

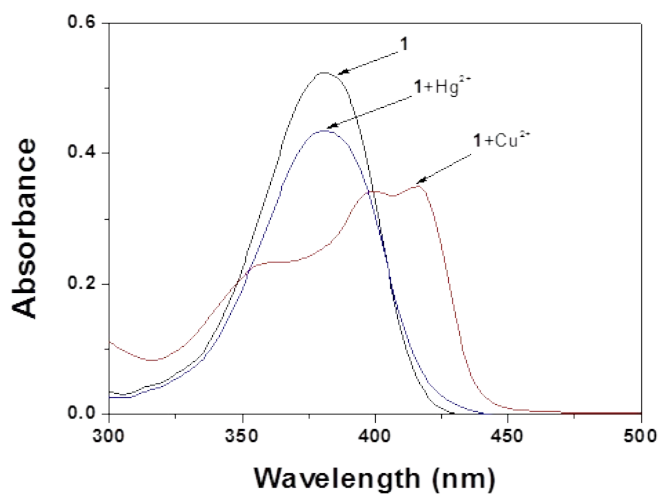
Supporting Information

Sequential detection of mercury (II) and thiol-containing amino acids by a fluorescent chemosensor

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(a)



(b)

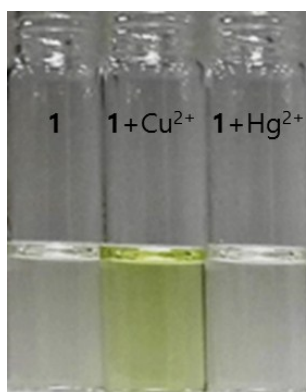


Fig. S1 (a) Changes in the UV-vis spectra of receptor **1** (10 μM) upon addition of Cu^{2+} (1 equiv) and Hg^{2+} (1 equiv), respectively, in DMSO/bis-tris buffer (8/2, v/v). The bathochromic shift of **1**- Cu^{2+} complex might be explained through the internal charge transfer (ICT) and ligand-to-metal charge-transfer (LMCT) in the molecule, which has the push-pull effect of the electron-donating and the electron-withdrawing group. Therefore, **1**- Cu^{2+} complex showed a color change, whereas **1**- Hg^{2+} complex did not. (b) The color changes of **1** (30 μM) upon the addition of Cu^{2+} (1 equiv) and Hg^{2+} (1 equiv), respectively, in DMSO/bis-tris buffer (8/2, v/v).

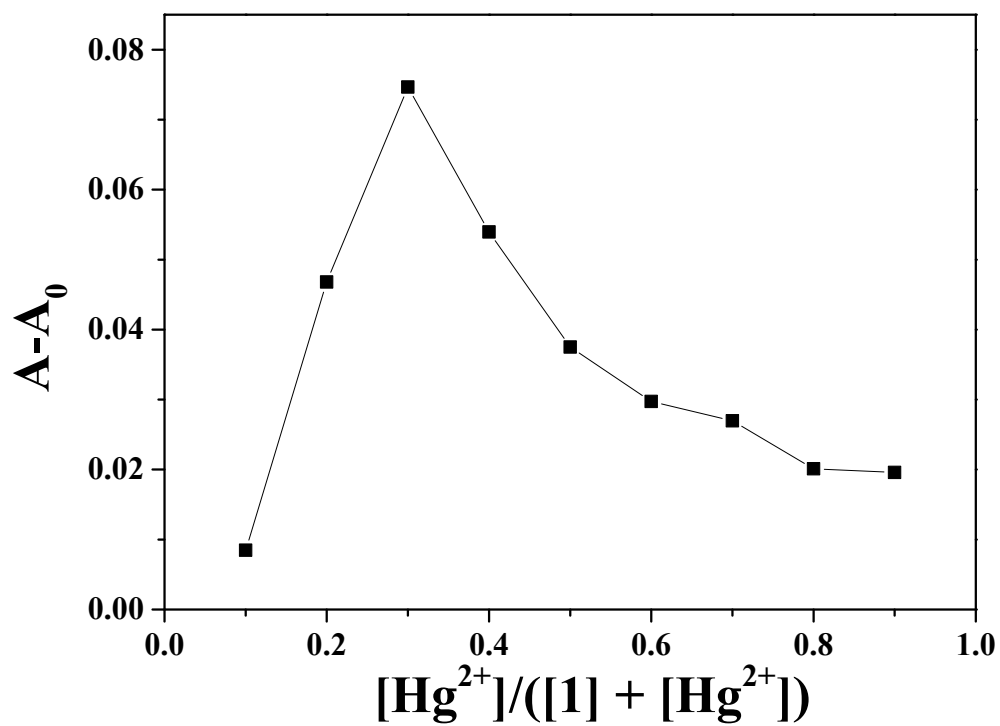


Fig. S2 Job plot for the binding of **1** with Hg^{2+} . Absorbance at 385 nm was plotted as a function of the molar ratio $[\text{Hg}^{2+}]/([\mathbf{1}] + [\text{Hg}^{2+}])$. The total concentrations of mercury ions with receptor **1** were 30 μM .

