

Effect of phenol on the synthesis of benzoxazine

Qin Zhang, Po Yang*, Yuyuan Deng, Chengxi Zhang, Rongqi Zhu and Yi Gu*

*State Key Laboratory of Polymer Materials Engineering (Sichuan University),
College of Polymer Science and Engineering, Sichuan University, Chengdu 610065,
China*

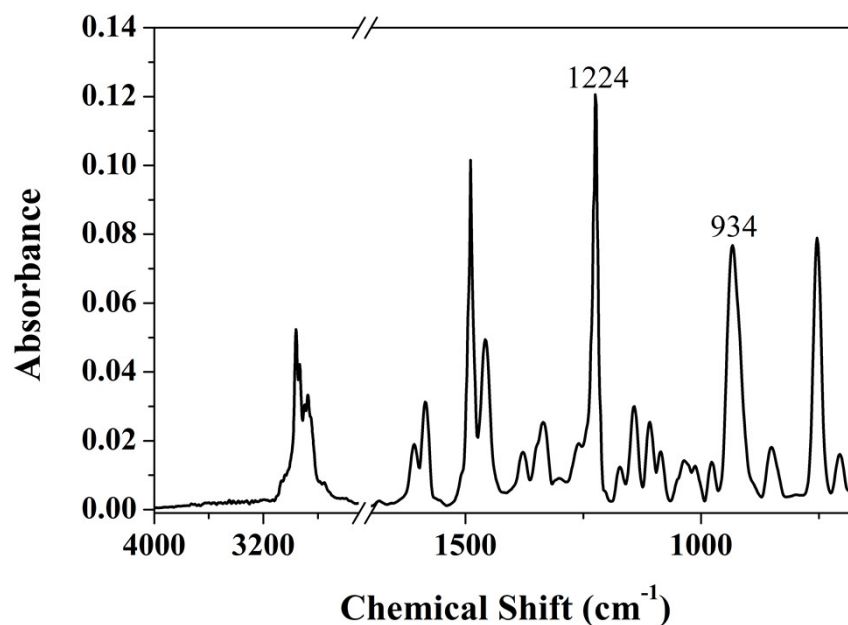


Fig. S1 FTIR spectrum of 3,4-dihydro-2H-3-propyl-1,3-benzoxazine.

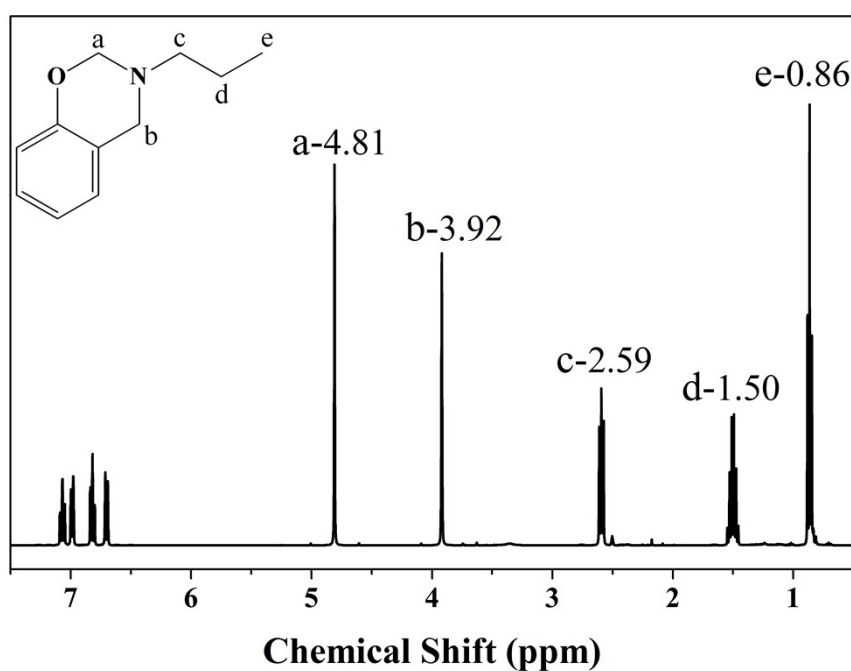


Fig. S2 ¹H NMR spectrum of 3,4-dihydro-2H-3-propyl-1,3-benzoxazine.

