

Electronic Supplementary Information (ESI) for:

ACdCO<sub>3</sub>F (A = K and Rb): new noncentrosymmetric materials with  
remarkably strong second-harmonic generation (SHG) responses  
enhanced via  $\pi$ -interaction

Guohong Zou,<sup>a</sup> Gnu Nam,<sup>b</sup> Hyung Gu Kim,<sup>a</sup> Hongil Jo,<sup>a</sup> Tae-Soo You,<sup>b,\*</sup> and Kang Min Ok<sup>a,\*</sup>

<sup>a</sup>Department of Chemistry, Chung-Ang University, Seoul, 156-756, Republic of Korea

<sup>b</sup>Department of Chemistry, Chungbuk National University, Cheongju, Chungbuk, 361-763, Republic of Korea.

\*Corresponding Author:

Professor Kang Min Ok

*Fax:* +82-2-825-4736; *Tel:* +82-2-820-5197; *E-mail:* kmok@cau.ac.kr

**Table S1.** Atomic coordinates and displacement parameters for KCdCO<sub>3</sub>F.

**Table S2.** Atomic coordinates and displacement parameters for RbCdCO<sub>3</sub>F.

**Table S3.** Selected bond lengths (Å) and angles (deg) for KCdCO<sub>3</sub>F.

**Table S4.** Selected bond lengths (Å) and angles (deg) for RbCdCO<sub>3</sub>F.

**Fig. S1** Powder X-ray diffraction patterns for the thermal decomposition products of KCdCO<sub>3</sub>F (red line) and RbCdCO<sub>3</sub>F (green line).

**Fig. S2** IR spectra of KCdCO<sub>3</sub>F and KCdCO<sub>3</sub>F

**Fig. S3** Selected COHP curves of KCdCO<sub>3</sub>F representing (a) the C-O interaction in the [CO<sub>3</sub>]<sup>2-</sup> unit, (b) the Cd-C and (c) the Cd-O interactions on the [Cd(CO<sub>3</sub>)] layer. EF is indicated as the vertical dashed line and the energy reference (0 eV).

**Table S1.** Atomic coordinates and displacement parameters for KCdCO<sub>3</sub>F.

<i>atom</i>	<i>x</i>	<i>y</i>	<i>z</i>	<i>U<sub>iso</sub></i>
Cd(1)	0	0	0.5	0.01706
K(1)	0.6667	0.3333	0	0.01433
C(1)	0.3333	0.6667	0.5	0.01974
F(1)	0	0	0	0.01574
O(1)	0.4743(3)	0.5257(3)	0.5	0.02088

**Table S2.** Atomic coordinates and displacement parameters for RbCdCO<sub>3</sub>F.

<i>atom</i>	<i>x</i>	<i>y</i>	<i>z</i>	<i>U<sub>iso</sub></i>
Cd(1)	0	0	0.5	0.01104
K(1)	0.6667	0.3333	0	0.01372
C(1)	0.3333	0.6667	0.5	0.01115
F(1)	0	0	0	0.02637
O(1)	0.4724(4)	0.5276(4)	0.5	0.02522

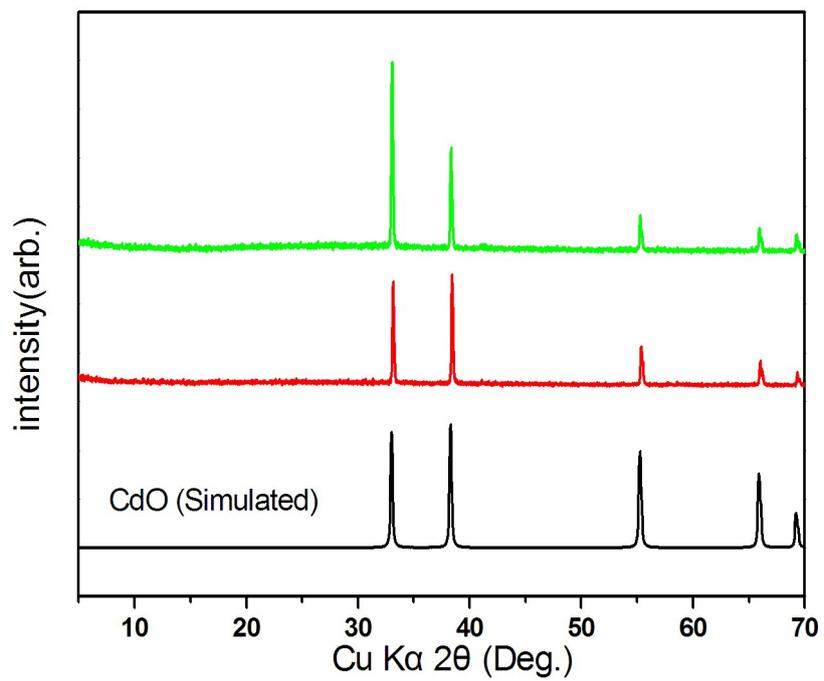
**Table S3.** Selected bond lengths (Å) and angles (deg) for KCdCO<sub>3</sub>F.

