

Supporting Information

Bioinspired Design of Redox-Active Ligands for Multielectron Catalysis: Effects of Positioning Pyrazine Reservoirs on Cobalt for Electro- and Photocatalytic Generation of Hydrogen from Water

Jonah W. Jurss,^{a,c,e} Rony S. Khnayzer,^{f,g} Julien A. Panetier,^{a,d} Karim A. El Roz,^f Eva M. Nichols,^{a,e} Martin Head-Gordon,*^{a,e} Jeffrey R. Long,*^{a,d} Felix N. Castellano,^{f,j} Christopher J. Chang*^{a,b,e,h}

^aDepartment of Chemistry and ^bDepartment of Molecular and Cell Biology, University of California, Berkeley, California 94720, USA; ^cDepartment of Chemistry and Biochemistry, University of Mississippi, University, MS 38677, USA; ^dMaterials Sciences Division and ^eChemical Sciences Division, Lawrence Berkeley National Laboratory, Berkeley, California 94720, USA; ^fDepartment of Chemistry, North Carolina State University, Raleigh, NC 27695-8204; ^gDepartment of Natural Sciences, Lebanese American University, Chouran, Beirut 1102-2801, Lebanon; ^hHoward Hughes Medical Institute, University of California, Berkeley, California 94720, USA.

chrischang@berkeley.edu, jrlong@berkeley.edu, fncastel@ncsu.edu, mhg@cchem.berkeley.edu

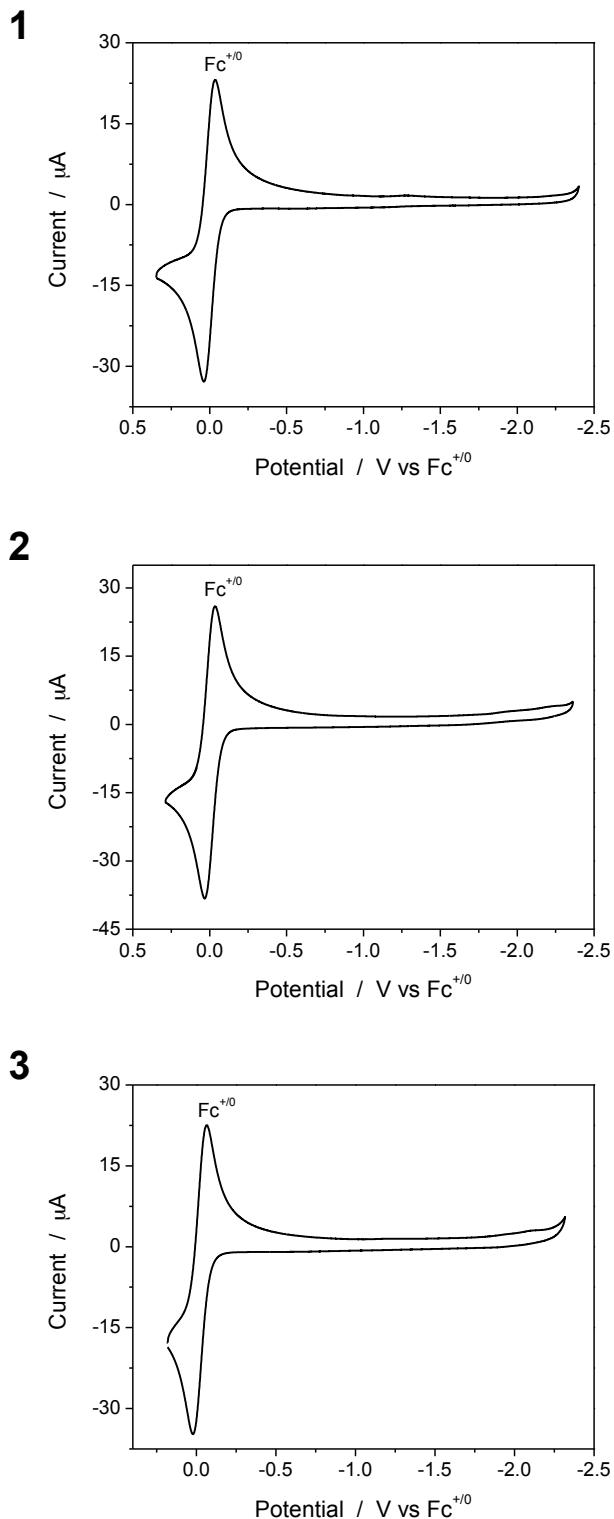


Figure S1. Cyclic voltammograms of ligands **1**, **2**, and **3** at 1 mM concentrations in 0.1 M Bu_4PF_6 CH_3CN with the ferrocene peak included as the internal reference, scan rate = 100 mV/s.

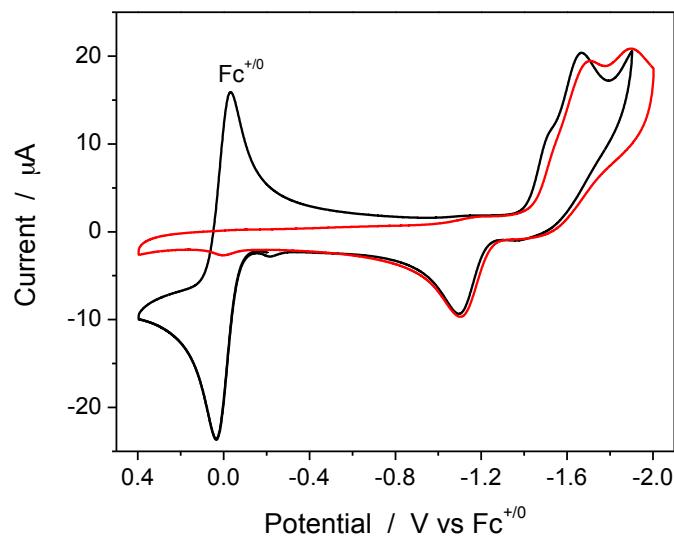


Figure S2. Cyclic voltammograms of 1 mM $[(ax\text{-PY4PZMe}_2)\text{Zn}(\text{OH}_2)](\text{OTf})_2$ (**1-Zn**) in 0.1 M $n\text{Bu}_4\text{PF}_6$ CH_3CN with and without the internal ferrocene reference, scan rate = 100 mV/s.

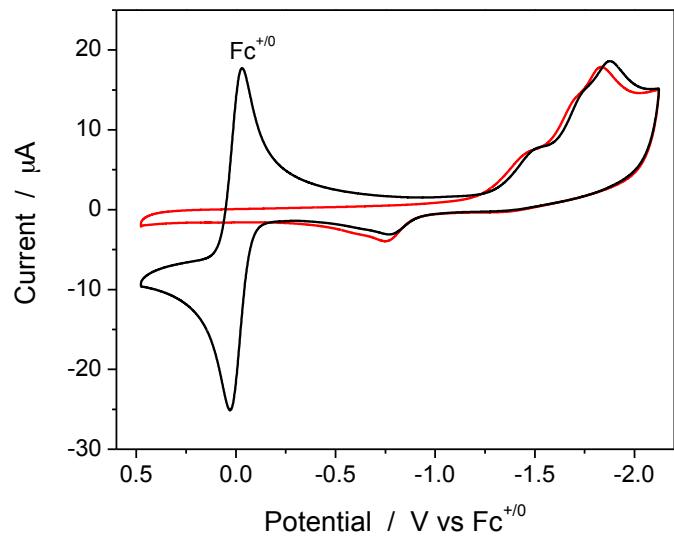


Figure S3. Cyclic voltammograms of 1 mM $[(eq\text{-PY4PZMe}_2)\text{Zn}(\text{OH}_2)](\text{OTf})_2$ (**2-Zn**) in 0.1 M $n\text{Bu}_4\text{PF}_6$ CH_3CN with and without the internal ferrocene reference, scan rate = 100 mV/s.

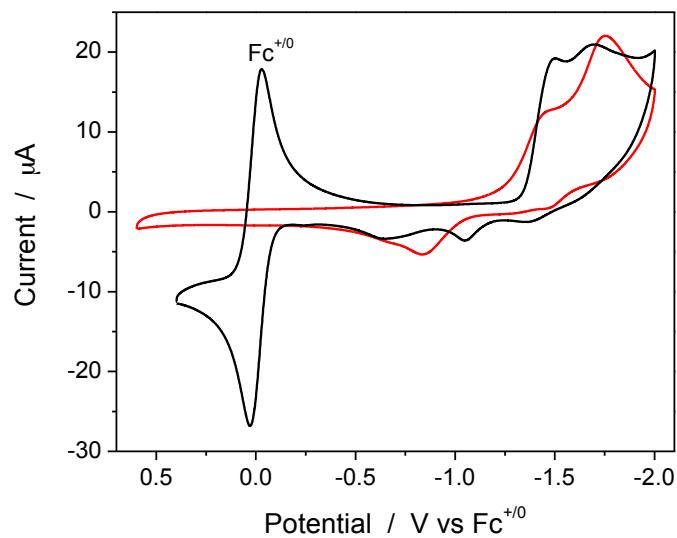


Figure S4. Cyclic voltammograms of 1 mM $[(\text{PY3PZ2Me}_2)\text{Zn}(\text{OH}_2)](\text{OTf})_2$ (**3-Zn**) in 0.1 M $n\text{Bu}_4\text{PF}_6$ CH_3CN with and without the internal ferrocene reference, scan rate = 100 mV/s.

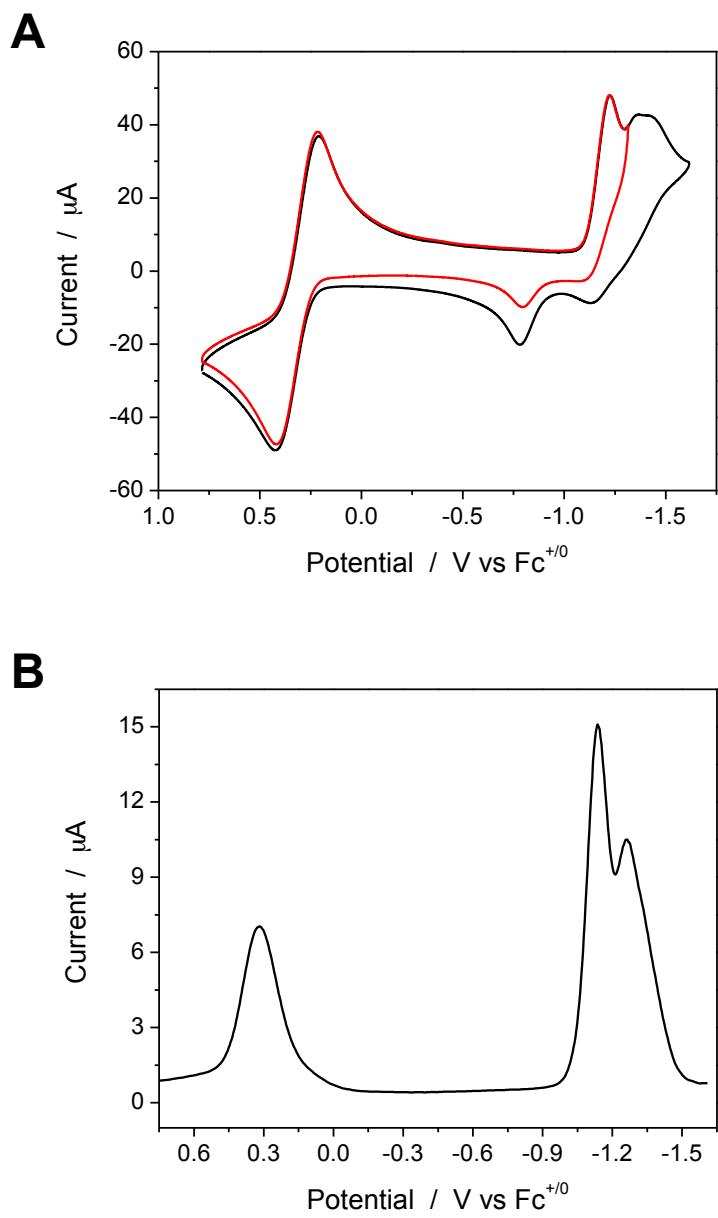


Figure S5. **A.** Cyclic voltammograms of 1 mM $[(ax\text{-PY4PZMe}_2)\text{Co}(\text{CH}_3\text{CN})](\text{OTf})_2$ (**1-Co**) in 0.1 M $n\text{Bu}_4\text{PF}_6$ CH_3CN , scan rate = 100 mV/s. **B.** Square wave voltammogram of the same solution.

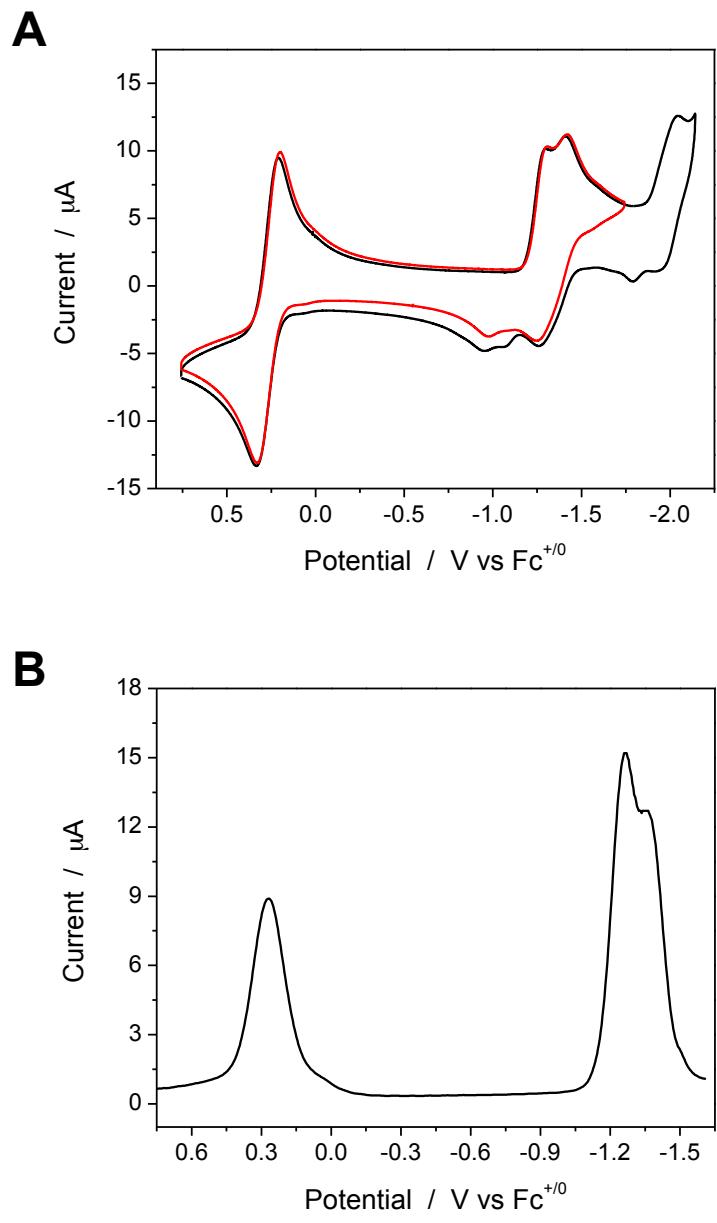


Figure S6. **A.** Cyclic voltammograms of 1 mM $[(eq\text{-PY4PZMe}_2)\text{Co}(\text{CH}_3\text{CN})](\text{OTf})_2$ (**2-Co**) in 0.1 M $n\text{Bu}_4\text{PF}_6$ CH_3CN , scan rate = 100 mV/s. **B.** Square wave voltammogram of the same solution.

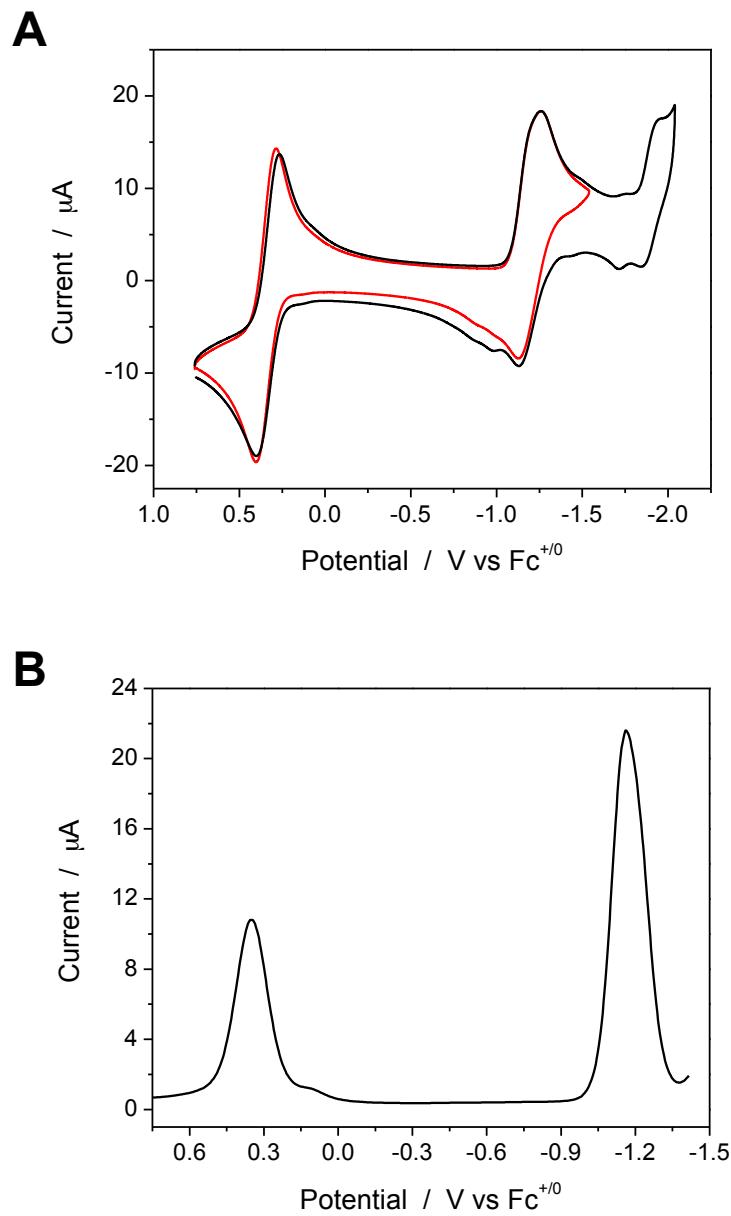


Figure S7. **A.** Cyclic voltammograms of 1 mM $[(\text{PY3PZ2Me}_2)\text{Co}(\text{CH}_3\text{CN})](\text{OTf})_2$ (**3-Co**) in 0.1 M $n\text{Bu}_4\text{PF}_6$ CH_3CN , scan rate = 100 mV/s. **B.** Square wave voltammogram of the same solution.

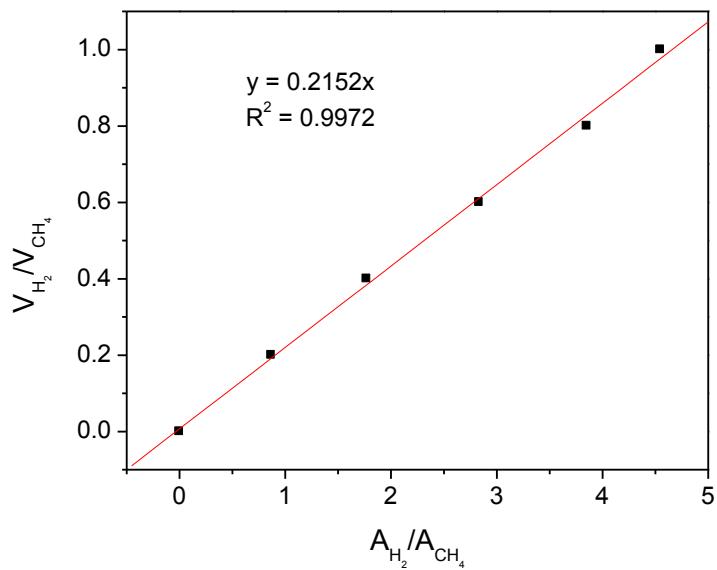


Figure S8. Calibration curve for H_2 quantification by gas chromatography using CH_4 (5 mL) as an internal standard (V: gas volume; A: integrated area of peak signal in the gas chromatogram).

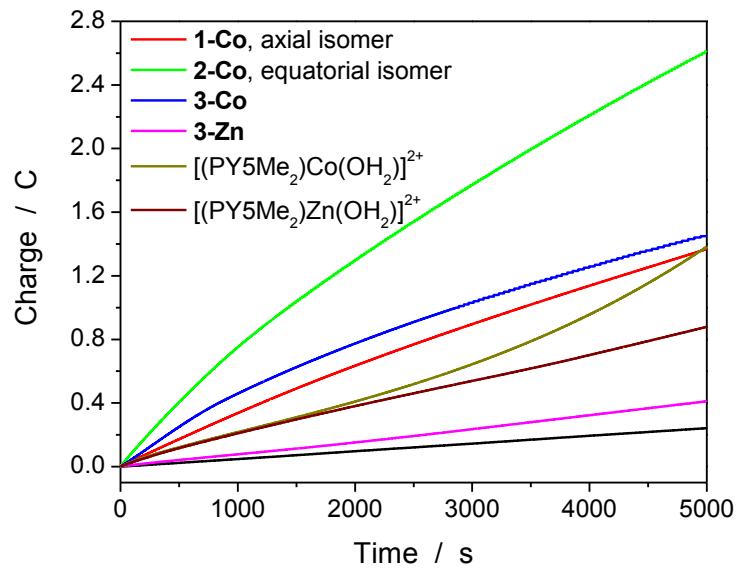


Figure S9. Controlled potential electrolyses of 1 mM complexes in 0.1 M $n\text{Bu}_4\text{PF}_6$ CH_3CN with 100 eq. of chloroacetic acid at a fixed potential of -1.5 V vs Fc^+/Fc . The charge-time profile of **3-Zn** is representative of the series of zinc complexes. The black trace is the background under these conditions. Note: $[(\text{PY5Me}_2)\text{Co}(\text{OH}_2)](\text{OTf})_2$ appears to adsorb onto the electrode during electrolysis and its activity begins to increase over time.

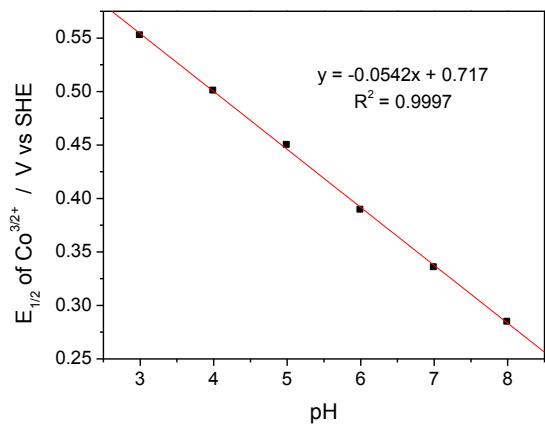
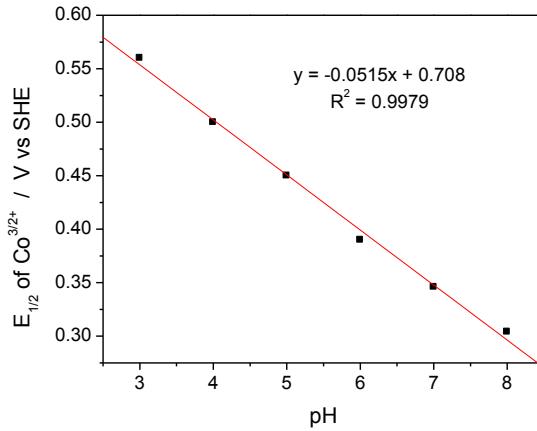
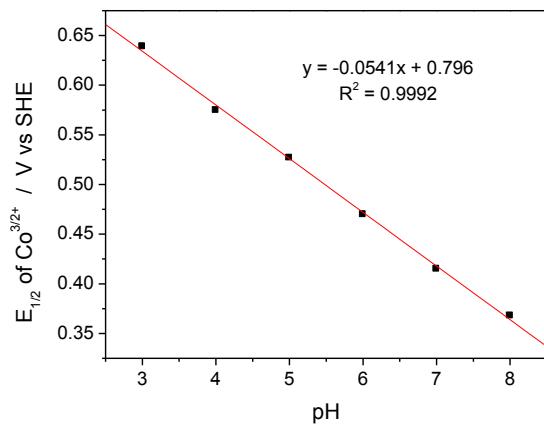
1-Co**2-Co****3-Co**

Figure S10. $E_{1/2}$ of Co(III/II) couple versus pH for **1-Co**, **2-Co**, and **3-Co**. *Conditions:* 0.9 mM catalyst, 0.03 M buffer, 0.1 M KNO_3 , glassy carbon electrode (3 mm dia).

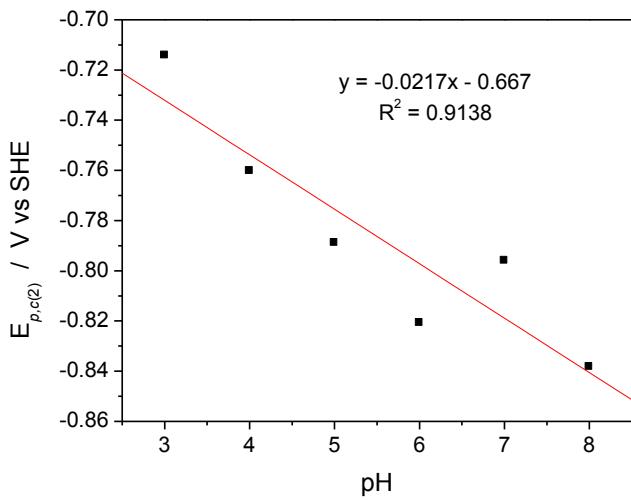


Figure S11. Second reductive peak potential ($E_{p,c(2)}$) vs pH for **1-Co**. *Conditions:* 0.9 mM catalyst, 0.03 M buffer, 0.1 M KNO_3 , glassy carbon electrode (3 mm dia).

1-Zn

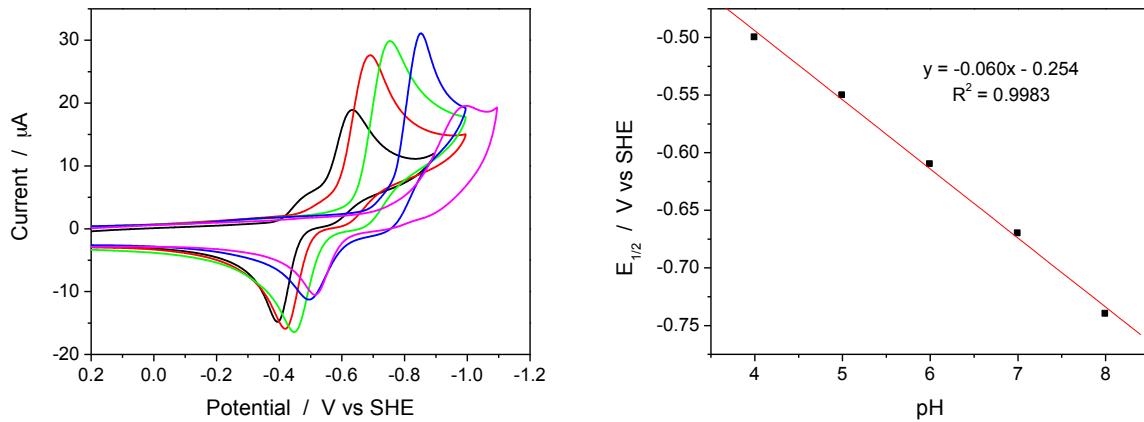


Figure S12. $E_{1/2}$ of skewed ligand-based redox couple versus pH for **1-Zn**. *Conditions:* 0.9 mM complex, 0.03 M buffer, 0.1 M KNO_3 , glassy carbon electrode (3 mm dia).

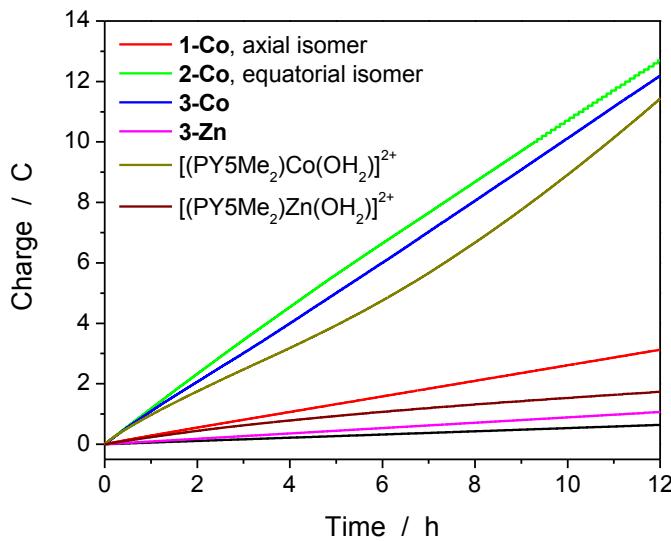


Figure S13. Long-term controlled potential electrolyses in 1M pH 7 KPBS with 10 μM complex at a fixed potential of -1.0 V vs SHE with a Hg pool working electrode. The charge-time profile of **3-Zn** is representative of the series of zinc complexes. The black trace is the background under these conditions.

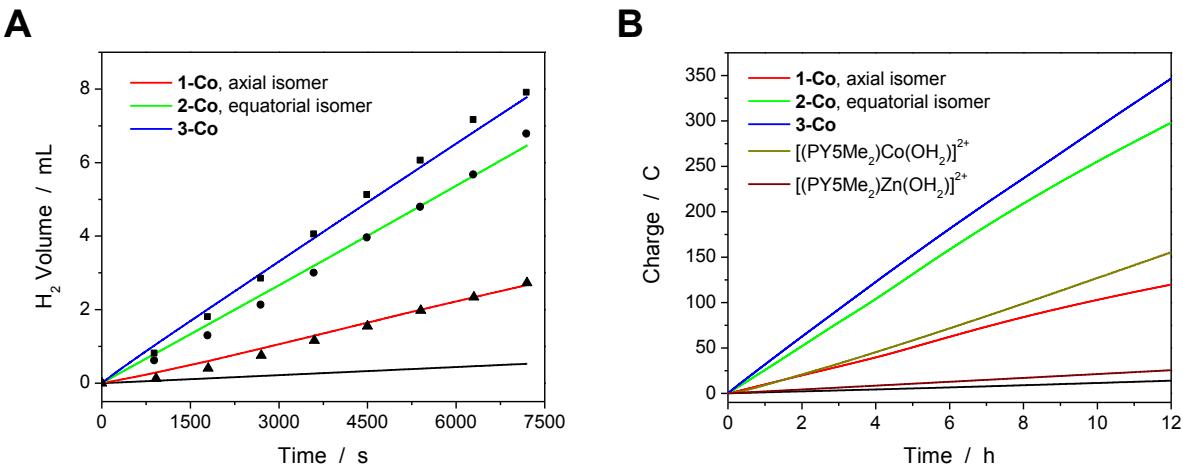


Figure S14. Controlled potential electrolyses in 1M pH 7 KPBS with 10 μM complex at a fixed potential of -1.2 V vs SHE using a Hg pool working electrode. **A.** Black symbols indicate quantified H_2 of headspace samples obtained at various time points over a 2 h period. **B.** Long-term charge-time profiles for 12 h electrolyses. The black traces are the background under these conditions.

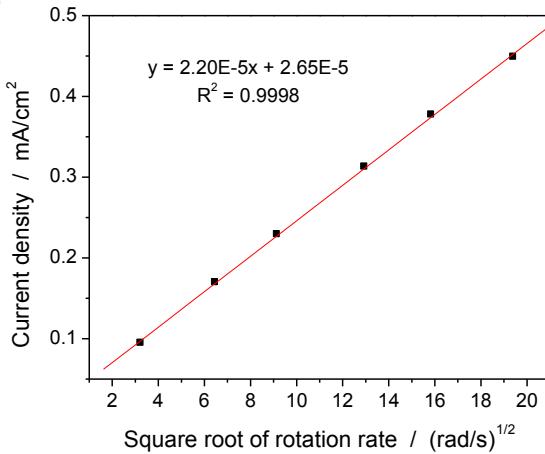
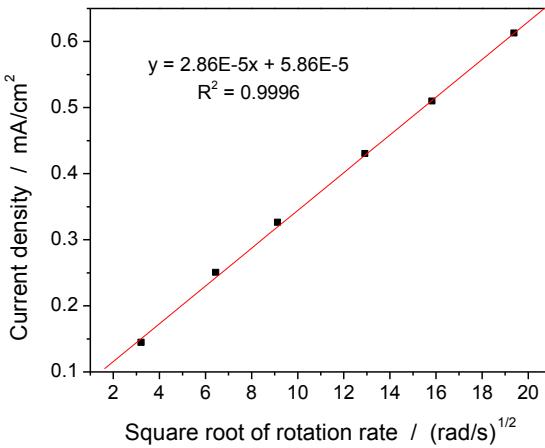
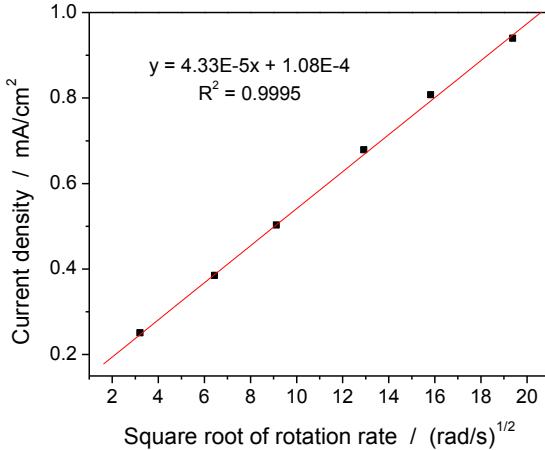
1-Co**2-Co****3-Co**

Figure S15. Levich plots are shown for catalysts **1-Co**, **2-Co**, and **3-Co** for current density at -0.85 V vs SHE versus the square root of the rotation rate. Data is shown in Figure 5. Conditions: 0.3 mM catalyst, 0.1 M pH 7 KPBS, 0.1 M KNO₃, glassy carbon electrode.