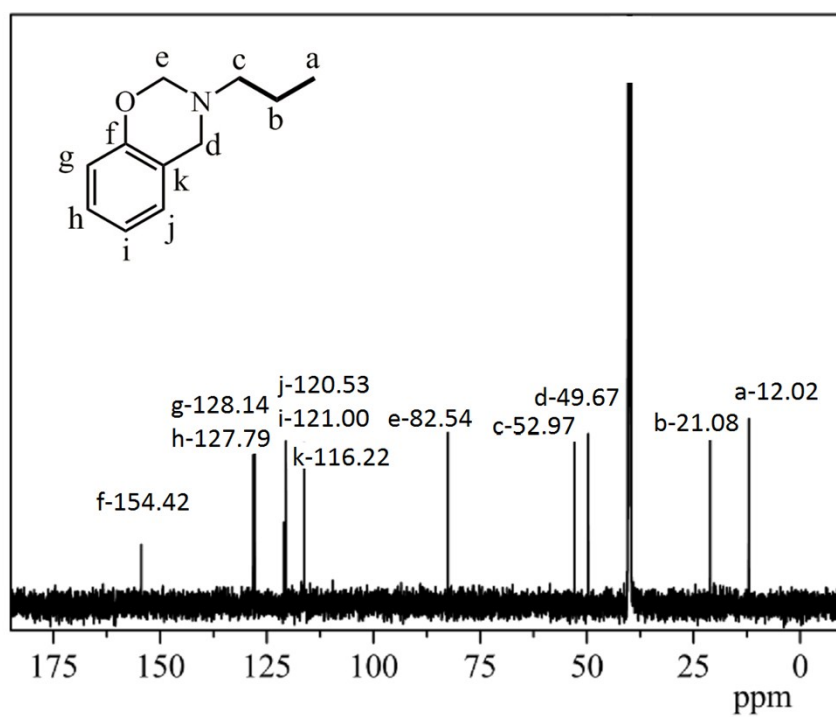
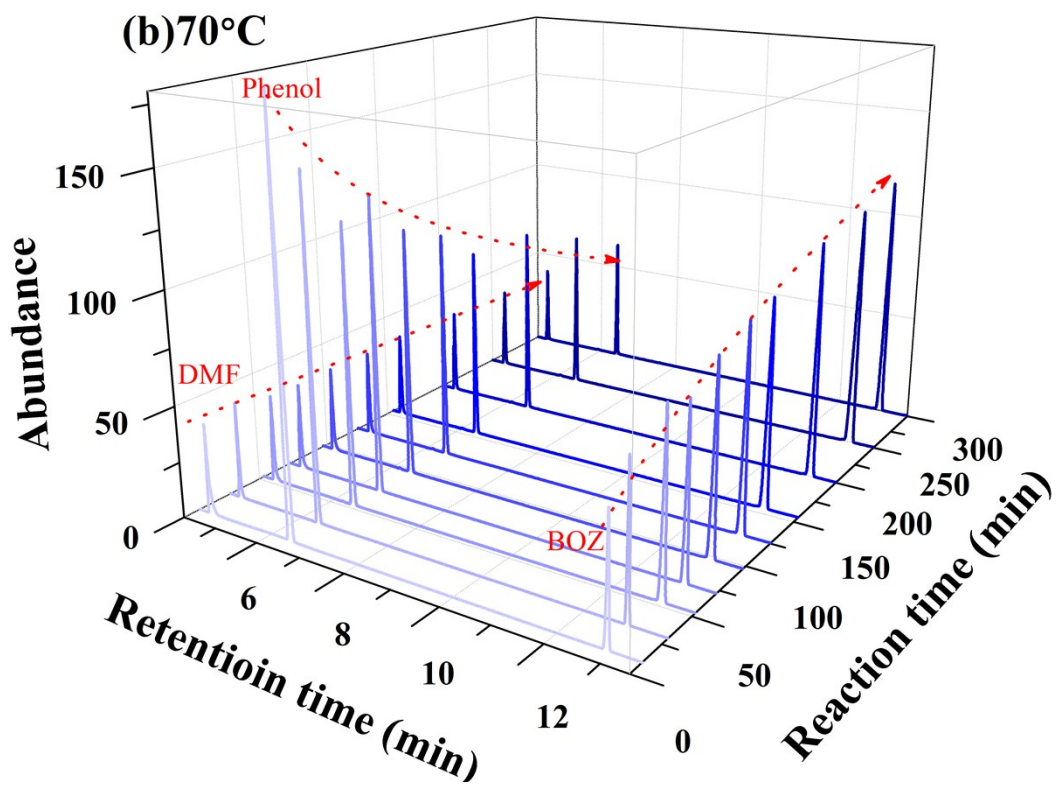