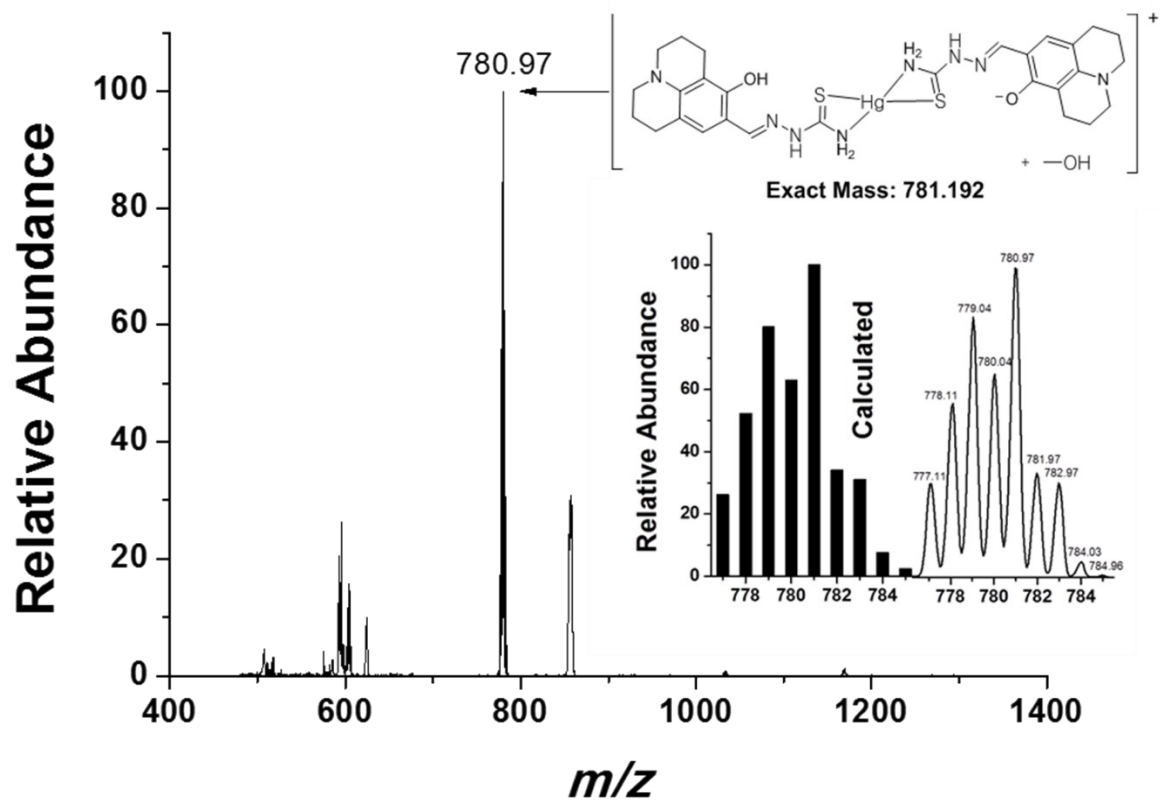


Fig. S3 Positive-ion electrospray ionization mass spectrum of **1** (100 μ M) upon addition of 1 equiv of $\text{Hg}(\text{NO}_3)_2$.

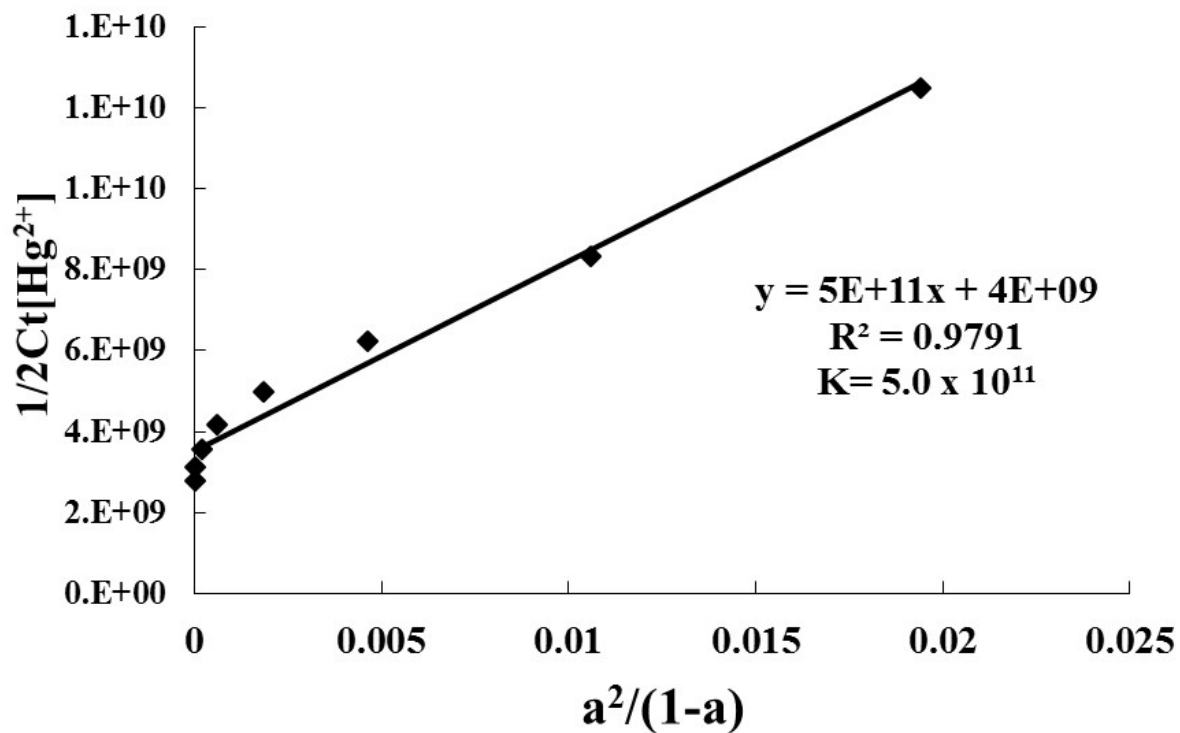


Fig. S4 Li's equation plot (absorbance at 382 nm) of **1**, assuming 1:2 stoichiometry for association between Hg^{2+} and **1**. 'Ct' means the concentration of **1**, and 'a' does $[(A_x - A_{max}) / (A_0 - A_{max})]$.

