

Figure S1. TEM image of Pd₂/HAP-6-S with Pd nanospheres of *ca.* 6 nm. The inset is the corresponding HRTEM image.

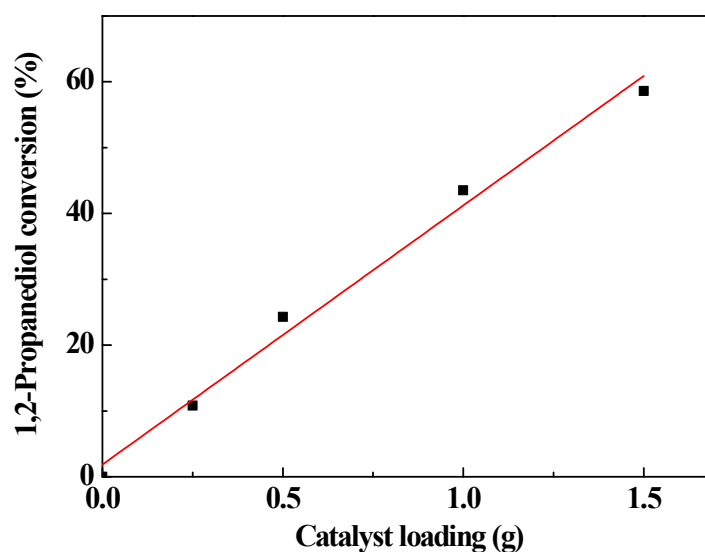
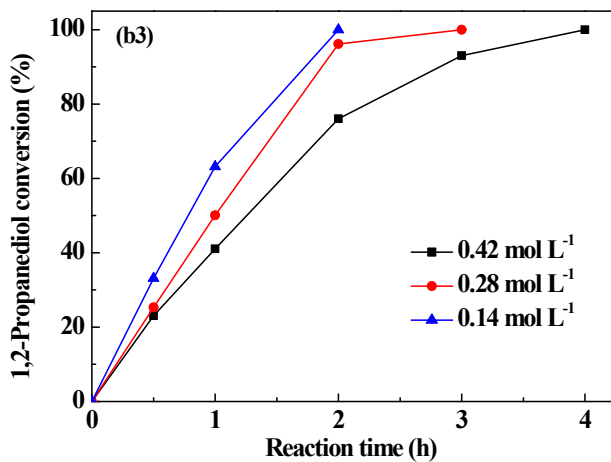
