Electronic Supplementary Information (ESI)

Substitution effect on photochromic properties of benzo[b]thiophene-1,1-dioxide

based diarylethenes

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Scheme S1 Synthetic routes for the photochromic compounds BTT-1, BTT-2, and BTT-4



Fig. S1 Simulated absorption spectra of the photochromic molecules BTT-1 to BTT-4: λ_{max} =250, 458 nm for BTT-1 and BTT-1c; λ_{max} =288, 529 nm for BTT-2 and BTT-2c; λ_{max} =253, 592 nm for BTT-3 and BTT-3c; λ_{max} =378, 611 nm for BTT-4 and BTT-4c.

	НОМО	LUMO
BTT-1		
BTT-1c		
BTT-2	A CONTRACTOR	
BTT-2c		
BTT-3		
BTT-3c		
BTT-4	The state of the s	and the second second
BTT-4c		

Table S1 The frontier orbitals of compounds BTT-1 to BTT-4 (isovalue = 0.04 a.u.)

Compound	open-ring isomer	closed-ring isomer	energy difference
BTT-1	-2118.05608805 a.u.	-2118.04609003 a.u.	–26.2 kJ/mol
BTT-2	-2501.58660643 a.u.	-2501.57554288 a.u.	–29.0 kJ/mol
BTT-3	-2266.06466067 a.u.	-2266.04126384 a.u.	-61.4 kJ/mol
BTT-4	-3691.46746019 a.u.	-3691.46254823 a.u.	–22.9 kJ/mol

Table S2 Computed self-consistent field energies for the photochromic molecules.

Table S3 Crystal data of BTT-1, BTT-2, and BTT-3

	Compound		
	BTT-1	BTT-2	BTT-3
Formula	$C_{20}H_{18}O_2S_3$	$C_{30}H_{22}O_2S_3$	$C_{20}H_{14}O_4S_3$
Formula weight	386.55	510.66	414.49
Temperature	293 К	573 k	166 K
Crystal system	prismatic	orthorhombic	orthorhombic
Space group	P2(1)/n	P212121	C2/c
Unit cell dimension a (Å)	9.2440(11)	9.973(4)	19.5864(18)
b (Å)	21.127(3)	12.164(4)	12.1795(11)
c (Å)	10.5457(12)	21.090(7)	15.8503(15)
α (°)	90.00	90.00	90.00
β (°)	114.609(2)	90.00	93.133(2)
γ (°)	90.00	90.00	90.00
Volume	1871.3(4)	2558.3(16)	3775.5(6)
Z	4	4	8
Density calc. (g/cm ³)	1.372	1.326	1.458
Goodness of fit on p ²	0.973	1.016	1.024
Final R1	0.0596	0.0419(2991)	0.0378(3365)
wR2(I> <i>2S(I)</i>)	0.1395(3462)	0.0906(4187)	0.1039(4295)
R1 (all data)	0.0915	0.0716	0.0527
wR2 (all data)	0.1267	0.0795	0.0953



Fig. S2 ORTEP drawings showing 50% probability displacement ellipsoids: (A) BTT-1, (B) BTT-2, and (c) BTT-3.



Fig. S3 (A) Changes in the absorbance of PMMA films upon alternative irradiation with UV (365 nm) and visible light (>450 nm) light; (A) PMMA-BTT-1, (B) PMMA-BTT-2, (C) PMMA-BTT-3, and (D) PMMA-BTT-4.



Fig. S4 Fluorescence spectra changes upon UV irradiation at 365 nm with the excitation at the isobestic point in ethyl acetate: (A) BTT-2, 2.15×10^{-5} mol L⁻¹ and (B) BTT-3, 2.10×10^{-5} mol L⁻¹

Fig. S5-S16 NMR(¹H, ¹³C) spectra and High resolution mass spectrum of BTT-1 to BTT-4



Fig. S5 ¹H NMR spectra of BTT-1 in CDCl₃ (400 MHz)



Fig. S6 ¹³C NMR spectrum of BTT-1 in CDCl₃ (100 MHz)

Elemental Composition Report

Single Mass Analysis

Tolerance = 2.0 mDa / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 4

Monoisotopic Mass, Even Electron Ions 367 formula(e) evaluated with 4 results within limits (up to 1 best isotopic matches for each mass) Elements Used: C: 20-32 H: 0-1000 O: 0-40 S: 0-3 K: 0-1 ZHU-WH LCT Premier Key Lab for Advanced Materials --- ECUST 1: TOF MS ES+

ZWH-CSJ-03 27 (0.841) Cm (18:31)







Fig. S8 ¹H NMR spectra of BTT-2 in CDCl₃ (400 MHz)

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Elemental Composition Report

Single Mass Analysis

Tolerance = 3.0 mDa / DBE: min = -1.5, max = 50.0 Element prediction: Off Number of isotope peaks used for i-FIT = 2

Monoisotopic Mass, Even Electron Ions 19 formula(e) evaluated with 1 results within limits (up to 1 best isotopic matches for each mass) Elements Used:

C: 0-30 H: 0-23 O: 0-2 S: 0-3 K: 0-1





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Fig. S11 ¹H NMR spectra of BTT-3 in CDCl₃ (400 MHz)



Fig. S12 ¹³C NMR spectrum of BTT-3 in CDCl₃ (100 MHz)

Elemental Composition Report

Single Mass Analysis Tolerance = 50.0 mDa / DBE: min = -1.5, max = 100.0 Element prediction: Off Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions 47 formula(e) evaluated with 15 results within limits (up to 1 closest results for each mass) Elements Used: C: 0-26 H: 0-30 O: 4-6 S: 0-3 K: 0-1







Fig. S14 ¹H NMR spectra of BTT-4 in DMSO- d_6 (400 MHz)







