

Single graphene nanoplatelets: capacitance,
potential of zero charge and diffusion coefficient

Supporting Information

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Supporting Information

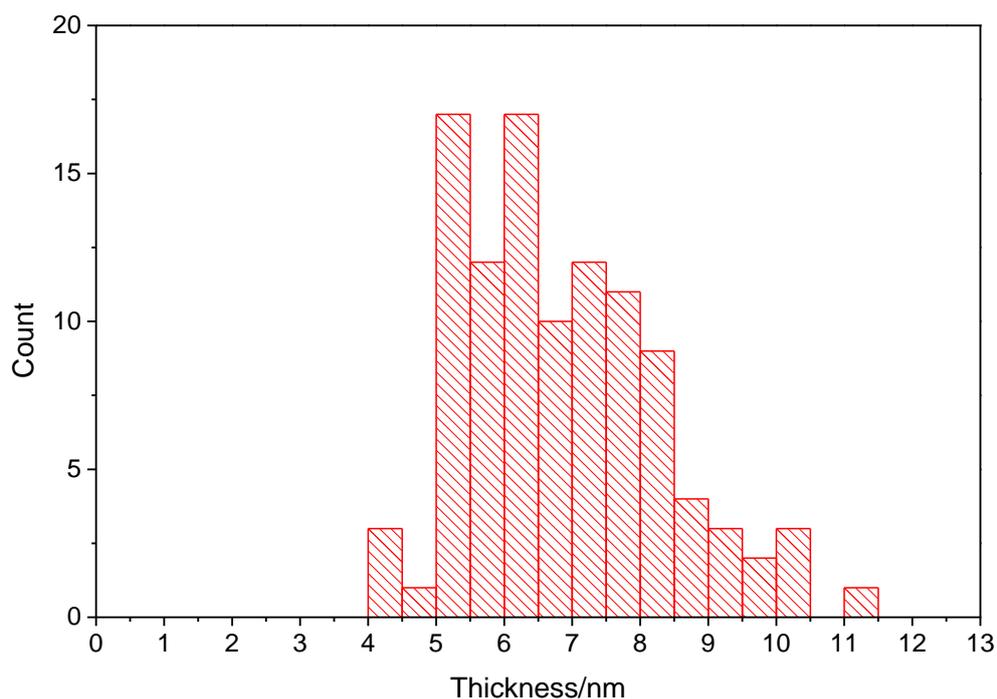


Figure S1: Histogram of thickness of graphene nanoparticles as inferred from SEM analysis with an average thickness of 7.1 ± 2.1 nm. ($n = 108$, bin size = 0.5 nm)

