

Supplementary Materials

Table A1

IGC Conditions	
Injector temperature (°C)	150
Detector temperature (°C)	200
Column temperatures (°C)-surface prop.	40, 50, 60
Column temperatures (°C)-thermodynamic prop.	90, 100, 110
Column length (cm)	25
Column O.D. (inch)	1/4
Column I.D. (inch)	1/6
Mass loaded (g)	2.19
Flow rate (ml/min)	25

Table A2

Okra	
Moisture content	8-10 %
Protein content	9 %
Zeta potential (25°C)	-14.45 mV

Table A3

Probe	$-\Delta H^{\text{sp}}$ (Jmol⁻¹)	ΔS^{sp} (Jmol⁻¹K⁻¹)	R²
Tetrahydrofuran	9212	-17.8	0.91
Chloroform	15017	-33.9	0.96
Acetonitrile	15057	-38.4	0.99
<i>n</i> -Butanol	14641	-36.3	0.90
Ethanol	12712	-29.9	0.95
Ethyl acetate	6307	-15.7	0.99
Pyridine	6662	-7.5	0.99
Methyl acetate	7609	-18.9	0.98

Table A4

Probes	90 °C	100 °C	110 °C
<i>n-Heptane</i>	18.46	19.02	20.43
<i>n-Octane</i>	15.91	16.55	17.31
<i>n-Nonane</i>	13.77	14.41	15.69
<i>n-Decane</i>	11.38	12.74	13.84
<i>n-Undecane</i>	9.22	10.35	11.39
<i>n-Dodecane</i>	7.23	8.43	9.31
<i>Tetrahydrofuran</i>	19.98	20.69	21.24
<i>Chloroform</i>	18.55	19.53	22.24
<i>Acetonitrile</i>	21.66	22.81	22.98
<i>n-Butanol</i>	15.61	17.06	17.96
<i>Ethanol</i>	22.02	23.22	24.63
<i>Ethyl acetate</i>	20.91	21.98	24.14
<i>Pyridine</i>	14.50	15.75	16.98
<i>Methyl acetate</i>	23.68	25.08	25.00
<i>1,4 Dioxane</i>	17.24	18.23	19.55
<i>1-Propanol</i>	19.74	21.37	22.13
<i>2-Pentanone</i>	18.19	19.15	20.33
<i>Cyclopentanone</i>	15.25	16.44	17.28
<i>1-Nitropropane</i>	16.31	16.70	18.18
<i>Methanol</i>	19.08	19.76	20.39

Table A5

Probes	90 °C	100 °C	110 °C
<i>n-Heptane</i>	-50.84	-50.97	-53.33
<i>n-Octane</i>	-69.84	-69.69	-69.85
<i>n-Nonane</i>	-95.34	-94.49	-95.36
<i>n-Decane</i>	-123.19	-123.55	-123.19
<i>n-Undecane</i>	-108.41	-108.52	-108.41
<i>n-Dodecane</i>	-103.99	-104.43	-103.99
<i>Tetrahydrofuran</i>	-63.27	-63.48	-63.27
<i>Chloroform</i>	-183.69	-181.39	-183.74
<i>Acetonitrile</i>	-66.53	-67.82	-66.51
<i>n-Butanol</i>	-117.34	-118.07	-117.34
<i>Ethanol</i>	-130.17	-129.90	-130.18
<i>Ethyl acetate</i>	-160.94	-159.51	-160.97
<i>Pyridine</i>	-123.69	-123.73	-123.70
<i>Methyl acetate</i>	-66.55	-68.50	-66.52
<i>1,4 Dioxane</i>	-115.27	-114.84	-115.28
<i>1-Propanol</i>	-119.90	-121.06	-119.89
<i>2-Pentanone</i>	-106.94	-106.64	-106.95
<i>Cyclopentanone</i>	-101.14	-101.60	-101.13
<i>1-Nitropropane</i>	-92.80	-91.36	-92.83
<i>Methanol</i>	-65.68	-56.65	-56.81

Table A6

Probes	90 °C	100 °C	110 °C
<i>n-Heptane</i>	19.37	19.04	19.60
<i>n-Octane</i>	19.34	19.05	18.90
<i>n-Nonane</i>	19.69	19.34	19.65
<i>n-Decane</i>	19.80	20.13	20.21
<i>n-Undecane</i>	20.08	20.11	20.08
<i>n-Dodecane</i>	20.51	20.57	20.32
<i>Tetrahydrofuran</i>	17.96	17.82	17.53
<i>Chloroform</i>	16.16	16.30	18.19
<i>Acetonitrile</i>	21.17	21.46	20.78
<i>n-Butanol</i>	18.96	19.22	18.97
<i>Ethanol</i>	20.82	20.92	21.25
<i>Ethyl acetate</i>	19.87	20.05	21.32
<i>Pyridine</i>	16.94	17.25	17.55
<i>Methyl acetate</i>	20.85	21.38	20.46
<i>1,4 Dioxane</i>	18.44	18.51	18.93
<i>1-Propanol</i>	20.67	21.16	20.80
<i>2-Pentanone</i>	19.45	19.49	19.77
<i>Cyclopentanone</i>	19.16	19.39	19.28
<i>1-Nitropropane</i>	20.36	19.78	20.30
<i>Methanol</i>	16.45	16.09	15.69

