

Supplementary Materials

Efficient conversion of glucose to 5-hydroxymethylfurfural by using bifunctional partially hydroxylated AlF₃

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Table S1 Acid distribution of AlF₃-150, AlF₃-250, and AlF₃-350

Catalyst	Acid distribution (mmol/g) ^a	
	Brønsted acid	Lewis acid
AlF ₃ -150	0.0305	0.0669
AlF ₃ -250	0.0142	0.0434
AlF ₃ -350	0.0081	0.0276

^a Deduced from the intensity of the band of Brønsted acid sites located at 1540 cm⁻¹ and Lewis acid sites at 1450 cm⁻¹ by using FT-IR of adsorbed pyridine.

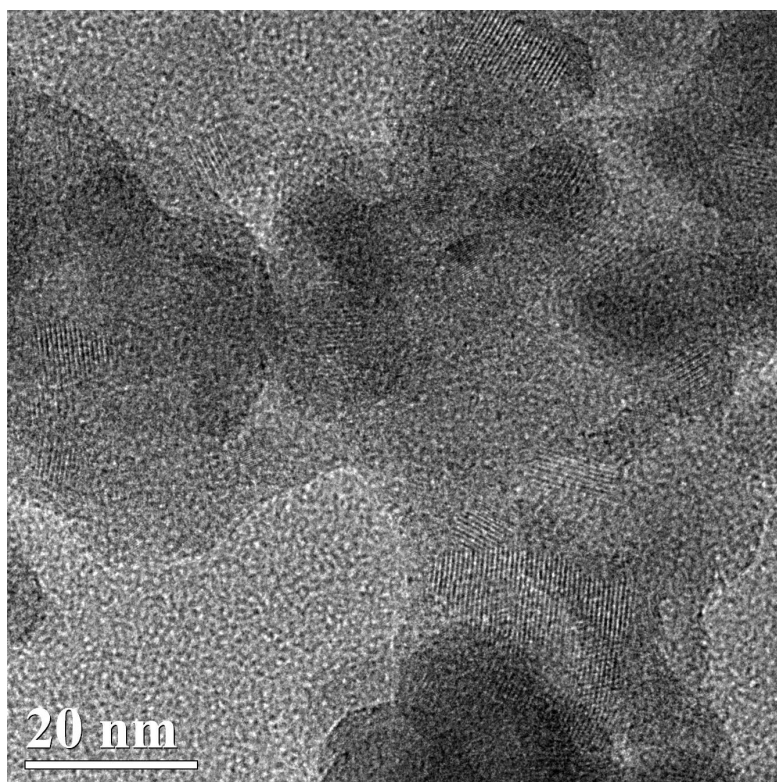


Fig. S1. TEM image of AlF₃-150 sample

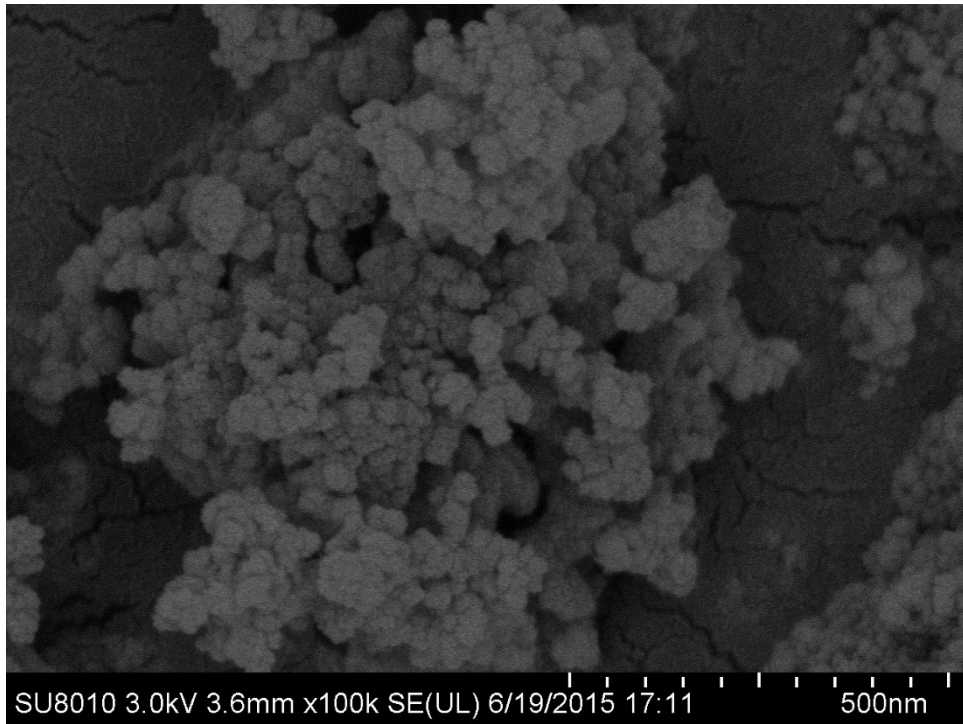


Fig. S2. SEM image of AlF₃-150 sample

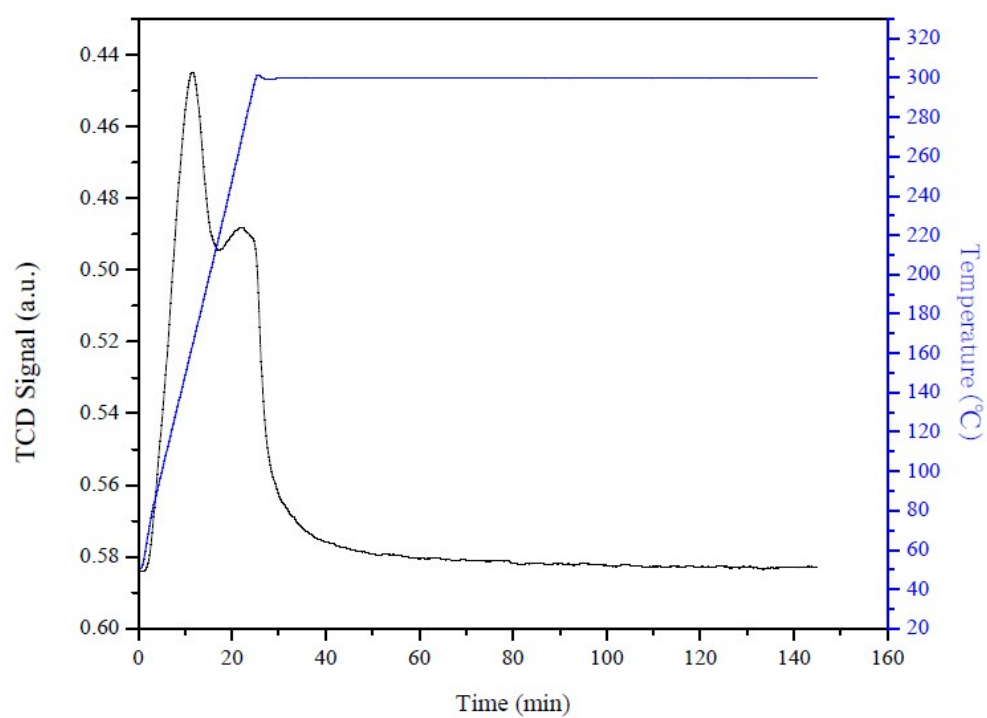


Fig. S3. NH₃-TPD profile of AlF₃-150 sample