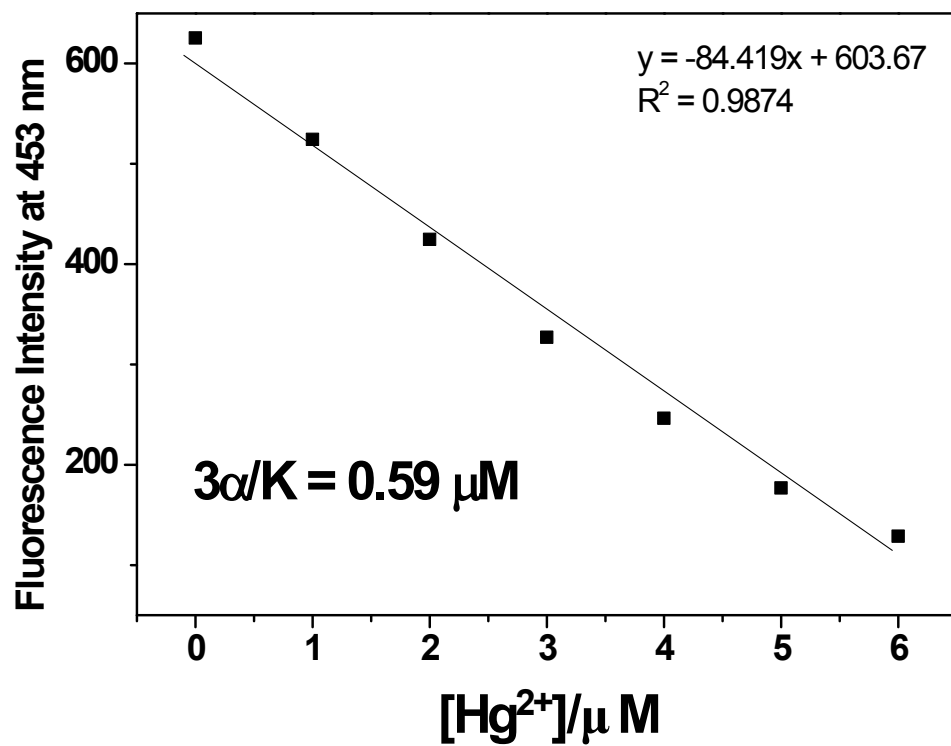


Fig. S5 Determination of the detection limit based on change in the ratio (absorbance at 453 nm) of **1** (10 μM) with Hg^{2+} .

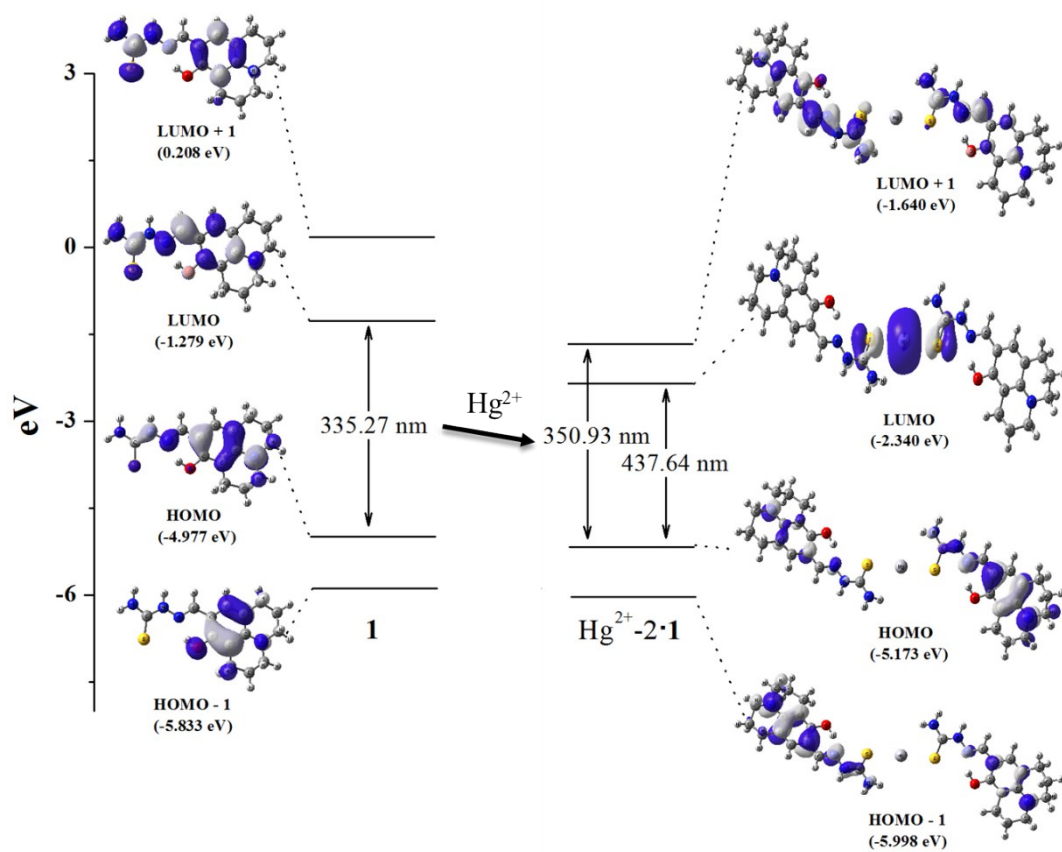


Fig. S6 Frontier molecular orbitals and their energies of **1** and Hg^{2+} -**2·1** complex.

