Supplementary Material

Microwave assisted non-solvothermal synthesis of metal-organic frameworks

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Effect of solvent



Fig. S1 The PXRD patterns of samples **2**, **5** and **6** compared with simulated diffraction patterns from the crystallographic data.¹

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Fig. S3 Adsorption (filled symbol) and desorption (empty symbol) isotherms for nitrogen at (-196)^oC K of HKUST-1 samples **2**, **5** and **6**.



Fig. S4 TGA traces at 10°C/min of samples 2, 5 and 6.



Fig. S5 The PXRD patterns of samples 7, 3 and 8 compared with simulated diffraction patterns from the crystallographic data.¹



Fig. S6 The FT-IR spectra of samples 3 and 7.



Fig. S7 SEM micrograph of sample 7.



Fig. S8 Nitrogen adsorption (filled symbol)/desorption (empty symbol) isotherms of samples 7, 3 and 8 at (-196)°C.



Fig. S9 The TGA (left) and DTG (right) traces of samples 7, 3 and 8 at 10°C/min.

Effect of reaction time



Fig. S10 The PXRD patterns of samples **9**, **10**, **3** and **11** compared with simulated diffraction patterns from the crystallographic data.¹



Fig. S11 The FT-IR spectra of samples 9, 10 and 3.



Fig. S12 SEM micrographs of samples 9 and 10.



Fig. S13 Adsorption (filled symbol) and desorption (empty symbol) isotherms for nitrogen at (-196)°C of HKUST-1 samples 9, 3, 10 and 11.



Fig. S14 The FT-IR spectra of samples 14 and 3.



Fig. S15 Adsorption (filled symbol) and desorption (empty symbol) isotherms for nitrogen at (-196)°C of HKUST-1 samples 3, 12, 13 and 14.



Fig. S16 TGA (left) and DTG (right) traces at 10°C/min of samples 3, 12, 13 and 14.

Mother liquors



Fig. S17 The FT-IR spectra of samples 14. 14(2) and 14(4).



Fig. S18 Adsorption (filled symbol) and desorption (empty symbol) isotherms for nitrogen at (-196)°C of HKUST-1 samples **14(2)** and **14(4)**.



Fig. S19 The comparison between TGA (left) and DTG (right) traces at 10°C/min of samples 14, 14(2) and 14(4).

¹S.S.Y. Chui, S.M.F. Lo, J.P.H. Charmant, A. Guy Orpen, I.D. Williams, *Science*, 1999, 283, 1148-1150.