

## Supporting Information

### Synthesis of Porous $\text{MnCo}_2\text{O}_4$ Microspheres with Yolk-Shell Structure Induced by Concentration Gradient and the Effect on their Performance in Electrochemical Energy Storage

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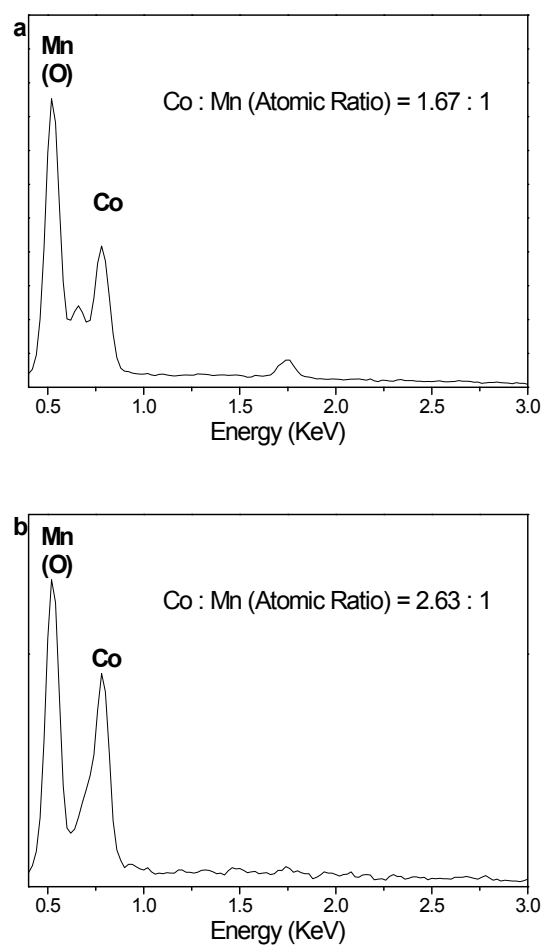
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**Fig. S1.** Representative EDX spectra in Fig. 4a. (a) Core and (b) Shell.

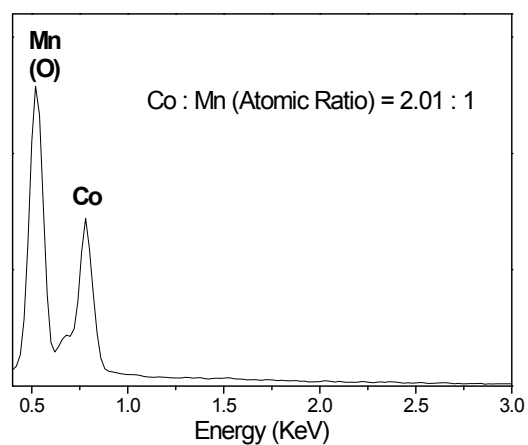


Fig. S2. Representative EDX spectrum in Fig. 5a.