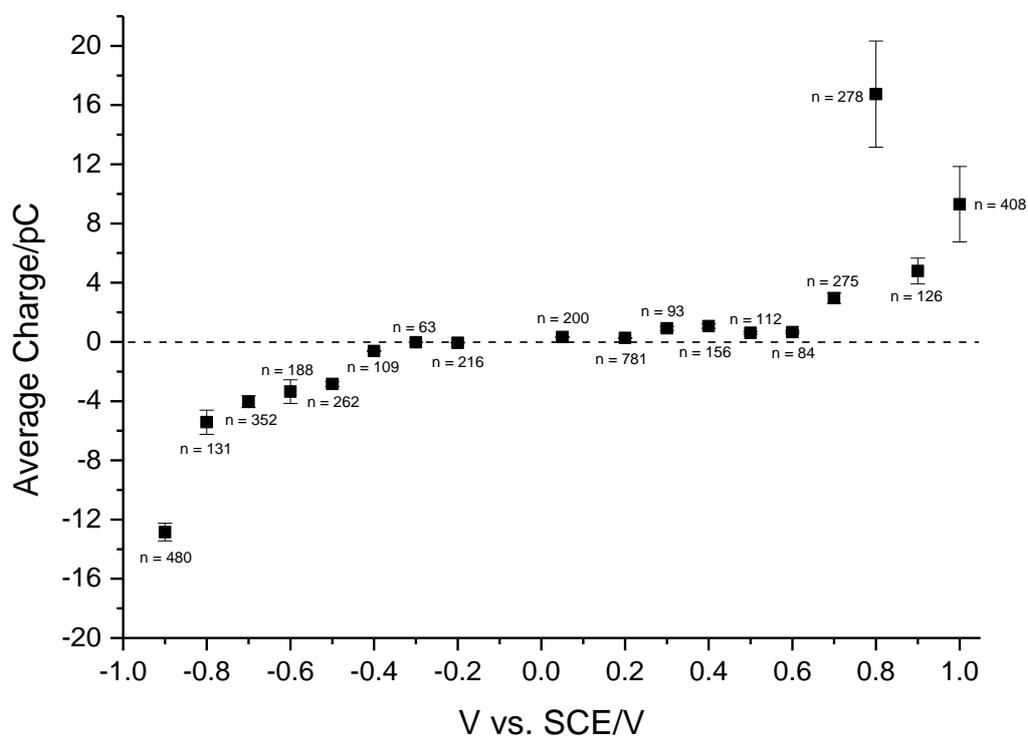


Figure S2: Plot of average impact charge vs. applied potential (vs. SCE) in 5.9×10^{-13} mol dm^{-3} suspension. The n numbers indicate the number of impacts involved in generating each data point and its error bars.

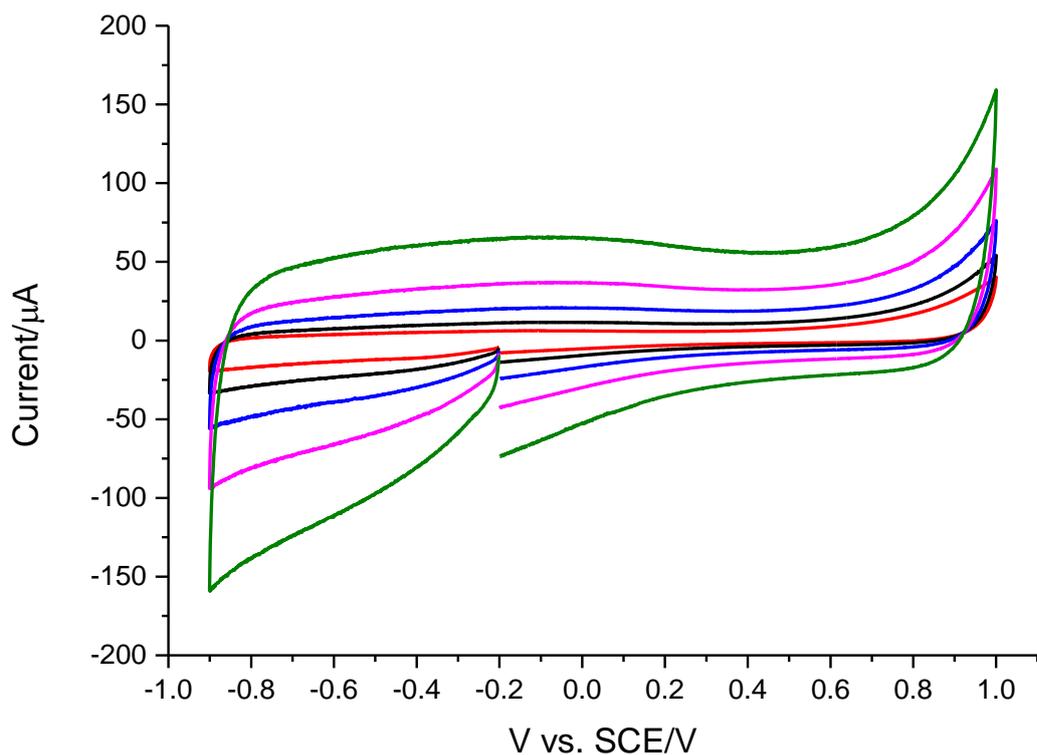


Figure S3: The voltammetric response of Graphene Nanoplatelets drop cast ($7.9 \mu\text{g}$) onto a glassy carbon electrode (BAS Technicol, USA, diameter 3 mm) in supporting electrolyte (0.1M KCl, 50 mM potassium monophosphate, 50 mM potassium diphosphate) recorded as a function of scan rates (25 mV s^{-1} , red line; 50 mV s^{-1} , black line; 100 mV s^{-1} , blue line; 200 mV s^{-1} , magenta line; 400 mV s^{-1} , green line).

