

Electronic Supplementary Information (ESI) for

**Interconnected Mesoporous NiO Sheets Deposited onto TiO<sub>2</sub>  
Nanosheet Arrays as Binder-Free Anode Materials with  
Enhanced Performance for Lithium Ion Batteries**

*Guojian Li, Hao Hu, Qiancheng Zhu and Ying Yu\**

Institute of Nanoscience and Nanotechnology, College of Physical Science and  
Technology, Central China Normal University, 430079, China.

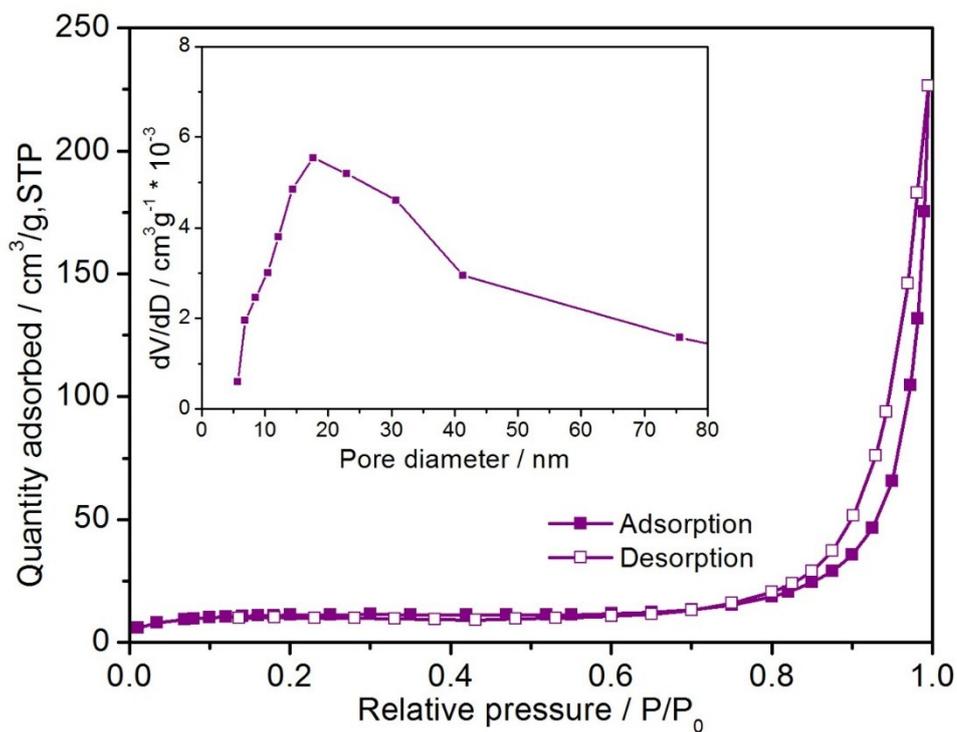
\* Corresponding author. Tex: 86-27-67867037. E-mail: [yuying01@mail.ccnu.edu.cn](mailto:yuying01@mail.ccnu.edu.cn).

