

**Shifting UV-Vis spectrum through rational structural modifications of zinc porphyrin  
photoactive compounds**

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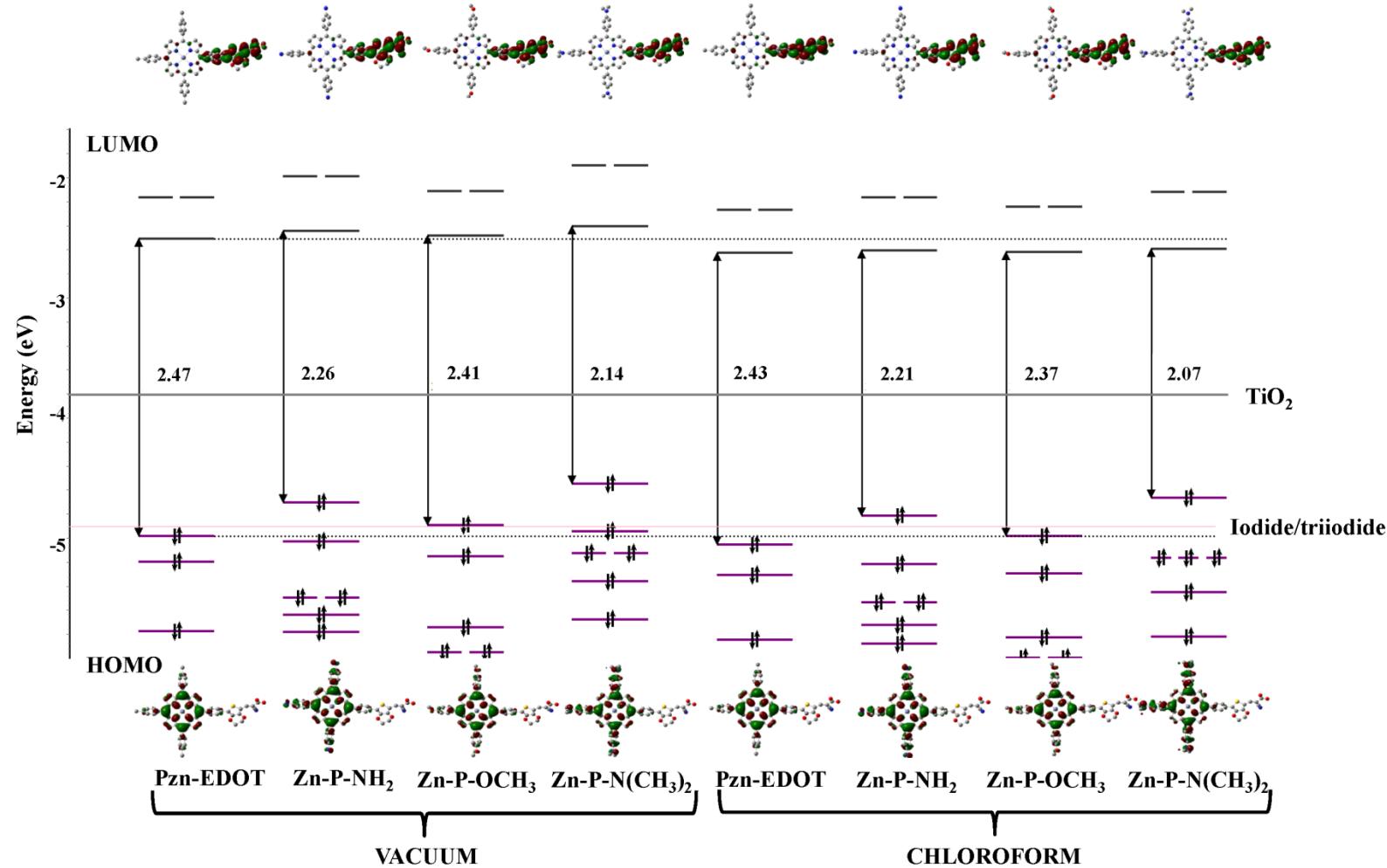
