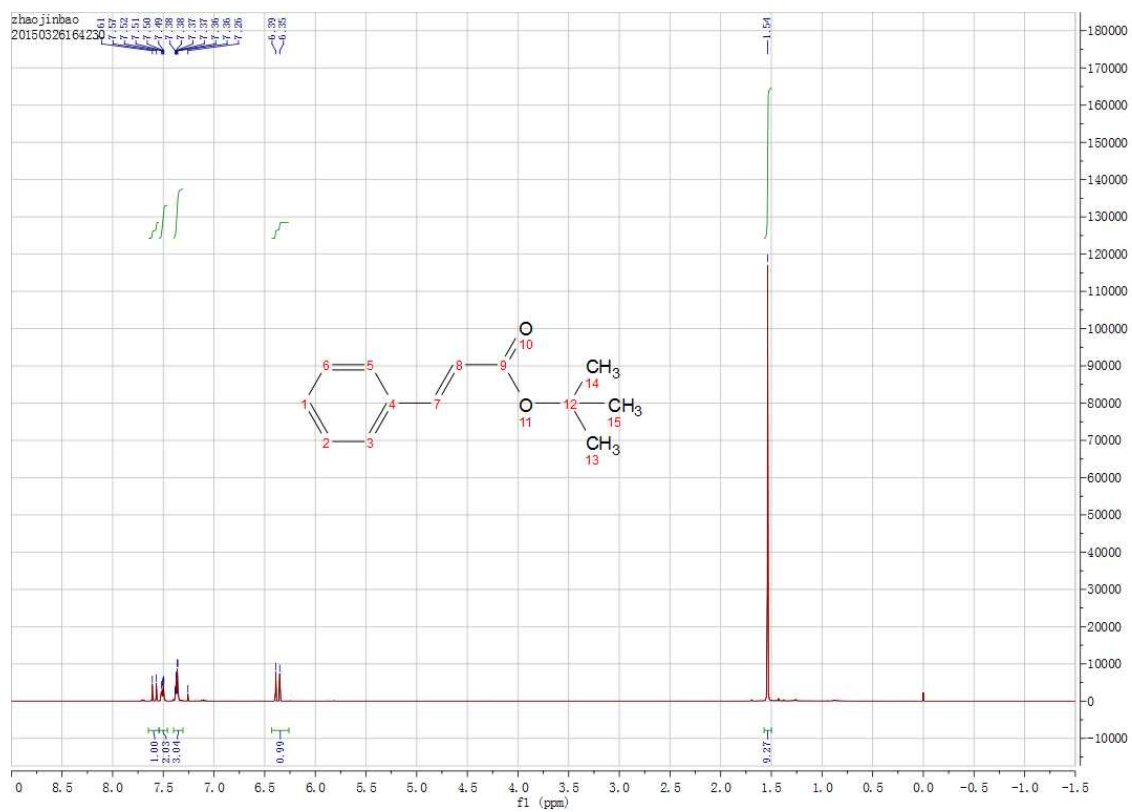
Entry 7. ^1H NMR (400 MHz, CDCl_3) δ 7.70 (d, J = 16.0 Hz, 1H), 7.61 – 7.46 (m, 2H), 7.44 – 7.27 (m, 3H), 6.45 (d, J = 16.0 Hz, 1H), 3.81 (s, 3H).



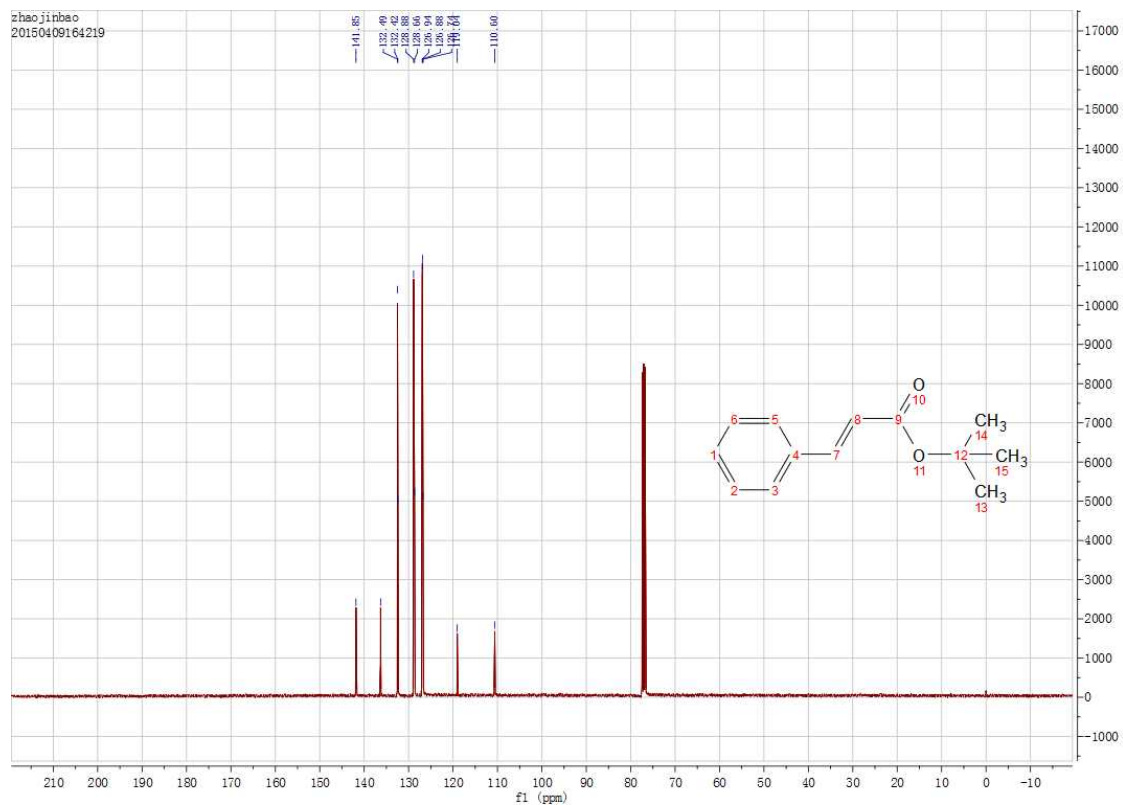
^{13}C NMR (101 MHz, CDCl_3) δ 167.43 , 144.89 , 134.47 , 130.30 , 128.92 , 128.09 , 117.88 , 51.70 .