Cd1—F1	2.2139(9)	K1—F1	2.9612(12)
Cd1—O1	2.5742(10)	K1—O1	2.7967(9)
C1—O1	1.2524(5)	Cd1—K1—O1	90.87(3)
Cd1—K1—F1	113.609(5)	F1—K1—F1	120
K1—F1—K1	120	O1—C1—O1	120.00(2)

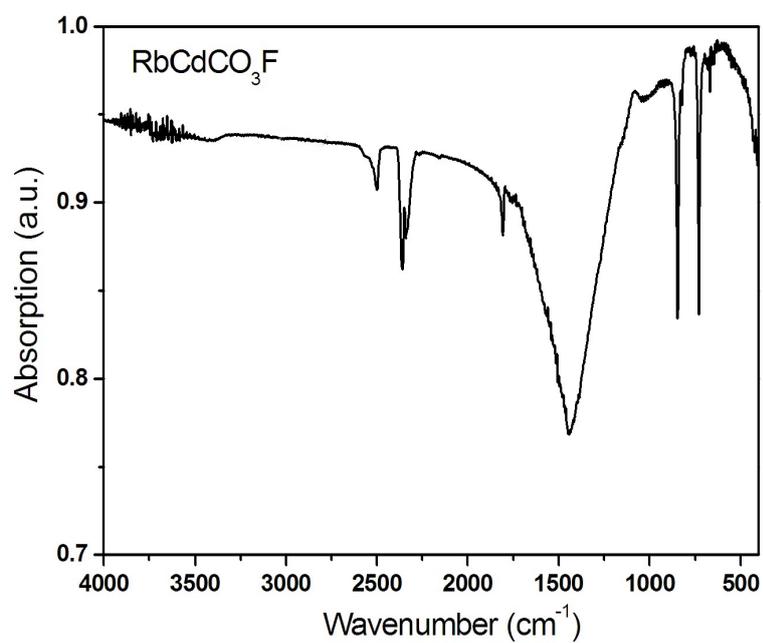
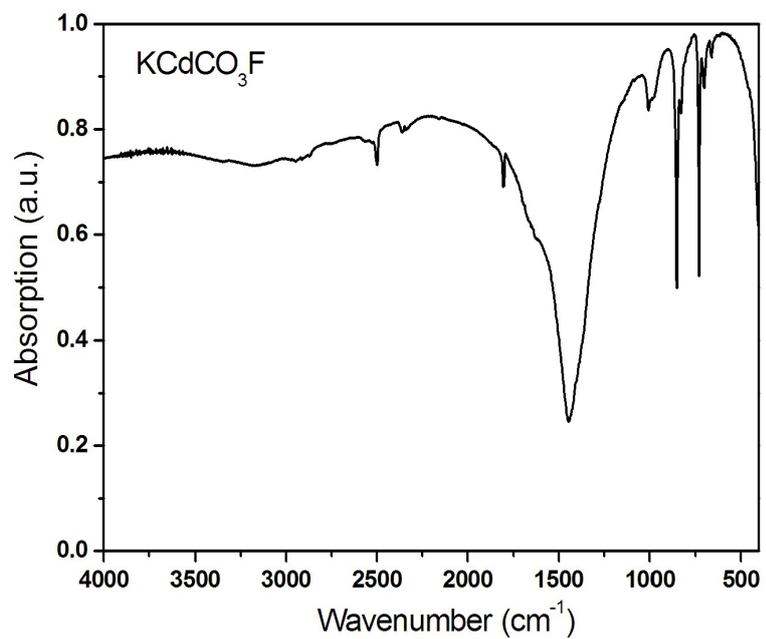
**Table S4.** Selected bond lengths (Å) and angles (deg) for RbCdCO<sub>3</sub>F.

Cd1—F1	2.2646(9)	Rb1—F1	3.0082(12)
Cd1—O1	2.6167(10)	Rb1—O1	2.8641(8)
C1—O1	1.2549(5)	Cd1—Rb1—O1	90.767(3)
Cd1—Rb1—F1	113.548(5)	F1—Rb1—F1	120
Rb1—F1—Rb1	120	O1—C1—O1	120.00(2)

**Fig. S1.** Powder X-ray diffraction patterns for the thermal decomposition products of  $\text{KCdCO}_3\text{F}$  (red line) and  $\text{RbCdCO}_3\text{F}$  (green line).



**Fig. S2.** IR spectra of  $\text{KCdCO}_3\text{F}$  and  $\text{RbCdCO}_3\text{F}$



**Fig. S3** Selected COHP curves of  $\text{KCdCO}_3\text{F}$  representing (a) the C-O interaction in the  $[\text{CO}_3]^{2-}$  unit, (b) the Cd-C and (c) the Cd-O interactions on the  $[\text{Cd}(\text{CO}_3)]$  layer.  $E_F$  is indicated as the vertical dashed line and the energy reference (0 eV).