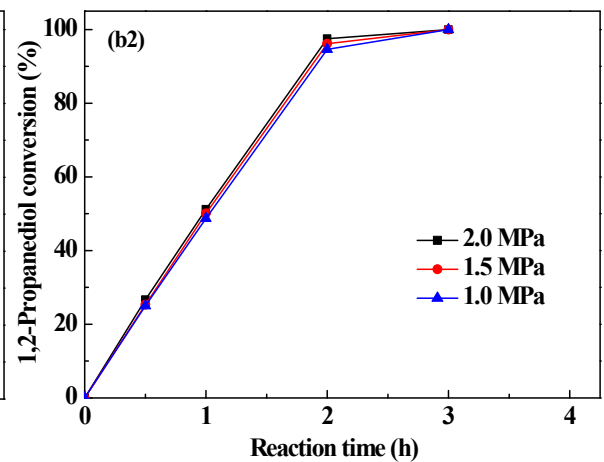
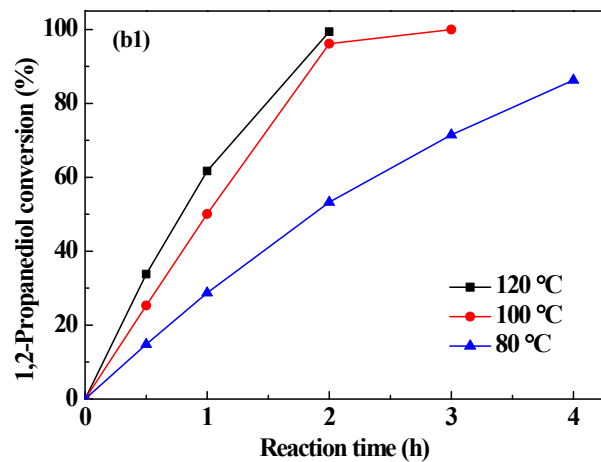
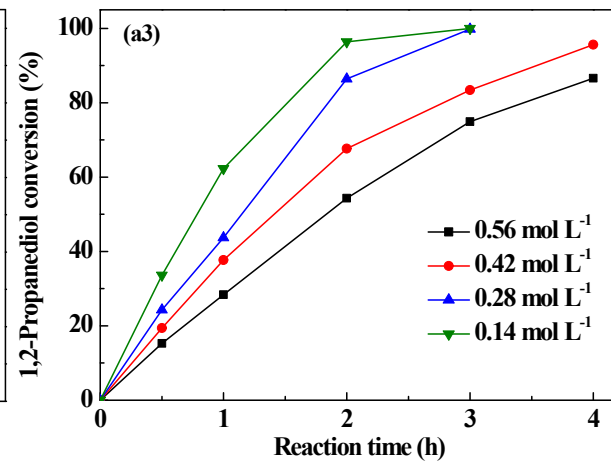
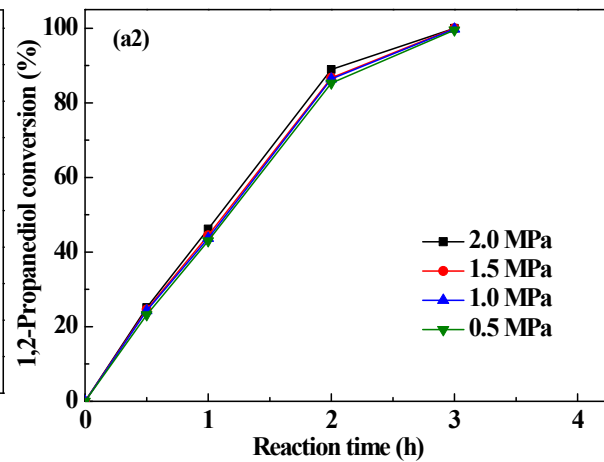
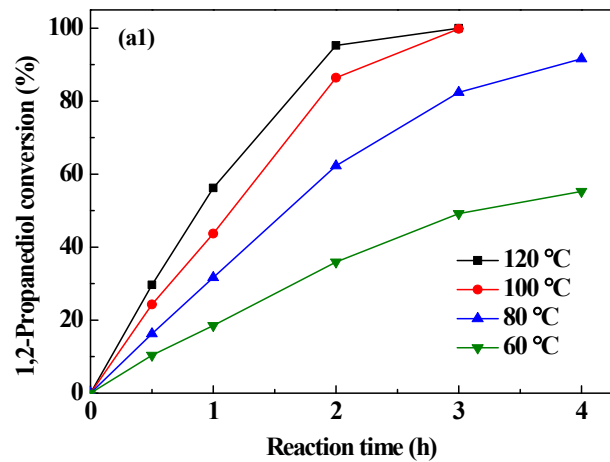