Table A7

Probes	90 °C			100 °C			110 °C		
	δ_d	δ_p	δ_h	δ_d	δ_p	δ_h	δ_d	δ_p	δ_h
<i>n-Octane</i>	14.9	0	0	14.8	0	0	14.7	0	0
<i>n-Nonane</i>	15.1	0	0	15.0	0	0	14.9	0	0
<i>n-Decane</i>	15.2	0	0	15.1	0	0	15.0	0	0
<i>n-Undecane</i>	15.5	0	0	15.4	0	0	15.3	0	0
<i>n-Dodecane</i>	15.5	0	0	15.4	0	0	15.3	0	0
<i>Tetrahydrofuran</i>	16.1	5.4	7.6	15.9	5.4	7.6	15.8	5.4	7.5
<i>Chloroform</i>	17.0	3.0	5.5	16.9	2.9	5.4	16.8	2.9	5.4
<i>Acetonitrile</i>	14.5	17.1	5.8	14.4	16.9	5.7	14.3	16.8	5.7
<i>n-Butanol</i>	15.4	5.5	15.2	15.3	5.4	15.1	15.2	5.4	15.0
<i>Ethanol</i>	15.1	8.4	18.6	15.0	8.4	18.4	14.9	8.3	18.3
<i>Ethyl acetate</i>	15.0	5.0	6.9	14.9	5.0	6.8	14.8	5.0	6.7
<i>Pyridine</i>	18.3	8.5	5.7	18.2	8.4	5.7	18.1	8.4	5.6
<i>1,4 Dioxane</i>	18.3	1.7	7.1	18.2	1.7	7.1	18.1	1.7	7.0
<i>1-Propanol</i>	15.4	6.5	16.7	15.3	6.5	16.6	15.2	6.4	16.5
<i>2-Pentanone</i>	15.3	7.3	4.5	15.2	7.2	4.5	15.1	7.2	4.4
<i>Cyclopentanone</i>	17.3	11.5	5.0	17.2	11.4	5.0	17.1	11.4	5.0
<i>1-Nitropropanone</i>	15.9	11.8	5.3	15.8	11.7	5.2	15.7	11.6	5.2
<i>Methanol</i>	14.5	11.8	21.3	14.3	11.7	21.2	14.2	11.6	21.0
<i>Water</i>	15.3	15.8	41.8	15.3	15.8	41.7	15.3	15.8	41.6

List of Supplementary Tables

Table A1: Chromatographic conditions and column specifications.

Table A2: Characterization of okra gum.

Table A3: Specific components of the enthalpy and entropy of adsorption of polar probes on the surface of okra.

Table A4: The molar free energies of sorption ΔG_1^s (kJ mol⁻¹), of various probes on okra extract at 90, 100 and 110 °C.

Table A5: The entropies of sorption ΔS_1^s (J mol⁻¹ K⁻¹), of various probes on okra extract at 90, 100 and 110 °C.

Table A6: The molar free energies of mixing, ΔG_1^∞ (kJ mol⁻¹), of various probes on okra extract at 90, 100 and 110 °C.

Table A7: Partial solubility parameters (MPa)^{0.5} of the probes at 90, 100 and 110 °C.