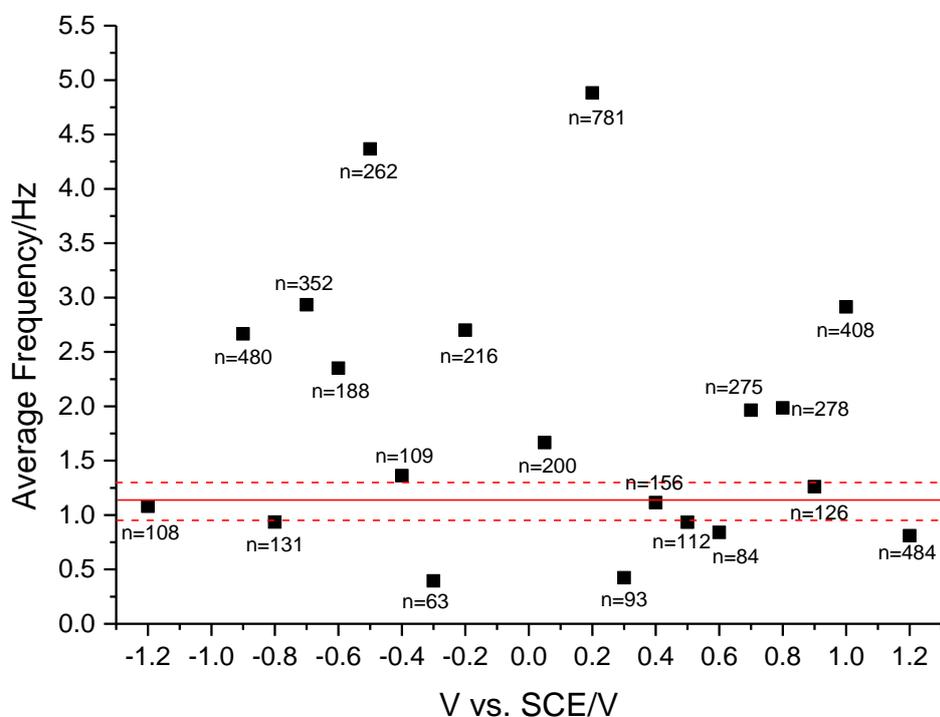


Figure S4: Plot of average impact frequency vs. potential applied (vs. SCE) in $5.91 \times 10^{-13} \text{ mol dm}^{-3}$ suspension. The wide scatter indicates a lack of frequency bias with respect to the potential applied. The numbers n indicate the number of impacts the dots plotted are based on.