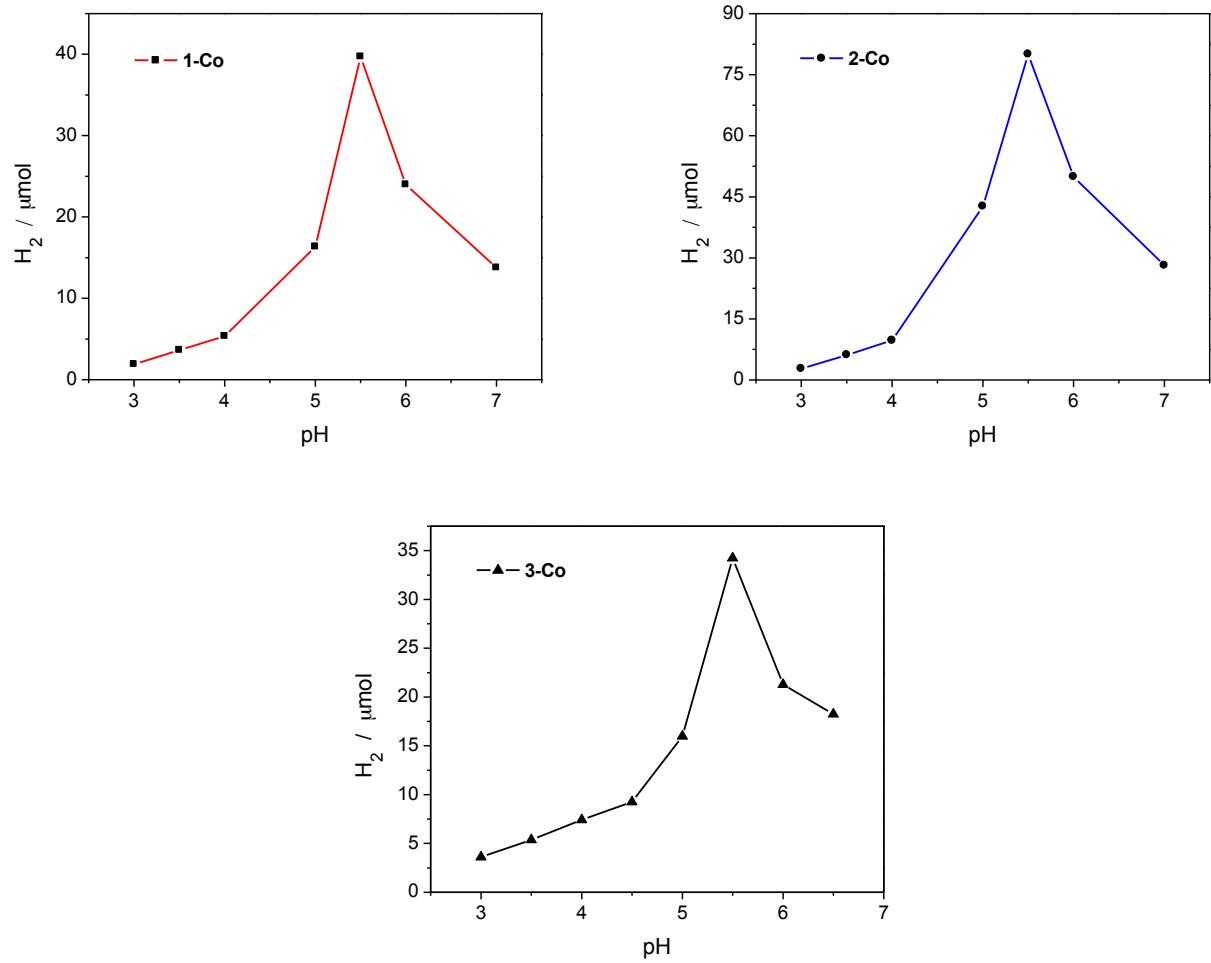


Figure S16. Final accumulated H_2 measured after 18 h of photocatalysis as a function of pH.
Conditions: $2.0 \times 10^{-5} \text{ M}$ Co(II) catalyst, $3.3 \times 10^{-4} \text{ M}$ $[\text{Ru}(\text{bpy})_3]^{2+}$ and 0.3 M $\text{H}_2\text{A}/\text{HA}^-$.

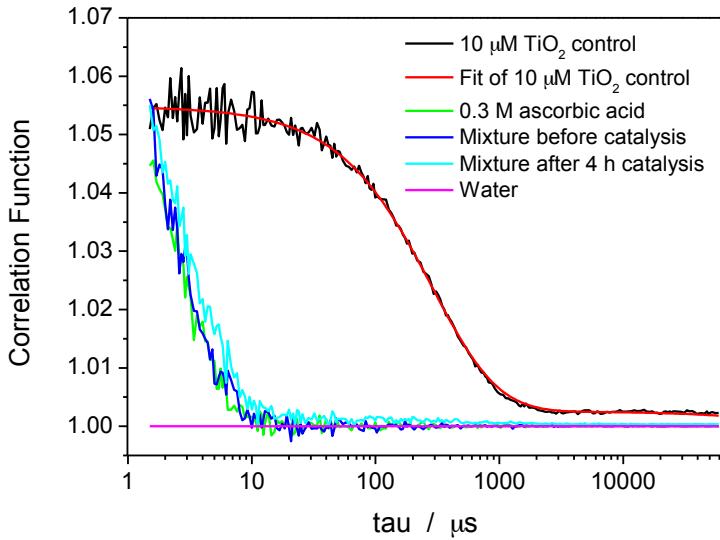


Figure S17. Dynamic Light Scattering (DLS) before and after 4 hours of irradiation under 452 ± 10 nm (540 mW). *Conditions:* 3.3×10^{-4} M $[\text{Ru}(\text{bpy})_3]^{2+}$, 2×10^{-5} M **2-Co**, 0.3 M ascorbic acid at pH 5.5. Degussa p25 titanium dioxide (1.0×10^{-5} M) was used as the standard with its autocorrelation fit to detect any particles by light scattering. The absence of nanoparticles in the mixture above 0.5 nm in radius, before and after catalysis, resulted in no light scattering.

Computational Results

All free energies (calculated at 298.15 K and 1 atm) include zero-point-vibrational energy and are corrected using the modified harmonic oscillator approximation proposed by Grimme.¹ This method interpolates between vibrational S_v and rotational S_R approximations to the entropy contribution of low-lying modes. The entropy S for low-frequency modes is given by:

$$S = w(\omega)S_v + [1 - w(\omega)]S_R \quad (1)$$

where the $w(\omega)$ is the Head-Gordon damping function with $\alpha = 4$:

$$w(\omega) = \frac{1}{1 + (\omega_0/\omega)^\alpha} \quad (2)$$

As proposed by Grimme, a cutoff value of $\omega_0 = 100$ cm⁻¹ was chosen. In this case, the vibrational entropy for all modes with frequencies lower than 100 cm⁻¹ is replaced by a corresponding free-rotor entropy.

Table S1. Experimental and calculated (B3LYP, B3LYP-D2 and ω B97X-D) bond lengths (Ångströms) in the gas phase for **1-Co** ($S = 3/2$) and **1-Zn** ($S = 0$).

| Complexes | Bond Lengths | M-N_{ax} | M-O | avg M-N_{eq} |
|--|--------------------------|-------------------------|------------|-----------------------------|
| 1-Co [(<i>ax</i> -PY4PZMe ₂)Co(OH ₂)] ²⁺ | Exp. | 2.1050(13) | 2.0342(12) | 2.1415(13) |
| | Calc. (B3LYP) | 2.109 | 2.182 | 2.177 |
| | Calc. (B3LYP-D2) | 2.110 | 2.136 | 2.157 |
| | Calc. (ω B97X-D) | 2.103 | 2.161 | 2.157 |
| 1-Zn [(<i>ax</i> -PY4PZMe ₂)Zn(OH ₂)] ²⁺ | Exp. | 2.147(3) | 2.039(3) | 2.160(4) |
| | Calc. (B3LYP) | 2.117 | 2.240 | 2.190 |
| | Calc. (B3LYP-D2) | 2.121 | 2.189 | 2.170 |
| | Calc. (ω B97X-D) | 2.108 | 2.223 | 2.166 |

Table S2. Experimental and calculated (B3LYP, B3LYP-D2 and ω B97X-D) bond lengths (Ångströms) in the gas phase for **2-Co** ($S = 3/2$) and **2-Zn** ($S = 0$).

| Complexes | Bond Lengths | M-N_{ax} | M-O | avg M-N_{eq} |
|--|--------------------------|-------------------------|------------|-----------------------------|
| 2-Co [(<i>eq</i> -PY4PZMe ₂)Co(OH ₂)] ²⁺ | Exp. | 2.099(2) | 2.0316(19) | 2.127(2) |
| | Calc. (B3LYP) | 2.108 | 2.193 | 2.171 |
| | Calc. (B3LYP-D2) | 2.108 | 2.147 | 2.151 |
| | Calc. (ω B97X-D) | 2.102 | 2.172 | 2.152 |
| 2-Zn [(<i>eq</i> -PY4PZMe ₂)Zn(OH ₂)] ²⁺ | Exp. | 2.1368(19) | 2.0608(17) | 2.150(2) |
| | Calc. (B3LYP) | 2.112 | 2.267 | 2.185 |
| | Calc. (B3LYP-D2) | 2.118 | 2.208 | 2.165 |
| | Calc. (ω B97X-D) | 2.105 | 2.239 | 2.162 |

Table S3. Experimental and calculated (B3LYP, B3LYP-D2 and ω B97X-D) bond lengths (Ångströms) in the gas phase for **3-Co** ($S = 3/2$) and **3-Zn** ($S = 0$).

| Complexes | Bond Lengths | M-N_{ax} | M-O | avg M-N_{eq} |
|---|--------------------------|-------------------------|------------|-----------------------------|
| 3-Co [(PY3PZ2Me ₂)Co(OH ₂)] ²⁺ | Exp. | 2.094(3) | 2.016(3) | 2.113(3) |
| | Calc. (B3LYP) | 2.111 | 2.189 | 2.171 |
| | Calc. (B3LYP-D2) | 2.111 | 2.142 | 2.151 |
| | Calc. (ω B97X-D) | 2.104 | 2.167 | 2.153 |
| 3-Zn [(PY3PZ2Me ₂)Zn(OH ₂)] ²⁺ | Exp. | 2.141(2) | 2.0474(18) | 2.157(2) |
| | Calc. (B3LYP) | 2.114 | 2.254 | 2.187 |
| | Calc. (B3LYP-D2) | 2.119 | 2.200 | 2.166 |
| | Calc. (ω B97X-D) | 2.105 | 2.233 | 2.164 |

Table S4. Calculated relative energies between the doublet and quartet states for **1'-Co**, **2'-Co** and **3'-Co** using the B3LYP/BS2 functional with and without dispersion corrections in the gas phase and in acetonitrile (SWIG C-PCM approach). The Mulliken spin population is given for cobalt while $\langle S^2 \rangle$ corresponds to the spin contamination.

| Complexes | Multiplicity | $\langle S^2 \rangle$ | $\rho(\text{Co})$ | ΔG_{B3LYP} | $\Delta G_{\text{B3LYP/C-PCM}}$ | $\Delta G_{\text{B3LYP-D2}}$ | $\Delta G_{\text{B3LYP-D2/C-PCM}}$ |
|---|----------------|-----------------------|-------------------|---------------------------|---------------------------------|------------------------------|------------------------------------|
| 1'-Co | S = 1/2 | 0.76 | 0.95 | 0.0 | 0.0 | 0.0 | 0.0 |
| $[(ax\text{-PY4PZMe}_2)\text{Co}(\text{CH}_3\text{CN})]^{2+}$ | S = 3/2 | 3.76 | 2.70 | -3.7 | -3.5 | -2.0 | -1.8 |
| 2'-Co | S = 1/2 | 0.76 | 0.95 | 0.0 | 0.0 | 0.0 | 0.0 |
| $[(eq\text{-PY4PZMe}_2)\text{Co}(\text{CH}_3\text{CN})]^{2+}$ | S = 3/2 | 3.76 | 2.70 | -3.5 | -3.5 | -1.8 | -1.9 |
| 3'-Co | S = 1/2 | 0.76 | 0.95 | 0.0 | 0.0 | 0.0 | 0.0 |
| $[(PY3PZ2\text{Me}_2)\text{Co}(\text{CH}_3\text{CN})]^{2+}$ | S = 3/2 | 3.76 | 2.70 | -3.5 | -3.5 | -1.7 | -1.8 |

Table S5. Calculated relative energies between the singlet and triplet states for the five- and six-coordinate species of **1'-Co+e⁻**, **2'-Co+e⁻** and **3'-Co+e⁻** using the B3LYP/BS2 functional with and without dispersion corrections in the gas phase and in acetonitrile (SWIG C-PCM approach). The Mulliken spin population is given for cobalt while $\langle S^2 \rangle$ corresponds to the spin contamination.

| Complexes | Multiplicity | $\langle S^2 \rangle$ | $\rho(\text{Co})$ | ΔG_{B3LYP} | $\Delta G_{\text{B3LYP/C-PCM}}$ | $\Delta G_{\text{B3LYP-D2}}$ | $\Delta G_{\text{B3LYP-D2/C-PCM}}$ |
|---|--------------|-----------------------|-------------------|---------------------------|---------------------------------|------------------------------|------------------------------------|
| 1'-Co+e⁻ | S = 0 | 1.08 | 0.33 | 0.0 | 0.0 | 0.0 | 0.0 |
| $[(ax\text{-PY4PZMe}_2)\text{Co}(\text{CH}_3\text{CN})]^{1+}$ | S = 1 | 2.21 | 2.14 | -18.0 | -16.9 | -17.7 | -16.6 |
| 1'-Co+e⁻ | S = 0 | 0.89 | 0.03 | +1.3 | -0.9 | +7.9 | +5.7 |
| $[(ax\text{-PY4PZMe}_2)\text{Co}]^{1+}$ | S = 1 | 2.14 | 2.08 | -12.8 | -15.9 | -5.5 | -8.6 |
| 2'-Co+e⁻ | S = 0 | 1.01 | 0.08 | 0.0 | 0.0 | 0.0 | 0.0 |
| $[(eq\text{-PY4PZMe}_2)\text{Co}(\text{CH}_3\text{CN})]^{1+}$ | S = 1 | 2.14 | 2.07 | -16.5 | -16.5 | -16.5 | -16.4 |
| 2'-Co+e⁻ | S = 0 | 0.89 | 0.16 | +1.8 | -2.3 | +8.2 | +4.1 |
| $[(eq\text{-PY4PZMe}_2)\text{Co}]^{1+}$ | S = 1 | 2.12 | 2.06 | -11.6 | -16.0 | -4.5 | -9.0 |
| 3'-Co+e⁻ | S = 0 | 1.01 | 0.05 | 0.0 | 0.0 | 0.0 | 0.0 |
| $[(PY3PZ2\text{Me}_2)\text{Co}(\text{CH}_3\text{CN})]^{1+}$ | S = 1 | 2.16 | 2.10 | -16.5 | -16.6 | -16.4 | -16.6 |
| 3'-Co+e⁻ | S = 0 | 0.87 | 0.15 | +2.3 | -2.0 | +8.7 | +4.3 |
| $[(PY3PZ2\text{Me}_2)\text{Co}]^{1+}$ | S = 1 | 2.12 | 2.06 | -11.0 | -15.8 | -4.0 | -8.7 |

Table S6. Calculated relative energies between the doublet and quartet states for the five- and six-coordinate species of **1'-Co+2e⁻**, **2'-Co+2e⁻** and **3'-Co+2e⁻** using the B3LYP/BS2 functional with and without dispersion corrections in the gas phase and in acetonitrile (SWIG C-PCM approach). The Mulliken spin population is given for cobalt while $\langle S^2 \rangle$ corresponds to the spin contamination.

| Complexes | Multiplicity | $\langle S^2 \rangle$ | $\rho(\text{Co})$ | ΔG_{B3LYP} | $\Delta G_{\text{B3LYP/C-PCM}}$ | $\Delta G_{\text{B3LYP-D2}}$ | $\Delta G_{\text{B3LYP-D2/C-PCM}}$ |
|--|--------------|-----------------------|-------------------|---------------------------|---------------------------------|------------------------------|------------------------------------|
| 1'-Co+2e⁻ | S = 1/2 | 1.83 | 2.00 | 0.0 | 0.0 | 0.0 | 0.0 |
| $[(ax\text{-PY4PZMe}_2)\text{Co}(\text{CH}_3\text{CN})]^0$ | S = 3/2 | 3.86 | 2.08 | -0.8 | -0.9 | -1.0 | -1.1 |
| 1'-Co+2e⁻ | S = 1/2 | 1.94 | 2.09 | +0.7 | -0.1 | +7.4 | +6.6 |
| $[(ax\text{-PY4PZMe}_2)\text{Co}]^0$ | S = 3/2 | 3.86 | 2.08 | -0.6 | -2.0 | +6.1 | +4.7 |
| 2'-Co+2e⁻ | S = 1/2 | 1.94 | 2.10 | 0.0 | 0.0 | 0.0 | 0.0 |
| $[(eq\text{-PY4PZMe}_2)\text{Co}(\text{CH}_3\text{CN})]^0$ | S = 3/2 | 3.87 | 2.09 | -0.2 | 0.0 | -0.2 | 0.0 |
| 2'-Co+2e⁻ | S = 1/2 | 2.00 | 2.18 | -2.1 | -3.2 | +5.3 | +4.2 |
| $[(eq\text{-PY4PZMe}_2)\text{Co}]^0$ | S = 3/2 | 3.94 | 2.16 | -1.5 | -3.5 | +5.6 | +3.5 |
| 3'-Co+2e⁻ | S = 1/2 | 2.00 | 2.15 | 0.0 | 0.0 | 0.0 | 0.0 |
| $[(PY3PZ2Me_2)\text{Co}(\text{CH}_3\text{CN})]^0$ | S = 3/2 | 3.90 | 2.11 | -0.4 | -0.9 | -0.5 | -1.0 |
| 3'-Co+2e⁻ | S = 1/2 | 1.98 | 2.15 | -0.9 | -1.9 | +6.4 | +5.5 |
| $[(PY3PZ2Me_2)\text{Co}]^0$ | S = 3/2 | 3.94 | 2.16 | -1.2 | -3.3 | +5.9 | +3.8 |

DFT calculations suggest that the cobalt(II) complexes **1'-Co**, **2'-Co** and **3'-Co** have a quartet ground-state (Table S4). The first reduction yields a Co^{1+} species (Figures S21-26) with a triplet ground-state (Table S5). Dissociation of the acetonitrile solvent molecule has also been considered after the first reduction. Calculations suggest that formation of a five-coordinate species is higher in energy than the six-coordinate complexes in solution ($\Delta G_{\text{B3LYP/C-PCM}} \sim +1.0$ kcal/mol; $\Delta G_{\text{B3LYP-D2/C-PCM}} \sim +8.0$ kcal/mol). The second reduction is computed to be ligand-centered (see below) for both doublet and quartet states in which five- and six-coordinate species are competitive (Table S6). For instance, DFT calculations performed using the B3LYP functional (S = 3/2) suggest that formation of the five-coordinate complex are lower in energy ($\Delta G_{\text{B3LYP/C-PCM}} = -2.0$ kcal/mol for **1'-Co+2e⁻**; $\Delta G_{\text{B3LYP/C-PCM}} = -3.5$ kcal/mol for **2'-Co+2e⁻**; $\Delta G_{\text{B3LYP/C-PCM}} = -3.3$ kcal/mol for **3'-Co+2e⁻**). However, calculations using the B3LYP-D2 functional (S = 3/2) show that the six-coordinate complexes are slightly higher in energy ($\Delta G_{\text{B3LYP-D2/C-PCM}} = +4.7$ kcal/mol for **1'-Co+2e⁻**; $\Delta G_{\text{B3LYP-D2/C-PCM}} = +3.5$ kcal/mol for **2'-Co+2e⁻**; $\Delta G_{\text{B3LYP-D2/C-PCM}} = +3.8$ kcal/mol for **3'-Co+2e⁻**). These observations suggest that a competition between five- and six-coordinate species may play a role in catalysis.

Table S7. Calculated redox potentials (V vs. Fc⁺/Fc) for the Co²⁺ complexes (S = 3/2) and their one- (S = 1) and two-electron reduced species (S = 1/2 and 3/2) in solution (acetonitrile *via* C-PCM approach). In the case of the second reduction, the redox potentials were calculated for the five-coordinate species after dissociation of the acetonitrile molecule. Redox potentials for the six-coordinate species are given in the main text.

| Complex | $E_{p1,c}$ (S = 1) | | $E_{p2,c}$ (S = 1/2 and 3/2) | | |
|---|--------------------|--------------------|------------------------------|---------------------|---------------------|
| | exptl | calcd | exptl | calcd | calcd |
| 1'-Co , [(ax-PY4PZMe ₂)Co(CH ₃ CN)] ²⁺ | -1.22 | -1.28 | -1.40 | -1.44 ^b | -1.43 ^b |
| 2'-Co , [(eq-PY4PZMe ₂)Co(CH ₃ CN)] ²⁺ | -1.30 | -1.34 | -1.42 | -1.36 ^b | -1.41 ^b |
| 3'-Co , [(PY3PZ2Me ₂)Co(CH ₃ CN)] ²⁺ | -1.18 | -1.18 ^a | -1.25 | -1.25 ^{ab} | -1.25 ^{ab} |

^aThis redox potential was used as reference in the isodesmic reactions, so it agrees by construction, and all other reduction potentials are calculated relative to this value.

^bDissociation of the acetonitrile molecule.

Edmiston-Ruedenberg localized orbitals for **1'-Co**, **2'-Co** and **3'-Co**.

In order to gain information on the oxidation state of the cobalt complexes **1'-Co**, **2'-Co** and **3'-Co**, localized orbital bonding analysis (LOBA) using the Edmiston-Ruedenberg localized orbitals and the Löwdin population analysis were employed (Figures S18-20). LOBA calculations suggest that **1'-Co**, **2'-Co** and **3'-Co** have an oxidation state of +2.

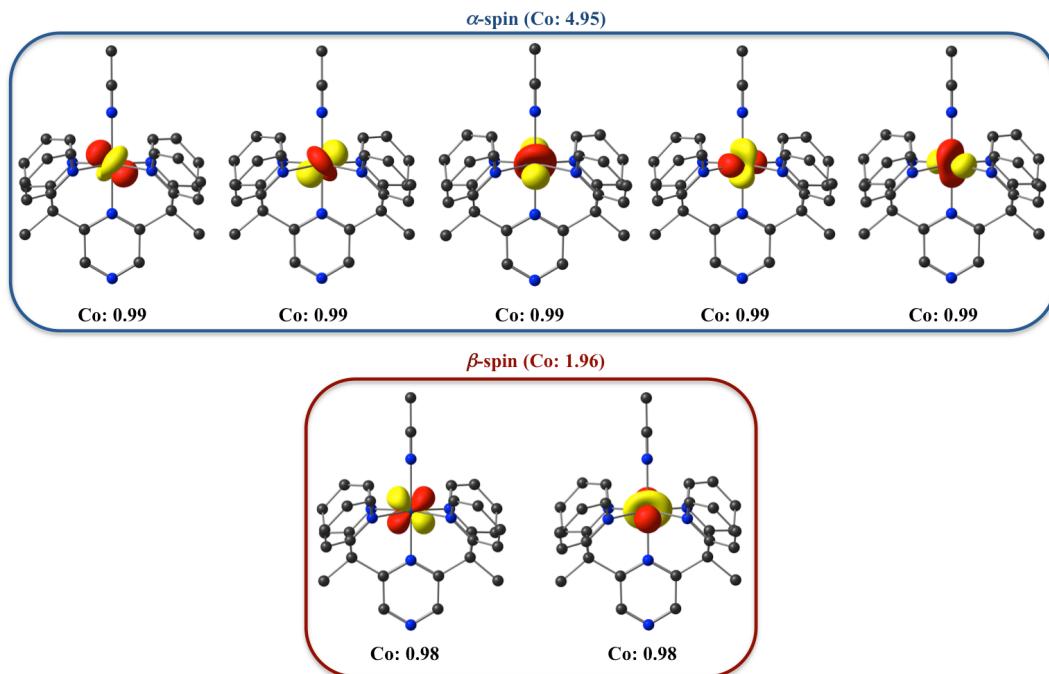


Figure S18. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for **1'-Co** (Co²⁺, S = 3/2) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

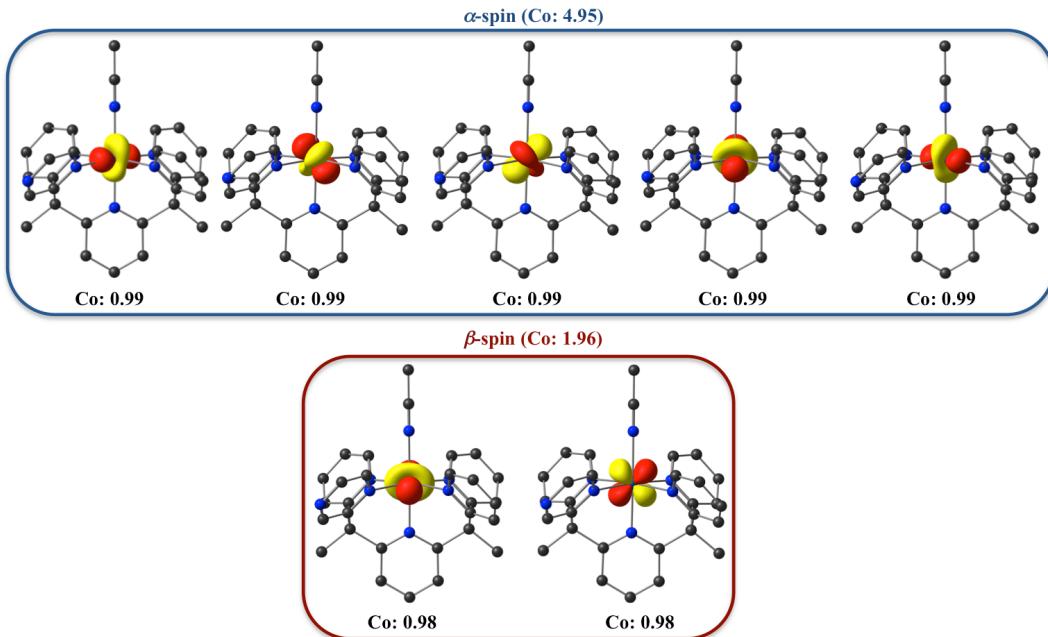


Figure S19. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for **2'-Co** (Co^{2+} , $S = 3/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

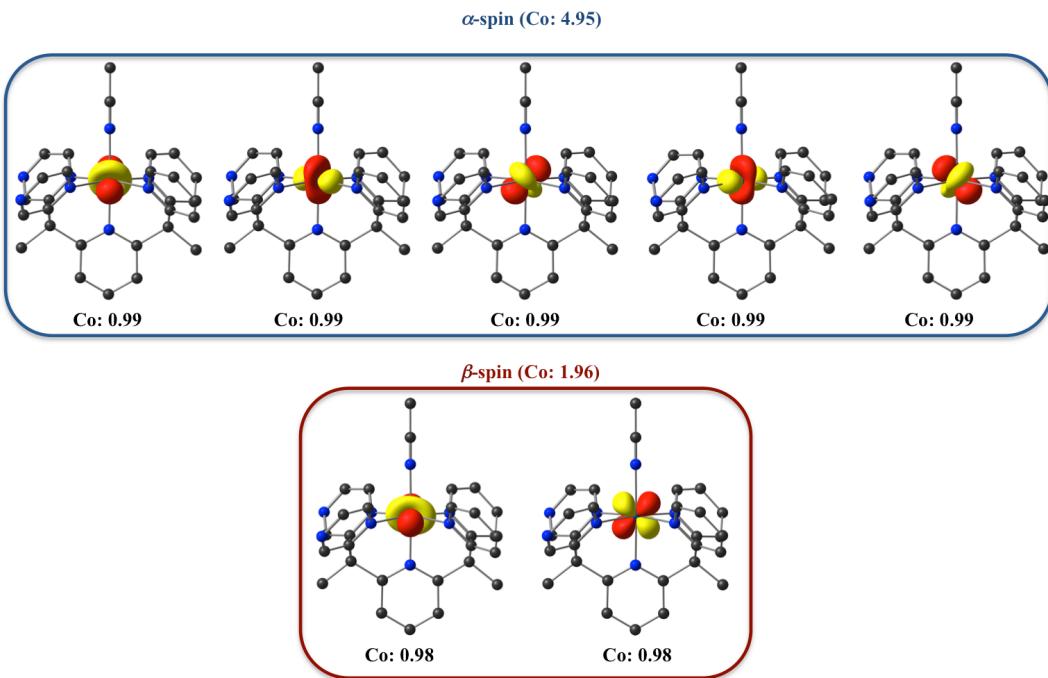


Figure S20. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for **3'-Co** (Co^{2+} , $S = 3/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

Edmiston-Ruedenberg localized orbitals and canonical molecular orbitals for 1'-Co+e⁻, 2'-Co+e⁻ and 3'-Co+e⁻ obtained from 1'-Co, 2'-Co and 3'-Co after one-electron reduction.

Edmiston-Ruedenberg localized orbitals (Figures S21-23) and canonical molecular orbitals (Figures S24-26) have been computed for the one-electron reduced species (**1'-Co+e⁻**, **2'-Co+e⁻** and **3'-Co+e⁻** obtained from **1'-Co**, **2'-Co** and **3'-Co**). DFT calculations ($S = 1$) suggest that the first reduction is mainly metal-centered.