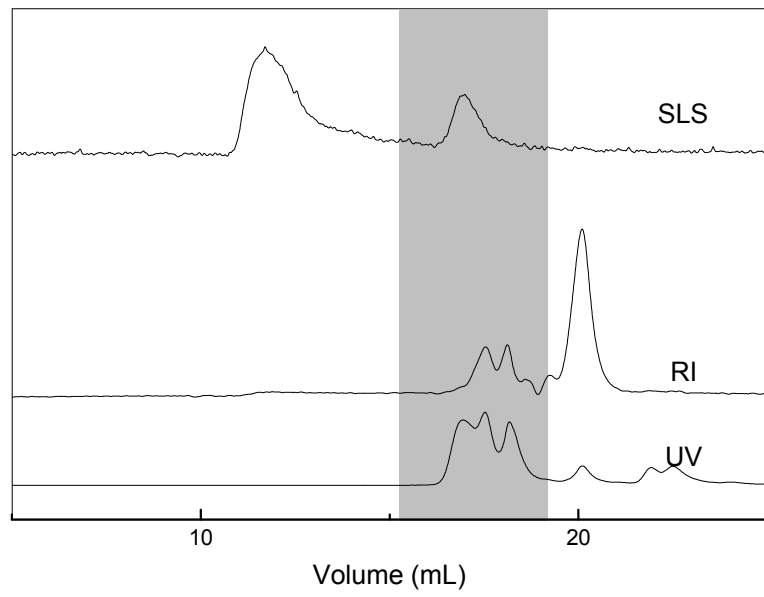


Fig. A1

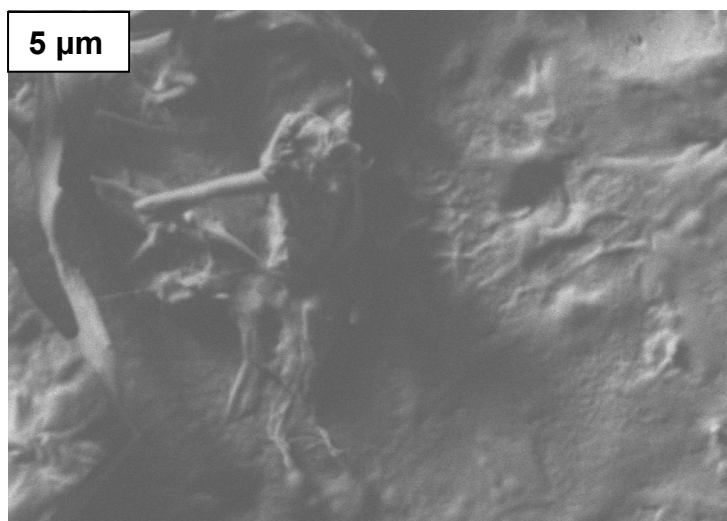


Fig. A2

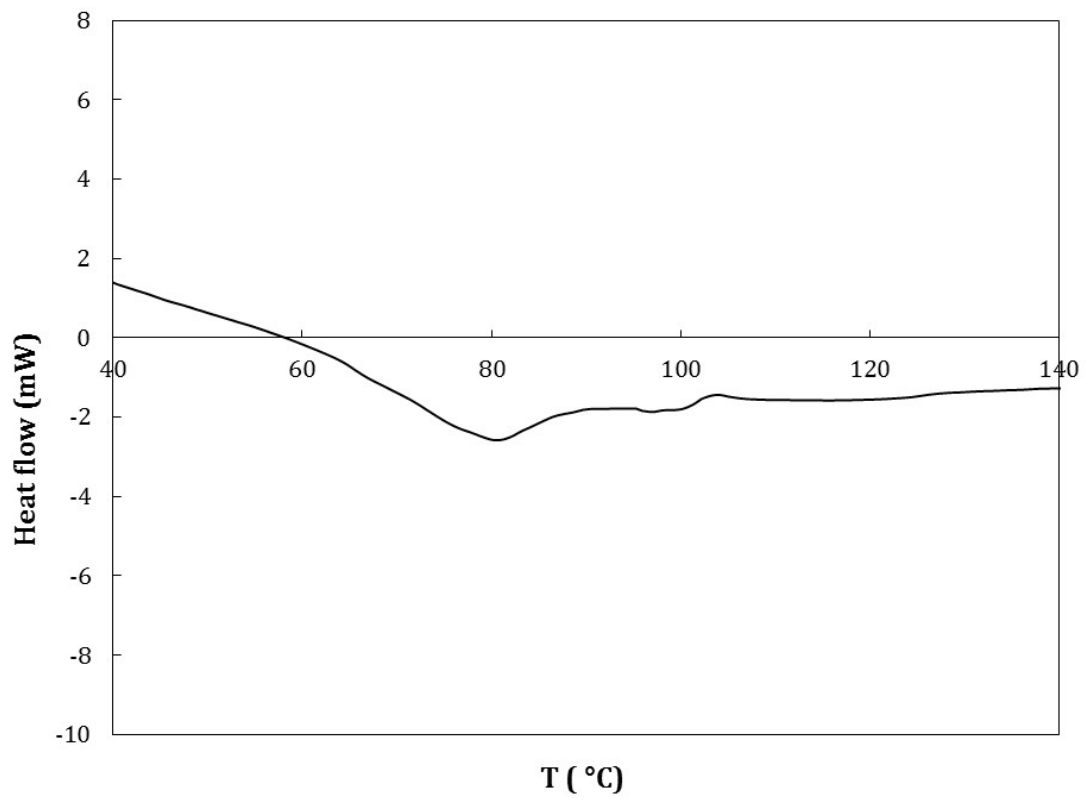


Fig. A3

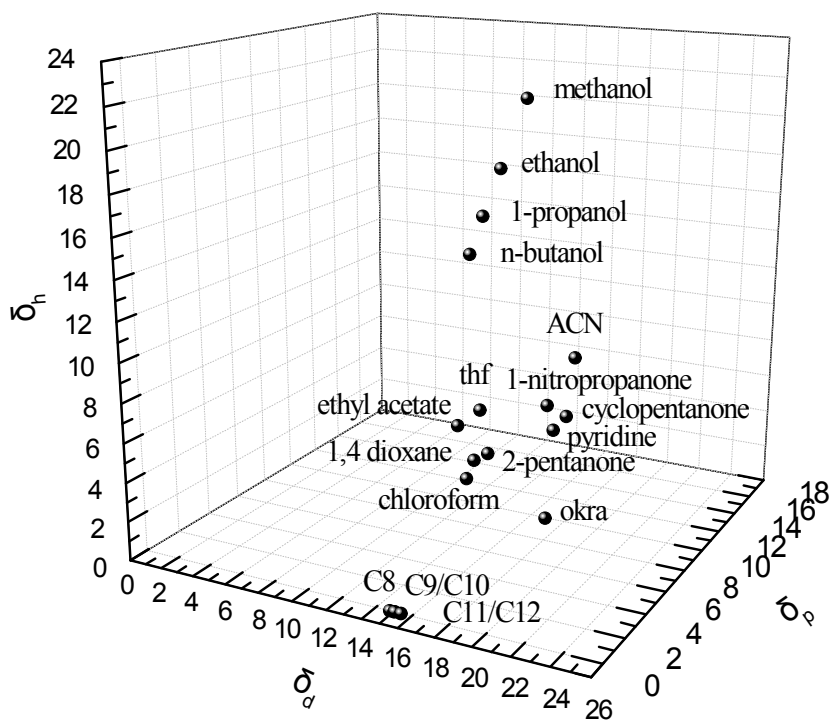


Fig. A4

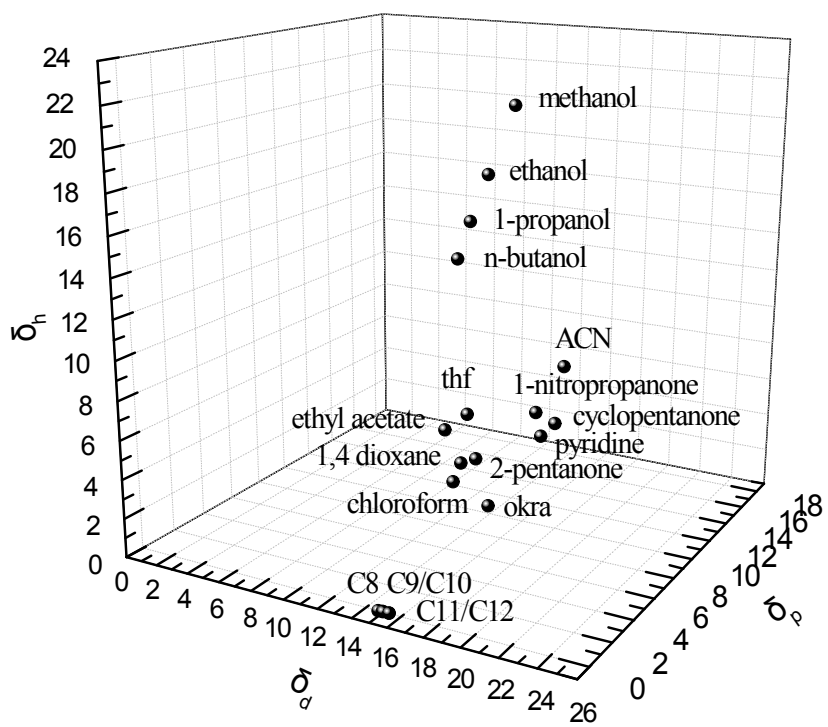


Fig. A5

List of Supplementary Figures

Fig. A1: Size exclusion chromatograms for the okra extract, as taken by a SEC–MALLS/UV/RI setup. The data presented for SLS (MALLS) are scattering intensities recorded at 90 °C, in an abstract intensity y-scale. The highlighted region corresponds to a region of strong absorbance at 280 nm, as recorded by a UV detector

Fig. A2: Scanning electron microscopy (SEM) image (bar = 5 μm) for okra extract.

Fig. A3: Differential scanning calorimetry (DSC) of okra gum extract.

Fig. A4: Partial solubility parameters in three dimensions (Hansen space) of okra at 100°C.

Fig. A5: Partial solubility parameters in three dimensions (Hansen space) of okra at 110°C.