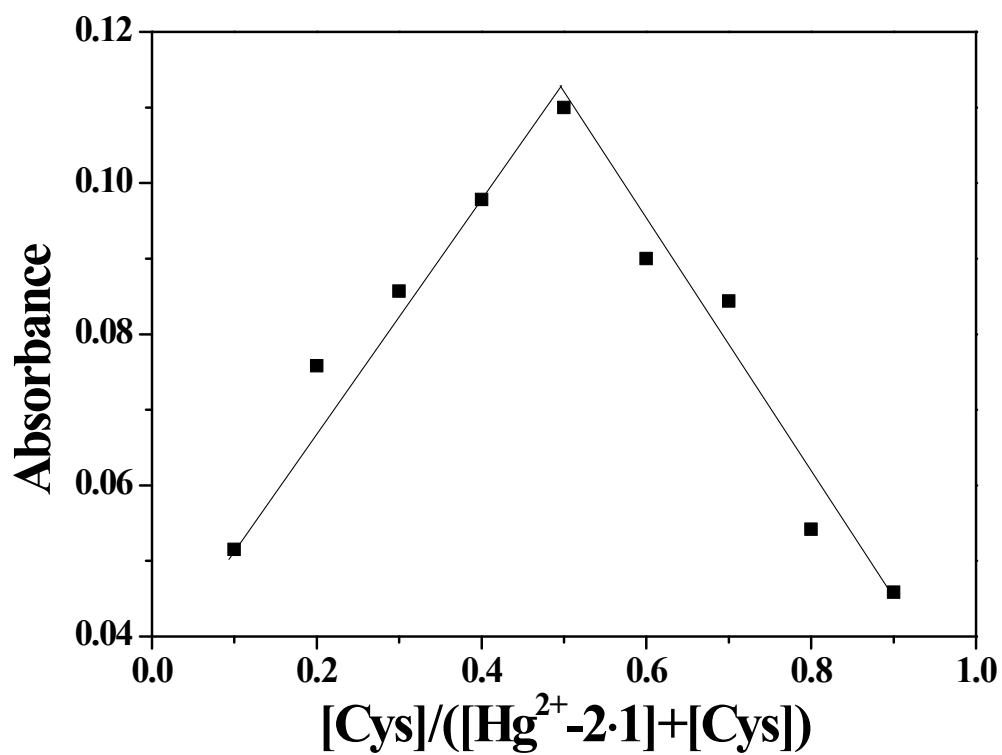


Fig. S7 Job plot for the binding of Hg²⁺-2·1 with Cys. Absorbance at 382 nm was plotted as a function of the molar ratio $[Cys]/([Hg^{2+}-2\cdot 1] + [Cys])$. The total concentrations of Cys with Hg²⁺-2·1 were 10 μ M.

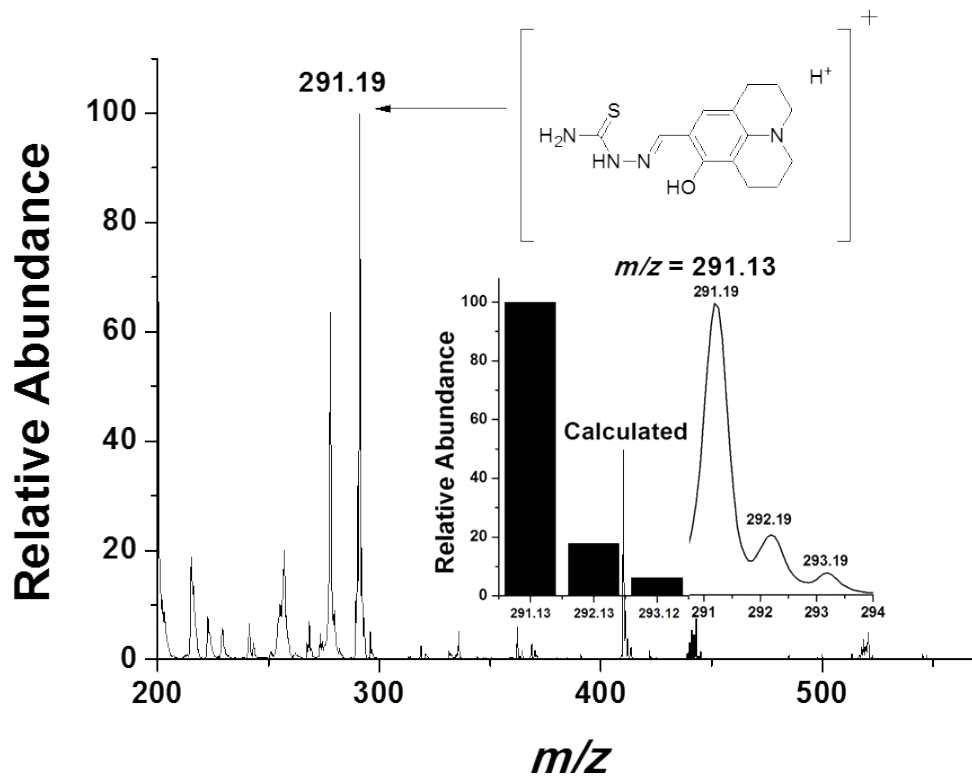


Fig. S8 Positive-ion electrospray ionization mass spectrum of $\text{Hg}^{2+}\text{-2}\cdot\mathbf{1}$ (100 μM) upon addition of 1 equiv of Cys.

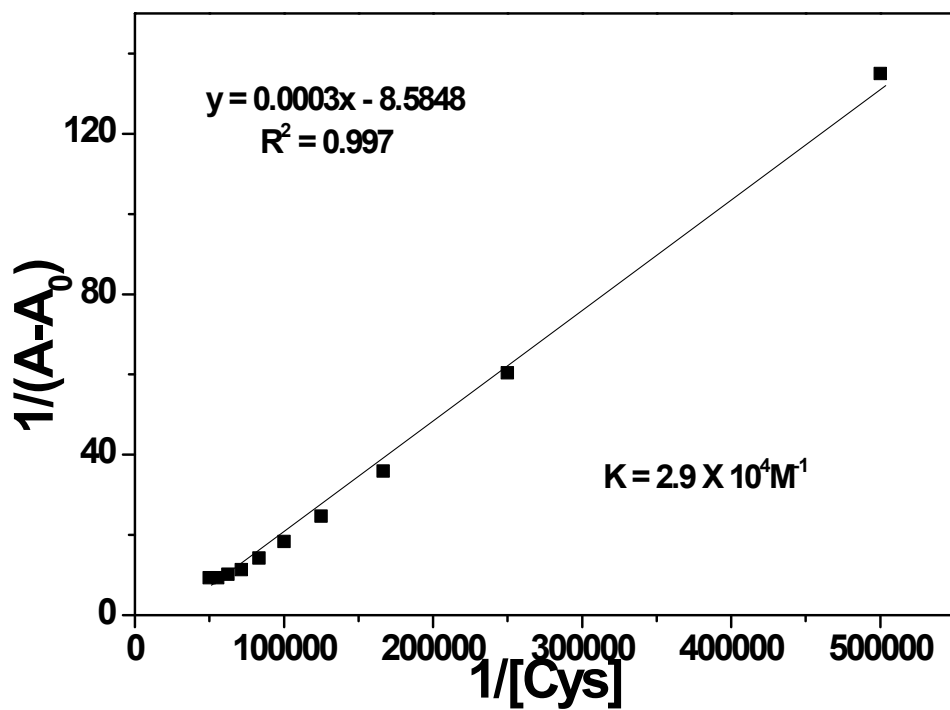


Fig. S9 Benesi-Hildebrand plot (absorbance at 382 nm) of **1**, assuming a 1:1 stoichiometry for association between **1** and Hg^{2+} .