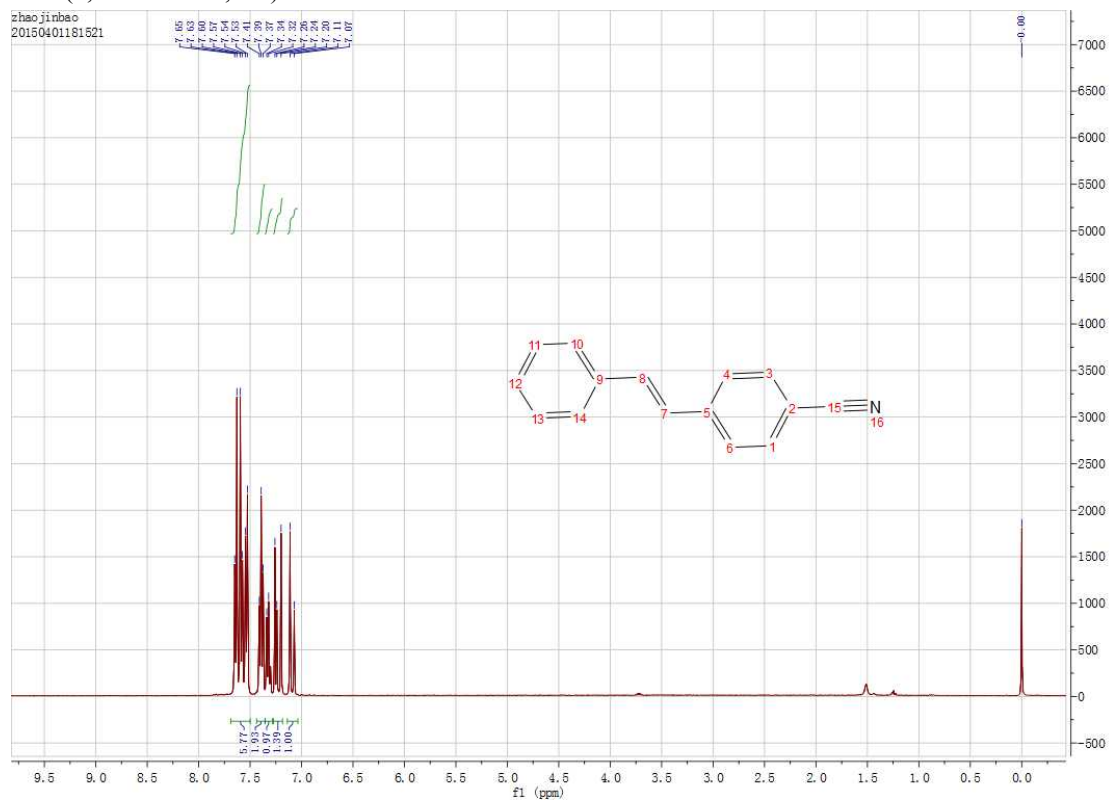
Entry 8. ^1H NMR (400 MHz, CDCl_3) δ 7.59 (d, $J = 16.0$ Hz, 1H), 7.51 (dd, $J = 6.7, 3.3$ Hz, 2H), 7.40 – 7.31 (m, 3H), 6.37 (d, $J = 16.0$ Hz, 1H), 1.54 (s, 9H).