## Measurement of the Active Material

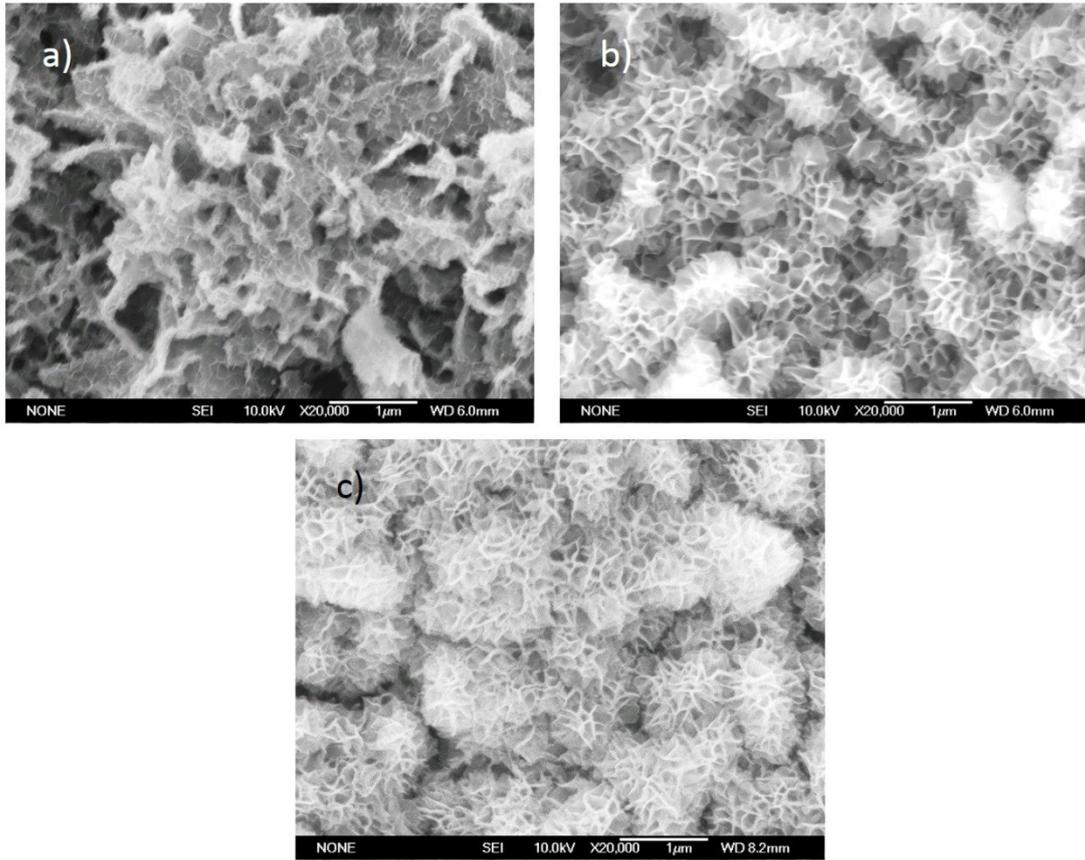
First, the sample of  $\text{TiO}_2@\text{NiO}$  growth on Ti substrate was weighed by an electronic balance, the result marked as  $M_1$ . Then the sample was immersed in dilute hydrochloric acid (1 M) in a beaker for 3 hours at room temperature. After that, ultrasonication for a certain period of time was employed to completely eliminate the active material attached to the substrate. Finally, the substrate was dried, followed by weighing and marking its weigh as  $M_2$ . The weight of the active materials was the difference between  $M_1$  and  $M_2$  ( $\Delta M = M_1 - M_2$ ). Several samples were treated as above at a batch to obtain an average weight for narrowing down the error. Note that metallic titanium is insoluble in cold dilute hydrochloric acid. Hence, the weight of titanium substrate could be regarded as invariance. The weight of the active material for different electrodes are shown in Table S1.

Table S1. Weight of the active material for different electrodes

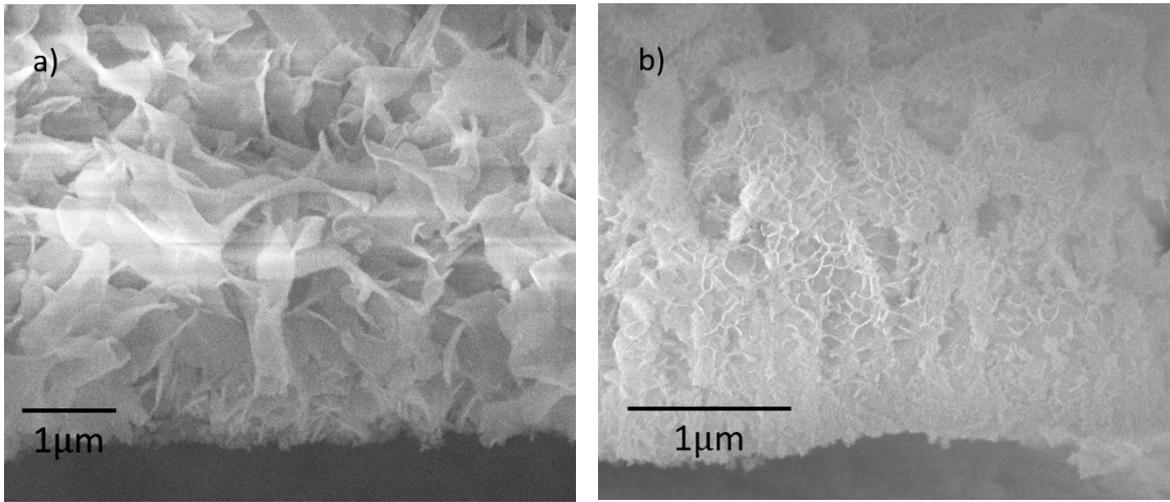
Electrodes	Mass Density ( mg/cm <sup>2</sup> )
TiO <sub>2</sub>	0.1812
TiO <sub>2</sub> +NiO (8 min)	0.2814
TiO <sub>2</sub> +NiO (10 min)	0.3527
TiO <sub>2</sub> +NiO (12 min)	0.4016



