

**Elucidation of hydrogen bonding formation by computational, FT-IR
Spectroscopic and theoretical study between benzylalcohol with isomeric
cresols**

Supplementary Information

Table S1:

Standard deviation $\sigma(V^E)$ and values of constants ($a_0, a_1, a_2; b_0, b_1, b_2$) for Redlich-Kister, Eq. (2) and Hwang et al., Eq. (3) for benzylalcohol (1) with *o,m,p*-cresols(2)

Temperature	Redlich-Kister				Hwang			
	a_0	a_1	a_2	$\sigma(V^E)$	b_0	b_1	b_2	$\sigma(V^E)$
	<hr/>				<hr/>			
	$\text{cm}^{-3} \cdot \text{mol}^{-1}$				$\text{cm}^{-3} \cdot \text{mol}^{-1}$			
Benzylalcohol(1) + <i>o</i> -cresol (2)								
303.15K	-0.195	0.003	0.053	0.001	-0.212	0.074	0.067	0.001
308.15K	-0.215	0.003	0.046	0.001	-0.229	0.065	0.058	0.001
313.15K	-0.234	-0.005	0.035	0.001	-0.245	0.041	0.050	0.001
318.15K	-0.268	-0.008	0.032	0.001	-0.278	0.033	0.050	0.001
323.15K	-0.325	-0.003	0.044	0.001	-0.339	0.056	0.061	0.001
Benzylalcohol(1)+ <i>m</i> -cresol (2)								
303.15K	-0.085	-0.008	-0.015	0.001	-0.080	-0.030	-0.010	0.001
308.15K	-0.089	-0.004	-0.048	0.001	-0.073	-0.069	-0.059	0.001
313.15K	-0.094	-0.004	-0.063	0.001	-0.073	-0.089	-0.079	0.001
318.15K	-0.106	-0.005	-0.066	0.001	-0.084	-0.094	-0.082	0.001

323.15K	-0.113	-0.001	-0.083	0.001	-0.085	-0.112	-0.109	0.001
Benzylalcohol (1)+ <i>p</i> -cresol (2)								
303.15K	-0.009	-0.090	0.014	0.001	-0.013	-0.088	0.126	0.001
308.15K	-0.003	-0.096	0.041	0.001	-0.017	-0.060	0.170	0.001
313.15K	-0.0003	-0.096	0.053	0.001	-0.018	-0.043	0.186	0.001
318.15K	0.010	-0.099	0.063	0.001	-0.011	-0.034	0.202	0.001
323.15K	0.023	-0.109	0.076	0.001	-0.002	-0.029	0.231	0.001

Table S2:

Standard deviation σ (κ_s^E) and values of constants ($a_0, a_1, a_2; b_0, b_1, b_2$) for Redlich–Kister and Hwang et al. for excess isentropic compressibility data.

Temp.	Function	Redlich-Kister				Hwang			
		a_0	a_1	a_2	σ (κ_s^E)	b_0	b_1	b_2	σ (κ_s^E)
		TPa ⁻¹				TPa ⁻¹			
Benzylalcohol(1)+ <i>o</i> -cresol (2)									
303.15K	κ_s^E	1.150	0.052	0.119	0.004	1.110	0.221	0.097	0.004
313.15K	κ_s^E	1.040	0.030	0.011	0.003	1.036	0.050	-0.021	0.003
Benzylalcohol(1)+ <i>m</i> -cresol (2)									
303.15K	κ_s^E	-2.248	0.035	0.035	0.002	-2.260	0.088	0.005	0.002
313.15K	κ_s^E	-2.590	0.020	-0.199	0.004	-2.524	-0.242	-0.289	0.004
Benzylalcohol(1)+ <i>p</i> -cresol (2)									
303.15K	κ_s^E	-4.056	-0.322	-0.007	0.004	-4.054	-0.393	0.374	0.004
313.15K	κ_s^E	-4.676	-0.413	0.048	0.002	-4.692	-0.428	0.556	0.003

Table S3:

Distances (Å) and angles ($^\circ$) of the hydrogen bonds for all hydrogen bond associations at the B3LYP/6-311++G (d, p).