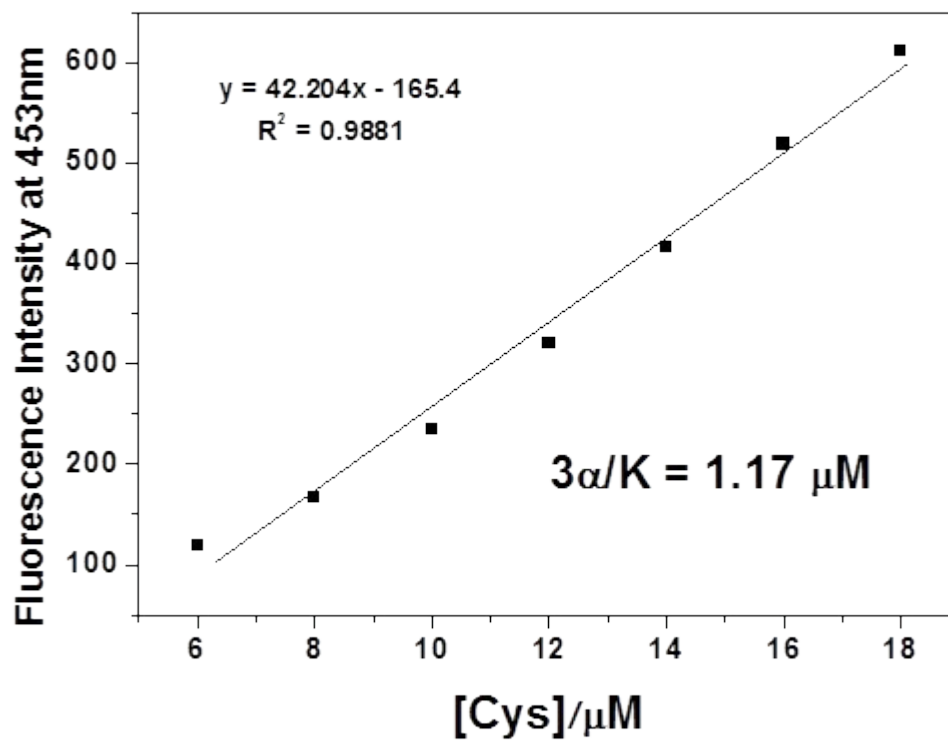


Fig. S10 Determination of the detection limit based on change in the ratio (absorbance at 453 nm) of Hg^{2+} -2·1 (10 μM) with Cys.

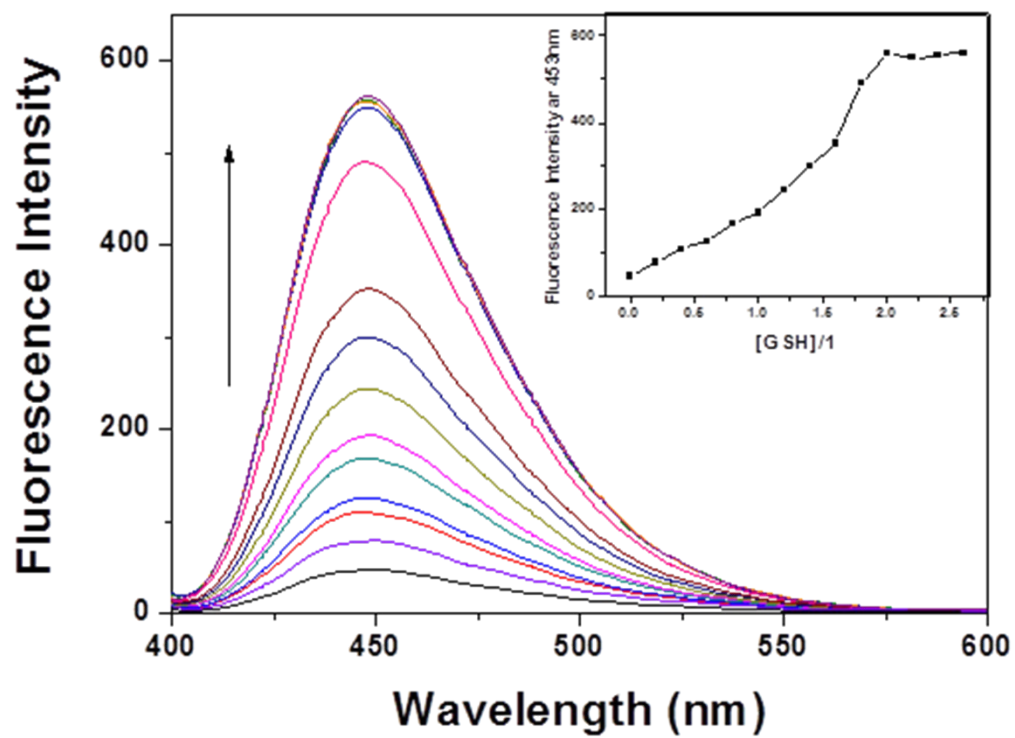


Fig. S11 Fluorescence spectral changes of Hg²⁺-2·1 in the presence of different concentrations of GSH in DMSO/bis-tris buffer (8/2, v/v). Inset: Fluorescence intensity at 453 nm versus the number of equiv of GSH added.