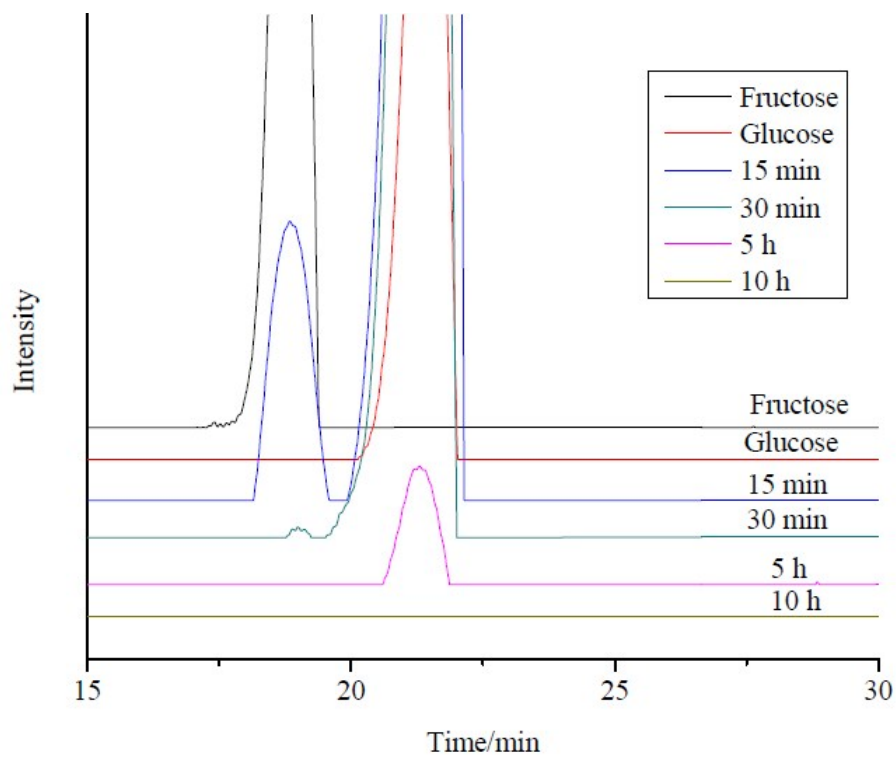


Fig. S4. HPLC images of the reaction mixture after reacting for 15 min, 30 min, 5 h and 10 h (Reaction conditions: 50 mg glucose, 1.0 g DMSO, 20 mg AlF_3 -150, and 140 °C).

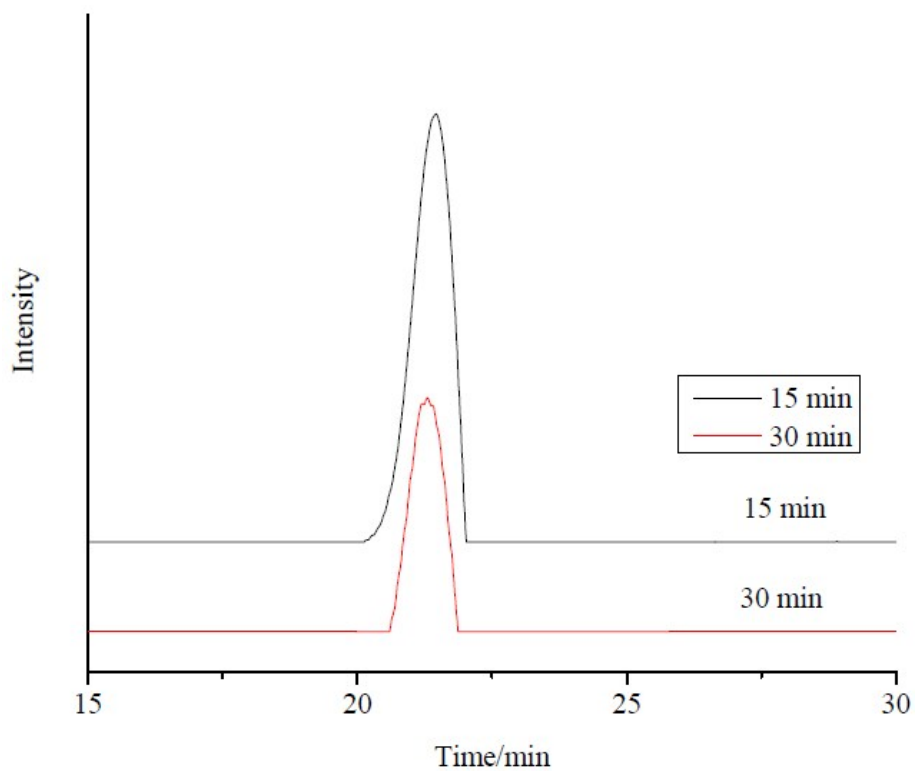


Fig. S5. HPLC images of the reaction mixture catalyzed by HCl for 15 and 30 min (Reaction conditions: 50 mg glucose, 1.0 g DMSO, 20 mg 0.01wt% HCl, and 140 °C).

