Supplementary file:



Fig. S1: SEM images of inactivated Bio-mass (a) Bio-mass (b) Bio-mass loaded with F^-



Fig. S2: TGA curve of pure algae



b

Fig.S3 (a) Experimental FT-IR spectra of the algae and fluoride loaded algae (lower curve). (b) DFT computed FT-IR spectra of the optimized structure of Phycocyanobilin2 loaded with fluoride.



Fig. S4: optimized structure of Phycocyanobilin2 loaded with fluoride (Energy: -73680.85 eV; Dipole moment: 12.6 Debye; Hydrogen bond length in NH-----F(86): 1.52 A; C-OH---F(84): 1.03 A and C-OH---F(82)---HN: 1.22 & 1.71 A).



Fig. S5: (a) Langmuir isotherm (Plot of C_e vs C_e/q_e) and (b) Freundlich isotherm (log C_e vs log q_e)



Fig. S6 Break-through curve



Fig. S7: Plot of 1000/T vs $\log K_d$

Peak position	Nature of peak	Peak assignment	Ref.			
3132-3650 cm ⁻¹	Sharp and broad	v (N-H) in heterocyclic pyrydyl ring and (C-O) in				
	peak extend up	carboxylic acid; Formation of hydrogen bonds (Fig.9b)				
	to 3000 cm ⁻¹	by electronegative fluoride. [NH stretching in NHF				
		(85) (DFT) at 3191.3 cm ⁻¹ and NH stretching in				
		N(9)HF(82) (DFT) at 3456.9 cm ⁻¹].				
2440 cm ⁻¹	Sharp and broad	d ν (H-O) stretching in carboxylic acid. This new peak appears at 2440 cm ⁻¹ (adjacent to C-O in –COOH ir				
	peak					
		pure algae appears 2338 cm ⁻¹) in the loaded exchanger				
		only. [Peak at 2447.3 cm ⁻¹ (DFT) was found to be				
		present for ν (OH) stretching mode in F(84)HOOC				
		confirming the F ⁻ intake].				
1616 cm ⁻¹ ; 1725	Sharp and	v (N-H) in heterocyclic pyrydyl ring and v (C-O) in	[15,			
cm ⁻¹	broad. A new	carboxylic acid; Formation of hydrogen bonds (Fig.9b)	22]			
	peak appears at	by electronegative fluoride. A new peak appears at				
	1725 cm ⁻¹	1725 cm ⁻¹ in the loaded exchanger. [Peak at 1732.3				
		cm ⁻¹ (DFT) was found to be present for v (NH)				
		bending and Peak at 1807.8 cm ⁻¹ (DFT) for ν (OH)				
		bending mode in NHF(82)HOOC].				

Table S1: FT-IR Peak	positions and	Peak assignment
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Table S2: Tolerance limits of foreign ions on the quantitative sorption (>93%) of fluoride; Experimental conditions: Biomass: 1.0 g; volume of solution: 100 mL; Concentration: 0.02 mg mL⁻¹; pH: Neutral pH.

Concentrations of foreign ions: 50-100 mM L ⁻¹									
I-	Cl	Br -	SO4 ²⁻	PO ₄ ³⁻	CH ₃ COO	ClO ₄ -			
30	20	20	50	75	40	40			
60	40	40	60	85	70	60			
90	80	80	100	95	100	80			