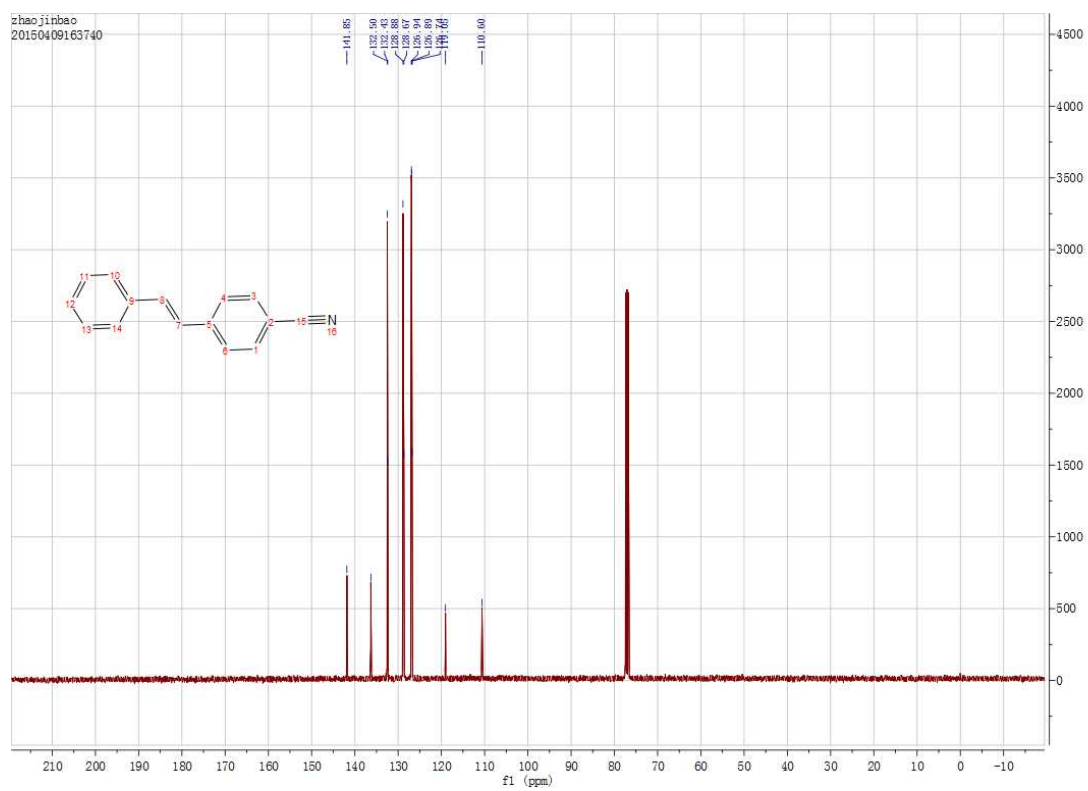
^{13}C NMR (101 MHz, CDCl_3) δ 141.85 , 136.31 , 132.46 , 128.88 , 128.66 , 127.04, 126.55, 119.04 , 110.60 .



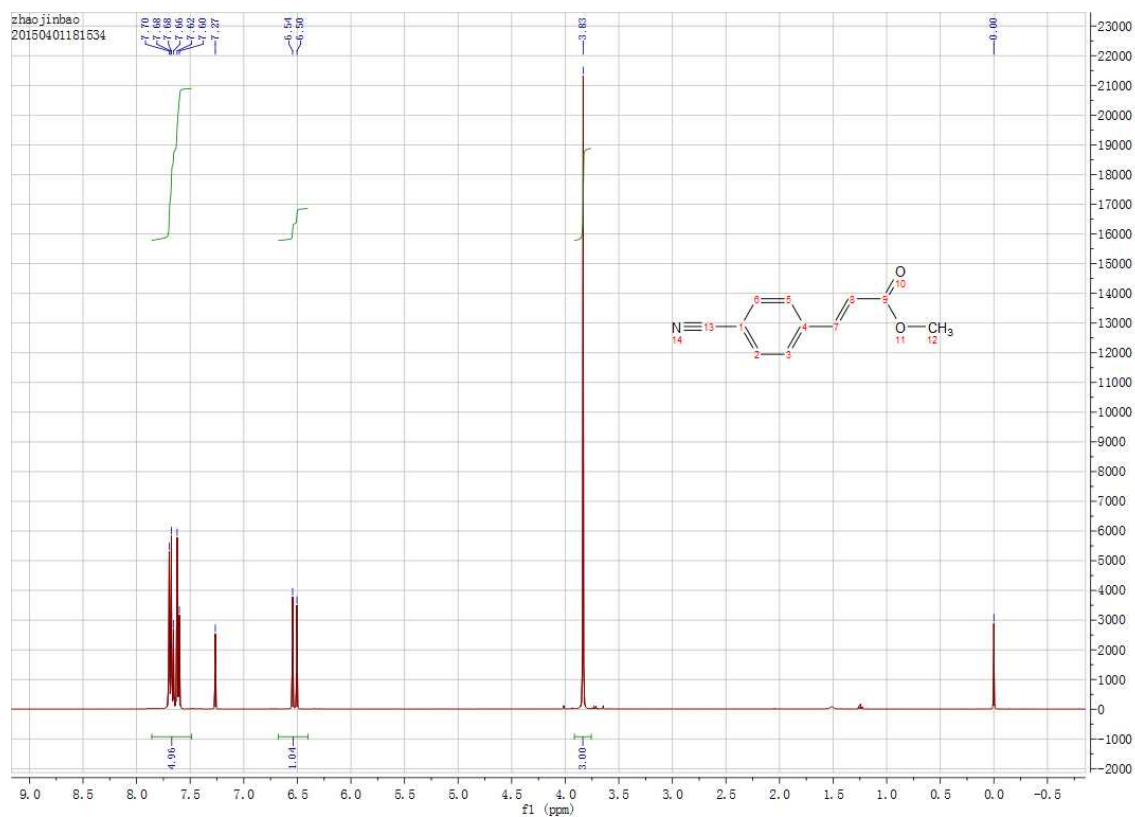
Entry 9. ^1H NMR (400 MHz, CDCl_3) δ 7.69 – 7.50 (m, 6H), 7.32 (ddd, $J = 48.4, 19.4, 11.8$ Hz, 4H), 7.09 (d, $J = 16.3$ Hz, 1H).



