

## Supplemental Material

### Synthesis, structure, and properties of a 3D porous Zn(II) MOF constructed from a terpyridine-based ligand

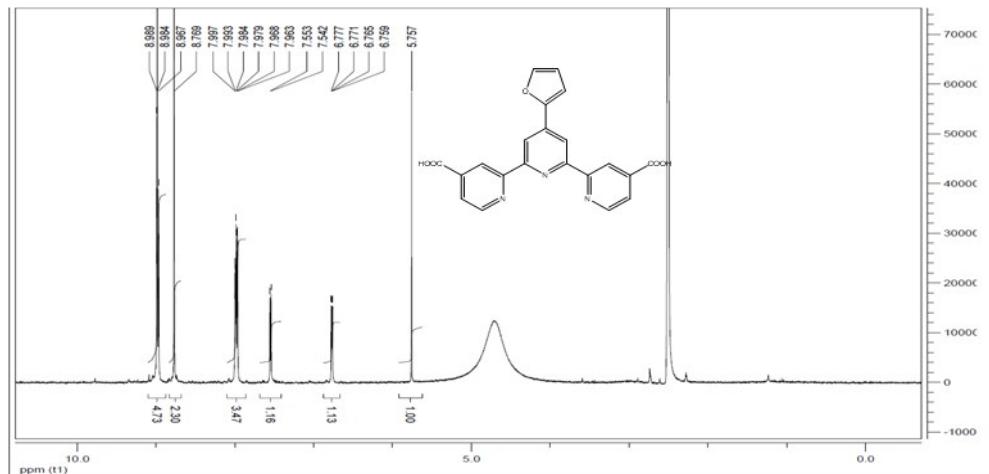
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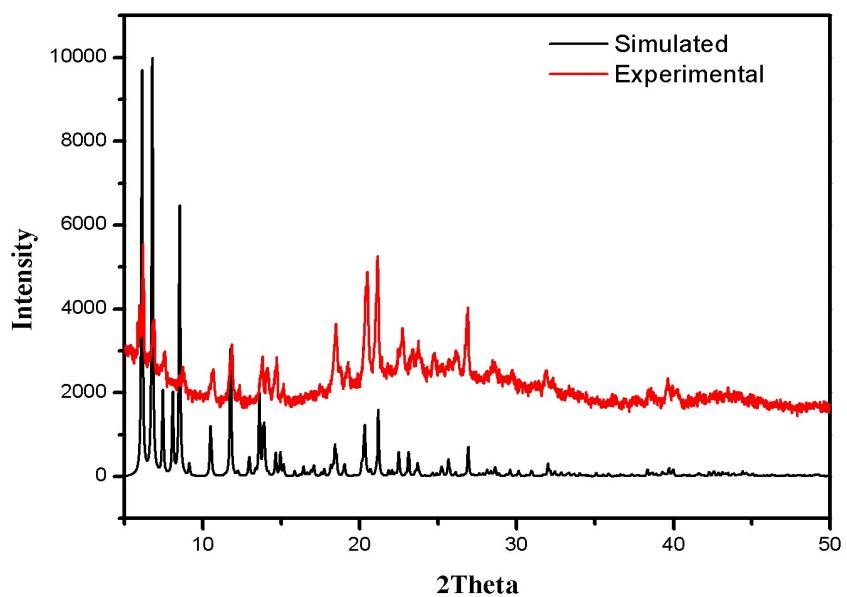
The 4'-(furan-2-yl)-[2,2':6',2"-terpyridine]-4,4"-dicarboxylic acid was prepared according to the reported procedures.<sup>[1]</sup>



