

SUPPORTING INFORMATION

Three Way Electron Transfer of C-N-S Tri Doped Two-Phase Junction of TiO₂ Nanoparticles for Efficient Visible Light Photocatalytic Dye Degradation

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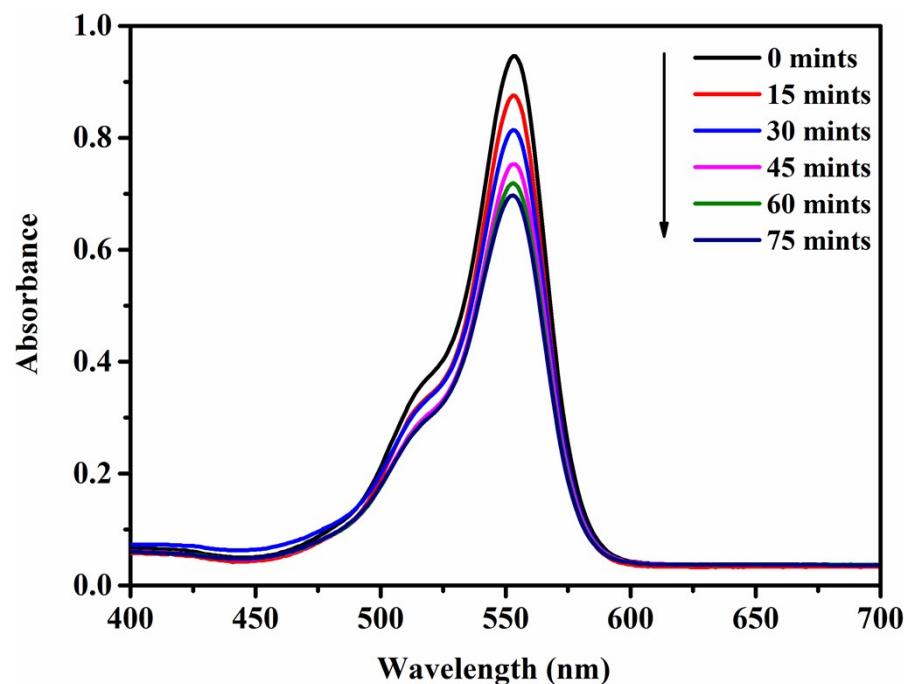


Figure S1. Adsorption – Desorption equilibrium determination of D4 sample under dark with different run time

