

Electronic Supplementary Information

Electrodeposition of ultrathin nickel-cobalt double hydroxide nanosheets on nickel foam as high-performance supercapacitor electrodes

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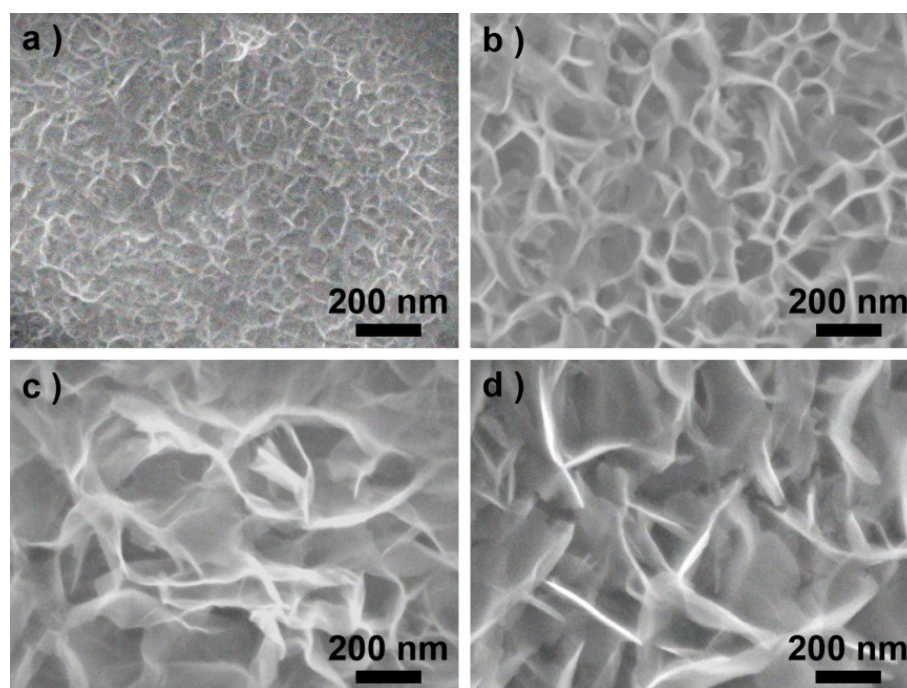


Figure S1 SEM images of electrodeposited Ni-Co DH nanosheets in solution with different Ni/Co molar ratios: (a) 1/0, (b) 2/1 (c) 1/2, and (d) 0/1.

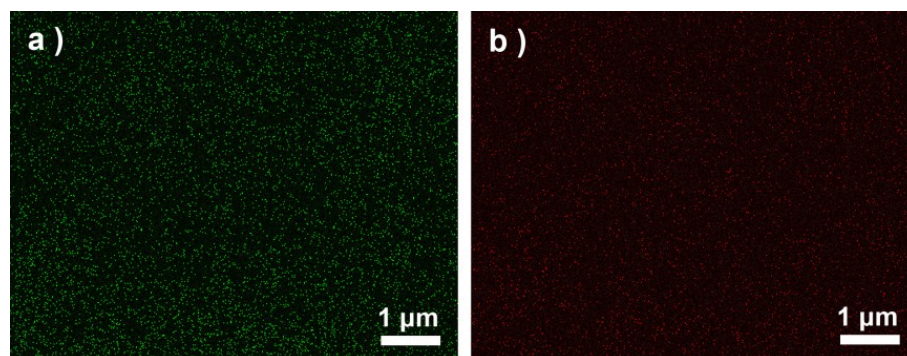


Figure S2 EDS mapping images of Ni-Co DH nanosheets obtained in solution of Ni/Co (1/1): (a) Ni, (b) Co. The green color is Ni and the red color indicates Co.

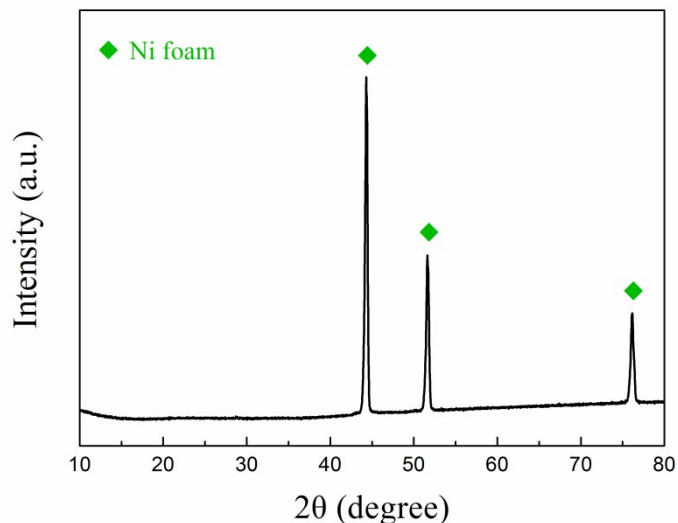


Figure S3 XRD patterns of Ni foam supported Ni-Co DHs obtained in solution of Ni/Co (1/1).

