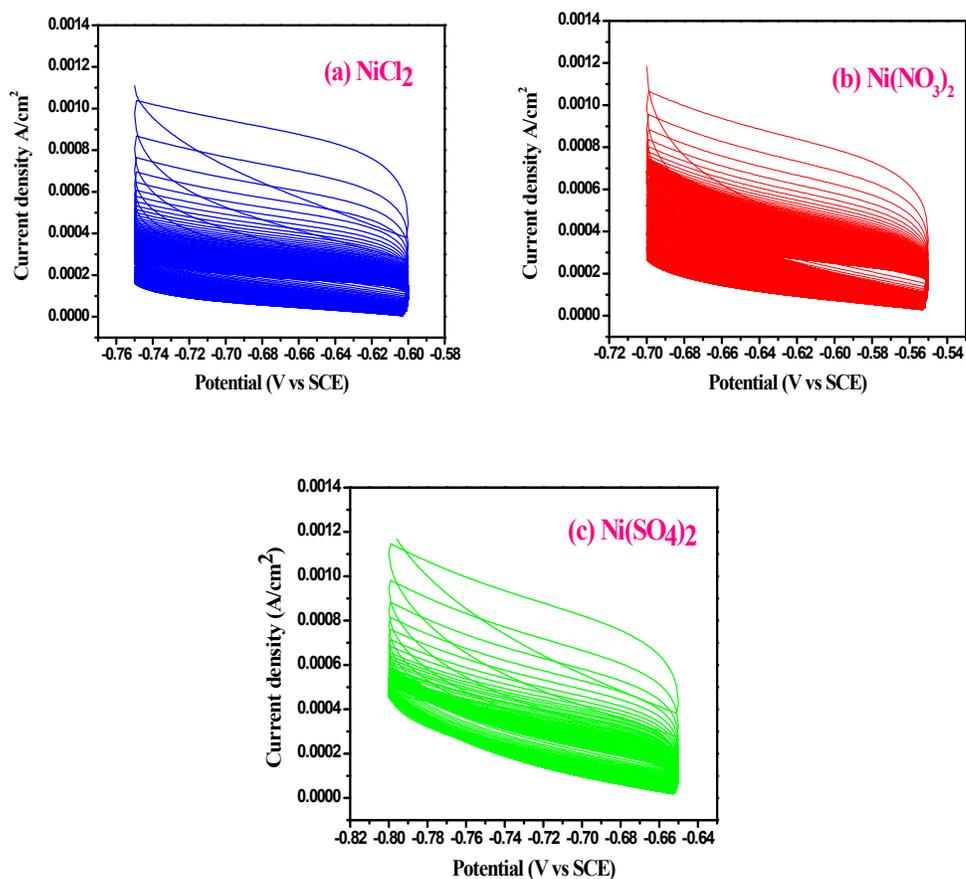


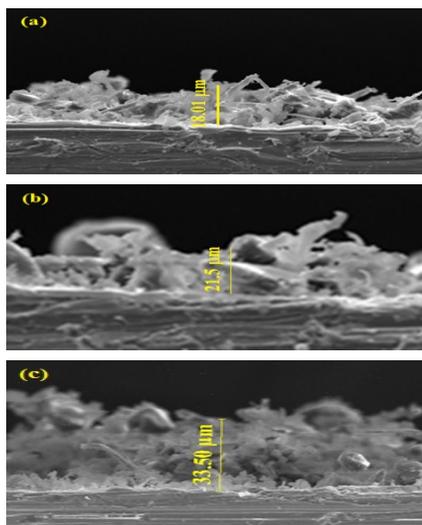
## Electronic Supporting Information

### Nanomorphology-dependent pseudocapacitive properties of NiO electrodes engineered through controlled potentiodynamic electrodeposition process

