

## Supporting Information for:

### **Theoretical exploration of Seleno and Tellurophenols as a promising alternative to sulfur ligands in the anchoring to gold (111) materials**

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**Table S1.** Selected geometric parameters for the *para* substituted selenophenol-gold complexes<sup>a</sup> (distances in Å and angles in deg).

<b>Au<sub>42</sub>-Selenophenol (SeH)</b>	<b>D<sub>top</sub></b>	<b>D<sub>C-Se</sub></b>	<b>θ</b>	<b>ω</b>
Au <sub>42</sub> -SeHC <sub>6</sub> H <sub>4</sub> NH <sub>2</sub>	2.70	1.93	87.6	95.3
Au <sub>42</sub> -SeHC <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	2.69	1.93	88.4	94.0
Au <sub>42</sub> -SeHC <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>	2.57	1.89	88.5	90.9
Au <sub>42</sub> -SeHC <sub>6</sub> H <sub>5</sub>	2.70	1.93	90.3	93.3
Au <sub>42</sub> -SeHC <sub>6</sub> H <sub>4</sub> F	2.68	1.93	90.3	94.2
Au <sub>42</sub> -SeHC <sub>6</sub> H <sub>4</sub> Cl	2.72	1.93	88.3	95.9
Au <sub>42</sub> -SeHC <sub>6</sub> H <sub>4</sub> OCOCH <sub>3</sub>	2.70	1.93	90.6	93.1
Au <sub>42</sub> -SeHC <sub>6</sub> H <sub>4</sub> CF <sub>3</sub>	2.70	1.93	91.1	94.3
Au <sub>42</sub> -SeHC <sub>6</sub> H <sub>4</sub> CN	2.60	1.91	86.9	93.2
Au <sub>42</sub> -SeHC <sub>6</sub> H <sub>4</sub> NO <sub>2</sub>	2.75	1.92	88.6	96.7

a The nomenclature used on the column headings for the definition of the geometric parameters is detailed in Figure 2 from the manuscript.

**Table S2.** Selected geometric parameters for the *para* substituted telurophenol-gold complexes<sup>a</sup> (distances in Å and angles in deg).

<b>Au<sub>42</sub>-Telurophenol (TeH)</b>	<b>D<sub>top</sub></b>	<b>D<sub>C-Te</sub></b>	<b>θ</b>	<b>ω</b>
Au <sub>42</sub> -TeHC <sub>6</sub> H <sub>4</sub> NH <sub>2</sub>	2.77	2.12	87.24	95.83
Au <sub>42</sub> -TeHC <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	2.76	2.12	88.82	93.08
Au <sub>42</sub> -TeHC <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>	2.75	2.12	88.83	91.21
Au <sub>42</sub> -TeHC <sub>6</sub> H <sub>5</sub>	2.77	2.13	89.50	93.54
Au <sub>42</sub> -TeHC <sub>6</sub> H <sub>4</sub> F	2.75	2.13	89.50	93.78
Au <sub>42</sub> -TeHC <sub>6</sub> H <sub>4</sub> Cl	2.77	2.12	87.97	95.28
Au <sub>42</sub> -TeHC <sub>6</sub> H <sub>4</sub> OCOCH <sub>3</sub>	2.76	2.13	89.47	93.87
Au <sub>42</sub> -TeHC <sub>6</sub> H <sub>4</sub> CF <sub>3</sub>	2.76	2.13	89.86	94.90
Au <sub>42</sub> -TeHC <sub>6</sub> H <sub>4</sub> CN	2.77	2.12	87.69	91.56
Au <sub>42</sub> -TeHC <sub>6</sub> H <sub>4</sub> NO <sub>2</sub>	2.79	2.12	88.11	96.98

a The nomenclature used on the column headings for the definition of the geometric parameters is detailed in Figure 2 from the manuscript.

**Table S3.** Selected geometric parameters for the anionic *para* substituted selenophenolate-gold complexes<sup>a</sup> (distances in Å and angles in deg).