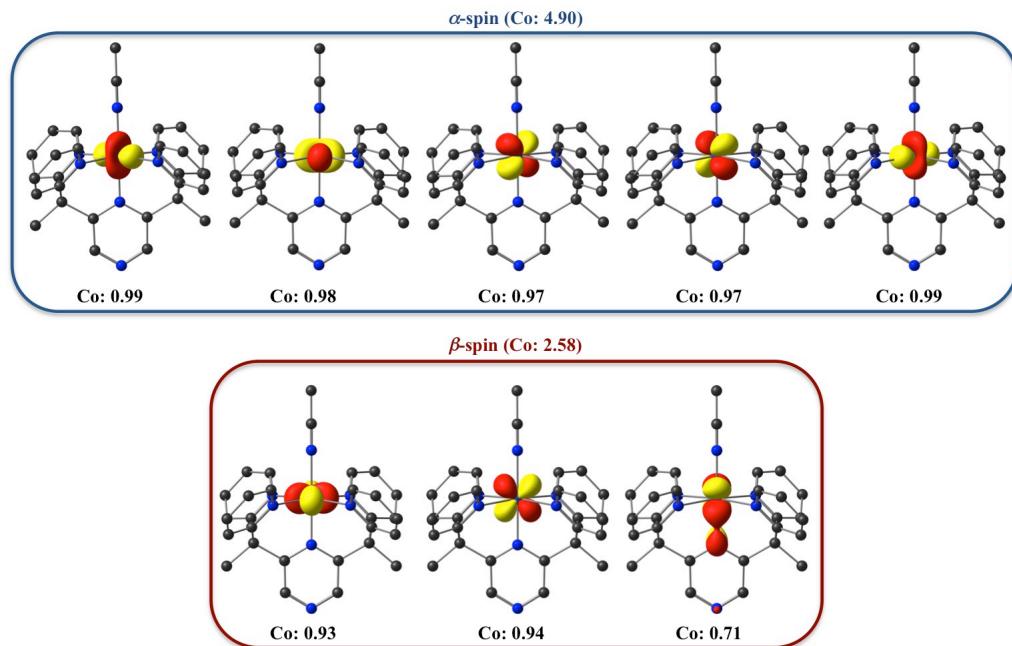


Figure S21. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for **1'-Co+e⁻** ($S = 1$) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

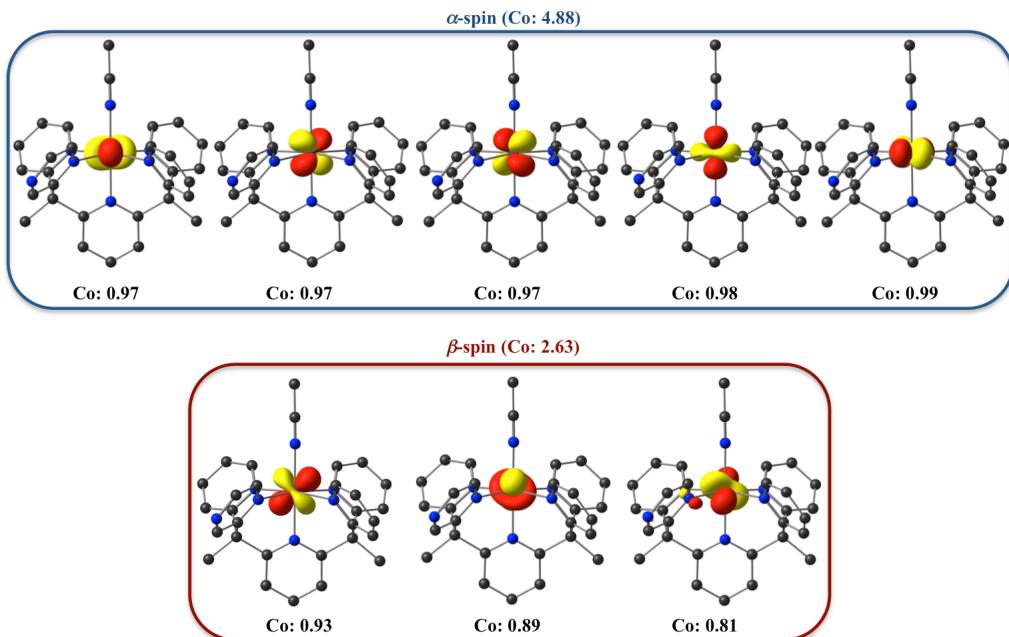


Figure S22. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $2'\text{-Co}+\text{e}^-$ ($S = 1$) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

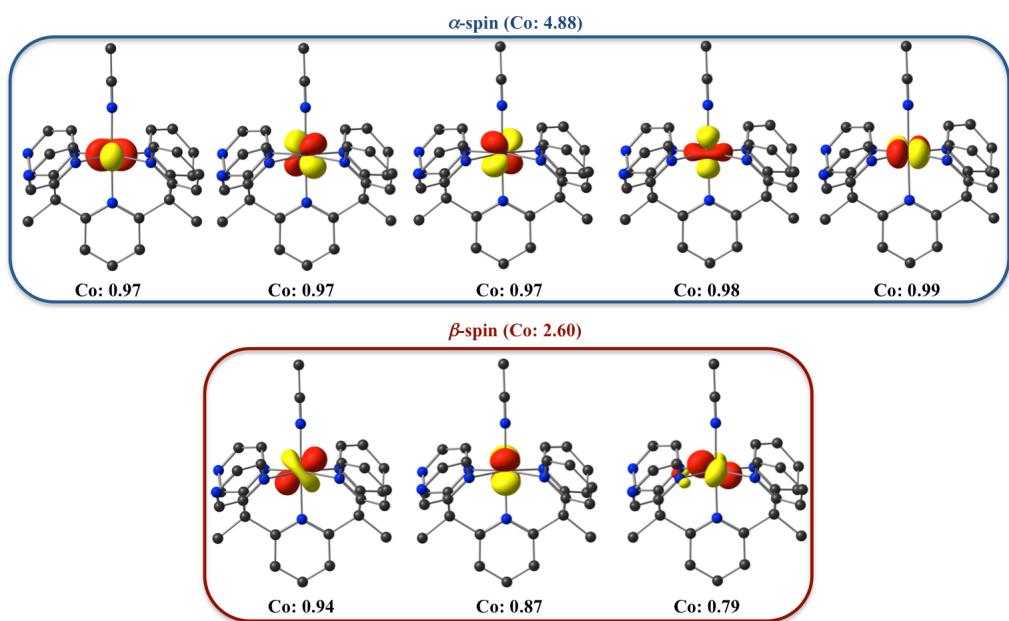


Figure S23. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $3'\text{-Co}+\text{e}^-$ ($S = 1$) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

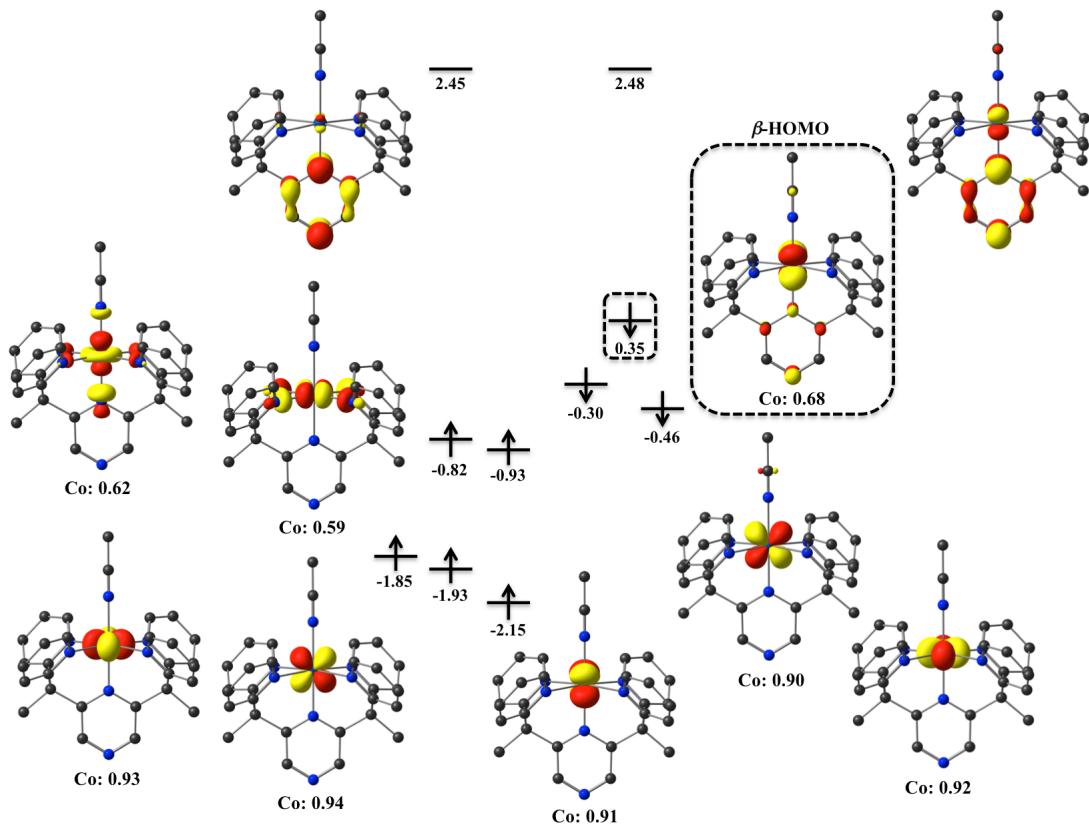


Figure S24. Isosurface (0.07 au) plots of the canonical molecular orbitals for **1'-Co+e⁻** (S = 1) using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of **3'-Co+e⁻** (S = 1).

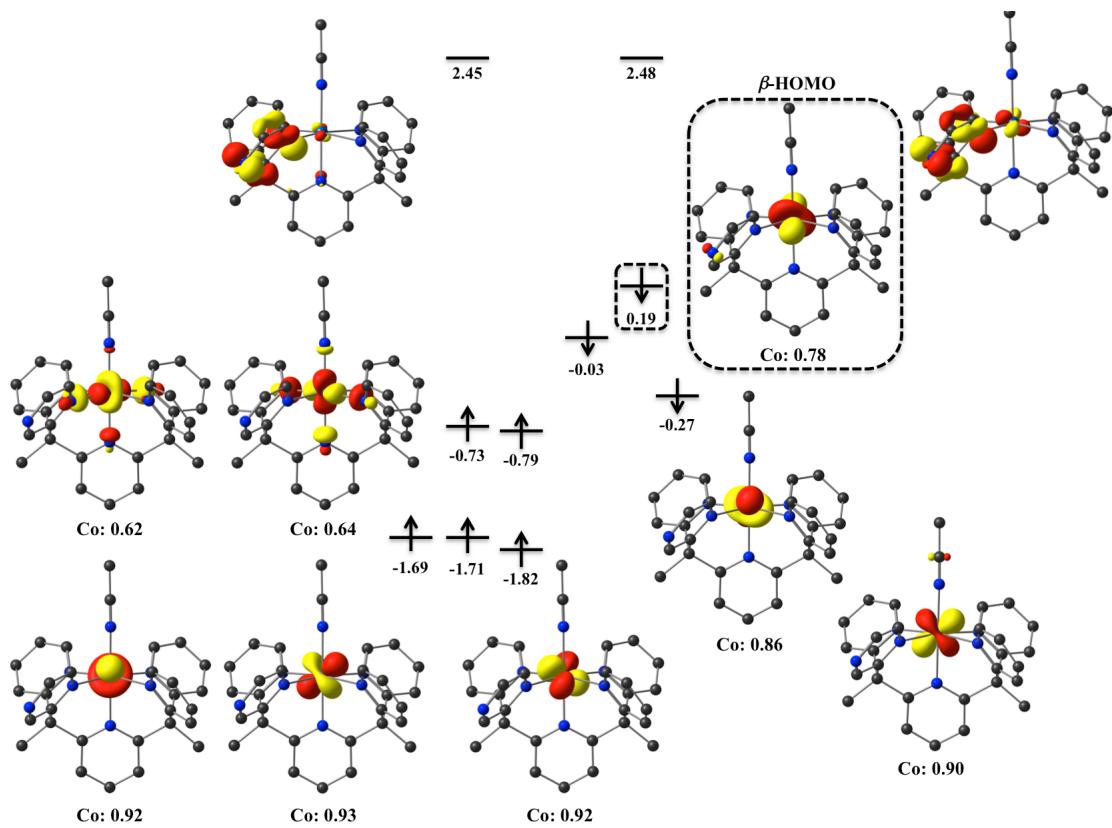


Figure S25. Isosurface (0.07 au) plots of the canonical molecular orbitals for **2'-Co+e⁻** (S = 1) using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of **3'-Co+e⁻** (S = 1).

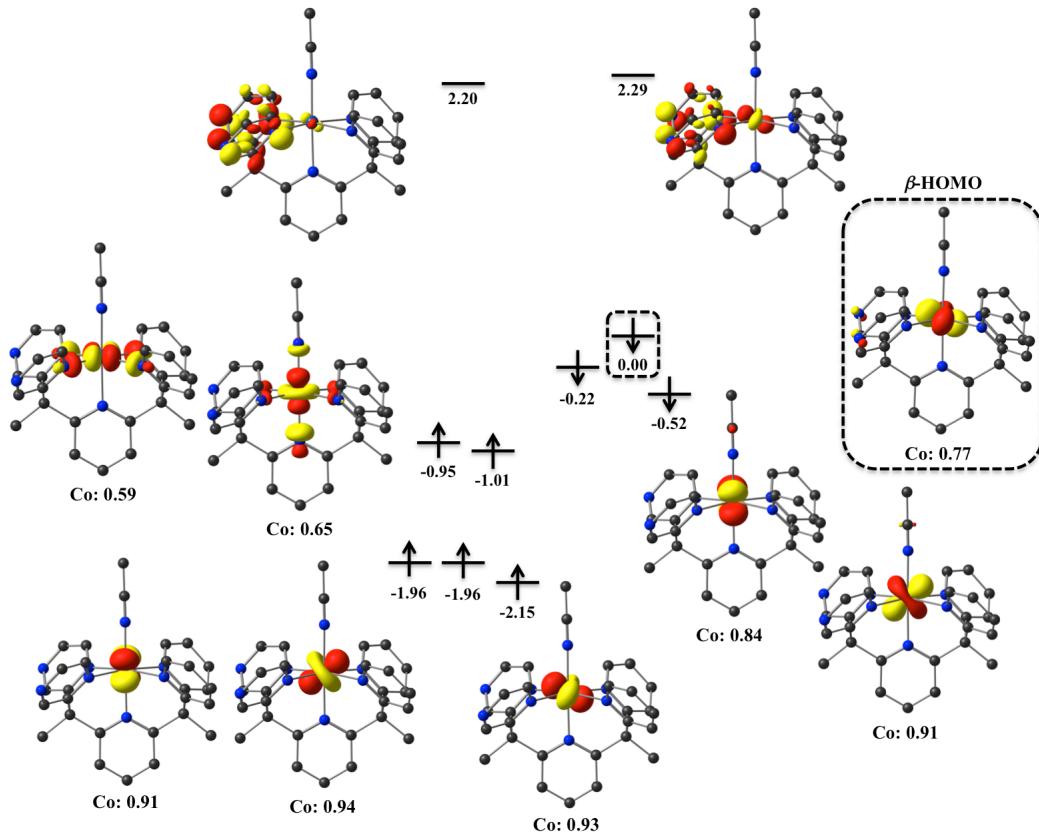


Figure S26. Isosurface (0.07 au) plots of the canonical molecular orbitals for **3'-Co+e⁻** ($S = 1$) using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital).

Edmiston-Ruedenberg localized orbitals and canonical molecular orbitals for the six-coordinate species **1'-Co+2e⁻, **2'-Co+2e⁻** and **3'-Co+2e⁻** obtained after two-electron reduction.**

Edmiston-Ruedenberg localized orbitals (Figures S27-32) and canonical molecular orbitals (Figures S33-38) have been computed for the two-electron reduced species to give **1'-Co+2e⁻**, **2'-Co+2e⁻** and **3'-Co+2e⁻**. As shown in Table S6, the doublet and quartet states are degenerate, however, the second reduction is computed to be ligand-centered in both states for all three species. Interestingly, the LUMOs (lowest unoccupied molecular orbitals) are computed to be ligand-centered. This suggests that the third reduction will be ligand-centered. Similar conclusions have been observed for the five-coordinate species (see below).

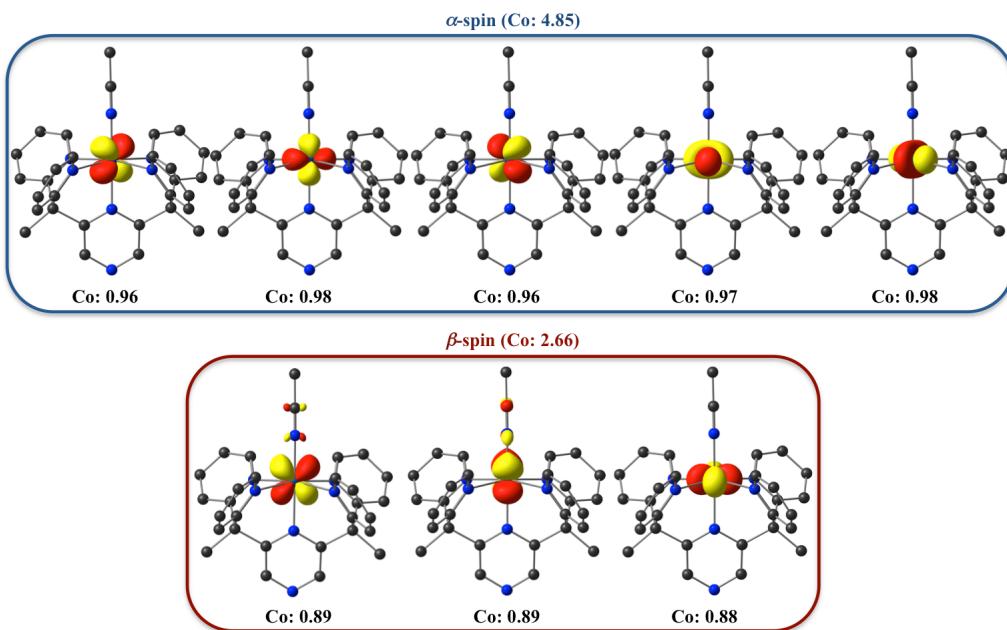


Figure S27. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $1'\text{-Co}+2\text{e}^-$ ($S = 1/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

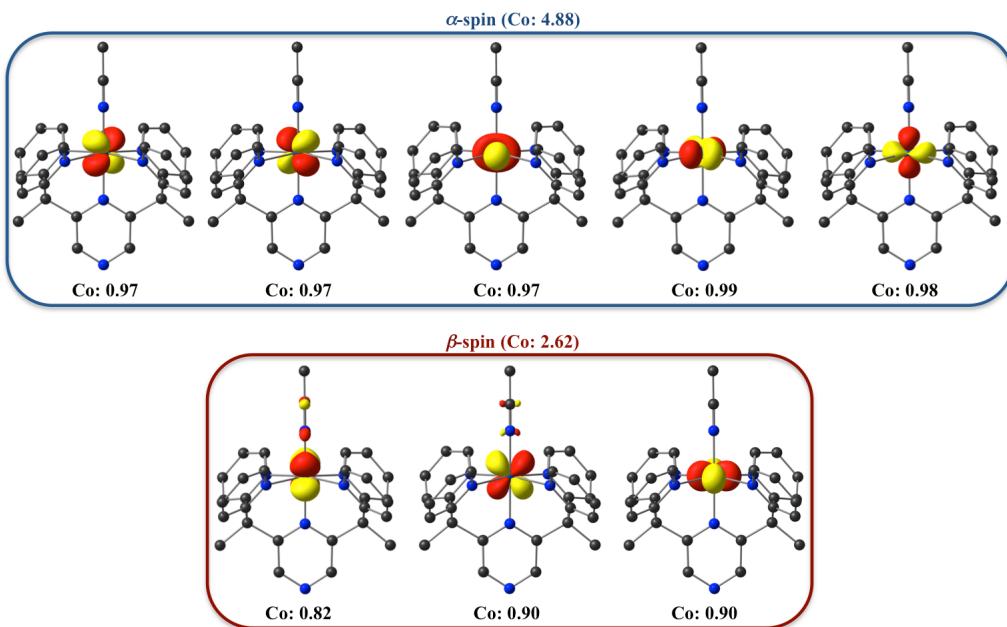


Figure S28. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $1'\text{-Co}+2\text{e}^-$ ($S = 3/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

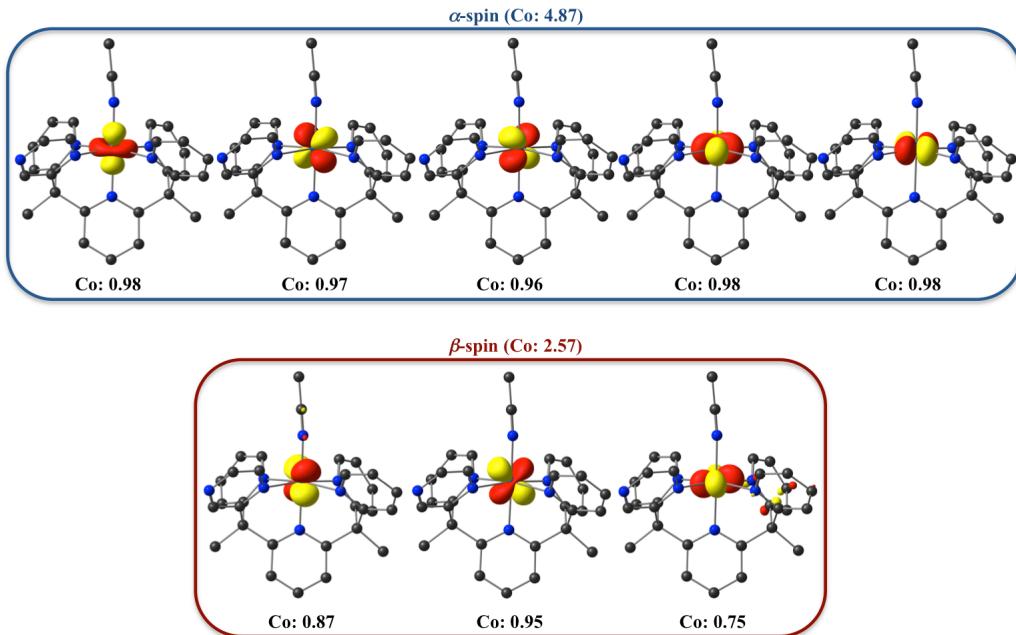


Figure S29. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $2'\text{-Co}+2\text{e}^-$ ($S = 1/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

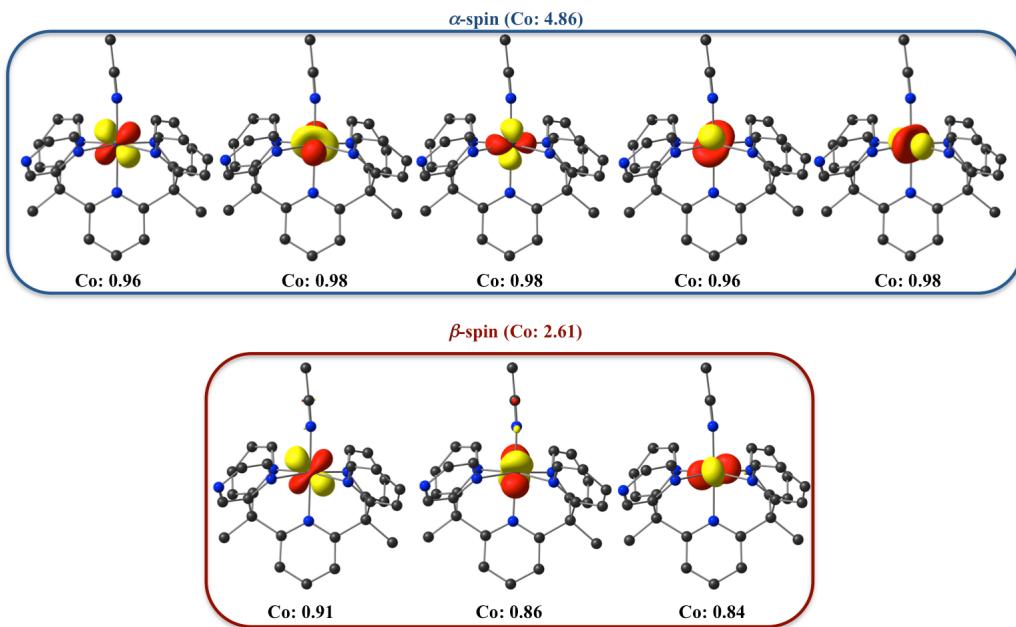


Figure S30. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $2'\text{-Co}+2\text{e}^-$ ($S = 3/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

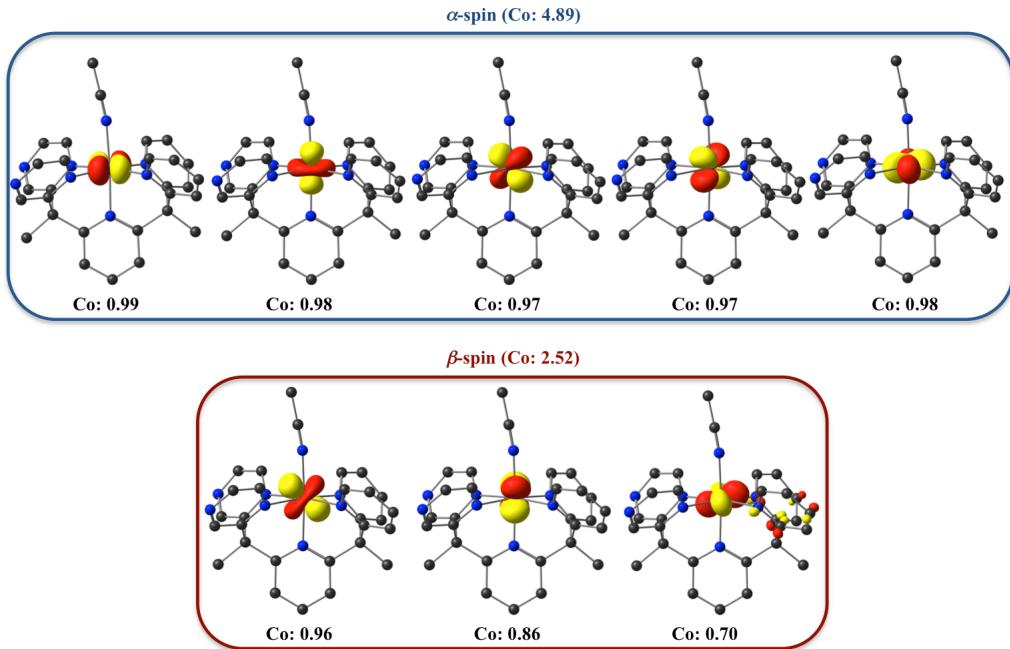


Figure S31. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $3'\text{-Co}+2\text{e}^-$ ($S = 1/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

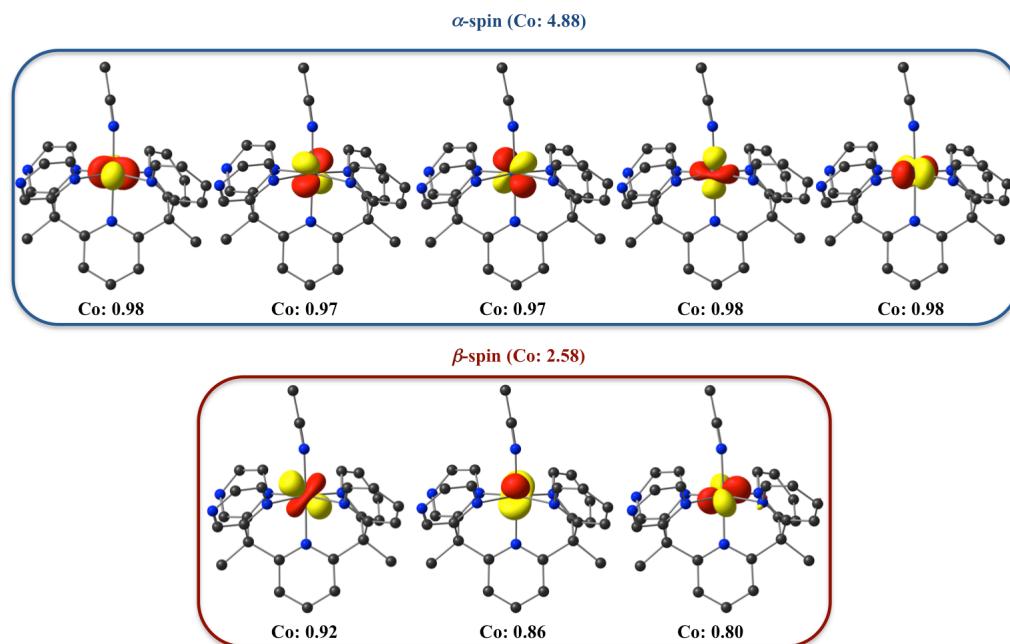


Figure S32. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $3'\text{-Co}+2\text{e}^-$ ($S = 3/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt.

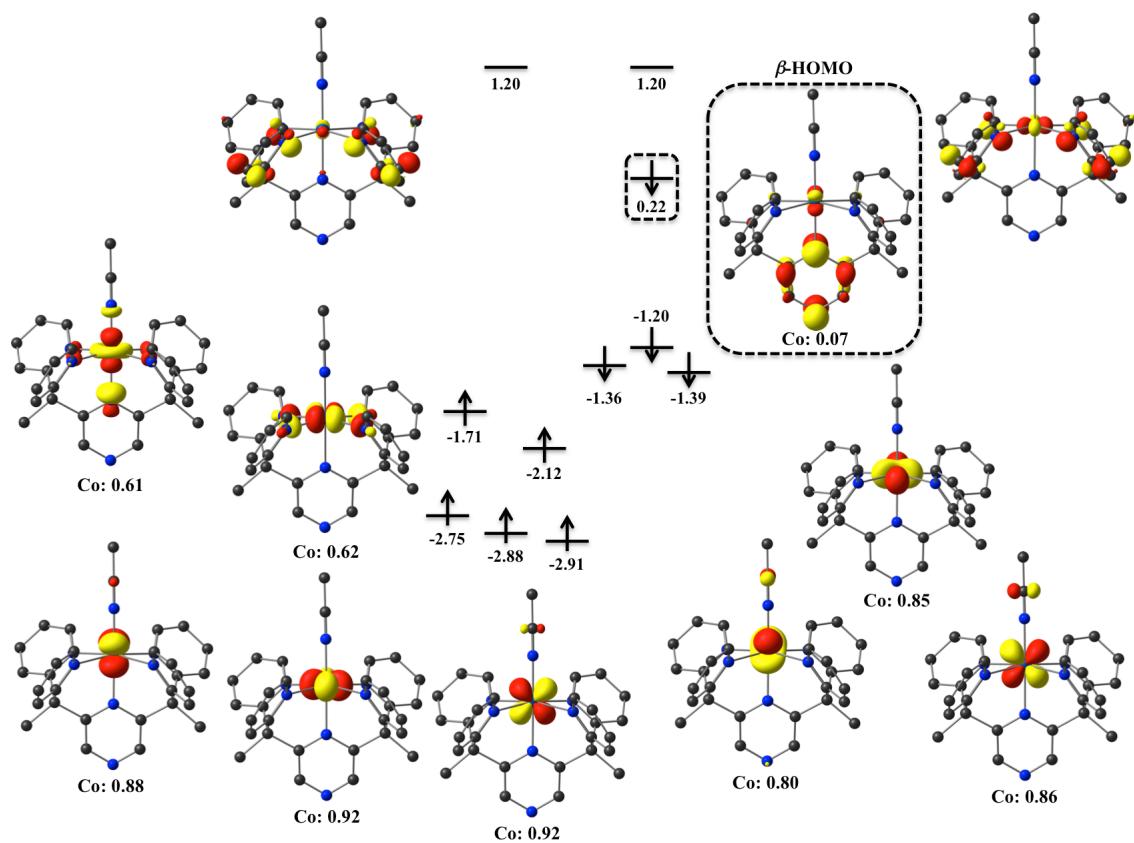


Figure S33. Isosurface (0.07 au) plots of the canonical molecular orbitals for **1'-Co+2e⁻** ($S = 1/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of the six-coordinate species **3'-Co+2e⁻** ($S = 1/2$).

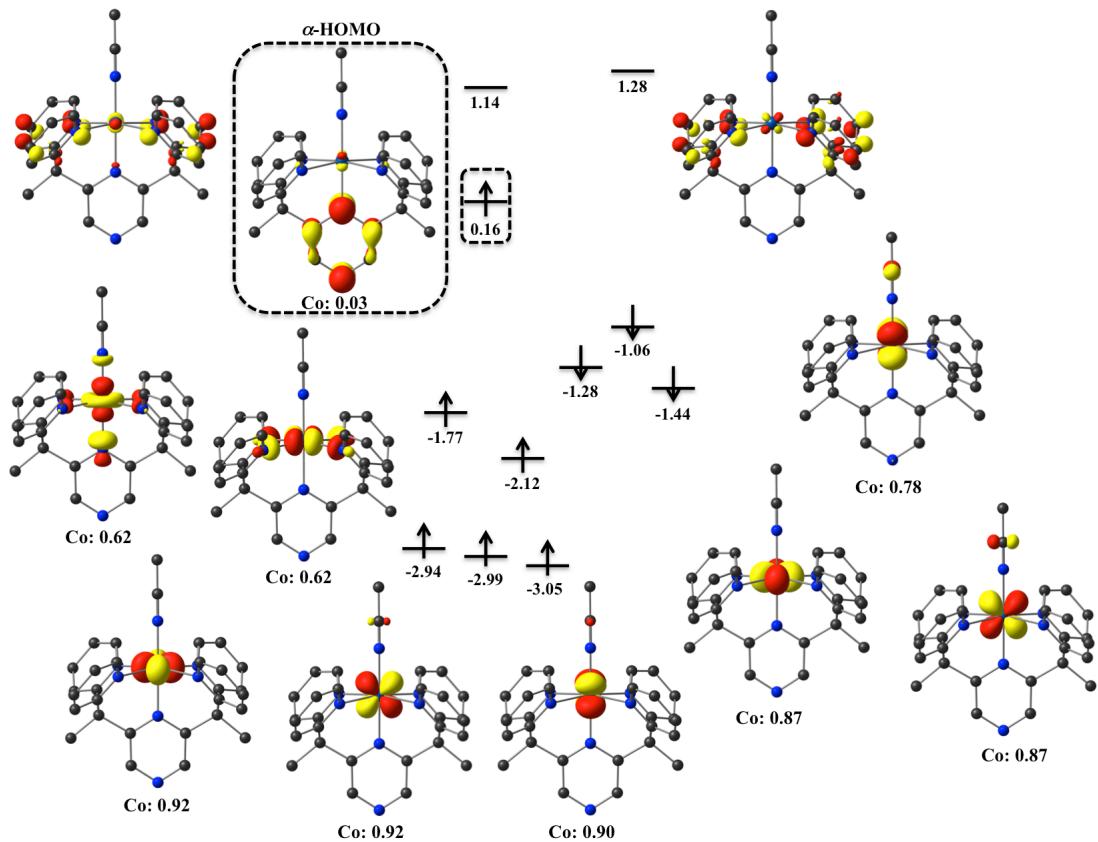


Figure S34. Isosurface (0.07 au) plots of the canonical molecular orbitals for **1'-Co+2e⁻** ($S = 3/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of the six-coordinate species **3'-Co+2e⁻** ($S = 1/2$).

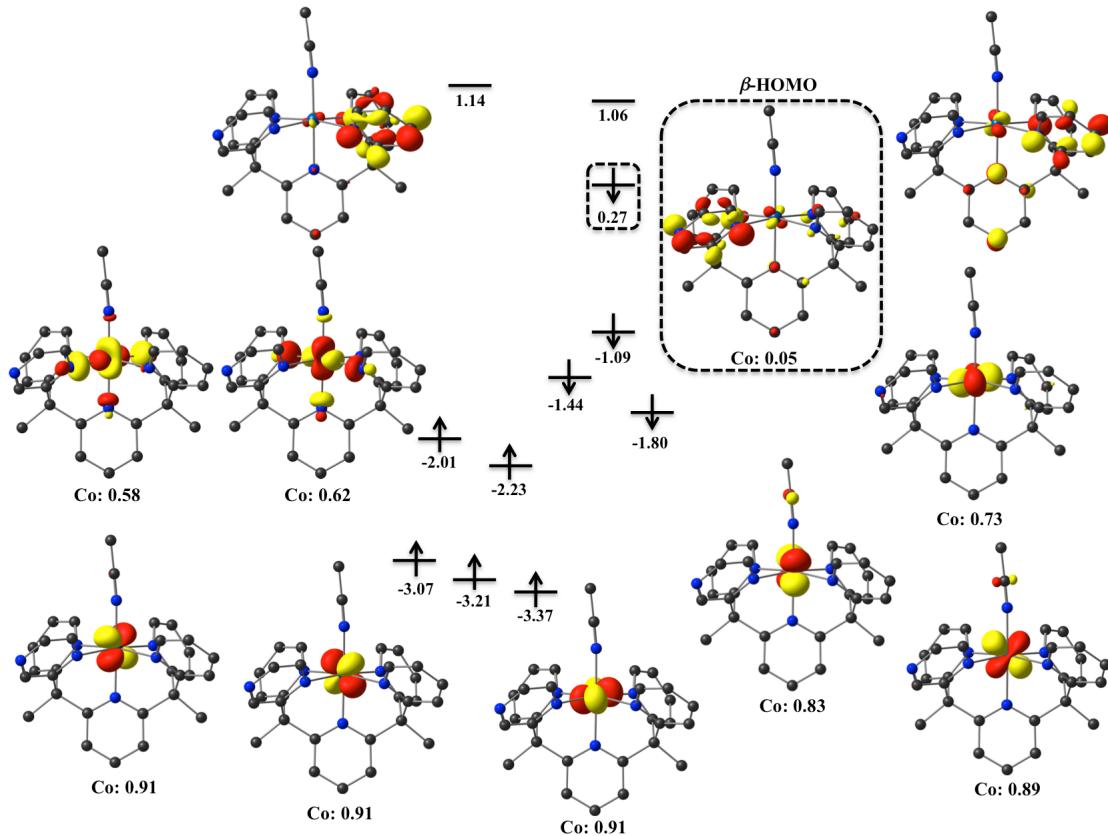


Figure S35. Isosurface (0.07 au) plots of the canonical molecular orbitals for **2'-Co+2e⁻** (S = 1/2) using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of the six-coordinate species **3'-Co+2e⁻** (S = 1/2).