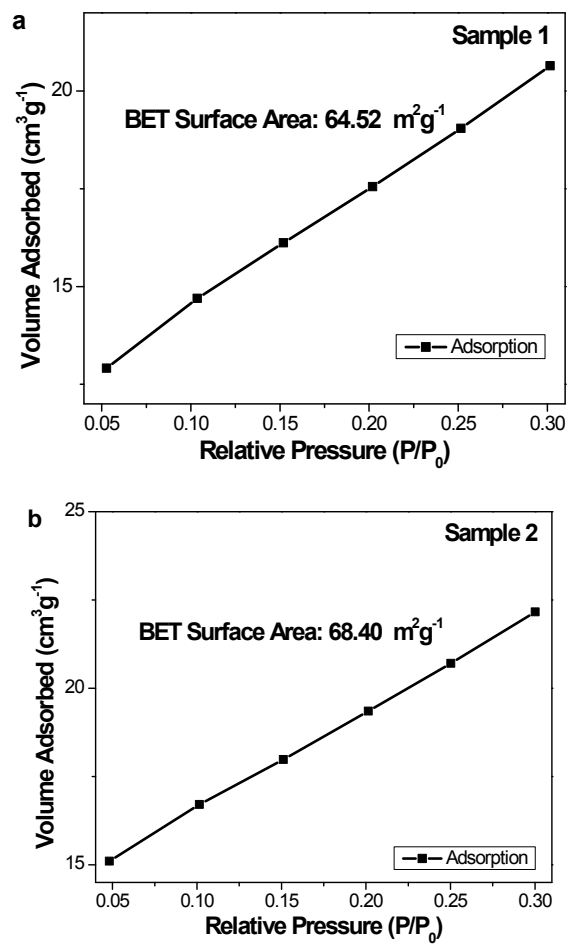
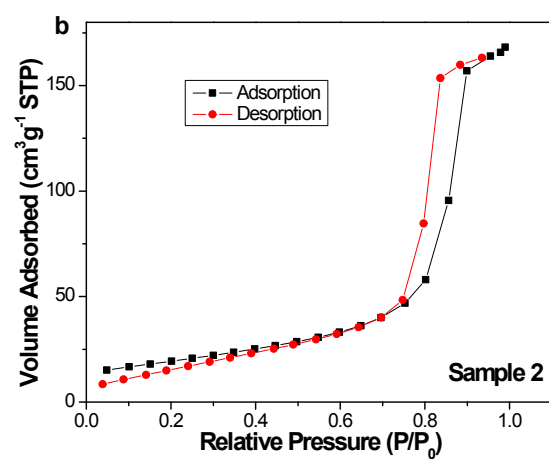
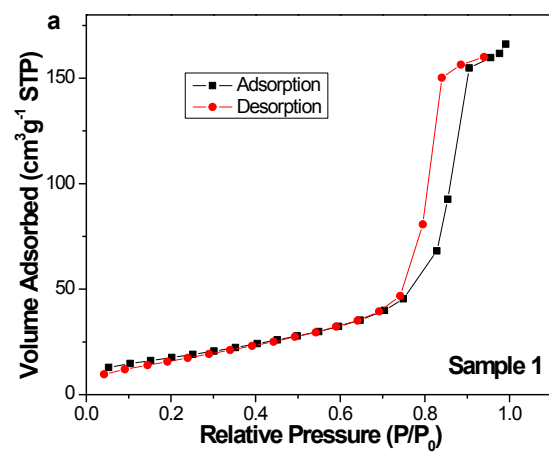
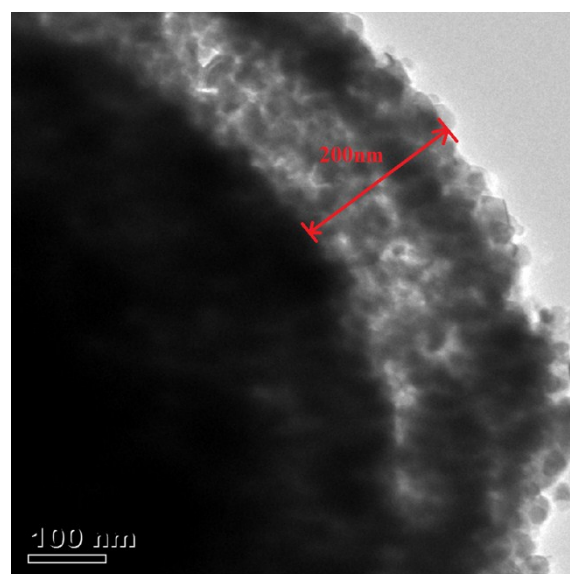


Fig. S3.  $N_2$  adsorption isotherms of  $\text{MnCo}_2\text{O}_4$  samples.