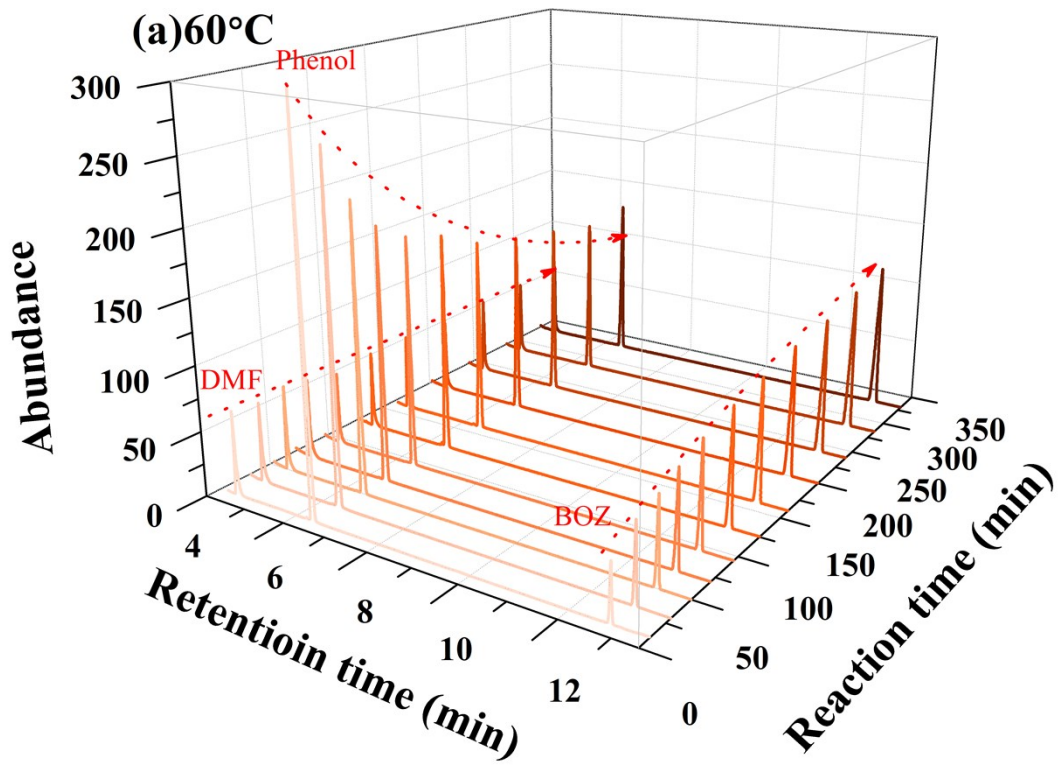