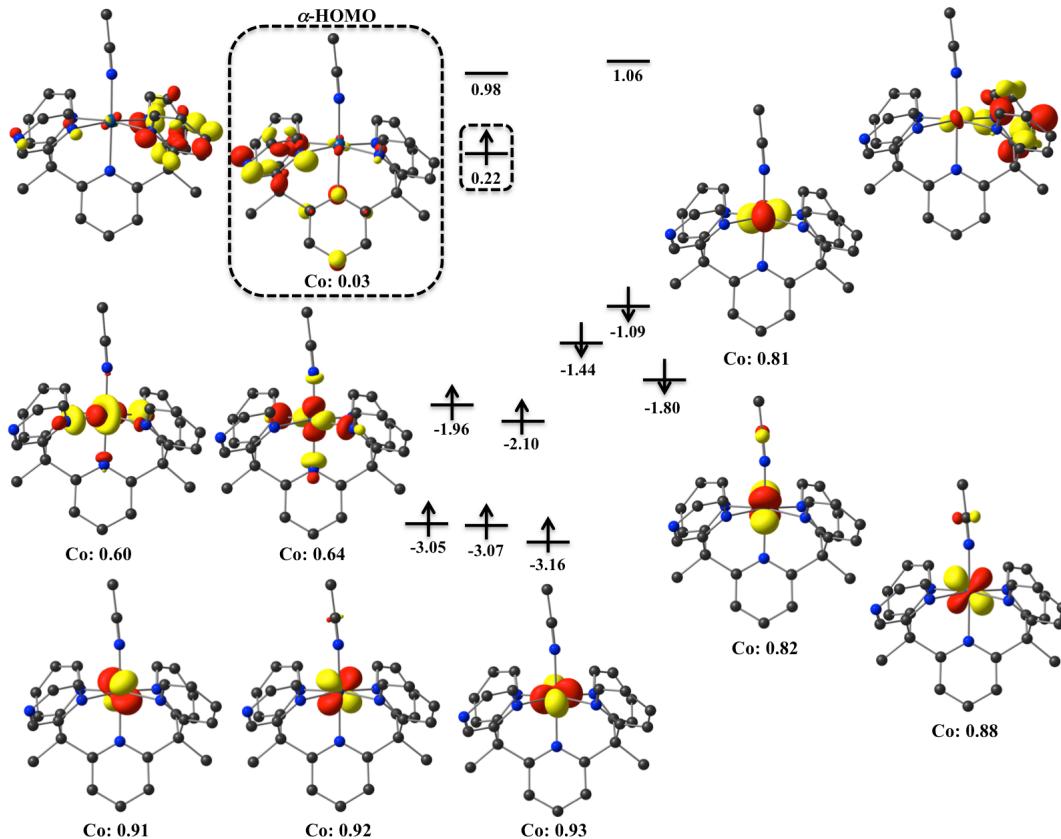


Figure S36. Isosurface (0.07 au) plots of the canonical molecular orbitals for **2'-Co+2e⁻** ($S = 3/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of the six-coordinate species **3'-Co+2e⁻** ($S = 1/2$).

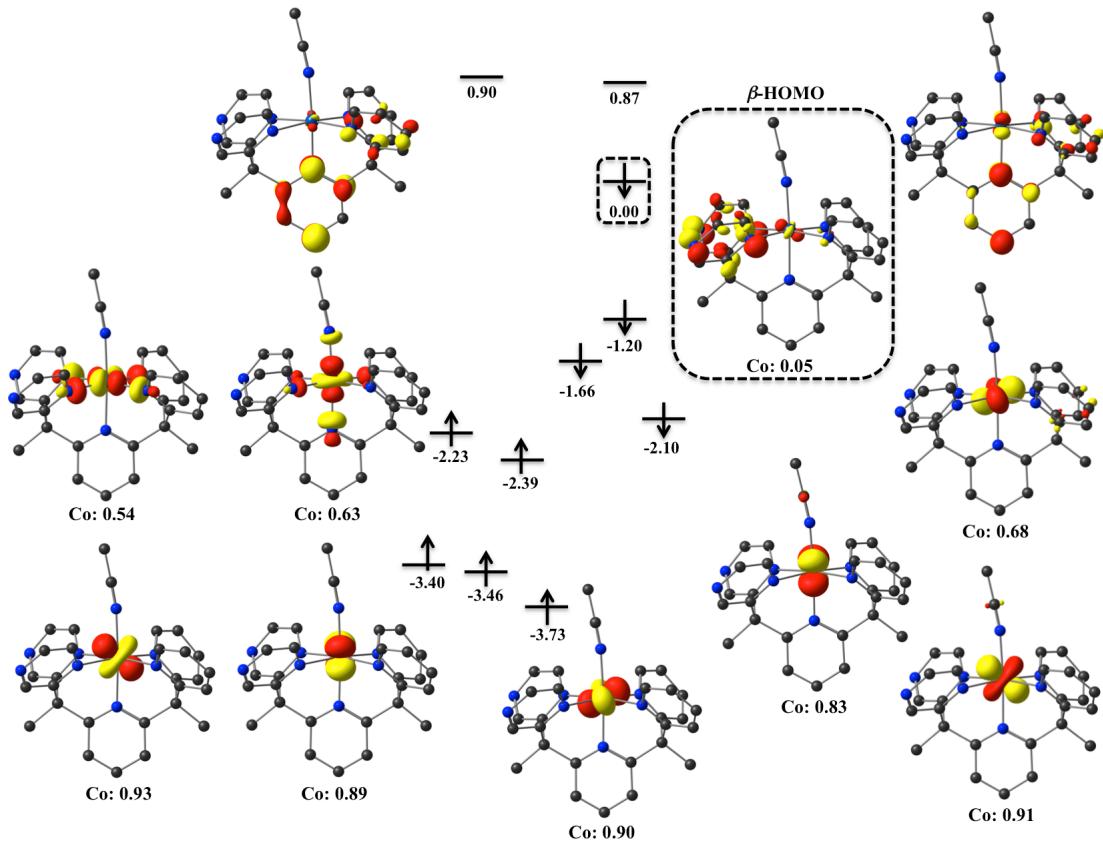


Figure S37. Isosurface (0.07 au) plots of the canonical molecular orbitals for **3'-Co+2e⁻** ($S = 1/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital).

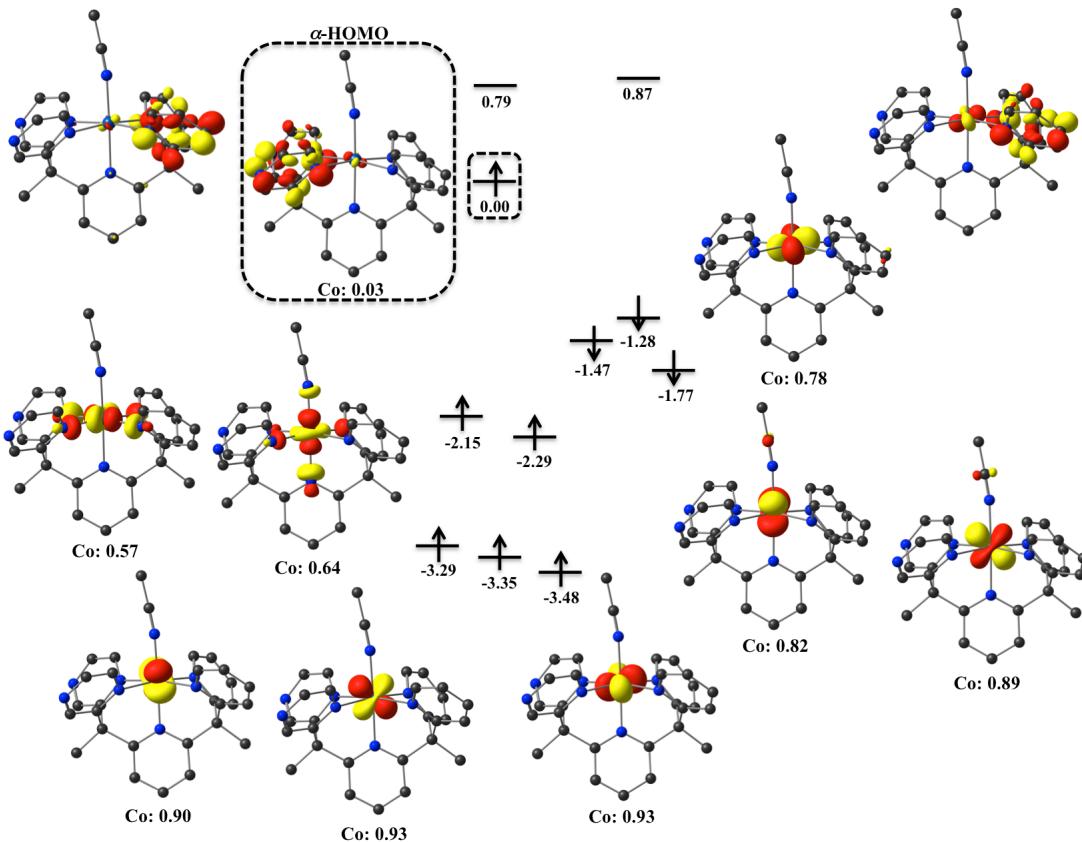


Figure S38. Isosurface (0.07 au) plots of the canonical molecular orbitals for **3'-Co+2e⁻** ($S = 3/2$) using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of the six-coordinate species **3'-Co+2e⁻** ($S = 1/2$).

Edmiston-Ruedenberg localized orbitals and canonical molecular orbitals for the five-coordinate species 1'-Co+2e⁻, 2'-Co+2e⁻ and 3'-Co+2e⁻ obtained after two-electron reduction and dissociation of the apical acetonitrile molecule.

Edmiston-Ruedenberg localized orbitals (Figures S39-44) and canonical molecular orbitals (Figures S45-50) have been computed for the two-electron reduced species after dissociation of the solvent molecule. Again, in this case the doublet and quartet states are degenerate and the second reduction is computed to be ligand-centered. As shown for the six-coordinate species, the LUMOs (lowest unoccupied molecular orbitals) are also calculated to be ligand-based.

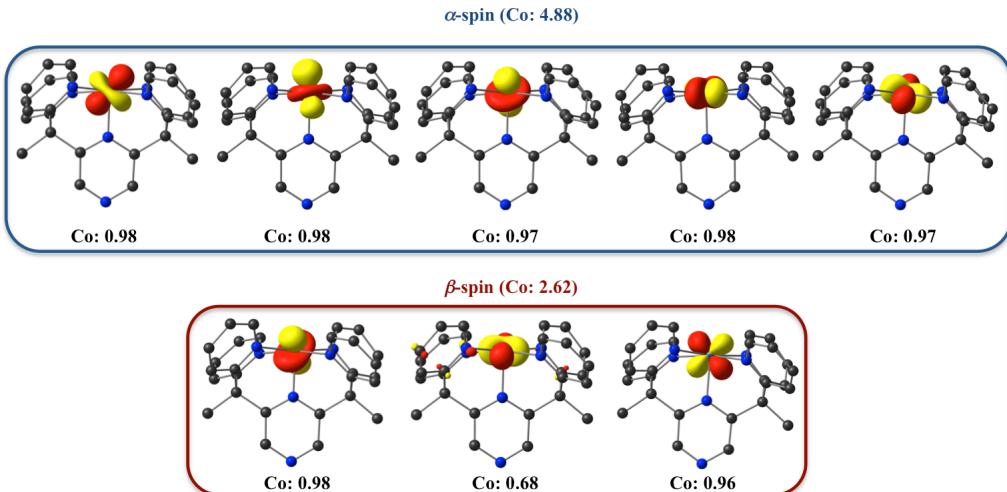


Figure S39. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $1'\text{-Co}+2\text{e}^-$ ($S = 1/2$) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt.

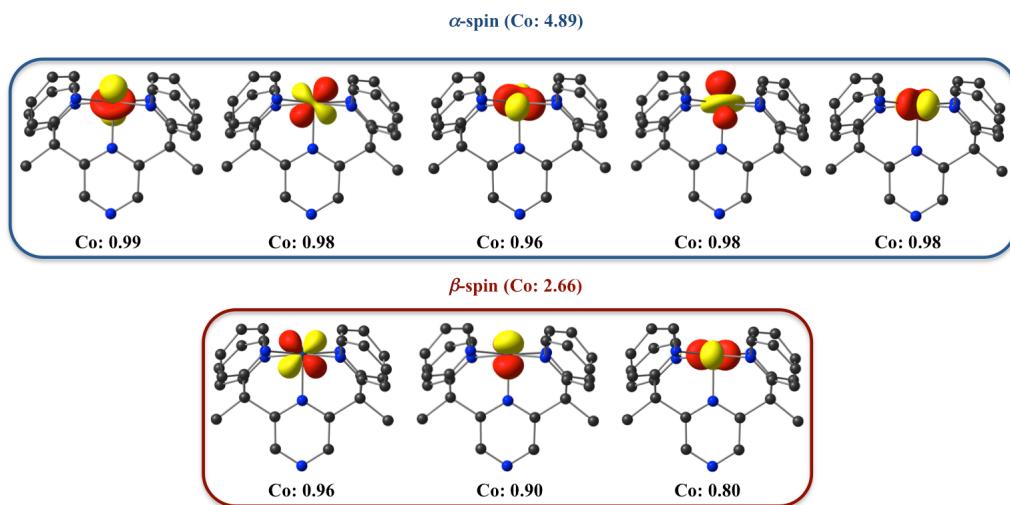


Figure S40. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $1'\text{-Co}+2\text{e}^-$ ($S = 3/2$) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt.

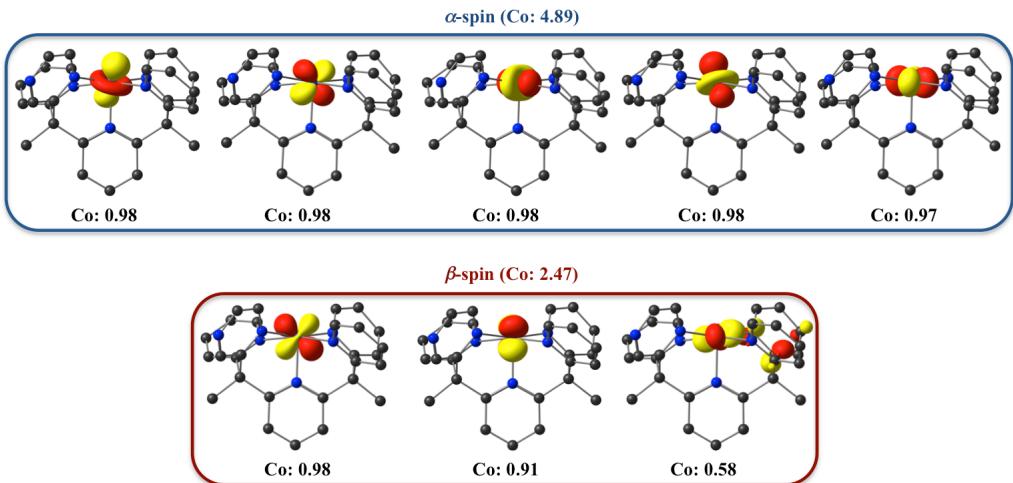


Figure S41. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $2'\text{-Co}+2\text{e}^-$ ($S = 1/2$) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt.

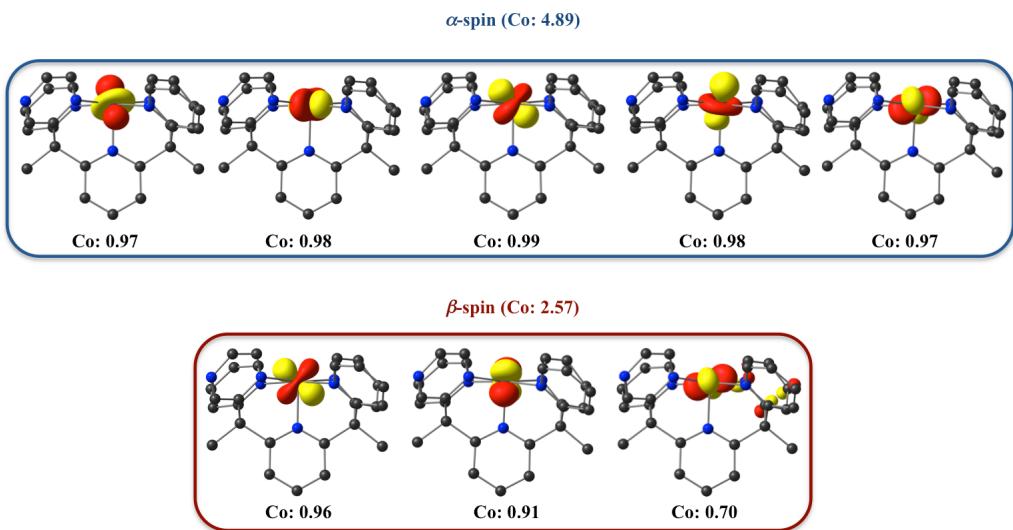


Figure S42. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $2'\text{-Co}+2\text{e}^-$ ($S = 3/2$) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt.

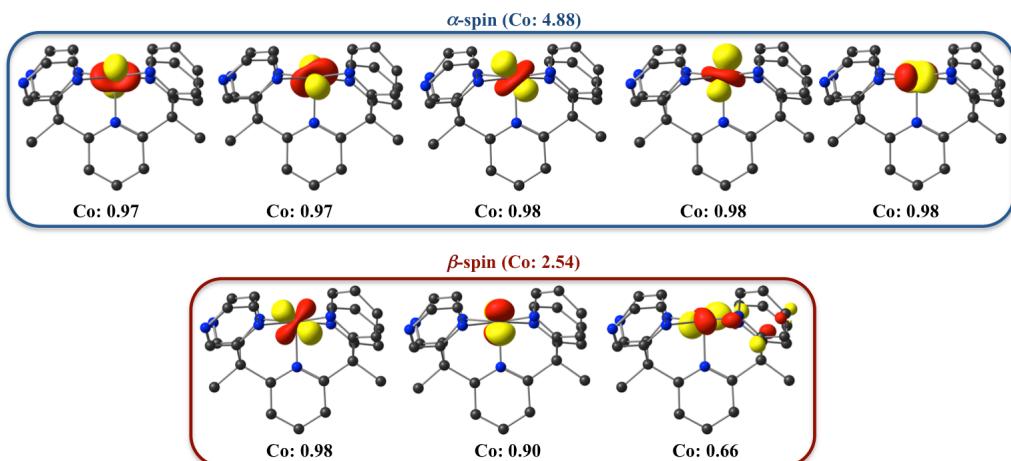


Figure S43. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $3'\text{-Co}+2\text{e}^-$ ($S = 1/2$) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt.

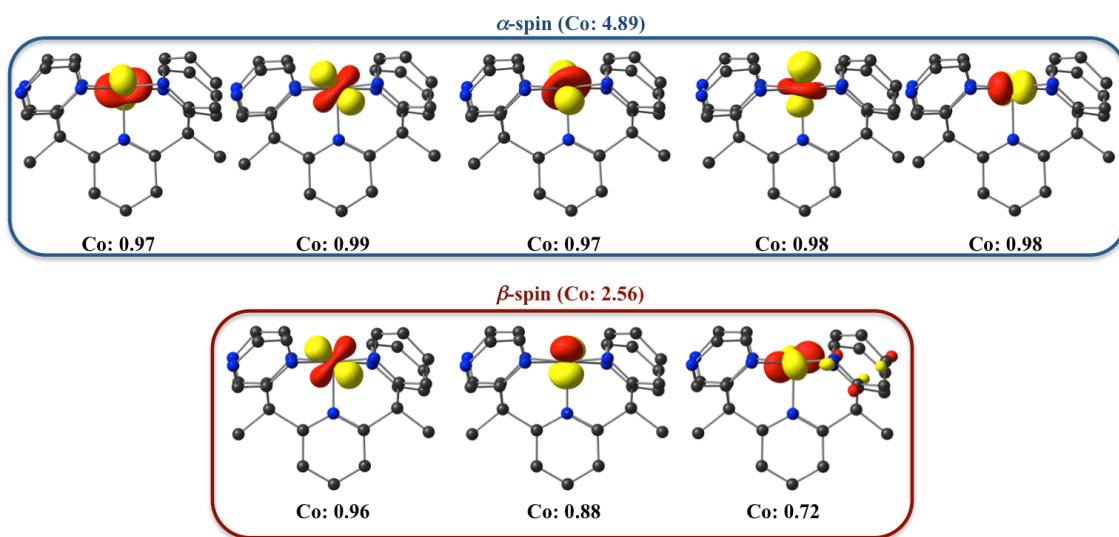


Figure S44. Isosurface (0.07 au) plots of the Edmiston-Ruedenberg localized orbitals for $3'\text{-Co}+2\text{e}^-$ ($S = 3/2$) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt.

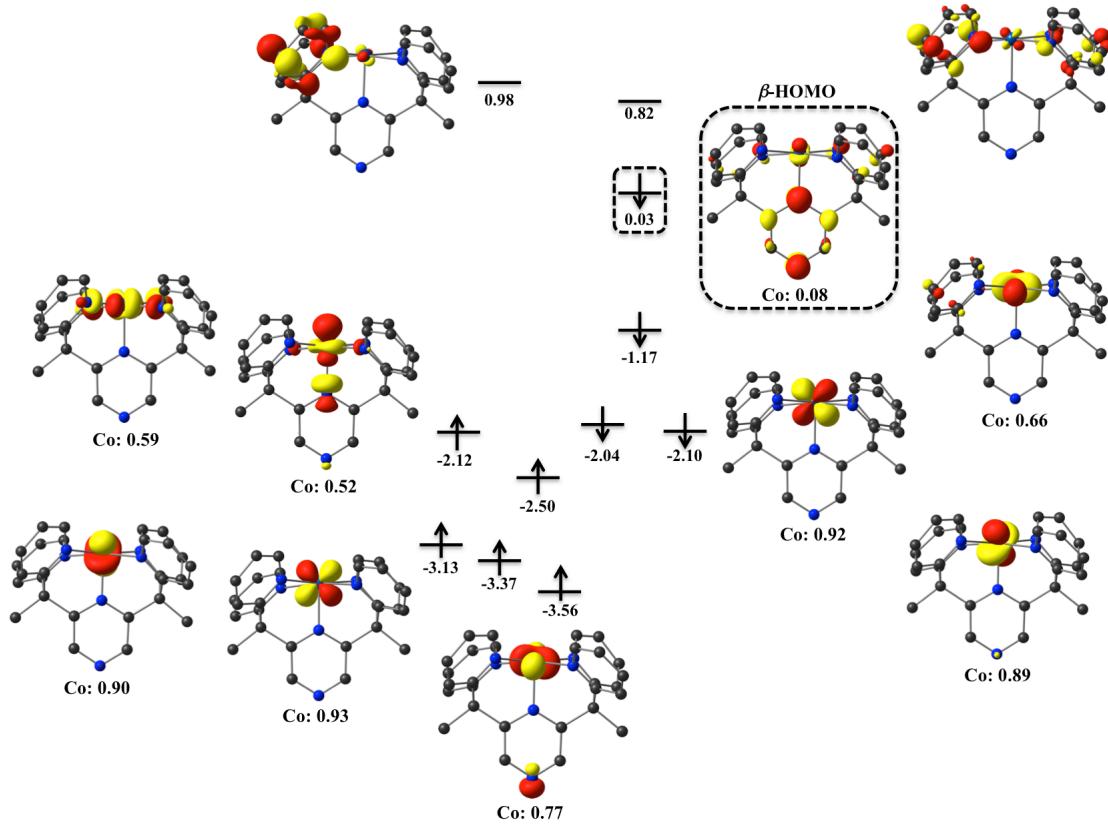


Figure S45. Isosurface (0.07 au) plots of the canonical molecular orbitals for **1'-Co+2e⁻** ($S = 1/2$) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of the six-coordinate species **3'-Co+2e⁻** ($S = 1/2$).

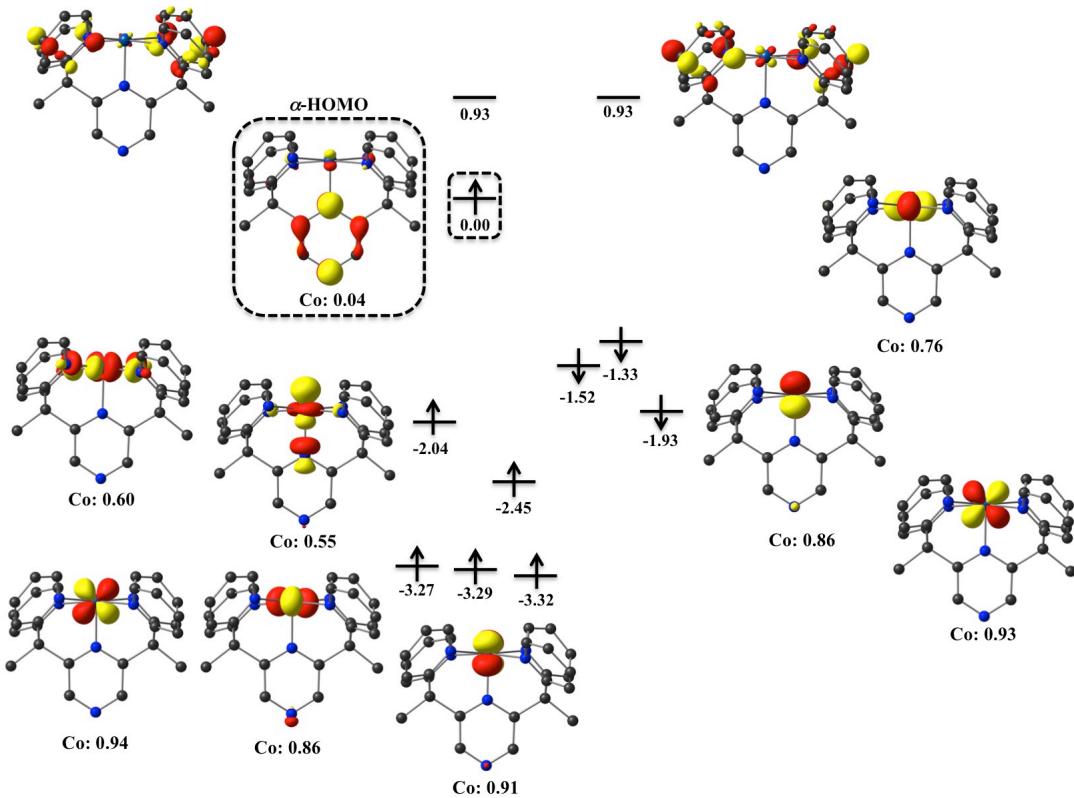


Figure S46. Isosurface (0.07 au) plots of the canonical molecular orbitals for **1'-Co+2e⁻** ($S = 3/2$) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of the six-coordinate species **3'-Co+2e⁻** ($S = 1/2$).

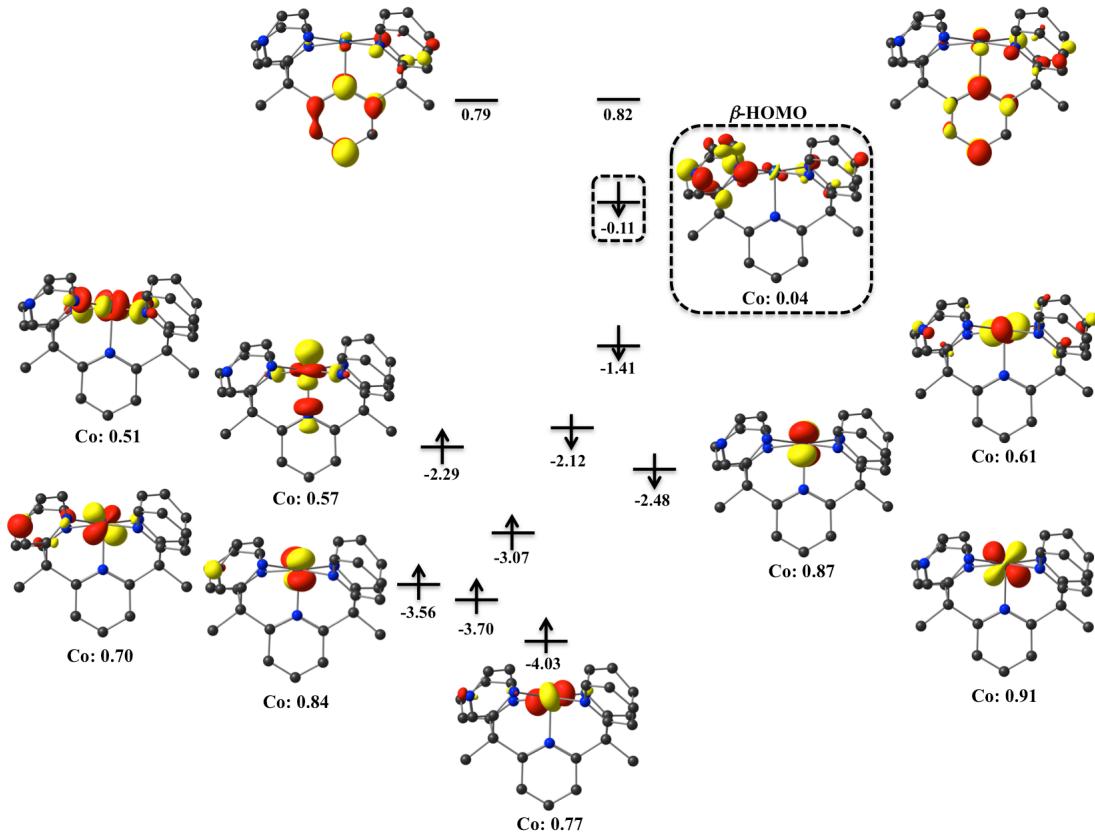


Figure S47. Isosurface (0.07 au) plots of the canonical molecular orbitals for **2'-Co+2e⁻** ($S = 1/2$) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of the six-coordinate species **3'-Co+2e⁻** ($S = 1/2$).

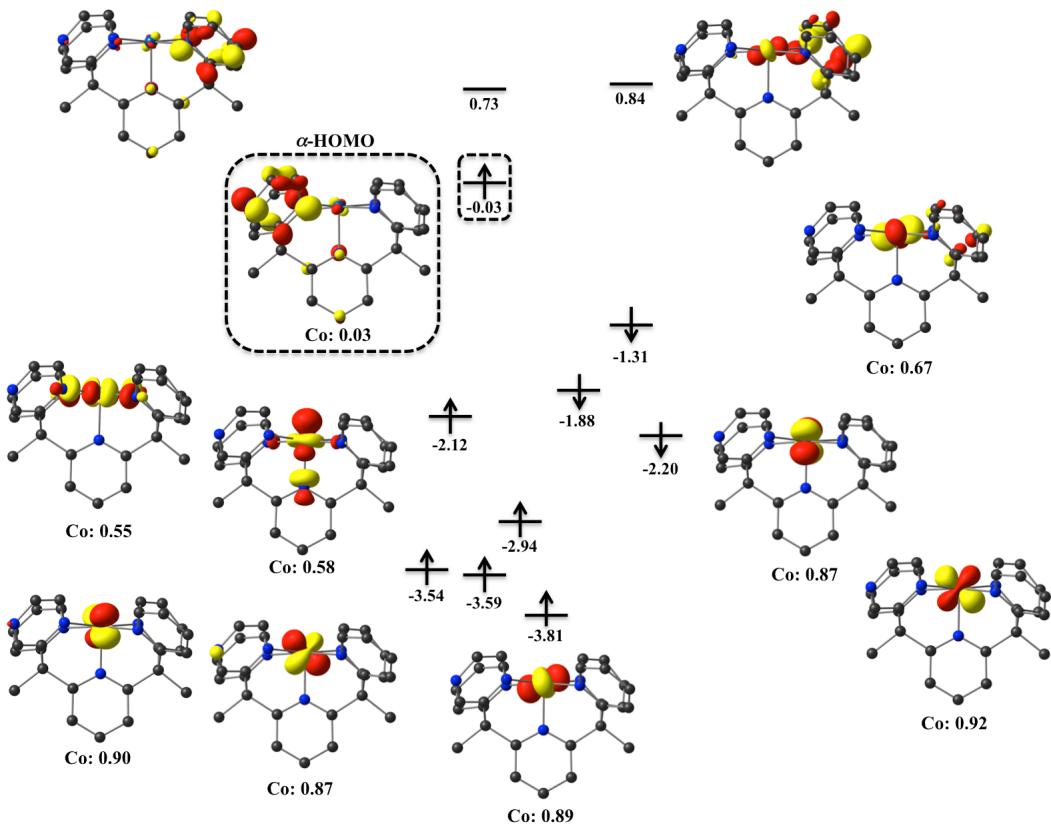


Figure S48. Isosurface (0.07 au) plots of the canonical molecular orbitals for **2'-Co+2e⁻** ($S = 3/2$) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of the six-coordinate species **3'-Co+2e⁻** ($S = 1/2$).

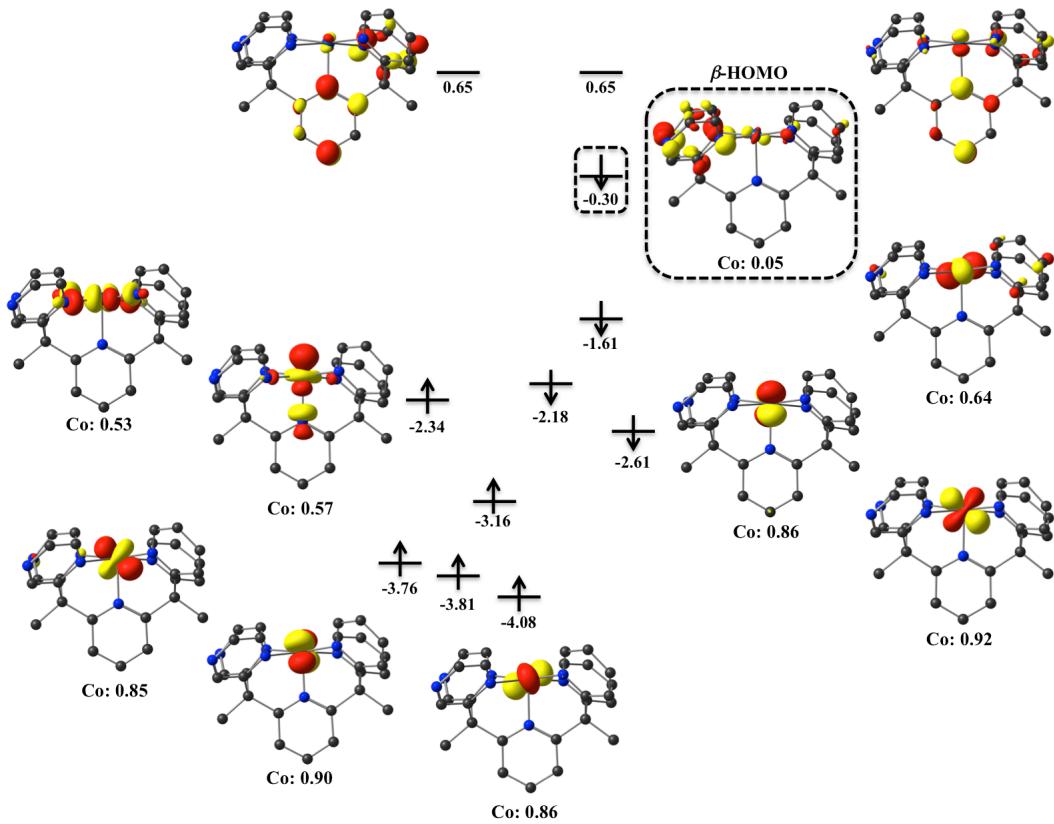


Figure S49. Isosurface (0.07 au) plots of the canonical molecular orbitals for **3'-Co+2e⁻** (S = 1/2) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of the six-coordinate species **3'-Co+2e⁻** (S = 1/2).