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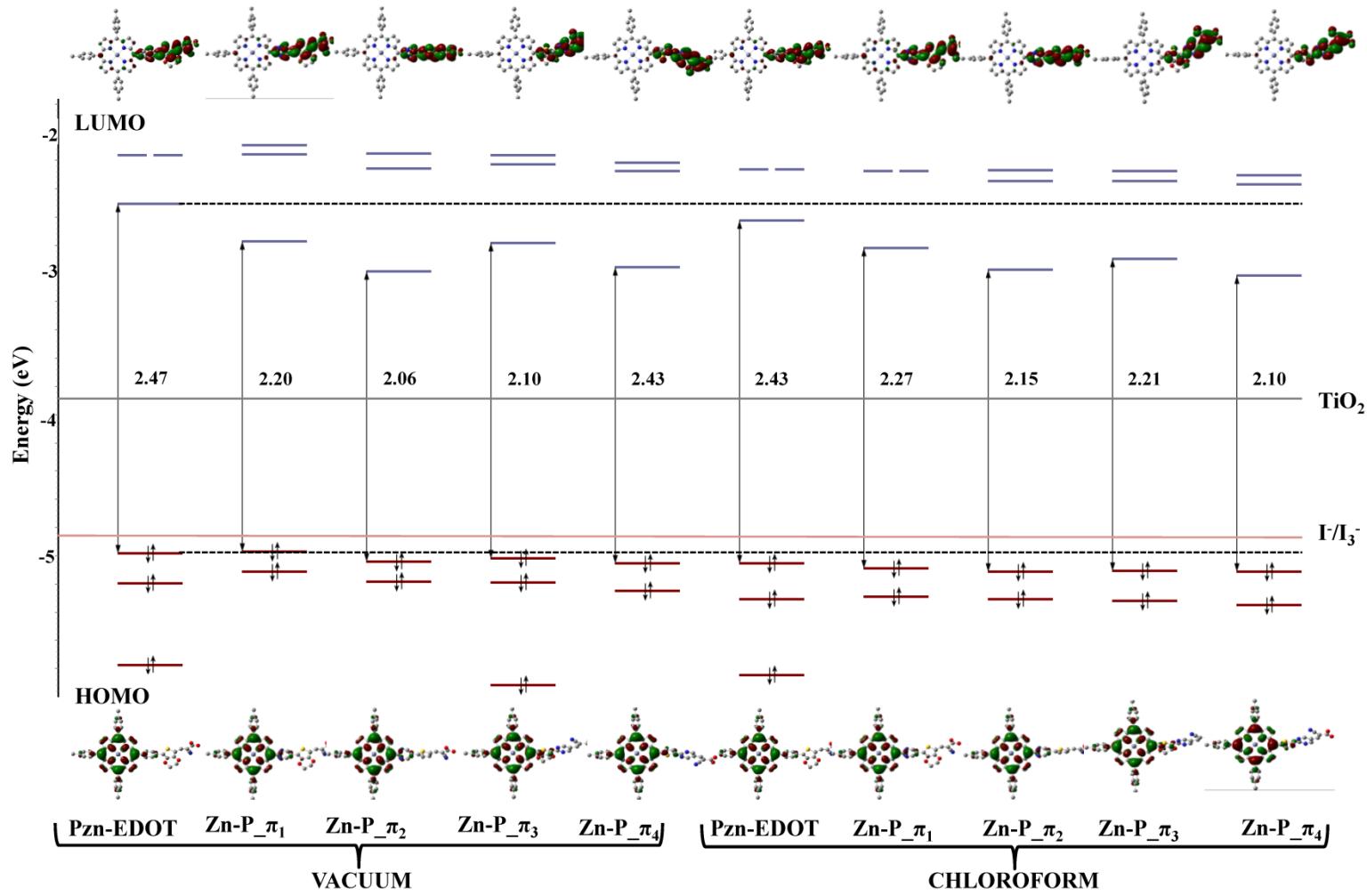
**Table S1. Selected Calculated Bond Lengths ( $\text{\AA}$ ) for the Four Oxidation States of the Porphyrin macrocycles (in Chloroform)**