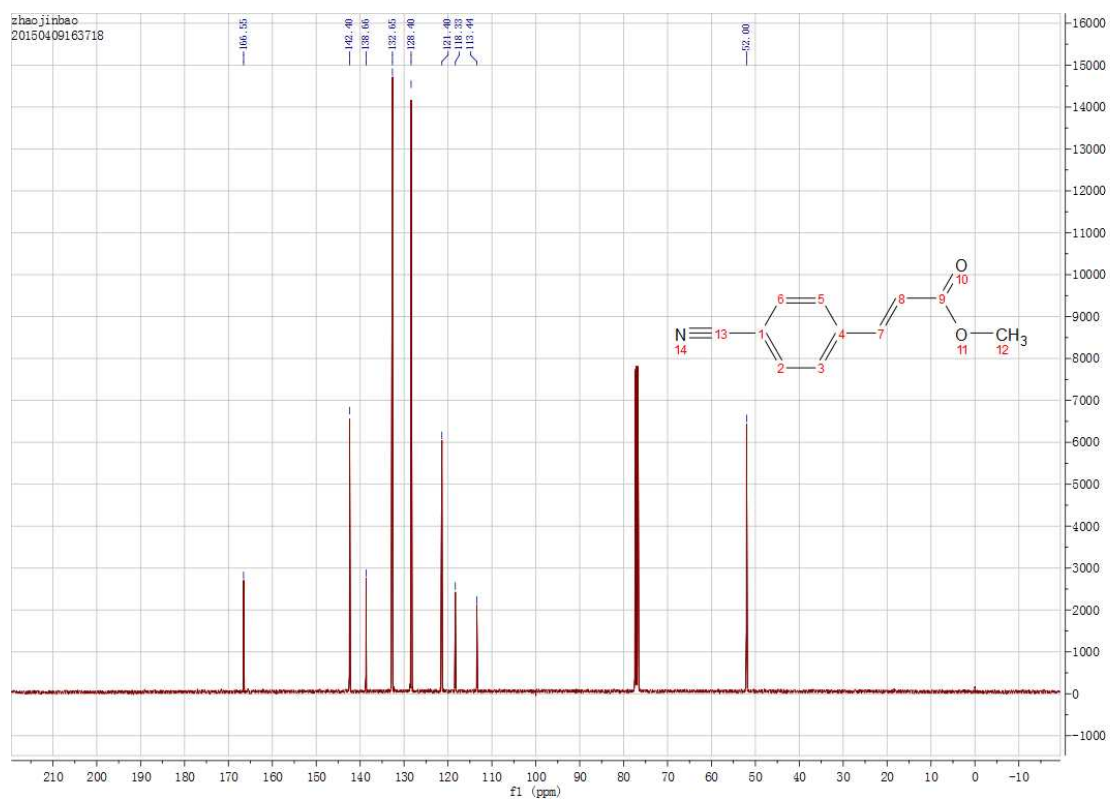
^{13}C NMR (101 MHz, CDCl_3) δ 141.85, 136.31, 132.47 (d, $J = 7.3$ Hz), 128.88, 128.67, 127.04, 126.60, 119.05, 110.60.



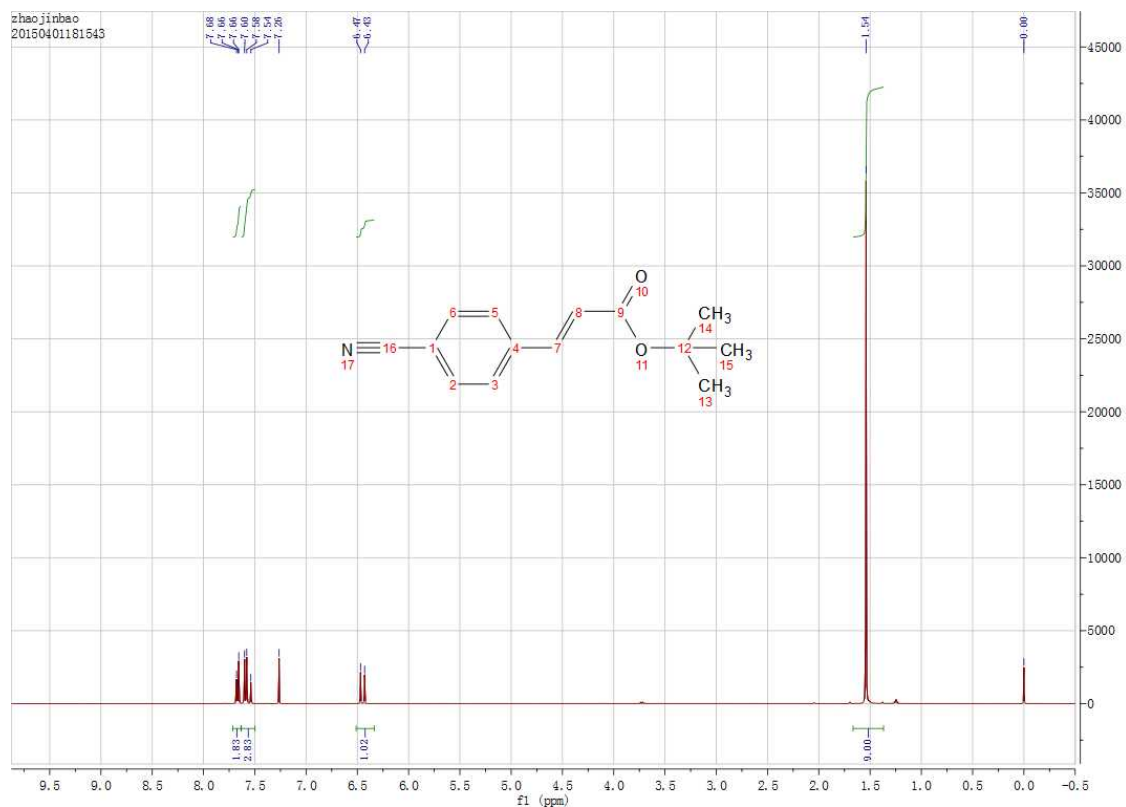
Entry 10. ^1H NMR (400 MHz, CDCl_3) δ 7.68 (dd, $J = 9.1, 7.3$ Hz, 3H), 7.61 (d, $J = 8.3$ Hz, 2H), 6.52 (d, $J = 16.0$ Hz, 1H), 3.83 (s, 3H).