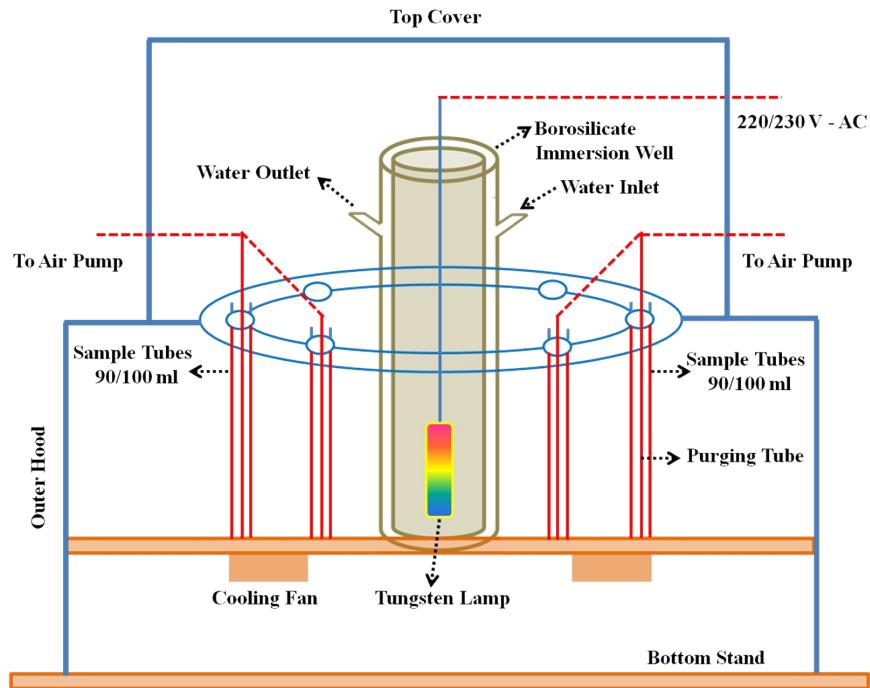


Figure S2. Schematic picture of photocatalytic reactor

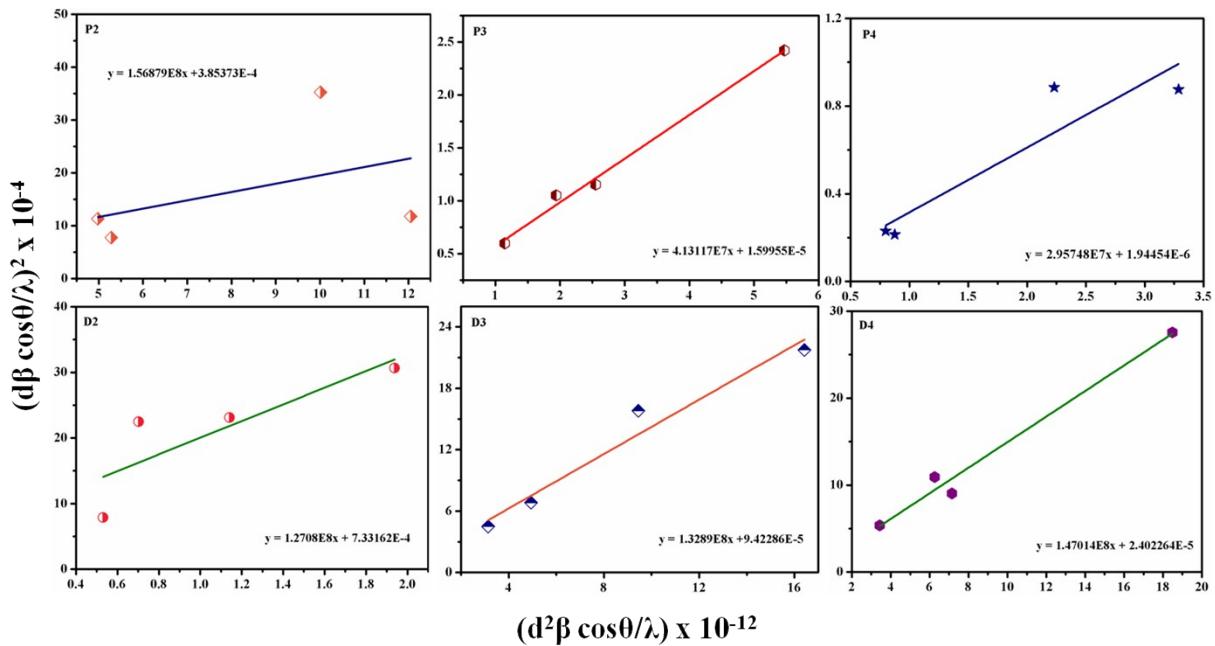


Figure S3. SSP – plot of undoped and C-N-S doped TiO₂ nanoparticles with different annealing temperature.

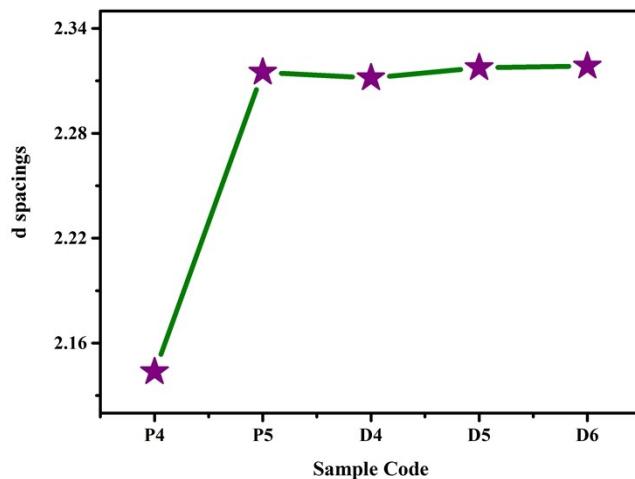


Figure S4. d-spacing variation of undoped and C-N-S doped TiO_2 nanoparticles.