The left diagram shows a porphyrin macrocycle with a central Zn atom coordinated to four nitrogen atoms (N<sup>3</sup>, N<sup>4</sup>, N<sup>11</sup>, N<sup>14</sup>). It features four phenyl groups at the 5, 8, 11, and 14 positions. Substituents include a methyl group at C<sup>10</sup> and a phenyl group at C<sup>15</sup>. Carbons are labeled with Greek subscripts α, β, and γ. The right diagram shows a substituted acrylate group: R'-CH=CH-C(=O)-C(=O)OCH<sub>2</sub>-N<sub>cy</sub> with carbons labeled α, β, and γ.

Bond	Bond length ( $\text{\AA}$ )				
	Chloroform				
	Pzn-EDOT	Zn-P- $\pi_1$	Zn-P- $\pi_2$	Zn-P- $\pi_3$	Zn-P- $\pi_4$
C <sub>R</sub> -C <sub>m</sub> <sup>5</sup>	1.495	1.493	1.492	1.481	1.481
C <sub>m</sub> <sup>5</sup> -C <sub>α</sub> <sup>6</sup>	1.409	1.410	1.410	1.409	1.412
C <sub>α</sub> <sup>6</sup> -C <sub>β</sub> <sup>7</sup>	1.446	1.447	1.448	1.446	1.446
C <sub>β</sub> <sup>7</sup> -C <sub>β</sub> <sup>8</sup>	1.361	1.359	1.359	1.360	1.360
C <sub>β</sub> <sup>8</sup> -C <sub>α</sub> <sup>9</sup>	1.445	1.446	1.446	1.446	1.445
C <sub>α</sub> <sup>9</sup> -C <sub>m</sub> <sup>10</sup>	1.408	1.405	1.405	1.406	1.406
C <sub>m</sub> <sup>10</sup> -C <sub>α</sub> <sup>11</sup>	1.409	1.411	1.411	1.411	1.410
C <sub>α</sub> <sup>11</sup> -C <sub>β</sub> <sup>12</sup>	1.445	1.444	1.444	1.444	1.445
C <sub>β</sub> <sup>12</sup> -C <sub>α</sub> <sup>13</sup>	1.361	1.362	1.362	1.362	1.362
C <sub>β</sub> <sup>13</sup> -C <sub>α</sub> <sup>14</sup>	1.445	1.444	1.444	1.444	1.445
C <sub>α</sub> <sup>14</sup> -C <sub>m</sub> <sup>15</sup>	1.409	1.409	1.408	1.409	1.408
C <sub>α</sub> <sup>6</sup> -N <sup>3</sup>	1.376	1.374	1.374	1.374	1.374
C <sub>α</sub> <sup>9</sup> -N <sup>3</sup>	1.378	1.380	1.380	1.379	1.379
C <sub>α</sub> <sup>11</sup> -N <sup>4</sup>	1.377	1.375	1.375	1.376	1.376
C <sub>α</sub> <sup>14</sup> -N <sup>4</sup>	1.377	1.378	1.378	1.377	1.378
N <sup>3</sup> -Zn	2.051	2.050	2.051	2.050	2.052
N <sup>4</sup> -Zn	2.049	2.051	2.051	2.047	2.050
C <sub>R</sub> -C <sub>ac</sub> <sup>α</sup>	1.419	1.422	1.428	1.440	1.443
C <sub>ac</sub> <sup>α</sup> -C <sub>ac</sub> <sup>β</sup>	1.376	1.373	1.370	1.365	1.493
C <sub>ac</sub> <sup>β</sup> -C <sub>ac</sub> <sup>γ</sup>	1.487	1.490	1.487	1.491	1.426
C <sub>ac</sub> <sup>β</sup> -C <sub>Cy</sub>	1.424	1.425	1.424	1.426	1.648
C <sub>Cy</sub> -N <sub>Cy</sub>	1.165	1.165	1.165	1.165	1.215
C <sub>ac</sub> <sup>γ</sup> -O <sup>1</sup>	1.219	1.218	1.216	1.216	1.346
C <sub>ac</sub> <sup>γ</sup> -O <sup>2</sup>	1.351	1.350	1.348	1.348	