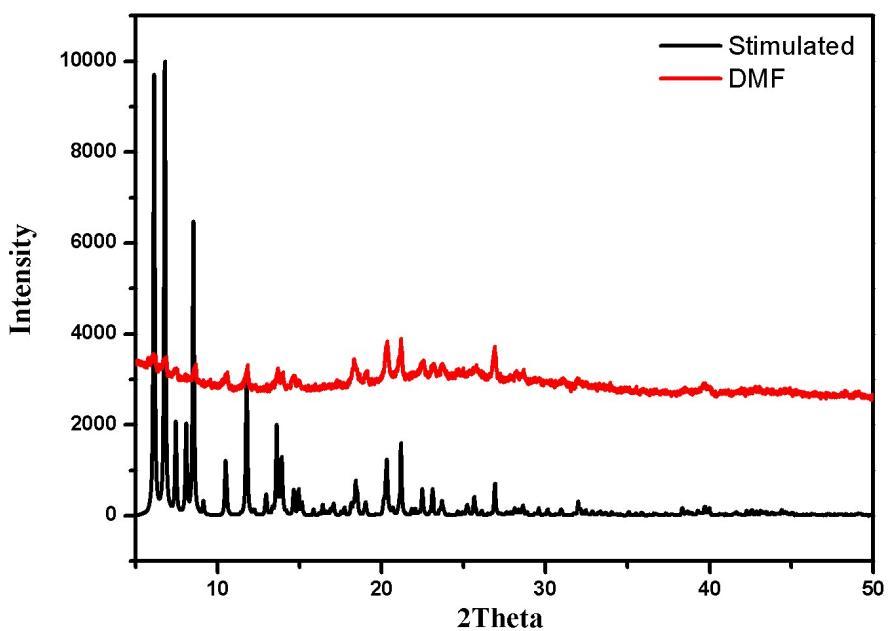
**Fig. S1** The  $^1\text{H}$  NMR of H<sub>2</sub>L ligand.

**Table S1** Selected bond lengths (Å) and angles (°) for **1**

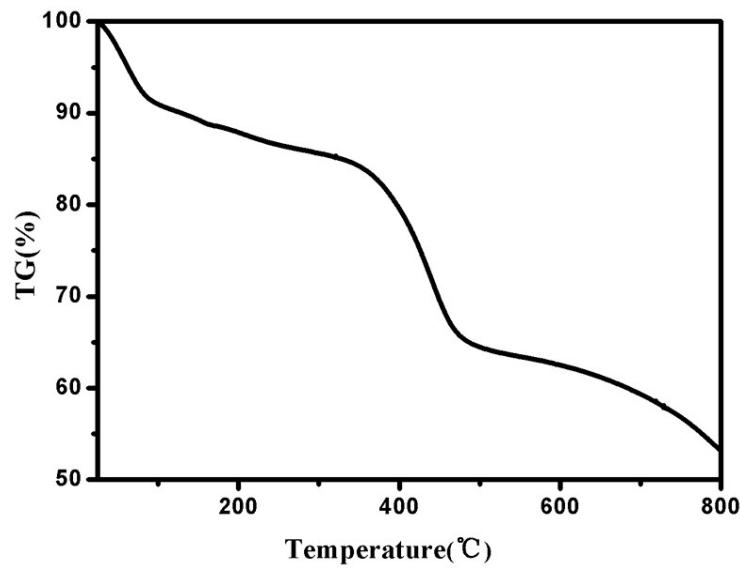
Zn1-O5	2.256 (9)	Zn1-O6	2.235 (6)	Zn1-O7	1.955 (4)
Zn1-N1	2.175 (4)	Zn1-N2	2.071 (4)	Zn1-N3	2.159 (4)
Zn2-O3	1.973 (3)	Zn2-O1	1.945 (3)	Zn2-N4	2.176 (4)
Zn2-N5	2.082(4)	Zn2-N6	2.224(4)		
O3-Zn2-N5	109.88(14)	O3-Zn2-N4	93.93 (16)	O3-Zn2-N6	99.57 (16)
O1-Zn2-O3	124.10(17)	O1-Zn2-N5	125.44 (16)	O1-Zn2-N4	106.16 (16)
O1-Zn2-N6	88.26 (17)	N5-Zn2-N4	75.30 (16)	N5-Zn2-N6	74.87 (16)
N4-Zn2-N6	149.93(17)	N3-Zn1-N1	150.00 (18)	N3-Zn1-O5	104.5 (2)
N3-Zn1-O6	93.03 (18)	O7-Zn1-N3	94.96 (16)	O7-Zn1-N2	132.84 (16)
O7-Zn1-N1	99.44 (16)	O7-Zn1-O5	87.17 (2)	O7-Zn1-O6	142.3 (3)
N2-Zn1-N3	75.11 (15)	N2-Zn1-N1	75.79 (16)	N2-Zn1-O5	139.97 (18)
N2-Zn1-O6	84.8 (3)	N1-Zn1-O5	102.4 (2)	N1-Zn1-O6	91.47 (17)
O6-Zn1-O5	55.2 (3)				