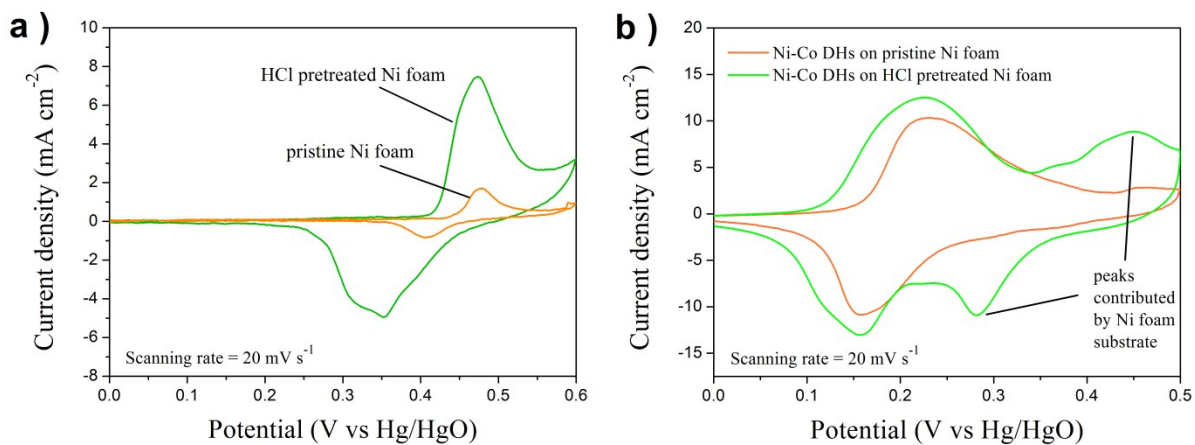
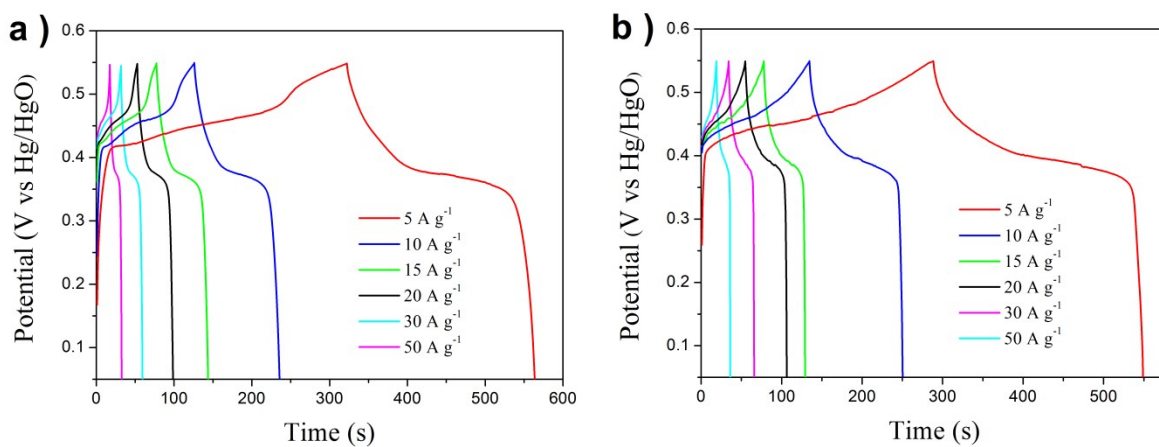


Figure S4 Comparison of CV curves of (a) pristine Ni foam and HCl pretreated Ni foam and (b) Ni-Co DHs deposited in the solution of Ni/Co (1/1) on Ni foam substrates without and with HCl pretreatment.