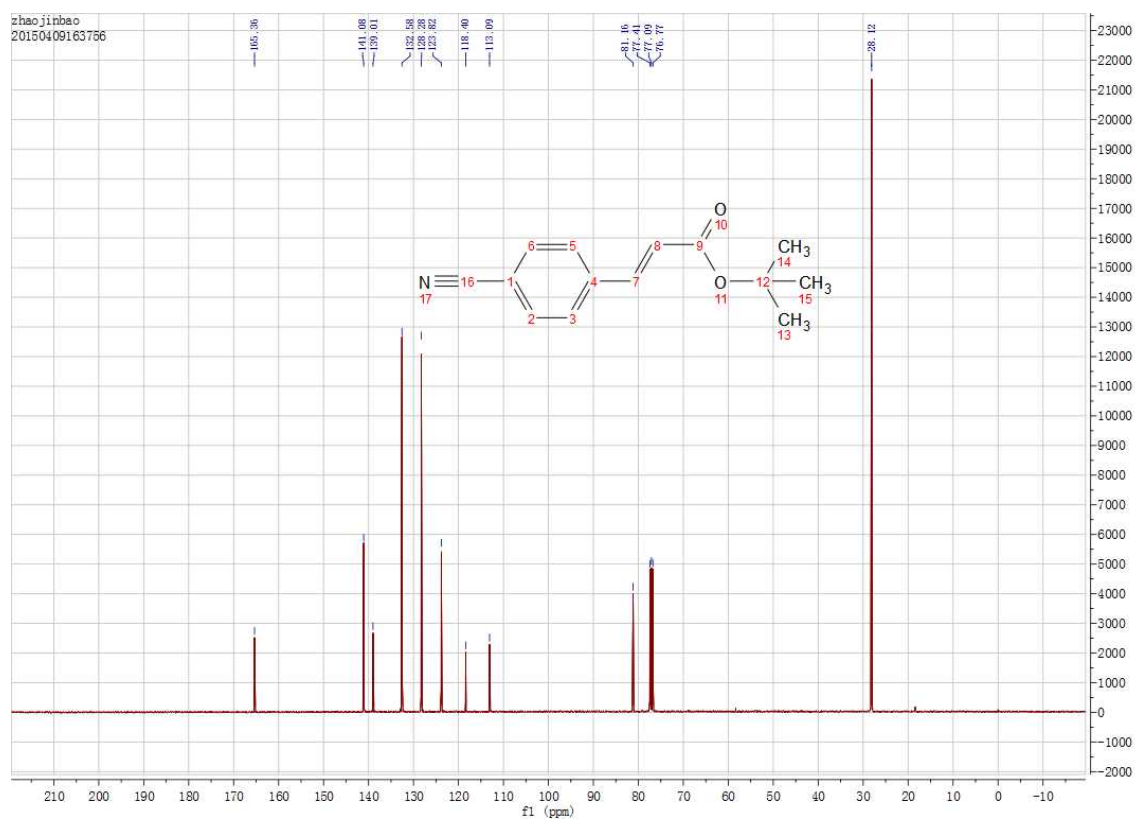
^{13}C NMR (101 MHz, CDCl_3) δ 166.55 , 142.40 , 138.66 , 132.65 , 128.40 , 121.40 , 118.33 , 113.44 , 52.00 .