Complexes	X-H...Y	R(X-H)	ΔR (X-H)	R(H...Y)	ΔR (H...Y)	R(X...Y)	\angle XHY
I	O2-H16...O1	0.9732	0.0115	1.8812	0.8388	2.8392	167.535
II	O2-H16...O1	0.9707	0.0080	1.9322	0.7878	2.8854	166.676
III	O2-H16...O1	0.9704	0.0081	1.9161	0.8040	2.8820	173.273
IV	O2-H16...O1	0.9706	0.0078	1.9163	0.8037	2.8837	174.334
V	O1-H8...O2	0.9687	0.0070	2.0369	0.6831	2.9537	157.164
Va	O2-H13...O1	0.9751	0.0125	1.8434	0.8766	2.8150	174.002
VI	O2-H13...O1	0.9687	0.0070	2.0363	0.6838	2.9496	156.413
VIa	O2-H13...O1	0.9751	0.0129	1.8453	0.8748	2.8149	172.504
VII	O1-H8...O2	0.9689	0.0072	1.9686	0.7514	2.9263	169.337
VIIa	O2-H13...O1	0.9748	0.0121	1.8596	0.8604	2.8223	168.798

Table S4:

Topological Properties (in a. u.) of the BCPs of intermolecular H-bonds in Benzyl alcohol and cresol complexes obtained from the B3LYP/6-311++ G (d, p) level calculations

Complexes	X-H...Y	ρ	$\nabla^2\rho$	H(r)	$ V(r) /G(r)$
I	O2-H16...O1	0.028676	0.103512	0.001596	0.93427
II	O2-H16...O1	0.024337	0.091802	0.002273	0.89008
III	O2-H16...O1	0.024804	0.093081	0.002283	0.88672
IV	O2-H16...O1	0.025173	0.095618	0.002287	0.89421
V	O1-H8...O2	0.021904	0.082080	0.001517	0.92017
Va	O2-H13...O1	0.031167	0.109377	0.000018	0.99921
VI	O2-H13...O1	0.022112	0.083770	0.001680	0.91159
Via	O2-H13...O1	0.030992	0.105946	0.001011	0.96161
VII	O1-H8...O2	0.022417	0.086099	0.002519	0.86746
VIIa	O2-H13...O1	0.030107	0.091774	0.001139	0.95507

Table S5:

Interaction energy corrected with BSSE (ΔE . kJ. mol⁻¹) for all dimers at the B3LYP/6-311++G(d, p) level.

Complexes	X-H...Y	ΔE	BSSE	ΔE^{cp}
I	O2-H16...O1	-24.3720	0.73	-23.6420
II	O2-H16...O1	-20.3062	0.76	-19.5462
III	O2-H16...O1	-20.7755	0.81	-19.9655
IV	O2-H16...O1	-17.5275	0.80	-16.7275
V	O1-H8...O2	-25.2681	0.86	-24.4081
Va	O2-H13...O1	-34.6280	0.91	-33.7180
VI	O2-H13...O1	-24.8573	0.83	-24.0273
Via	O2-H13...O1	-33.6073	0.93	-32.6773
VII	O1-H8...O2	-20.5784	0.79	-19.7884
VIIa	O2-H13...O1	-26.6809	0.94	-25.7409

Table S6:

Second-perturbation energies ($E(2) / \text{kJ} \cdot \text{mol}^{-1}$) of hydrogen bonds for all hydrogen bond associations obtained by NBO analysis at the B3LYP/6-311++G (d, p) level.

Complexes	Donor NBO(i)	Acceptor NBO(j)	$E(2)$
I	LP(1) O1	BD*(1)O2-H16	1.95
II	LP(1) O1	BD*(1)O2-H16	1.14
III	LP(1) O1	BD*(1)O2-H16	1.75
IV	LP(1) O1	BD*(1)O2-H16	1.52
V	LP(1) O2	BD*(1)O1-H8	0.60
Va	LP(1) O1	BD*(1)O2-H13	2.58
VI	LP(1) O1	BD*(1)O2-H13	0.46
Via	LP(1) O1	BD*(1)O2-H13	2.66
VII	LP(1) O2	BD*(1)O1-H8	0.34
VIIa	LP(1) O1	BD*(1)O2-H13	2.44