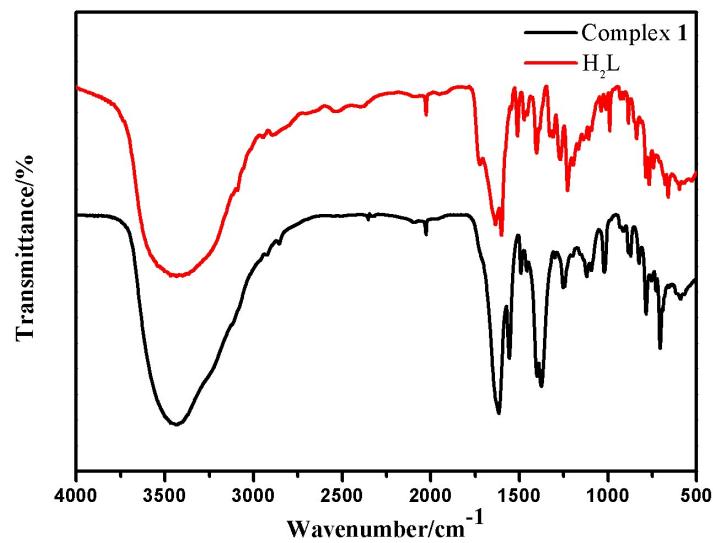
**Fig. S2** The powder XRD pattern and the simulated one from the single-crystal diffraction data for complex **1**.



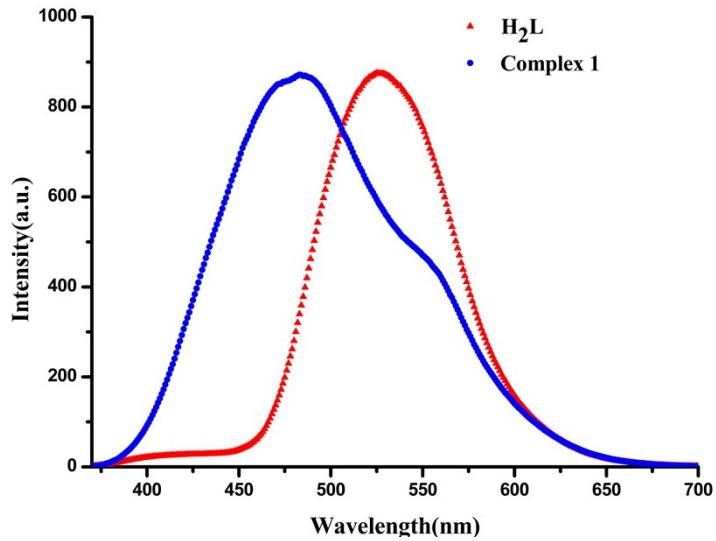
**Fig. S3** The PXRD patterns of complex **1** treated in DMF.