**Fig. S4.** N<sub>2</sub> adsorption/desorption isotherms of MnCo<sub>2</sub>O<sub>4</sub> samples.



**Fig. S5.** TEM image of partial enlarged detail of Sample 1 microsphere.

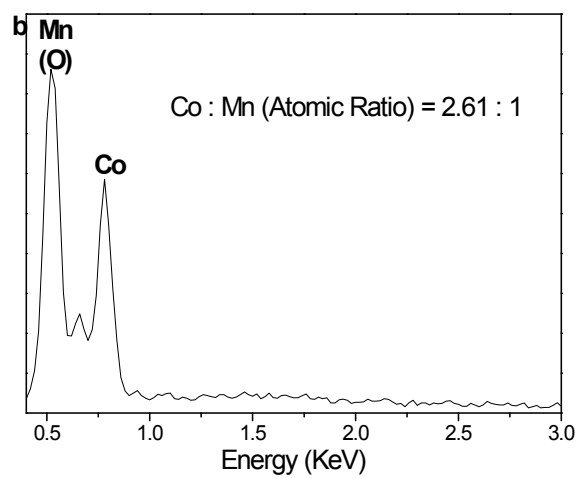
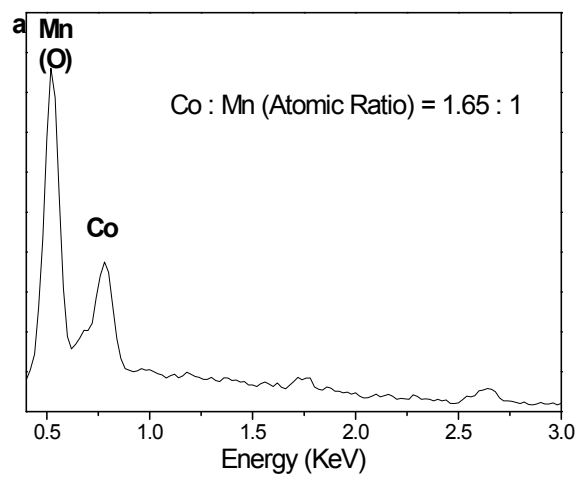


Fig. S6. Representative EDX spectra in Fig. 11a. (a) Core and (b) Shell.

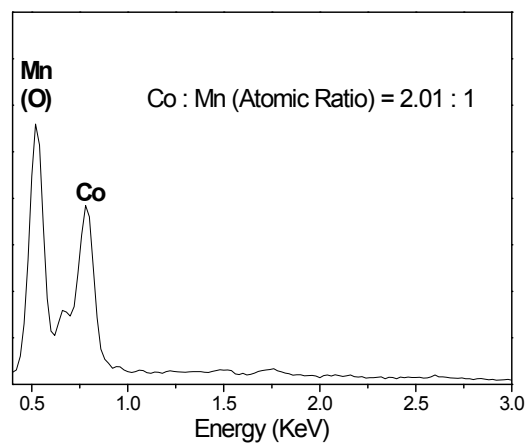
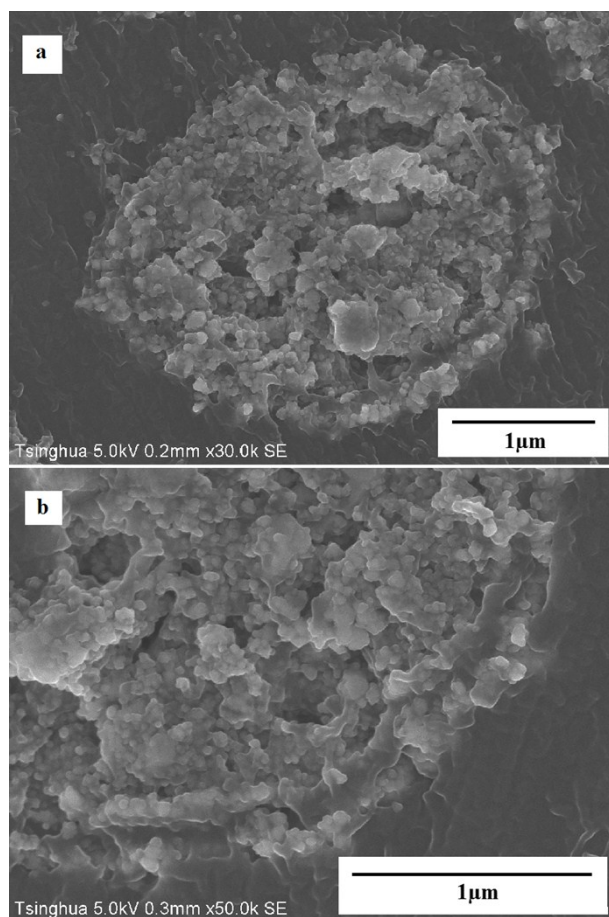
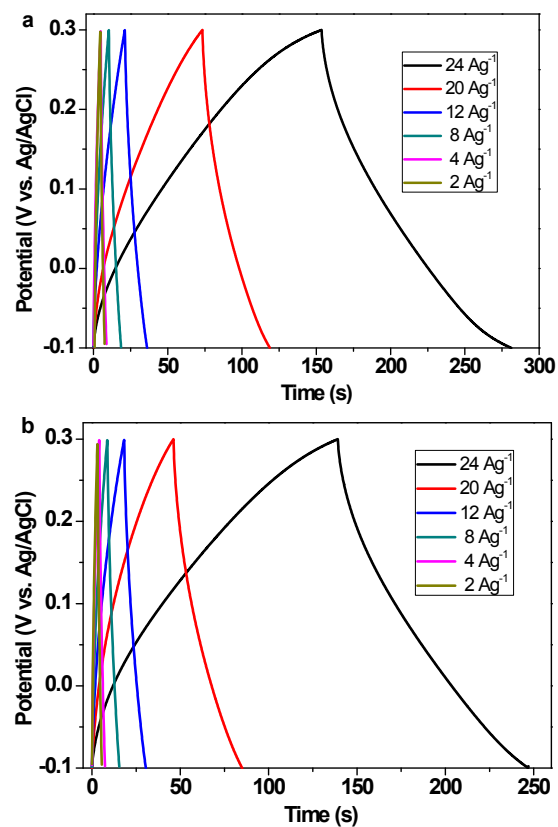