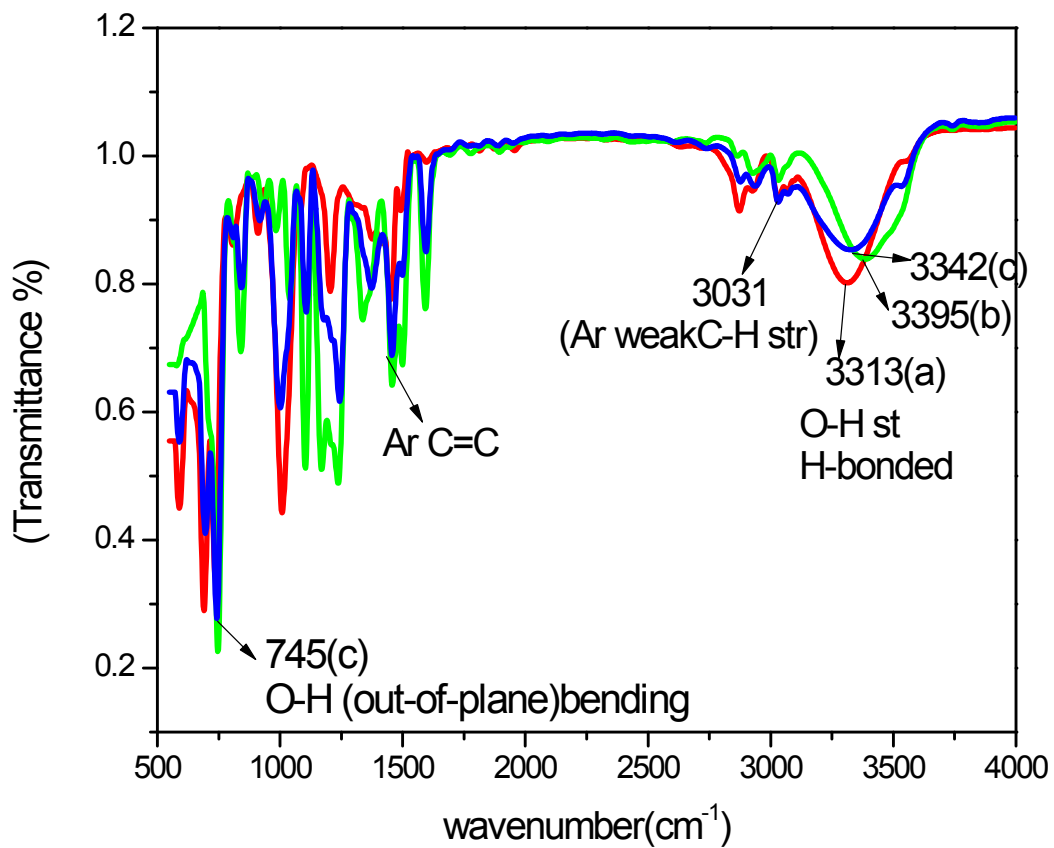


Figure S1: Normalized FT-IR spectra of (benzylalcohol + *o*-cresol) binary mixture over the range (4000–500 cm⁻¹). Peak a: Benzylalcohol; Peak b: *o*-cresol; Peak c: Benzylalcohol with *o*-cresol.

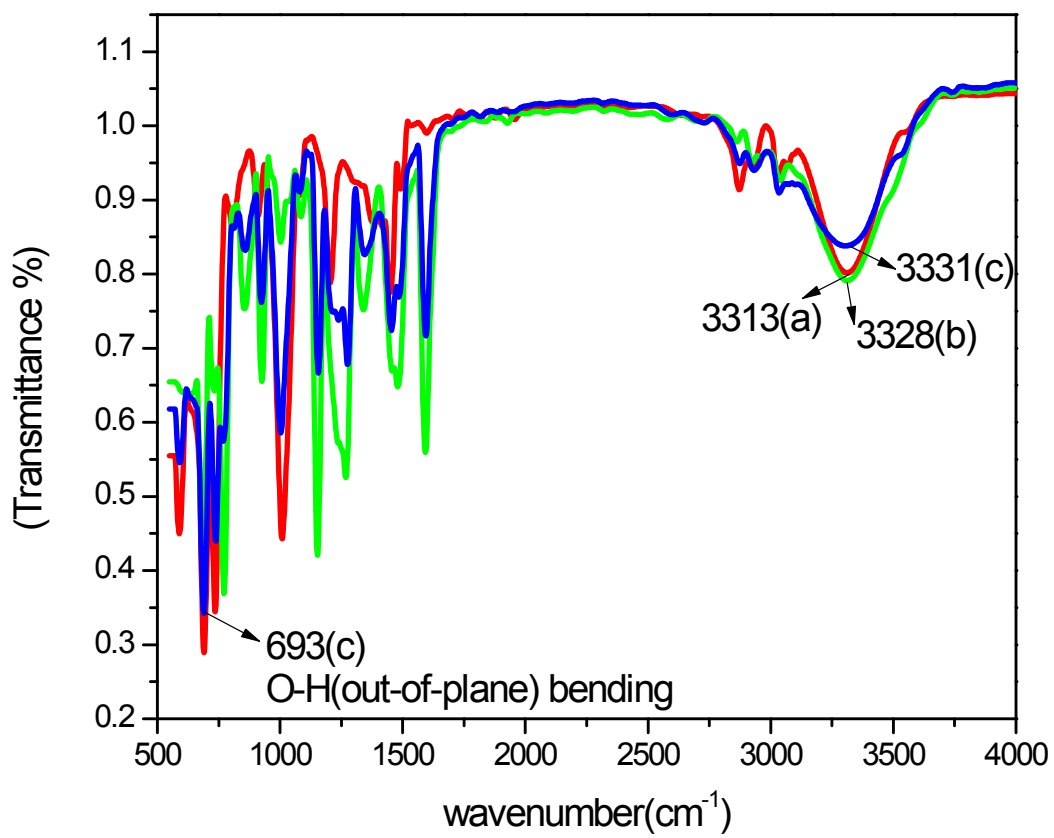


Figure S2: Normalized FT-IR spectra of (Benzylalcohol + *m*-cresol) binary mixture over the range (4000–500 cm⁻¹). Peak a: Benzylalcohol; Peak b: *m*-cresol; Peak c: Benzylalcohol with *m*-cresol.

