Supplementary Information

Hydrothermal Self-Assembly of Graphene Foams with Controllable Pore Size

Wei Deng, aQile Fang, Xufeng Zhou, *a Hailiang Caoa and Zhaoping Liu*a

^aNingbo Institute of Materials Technology and Engineering (NIMTE), Chinese Academy of Sciences (CAS), Ningbo 315201, P. R. China.E-mail: zhouxf@nimte.ac.cn (X. Zhou), liuzp@nimte.ac.cn (Z. Liu).



Figure S1. Optical photographs of aqueous dispersions of L-GOs (a), M-GOs(b) and S-GOs (c).



Figure S2. Polarized-light optical microscopy (POM) images of aqueous dispersions of L-GOs (a), M-GOs (b) and S-GOs (c).



Figure S3. Optical photographs of L-GF (a), M-GF(b) and S-GF (c) prepared from 4 mg mL⁻¹ GOs suspensions for 12 h.



Figure S4. SEM images of interior pores taken along horizontal (a, c, e) and vertical (b, d, f) directions of the as-prepared L-, M- and S-GFs.



Figure S5. Optical photographs of the products prepared from 0.5 mgmL⁻¹ (a), 1.0 mgmL⁻¹ (b) and 2.0 mgmL⁻¹(c) S-GOs suspensions.



Figure S6. SEM images of the products prepared from 0.5 mg mL⁻¹ (a), 1.0 mg mL⁻¹

(b) and 2.0 mg mL⁻¹ (c) S-GOs suspensions.



Figure S7. (a) SEM image of graphene foam prepared by a mixture of L-GOs and S-GOs with a mass ratio of L-GOs: S-GOs= 1:5. (b, c and d) SEM images with different magnifications of graphene foam prepared by a mixture of L-GOs and S-GOs with a mass ratio of L-GOs: S-GOs = 1:2.



Figure S8. The corresponding relationship between size of GO sheets and pore sizes of GFs.

| determined by XPS. | | | | | | | |
|--------------------|-------------------------------|-------|-------|-------------------|------|------|--------|
| Sample | Elemental compositions (wt %) | | | C1s deconvolution | | | |
| | C/O | С | 0 | C-C | C-0 | С=О | C(O)OH |
| L-rGO | 4.87 | 82.95 | 17.05 | 64.9 | 13.4 | 9.5 | 12.2 |
| M-rGO | 4.69 | 82.42 | 17.58 | 64.3 | 12.7 | 9.8 | 13.2 |
| S-rGO | 4.84 | 82.87 | 17.13 | 64.6 | 13.2 | 10.2 | 12.0 |

Table S1. C and O contents and speciation in the (L-, M- and S-) rGO sheets