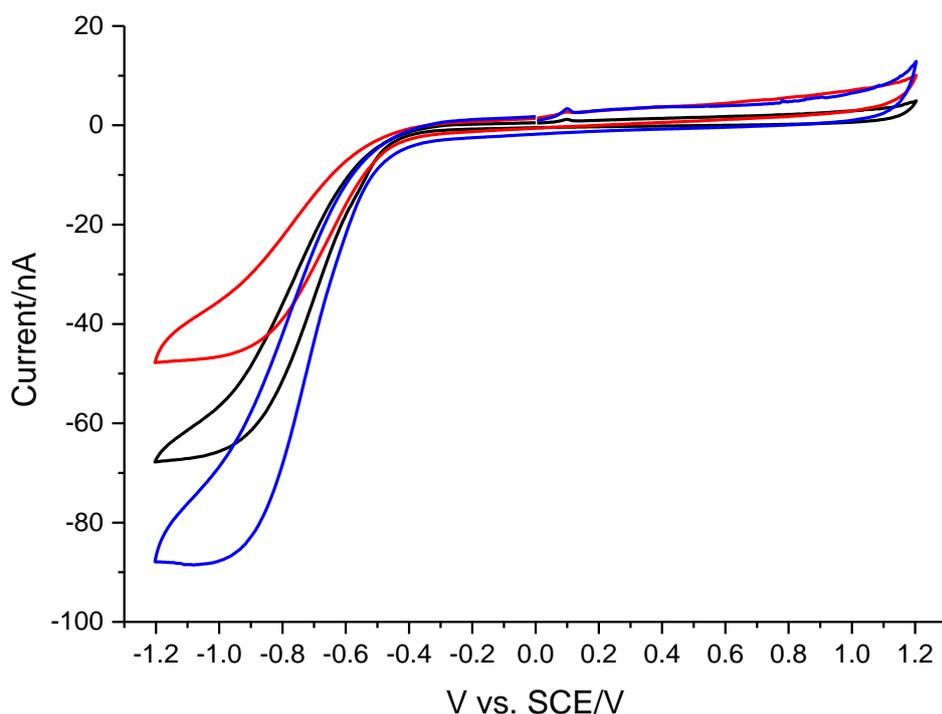


Figure S5: The voltammetric response of a cylindrical carbon fibre wire electrode in supporting electrolyte (0.1M KCl, 50 mM potassium monophosphate, 50 mM potassium diphosphate) recorded as a function of scan rates (25 mV s⁻¹, red line; 50 mV s⁻¹, black line; 100 mV s⁻¹, blue line). The reduction current likely indicates 50 μM of oxygen in the solution due to limitations of degassing.

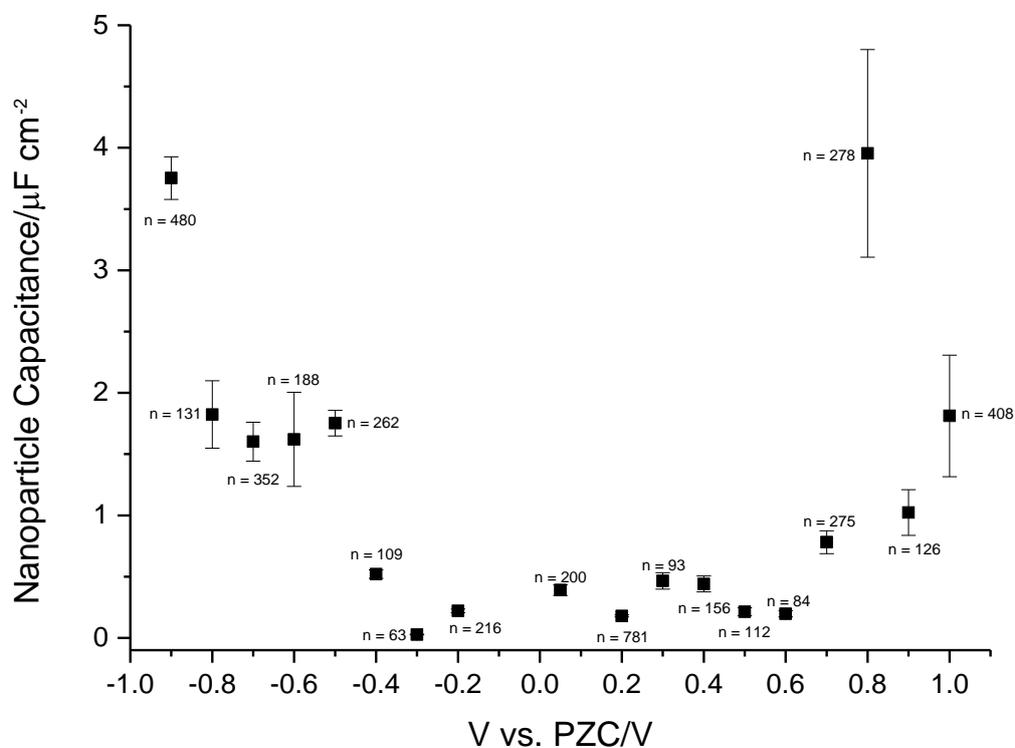


Figure S6: Plot of nanoparticle capacitance vs. PZC in 5.91×10^{-13} mol dm⁻³ suspension. The plot shows a non-linear relationship of potential (vs. PZC) to nanoparticle capacitance. The n numbers shows the number of impacts involved in generating each data point and error bars.