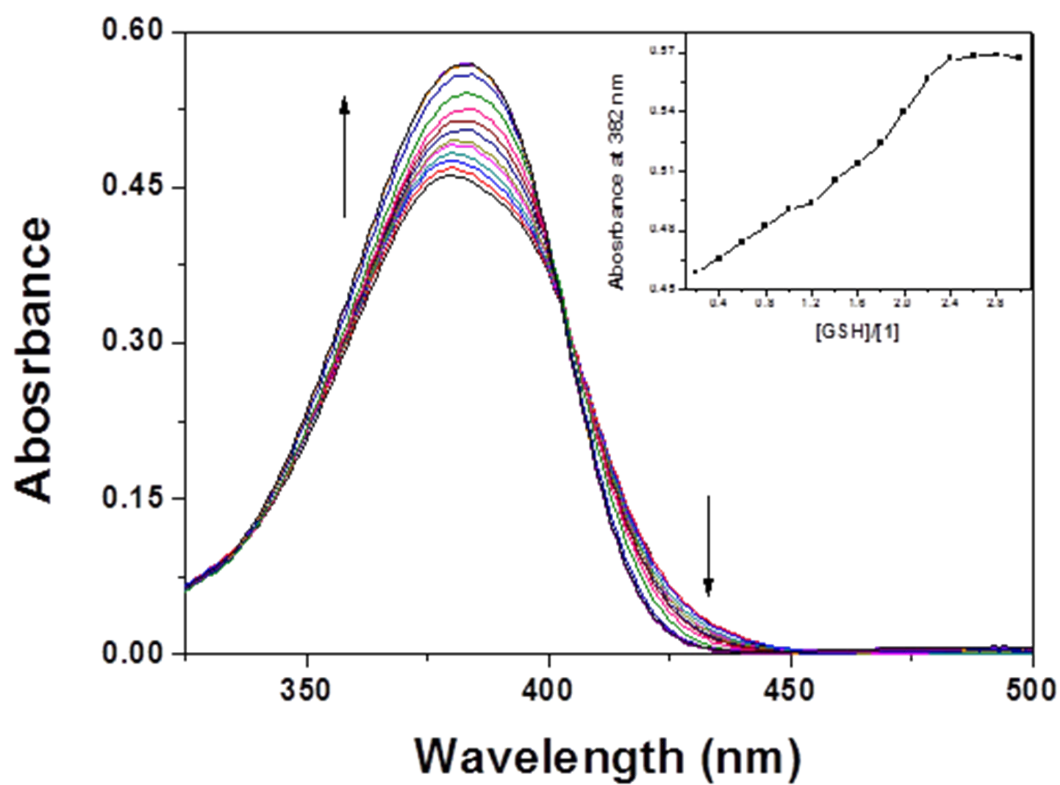


Fig. S12 Absorption spectral changes of $\text{Hg}^{2+}\text{-}2\cdot\mathbf{1}$ after addition of increasing amounts of GSH in DMSO/bis-tris buffer (8/2, v/v). Inset: Absorption at 382 nm versus the number of equiv of GSH added.

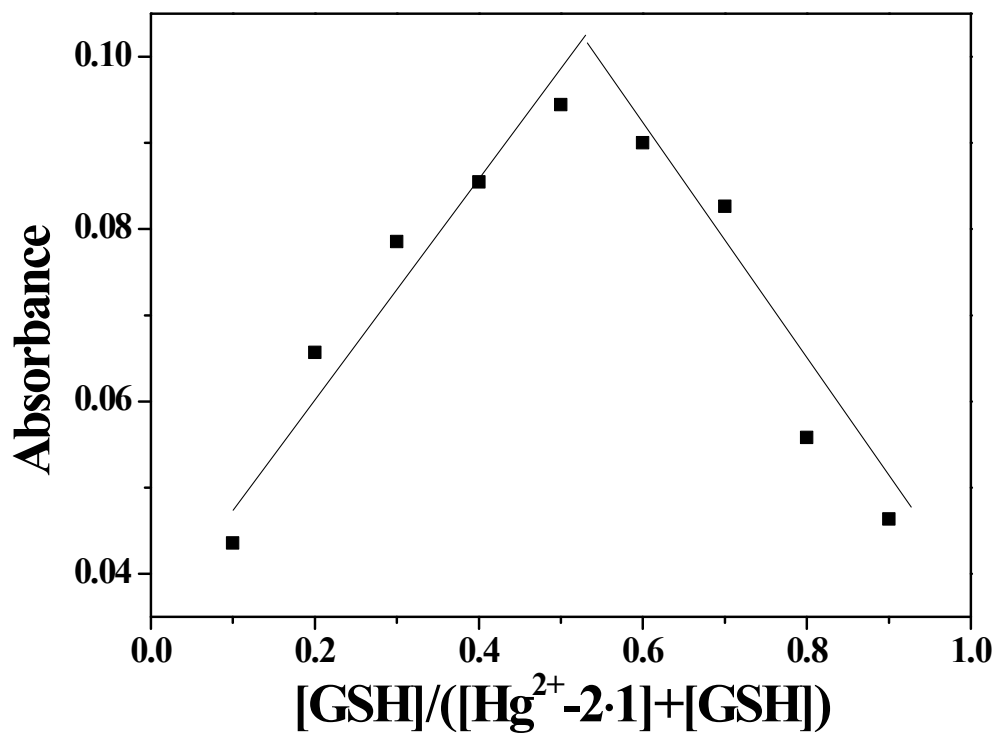


Fig. S13 Job plot for the binding of $\text{Hg}^{2+}\text{-}2\cdot\mathbf{1}$ with Cys. Absorbance at 382 nm was plotted as a function of the molar ratio $[\text{GSH}]/([\text{Hg}^{2+}\text{-}2\cdot\mathbf{1}] + [\text{GSH}])$. The total concentrations of Cys with $\text{Hg}^{2+}\text{-}2\cdot\mathbf{1}$ were 10 μM .