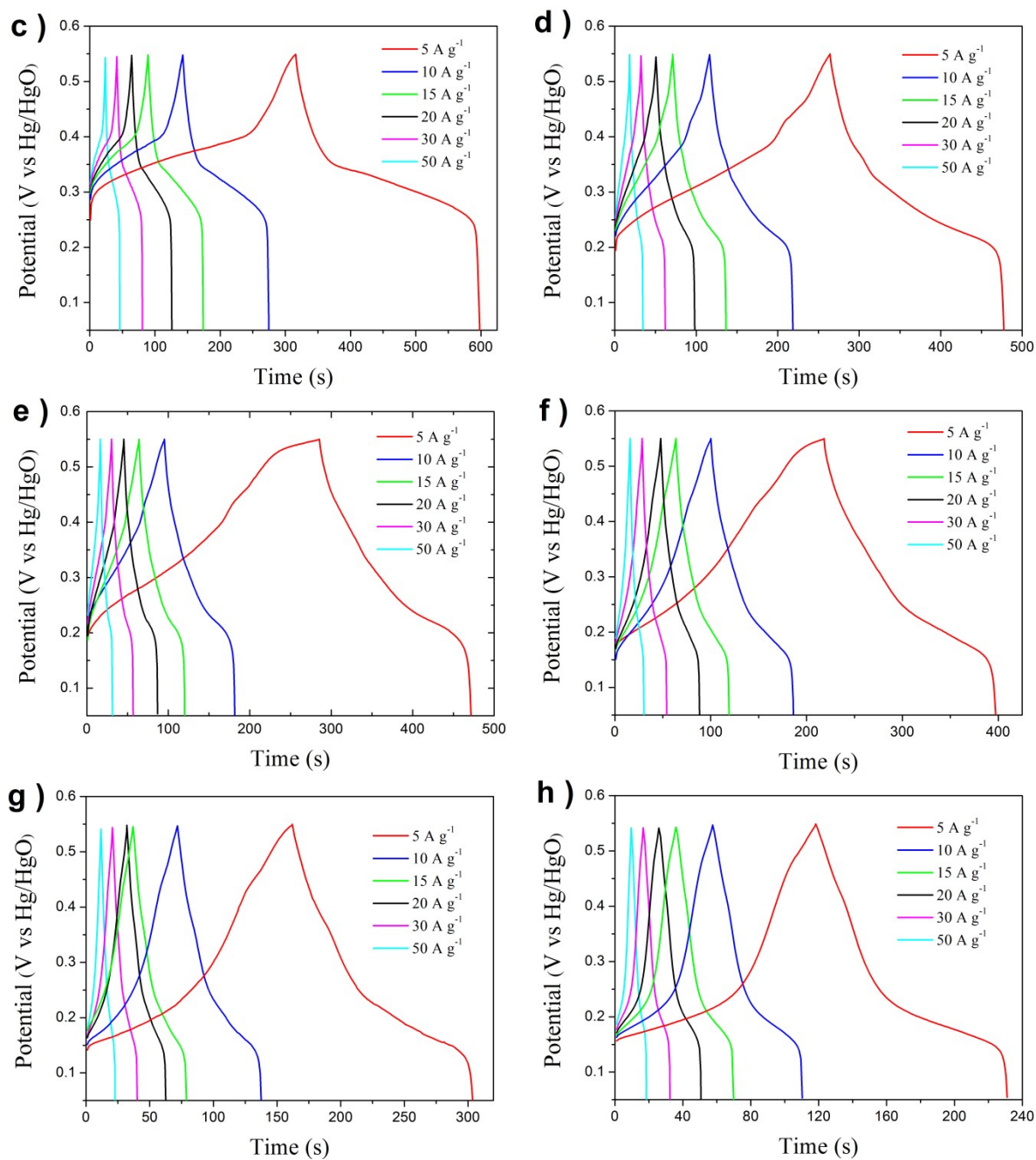


Figure S5 Galvanostatic charge/discharge curves at different current density of Ni foam supported Ni-Co DHs formed with different Ni/Co feeding molar ratios: (a) 1/0, (b) 9/1, (c) 4/1, (d) 3/2, (e) 1/1, (f) 1/2, (g) 1/3, and (h) 0/1.