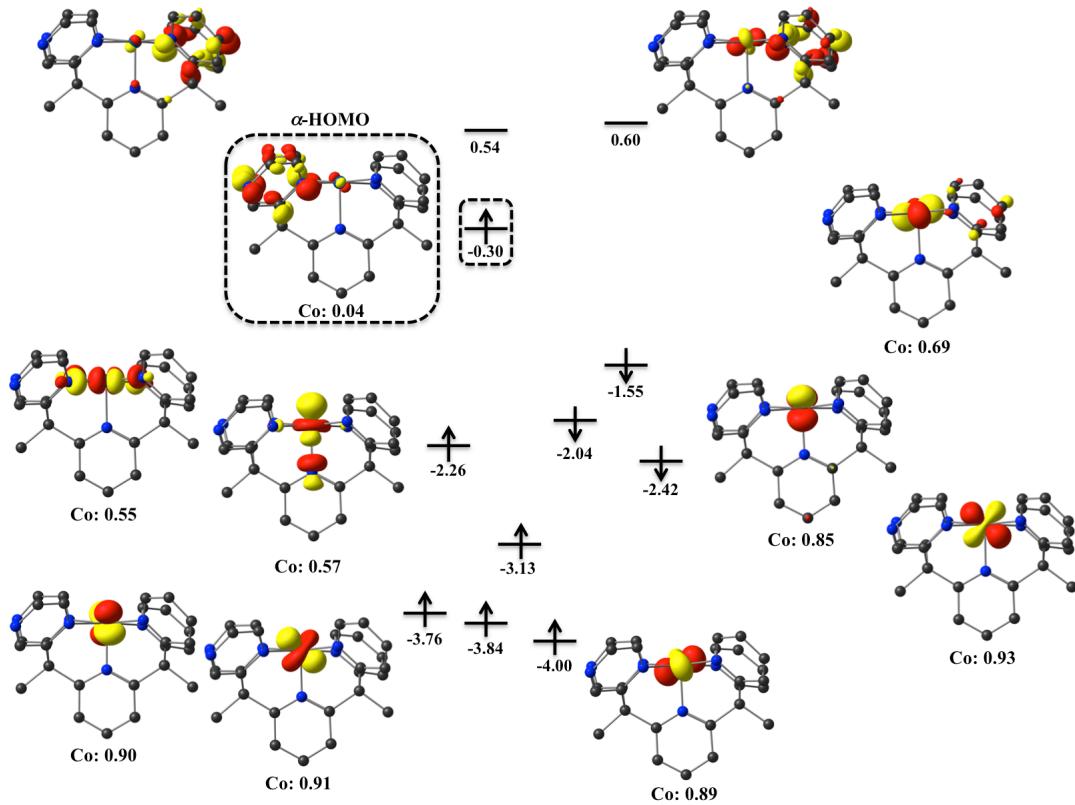


Figure S50. Isosurface (0.07 au) plots of the canonical molecular orbitals for **3'-Co+2e⁻** ($S = 3/2$) after dissociation of the acetonitrile molecule using the B3LYP functional. The Löwdin population analyses are given for cobalt. All energies (in eV) are relative to the HOMO (highest occupied molecular orbital) of the six-coordinate species **3'-Co+2e⁻** ($S = 1/2$).

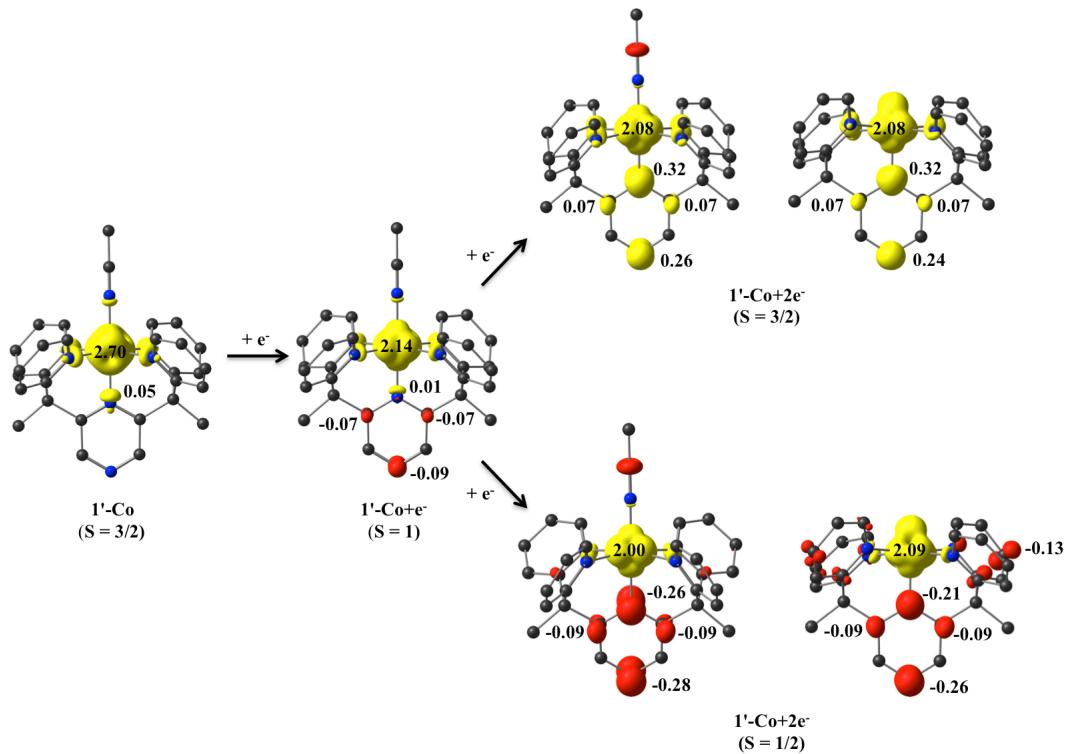


Figure S51. Isosurface (0.007 au) plots of the Mulliken spin population for **1'-Co** and its one and two-electron reduced species.

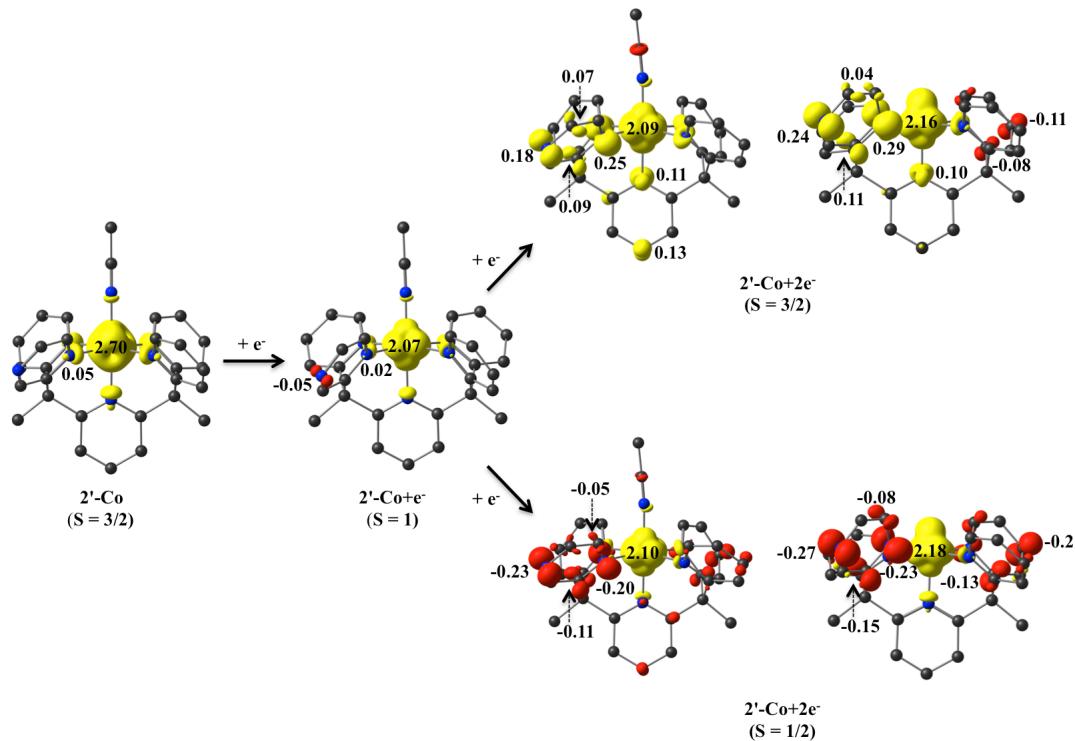


Figure S52. Isosurface (0.007 au) plots of the Mulliken spin population for **2'-Co** and its one and two-electron reduced species.

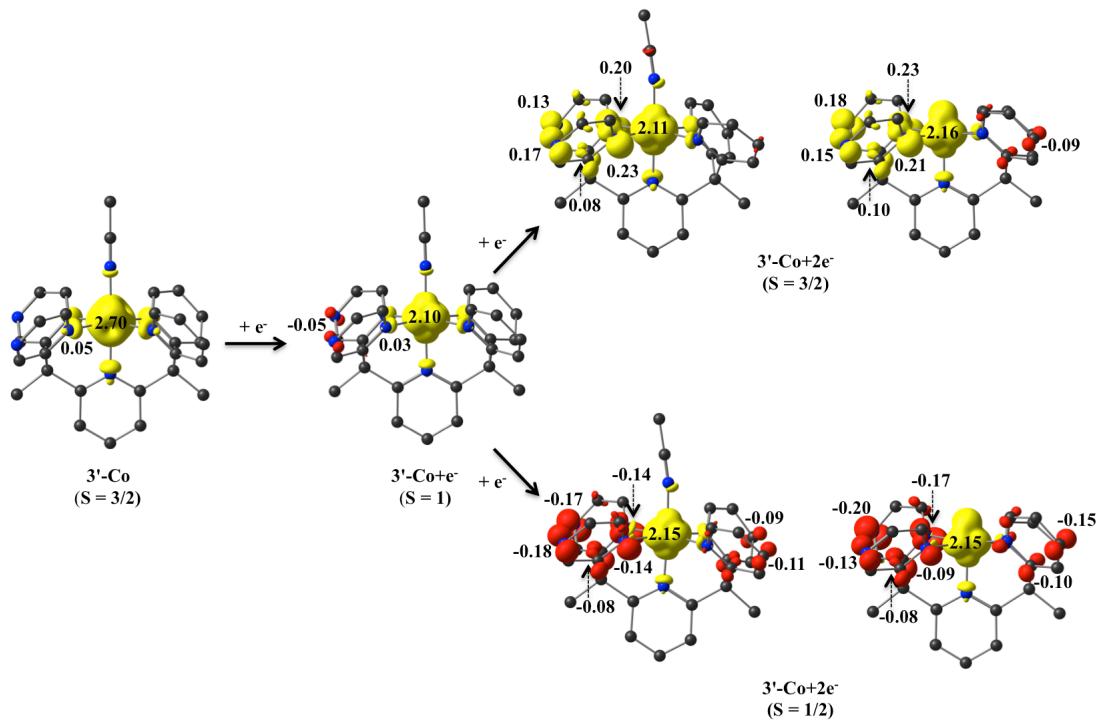
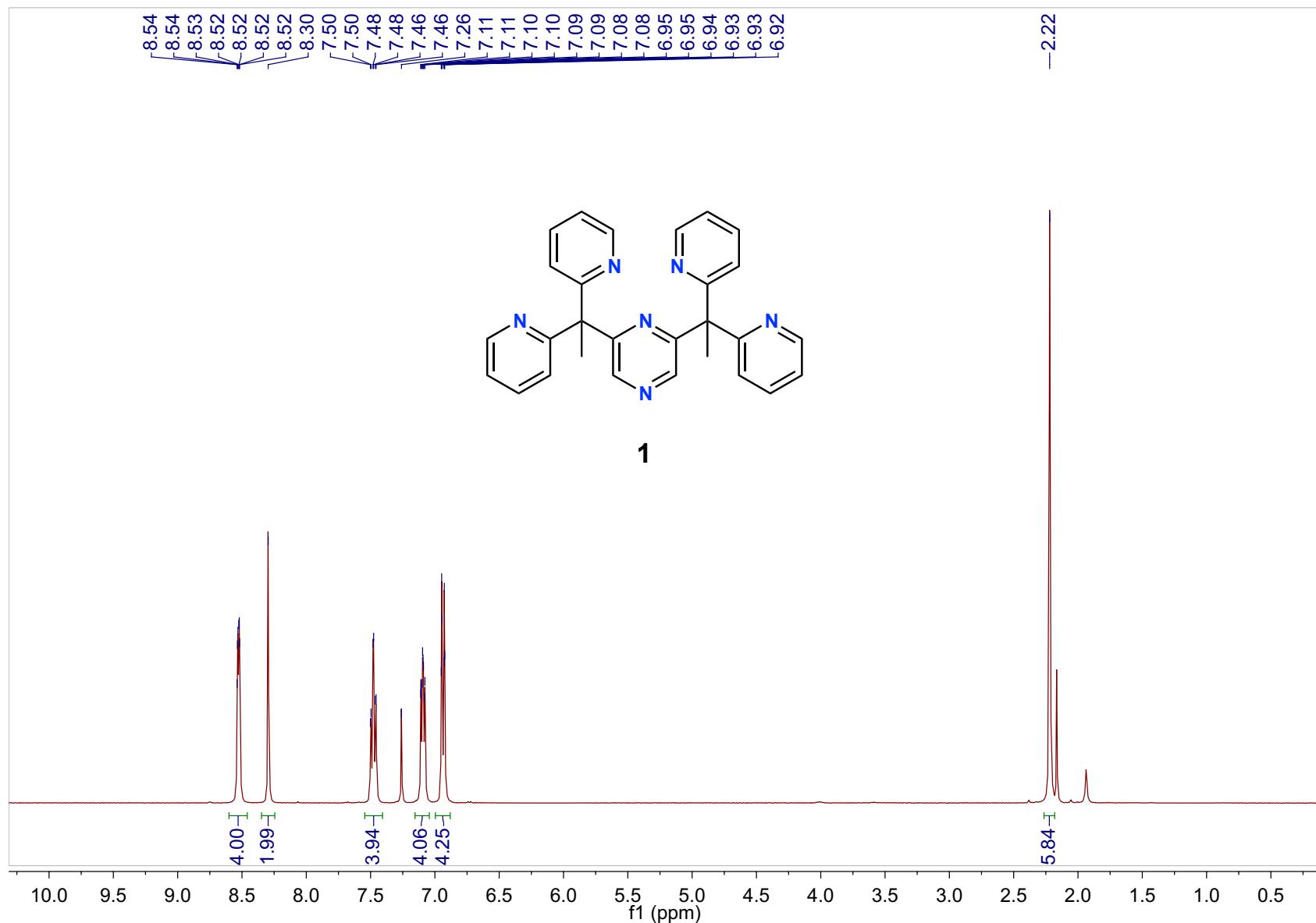
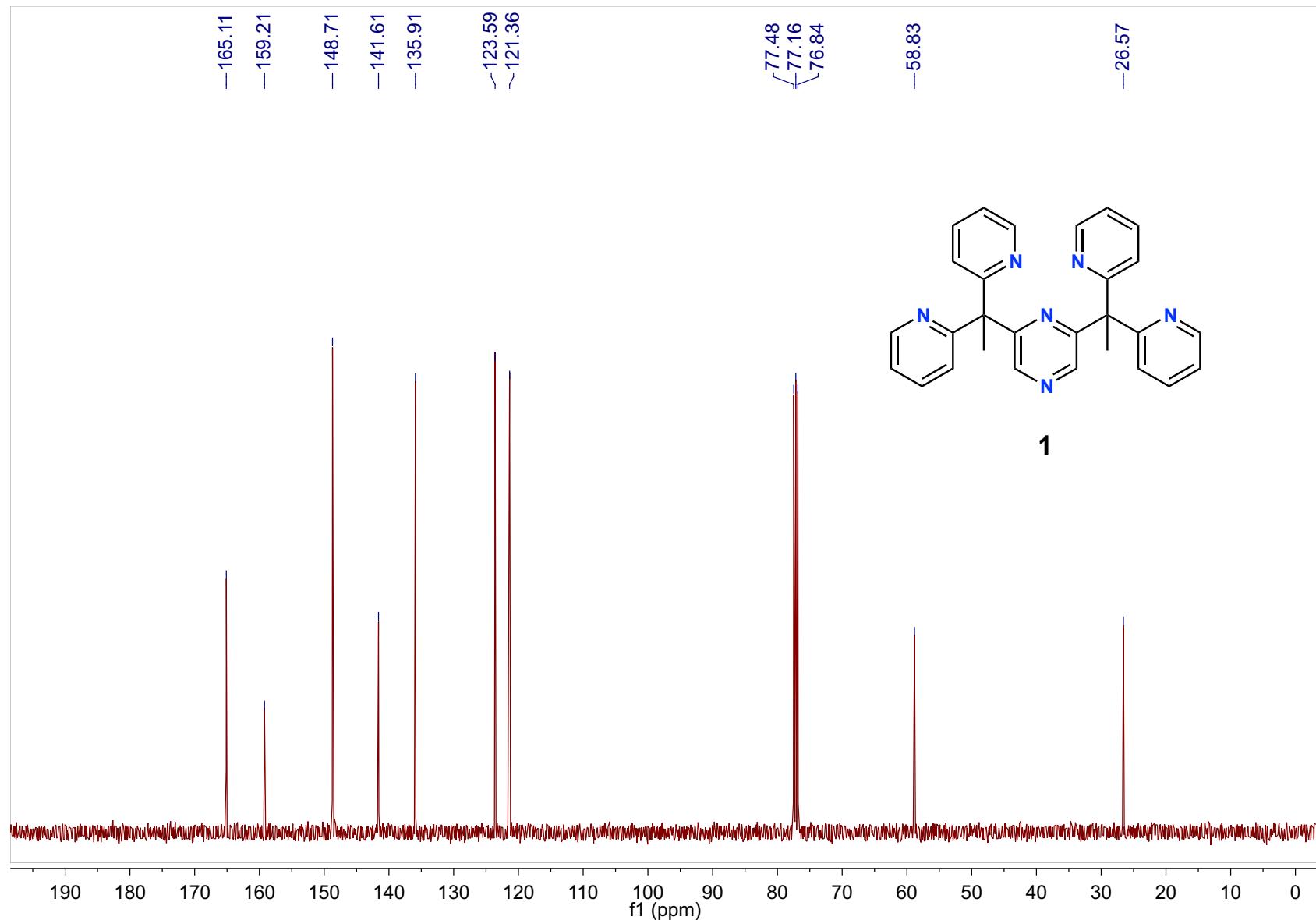


Figure S53. Isosurface (0.007 au) plots of the Mulliken spin population for **3'-Co** and its one and two-electron reduced species.

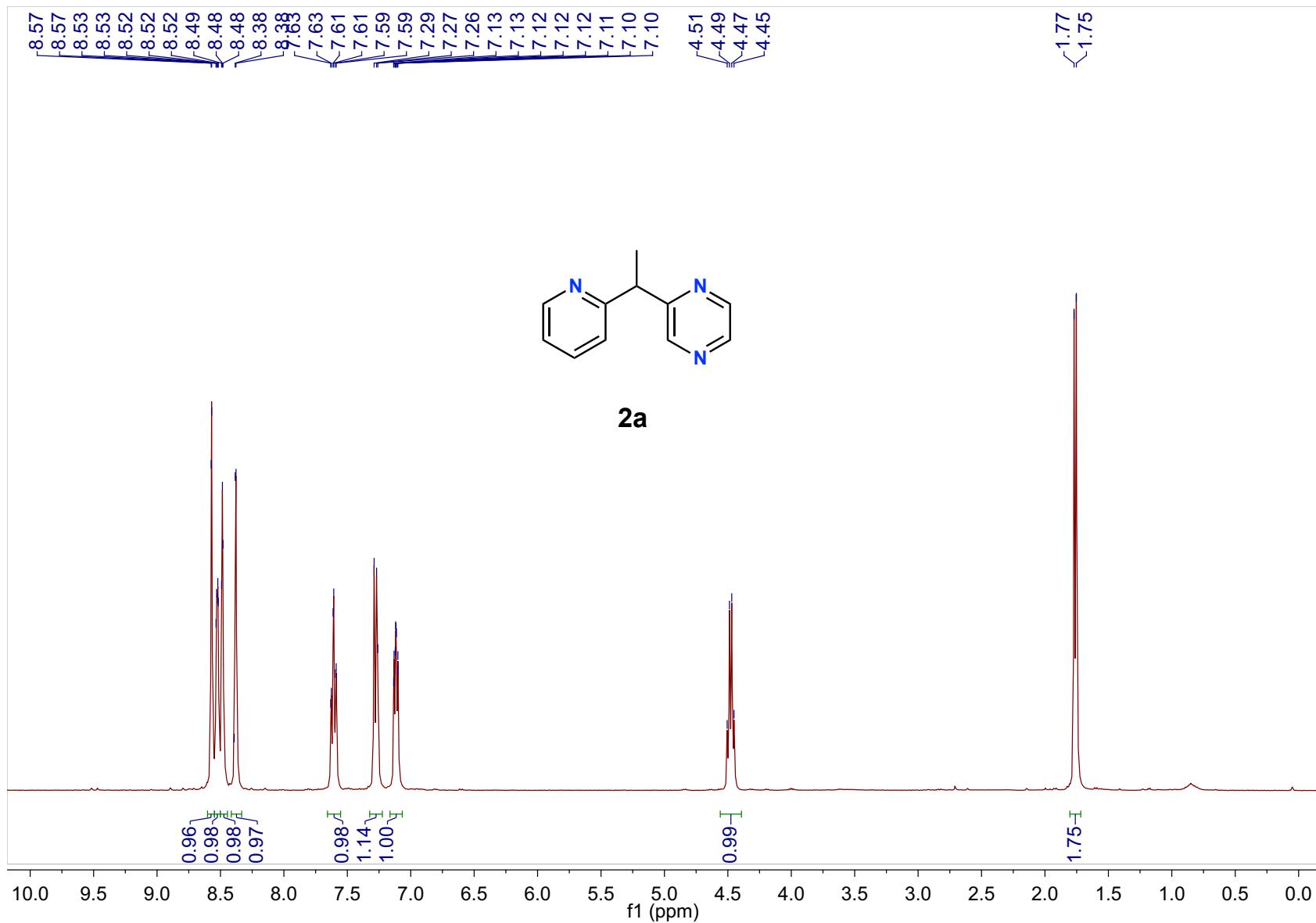
400 MHz ^1H NMR (CDCl_3)



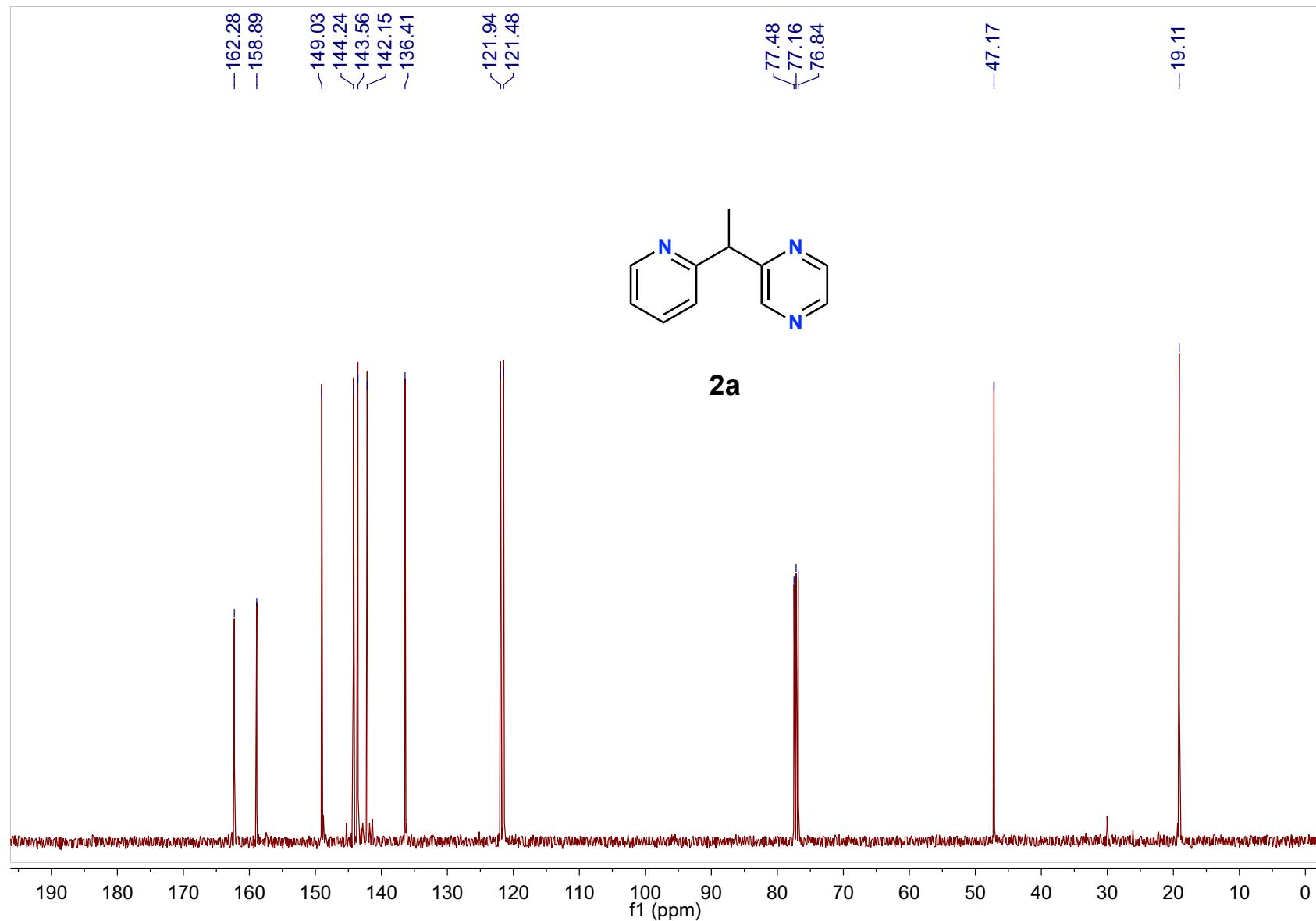
100 MHz ^{13}C NMR (CDCl_3)



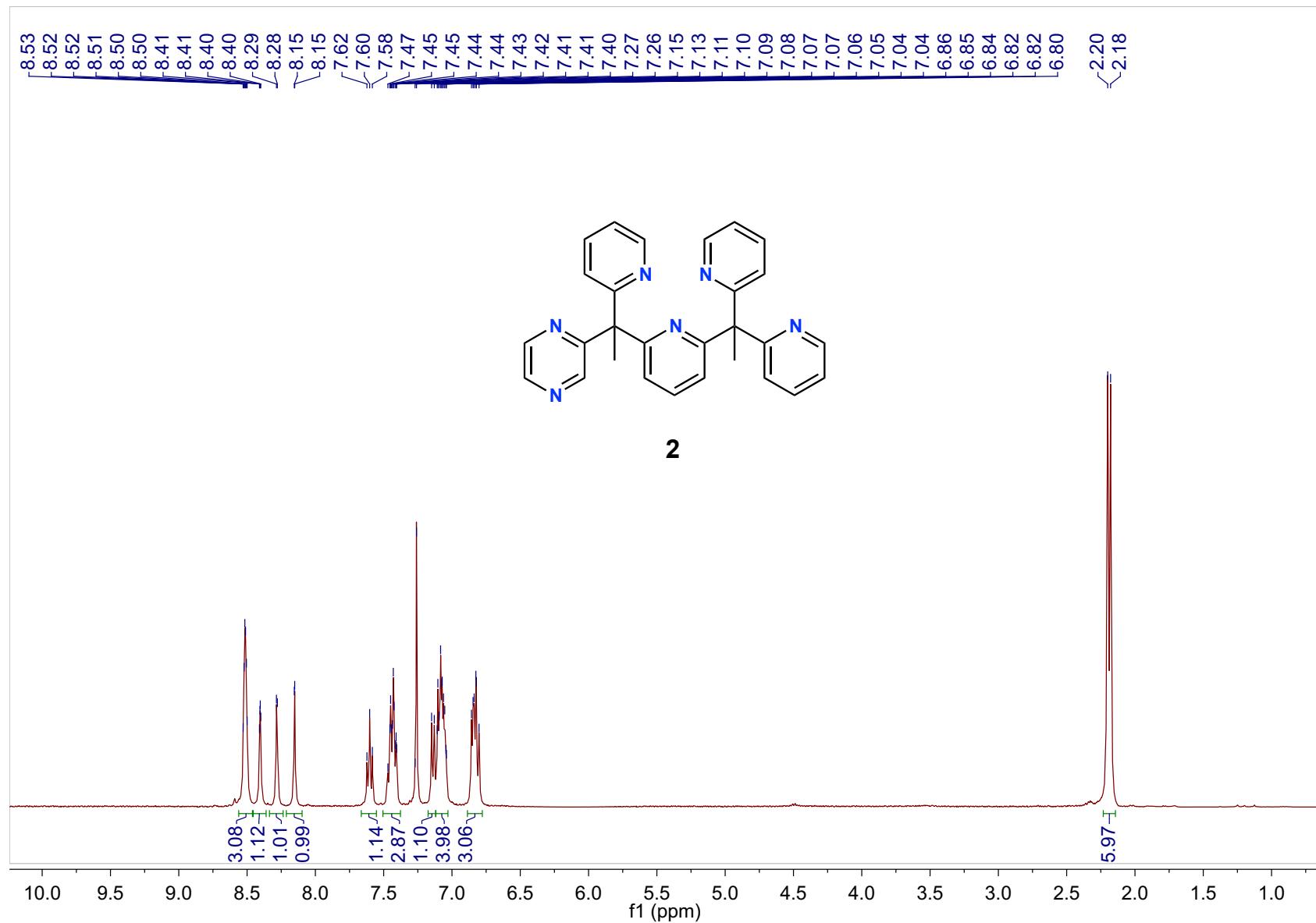
400 MHz ^1H NMR (CDCl_3)



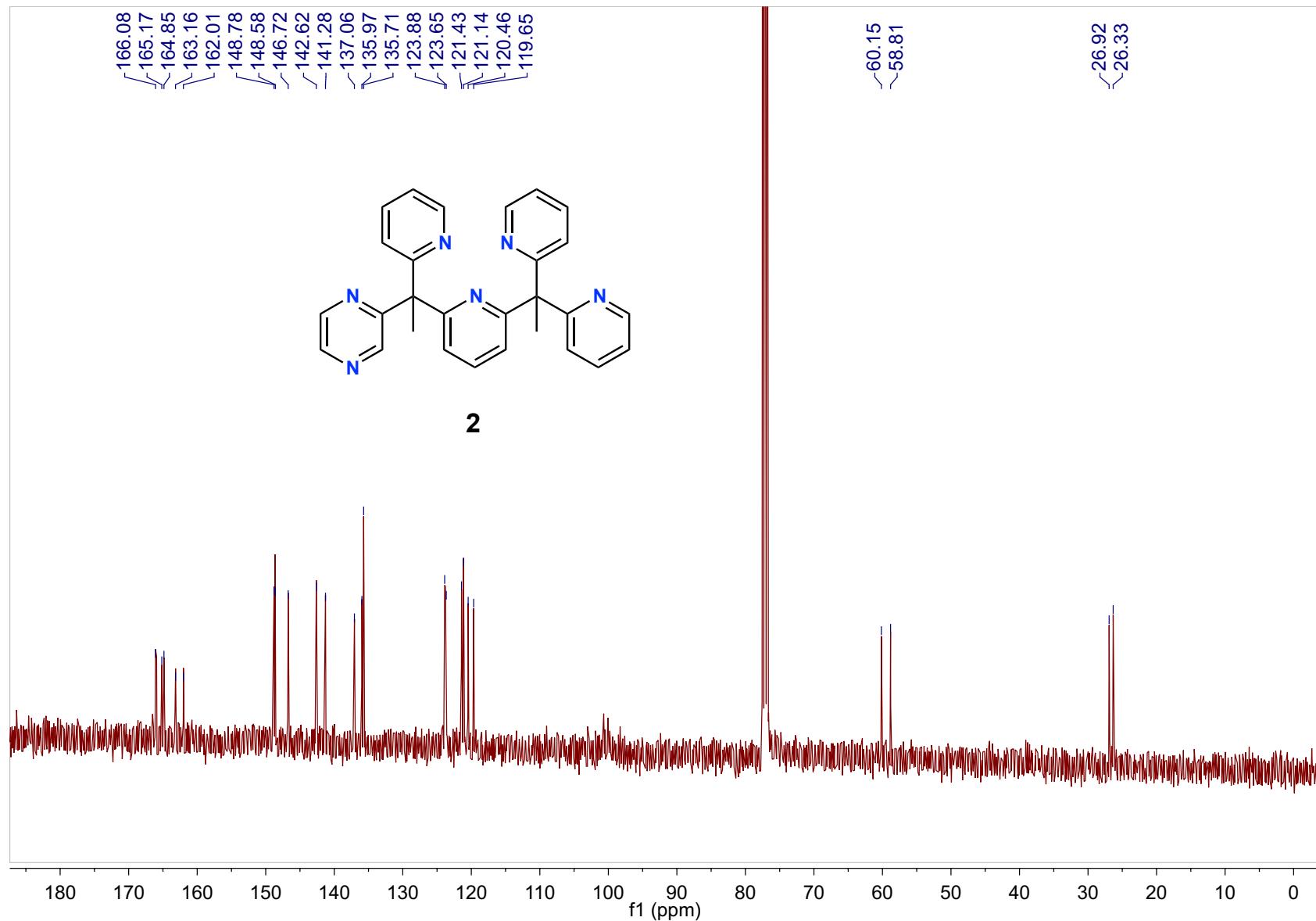
100 MHz ^{13}C NMR (CDCl_3)