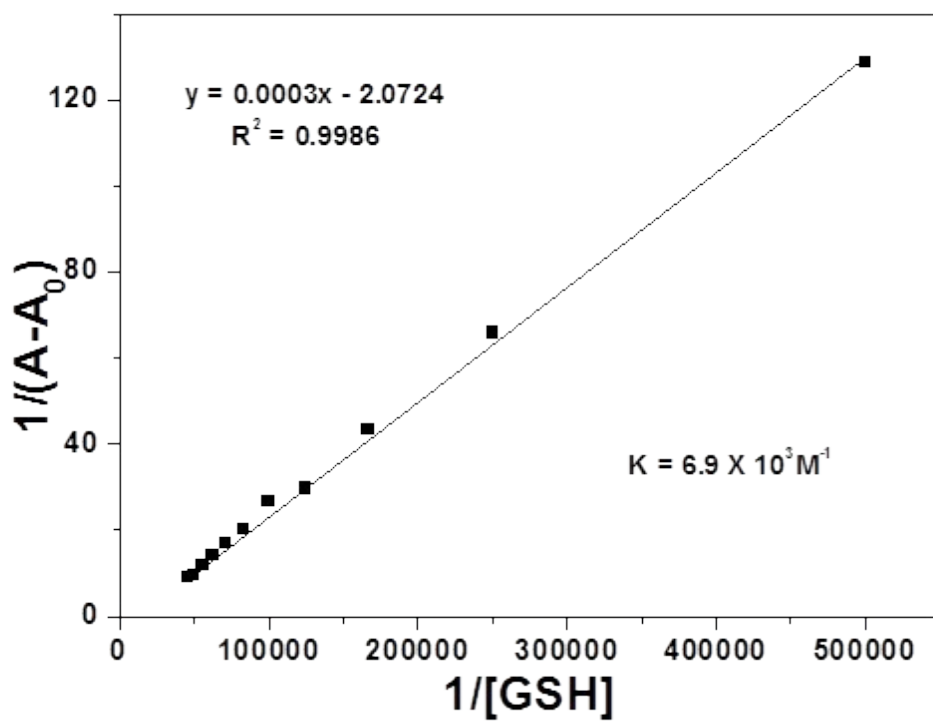


Fig. S14 Benesi-Hildebrand plot (absorbance at 382 nm) of **1**, assuming a 1:1 stoichiometry for association between $\text{Hg}^{2+} \cdot 2 \cdot \mathbf{1}$ and GSH.

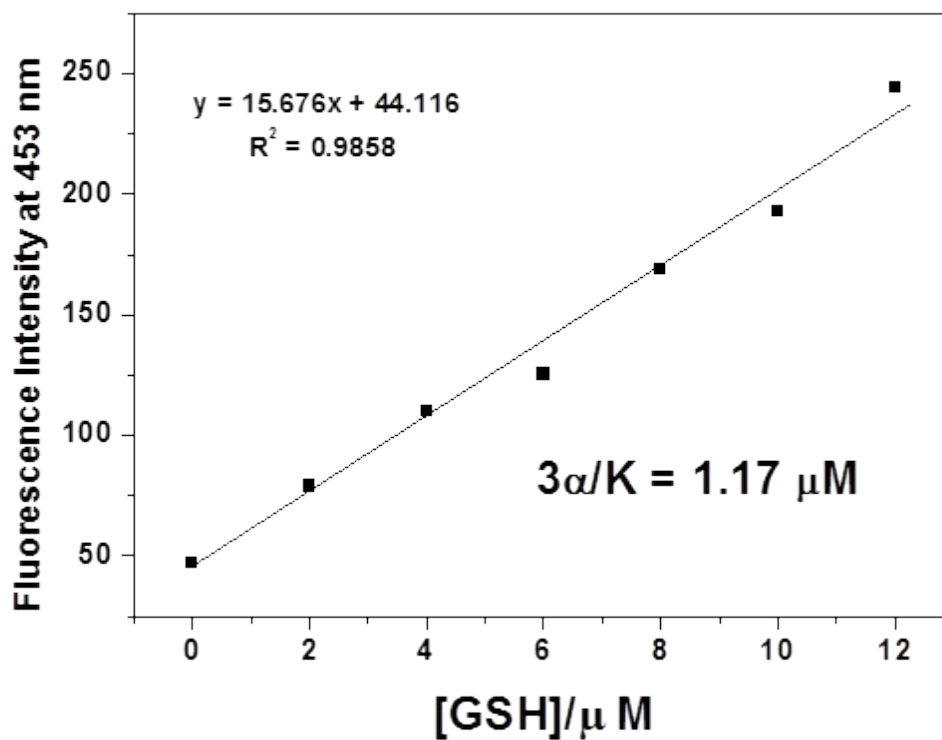


Fig. S15 Determination of the detection limit based on change in the ratio (absorbance at 453 nm) of Hg^{2+} -2·1 (10 μM) with GSH.

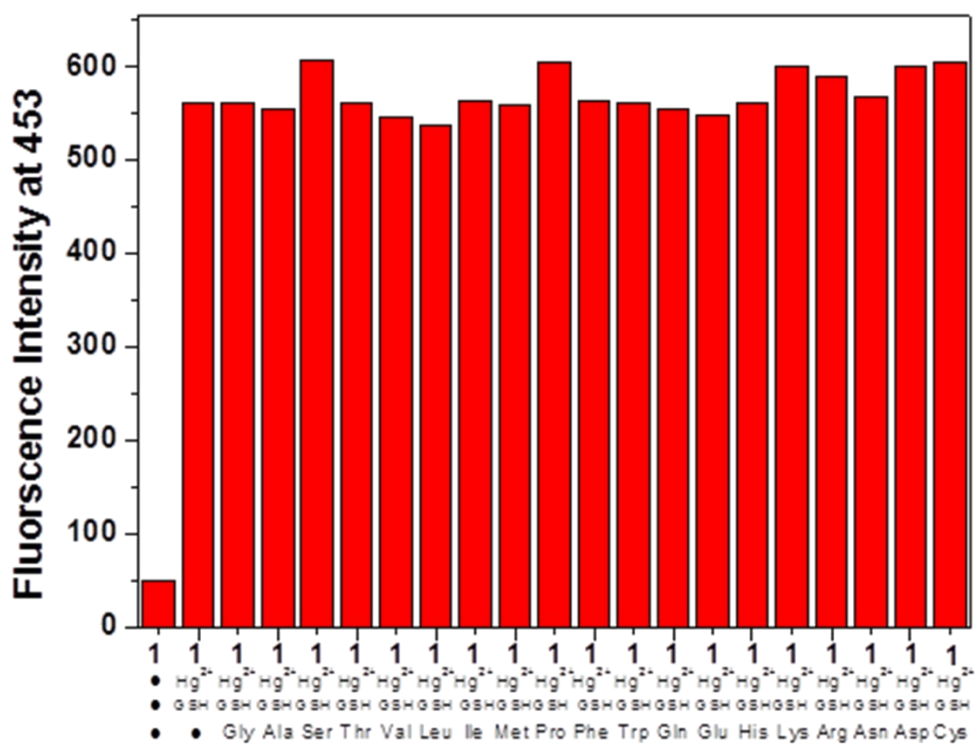
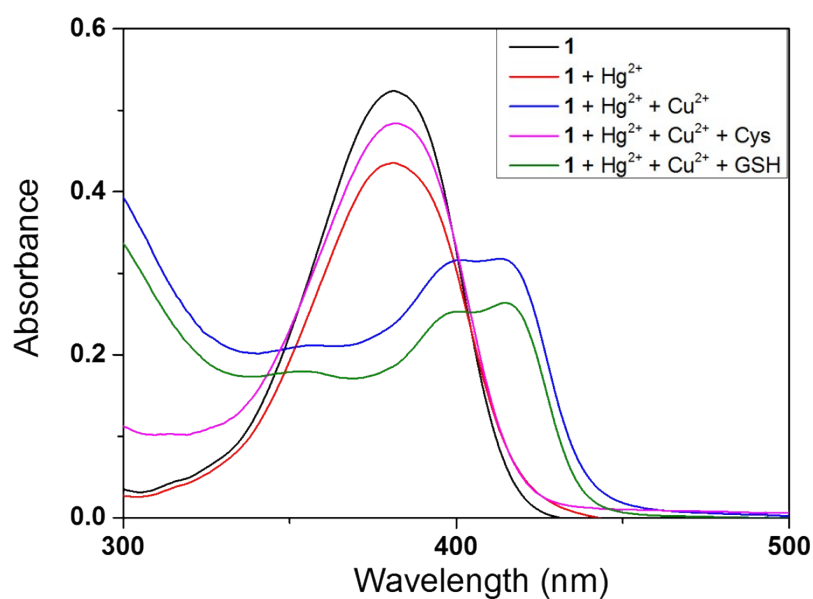