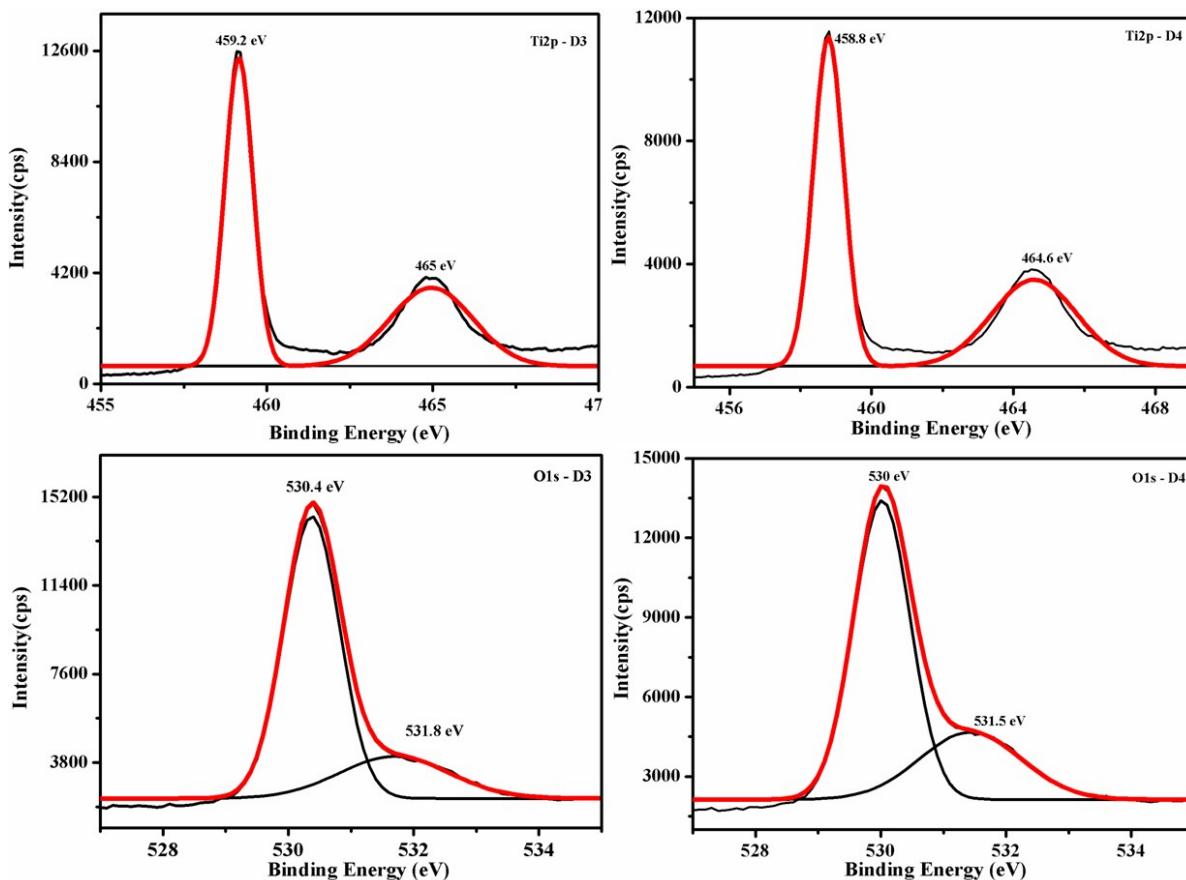


Figure S5. Ti 2p and O 1s XPS spectra of D3 and D4.