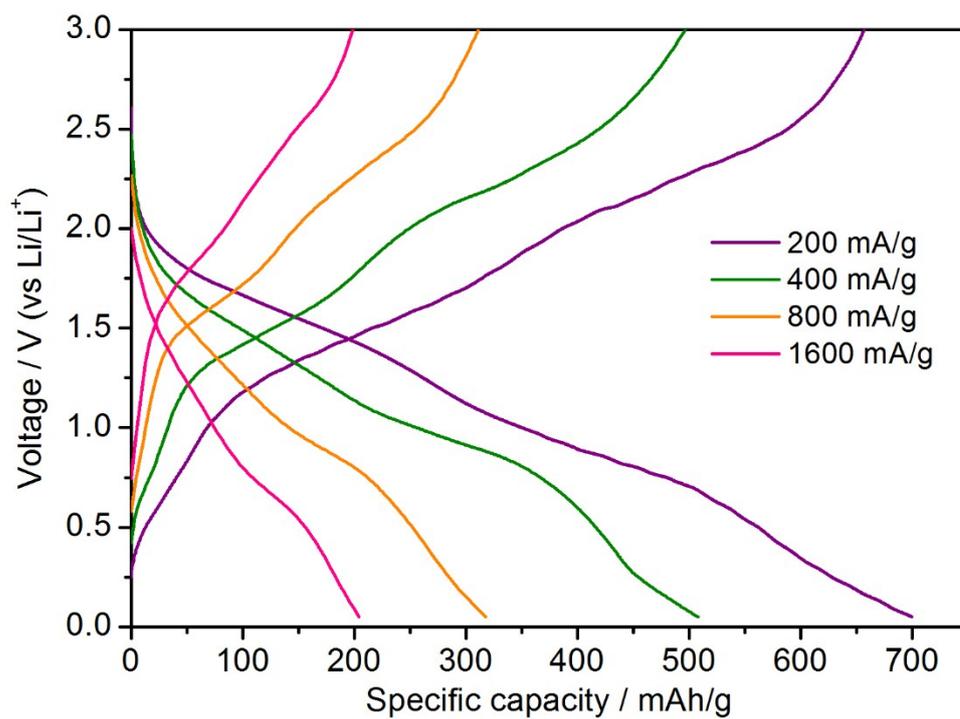
**Fig. S1** Nitrogen adsorption-desorption isotherms for bare TiO<sub>2</sub> backbone with the corresponding BJH pore size distribution in the inset. With calculation, the BET specific surface area for TiO<sub>2</sub> was 39.7 m<sup>2</sup>/g and the average pore width was 16.3 nm.