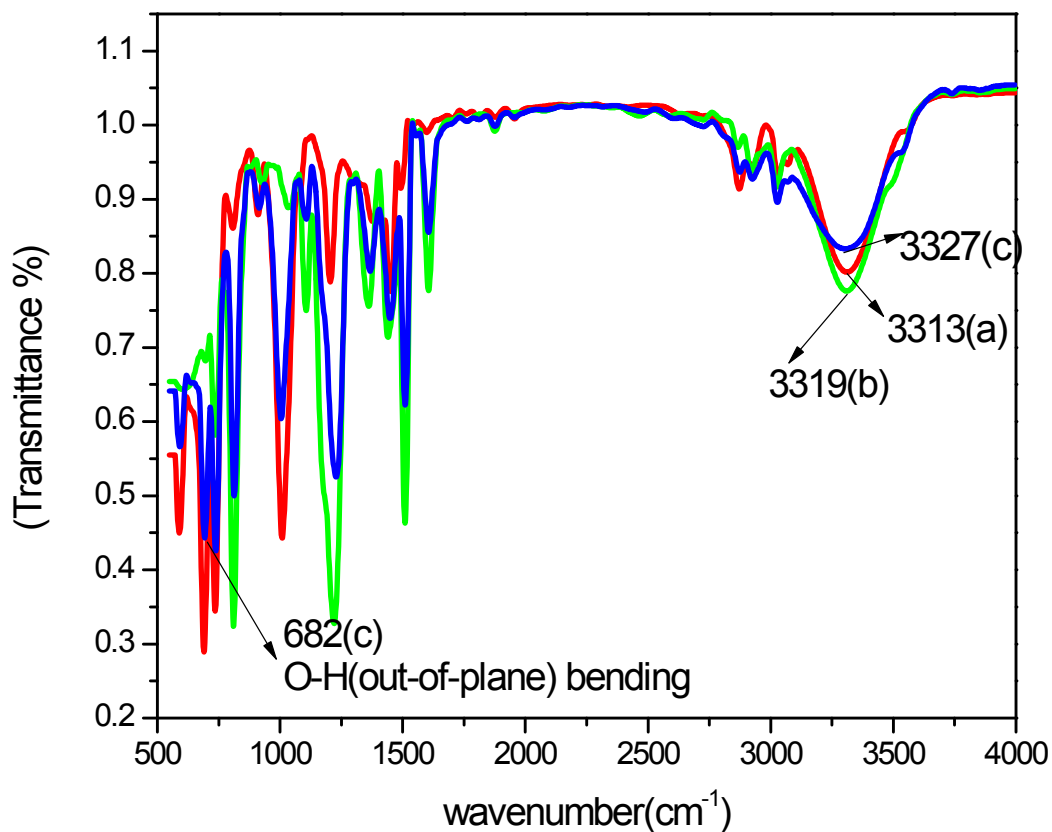
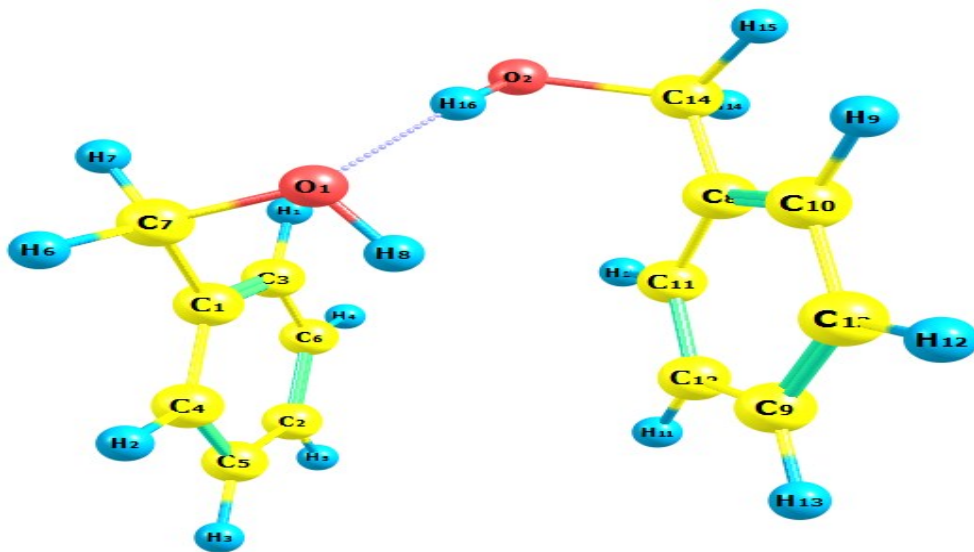