Fig. S3 ¹³C NMR spectrum of 3,4-dihydro-2H-3-propyl-1,3-benzoxazine.



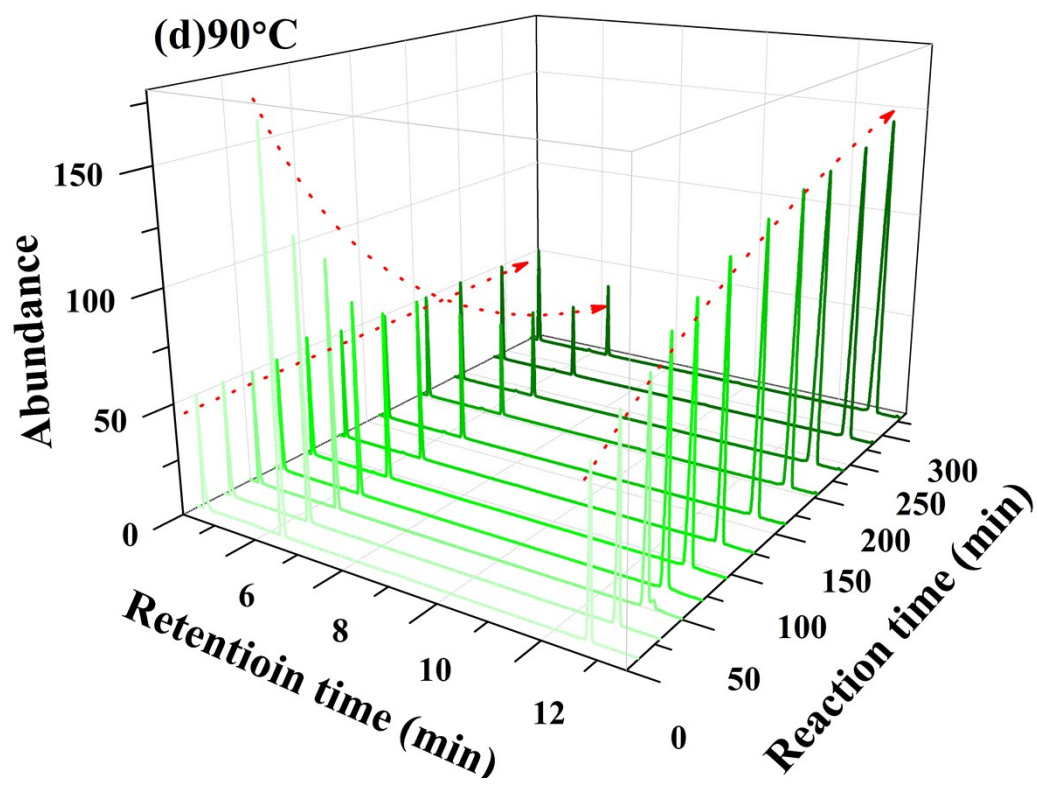
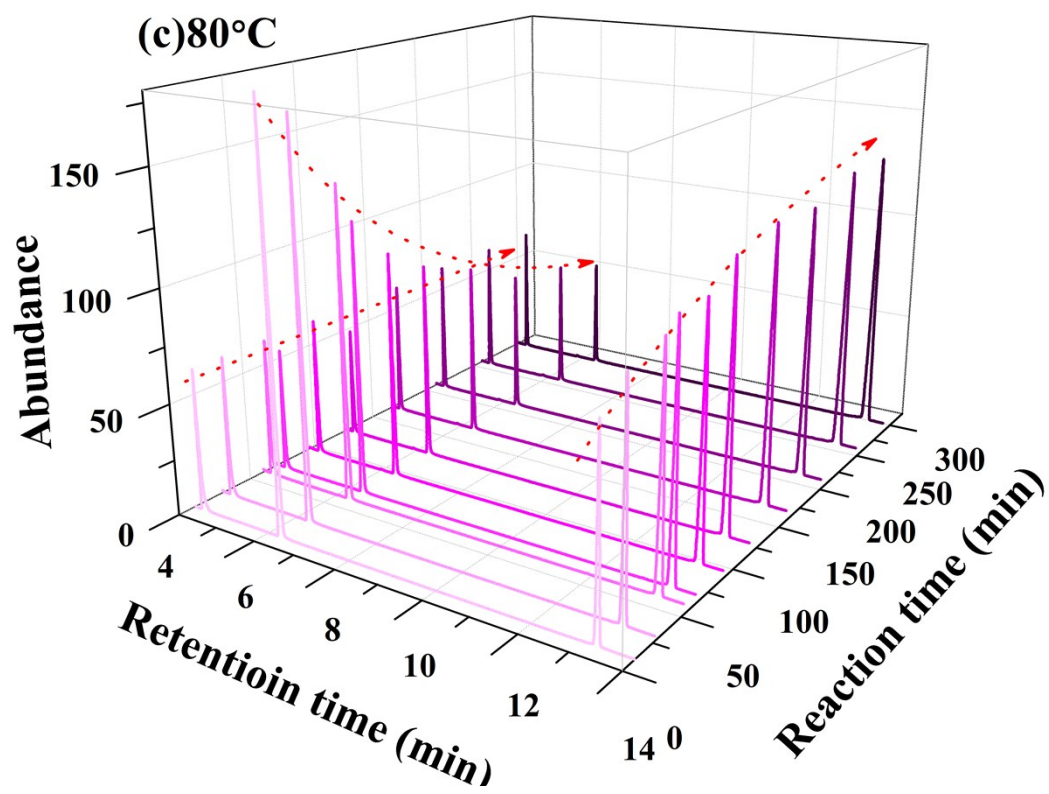