<b>Au<sub>42</sub>-Selenophenolate (Se<sup>-</sup>)</b>	<b>D<sub>1</sub></b>	<b>D<sub>2</sub></b>	<b>D<sub>C-Se</sub></b>	<b>α</b>	<b>β</b>
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> NH <sub>2</sub>	2.61	2.59	1.96	67.7	89.9
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	2.60	2.63	1.96	67.3	88.1
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>	2.60	2.60	1.97	67.6	90.3
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>5</sub>	2.62	2.59	1.97	67.5	90.7
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> F	2.61	2.59	1.97	67.6	88.7
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> Cl	2.62	2.60	1.97	67.4	89.3
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> OCOCH <sub>3</sub>	2.63	2.60	1.97	67.3	90.2
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> CF <sub>3</sub>	2.63	2.60	1.96	67.4	88.2
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> CN	2.62	2.61	1.96	67.3	92.1
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> NO <sub>2</sub>	2.64	2.60	1.96	67.0	89.1

<sup>a</sup> The nomenclature used on the column headings for the definition of the geometric parameters is detailed on Figure 3 from the manuscript.

**Table S4.** Selected geometric parameters for the anionic *para* substituted telurophenolates-gold complexes<sup>a</sup> (distances in Å and angles in deg).

Au <sub>42</sub> -Telurophenolate (Te <sup>-</sup> )	D <sub>1</sub>	D <sub>2</sub>	D <sub>C-Te</sub>	α	β
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> NH <sub>2</sub>	2.70	2.72	2.16	64,69	87.40
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	2.69	2.73	2.17	64,55	85.16
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>	2.70	2.72	2.17	64,65	88.74
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>5</sub>	2.70	2.71	2.17	64,70	89.51
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> F	2.70	2.71	2.17	64,80	86.06
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> Cl	2.70	2.71	2.17	64,71	85.19
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> OCOCH <sub>3</sub>	2.70	2.71	2.17	64,66	88.74
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> CF <sub>3</sub>	2.68	2.71	2.17	65,03	80.81
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> CN	2.71	2.72	2.17	64,54	92.22
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> NO <sub>2</sub>	2.70	2.71	2.17	64,66	85.42

a The nomenclature used on the column headings for the definition of the geometric parameters is detailed on Figure 3 from the manuscript.

**Table S5.** Selected geometric parameters for the radical *para* substituted selenophenolate-gold complexes<sup>a</sup> (distances in Å and angles in deg).