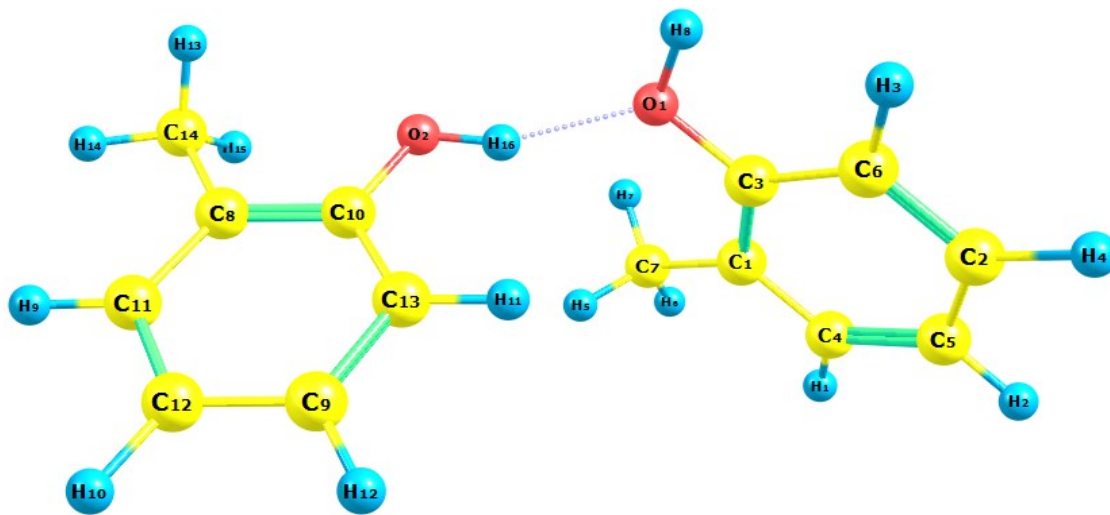
Figure S3: Normalized FT-IR spectra of (Benzylalcohol + *p*-cresol) binary mixture over the range (4000–500 cm⁻¹). Peak a: Benzylalcohol; Peak b: *p*-cresol; Peak c: Benzylalcohol with *p*-cresol.

Figure S4:

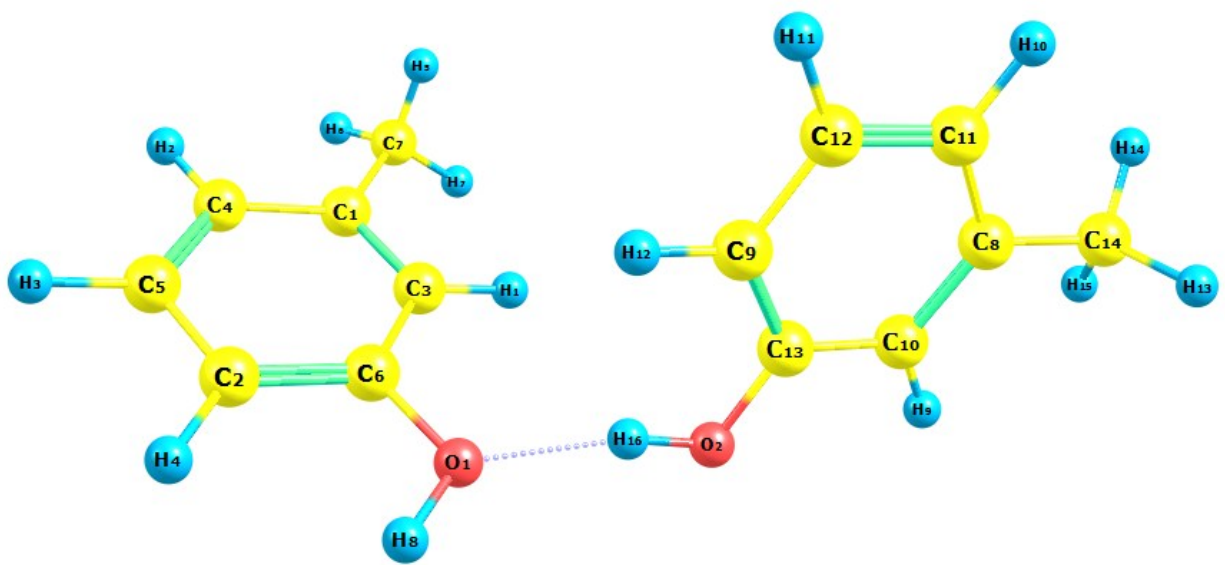
Geometrical optimized structures using B3LYP/6-311++G (d, p) of self-associations **I** (Benzylalcohol–Benzylalcohol), **II** (o-cresol – o-cresol), **III** (p-cresol – p-cresol), **IV** (m-cresol –m-cresol) and of cross-associations **V**, **Va** (Benzylalcohol + o-cresol) **VI**, **VIa** (Benzylalcohol + p-cresol) and **VII**, **VIIa** (Benzylalcohol + m-cresol).