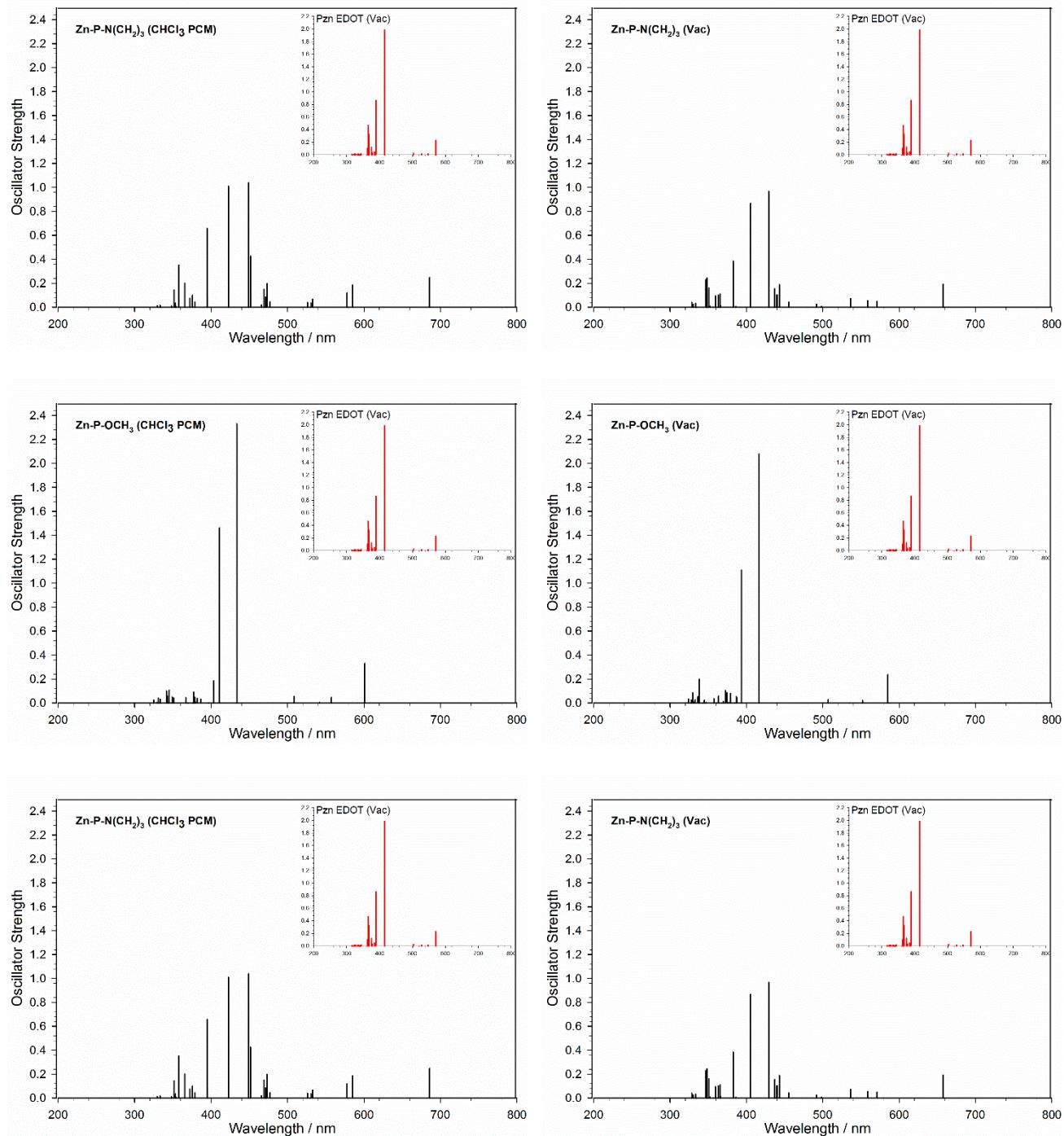
**Figure S2.** Single point calculated frontier MO (Molecular orbitals) using B3LYP/6-31G\* model in vacuum and Chloroform along with the comparison of charge density of HOMOs and LUMOs with the reference dye Pzn-EDOT and *meso*-substituted zinc porphyrin dyes.