Fig. S4 GC spectra of the reaction of n-propylamine, phenol and aqueous formaldehyde with different time at different temperatures.

(a) 60 °C, (b) 70 °C, (c) 80 °C, (d) 90 °C.

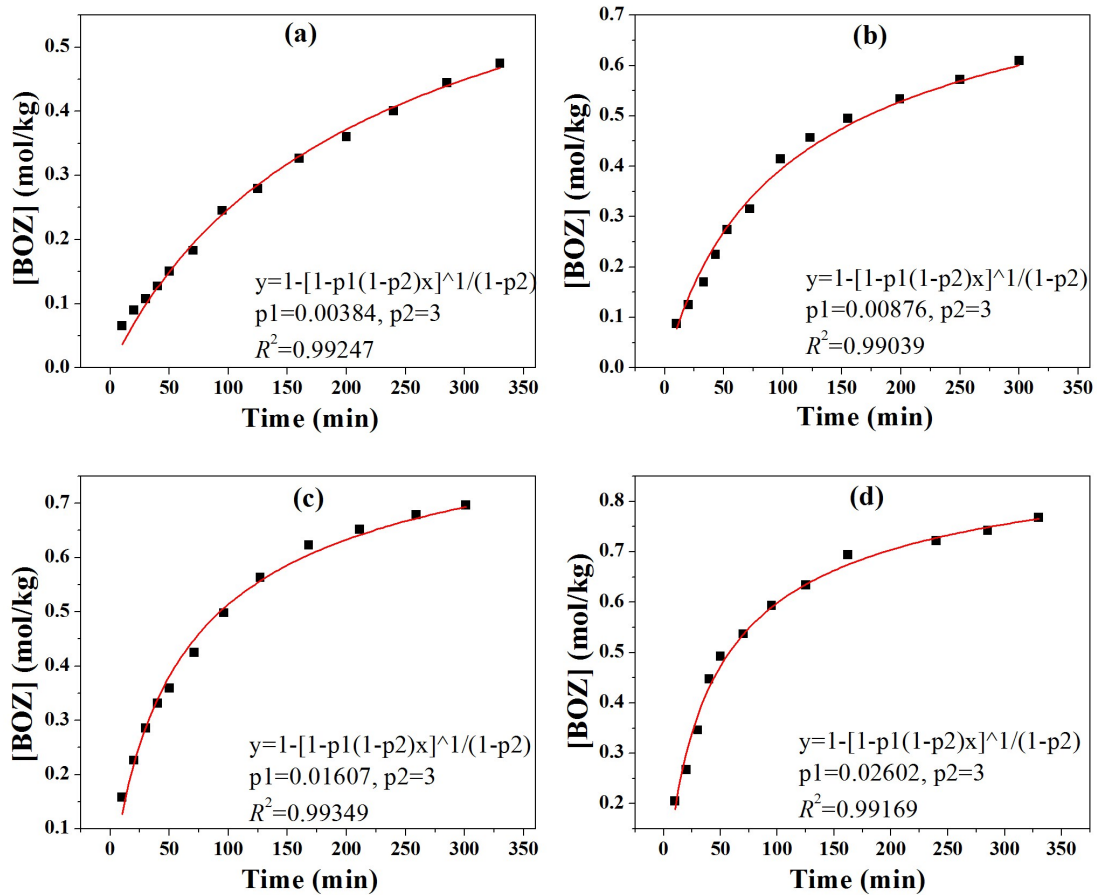