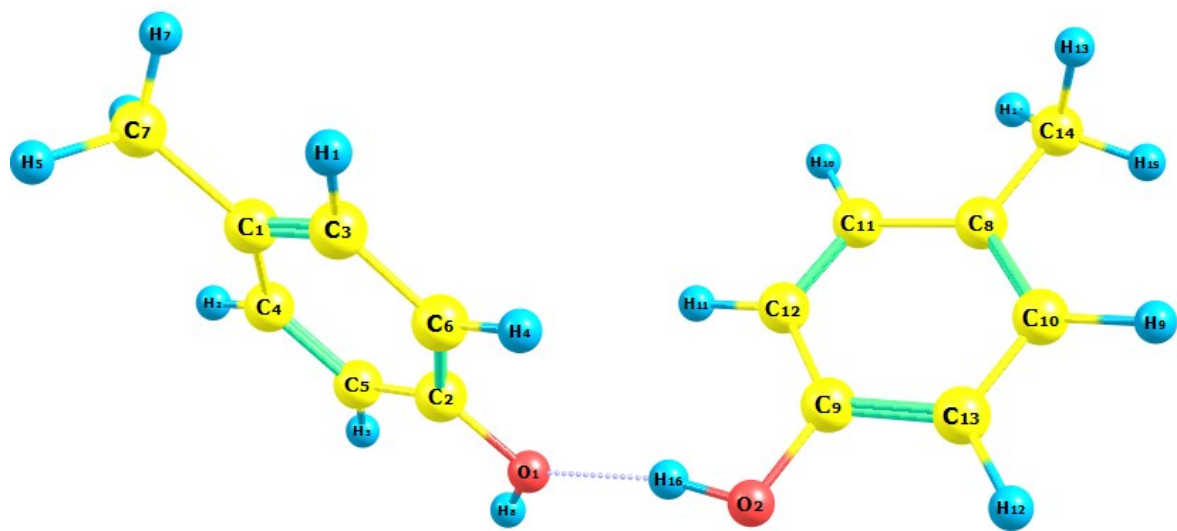
I



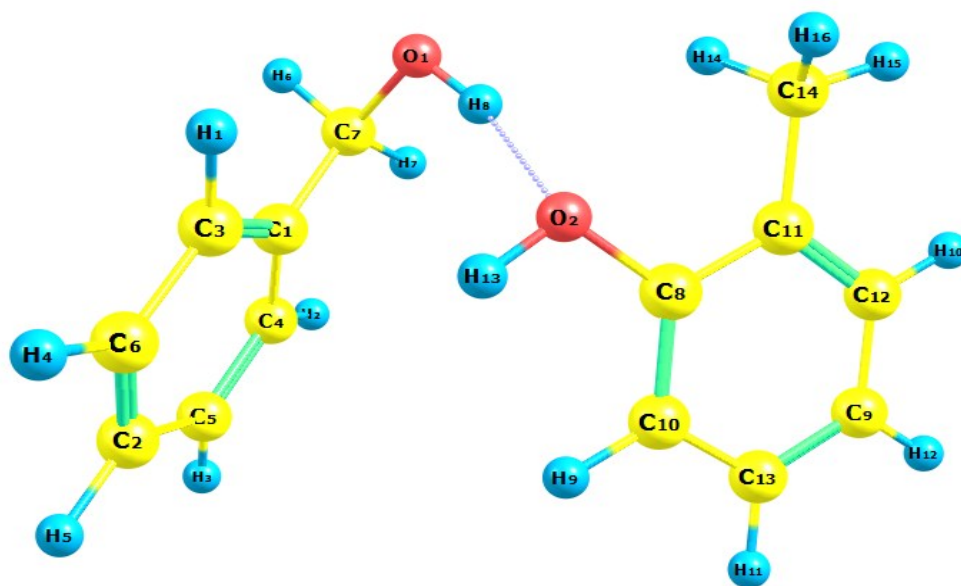
II



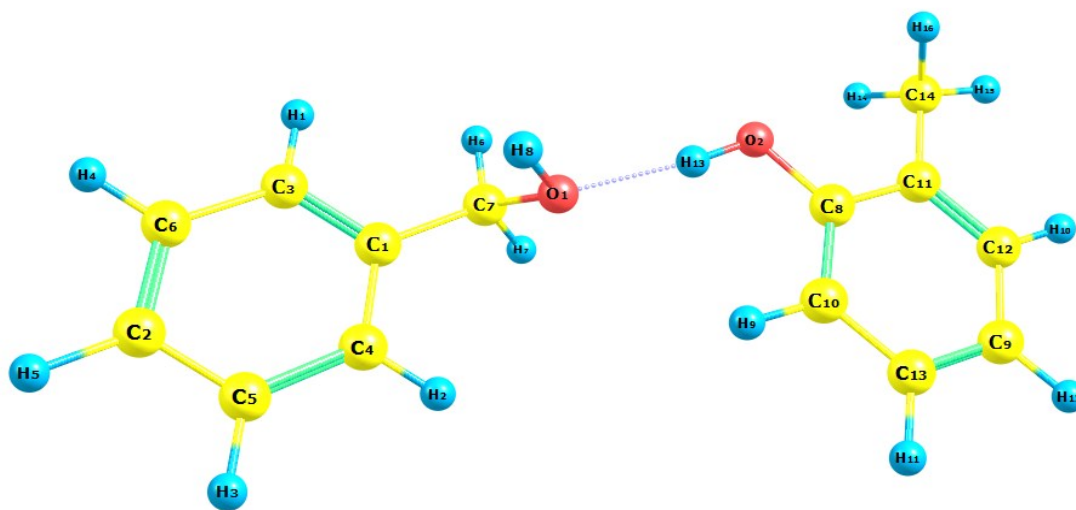