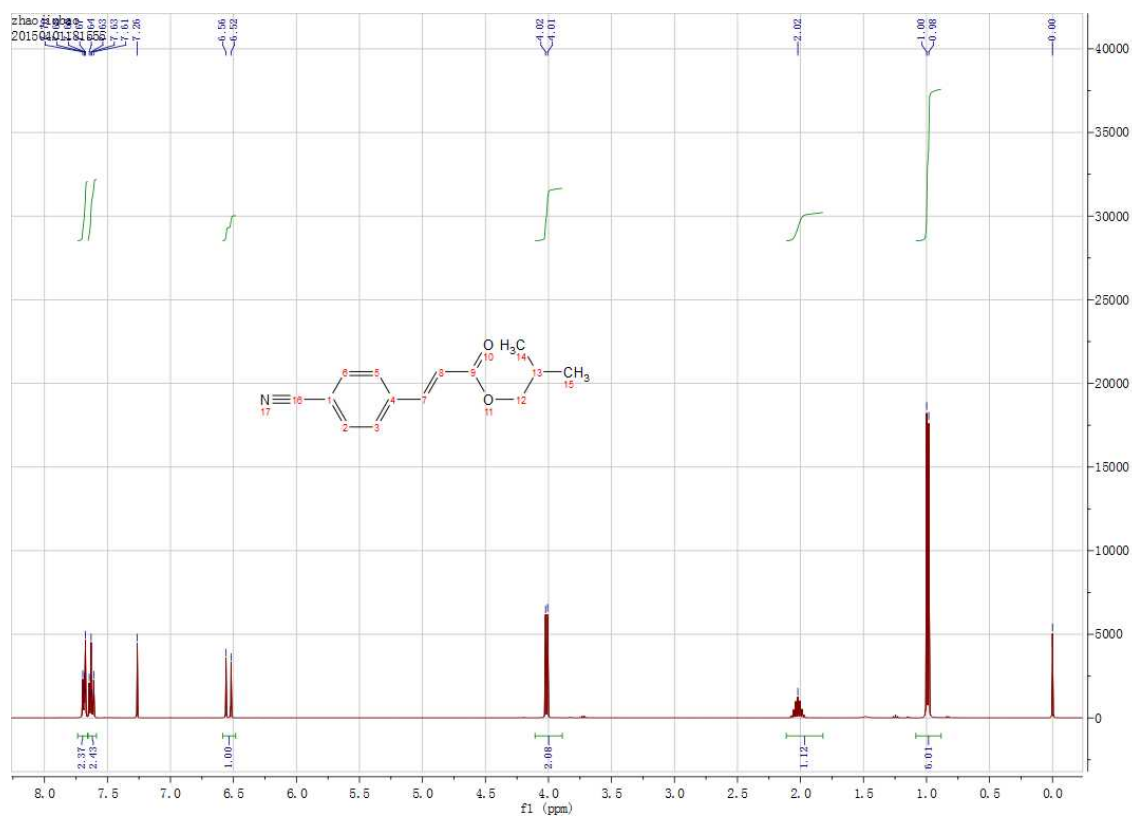
Entry 11. ^1H NMR (400 MHz, CDCl_3) δ 7.71 – 7.64 (m, 2H), 7.63 – 7.50 (m, 3H), 6.45 (d, J = 16.0 Hz, 1H), 1.54 (s, 9H).



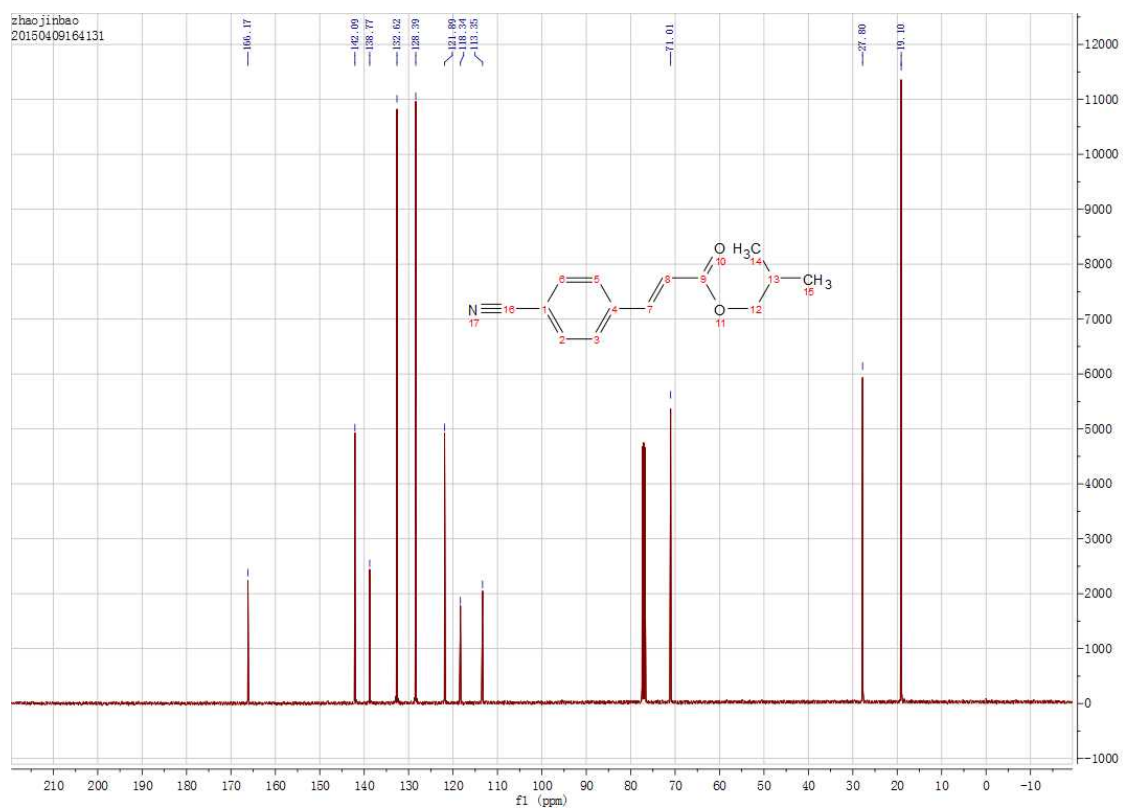
$^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 165.36, 141.08, 139.01, 132.58, 128.28, 123.82, 118.40, 113.09, 81.16, 77.41, 77.09, 76.77, 28.12.