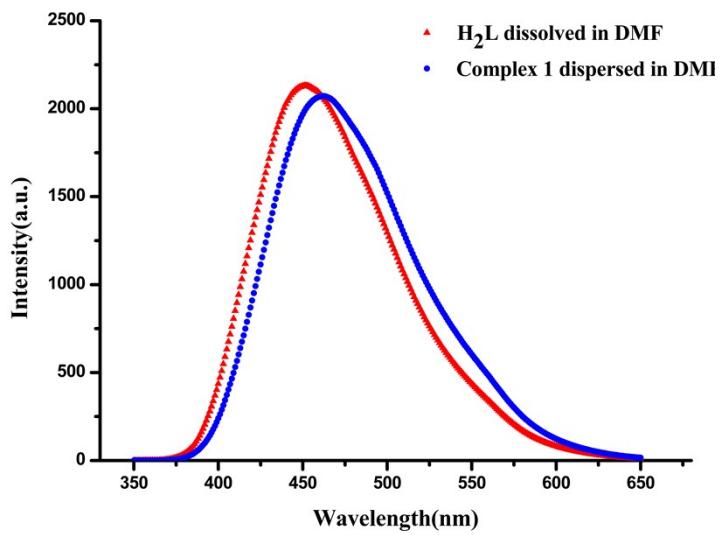
**Fig. S4** TG curve of complex **1**.



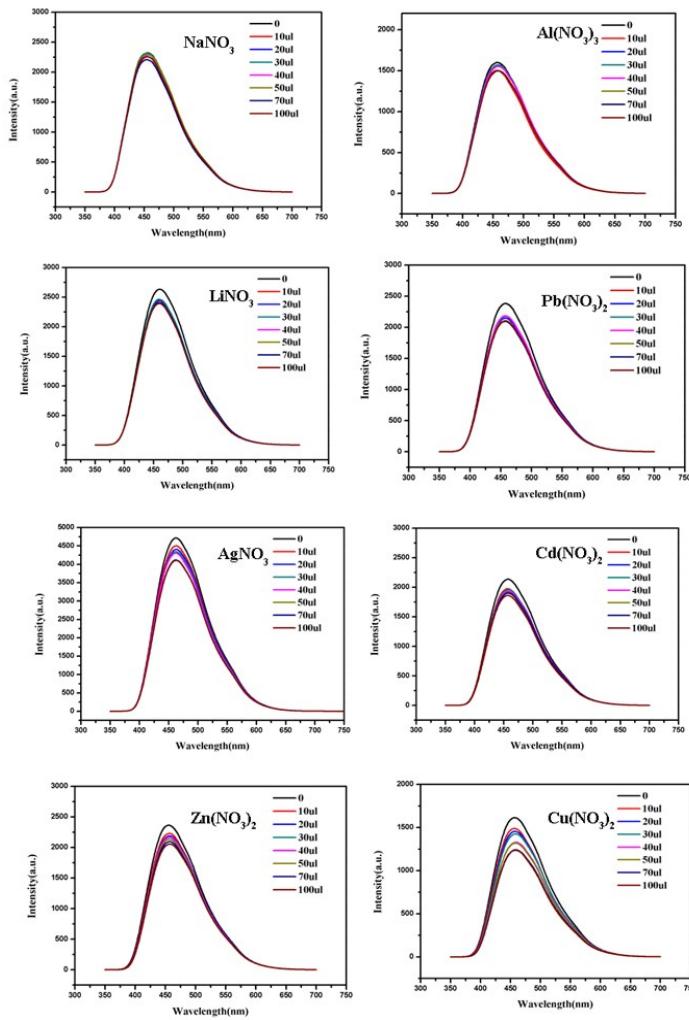
**Fig. S5** IR spectrum of complex **1**.



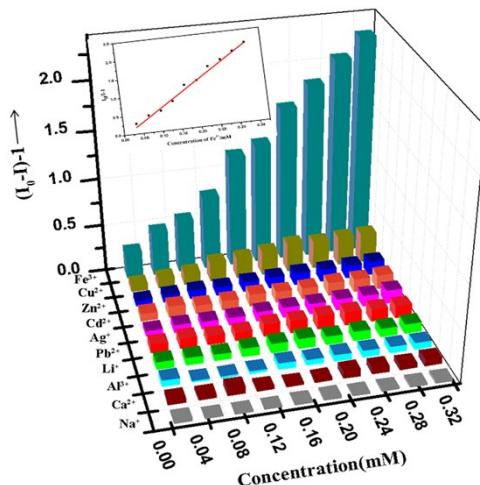
**Fig. S6** Room-temperature emission spectra of free ligand and complex **1**.