III



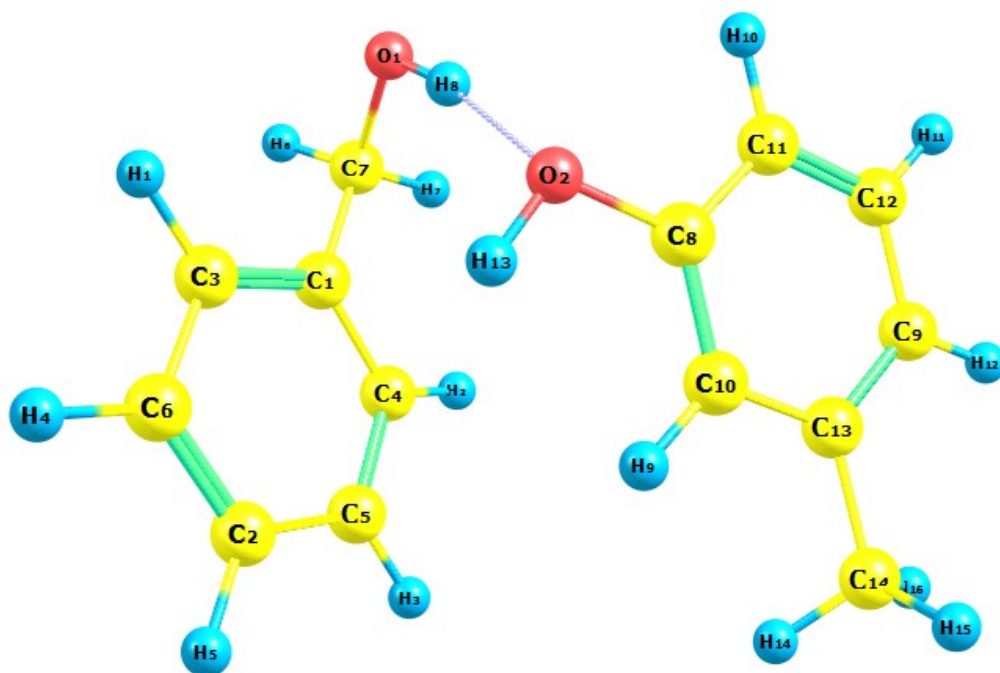
IV



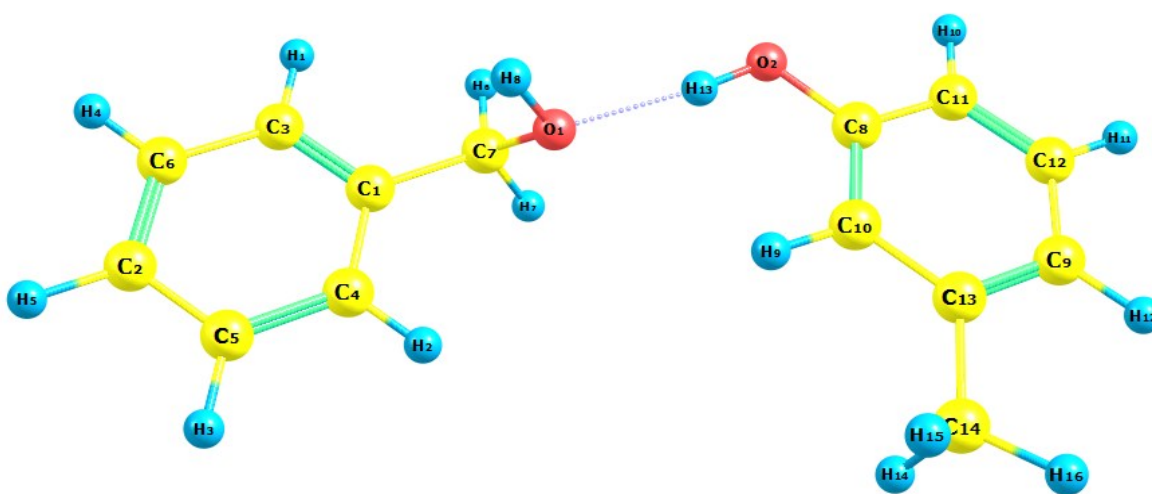