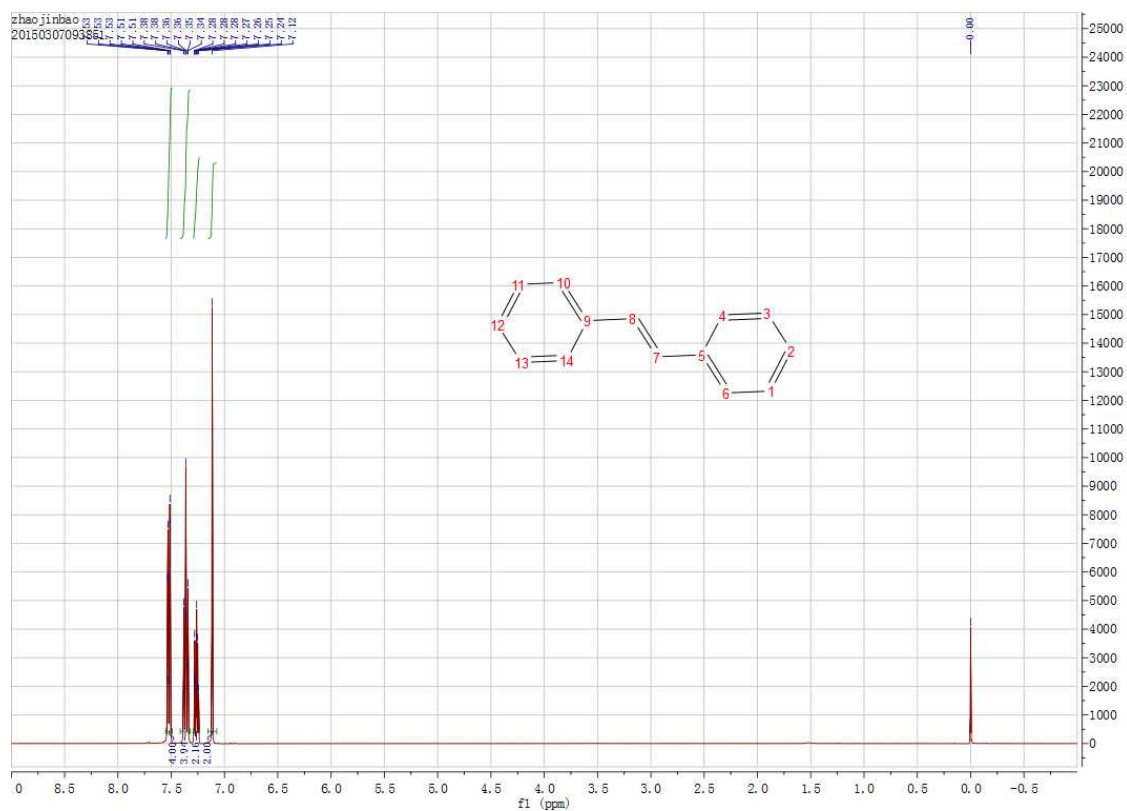
Entry 12. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.68 (dd, $J = 5.2, 3.2$ Hz, 2H), 7.63 (dd, $J = 8.0, 6.2$ Hz, 2H), 6.54 (d, $J = 16.0$ Hz, 1H), 4.01 (d, $J = 6.7$ Hz, 2H), 2.02 (s, 1H), 0.99 (d, $J = 6.7$ Hz, 6H).



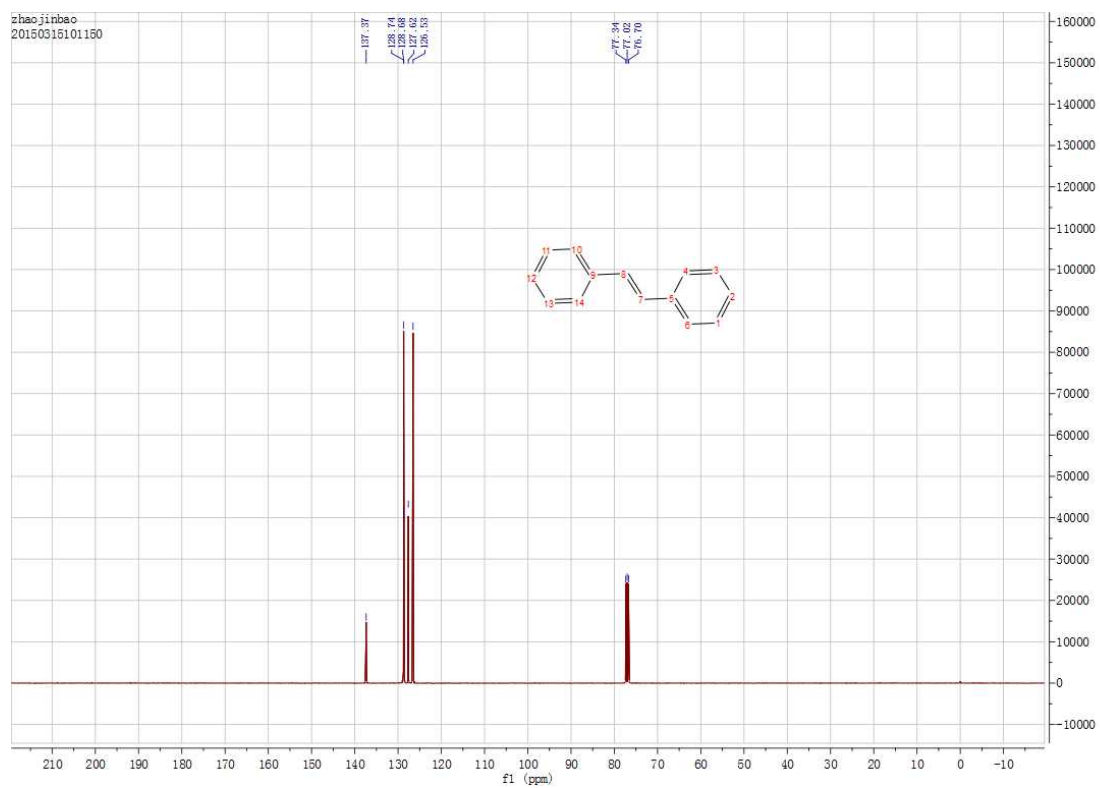
$^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 166.17, 142.09, 138.77, 132.62, 128.39, 121.89, 118.34, 113.35, 71.01, 27.80, 19.10.