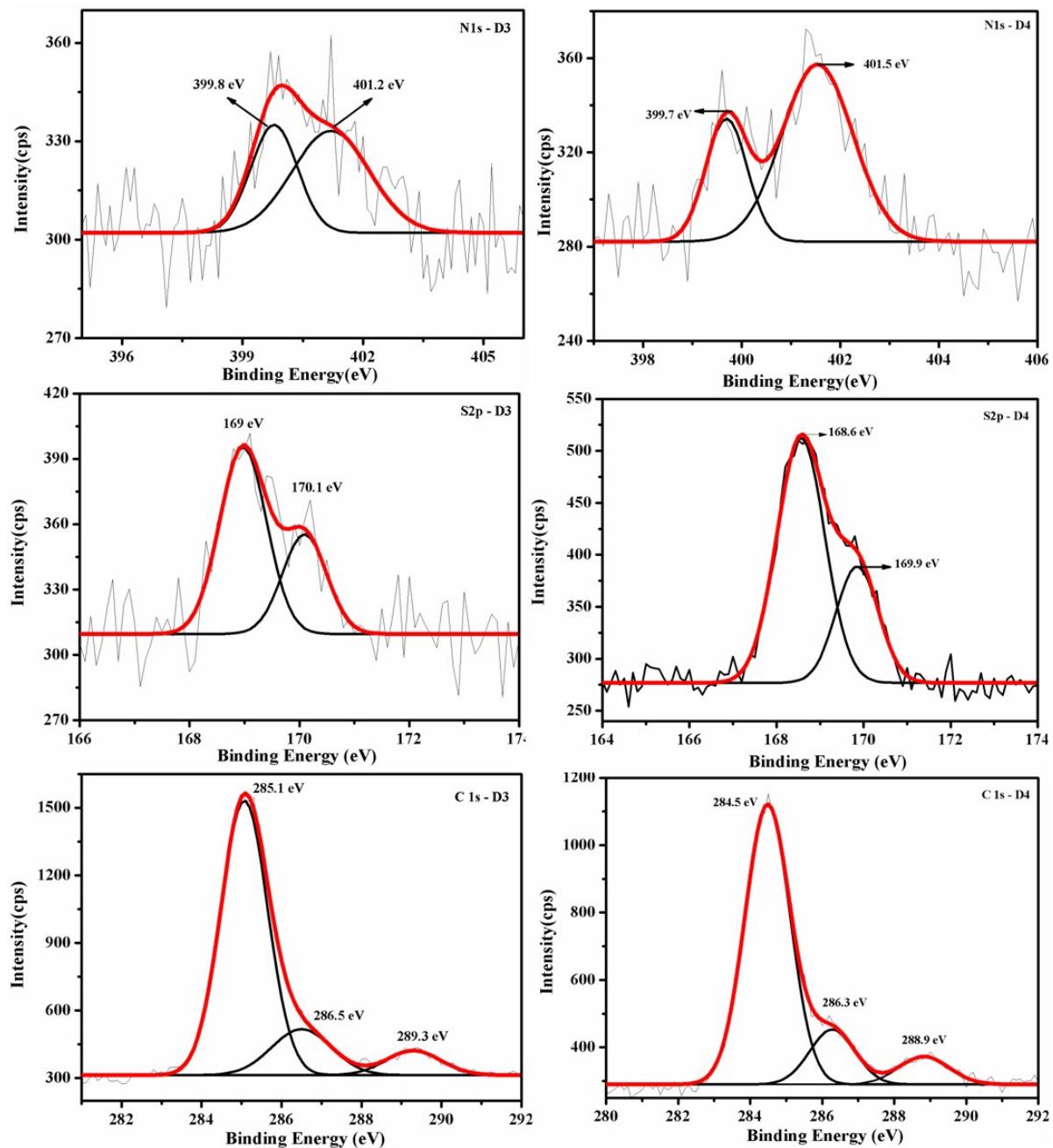


Figure S6. N 1s, S 2p and C 1s XPS spectra of D3 and D4.