Fig. S16 Competitive selectivity of Hg²⁺-2·1 toward GSH (2 equiv) in the presence of other amino acids (2 equiv) in DMSO/bis-tris buffer (8/2, v/v).

(a)



(b)

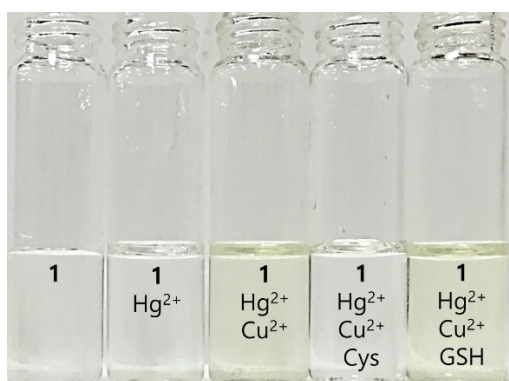
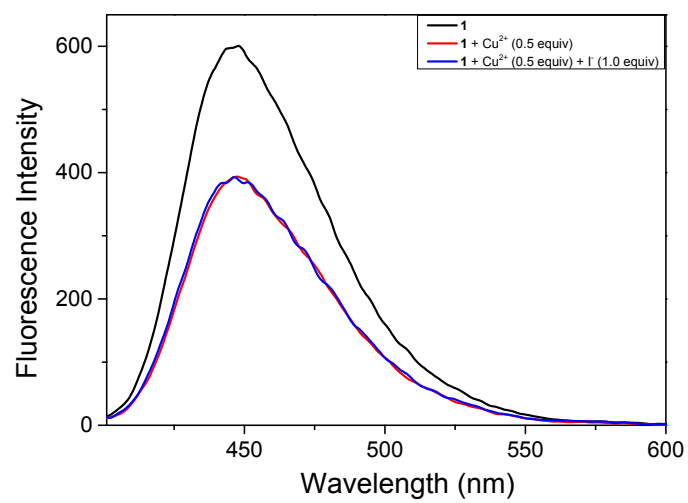


Fig. S17 (a) Changes in the UV-vis spectra of **1** (10 μ M), **1**-Hg²⁺ (1 equiv), **1**-Hg²⁺-Cu²⁺ (1 equiv), **1**-Hg²⁺-Cu²⁺ (1 equiv)-Cys (13 equiv), and **1**-Hg²⁺-Cu²⁺ (1 equiv)-GSH (13 equiv), respectively. (b) Color changes of **1** (10 μ M), **1**-Hg²⁺ (1 equiv), **1**-Hg²⁺-Cu²⁺ (1 equiv), **1**-Hg²⁺-

Cu^{2+} (1 equiv)-Cys (13 equiv), and $1\text{-Hg}^{2+}\text{-Cu}^{2+}$ (1 equiv)-GSH (13 equiv), respectively.

(a)



(b)

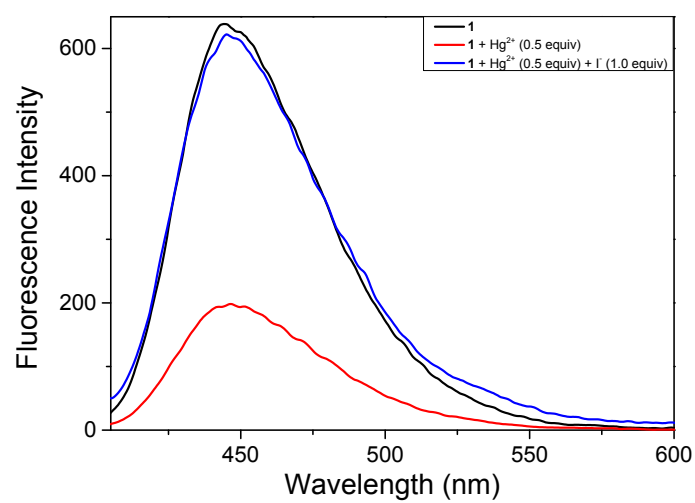


Fig. S18 Recovery tests of **1** with (a) **1**-Cu²⁺ (0.5 equiv) complex and (b) **1**-Hg²⁺ (0.5 equiv) complex in presence of I⁻ in DMSO/bis-tris buffer (8/2, v/v).