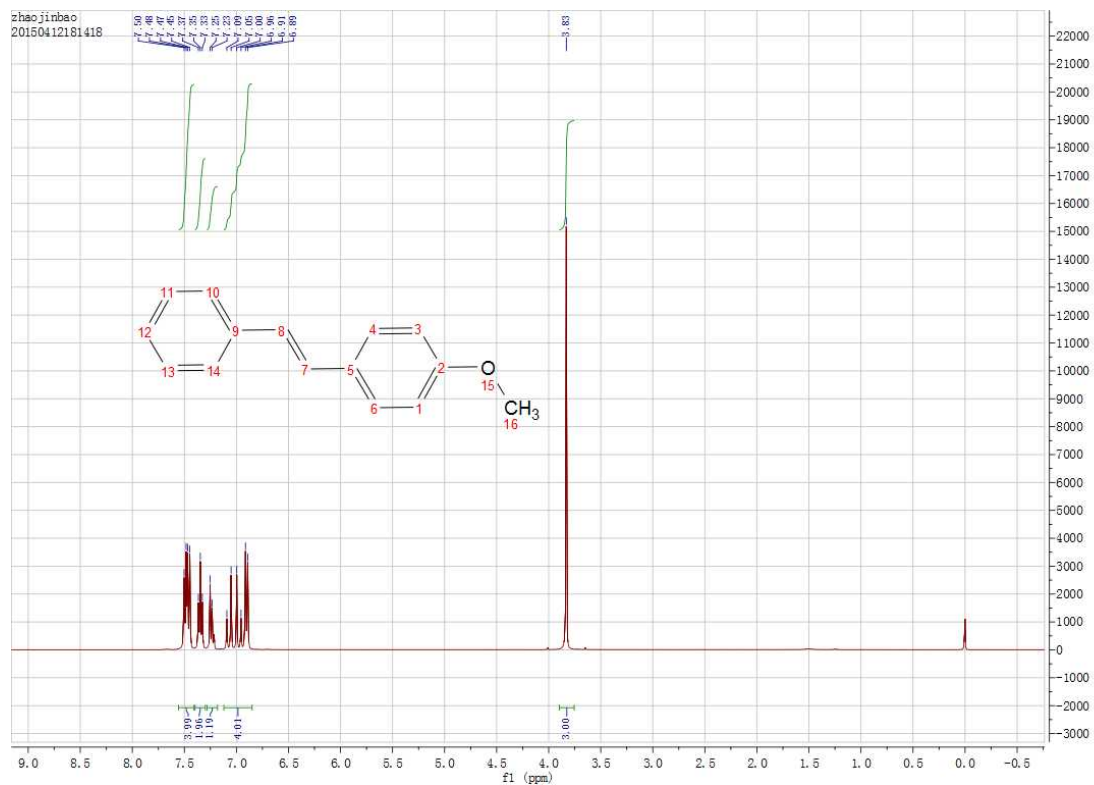
Entry 13. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.55 – 7.49 (m, 4H), 7.42 – 7.32 (m, 4H), 7.29 – 7.24 (m, 2H), 7.12 (s, 2H).



^{13}C NMR (101 MHz, CDCl_3) δ 137.37, 128.71 (d, $J = 5.4$ Hz), 127.62, 126.53, 77.34, 77.02, 76.70.



Entry 14. ^1H NMR (400 MHz, CDCl_3) δ 7.48 (dd, $J = 13.7, 8.2$ Hz, 4H), 7.35 (t, $J = 7.6$ Hz, 2H), 7.24 (d, $J = 7.3$ Hz, 1H), 7.12 – 6.85 (m, 4H), 3.83 (s, 3H).



^{13}C NMR (101 MHz, CDCl_3) δ 159.37 , 137.71 , 130.21 , 128.66 , 128.27 , 127.75 , 127.23 , 126.68 , 126.29 , 114.19 , 55.33 .

