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Supporting Information:

Fabrication of non-enzymatic Ni(II) loaded ZSM-5 nanozeolite and multi-walled carbon nanotubes paste electrode as glucose electrochemical sensor

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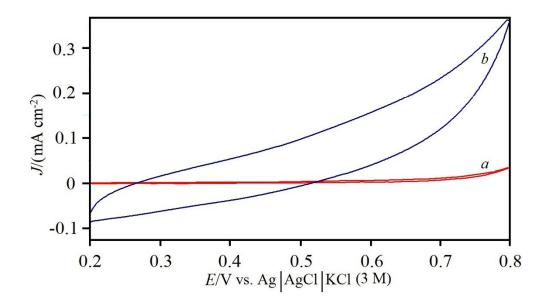


Fig. S1 The cyclic voltammograms of (a) MW/CPE and (b) ZSM-5/CPE in 0.1 M NaOH before immersion in 0.5 M NiCl₂ solution.

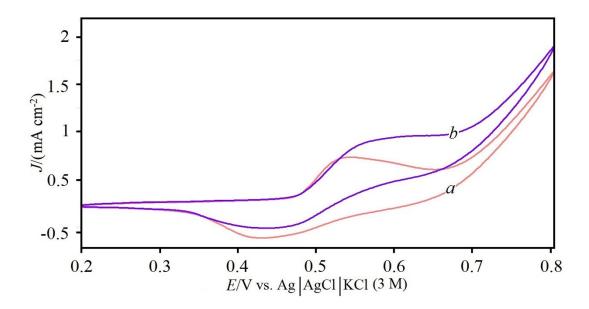


Fig. S2 The cyclic voltammograms of Ni-MW/CPE (a) in the absence and (b) in the presence of 0.005 M glucose in 0.1 M NaOH at scan rate of 20 mV s⁻¹.

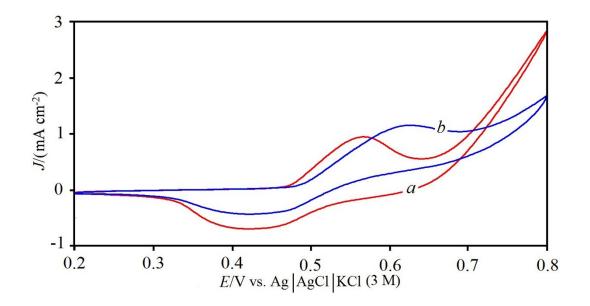


Fig. S3 The cyclic voltammograms of Ni-ZSM-5/CPE (a) in the absence and (b) in the presence of 0.005 M glucose in 0.1 M NaOH at scan rate of 20 mV s⁻¹.

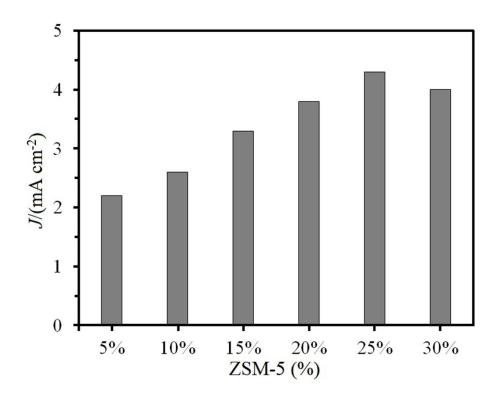


Fig. S4 The effect of ZSM-5 percentage together with 10 wt% of MWCNTs on the current densities of electrocatalytic oxidation of 0.005 M glucose in 0.1 M NaOH at scan rate of 20 mV $\rm s^{-1}$.

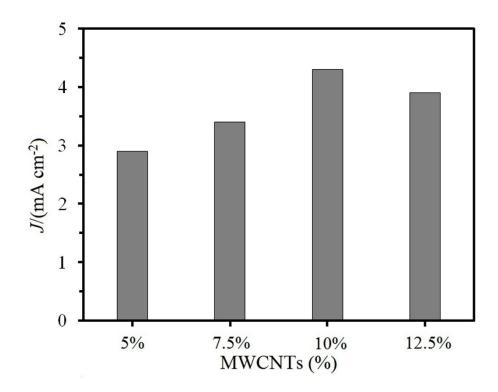


Fig. S5 The effect of MWCNTs percentage together with 25 wt% of ZSM-5 on the current densities of electrocatalytic oxidation of 0.005 M glucose in 0.1 M NaOH at scan rate of 20 mV $\rm s^{-1}$.

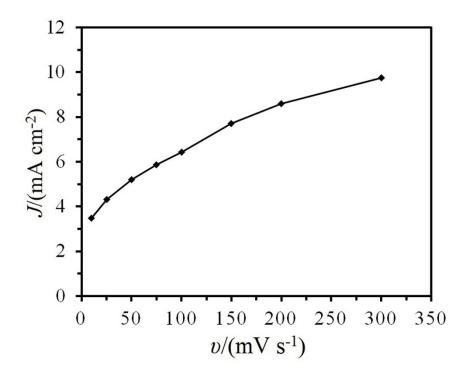


Fig. S6 The Variation of J_{pa} vs. scan rates (v) for electrocatalytic oxidation of 0.005 M glucose in 0.1 M NaOH at the surface of Ni-MW-ZSM-5/CPE.