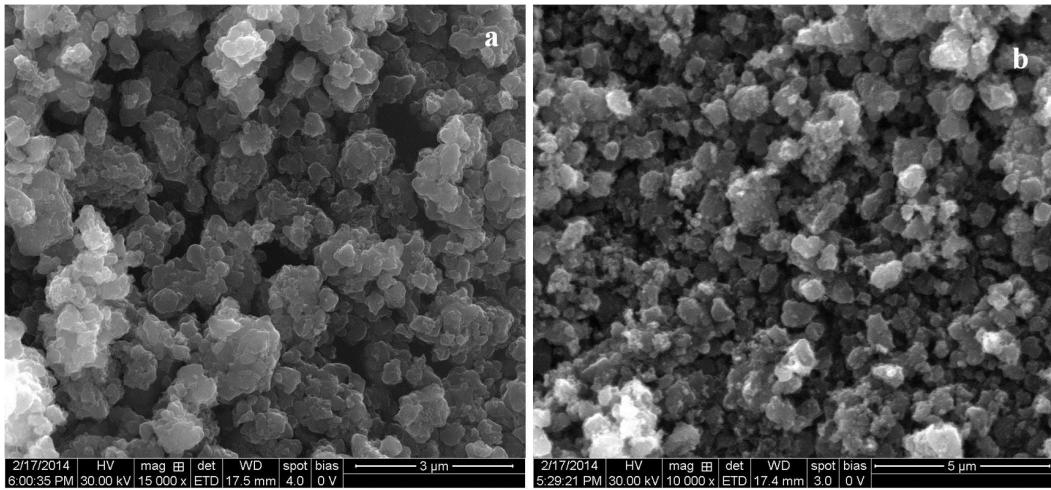


Figure S7. FESEM images of (a) P6 and (b) D6 catalysts.

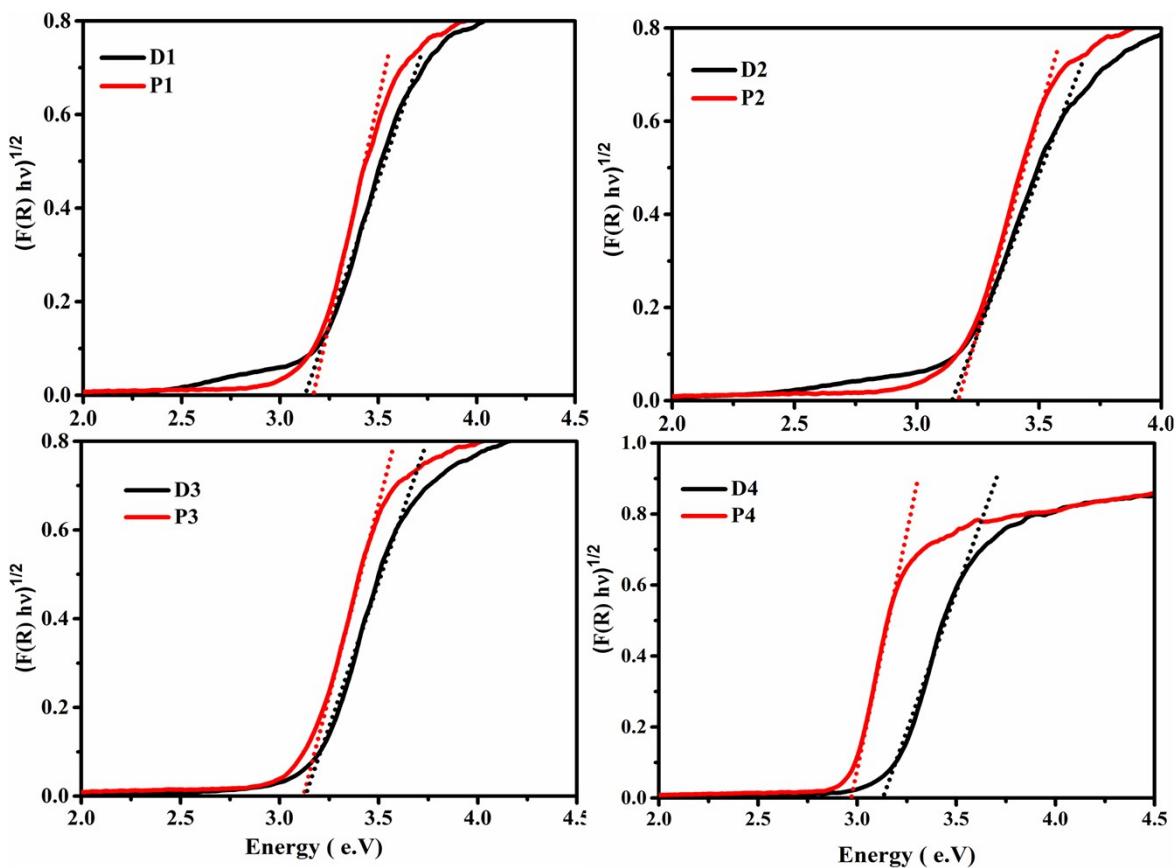


Figure S8. Bandgap determination of undoped and C-N-S doped TiO_2 nanoparticles with different annealing temperature using Kubelka – Munk function