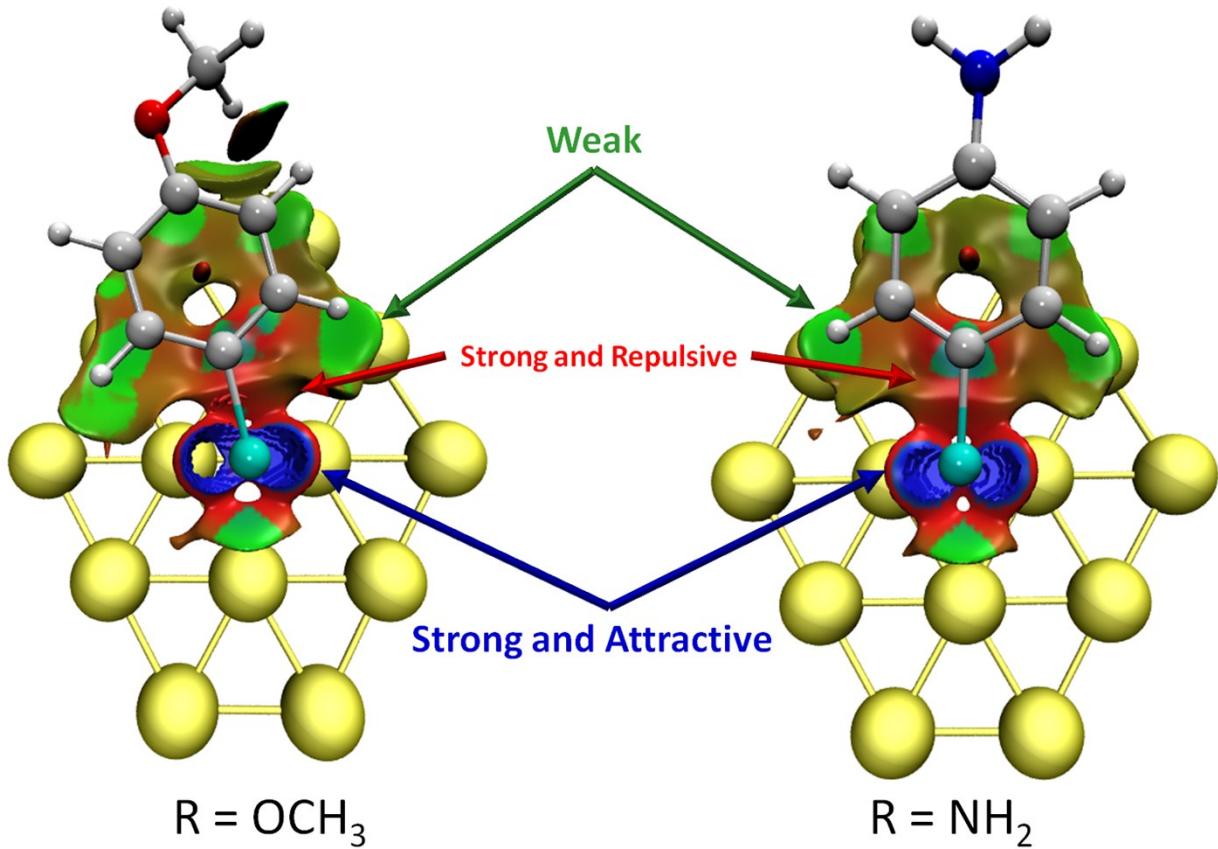
<b>Au<sub>42</sub>-Selenophenolate (Se<sup>•</sup>)</b>	<b>D<sub>1</sub></b>	<b>D<sub>2</sub></b>	<b>D<sub>C-Se</sub></b>	<b>α</b>	<b>β</b>
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> NH <sub>2</sub>	2.58	2.60	1.96	68.0	90.0
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	2.59	2.62	1.96	67.4	87.9
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>	2.60	2.58	1.97	68.0	90.2
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>5</sub>	2.59	2.59	1.97	67.9	89.6
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> F	2.59	2.59	1.97	67.9	88.6
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> Cl	2.60	2.60	1.97	67.7	89.1
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> OCOCH <sub>3</sub>	2.60	2.60	1.97	67.7	89.5
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> CF <sub>3</sub>	2.60	2.60	1.97	67.7	87.1
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> CN	2.62	2.58	1.97	67.6	90.7
Au <sub>42</sub> -SeC <sub>6</sub> H <sub>4</sub> NO <sub>2</sub>	2.60	2.60	1.96	67.5	87.6

<sup>a</sup> The nomenclature used on the column headings for the definition of the geometric parameters is detailed on Figure 3 from the manuscript.

**Table S6.** Selected geometric parameters for the radical *para* substituted telurophenolates-gold complexes<sup>a</sup> (distances in Å and angles in deg).

Au <sub>42</sub> -Telurophenolate (Te <sup>•</sup> )	D <sub>1</sub>	D <sub>2</sub>	D <sub>C-Te</sub>	α	β
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> NH <sub>2</sub>	2.69	2.72	2.16	64.73	89.46
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub>	2.70	2.71	2.17	64.73	88.00
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>	2.69	2.72	2.17	64.70	88.89
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>5</sub>	2.70	2.71	2.17	64.76	89.03
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> F	2.69	2.71	2.17	64.80	86.53
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> Cl	2.70	2.71	2.17	64.68	86.99
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> OCOCH <sub>3</sub>	2.70	2.72	2.17	64.68	88.36
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> CF <sub>3</sub>	2.70	2.72	2.17	64.67	82.37
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> CN	2.69	2.73	2.17	64.60	89.25
Au <sub>42</sub> -TeC <sub>6</sub> H <sub>4</sub> NO <sub>2</sub>	2.71	2.72	2.17	64.52	83.53

a The nomenclature used on the column headings for the definition of the geometric parameters is detailed on Figure 3 from the manuscript.



**Figure S1.** Graphical representation of the non-covalent interactions between the radical ligands (carbon-grey, nitrogen-blue, hydrogen-white, cyan-chalcogen spheres) and the gold surface (yellow spheres). At the left side there is the ligand with the strongest interaction strength ( $-\text{OCH}_3$ ) and at the right side of the picture the ligand with the weakest ( $-\text{NH}_2$ ).