V



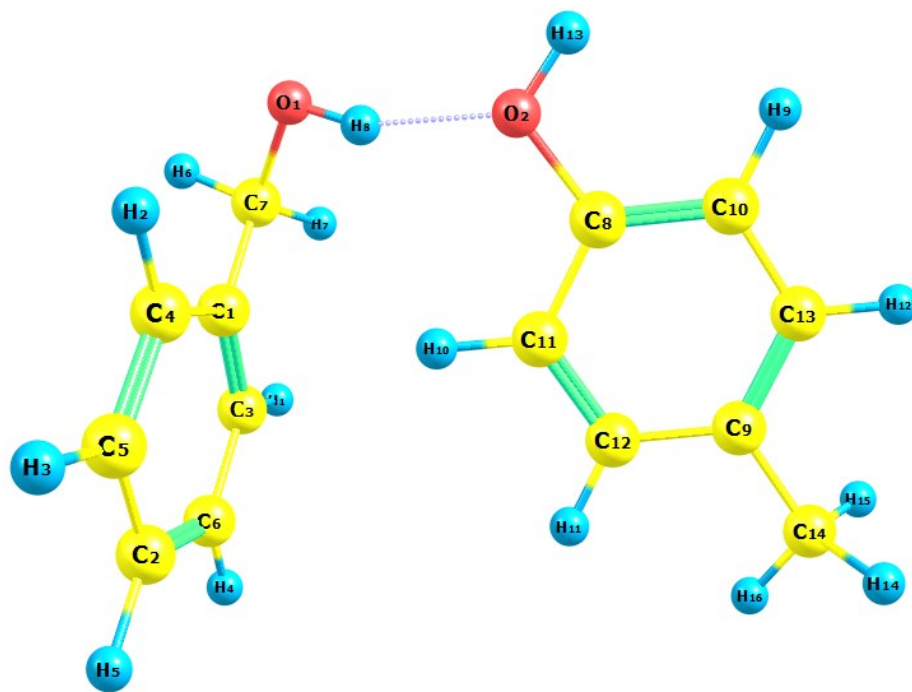
Va



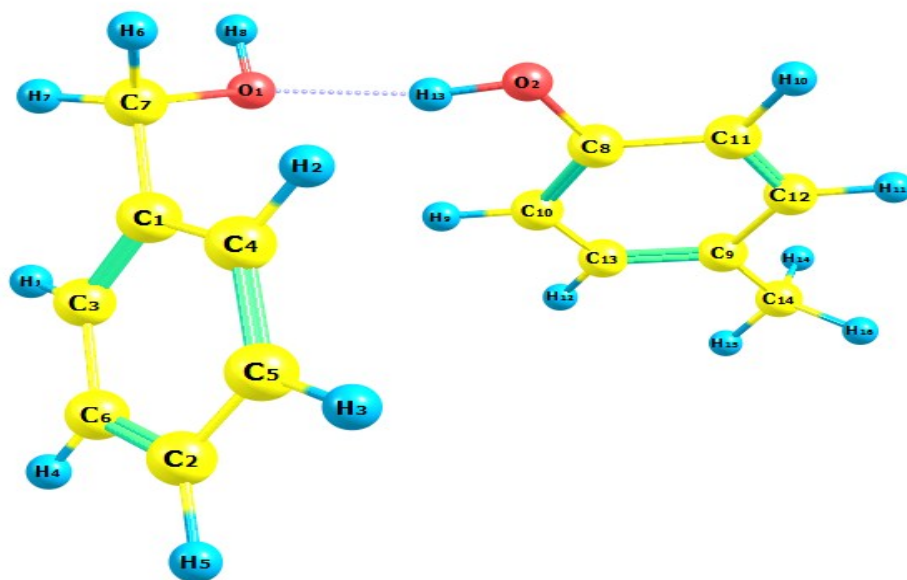
VI



VIa



VII



VIIa