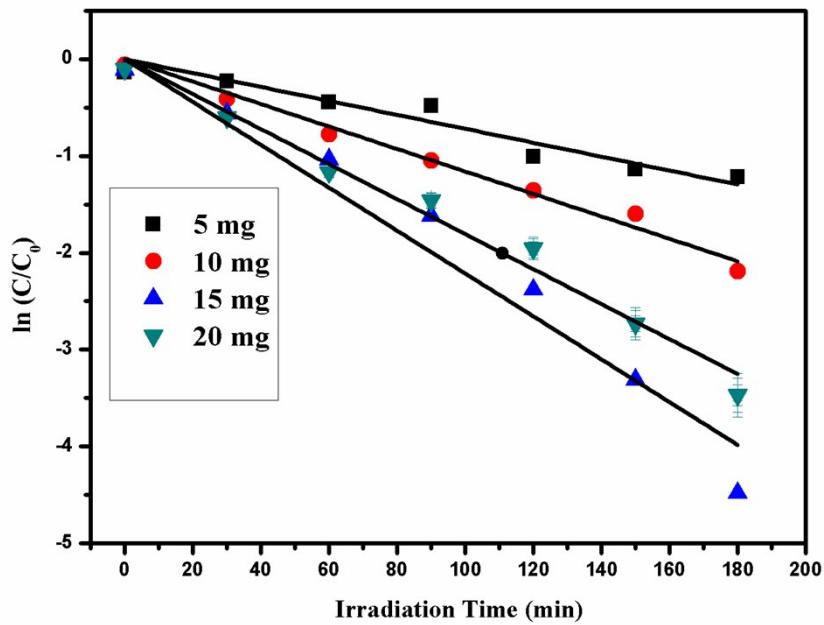


Figure S9. Kinetics of catalyst optimization of D4 sample

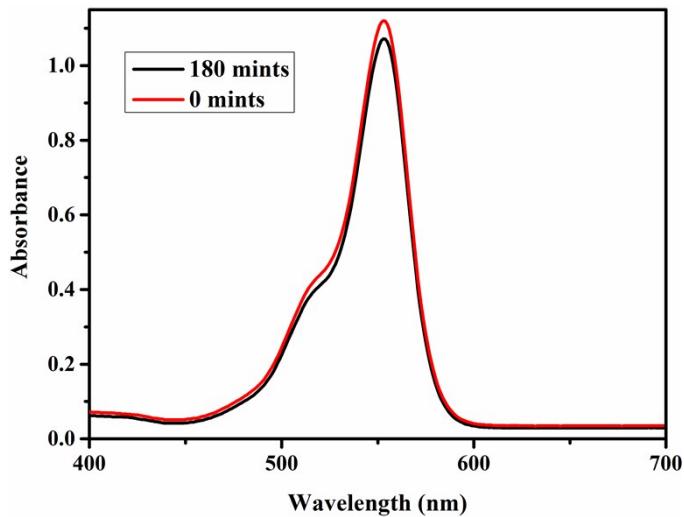


Figure S10. Absorbance spectra of Rh B without catalyst

Table S1. Lattice parameters and the grain size of the synthesized TiO₂ nanoparticles

Sample	Lattice parameter (Å)		Anatase	Rutile
	a	c	(%)	(%)
P1	a = 3.79	c = 9.45	100	-
P2	a = 3.79	c = 9.41	100	-
P3	a = 3.78	c = 9.51	100	-
P4	a = 4.59	c = 2.96	14	86
D1	a = 3.79	c = 9.42	100	-
D2	a = 3.78	c = 9.41	100	-
D3	a = 3.81	c = 9.48	100	-
D4	a = 3.79	c = 9.56	88	12

Table S2. Atomic percentage of C-N-S doped TiO₂ nanoparticles with different annealing temperatures.

Sample	Atomic concentration (%)				
	Ti 2p	O 1s	N 1s	S 2p	C 1s
D3	19.46	46.35	1.42	1.68	31.09
D4	21.73	55.33	1.51	2.04	19.4