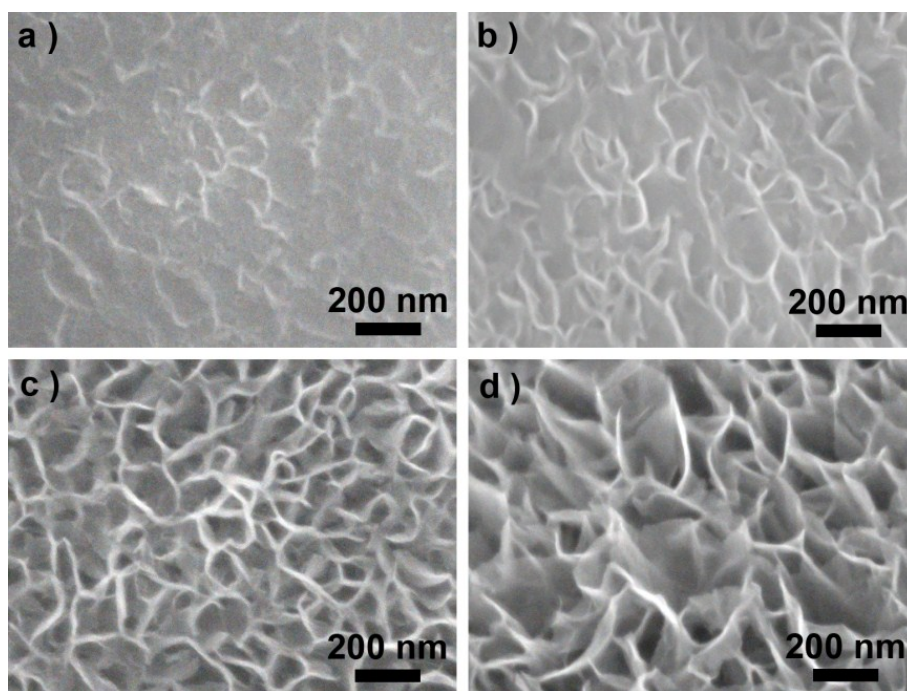


Figure S6 SEM images of Ni-Co DHs after 2000 cycles of charge/discharge tests: (a) Ni/Co (1/0), (b) Ni/Co (4/1), (c) Ni/Co (1/1), (d) Ni/Co (0/1).

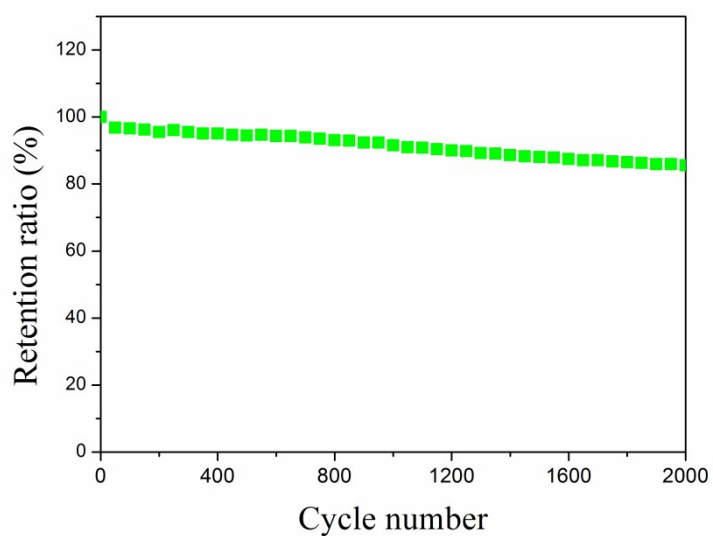


Figure S7 Cycle performance of Ni-Co DHs formed in solution of Ni/Co (1/1) without immersing pretreatment.

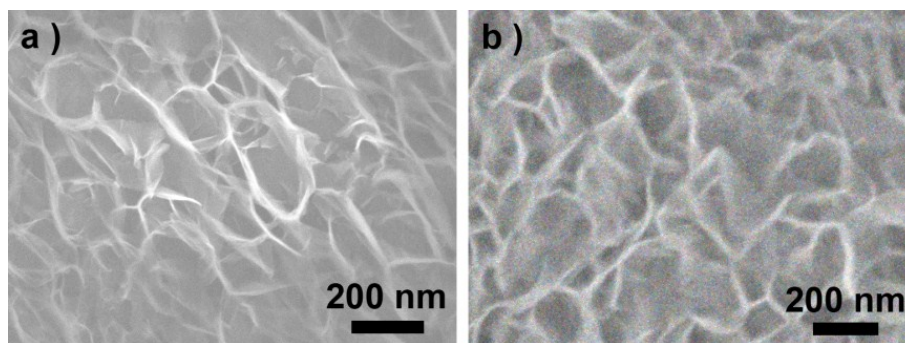


Figure S8 SEM images of Ni-Co DHs formed in solution of Ni/Co (1/1) without pre-immersing process (a) before and (b) after 2000 cycles of galvanostatic charge-discharge tests.