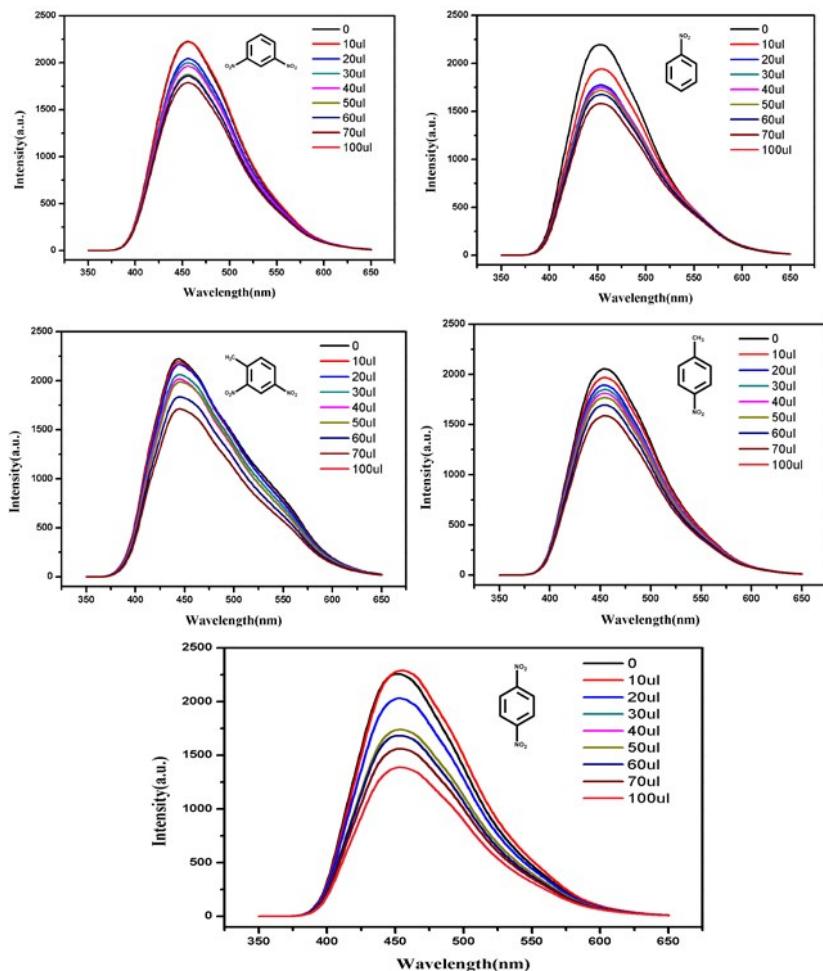
**Fig. S7** The emission spectra for  $\text{H}_2\text{L}$  dissolved in DMF and complex **1** dispersed in DMF at room temperature.



**Fig. S8** Emission intensity of complex **1** dispersed in DMF upon incremental addition of a  $M(NO_3)_x$  ( $M = Cu^{2+}, Na^+, Al^{3+}, Ag^+, Cd^{2+}, Li^+, Zn^{2+}, Pb^{2+}$ ) solution (1mM) in DMF.



**Fig. S9** Corresponding Stern-Volmer plots of analytes. Insert: Stern–Volmer plot of  $I_0/I-1$  versus the  $Fe^{3+}$  concentration in DMF.



**Fig. S10** Effect on the emission spectra of complex **1** dispersed in DMF upon incremental addition of different analyte solutions (1,3-DNB, 2,4-DNT, 1,4-DNB, NB, 1-M-4-NB) solution (1 mM) in DMF

## References:

- [1] J. Dehaut, J. Husson, Laurent Guyard, *Green Chem.*, 2011, **13**, 3337 .