# **Supporting Information:**

Title: Durable and Modified Foam for Cleanup of Oil Contaminations and

Separation of Oil-water Mixtures.

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Information on EDS spectra, stress-strain curves, situation of immersed in the water bath, optical image after floating on HCl and NaOH solution, experiment of removing deionized water, collection of oil and organic solvent, weight percentage of the asprepared samples; three videos showing the magnetic property, unsinkable property and oil absorption process of the sample.

#### Figure:

Figure. S1 EDS spectra of (a) oleic acid coated Fe<sub>3</sub>O<sub>4</sub>, (b) DMPF and (c)DMMPF.

Figure. S2 Stress-strain curves at 80% strain of (a) pure foam, (b) DMPF, and (c) DMMPF in the process of repeated mechanical compression.

Figure S3. Pictures of piece of (a) pure foam and (b) as-prepared foam placed on a water bath. (c) and (d) The pictures of DMMPF immersed in the water bath by an external force.

Figure S4. Optical image of a water droplet on DMMPF after floating on 1.0 M (a) HCl solution and (b) NaOH solution for 72h.

Figure S5. Experiment of removing deionized water with DMMPF under extra force.

Figure. S6 (a) Stress-strain curves of DMMPF after absorption-desorption 60 cycles

(b) and (c) SEM images of DMMPF after absorbing organic liquid after 60 separation cycles at different magnification.

Figure S7. Collection of (a) lubricating oil and (b) benzene from the as-prepared DMMPF by mechanical squeezing.

#### Table:

Table S1 Weight percentage accounted for the element of the as-prepared samples Table S2 Comparison of similar oil absorbing foams/sponges<sup>a</sup>

#### Video:

- Video S1: The exhibition of magnetic property of the durable and modified magnetic polystyrene foam.
- Video S2: The display of unsinkable property of the durable and modified magnetic polystyrene foam by external force.
- Video S3: Removal of lubrication oil from the water surface by the durable and modified magnetic polystyrene foam. The lubrication oil was labeled by Sudan I for clear observation.

### **Figure caption:**



Fig. S1 EDS spectra of (a) oleic acid coated  $Fe_3O_4$ , (b) DMPF and (c)DMMPF.



**Fig. S2** Stress-strain curves at 80% strain of (a) pure foam, (b) DMPF, and (c) DMMPF in the process of repeated mechanical compression.



**Fig. S3.** Pictures of piece of (a) pure foam and (b) as-prepared foam placed on a water bath. (c) and (d) The pictures of DMMPF immersed in the water bath by an external force.



**Fig. S4.** Optical image of a water droplet on DMMPF after floating on 1.0 M (a) HCl solution and (b) NaOH solution for 72h.



Fig. S5. Experiment of removing deionized water with DMMPF under extra force.



**Fig. S6** (a) Stress-strain curves of DMMPF after absorption-desorption 60 cycles (b) and (c) SEM images of DMMPF after absorbing organic liquid after 60 separation cycles at different magnification.



**Fig. S7.** Collection of (a) lubricating oil and (b) benzene from the as-prepared DMMPF by mechanical squeezing.

## Table caption:

| Samples  | С (%) | <b>O(%)</b> | Si(%) | Fe(%) | Au(%) |
|--|-------|-------------|-------|-------|-------|
| Oleic acid coated Fe <sub>3</sub> O <sub>4</sub> | 14.45 | 3.50        | 0     | 38.29 | 43.76 |
| DMPF   | 34.40 | 6.10        | 0     | 21.94 | 37.56 |
| DMMPF  | 22.85 | 14.72       | 14.62 | 38.56 | 9.24  |

Table S1 Weight percentage accounted for the element of the as-prepared samples

| Table | S2 | Com | parison | of | sim | nilar | oil | absor | bing | foams/ | spong | esa |
|-------|----|-----|---------|----|-----|-------|-----|-------|------|--------|-------|-----|
|       |    |     |         |    |     |       |     |       | 0    |        |       |     |

| Absorbents                          | Absorbency (g/g) | Cost | Mechanical property | Wettabilit        | Ref.      |
|-------------------------------------|------------------|------|---------------------|-------------------|-----------|
|                                     |                  |      |                     | У                 |           |
| PU sponges                          | 39.2             | -    |                     | $\checkmark$      | 36        |
| PDMS sponge                         | 12               | +    | -                   | $\checkmark$      | 21        |
| Graphene oxide foam                 | 100              | ++   | ++                  | $\sqrt{\sqrt{1}}$ | 31        |
| RGO sponge                          | 45               | ++   | -                   | $\sqrt{\sqrt{1}}$ | 29        |
| PU soft foam                        | 22.4             | +    | +                   | $\checkmark$      | 30        |
| Spongy graphene                     | 20               | ++   | +                   | $\sqrt{\sqrt{1}}$ | 22        |
| PUF foam                            | 49.9             | +    | +                   | $\sqrt{\sqrt{1}}$ | 19        |
| PUR foam                            | 4.5              | +    |                     | $\sqrt{\sqrt{2}}$ | 38        |
| PFA/SiO <sub>2</sub> -coated sponge | 63               | ++   | -                   | $\sqrt{\sqrt{1}}$ | 37        |
| PU-CNT-PDA-ODA spong                | ge 35.0          | +    | -                   | $\sqrt{\sqrt{2}}$ | 35        |
| MPG foam                            | 27.0             | ++   | -                   | $\sqrt{\sqrt{2}}$ | 43        |
| DMMPF                               | 40.1             |      | +                   | $\sqrt{\sqrt{2}}$ | This work |

aNoting: "- -"very low, "-" low, "+"high, "++"very high, " $\sqrt{}$ " hydrochloricity, " $\sqrt{\sqrt{}}$ " high

hydrochloricity, " $\sqrt{\sqrt{\sqrt{1}}}$ " superhydrophobicity.