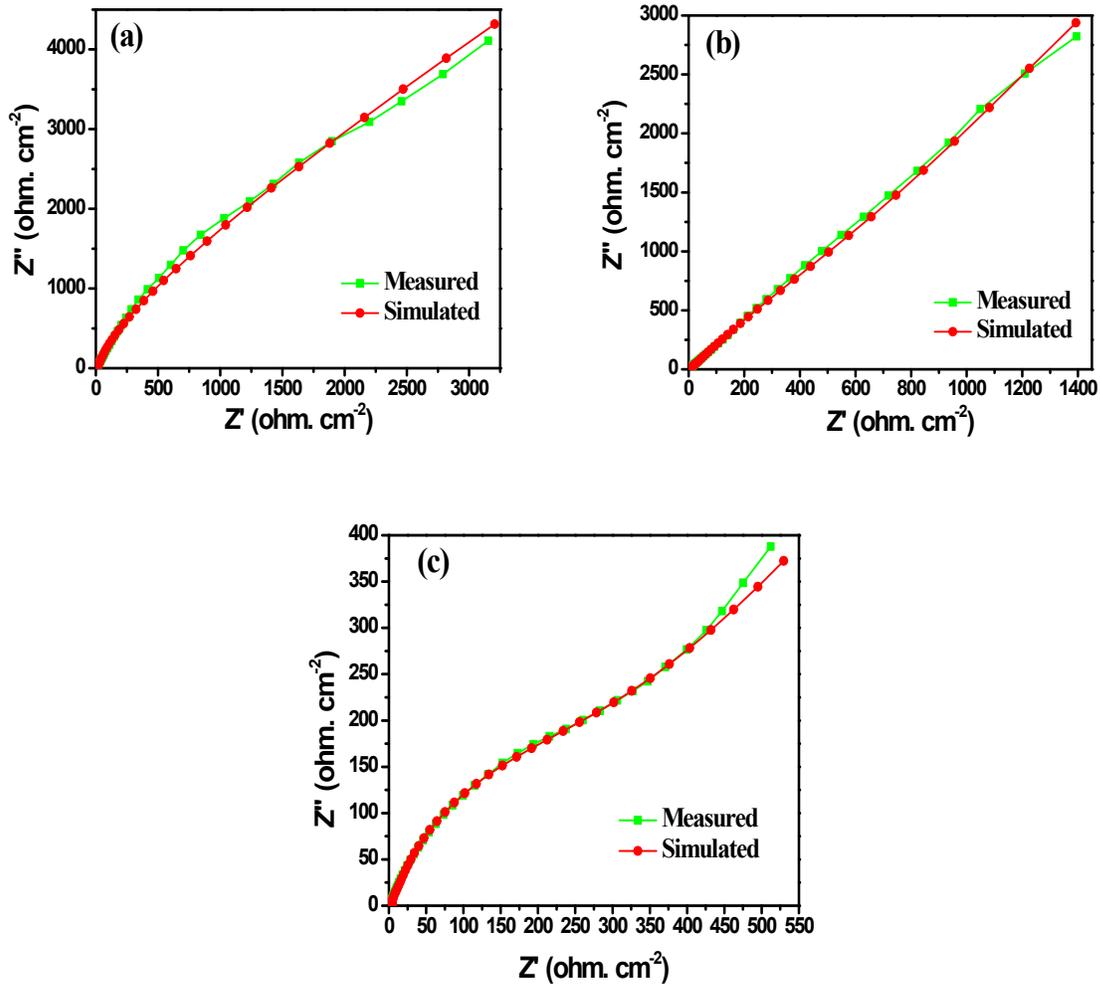
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**Figure S1:** Cyclic voltammetry curves at  $100 \text{ mV}\cdot\text{s}^{-1}$  for 30 minutes during the potentiodynamic electrodeposition for different precursors (a) Nickel chloride ( $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ ), (b) Nickel nitrate ( $\text{Ni}(\text{NO}_3)_2$ ) and (c) Nickel sulphate ( $\text{Ni}(\text{SO}_4)_2$ ).



**Figure S2:** Cross sectional SEM images of (a) NiO-C, (b) NiO-N and (c) NiO-S samples.



**Figure S3:** The measured and simulated data for matched equivalent circuit for (a) NiO-N, (b) NiO-C and (c) NiO-S electrodes, which indicates a very good agreement between measured and simulated data for matched equivalent circuit.

**Table S1:** Electrochemical circuitry parameters obtained from EIS study for matched equivalent circuit.

Parameters	Electrode Type		
	NiO-N	NiO-C	NiO-S
$R_s(\Omega/\text{cm}^2)$	1.34	1.29	1.27
$R_1(\Omega/\text{cm}^2)$	21.11	10.17	4.35
$R_2(\Omega/\text{cm}^2)$	0.28	0.21	0.10
$R_3(\Omega/\text{cm}^2)$	0.001	0.001	0.001
CPE-1( $\mu\text{F}$ )	0.227	0.349	0.455
CPE-2( $\mu\text{F}$ )	0.3427	0.533	0.841
Q	0.000158	0.000258	0.000961
W	0.000144	0.000207	0.000145