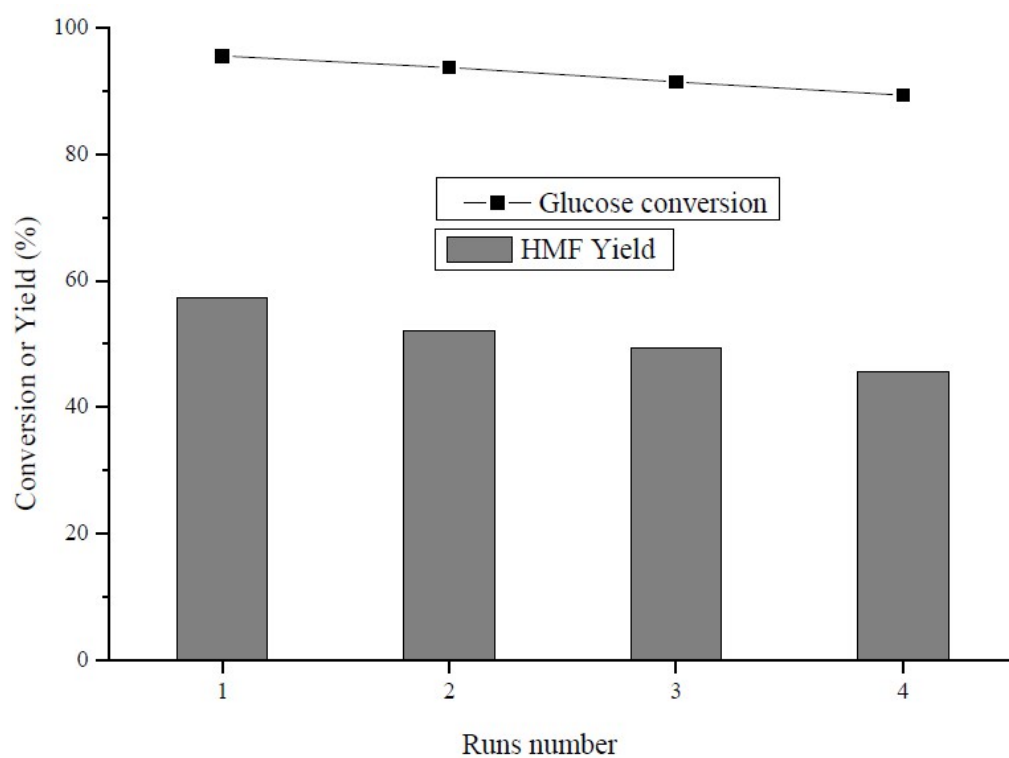


Fig. S6. Recyclability of AlF₃-150 in glucose-to-HMF conversion (Reaction conditions: 50 mg glucose, 1.0 g DMSO, 20 mg catalyst, 140 °C, and 10 h).