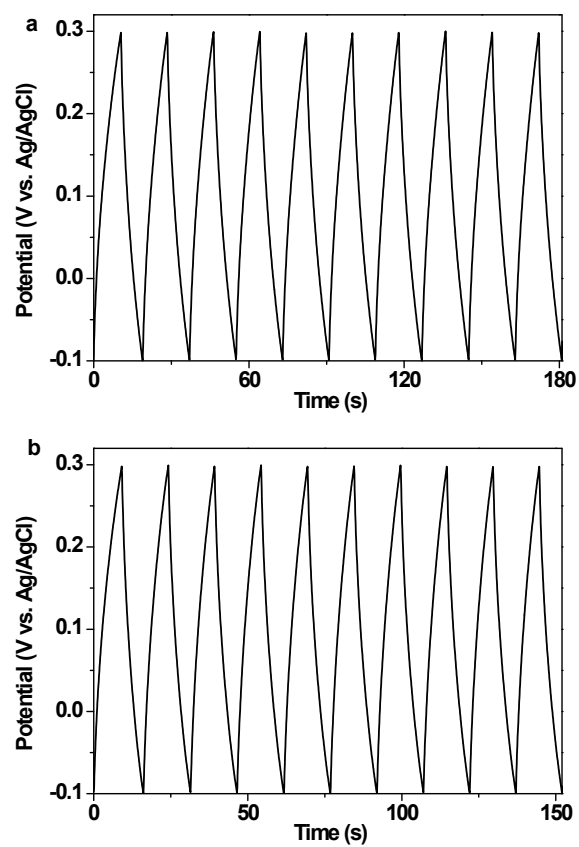
Fig. S7. Representative EDX spectrum in Fig. 12a.



**Fig. S8.** SEM images of the cross section of Sample 1 (mixtures of MnCo<sub>2</sub>O<sub>4</sub>/ATB/PVDF) after 40 cycles at 0.2 Ag<sup>-1</sup>.



**Fig. S9.** Galvanostatic charge-discharge curves of MnCo<sub>2</sub>O<sub>4</sub> electrodes at various current densities. (a) Sample 1 and (b) Sample 2.



**Fig. S10.** First ten galvanostatic charge-discharge curves of  $\text{MnCo}_2\text{O}_4$  electrodes in the range of -0.10~0.30 V at  $12 \text{ Ag}^{-1}$ . (a) Sample 1 and (b) Sample 2.