Fig. S5 Concentrations of benzoxazine versus reaction time at (a) 60 °C, (b) 70 °C, (c) 80 °C, (d) 90 °C.

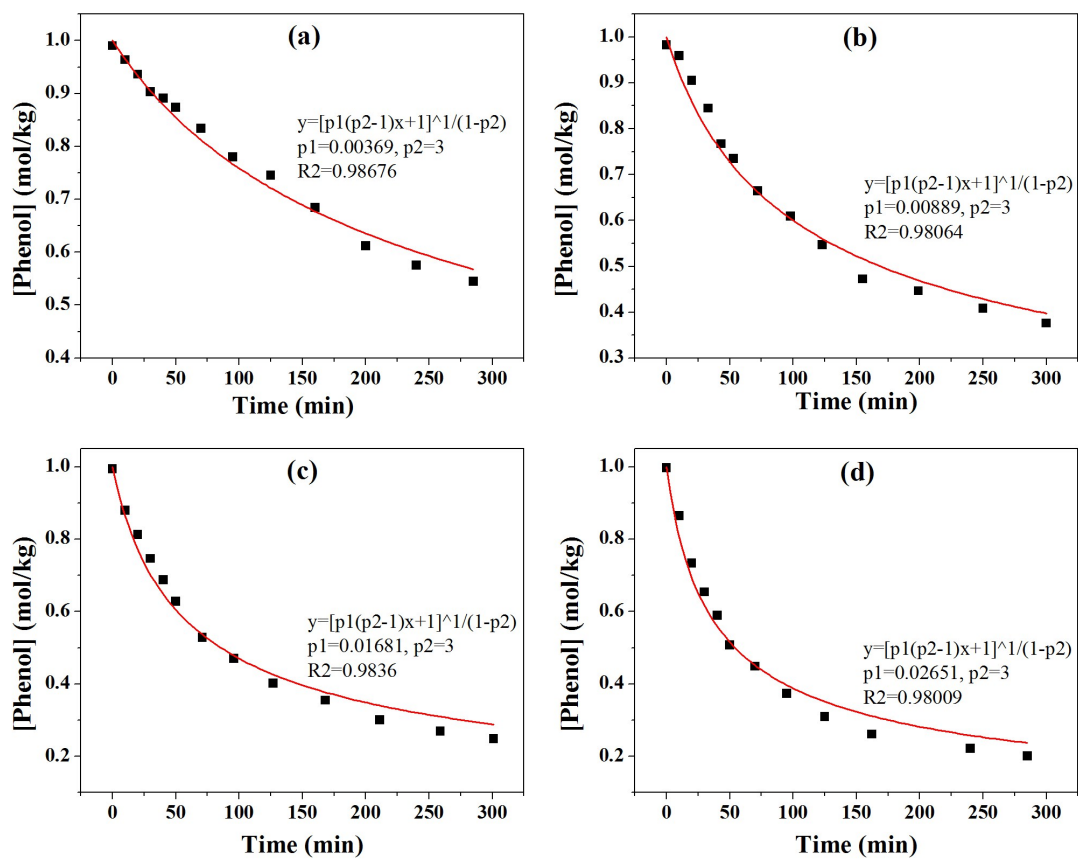


Fig. S6 Concentrations of phenol versus reaction time at (a) 60 °C, (b) 70 °C, (c) 80 °C, (d) 90 °C.