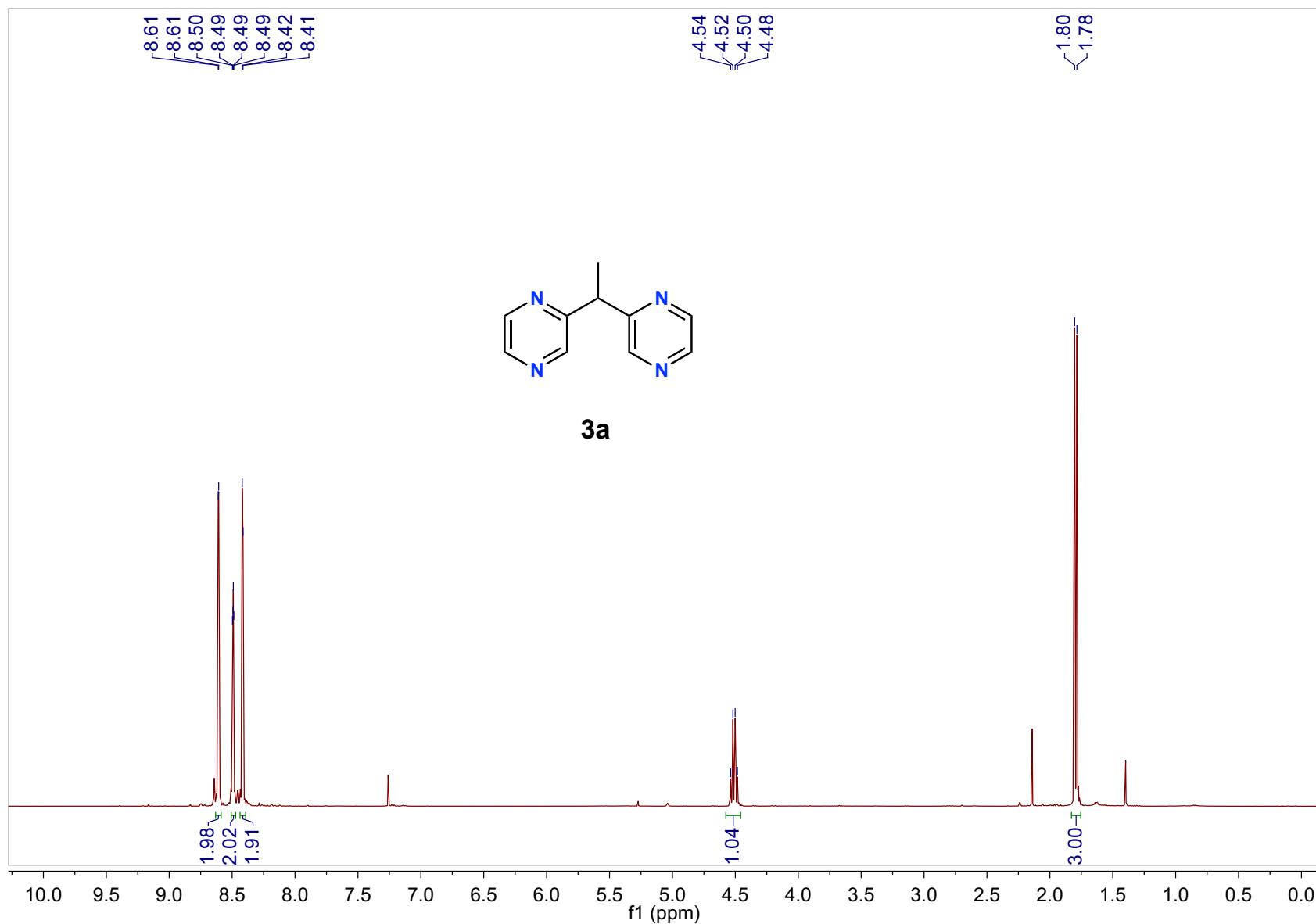
400 MHz ^1H NMR (CDCl_3)



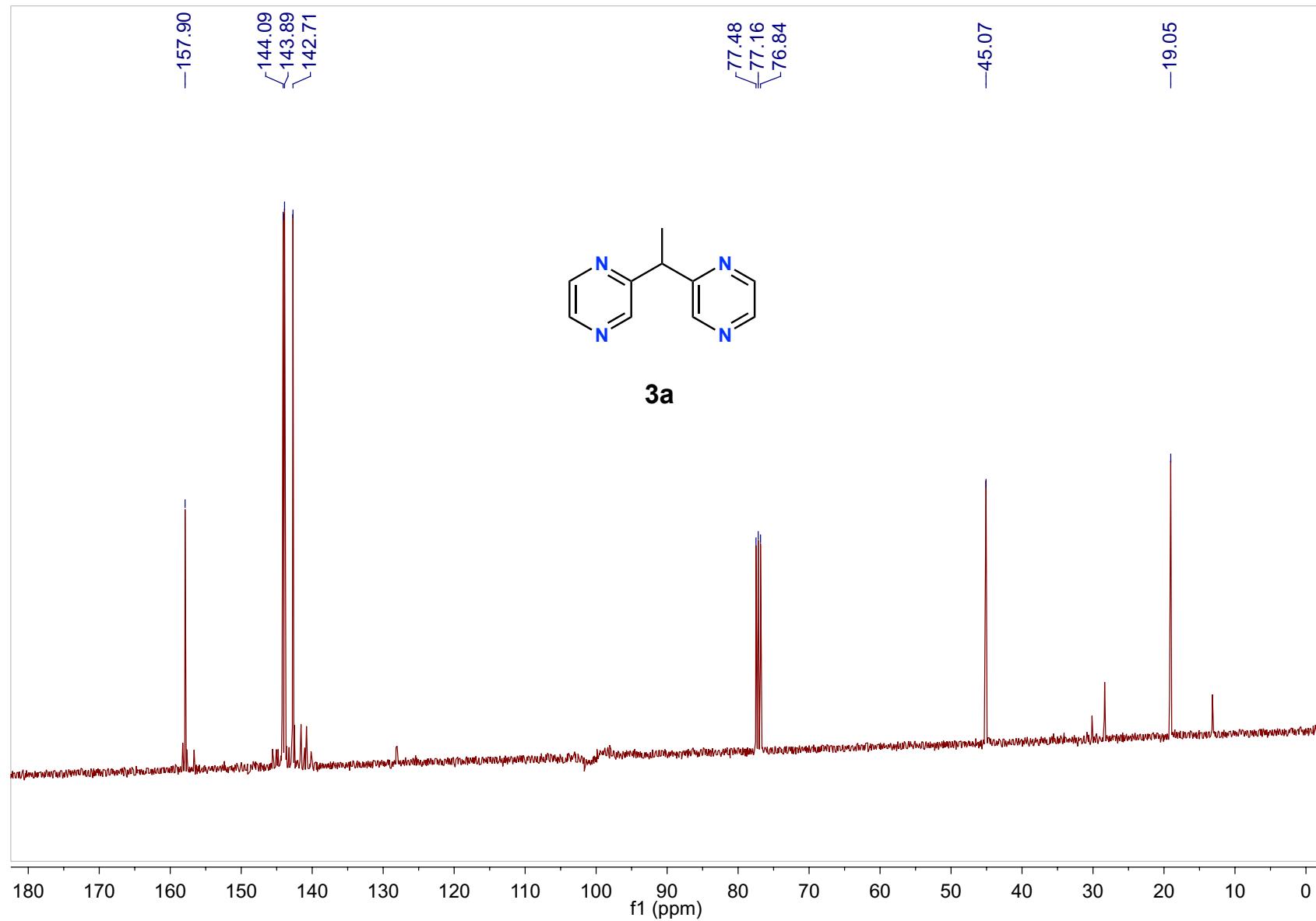
100 MHz ^{13}C NMR (CDCl_3)



400 MHz ^1H NMR (CDCl_3)



100 MHz ^{13}C NMR (CDCl_3)



COMPUTED STRUCTURES AND ENERGIES
Acetonitrile (S = 0)

Energy = -132.654863
 Enthalpy 0K = -132.609807
 Free Energy 298K = -132.633844

C 7.53820 2.87961 9.05531
 N 7.35471 3.97134 9.39570
 C 7.76952 1.50358 8.62616
 H 7.74129 0.82322 9.49077
 H 6.99555 1.19139 7.90870
 H 8.75360 1.41432 8.14150

1-Co (S = 3/2)

Energy = -2867.756390
 Enthalpy 0K = -2867.259050
 Free Energy 298K = -2867.326118

H 6.43501 0.74956 5.34748
 H 6.41484 2.30667 5.42430
 C 7.42130 -1.86318 0.79716
 H 8.01657 -2.57896 0.23673
 C 8.02685 -0.86415 1.57391
 C 10.97646 2.95428 0.22184
 H 11.35661 3.88758 -0.19231
 C 7.97036 4.29821 1.80543
 C 10.08383 0.58861 1.18452
 C 9.59643 -0.19100 5.41614
 H 9.09973 0.52203 6.07549
 C 5.92836 -0.08310 2.20433
 H 5.36287 0.63859 2.79507
 C 5.88933 3.42498 2.37011
 H 5.33993 2.64330 2.89629
 C 5.26024 -1.05655 1.46919
 H 4.16973 -1.09958 1.47212
 C 10.23890 -1.91096 0.92499
 H 9.99584 -1.84534 -0.14245
 H 9.90219 -2.89105 1.28347
 H 11.33127 -1.87992 1.01810
 C 9.87447 4.24234 3.45837
 C 10.34348 4.28853 6.19220
 H 10.48563 4.27199 7.27413
 C 9.92727 -0.91798 3.23198
 C 9.52009 4.22352 1.94320
 C 10.05729 2.92665 1.28645
 C 10.99636 -2.08769 5.06745
 H 11.64134 -2.89092 5.43132
 C 9.56562 3.31776 5.56975
 H 9.08587 2.53930 6.16444
 C 6.03001 -1.95791 0.73758
 H 5.55951 -2.73616 0.13213
 C 11.00238 0.67456 0.12258
 H 11.40307 -0.21039 -0.37077
 C 9.57444 -0.77217 1.72344
 C 10.39258 -1.19754 5.95260
 H 10.53249 -1.27462 7.03220
 C 10.15897 5.44169 1.24798
 H 11.25187 5.42488 1.33777
 H 9.80317 6.37915 1.69174
 H 9.91483 5.46553 0.17901
 C 5.20006 4.44780 1.72732
 H 4.10899 4.47038 1.73785

C 5.94977 5.42527 1.07716
 H 5.46231 6.24603 0.54581
 C 7.34283 5.35143 1.12344
 H 7.92205 6.12605 0.62810
 C 10.92621 5.26585 5.38879
 H 11.55628 6.04644 5.82152
 C 10.68296 5.24462 4.01463
 H 11.12637 6.01934 3.39475
 C 10.75501 -1.94939 3.69977
 H 11.21514 -2.65653 3.01493
 N 9.38068 -0.03971 4.09815
 N 7.26732 0.02575 2.24714
 N 9.63962 1.73088 1.73795
 N 11.43253 1.84175 -0.33963
 N 9.34827 3.27965 4.24391
 N 7.23020 3.33769 2.39857
 Co 8.24504 1.64745 3.31842
 O 6.81448 1.55402 4.96292

1'-Co (S = 1/2)

Energy = -2924.051520
 Enthalpy 0K = -2923.530613
 Free Energy 298K = -2923.604277
 C 7.41806 -1.69830 0.75410
 H 7.99318 -2.39994 0.15610
 C 8.06097 -0.74038 1.54964
 C 11.05156 2.96171 0.15895
 H 11.43332 3.89967 -0.24613
 C 8.00751 4.17967 1.77472
 C 10.15014 0.60482 1.11061
 C 9.51006 -0.06776 5.36265
 H 8.95398 0.61238 6.00195
 C 5.99592 0.03461 2.28045
 H 5.45746 0.71139 2.93820
 C 5.96082 3.30144 2.43731
 H 5.43842 2.55719 3.03211
 C 5.29589 -0.90358 1.53030
 H 4.20664 -0.94318 1.58322
 C 10.23294 -1.89966 0.93088
 H 9.99985 -1.85456 -0.14000
 H 9.86562 -2.85889 1.31638
 H 11.32594 -1.89502 1.02388
 C 9.88584 4.12693 3.41955
 C 10.25124 4.14835 6.16496
 H 10.33642 4.12842 7.25270
 C 9.93760 -0.79712 3.19839
 C 9.55353 4.17126 1.90511
 C 10.12439 2.92166 1.21473
 C 10.97707 -1.92649 5.07793
 H 11.63065 -2.70960 5.46895
 C 9.47165 3.20152 5.51006
 H 8.92712 2.45818 6.08580
 C 6.02573 -1.77477 0.72705
 H 5.52718 -2.52072 0.10387
 C 11.07670 0.68022 0.05619
 H 11.47847 -0.20878 -0.43163
 C 9.60632 -0.71347 1.68532
 C 10.30898 -1.05343 5.93121
 H 10.39658 -1.12752 7.01641
 C 10.15389 5.43322 1.25973
 H 11.24689 5.44245 1.35191
 H 9.76825 6.34649 1.72976
 H 9.91976 5.48017 0.18909
 C 5.23993 4.29260 1.78093

| | | | | | | | |
|------------------|----------|----------|----------------|----|----------|----------|----------|
| H | 4.15049 | 4.30729 | 1.84205 | H | 10.54416 | -1.28319 | 7.03244 |
| C | 5.94939 | 5.24831 | 1.05972 | C | 10.15982 | 5.43935 | 1.25298 |
| H | 5.43394 | 6.04002 | 0.51123 | H | 11.25222 | 5.42393 | 1.34854 |
| C | 7.34305 | 5.19537 | 1.07411 | H | 9.80116 | 6.37850 | 1.69069 |
| H | 7.90210 | 5.96050 | 0.54235 | H | 9.92076 | 5.45829 | 0.18285 |
| C | 10.90314 | 5.10561 | 5.39332 | C | 5.20103 | 4.44426 | 1.69616 |
| H | 11.54158 | 5.86331 | 5.85334 | H | 4.10990 | 4.46475 | 1.69391 |
| C | 10.70189 | 5.09928 | 4.01323 | C | 5.95585 | 5.42315 | 1.05502 |
| H | 11.17976 | 5.86684 | 3.41030 | H | 5.47344 | 6.24364 | 0.51866 |
| C | 10.77321 | -1.80207 | 3.70390 | C | 7.34826 | 5.35079 | 1.11663 |
| H | 11.26519 | -2.50359 | 3.03552 | H | 7.93159 | 6.12660 | 0.62829 |
| N | 9.35116 | 0.09166 | 4.03663 | C | 10.91512 | 5.28223 | 5.39432 |
| N | 7.33592 | 0.15010 | 2.26897 | H | 11.53897 | 6.06813 | 5.82638 |
| N | 9.70036 | 1.73602 | 1.66053 | C | 10.66971 | 5.25935 | 4.02086 |
| N | 11.51208 | 1.84886 | -0.40204 | H | 11.10474 | 6.03834 | 3.40057 |
| N | 9.31725 | 3.15616 | 4.17478 | C | 10.74141 | -1.96643 | 3.70114 |
| N | 7.30260 | 3.21248 | 2.41006 | H | 11.19015 | -2.67931 | 3.01487 |
| Co | 8.29132 | 1.64999 | 3.26248 | N | 9.38795 | -0.04276 | 4.10314 |
| C | 5.95397 | 1.48719 | 5.93223 | N | 7.26031 | 0.02658 | 2.24122 |
| N | 6.71416 | 1.54784 | 5.05965 | N | 9.63659 | 1.72987 | 1.75874 |
| C | 4.99787 | 1.41027 | 7.02833 | N | 11.44076 | 1.84092 | -0.31696 |
| H | 5.52765 | 1.26403 | 7.98220 | N | 9.35208 | 3.28324 | 4.25100 |
| H | 4.41321 | 2.34119 | 7.08656 | N | 7.22440 | 3.33617 | 2.38961 |
| H | 4.31118 | 0.56415 | 6.87218 | Co | 8.22632 | 1.64582 | 3.35380 |
| | | | | C | 6.00753 | 1.50079 | 5.83811 |
| | | | | N | 6.76385 | 1.55385 | 4.96383 |
| | | | | C | 5.05477 | 1.43394 | 6.93509 |
| Energy | | | = -2924.055978 | H | 5.58867 | 1.29326 | 7.88775 |
| Enthalpy 0K | | | = -2923.536544 | H | 4.47332 | 2.36744 | 6.98706 |
| Free Energy 298K | | | = -2923.611360 | H | 4.36705 | 0.58727 | 6.78538 |
| C | 7.42395 | -1.86090 | 0.79014 | | | | |
| H | 8.02276 | -2.57708 | 0.23424 | | | | |
| C | 8.02395 | -0.86282 | 1.57258 | | | | |
| C | 10.98144 | 2.95193 | 0.24416 | | | | |
| H | 11.36287 | 3.88571 | -0.16732 | | | | |
| C | 7.96943 | 4.29702 | 1.80375 | | | | |
| C | 10.08259 | 0.58984 | 1.20161 | | | | |
| C | 9.60754 | -0.19545 | 5.41949 | | | | |
| H | 9.11964 | 0.52083 | 6.08151 | | | | |
| C | 5.92273 | -0.08156 | 2.18747 | | | | |
| H | 5.35356 | 0.63803 | 2.77613 | | | | |
| C | 5.88489 | 3.42315 | 2.34763 | | | | |
| H | 5.33111 | 2.64264 | 2.86963 | | | | |
| C | 5.25920 | -1.05178 | 1.44374 | | | | |
| H | 4.16866 | -1.09239 | 1.43623 | | | | |
| C | 10.23694 | -1.90879 | 0.92711 | | | | |
| H | 9.99828 | -1.83718 | -0.14087 | | | | |
| H | 9.89728 | -2.89027 | 1.27875 | | | | |
| H | 11.32890 | -1.88009 | 1.02519 | | | | |
| C | 9.87005 | 4.24930 | 3.46545 | | | | |
| C | 10.34200 | 4.29945 | 6.19759 | | | | |
| H | 10.48655 | 4.28308 | 7.27915 | | | | |
| C | 9.92308 | -0.92604 | 3.23612 | | | | |
| C | 9.51840 | 4.22406 | 1.95114 | | | | |
| C | 10.05754 | 2.92381 | 1.30467 | | | | |
| C | 10.98835 | -2.10541 | 5.06744 | | | | |
| H | 11.62670 | -2.91480 | 5.42927 | | | | |
| C | 9.56927 | 3.32409 | 5.57596 | | | | |
| H | 9.09422 | 2.54442 | 6.17246 | | | | |
| C | 6.03327 | -1.95352 | 0.71791 | | | | |
| H | 5.56715 | -2.73021 | 0.10710 | | | | |
| C | 11.00553 | 0.67522 | 0.14337 | | | | |
| H | 11.40579 | -0.21028 | -0.34898 | | | | |
| C | 9.57100 | -0.77347 | 1.72954 | | | | |
| C | 10.39848 | -1.20731 | 5.95357 | | | | |

| | | | | | | | |
|----|----------|----------|----------|----|----------|----------|----------|
| C | 12.45743 | 7.97497 | 18.32143 | C | 3.25085 | 1.94924 | 8.86043 |
| C | 9.45388 | 7.49168 | 21.39980 | H | 2.16906 | 1.90295 | 8.71501 |
| H | 8.57335 | 8.13402 | 21.43899 | C | 3.94385 | 0.83438 | 9.31936 |
| C | 9.46456 | 6.28193 | 22.08663 | H | 3.39300 | -0.07747 | 9.52852 |
| H | 8.59372 | 5.96607 | 22.66358 | C | 5.33042 | 0.91374 | 9.49821 |
| C | 10.61549 | 5.50191 | 22.00600 | C | 8.60048 | 1.21962 | 12.36680 |
| H | 10.67795 | 4.53899 | 22.51841 | H | 9.40364 | 1.94948 | 12.25381 |
| C | 11.70174 | 5.97100 | 21.26489 | C | 8.39744 | 0.58834 | 13.58906 |
| H | 12.59987 | 5.36170 | 21.21590 | H | 9.02887 | 0.82753 | 14.44651 |
| C | 12.83782 | 7.78565 | 19.81423 | C | 6.61990 | -0.63216 | 12.52844 |
| C | 14.02342 | 6.80373 | 19.90202 | H | 5.83315 | -1.37883 | 12.59378 |
| H | 13.76418 | 5.82623 | 19.47780 | C | 6.89337 | 0.03214 | 11.32277 |
| H | 14.32940 | 6.64485 | 20.94292 | C | 6.14220 | -0.31498 | 10.00432 |
| H | 14.89447 | 7.18080 | 19.35265 | C | 5.17979 | -1.48974 | 10.26682 |
| C | 13.24656 | 9.14714 | 20.45615 | H | 4.64949 | -1.78526 | 9.35378 |
| C | 14.53468 | 9.36097 | 20.96945 | H | 4.42544 | -1.22723 | 11.01798 |
| H | 15.30715 | 8.60114 | 20.88861 | H | 5.72345 | -2.36808 | 10.63489 |
| C | 14.84474 | 10.56328 | 21.60788 | C | 7.19411 | -0.74186 | 8.93791 |
| H | 15.84870 | 10.72732 | 22.00645 | C | 7.17532 | -2.01754 | 8.35360 |
| C | 13.85480 | 11.53410 | 21.74208 | H | 6.39940 | -2.73755 | 8.59852 |
| H | 14.04052 | 12.47943 | 22.25478 | C | 8.16754 | -2.38825 | 7.44486 |
| C | 12.60525 | 11.26182 | 21.19535 | H | 8.15117 | -3.38218 | 6.99157 |
| H | 11.79404 | 11.98582 | 21.28640 | C | 9.17663 | -1.47833 | 7.13848 |
| C | 11.62455 | 7.20743 | 20.60661 | H | 9.98440 | -1.72429 | 6.44706 |
| N | 10.44324 | 11.58473 | 18.22180 | C | 9.12456 | -0.22767 | 7.74433 |
| N | 8.55360 | 9.47763 | 18.44262 | H | 9.89757 | 0.51217 | 7.53369 |
| N | 11.44776 | 8.79886 | 17.99707 | N | 8.19371 | 3.31529 | 7.64366 |
| N | 12.78696 | 7.47205 | 16.00912 | N | 7.81712 | 4.12375 | 10.30868 |
| N | 10.49135 | 7.93952 | 20.67310 | N | 5.98265 | 2.06495 | 9.22749 |
| N | 12.31261 | 10.11765 | 20.55426 | N | 7.86095 | 0.97017 | 11.27231 |
| O | 9.31130 | 10.89622 | 21.16352 | N | 8.16060 | 0.13976 | 8.60582 |
| Zn | 10.40886 | 9.81991 | 19.53383 | O | 10.25083 | 2.22332 | 9.73281 |
| H | 8.60799 | 11.55332 | 21.05315 | Co | 8.07714 | 2.12879 | 9.45725 |
| H | 9.45352 | 10.79855 | 22.11685 | H | 10.82835 | 2.97993 | 9.55113 |

2-Co (S = 3/2)

| | |
|------------------|----------------|
| Energy | = -2867.755038 |
| Enthalpy 0K | = -2867.257601 |
| Free Energy 298K | = -2867.324549 |

| | | | |
|---|----------|---------|----------|
| C | 9.17942 | 3.14031 | 6.74661 |
| H | 9.97745 | 2.45284 | 7.03031 |
| C | 9.22501 | 3.79354 | 5.51977 |
| H | 10.05109 | 3.61518 | 4.82920 |
| C | 8.18648 | 4.67029 | 5.21372 |
| H | 8.16337 | 5.20191 | 4.25954 |
| C | 7.17620 | 4.87679 | 6.15427 |
| H | 6.38022 | 5.57781 | 5.91804 |
| C | 7.20438 | 4.19409 | 7.37964 |
| C | 6.14215 | 4.45566 | 8.48738 |
| C | 5.18365 | 5.57070 | 8.02321 |
| H | 4.42972 | 5.79436 | 8.78752 |
| H | 4.65257 | 5.28405 | 7.10793 |
| H | 5.73007 | 6.49821 | 7.81318 |
| C | 6.88069 | 4.92452 | 9.76755 |
| C | 6.63222 | 6.17631 | 10.36857 |
| H | 5.87096 | 6.85448 | 9.98308 |
| C | 8.24731 | 5.81969 | 11.93229 |
| H | 8.81484 | 6.17763 | 12.79759 |
| C | 8.50347 | 4.56863 | 11.37061 |
| H | 9.27739 | 3.92064 | 11.78299 |
| C | 5.33003 | 3.15973 | 8.78066 |
| C | 3.94426 | 3.12365 | 8.58761 |
| H | 3.39513 | 3.98918 | 8.22931 |

| | | | |
|----|----------|----------|----------|
| C | 3.25085 | 1.94924 | 8.86043 |
| H | 2.16906 | 1.90295 | 8.71501 |
| C | 3.94385 | 0.83438 | 9.31936 |
| H | 3.39300 | -0.07747 | 9.52852 |
| C | 5.33042 | 0.91374 | 9.49821 |
| C | 8.60048 | 1.21962 | 12.36680 |
| H | 9.40364 | 1.94948 | 12.25381 |
| C | 8.39744 | 0.58834 | 13.58906 |
| H | 9.02887 | 0.82753 | 14.44651 |
| C | 6.61990 | -0.63216 | 12.52844 |
| H | 5.83315 | -1.37883 | 12.59378 |
| C | 6.89337 | 0.03214 | 11.32277 |
| C | 6.14220 | -0.31498 | 10.00432 |
| C | 5.17979 | -1.48974 | 10.26682 |
| H | 4.64949 | -1.78526 | 9.35378 |
| H | 4.42544 | -1.22723 | 11.01798 |
| H | 5.72345 | -2.36808 | 10.63489 |
| C | 7.19411 | -0.74186 | 8.93791 |
| C | 7.17532 | -2.01754 | 8.35360 |
| H | 6.39940 | -2.73755 | 8.59852 |
| C | 8.16754 | -2.38825 | 7.44486 |
| H | 8.15117 | -3.38218 | 6.99157 |
| C | 9.17663 | -1.47833 | 7.13848 |
| H | 9.98440 | -1.72429 | 6.44706 |
| C | 9.12456 | -0.22767 | 7.74433 |
| H | 9.89757 | 0.51217 | 7.53369 |
| N | 8.19371 | 3.31529 | 7.64366 |
| N | 7.81712 | 4.12375 | 10.30868 |
| N | 5.98265 | 2.06495 | 9.22749 |
| N | 7.86095 | 0.97017 | 11.27231 |
| N | 8.16060 | 0.13976 | 8.60582 |
| O | 10.25083 | 2.22332 | 9.73281 |
| Co | 8.07714 | 2.12879 | 9.45725 |
| H | 10.82835 | 2.97993 | 9.55113 |
| H | 10.82309 | 1.48662 | 9.99538 |
| C | 7.36813 | -0.34704 | 13.67142 |
| H | 7.15353 | -0.86411 | 14.60960 |
| N | 7.30173 | 6.60919 | 11.43332 |

2'-Co (S = 1/2)

| | | | |
|------------------|----------------|---------|----------|
| Energy | = -2924.050471 | | |
| Enthalpy 0K | = -2923.529607 | | |
| Free Energy 298K | = -2923.603495 | | |
| C | 9.15714 | 3.01032 | 6.87134 |
| H | 9.96718 | 2.36356 | 7.19643 |
| C | 9.22222 | 3.62809 | 5.62772 |
| H | 10.07238 | 3.44265 | 4.96919 |
| C | 8.18220 | 4.47896 | 5.26501 |
| H | 8.17334 | 4.97852 | 4.29356 |
| C | 7.15889 | 4.70985 | 6.18383 |
| H | 6.36505 | 5.40372 | 5.92073 |
| C | 7.16554 | 4.06801 | 7.42972 |
| C | 6.10389 | 4.39826 | 8.51080 |
| C | 5.21440 | 5.56582 | 8.04535 |
| H | 4.46312 | 5.81862 | 8.80321 |
| H | 4.67333 | 5.30717 | 7.12739 |
| H | 5.80723 | 6.46703 | 7.84398 |
| C | 6.86373 | 4.79540 | 9.79580 |
| C | 6.63969 | 6.02721 | 10.44294 |
| H | 5.86240 | 6.71022 | 10.10049 |
| C | 8.31795 | 5.63618 | 11.91851 |
| H | 8.93317 | 5.97352 | 12.75896 |
| C | 8.55066 | 4.40005 | 11.31722 |

| | | | | | | | |
|----|----------|----------|----------|----|---------|----------|----------|
| H | 9.35934 | 3.76317 | 11.66428 | H | 5.74172 | 6.49958 | 7.82074 |
| C | 5.23612 | 3.15091 | 8.78526 | C | 6.89683 | 4.92952 | 9.77328 |
| C | 3.84904 | 3.13997 | 8.59897 | C | 6.65942 | 6.18745 | 10.36628 |
| H | 3.31021 | 4.01736 | 8.25059 | H | 5.90398 | 6.86946 | 9.97642 |
| C | 3.14706 | 1.96801 | 8.86950 | C | 8.27278 | 5.82738 | 11.92949 |
| H | 2.06388 | 1.93155 | 8.73252 | H | 8.84381 | 6.18544 | 12.79240 |
| C | 3.83568 | 0.84319 | 9.31622 | C | 8.51997 | 4.57253 | 11.37323 |
| H | 3.28490 | -0.06985 | 9.52613 | H | 9.29436 | 3.92462 | 11.78457 |
| C | 5.22373 | 0.92295 | 9.48223 | C | 5.34226 | 3.16403 | 8.79891 |
| C | 8.59580 | 1.27809 | 12.25353 | C | 3.95498 | 3.13551 | 8.61297 |
| H | 9.43075 | 1.95908 | 12.11016 | H | 3.40874 | 4.00556 | 8.26178 |
| C | 8.40559 | 0.66407 | 13.48589 | C | 3.25678 | 1.96382 | 8.88176 |
| H | 9.07476 | 0.88837 | 14.31821 | H | 2.17420 | 1.92285 | 8.74121 |
| C | 6.57587 | -0.51601 | 12.48614 | C | 3.94872 | 0.84432 | 9.32879 |
| H | 5.777657 | -1.24715 | 12.57019 | H | 3.39616 | -0.06738 | 9.53309 |
| C | 6.83609 | 0.12304 | 11.26583 | C | 5.33662 | 0.91774 | 9.50277 |
| C | 6.07591 | -0.26786 | 9.97101 | C | 8.55305 | 1.21839 | 12.40054 |
| C | 5.16848 | -1.48194 | 10.23983 | H | 9.34432 | 1.96337 | 12.30850 |
| H | 4.63089 | -1.78369 | 9.33291 | C | 8.34581 | 0.57210 | 13.61431 |
| H | 4.41299 | -1.25035 | 11.00024 | H | 8.96082 | 0.81556 | 14.48234 |
| H | 5.74785 | -2.34476 | 10.59164 | C | 6.60533 | -0.66901 | 12.51961 |
| C | 7.14095 | -0.62538 | 8.90031 | H | 5.83192 | -1.43068 | 12.56685 |
| C | 7.12814 | -1.86520 | 8.24641 | C | 6.87915 | 0.01591 | 11.32541 |
| H | 6.32528 | -2.57605 | 8.42085 | C | 6.13932 | -0.31982 | 9.99911 |
| C | 8.15612 | -2.21415 | 7.37102 | C | 5.17076 | -1.49308 | 10.24604 |
| H | 8.14166 | -3.18064 | 6.86209 | H | 4.64091 | -1.77561 | 9.32869 |
| C | 9.20756 | -1.32059 | 7.18947 | H | 4.41655 | -1.23605 | 10.99912 |
| H | 10.06160 | -1.55582 | 6.55223 | H | 5.70971 | -2.37810 | 10.60470 |
| C | 9.14825 | -0.10012 | 7.85210 | C | 7.19501 | -0.74374 | 8.93551 |
| H | 9.96664 | 0.60789 | 7.75477 | C | 7.16787 | -2.01410 | 8.33961 |
| N | 8.14403 | 3.18399 | 7.73920 | H | 6.38235 | -2.72799 | 8.57085 |
| N | 7.81006 | 3.96480 | 10.28623 | C | 8.16408 | -2.38729 | 7.43652 |
| N | 5.88282 | 2.05914 | 9.21648 | H | 8.14048 | -3.37648 | 6.97332 |
| N | 7.81339 | 1.05702 | 11.18171 | C | 9.18675 | -1.48646 | 7.15032 |
| N | 8.13092 | 0.26556 | 8.65249 | H | 9.99928 | -1.73552 | 6.46570 |
| Co | 8.01903 | 2.11741 | 9.46965 | C | 9.14260 | -0.24126 | 7.76792 |
| C | 7.34955 | -0.23401 | 13.61161 | H | 9.92852 | 0.48967 | 7.57737 |
| H | 7.14147 | -0.73169 | 14.56154 | N | 8.18742 | 3.29783 | 7.63608 |
| N | 7.34519 | 6.43446 | 11.49343 | N | 7.82750 | 4.12524 | 10.31707 |
| C | 11.56109 | 2.29700 | 9.88652 | N | 5.99543 | 2.06687 | 9.23925 |
| N | 10.41165 | 2.23108 | 9.75345 | N | 7.83230 | 0.96855 | 11.29491 |
| C | 13.00538 | 2.38015 | 10.05568 | N | 8.17288 | 0.13056 | 8.61961 |
| H | 13.51253 | 2.08448 | 9.12442 | Co | 8.11320 | 2.12632 | 9.47878 |
| H | 13.29812 | 3.41136 | 10.30631 | C | 7.33506 | -0.38442 | 13.67424 |
| H | 13.33058 | 1.70797 | 10.86457 | H | 7.11990 | -0.91686 | 14.60361 |