Table S1 Comparison of maximum (at low current density) and minimum (at high current density) C_s , cycle stability, and maximum (at low power density) and minimum (at high power density) energy density based on active materials of some reported supercapacitor electrodes of Ni-based oxides/hydroxides

Ref.	Electrode materials	C_s (F g ⁻¹)		Stability*	Energy density (Wh kg ⁻¹)	
		Maximum	Minimum		Maximum	Minimum
1	Ni(OH) ₂	3152 (4 A g ⁻¹)	280 (16 A g ⁻¹)	48% (after 300 cycles)	--	--
2	NiO	309 (1 A g ⁻¹)	221 (40 A g ⁻¹)	~ 91%**	--	--
3	Ni-Co LDHs***	2184 (1 A g ⁻¹)	1494 (20 A g ⁻¹)	88.5%	91.76 (0.826 kW kg ⁻¹)	~ 62 (~ 11 kW kg ⁻¹)
4	Ni-Co DH microspheres	2275.5 (1 A g ⁻¹)	1007.8 (25 A g ⁻¹)	~ 95%	--	--
5	Ni-Co LDHs/ZnO nanoflake	1624 (10 A g ⁻¹)	1311 (50 A g ⁻¹)	94 %	68.23 (2.75 kW kg ⁻¹)	48.32 (27.53 kW kg ⁻¹)
6	NiCo(OH) ₂ /graphene/carbon nanotube	2360 (0.5 A g ⁻¹)	2030 (20 A g ⁻¹)	~ 81%	--	--
7	NiCo ₂ O ₄ nanowire/Ni-Co DHs	--	--	~ 69%	~ 96 (~ 1 kW kg ⁻¹)	58.4 (41.3 kW kg ⁻¹)
8	Ni-Co LDHs	2682 (3 A g ⁻¹)	1706 (20 A g ⁻¹)	--	77.3 (0.623 kW kg ⁻¹)	~ 40 (~ 1.3 kW kg ⁻¹)
9	Ni-Mn LDHs	881 (1 A g ⁻¹)	403 (10 A g ⁻¹)	88% (after 500 cycles)	--	--

10	Ni-Al LDHs/graphene	1255.8 (1 A g ⁻¹)	755.6 (6 A g ⁻¹)	~ 79%****	--	--
11	Ni-Al LDHs/graphene	1329 (3.6 A g ⁻¹)	851 (18 A g ⁻¹)	91% (after 500 cycles)	--	--
12	Ni-Al LDH/carbon nanotube	1500 (1 A g ⁻¹)	1054 (10 A g ⁻¹)	50%	--	--
This wor -k	Ni-Co DHs (electrodeposition)	3028 (2 A g ⁻¹)	2225 (50 A g ⁻¹)	94%	127.22 (0.605 kW kg ⁻¹)	93.48 (15.125 kW kg ⁻¹)

* C_s retention ratios after 2000 cycles galvanostatic charge-discharge tests

** based on the maximum C_s

*** LDHs (layered double hydroxides)

**** after 1500 cycles, based on the maximum C_s

References

1. G. Yang, C. Xu and H. Li, *Chem. Commun.*, 2008, **0**, 6537-6539.
2. X. Xia, J. Tu, X. Wang, C. Gu and X. Zhao, *J. Mater. Chem.*, 2011, **21**, 671-679.
3. X. Zheng, Z. Gu, Q. Hu, B. Geng and X. Zhang, *RSC Adv.*, 2015, **5**, 17007-17013.
4. T. Yan, Z. Li, R. Li, Q. Ning, H. Kong, Y. Niu and J. Liu, *J. Mater. Chem.*, 2012, **22**, 23587-23592.
5. N. T. H. Trang, H. Van Ngoc, N. Lingappan and D. J. Kang, *Nanoscale*, 2014, **6**, 2434-2439.
6. Y. Cheng, H. Zhang, C. V. Varanasi and J. Liu, *Energy Environ. Sci.*, 2013, **6**, 3314-3321.
7. L. Huang, D. Chen, Y. Ding, S. Feng, Z. L. Wang and M. Liu, *Nano Lett.*, 2013, **13**, 3135-3139.
8. H. Chen, L. Hu, M. Chen, Y. Yan and L. Wu, *Adv. Funct. Mater.*, 2014, **24**, 934-942.
9. H. Sim, C. Jo, T. Yu, E. Lim, S. Yoon, J. H. Lee, J. Yoo, J. Lee and B. Lim, *Chem. Eur. J.*, 2014, **20**, 14880-14884.
10. L. Zhang, J. Wang, J. Zhu, X. Zhang, K. San Hui and K. N. Hui, *J. Mater. Chem. A*, 2013, **1**, 9046-9053.
11. J. Xu, S. Gai, F. He, N. Niu, P. Gao, Y. Chen and P. Yang, *J. Mater. Chem. A*, 2014, **2**, 1022-1031.
12. M. Li, F. Liu, J. Cheng, J. Ying and X. Zhang, *J. Alloys Compd.*, 2015, **635**, 225-232.