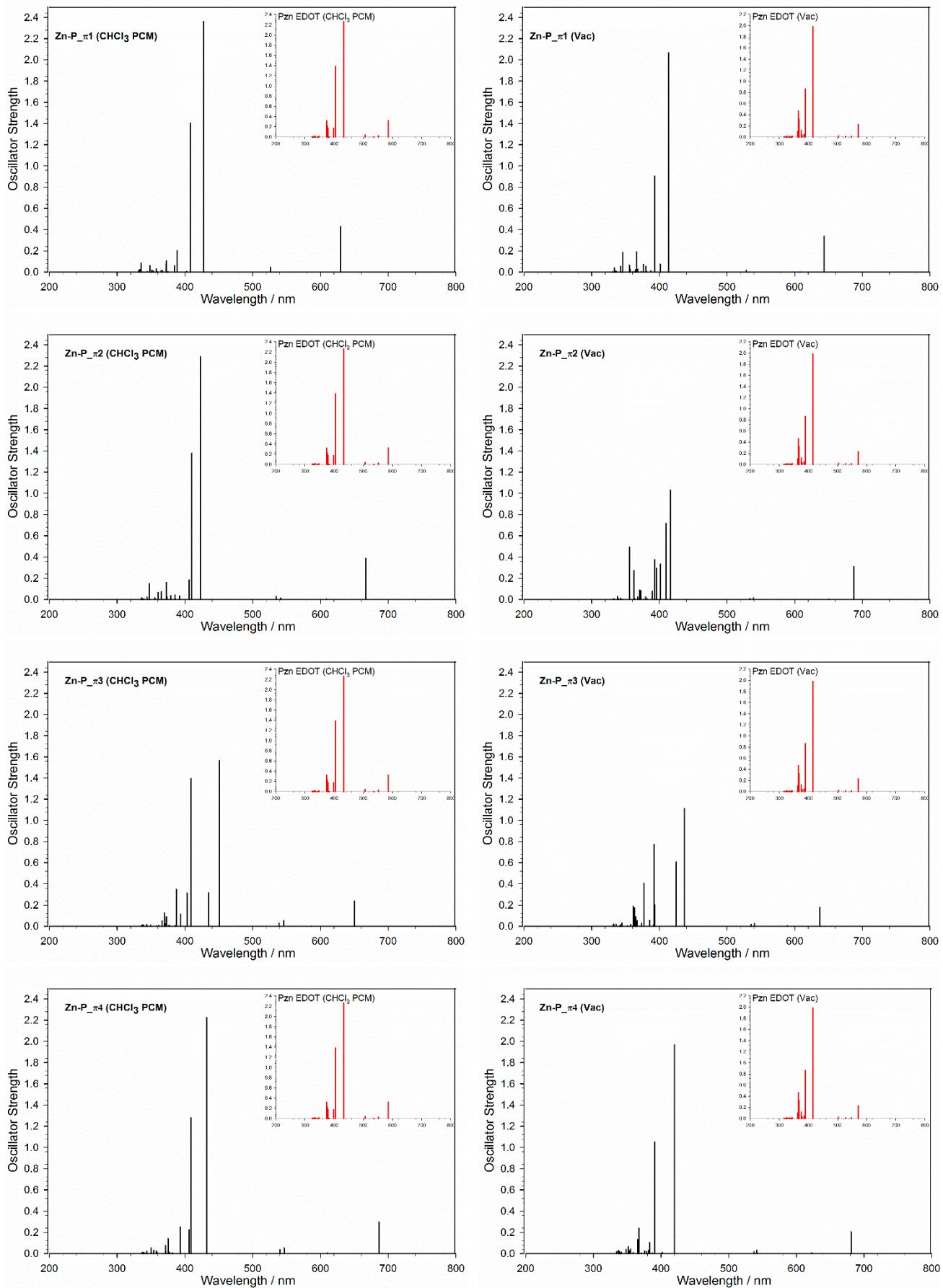
**Fig. S3** Single point calculated frontier MO (Molecular orbitals) using B3LYP/6-31G\* model in vacuum and Chloroform along with the comparison of charge density of HOMOs and LUMOs with the reference dye Pzn-EDOT and  $\pi$ -substituted zinc porphyrin dyes.

CHCl3

Vacuum



**Fig. S4.** Calculated excitations of three examples of meso-substituted porphyrins (black vertical lines) in polarised continuum and vacuum. The y axes for are scaled to the maximum oscillator strength for the examples studied. Inset: calculated excitations for the parent EDOT porphyrin (red vertical lines).



**Fig. S5.** Calculated excitations of four examples of Pi Bridge porphyrins (black vertical lines) in polarised continuum and vacuum. The y axes for are scaled to the maximum oscillator strength for the examples studied. Inset: calculated excitations for the parent EDOT porphyrin (red vertical lines).