Figure S2. Catalyst loading vs 1,2-propanediol conversion. Reaction conditions: NaOH concentration, 0.56 mol L⁻¹; O₂ pressure, 1 MPa; 1,2-propanediol aqueous solution, 0.28 mol L⁻¹, 200 mL; reaction temperature, 100 °C; reaction time, 0.5 h.



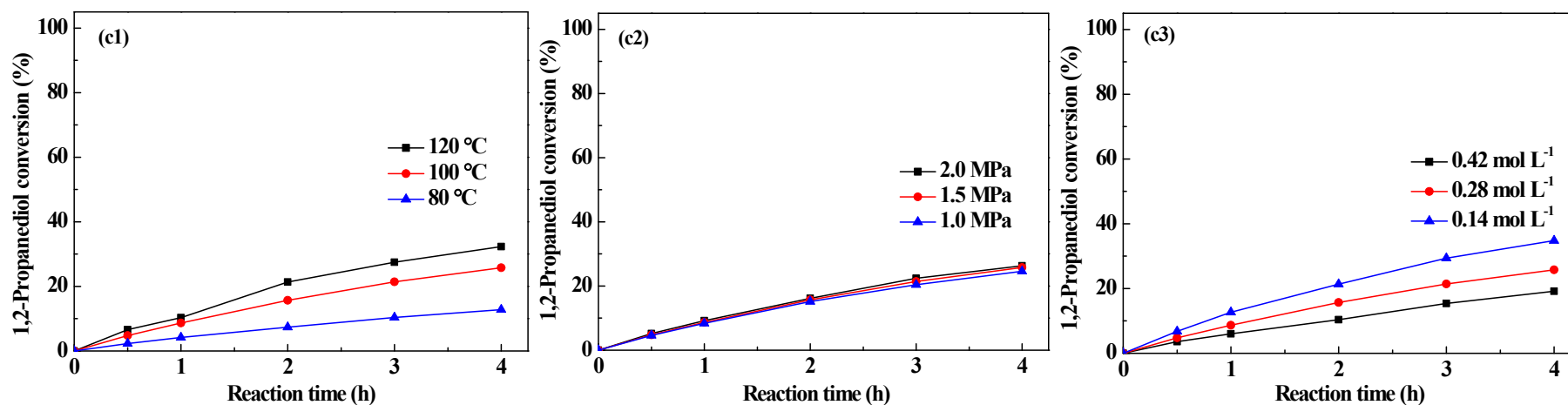


Figure S3. Effect of (a1-c1) reaction temperature, (a2-c2) O₂ pressure, and (a3-c3) 1,2-propanediol concentration on the conversion of 1,2-propanediol over (a) Pd₁Ag₁/HAP, (b) Pd₂/HAP-6, and (c) Ag₂/HAP catalysts. Reaction conditions: (a1-c1), 1,2-propanediol aqueous solution, 0.28 mol L⁻¹, 200 mL; NaOH concentration, 0.56 mol L⁻¹; O₂ pressure, 1.0 MPa; catalyst, 0.5 g. (a2-c2), 1,2-propanediol aqueous solution, 0.28 mol L⁻¹, 200 mL; NaOH concentration, 0.56 mol L⁻¹; reaction temperature, 100 °C; catalyst, 0.5 g. (a3-c3) 1,2-propanediol aqueous solution, 200 mL; NaOH concentration, 0.56 mol L⁻¹; reaction temperature, 100 °C; O₂ pressure, 1.0 MPa; catalyst, 0.5 g.

Table S1. Recycling performances of Pd₁Ag₁/HAP and Pd₂/HAP-6 catalysts in the catalytic oxidation of 1,2-propanediol^a

Catalysts	Runs	Amount of noble metals ^b		1,2-Propanediol Conversions (%)	Selectivities (%)			TON (h ⁻¹)
		(μmol g ⁻¹)			Lactic acid	Formic acid	Acetic acid	
		Pd	Ag					
Pd ₁ Ag ₁ /HAP	1	93.3	93.2	86.3	88.8	3.1	8.1	259
	2	92.8	92.0	84.2	87.8	3.4	8.8	258
	3	90.5	89.8	83.1	87.3	4.1	8.6	258
	4	87.0	88.5	79.6	87.0	4.2	8.8	254
Pd ₂ /HAP-6	1	189.2		96.2	86.2	4.5	9.3	285
	2	173.2		85.9	86.0	4.4	9.6	278
	3	164.7		77.4	85.8	4.7	9.5	263
	4	152.6		70.6	85.4	5.2	9.5	259

^aReaction conditions: 1,2-propanediol aqueous solution, 0.28 mol L⁻¹, 200 mL; NaOH concentration, 0.56 mol L⁻¹; catalyst, 0.5 g; O₂ pressure, 1.0 MPa; reaction temperature, 100 °C; reaction time, 2 h.

^bThe amount of Pd and Ag were detected by ICP.