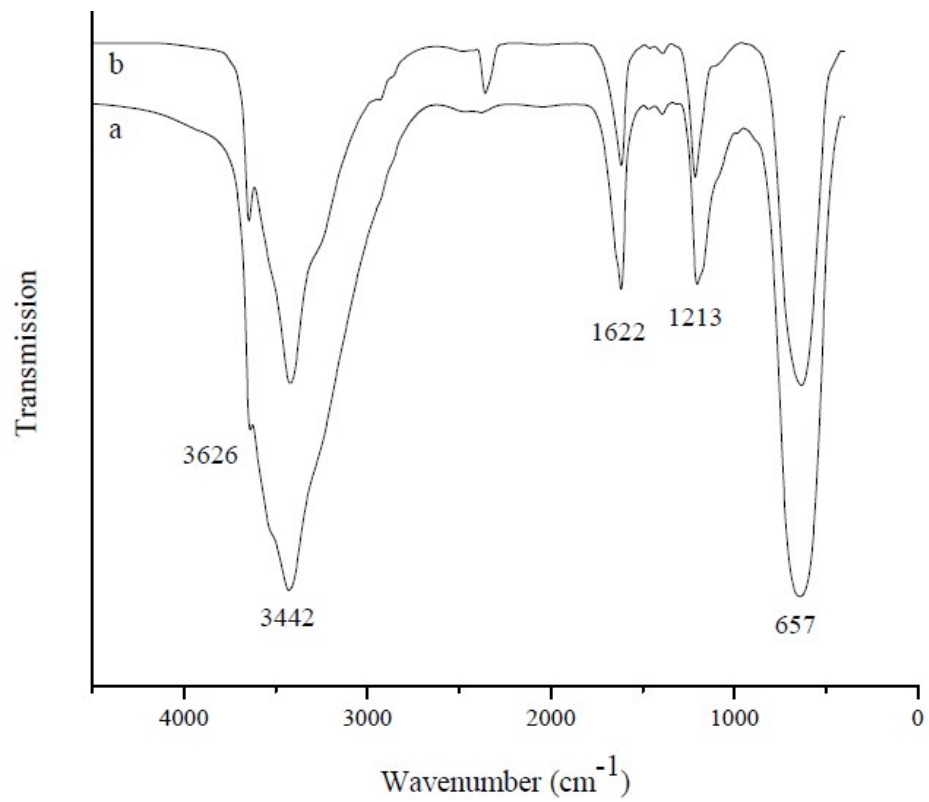


Fig. S7. FT-IR spectra of (a) fresh and (b) reused AlF₃-150 samples

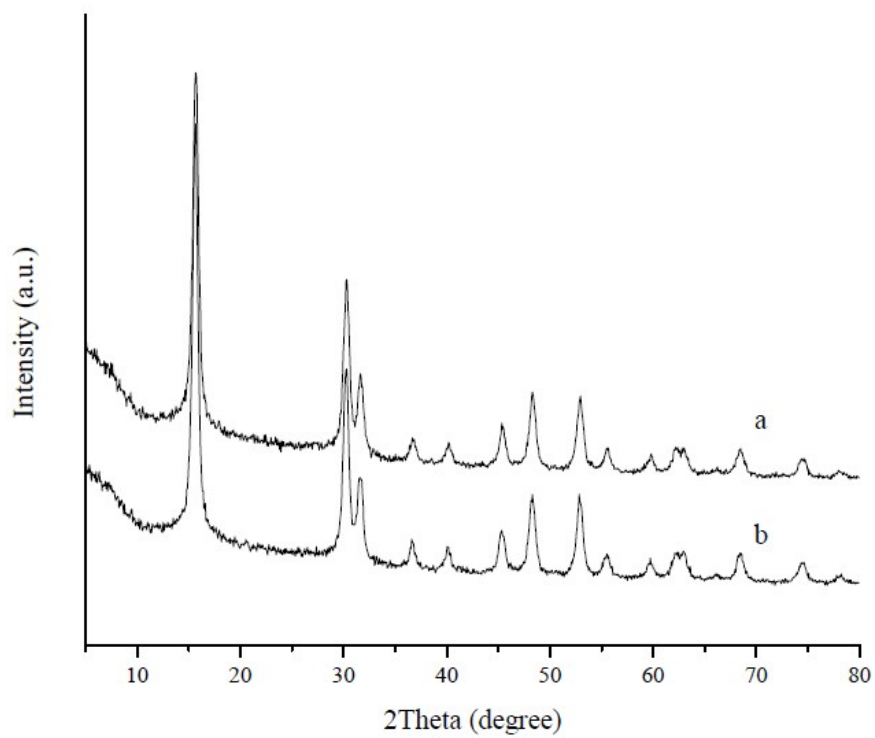


Fig. S8. XRD patterns of (a) fresh and (b) reused AlF₃-150 samples