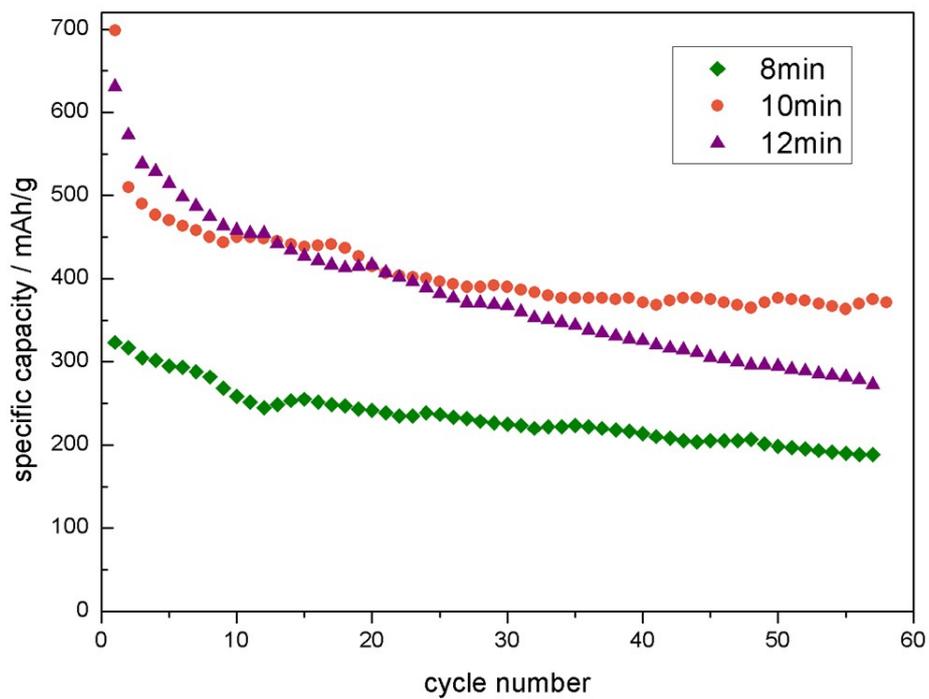
**Fig. S2** Chronological SEM images for  $\text{TiO}_2@\text{NiO}$  samples with different NiO deposition time of (a) 8 min, (b) 10 min and (c) 12 min.



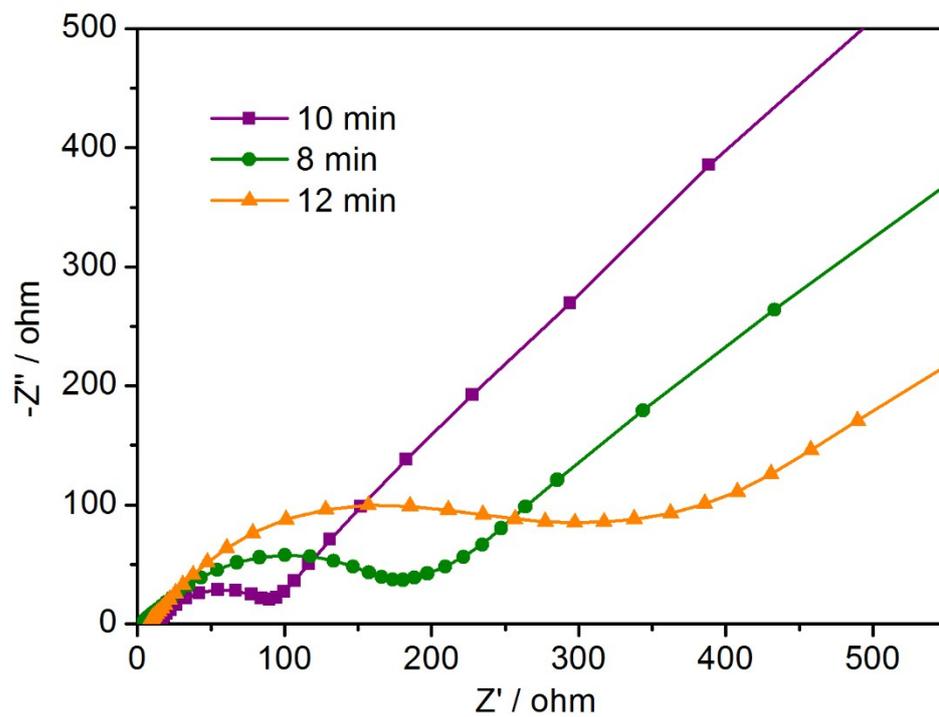
**Fig. S3** Cross sectional SEM images of (a) TiO<sub>2</sub> arrays and (b) TiO<sub>2</sub>@NiO arrays.



**Fig. S4** Charge/discharge curves for hybrid  $\text{TiO}_2@\text{NiO}$  electrode under different current densities.



**Fig. S5** Galvanostatic charge/discharge measurement for  $\text{TiO}_2@\text{NiO}$  electrodes with NiO deposited for 8, 10 and 12 min at 200 mA/g. The capacity mentioned here refers to the discharge capacity.



**Fig. S6** Nyquist plots of  $\text{TiO}_2@\text{NiO}$  electrodes with NiO deposited for 8, 10 and 12 min, scanning range from 100 kHz to 10 mHz.