2'-Co (S = 3/2)

| | |
|------------------|----------------|
| Energy | = -2924.054424 |
| Enthalpy 0K | = -2923.535028 |
| Free Energy 298K | = -2923.610196 |

| | | | |
|---|----------|---------|---------|
| C | 9.15638 | 3.10912 | 6.72533 |
| H | 9.94696 | 2.40911 | 6.99572 |
| C | 9.19419 | 3.76114 | 5.49736 |
| H | 10.00641 | 3.56857 | 4.79437 |
| C | 8.16665 | 4.65499 | 5.20639 |
| H | 8.13850 | 5.18829 | 4.25330 |
| C | 7.17221 | 4.87458 | 6.16053 |
| H | 6.38340 | 5.58692 | 5.93515 |
| C | 7.20616 | 4.18874 | 7.38405 |
| C | 6.15326 | 4.45819 | 8.49864 |
| C | 5.19441 | 5.57407 | 8.03676 |
| H | 4.44608 | 5.80210 | 8.80523 |
| H | 4.65709 | 5.28626 | 7.12557 |

| | | | |
|----|----------|----------|----------|
| H | 5.74172 | 6.49958 | 7.82074 |
| C | 6.89683 | 4.92952 | 9.77328 |
| C | 6.65942 | 6.18745 | 10.36628 |
| H | 5.90398 | 6.86946 | 9.97642 |
| C | 8.27278 | 5.82738 | 11.92949 |
| H | 8.84381 | 6.18544 | 12.79240 |
| C | 8.51997 | 4.57253 | 11.37323 |
| H | 9.29436 | 3.92462 | 11.78457 |
| C | 5.34226 | 3.16403 | 8.79891 |
| C | 3.95498 | 3.13551 | 8.61297 |
| H | 3.40874 | 4.00556 | 8.26178 |
| C | 3.25678 | 1.96382 | 8.88176 |
| H | 2.17420 | 1.92285 | 8.74121 |
| C | 3.94872 | 0.84432 | 9.32879 |
| H | 3.39616 | -0.06738 | 9.53309 |
| C | 5.33662 | 0.91774 | 9.50277 |
| C | 8.55305 | 1.21839 | 12.40054 |
| H | 9.34432 | 1.96337 | 12.30850 |
| C | 8.34581 | 0.57210 | 13.61431 |
| H | 8.96082 | 0.81556 | 14.48234 |
| C | 6.60533 | -0.66901 | 12.51961 |
| H | 5.83192 | -1.43068 | 12.56685 |
| C | 6.87915 | 0.01591 | 11.32541 |
| C | 6.13932 | -0.31982 | 9.99911 |
| C | 5.17076 | -1.49308 | 10.24604 |
| H | 4.64091 | -1.77561 | 9.32869 |
| H | 4.41655 | -1.23605 | 10.99912 |
| H | 5.70971 | -2.37810 | 10.60470 |
| C | 7.19501 | -0.74374 | 8.93551 |
| C | 7.16787 | -2.01410 | 8.33961 |
| H | 6.38235 | -2.72799 | 8.57085 |
| C | 8.16408 | -2.38729 | 7.43652 |
| H | 8.14048 | -3.37648 | 6.97332 |
| C | 9.18675 | -1.48646 | 7.15032 |
| H | 9.99928 | -1.73552 | 6.46570 |
| C | 9.14260 | -0.24126 | 7.76792 |
| H | 9.92852 | 0.48967 | 7.57737 |
| N | 8.18742 | 3.29783 | 7.63608 |
| N | 7.82750 | 4.12524 | 10.31707 |
| N | 5.99543 | 2.06687 | 9.23925 |
| N | 7.83230 | 0.96855 | 11.29491 |
| N | 8.17288 | 0.13056 | 8.61961 |
| Co | 8.11320 | 2.12632 | 9.47878 |
| C | 7.33506 | -0.38442 | 13.67424 |
| H | 7.11990 | -0.91686 | 14.60361 |
| N | 7.33275 | 6.62179 | 11.42789 |
| C | 11.43830 | 2.26385 | 9.85682 |
| N | 10.29064 | 2.21329 | 9.71614 |
| C | 12.88115 | 2.32561 | 10.03025 |
| H | 13.33462 | 2.90134 | 9.20855 |
| H | 13.12411 | 2.81860 | 10.98436 |
| H | 13.30410 | 1.30910 | 10.03595 |

2-Zn (S = 0)

| | | | |
|------------------|----------------|---------|---------|
| Energy | = -3264.355648 | | |
| Enthalpy 0K | = -3263.858693 | | |
| Free Energy 298K | = -3263.926790 | | |
| C | 4.07772 | 5.32198 | 6.14948 |
| H | 4.88991 | 4.59308 | 6.16290 |
| C | 3.79093 | 6.04180 | 4.99491 |
| H | 4.36451 | 5.87585 | 4.08141 |
| C | 2.75325 | 6.96980 | 5.05387 |
| H | 2.47310 | 7.55343 | 4.17373 |

| | | | | | | | |
|----|----------|----------|----------|---|----------|----------|----------|
| C | 2.08265 | 7.16488 | 6.26270 | C | 3.28553 | 8.63900 | 4.03592 |
| H | 1.29198 | 7.90900 | 6.30290 | H | 2.22508 | 8.50189 | 4.25446 |
| C | 2.43830 | 6.41729 | 7.39599 | C | 4.01723 | 9.69525 | 4.57428 |
| C | 1.78322 | 6.67286 | 8.78854 | H | 3.54776 | 10.41867 | 5.24502 |
| C | 0.83458 | 7.88314 | 8.67830 | C | 5.36398 | 9.82878 | 4.23351 |
| H | 1.36992 | 8.77460 | 8.33018 | H | 5.92369 | 10.66520 | 4.64317 |
| H | 0.38112 | 8.12422 | 9.64693 | C | 5.96947 | 8.90064 | 3.37258 |
| H | 0.01942 | 7.68736 | 7.97155 | C | 7.44857 | 9.05289 | 2.91117 |
| C | 2.90812 | 6.99742 | 9.82069 | C | 8.03545 | 10.34496 | 3.51185 |
| C | 2.96336 | 8.23286 | 10.48336 | H | 8.02043 | 10.31907 | 4.60789 |
| H | 2.21260 | 8.99895 | 10.31170 | H | 7.46081 | 11.22248 | 3.19205 |
| C | 3.99785 | 8.50358 | 11.38082 | H | 9.07512 | 10.49592 | 3.19809 |
| H | 4.03794 | 9.46714 | 11.89426 | C | 7.47084 | 9.15526 | 1.35771 |
| C | 4.97159 | 7.53334 | 11.60348 | C | 8.00451 | 10.27255 | 0.69737 |
| H | 5.80360 | 7.69825 | 12.29014 | H | 8.46295 | 11.08776 | 1.25050 |
| C | 4.85247 | 6.32927 | 10.91718 | C | 7.94682 | 10.35897 | -0.69441 |
| H | 5.59327 | 5.54262 | 11.06431 | H | 8.36349 | 11.22934 | -1.20673 |
| C | 0.96797 | 5.43038 | 9.26355 | C | 7.33938 | 9.32991 | -1.41072 |
| C | -0.39705 | 5.52754 | 9.56231 | H | 7.24928 | 9.35942 | -2.49798 |
| H | -0.93924 | 6.46344 | 9.46628 | C | 6.84545 | 8.24687 | -0.69203 |
| C | -1.07959 | 4.39546 | 9.99608 | H | 6.35325 | 7.42406 | -1.21250 |
| H | -2.14391 | 4.45499 | 10.23532 | C | 8.29263 | 7.83335 | 3.38601 |
| C | -0.40207 | 3.18667 | 10.12229 | C | 9.39445 | 8.00889 | 4.23254 |
| H | -0.94658 | 2.31001 | 10.46039 | H | 9.69333 | 8.99249 | 4.58177 |
| C | 0.96133 | 3.13428 | 9.80800 | C | 10.12722 | 6.89895 | 4.63859 |
| C | 5.03421 | 2.83520 | 11.42674 | H | 10.98923 | 7.02015 | 5.29878 |
| H | 5.83866 | 3.49644 | 11.10052 | C | 9.75503 | 5.63287 | 4.19824 |
| C | 5.16396 | 2.08626 | 12.59120 | H | 10.33229 | 4.77108 | 4.51959 |
| H | 6.06529 | 2.16390 | 13.20158 | C | 8.64758 | 5.50310 | 3.35328 |
| C | 4.10945 | 1.24513 | 12.93976 | C | 7.40693 | 4.78393 | -0.74119 |
| H | 4.15079 | 0.64222 | 13.85000 | H | 6.71849 | 5.42878 | -1.28842 |
| C | 2.99912 | 1.16239 | 12.09744 | C | 8.23257 | 3.88518 | -1.41747 |
| H | 2.19371 | 0.48379 | 12.36479 | H | 8.21017 | 3.80887 | -2.50955 |
| C | 2.94201 | 1.93522 | 10.92773 | C | 9.07705 | 3.18050 | 0.57596 |
| C | 1.76458 | 1.80155 | 9.91377 | H | 9.77176 | 2.51391 | 1.08740 |
| C | 0.81996 | 0.67626 | 10.38260 | C | 8.24138 | 4.06572 | 1.28797 |
| H | 1.35328 | -0.27807 | 10.46988 | C | 8.19941 | 4.10379 | 2.83766 |
| H | -0.00833 | 0.53177 | 9.67876 | C | 9.13837 | 3.01534 | 3.39742 |
| H | 0.38511 | 0.90514 | 11.36257 | H | 8.84275 | 2.01922 | 3.04550 |
| C | 2.34901 | 1.42261 | 8.52402 | H | 9.12138 | 2.99867 | 4.49365 |
| C | 2.00131 | 0.22528 | 7.86406 | H | 10.17565 | 3.18352 | 3.08426 |
| H | 1.28193 | -0.47453 | 8.28880 | C | 6.74611 | 3.80532 | 3.28941 |
| C | 3.40885 | 0.68950 | 6.13537 | C | 6.42376 | 2.71653 | 4.12590 |
| H | 3.84885 | 0.39817 | 5.17596 | H | 7.19396 | 2.06273 | 4.53532 |
| C | 3.76404 | 1.88707 | 6.75796 | C | 4.19966 | 3.18029 | 3.98029 |
| H | 4.48712 | 2.56084 | 6.29720 | H | 3.16664 | 2.92693 | 4.24042 |
| N | 3.40926 | 5.48386 | 7.30412 | C | 4.48926 | 4.27536 | 3.16550 |
| N | 3.85433 | 6.06308 | 10.05811 | H | 3.68969 | 4.90076 | 2.76697 |
| N | 1.60116 | 4.24683 | 9.39447 | N | 5.25665 | 7.86154 | 2.89026 |
| N | 3.95129 | 2.78231 | 10.63206 | N | 6.92235 | 8.14843 | 0.64651 |
| N | 3.23352 | 2.24859 | 7.93337 | N | 7.94930 | 6.59447 | 2.97090 |
| N | 2.52221 | -0.12878 | 6.69197 | N | 7.42444 | 4.88150 | 0.59579 |
| O | 5.88843 | 4.03555 | 8.52955 | N | 9.07356 | 3.10028 | -0.75177 |
| Zn | 3.67265 | 4.16450 | 8.99263 | N | 5.75050 | 4.59098 | 2.83821 |
| H | 6.47106 | 4.78185 | 8.32372 | N | 5.17333 | 2.41787 | 4.46720 |
| H | 6.44370 | 3.24178 | 8.51466 | O | 4.54983 | 6.12414 | 0.38098 |

3-Co (S = 3/2)

Energy = -2883.761320
Enthalpy 0K = -2883.275898
Free Energy 298K = -2883.342626

C 3.95357 7.74958 3.20156
H 3.41594 6.91376 2.75159

3'-Co (S = 1/2)

Energy = -2940.057569
Enthalpy 0K = -2939.548713
Free Energy 298K = -2939.621933

3'-Co (S = 3/2)

| | | | | | |
|----|----------|----------|----------|------------------|---------------------------|
| C | 4.02205 | 7.67581 | 3.08208 | Energy | = -2940.061515 |
| H | 3.48751 | 6.89853 | 2.54187 | Enthalpy 0K | = -2939.554100 |
| C | 3.33086 | 8.52981 | 3.93346 | Free Energy 298K | = -2939.628568 |
| H | 2.25752 | 8.40222 | 4.08354 | | |
| C | 4.04875 | 9.53776 | 4.57104 | | |
| H | 3.56058 | 10.22641 | 5.26445 | C | 3.95717 7.76088 3.20975 |
| C | 5.40586 | 9.67671 | 4.28114 | H | 3.41669 6.92517 2.76336 |
| H | 5.96001 | 10.48824 | 4.74475 | C | 3.29060 8.65366 4.04175 |
| C | 6.03777 | 8.79369 | 3.39460 | H | 2.23078 8.51697 4.26335 |
| C | 7.51092 | 9.00676 | 2.95540 | C | 4.02203 9.71362 4.57247 |
| C | 8.04410 | 10.33306 | 3.52604 | H | 3.55339 10.44116 5.23925 |
| H | 8.02951 | 10.32631 | 4.62254 | C | 5.36772 9.84529 4.22826 |
| H | 7.44335 | 11.18574 | 3.18555 | H | 5.92784 10.68486 4.63067 |
| H | 9.08215 | 10.50755 | 3.21860 | C | 5.97208 8.91064 3.37326 |
| C | 7.52719 | 9.05107 | 1.40426 | C | 7.45039 9.05592 2.91256 |
| C | 8.08373 | 10.13469 | 0.71079 | C | 8.04140 10.34641 3.51296 |
| H | 8.58996 | 10.93421 | 1.24479 | H | 8.02237 10.32227 4.60892 |
| C | 7.98342 | 10.21255 | -0.67812 | H | 7.47125 11.22579 3.19044 |
| H | 8.42107 | 11.05720 | -1.21515 | H | 9.08262 10.49270 3.20225 |
| C | 7.28782 | 9.21340 | -1.35320 | C | 7.48034 9.15526 1.35886 |
| H | 7.13514 | 9.24537 | -2.43326 | C | 8.01537 10.27473 0.70322 |
| C | 6.77897 | 8.15612 | -0.60808 | H | 8.46406 11.09265 1.26001 |
| H | 6.20212 | 7.37540 | -1.09747 | C | 7.97330 10.35946 -0.68909 |
| C | 8.38300 | 7.83514 | 3.45533 | H | 8.39201 11.23118 -1.19742 |
| C | 9.48200 | 8.01643 | 4.30413 | C | 7.38014 9.32616 -1.41040 |
| H | 9.77761 | 9.00139 | 4.65597 | H | 7.30456 9.35268 -2.49878 |
| C | 10.21264 | 6.90086 | 4.70523 | C | 6.88138 8.24285 -0.69539 |
| H | 11.07414 | 7.01615 | 5.36698 | H | 6.39942 7.41770 -1.22026 |
| C | 9.83715 | 5.63657 | 4.25687 | C | 8.28536 7.83276 3.39163 |
| H | 10.40790 | 4.76626 | 4.57060 | C | 9.38373 8.00920 4.24305 |
| C | 8.72839 | 5.52710 | 3.41062 | H | 9.68024 8.99310 4.59294 |
| C | 7.34145 | 4.94446 | -0.67082 | C | 10.11666 6.90113 4.65165 |
| H | 6.61271 | 5.56223 | -1.18774 | H | 10.97607 7.02255 5.31508 |
| C | 8.16299 | 4.07343 | -1.38561 | C | 9.74794 5.63671 4.20582 |
| H | 8.09858 | 4.01340 | -2.47689 | H | 10.32625 4.77516 4.52536 |
| C | 9.07718 | 3.35201 | 0.56175 | C | 8.64315 5.50682 3.35676 |
| H | 9.79350 | 2.68896 | 1.04721 | C | 7.41859 4.77766 -0.74261 |
| C | 8.24692 | 4.20479 | 1.31591 | H | 6.72632 5.41612 -1.29161 |
| C | 8.23108 | 4.17209 | 2.86023 | C | 8.25024 3.88223 -1.41560 |
| C | 9.12444 | 3.02796 | 3.37567 | H | 8.23151 3.80460 -2.50763 |
| H | 8.79039 | 2.05366 | 2.99646 | C | 9.09172 3.18423 0.58042 |
| H | 9.11590 | 2.98293 | 4.47140 | H | 9.78726 2.52098 1.09475 |
| H | 10.16656 | 3.17011 | 3.06514 | C | 8.25097 4.06801 1.28843 |
| C | 6.76755 | 3.94326 | 3.29873 | C | 8.20704 4.10551 2.83830 |
| C | 6.40235 | 2.87740 | 4.14492 | C | 9.15188 3.02193 3.39779 |
| H | 7.15525 | 2.21964 | 4.57907 | H | 8.86353 2.02494 3.04259 |
| C | 4.19889 | 3.37356 | 3.92982 | H | 9.13142 3.00205 4.49389 |
| H | 3.15297 | 3.14079 | 4.15465 | H | 10.18890 3.19720 3.08785 |
| C | 4.52712 | 4.44950 | 3.10565 | C | 6.75762 3.80032 3.29251 |
| H | 3.74635 | 5.05961 | 2.66057 | C | 6.43917 2.70106 4.11669 |
| N | 5.34364 | 7.76713 | 2.84726 | H | 7.21072 2.03787 4.50793 |
| N | 6.92776 | 8.04433 | 0.72432 | C | 4.21725 3.17764 4.00098 |
| N | 8.03618 | 6.61129 | 3.03370 | H | 3.18585 2.92781 4.27064 |
| N | 7.40271 | 5.03660 | 0.66662 | C | 4.50340 4.27911 3.19407 |
| N | 9.04513 | 3.29749 | -0.76601 | H | 3.70153 4.90953 2.80891 |
| N | 5.80142 | 4.75538 | 2.81620 | N | 5.25951 7.86958 2.89771 |
| N | 5.14120 | 2.60585 | 4.46540 | N | 6.94252 8.14598 0.64300 |
| Co | 6.32802 | 6.39716 | 1.73398 | N | 7.93976 6.59517 2.97484 |
| C | 3.55002 | 5.89723 | -0.41269 | N | 7.43288 4.87907 0.59296 |
| N | 4.45277 | 6.07318 | 0.29263 | N | 9.09294 3.10166 -0.74723 |
| C | 2.41538 | 5.67839 | -1.29908 | N | 5.76190 4.59375 2.85815 |
| H | 1.84323 | 6.61140 | -1.41799 | N | 5.19122 2.40295 4.46722 |
| H | 2.76709 | 5.34714 | -2.28833 | Co | 6.24993 6.37578 1.68725 |
| H | 1.75409 | 4.90307 | -0.88243 | C | 3.63085 5.97335 -0.35388 |

N 4.54672 6.10768 0.34103
 C 2.48093 5.80546 -1.22829
 H 1.69075 6.52093 -0.95243
 H 2.77539 5.98770 -2.27362
 H 2.08749 4.78069 -1.14230

3-Zn (S = 0)

Energy = -3280.361472
 Enthalpy 0K = -3279.877021
 Free Energy 298K = -3279.945654

C 6.93515 3.28673 14.26643
 H 7.25447 2.82851 15.20403
 C 5.82305 4.12061 14.23336
 H 5.26677 4.33873 15.14660
 C 5.45566 4.65808 13.00154
 H 4.59791 5.32951 12.91664
 C 6.18771 4.31381 11.86366
 H 5.87869 4.72136 10.90520
 C 7.29047 3.45217 11.96682
 C 8.08213 2.97736 10.70825
 C 7.40240 3.54681 9.44689
 H 6.35844 3.21817 9.37882
 H 7.91518 3.21843 8.53519
 H 7.40664 4.64315 9.45176
 C 8.04161 1.41806 10.64694
 C 7.47834 0.73867 9.55605
 H 7.08089 1.27723 8.70027
 C 7.41302 -0.65591 9.55428
 H 6.97403 -1.18133 8.70284
 C 7.90255 -1.35881 10.65264
 H 7.85977 -2.44800 10.70586
 C 8.45637 -0.62293 11.69470
 H 8.84500 -1.13117 12.57815
 C 9.55875 3.47827 10.75964
 C 10.07978 4.30593 9.75669
 H 9.47749 4.63315 8.91465
 C 11.40482 4.72187 9.83756
 H 11.82581 5.36641 9.06236
 C 12.19188 4.31334 10.91006
 H 13.22512 4.64380 10.96256
 C 11.62871 3.48943 11.89121
 C 11.47882 -0.56739 13.52825
 H 10.55608 -1.10794 13.74346
 C 12.69937 -1.23595 13.42634
 H 12.76266 -2.32204 13.55025
 C 13.74350 0.75450 13.05185
 H 14.68641 1.26480 12.85571
 C 12.52883 1.46187 13.17123
 C 12.46260 3.01307 13.11951
 C 13.89699 3.57274 13.01592
 H 14.39230 3.23191 12.09906
 H 14.51102 3.24953 13.86562
 H 13.89740 4.66905 13.00365
 C 11.81111 3.52444 14.43460
 C 12.48330 4.39631 15.31613
 H 13.48057 4.77663 15.09530
 C 10.74093 4.36439 16.78283
 H 10.30544 4.68977 17.73334
 C 10.04587 3.50970 15.92566
 H 9.05106 3.14610 16.18537
 N 7.65952 2.97917 13.17677
 N 8.53827 0.71837 11.69028
 N 10.34180 3.09780 11.79044

N 11.39760 0.76222 13.38150
 N 13.81962 -0.56853 13.16921
 N 10.57360 3.10555 14.76244
 N 11.95205 4.80860 16.46422
 O 8.65631 0.52279 14.88883
 Zn 9.49496 1.85454 13.27582
 H 7.76118 0.15172 14.89864
 H 9.08212 0.23488 15.71018

1'-Co+e- (S = 0)

Energy = -2924.288386
 Enthalpy 0K = -2923.772340
 Free Energy 298K = -2923.849517

C 7.51836 -2.22613 1.18595
 H 8.17337 -3.04351 0.89435
 C 8.04282 -1.01659 1.67007
 C 10.47345 2.95126 -0.01061
 H 10.72882 3.89519 -0.49325
 C 7.99751 4.44324 1.92144
 C 9.89742 0.59341 1.14918
 C 9.44188 -0.09921 5.43382
 H 8.76135 0.47407 6.06668
 C 5.89834 -0.18185 1.93818
 H 5.27883 0.65582 2.26357
 C 5.86920 3.55029 2.12460
 H 5.26646 2.67667 2.37895
 C 5.30327 -1.34181 1.45075
 H 4.21689 -1.41431 1.37398
 C 10.33568 -1.89341 1.04685
 H 9.98237 -2.00530 0.01471
 H 10.20437 -2.86078 1.54859
 H 11.41118 -1.67927 1.00676
 C 9.93552 4.09978 3.49985
 C 10.45329 3.94069 6.21265
 H 10.59891 3.86723 7.29208
 C 9.96785 -0.77765 3.28213
 C 9.53287 4.22187 2.01045
 C 9.87889 2.92221 1.25327
 C 11.30592 -1.55971 5.14307
 H 12.14068 -2.14785 5.53145
 C 9.42459 3.23952 5.58888
 H 8.75039 2.61108 6.17492
 C 6.13848 -2.39198 1.07390
 H 5.72648 -3.33001 0.69440
 C 10.49131 0.68666 -0.11179
 H 10.76098 -0.20648 -0.67630
 C 9.57356 -0.77504 1.78625
 C 10.47122 -0.84539 6.00154
 H 10.61122 -0.86194 7.08407
 C 10.27075 5.41582 1.37152
 H 11.34930 5.22496 1.30577
 H 10.12695 6.33084 1.96044
 H 9.90838 5.61506 0.35581
 C 5.25126 4.73757 1.74322
 H 4.16333 4.79721 1.67829
 C 6.06588 5.83117 1.45537
 H 5.63554 6.79125 1.16091
 C 7.44893 5.68064 1.54696
 H 8.08796 6.53195 1.32557
 C 11.28032 4.72864 5.41310
 H 12.11457 5.28599 5.84553
 C 11.01289 4.81537 4.04627
 H 11.64188 5.44760 3.42374

| | | | | | | | |
|---|----------------|----------|----------|------------------------------------|----------------|----------|----------|
| C | 11.04523 | -1.53222 | 3.77254 | H | 11.38096 | -1.87214 | 1.12416 |
| H | 11.67939 | -2.10773 | 3.10215 | H | 10.87347 | 4.15623 | 7.15592 |
| N | 9.21182 | -0.03879 | 4.11531 | H | 11.55368 | -2.78057 | 5.59029 |
| N | 7.22409 | -0.01744 | 2.04902 | H | 9.43888 | 2.41519 | 6.12508 |
| N | 9.55959 | 1.72731 | 1.82613 | H | 5.60650 | -2.67451 | 0.14345 |
| N | 10.77323 | 1.84987 | -0.70111 | H | 11.44862 | -0.21636 | -0.37259 |
| N | 9.18795 | 3.28969 | 4.27143 | H | 10.26192 | -1.22654 | 7.11840 |
| N | 7.19836 | 3.39978 | 2.21439 | H | 11.14527 | 5.45514 | 1.21851 |
| Co | 8.08203 | 1.65229 | 3.16973 | H | 9.68834 | 6.36159 | 1.66198 |
| C | 5.85767 | 1.51640 | 5.42558 | H | 9.74182 | 5.44742 | 0.14308 |
| N | 6.64321 | 1.57137 | 4.57480 | H | 4.10781 | 4.39066 | 2.32125 |
| C | 4.86855 | 1.44628 | 6.49357 | H | 5.28574 | 6.10085 | 0.86987 |
| H | 5.07924 | 2.20790 | 7.25999 | H | 7.74311 | 5.99645 | 0.67211 |
| H | 3.85984 | 1.62526 | 6.09054 | H | 11.76703 | 6.00439 | 5.67031 |
| H | 4.88712 | 0.45218 | 6.96605 | H | 11.14342 | 6.00990 | 3.28302 |
| | | | | H | 11.27464 | -2.54536 | 3.14985 |
| 1'-Cote⁻ (S = 0, 5-coordinates) | | | | 1'-Cote⁻ (S = 1) | | | |
| Energy | = -2791.618195 | | | Energy | = -2924.316085 | | |
| Enthalpy 0K | = -2791.147074 | | | Enthalpy 0K | = -2923.799349 | | |
| Free Energy 298K | = -2791.211067 | | | Free Energy 298K | = -2923.876237 | | |
| N | 7.28874 | 3.26612 | 2.60774 | C | 7.41994 | -1.89738 | 0.79875 |
| C | 7.94404 | 4.22102 | 1.90955 | H | 8.02731 | -2.61782 | 0.25743 |
| C | 7.22966 | 5.24102 | 1.26116 | C | 8.00857 | -0.89275 | 1.58666 |
| C | 5.84115 | 5.30689 | 1.37412 | C | 10.92878 | 2.94575 | 0.26740 |
| C | 5.18723 | 4.36134 | 2.16245 | H | 11.30524 | 3.88308 | -0.14072 |
| C | 5.95539 | 3.36639 | 2.75684 | C | 7.95152 | 4.32535 | 1.82062 |
| C | 9.49902 | 4.19313 | 1.92058 | C | 10.04098 | 0.59088 | 1.22686 |
| C | 10.04877 | 5.43580 | 1.19661 | C | 9.61505 | -0.20744 | 5.43775 |
| Co | 8.37953 | 1.65188 | 3.26918 | H | 9.12032 | 0.50602 | 6.09908 |
| N | 9.50978 | 3.19917 | 4.20839 | C | 5.90551 | -0.10587 | 2.16145 |
| C | 9.95227 | 4.19579 | 3.40894 | H | 5.33237 | 0.62065 | 2.74011 |
| C | 10.77452 | 5.21176 | 3.92185 | C | 5.86538 | 3.45022 | 2.32620 |
| C | 11.12860 | 5.21343 | 5.27026 | H | 5.30782 | 2.66409 | 2.83847 |
| C | 10.64313 | 4.19695 | 6.08987 | C | 5.24943 | -1.07585 | 1.40976 |
| C | 9.84052 | 3.21968 | 5.51025 | H | 4.15872 | -1.11081 | 1.38289 |
| C | 10.04881 | 2.92221 | 1.22693 | C | 10.22956 | -1.91057 | 0.95057 |
| C | 10.93498 | 2.95319 | 0.13853 | H | 9.98715 | -1.84260 | -0.11671 |
| N | 11.41683 | 1.83988 | -0.40971 | H | 9.90220 | -2.89579 | 1.30508 |
| C | 11.02809 | 0.67269 | 0.09899 | H | 11.32128 | -1.86740 | 1.04492 |
| C | 10.13084 | 0.59340 | 1.17578 | C | 9.85895 | 4.25731 | 3.48671 |
| N | 9.66797 | 1.73177 | 1.71299 | C | 10.37789 | 4.29255 | 6.20833 |
| C | 9.62027 | -0.74305 | 1.76917 | H | 10.53991 | 4.26875 | 7.28757 |
| C | 9.92800 | -0.84516 | 3.28996 | C | 9.91270 | -0.93633 | 3.25769 |
| C | 10.75435 | -1.85467 | 3.80856 | C | 9.49676 | 4.23179 | 1.97456 |
| C | 10.91095 | -1.99471 | 5.18724 | C | 10.01482 | 2.92073 | 1.32888 |
| C | 10.21109 | -1.13473 | 6.03202 | C | 11.02444 | -2.09535 | 5.07608 |
| C | 9.42857 | -0.14407 | 5.44911 | H | 11.68079 | -2.89308 | 5.43162 |
| N | 9.31635 | 0.03019 | 4.11967 | C | 9.57776 | 3.33391 | 5.59403 |
| C | 8.07505 | -0.81547 | 1.60256 | H | 9.09572 | 2.55739 | 6.19055 |
| C | 7.46841 | -1.82314 | 0.83557 | C | 6.03170 | -1.98756 | 0.70367 |
| C | 6.07929 | -1.89015 | 0.73885 | H | 5.57318 | -2.76754 | 0.09134 |
| C | 5.31565 | -0.94720 | 1.42318 | C | 10.95403 | 0.67935 | 0.16804 |
| C | 5.98586 | 0.02625 | 2.15697 | H | 11.35052 | -0.21007 | -0.32039 |
| N | 7.32402 | 0.10990 | 2.24117 | C | 9.55123 | -0.78307 | 1.75301 |
| C | 10.28907 | -1.91077 | 1.02177 | C | 10.43125 | -1.20270 | 5.96659 |
| H | 8.06611 | -2.56429 | 0.31126 | H | 10.59286 | -1.27132 | 7.04397 |
| H | 11.27140 | 3.88911 | -0.30919 | C | 10.14990 | 5.43981 | 1.27543 |
| H | 8.84486 | 0.53358 | 6.07470 | H | 11.24246 | 5.41023 | 1.36451 |
| H | 5.41773 | 0.77351 | 2.70959 | H | 9.80400 | 6.38257 | 1.71702 |
| H | 5.48532 | 2.62198 | 3.40199 | H | 9.90571 | 5.46241 | 0.20662 |
| H | 4.22462 | -0.96023 | 1.39731 | C | 5.18857 | 4.47162 | 1.66658 |
| H | 10.05899 | -1.87948 | -0.05025 | | | | |
| H | 9.94890 | -2.87897 | 1.40951 | | | | |

| | | | | | | | |
|----|----------|----------|----------|---|----------|----------|----------|
| H | 4.09731 | 4.48843 | 1.64534 | C | 10.31095 | -1.91054 | 1.00640 |
| C | 5.95115 | 5.45812 | 1.04461 | H | 8.11685 | -2.61985 | 0.25956 |
| H | 5.47604 | 6.28177 | 0.50661 | H | 11.32134 | 3.89141 | -0.23570 |
| C | 7.34117 | 5.38594 | 1.12835 | H | 8.79370 | 0.50396 | 6.05859 |
| H | 7.93276 | 6.16437 | 0.65387 | H | 5.36236 | 0.72694 | 2.52755 |
| C | 10.95452 | 5.26999 | 5.39987 | H | 5.39716 | 2.64962 | 3.18794 |
| H | 11.59807 | 6.04422 | 5.82411 | H | 4.22645 | -1.03993 | 1.19930 |
| C | 10.68673 | 5.25435 | 4.03146 | H | 10.10859 | -1.86981 | -0.07068 |
| H | 11.12621 | 6.02663 | 3.40577 | H | 9.96988 | -2.88627 | 1.37348 |
| C | 10.75712 | -1.96353 | 3.71387 | H | 11.39895 | -1.86272 | 1.13980 |
| H | 11.21020 | -2.66946 | 3.02313 | H | 10.72466 | 4.23112 | 7.19040 |
| N | 9.37398 | -0.06108 | 4.12705 | H | 11.43842 | -2.86280 | 5.58982 |
| N | 7.23991 | -0.00062 | 2.23889 | H | 9.34366 | 2.45937 | 6.12775 |
| N | 9.57414 | 1.72788 | 1.80125 | H | 5.66621 | -2.75473 | 0.01608 |
| N | 11.39560 | 1.84076 | -0.31044 | H | 11.47108 | -0.21074 | -0.32556 |
| N | 9.33728 | 3.29903 | 4.27548 | H | 10.15791 | -1.29418 | 7.11101 |
| N | 7.20172 | 3.36365 | 2.39104 | H | 11.17219 | 5.45250 | 1.25338 |
| Co | 8.22086 | 1.64599 | 3.33171 | H | 9.71133 | 6.37587 | 1.64125 |
| C | 6.06802 | 1.50428 | 5.76731 | H | 9.80249 | 5.44731 | 0.13346 |
| N | 6.83430 | 1.55657 | 4.89736 | H | 4.08421 | 4.45968 | 2.09211 |
| C | 5.10524 | 1.43856 | 6.85961 | H | 5.34193 | 6.19047 | 0.73686 |
| H | 5.62458 | 1.29145 | 7.81923 | H | 7.80234 | 6.06404 | 0.63752 |
| H | 4.52558 | 2.37286 | 6.91442 | H | 11.63260 | 6.08060 | 5.71753 |
| H | 4.41025 | 0.59869 | 6.70610 | H | 11.07950 | 6.06488 | 3.31432 |
| | | | | H | 11.20624 | -2.60417 | 3.14865 |

