Applications in Catalysis and SERS

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Figure S1. TEM images of empty wipe (1a, 1b), AgNP@wipe-1 (2a, 2b), AgNP@wipe-2 (3a, 3b) and AgNP@wipe-3 (4a, 4b). Samples 1a-4a were stained with 5% aqueous uranyl acetate first and then 2% lead citrate. Samples were washed with DI water after stain with each dye. Samples 1b-4b were unstained. Scale bar is 100 nm.





Figure S2. The size distribution histogram of the AgNPs calculated from TEM images of AgNP@wipe-1 (1a), AgNP@wipe-2 (1b) and AgNP@wipe-3 (1c).



Figure S3. UV-visible spectra of run on water after sonicating wipes for 1.5 hrs and no absorbance for AgNP was observed.



Figure S4. UV-visible spectra of 0.12 mM 4-nitrophenol in presence of 12 mM NaBH₄ without any catalyst. There is no visible conversion 4-nitrophenol to 4-aminophenol even after 35 minutes of reaction time.



Figure S5. (A) Plot of UV-visible absorbance of 0.12 mM 4-nitrophenol against wavelength. The absorption peak at 400 nm gradually decreases due to catalytic activity of AgNP@wipe-2 in presence 12 mM NaBH₄. (B) The pseudo-first-order kinetics of 4-nitrophenol with excess NaBH₄ in aqueous solutions is plotted as $\ln(A_t/A_0)$ vs time (s). (C) Percent conversion of 4-nitrophenol in five successive cycles.



Figure S6. (A) Plot of UV-visible absorbance of 0.12 mM 4-nitrophenol against wavelength. The absorption peak at 400 nm gradually decreases due to catalytic activity of AgNP@wipe-1 in presence 12 mM NaBH₄. (B) The pseudo-first-order kinetics of 4-nitrophenol with excess NaBH₄ in aqueous solutions is plotted as $\ln(A_t/A_0)$ vs time (s). (C) Percent conversion of 4-nitrophenol in three successive cycles.



Figure S7. (A) Raman spectra of AgNP@wipe-1 (a), AgNP@wipe-2 (c), and AgNP@wipe-3 (d) before catalytic reaction. SERS spectra of 4-aminophenol adsorbed on AgNP@wipe-1 (b), AgNP@wipe-2 (d) and AgNP@wipe-3 (f) after catalytic reaction completing 3 cycles, 5 cycles and 8 cycles, respectively. The characteristic signal for 4-aminophenol appears at 1228 cm⁻¹.