1'-Cote⁻ (S = 1, 5-coordinates)

| | |
|------------------|----------------|
| Energy | = -2791.640328 |
| Enthalpy 0K | = -2791.169446 |
| Free Energy 298K | = -2791.233758 |

| | | | |
|----|----------|----------|----------|
| N | 7.24128 | 3.30918 | 2.50589 |
| C | 7.93621 | 4.26491 | 1.85201 |
| C | 7.25928 | 5.30350 | 1.19231 |
| C | 5.86750 | 5.38146 | 1.24928 |
| C | 5.17002 | 4.42653 | 1.98653 |
| C | 5.90421 | 3.41245 | 2.59261 |
| C | 9.49205 | 4.21292 | 1.91552 |
| C | 10.07705 | 5.44260 | 1.19574 |
| Co | 8.33955 | 1.65169 | 3.20772 |
| N | 9.45253 | 3.24420 | 4.21272 |
| C | 9.90364 | 4.23736 | 3.41808 |
| C | 10.70027 | 5.26615 | 3.94630 |
| C | 11.01480 | 5.27961 | 5.30506 |
| C | 10.52029 | 4.26277 | 6.11874 |
| C | 9.74700 | 3.27166 | 5.52177 |
| C | 10.04689 | 2.92818 | 1.25120 |
| C | 10.96517 | 2.95445 | 0.19312 |
| N | 11.45903 | 1.84420 | -0.35293 |
| C | 11.04490 | 0.67805 | 0.14043 |
| C | 10.12081 | 0.59101 | 1.19010 |
| N | 9.63801 | 1.73230 | 1.72932 |
| C | 9.60811 | -0.75944 | 1.74974 |
| C | 9.87955 | -0.88596 | 3.27845 |
| C | 10.68322 | -1.91122 | 3.80258 |
| C | 10.81405 | -2.06404 | 5.18308 |
| C | 10.11851 | -1.19799 | 6.02454 |
| C | 9.36216 | -0.19017 | 5.43538 |
| N | 9.26720 | -0.01238 | 4.10662 |
| C | 8.06814 | -0.85713 | 1.53401 |
| C | 7.49713 | -1.87865 | 0.75762 |
| C | 6.11177 | -1.95943 | 0.61805 |
| C | 5.31568 | -1.01673 | 1.26467 |
| C | 5.95362 | -0.02826 | 2.00808 |
| N | 7.28645 | 0.06377 | 2.13581 |

2'-Cote⁻ (S = 0)

| | |
|------------------|----------------|
| Energy | = -2924.285843 |
| Enthalpy 0K | = -2923.769738 |
| Free Energy 298K | = -2923.847193 |

| | | | |
|---|---------|----------|----------|
| C | 8.98104 | 3.18256 | 6.49856 |
| H | 9.62709 | 2.30401 | 6.55638 |
| C | 9.14962 | 4.10823 | 5.47196 |
| H | 9.91693 | 3.95549 | 4.71069 |
| C | 8.31366 | 5.22327 | 5.45757 |
| H | 8.40662 | 5.98614 | 4.68101 |
| C | 7.34677 | 5.35188 | 6.45473 |
| H | 6.69490 | 6.22222 | 6.44739 |
| C | 7.23087 | 4.36444 | 7.44656 |
| C | 6.16674 | 4.45390 | 8.57343 |
| C | 5.17095 | 5.59159 | 8.26539 |
| H | 4.33593 | 5.59479 | 8.97731 |
| H | 4.74782 | 5.49177 | 7.25835 |
| H | 5.66156 | 6.57168 | 8.32175 |
| C | 6.88031 | 4.75010 | 9.91035 |
| C | 6.43119 | 5.73629 | 10.80737 |
| H | 5.51052 | 6.29251 | 10.62335 |
| C | 8.23211 | 5.41208 | 12.15079 |
| H | 8.81021 | 5.69553 | 13.03710 |
| C | 8.68176 | 4.40824 | 11.29780 |
| H | 9.62034 | 3.88518 | 11.48955 |
| C | 5.41113 | 3.10096 | 8.65923 |
| C | 4.11392 | 2.97495 | 8.14670 |
| H | 3.60844 | 3.81252 | 7.67335 |
| C | 3.46048 | 1.75142 | 8.24433 |
| H | 2.45157 | 1.62826 | 7.84428 |
| C | 4.10387 | 0.69063 | 8.87316 |
| H | 3.58708 | -0.26031 | 8.96842 |
| C | 5.40285 | 0.86866 | 9.36620 |
| C | 8.65667 | 1.27448 | 12.31740 |
| H | 9.57348 | 1.84072 | 12.14156 |
| C | 8.28554 | 0.91741 | 13.61099 |
| H | 8.90374 | 1.20470 | 14.46363 |

| | | | | | | | |
|----|----------|----------|----------|---|----------|----------|----------|
| C | 6.38915 | -0.18615 | 12.63529 | C | 6.77072 | 0.09821 | 11.32947 |
| H | 5.48334 | -0.77522 | 12.75761 | C | 7.17513 | 4.16686 | 7.44859 |
| C | 6.84817 | 0.18731 | 11.36147 | C | 7.19797 | 4.93063 | 6.27062 |
| C | 6.15140 | -0.29924 | 10.06248 | C | 8.15406 | 4.67641 | 5.28875 |
| C | 5.15156 | -1.42004 | 10.41315 | C | 9.08060 | 3.65879 | 5.50588 |
| H | 4.73704 | -1.88613 | 9.51110 | C | 8.99792 | 2.94466 | 6.69668 |
| H | 4.31106 | -1.03223 | 11.00246 | N | 8.07539 | 3.17694 | 7.64503 |
| H | 5.63480 | -2.21088 | 11.00054 | C | 5.22306 | 5.61155 | 8.18031 |
| C | 7.23098 | -0.84881 | 9.08979 | C | 5.20858 | -1.51249 | 10.15164 |
| C | 7.34028 | -2.21971 | 8.80330 | H | 9.71115 | 2.14655 | 6.89802 |
| H | 6.67577 | -2.94590 | 9.26512 | H | 9.85564 | 3.41918 | 4.77580 |
| C | 8.31524 | -2.66918 | 7.91312 | H | 8.17283 | 5.27021 | 4.37208 |
| H | 8.40187 | -3.73405 | 7.68474 | H | 6.47779 | 5.72799 | 6.10653 |
| C | 9.16817 | -1.73703 | 7.32502 | H | 4.46480 | 5.79663 | 8.95136 |
| H | 9.94361 | -2.03368 | 6.61632 | H | 4.69426 | 5.40014 | 7.24285 |
| C | 9.00693 | -0.39890 | 7.67305 | H | 5.79242 | 6.53907 | 8.04159 |
| H | 9.66817 | 0.36140 | 7.25296 | H | 5.86417 | 6.65967 | 10.30771 |
| N | 8.05596 | 3.30167 | 7.46024 | H | 9.13512 | 5.96340 | 12.73155 |
| N | 7.99178 | 4.04526 | 10.20439 | H | 9.61858 | 3.84200 | 11.48638 |
| N | 6.04027 | 2.05011 | 9.22579 | H | 3.36986 | 4.00188 | 8.19215 |
| N | 7.94527 | 0.95426 | 11.22776 | H | 2.15149 | 1.87576 | 8.53350 |
| N | 8.07229 | 0.03874 | 8.52822 | H | 3.36500 | -0.12429 | 9.33085 |
| Co | 8.14762 | 2.08390 | 9.33541 | H | 9.02674 | 2.20578 | 12.46370 |
| C | 7.10621 | 0.19193 | 13.77102 | H | 8.56293 | 1.05802 | 14.61063 |
| H | 6.75330 | -0.09597 | 14.76405 | H | 5.72188 | -1.39523 | 12.50968 |
| N | 7.08846 | 6.05844 | 11.92173 | H | 4.74857 | -1.80745 | 9.20029 |
| C | 11.33819 | 2.14563 | 9.45810 | H | 4.39586 | -1.29762 | 10.85617 |
| N | 10.18025 | 2.10070 | 9.38886 | H | 5.76614 | -2.37557 | 10.53598 |
| C | 12.79248 | 2.20292 | 9.54290 | H | 6.56173 | -2.63492 | 8.47341 |
| H | 13.15974 | 1.50100 | 10.30743 | H | 8.53636 | -3.21752 | 7.11578 |
| H | 13.24316 | 1.93318 | 8.57527 | H | 10.43959 | -1.55249 | 6.95635 |
| H | 13.12071 | 3.21940 | 9.80962 | H | 10.17410 | 0.64394 | 8.10122 |
| | | | | H | 6.83335 | -0.79451 | 14.62853 |

2'-Cote- (S = 0, 5-coordinates)

Energy = -2791.617500
 Enthalpy 0K = -2791.146291
 Free Energy 298K = -2791.209647

| | | | |
|----|---------|----------|----------|
| N | 7.43602 | 6.39814 | 11.60474 |
| C | 6.69339 | 6.00180 | 10.57111 |
| C | 6.93928 | 4.82095 | 9.84907 |
| N | 7.94863 | 4.00541 | 10.24310 |
| C | 8.74104 | 4.44355 | 11.24368 |
| C | 8.47921 | 5.62900 | 11.92074 |
| C | 6.14412 | 4.44267 | 8.57654 |
| C | 5.28771 | 3.16388 | 8.80095 |
| C | 3.91156 | 3.12648 | 8.54117 |
| C | 3.22396 | 1.93055 | 8.73384 |
| C | 3.90974 | 0.80459 | 9.18395 |
| C | 5.28434 | 0.90490 | 9.43658 |
| N | 5.92440 | 2.06680 | 9.23711 |
| C | 6.12586 | -0.28982 | 9.96811 |
| C | 7.28271 | -0.65144 | 8.99334 |
| C | 7.36022 | -1.90538 | 8.36673 |
| C | 8.47756 | -2.24162 | 7.60290 |
| C | 9.52234 | -1.32405 | 7.50189 |
| C | 9.36959 | -0.09325 | 8.12990 |
| N | 8.26833 | 0.25715 | 8.81988 |
| Co | 7.95930 | 2.14471 | 9.51617 |
| N | 7.67237 | 1.10671 | 11.34458 |
| C | 8.28644 | 1.40818 | 12.50096 |
| C | 8.02846 | 0.76308 | 13.70593 |
| C | 7.08015 | -0.25704 | 13.71015 |
| C | 6.45387 | -0.59195 | 12.51055 |

2'-Cote- (S = 1)

Energy = -2924.313407
 Enthalpy 0K = -2923.796819
 Free Energy 298K = -2923.873539

| | | | |
|---|---------|---------|----------|
| C | 9.07344 | 3.15985 | 6.60761 |
| H | 9.80359 | 2.36670 | 6.77749 |
| C | 9.15739 | 3.95409 | 5.46824 |
| H | 9.94280 | 3.77980 | 4.73046 |
| C | 8.21412 | 4.96790 | 5.31167 |
| H | 8.23126 | 5.62361 | 4.43804 |
| C | 7.24296 | 5.13892 | 6.29759 |
| H | 6.51443 | 5.93664 | 6.17925 |
| C | 7.22094 | 4.29124 | 7.41820 |
| C | 6.16745 | 4.45777 | 8.54991 |
| C | 5.18974 | 5.59072 | 8.17739 |
| H | 4.39657 | 5.69657 | 8.92796 |
| H | 4.70875 | 5.40166 | 7.21044 |
| H | 5.71086 | 6.55323 | 8.10351 |
| C | 6.89891 | 4.83435 | 9.86219 |
| C | 6.56755 | 5.98561 | 10.60287 |
| H | 5.73404 | 6.62590 | 10.31308 |
| C | 8.26950 | 5.61679 | 12.05715 |
| H | 8.84984 | 5.93370 | 12.93048 |
| C | 8.60517 | 4.46048 | 11.36060 |
| H | 9.45629 | 3.85096 | 11.66630 |
| C | 5.38132 | 3.12704 | 8.72377 |
| C | 4.03078 | 3.04634 | 8.35801 |
| H | 3.49976 | 3.90076 | 7.94857 |
| C | 3.35307 | 1.84347 | 8.51862 |

| | | | | | | | |
|----|----------|----------|----------|----|----------|----------|----------|
| H | 2.30264 | 1.75703 | 8.23186 | C | 9.36227 | -0.13762 | 8.07836 |
| C | 4.02746 | 0.75448 | 9.05932 | N | 8.27431 | 0.21493 | 8.78553 |
| H | 3.49260 | -0.18086 | 9.19444 | Co | 7.95971 | 2.14530 | 9.50812 |
| C | 5.37677 | 0.88709 | 9.41403 | N | 7.65889 | 1.04107 | 11.38296 |
| C | 8.62963 | 1.22009 | 12.35484 | C | 8.26100 | 1.32719 | 12.54731 |
| H | 9.50840 | 1.85105 | 12.20929 | C | 8.01431 | 0.64096 | 13.73275 |
| C | 8.31268 | 0.73871 | 13.62176 | C | 7.08861 | -0.39942 | 13.70057 |
| H | 8.93202 | 0.99658 | 14.48290 | C | 6.46805 | -0.71039 | 12.49061 |
| C | 6.46535 | -0.39848 | 12.59445 | C | 6.77541 | 0.02322 | 11.33293 |
| H | 5.60160 | -1.05152 | 12.68844 | C | 7.14747 | 4.18239 | 7.40326 |
| C | 6.86501 | 0.10759 | 11.34567 | C | 7.11375 | 4.92468 | 6.21147 |
| C | 6.15348 | -0.31275 | 10.02772 | C | 8.04651 | 4.67301 | 5.20556 |
| C | 5.16792 | -1.45799 | 10.33534 | C | 9.00455 | 3.68253 | 5.41062 |
| H | 4.70782 | -1.84829 | 9.41969 | C | 8.97136 | 2.98580 | 6.61495 |
| H | 4.35887 | -1.12066 | 10.99473 | N | 8.07145 | 3.21625 | 7.58310 |
| H | 5.67731 | -2.29306 | 10.83200 | C | 5.20360 | 5.61036 | 8.17772 |
| C | 7.22215 | -0.81425 | 9.01491 | C | 5.19620 | -1.53026 | 10.11020 |
| C | 7.26695 | -2.15465 | 8.59380 | H | 9.70401 | 2.20139 | 6.80905 |
| H | 6.54922 | -2.88477 | 8.95829 | H | 9.76220 | 3.44954 | 4.66032 |
| C | 8.24739 | -2.57234 | 7.69444 | H | 8.02250 | 5.24966 | 4.27803 |
| H | 8.28226 | -3.61333 | 7.36472 | H | 6.37126 | 5.70230 | 6.05248 |
| C | 9.17596 | -1.64134 | 7.23377 | H | 4.48156 | 5.81822 | 8.97734 |
| H | 9.96656 | -1.91490 | 6.53259 | H | 4.63537 | 5.37156 | 7.27067 |
| C | 9.06982 | -0.33490 | 7.70169 | H | 5.76325 | 6.53358 | 7.98458 |
| H | 9.78843 | 0.42174 | 7.38206 | H | 6.01388 | 6.80214 | 10.14320 |
| N | 8.13387 | 3.31041 | 7.55318 | H | 9.26967 | 6.09855 | 12.58332 |
| N | 7.90901 | 4.04712 | 10.28850 | H | 9.58574 | 3.84385 | 11.52925 |
| N | 6.02889 | 2.05306 | 9.22203 | H | 3.37257 | 4.03052 | 8.31765 |
| N | 7.91324 | 0.94682 | 11.25488 | H | 2.14630 | 1.91828 | 8.69701 |
| N | 8.12360 | 0.07494 | 8.55830 | H | 3.37310 | -0.09802 | 9.42226 |
| Co | 8.15478 | 2.12257 | 9.40029 | H | 8.97774 | 2.14946 | 12.53536 |
| C | 7.18561 | -0.07157 | 13.74361 | H | 8.53584 | 0.92053 | 14.64987 |
| H | 6.87570 | -0.46292 | 14.71536 | H | 5.75118 | -1.52695 | 12.46676 |
| N | 7.23267 | 6.37071 | 11.69169 | H | 4.72479 | -1.79247 | 9.15473 |
| C | 11.37592 | 2.21607 | 9.64282 | H | 4.39439 | -1.32445 | 10.82934 |
| N | 10.22096 | 2.15889 | 9.53643 | H | 5.74337 | -2.41125 | 10.46707 |
| C | 12.82574 | 2.29011 | 9.77588 | H | 6.55000 | -2.66745 | 8.45858 |
| H | 13.17788 | 1.57283 | 10.53325 | H | 8.50768 | -3.26188 | 7.08645 |
| H | 13.31129 | 2.05229 | 8.81674 | H | 10.41261 | -1.60336 | 6.89204 |
| H | 13.13096 | 3.30333 | 10.07989 | H | 10.16194 | 0.60472 | 8.02051 |
| | | | | H | 6.85150 | -0.96989 | 14.60154 |

2'-Cote⁻ (S = 1, 5-coordinates)

Energy = -2791.638231
 Enthalpy 0K = -2791.167411
 Free Energy 298K = -2791.231507

| | | | |
|---|---------|----------|----------|
| N | 7.59239 | 6.54930 | 11.43105 |
| C | 6.80978 | 6.11917 | 10.44138 |
| C | 6.97705 | 4.87382 | 9.80782 |
| N | 7.94210 | 4.03665 | 10.25604 |
| C | 8.76414 | 4.49386 | 11.22068 |
| C | 8.58993 | 5.74591 | 11.80037 |
| C | 6.14680 | 4.45542 | 8.56576 |
| C | 5.30704 | 3.17293 | 8.83061 |
| C | 3.92036 | 3.14771 | 8.63532 |
| C | 3.22706 | 1.95998 | 8.84949 |
| C | 3.92161 | 0.82538 | 9.25972 |
| C | 5.30644 | 0.90395 | 9.45546 |
| N | 5.95760 | 2.06369 | 9.23509 |
| C | 6.13284 | -0.31750 | 9.95414 |
| C | 7.28535 | -0.68568 | 8.97183 |
| C | 7.35044 | -1.94120 | 8.34588 |
| C | 8.45939 | -2.28459 | 7.57206 |
| C | 9.50599 | -1.37186 | 7.45389 |

3'-Cote⁻ (S = 0)

Energy = -2940.300145
 Enthalpy 0K = -2939.796030
 Free Energy 298K = -2939.873759

| | | | |
|---|---------|----------|---------|
| C | 3.98200 | 7.84154 | 3.37970 |
| H | 3.46370 | 6.90687 | 3.15671 |
| C | 3.29797 | 8.89537 | 3.97912 |
| H | 2.24543 | 8.78911 | 4.24819 |
| C | 4.00049 | 10.07574 | 4.21335 |
| H | 3.51339 | 10.93903 | 4.67273 |
| C | 5.34582 | 10.14034 | 3.85135 |
| H | 5.89296 | 11.06231 | 4.03129 |
| C | 5.96919 | 9.02634 | 3.26476 |
| C | 7.46898 | 9.04794 | 2.85728 |
| C | 8.13016 | 10.33560 | 3.38954 |
| H | 7.96914 | 10.45432 | 4.46781 |
| H | 7.71811 | 11.22402 | 2.89466 |
| H | 9.21345 | 10.32800 | 3.21534 |
| C | 7.57113 | 9.02632 | 1.30799 |
| C | 8.38368 | 9.93158 | 0.60551 |
| H | 9.02338 | 10.63862 | 1.12811 |

| | | | | | | | |
|----|----------|----------|----------|---|----------|----------|----------|
| C | 8.37112 | 9.93975 | -0.78964 | C | 10.13068 | 6.88959 | 4.73808 |
| H | 9.00187 | 10.64188 | -1.33984 | C | 9.38839 | 8.00259 | 4.34902 |
| C | 7.52571 | 9.05748 | -1.45985 | C | 8.32146 | 7.82762 | 3.45750 |
| H | 7.44993 | 9.04916 | -2.54882 | N | 8.02032 | 6.60384 | 2.99878 |
| C | 6.77316 | 8.17250 | -0.69226 | C | 7.45930 | 9.01399 | 2.94124 |
| H | 6.09419 | 7.46446 | -1.17150 | C | 7.59379 | 9.10957 | 1.39432 |
| C | 8.17354 | 7.80805 | 3.47215 | C | 8.13822 | 10.24687 | 0.77680 |
| C | 9.04307 | 7.94780 | 4.56219 | C | 8.20717 | 10.32698 | -0.61336 |
| H | 9.24938 | 8.91971 | 5.00168 | C | 7.71729 | 9.26452 | -1.36873 |
| C | 9.65537 | 6.81926 | 5.09679 | C | 7.19809 | 8.16733 | -0.68935 |
| H | 10.32872 | 6.90780 | 5.95228 | N | 7.14549 | 8.07321 | 0.64963 |
| C | 9.40996 | 5.57890 | 4.51795 | C | 5.95094 | 8.81067 | 3.26492 |
| H | 9.89942 | 4.69639 | 4.92128 | C | 5.25041 | 9.68476 | 4.11104 |
| C | 8.53471 | 5.49556 | 3.42799 | C | 3.87113 | 9.55254 | 4.27147 |
| C | 7.24613 | 4.89234 | -0.73714 | C | 3.20341 | 8.56160 | 3.55345 |
| H | 6.37916 | 5.34130 | -1.22470 | C | 3.96283 | 7.71589 | 2.75275 |
| C | 8.28019 | 4.33109 | -1.48159 | N | 5.30128 | 7.80208 | 2.64210 |
| H | 8.24751 | 4.33177 | -2.57648 | C | 7.95196 | 10.31854 | 3.59323 |
| C | 9.33381 | 3.73890 | 0.44053 | C | 8.22821 | 4.16564 | 1.24215 |
| H | 10.19830 | 3.26188 | 0.90515 | C | 9.07142 | 3.34908 | 0.46715 |
| C | 8.28664 | 4.26940 | 1.21643 | N | 8.99763 | 3.27418 | -0.86072 |
| C | 8.23686 | 4.12927 | 2.75461 | C | 8.04313 | 3.99555 | -1.45027 |
| C | 9.27357 | 3.07952 | 3.20657 | C | 7.20376 | 4.82773 | -0.71791 |
| H | 9.15198 | 2.13949 | 2.65355 | N | 7.31982 | 4.95477 | 0.61951 |
| H | 9.17279 | 2.84965 | 4.27445 | H | 3.47856 | 6.94368 | 2.15179 |
| H | 10.29786 | 3.43524 | 3.03861 | H | 2.11945 | 8.44470 | 3.60365 |
| C | 6.81612 | 3.67000 | 3.15919 | H | 3.32724 | 10.23326 | 4.93018 |
| C | 6.55110 | 2.37962 | 3.65758 | H | 5.76462 | 10.48041 | 4.64373 |
| H | 7.34628 | 1.64259 | 3.77737 | H | 7.86970 | 10.26906 | 4.68619 |
| C | 4.34112 | 2.85729 | 3.86777 | H | 7.36693 | 11.18130 | 3.25130 |
| H | 3.33306 | 2.54517 | 4.16118 | H | 9.00455 | 10.51078 | 3.35231 |
| C | 4.57536 | 4.13467 | 3.36213 | H | 8.51007 | 11.08025 | 1.36683 |
| H | 3.75572 | 4.84423 | 3.23351 | H | 8.63044 | 11.21245 | -1.09297 |
| N | 5.27485 | 7.89762 | 3.03084 | H | 7.73052 | 9.27745 | -2.45992 |
| N | 6.81798 | 8.12957 | 0.64626 | H | 6.80104 | 7.32358 | -1.25145 |
| N | 7.91626 | 6.59290 | 2.94578 | H | 9.65323 | 8.98064 | 4.74210 |
| N | 7.26476 | 4.90097 | 0.60516 | H | 10.96940 | 7.00401 | 5.42854 |
| N | 9.34069 | 3.77918 | -0.89149 | H | 10.38282 | 4.76619 | 4.54995 |
| N | 5.80124 | 4.54015 | 3.00673 | H | 6.40272 | 5.38470 | -1.20620 |
| N | 5.33109 | 1.97740 | 4.00916 | H | 7.93237 | 3.90896 | -2.53629 |
| Co | 6.16709 | 6.32976 | 1.77036 | H | 9.83824 | 2.72887 | 0.93312 |
| C | 3.50203 | 5.93210 | 0.04069 | H | 8.88051 | 2.05301 | 2.88887 |
| N | 4.45634 | 6.08493 | 0.68356 | H | 9.17084 | 2.96328 | 4.38183 |
| C | 2.30102 | 5.73878 | -0.76278 | H | 10.21501 | 3.21580 | 2.97605 |
| H | 1.50818 | 6.42933 | -0.43637 | H | 7.21735 | 2.04558 | 4.36340 |
| H | 2.51510 | 5.93036 | -1.82555 | H | 3.23365 | 3.13349 | 4.34356 |
| H | 1.93467 | 4.70568 | -0.65992 | H | 3.77656 | 5.13734 | 2.95106 |

3'-Cote⁻ (S = 0, 5-coordinates)

Energy = -2807.631215
 Enthalpy 0K = -2807.171979
 Free Energy 298K = -2807.235426

N 5.21357 2.48932 4.43007
 C 6.45671 2.75631 4.03865
 C 6.79495 3.87831 3.25713
 N 5.82140 4.73849 2.87884
 C 4.56633 4.45717 3.26608
 C 4.26512 3.33960 4.03942
 C 8.24428 4.14911 2.78963
 C 9.17931 3.03080 3.28696
 Co 6.42949 6.37906 1.71084
 C 8.71475 5.51637 3.36383
 C 9.79880 5.63025 4.24346

3'-Cote⁻ (S = 1)

Energy = -2940.328356
 Enthalpy 0K = -2939.823426
 Free Energy 298K = -2939.899840

C 3.95198 7.77134 3.22852
 H 3.41222 6.92830 2.79417
 C 3.29006 8.67286 4.05667
 H 2.23231 8.53666 4.28932
 C 4.02391 9.74071 4.56799
 H 3.56115 10.47709 5.22914
 C 5.36618 9.86583 4.21118
 H 5.92990 10.70978 4.59932
 C 5.96244 8.91664 3.36230
 C 7.44281 9.05242 2.90034
 C 8.03863 10.34131 3.49929

| | | | | | | | |
|----|----------|----------|----------|----|----------|----------|----------|
| H | 8.00842 | 10.32409 | 4.59523 | C | 4.29865 | 3.27964 | 4.17892 |
| H | 7.47974 | 11.22381 | 3.16466 | C | 8.23562 | 4.13331 | 2.80874 |
| H | 9.08475 | 10.47562 | 3.19929 | C | 9.19322 | 3.03461 | 3.30939 |
| C | 7.48340 | 9.15074 | 1.34712 | Co | 6.39932 | 6.36873 | 1.75422 |
| C | 8.05585 | 10.25673 | 0.69439 | N | 7.30957 | 4.91729 | 0.62429 |
| H | 8.52476 | 11.06199 | 1.25330 | C | 7.15649 | 4.75525 | -0.70348 |
| C | 8.02416 | 10.34166 | -0.69727 | C | 7.91392 | 3.83597 | -1.42228 |
| H | 8.47079 | 11.19987 | -1.20474 | N | 8.83164 | 3.07591 | -0.82528 |
| C | 7.40199 | 9.32479 | -1.41792 | C | 8.94915 | 3.19926 | 0.49622 |
| H | 7.32937 | 9.35151 | -2.50674 | C | 8.17994 | 4.09821 | 1.25856 |
| C | 6.86813 | 8.25802 | -0.70108 | C | 8.71052 | 5.51676 | 3.33671 |
| H | 6.35955 | 7.44599 | -1.22326 | C | 9.83346 | 5.64528 | 4.16331 |
| C | 8.27176 | 7.82682 | 3.38547 | C | 10.18764 | 6.90763 | 4.63083 |
| C | 9.35556 | 8.00317 | 4.25744 | C | 9.42370 | 8.01307 | 4.26664 |
| H | 9.64371 | 8.98638 | 4.61643 | C | 8.31558 | 7.83402 | 3.42832 |
| C | 10.08357 | 6.89561 | 4.67623 | N | 7.99053 | 6.60169 | 2.98951 |
| H | 10.93075 | 7.01508 | 5.35530 | C | 7.44419 | 9.03427 | 2.95647 |
| C | 9.72028 | 5.63477 | 4.21868 | C | 7.53879 | 9.16557 | 1.40610 |
| H | 10.29050 | 4.77015 | 4.54476 | C | 8.04038 | 10.32592 | 0.79429 |
| C | 8.62812 | 5.51026 | 3.34934 | C | 8.07526 | 10.42408 | -0.59682 |
| C | 7.41527 | 4.80069 | -0.75098 | C | 7.60109 | 9.35820 | -1.35759 |
| H | 6.71481 | 5.43315 | -1.29770 | C | 7.12675 | 8.23852 | -0.68047 |
| C | 8.25788 | 3.92282 | -1.42515 | N | 7.10151 | 8.13287 | 0.65673 |
| H | 8.24205 | 3.85848 | -2.51843 | C | 5.94165 | 8.84426 | 3.32658 |
| C | 9.09301 | 3.21449 | 0.56246 | C | 5.27558 | 9.73089 | 4.18832 |
| H | 9.78771 | 2.55205 | 1.07943 | C | 3.90204 | 9.60611 | 4.39963 |
| C | 8.24393 | 4.08574 | 1.27323 | C | 3.20480 | 8.60718 | 3.72283 |
| C | 8.19483 | 4.11240 | 2.82182 | C | 3.93364 | 7.74831 | 2.90688 |
| C | 9.14200 | 3.02956 | 3.37668 | N | 5.26448 | 7.83482 | 2.73903 |
| H | 8.86058 | 2.03447 | 3.01096 | C | 7.96574 | 10.32573 | 3.61406 |
| H | 9.11566 | 2.99996 | 4.47259 | H | 3.42614 | 6.95703 | 2.34977 |
| H | 10.17962 | 3.21428 | 3.07339 | H | 2.12426 | 8.48967 | 3.82196 |
| C | 6.74536 | 3.80120 | 3.27315 | H | 3.38553 | 10.29663 | 5.07019 |
| C | 6.43178 | 2.69187 | 4.08318 | H | 5.81012 | 10.52997 | 4.69497 |
| H | 7.20893 | 2.02528 | 4.45770 | H | 7.92102 | 10.25947 | 4.70827 |
| C | 4.21561 | 3.16926 | 3.98627 | H | 7.37202 | 11.19514 | 3.30656 |
| H | 3.18415 | 2.92054 | 4.25812 | H | 9.00861 | 10.52087 | 3.33694 |
| C | 4.49508 | 4.27497 | 3.18987 | H | 8.40392 | 11.16255 | 1.38492 |
| H | 3.69116 | 4.90672 | 2.81019 | H | 8.46399 | 11.32686 | -1.07353 |
| N | 5.24877 | 7.87194 | 2.90336 | H | 7.59597 | 9.38547 | -2.44864 |
| N | 6.92405 | 8.15563 | 0.63438 | H | 6.74875 | 7.38420 | -1.24348 |
| N | 7.92983 | 6.59494 | 2.95430 | H | 9.70510 | 8.99410 | 4.63868 |
| N | 7.41888 | 4.90429 | 0.58807 | H | 11.05795 | 7.03006 | 5.27937 |
| N | 9.11073 | 3.13751 | -0.76716 | H | 10.43316 | 4.78586 | 4.44998 |
| N | 5.75072 | 4.60739 | 2.84722 | H | 6.40363 | 5.36995 | -1.20123 |
| N | 5.18808 | 2.38109 | 4.44474 | H | 7.77552 | 3.72030 | -2.50242 |
| Co | 6.23819 | 6.32328 | 1.67148 | H | 9.69078 | 2.55116 | 0.96433 |
| C | 3.66315 | 5.94799 | -0.28228 | H | 8.88059 | 2.04359 | 2.95832 |
| N | 4.57707 | 6.10192 | 0.41632 | H | 9.22469 | 3.00409 | 4.40527 |
| C | 2.51493 | 5.75313 | -1.15893 | H | 10.21582 | 3.20825 | 2.95198 |
| H | 1.71144 | 6.45928 | -0.89874 | H | 7.28508 | 2.05904 | 4.48262 |
| H | 2.80211 | 5.92118 | -2.20837 | H | 3.27823 | 3.05693 | 4.50807 |
| H | 2.12984 | 4.72644 | -1.06104 | H | 3.75409 | 5.04664 | 3.05999 |

3'-Cote- (S = 1, 5-coordinates)

Energy = -2807.651837
Enthalpy 0K = -2807.192936
Free Energy 298K = -2807.257052

N 5.27428 2.46420 4.57489
C 6.50495 2.74495 4.15103
C 6.80114 3.84948 3.32869
N 5.80328 4.67946 2.95688
C 4.56195 4.38468 3.37305

1'-Co+2e- (S = 1/2)

Energy = -2924.423764
Enthalpy 0K = -2923.912361
Free Energy 298K = -2923.991039

C 7.48209 -2.09861 1.01702
H 8.12831 -2.87337 0.61171
C 8.02683 -0.97410 1.66801
C 10.54967 2.95124 0.04317
H 10.81749 3.89885 -0.42812

| | | | | | | | |
|----|----------|----------|----------|------------------|----------------|----------|----------|
| C | 7.97641 | 4.40314 | 1.90219 | Free Energy 298K | = -2791.355745 | | |
| C | 9.92437 | 0.59273 | 1.16950 | N | 7.28342 | 3.29069 | 2.70080 |
| C | 9.51492 | -0.12372 | 5.47228 | C | 7.93831 | 4.21753 | 1.94513 |
| H | 8.85194 | 0.44965 | 6.12487 | C | 7.20952 | 5.17186 | 1.21904 |
| C | 5.88102 | -0.17039 | 2.04748 | C | 5.81714 | 5.22811 | 1.30794 |
| H | 5.27094 | 0.62034 | 2.49085 | C | 5.16632 | 4.32129 | 2.15905 |
| C | 5.84587 | 3.53163 | 2.21304 | C | 5.94106 | 3.38339 | 2.82203 |
| H | 5.25076 | 2.69503 | 2.58719 | C | 9.49093 | 4.19597 | 1.95213 |
| C | 5.27237 | -1.24834 | 1.41347 | C | 10.02391 | 5.43798 | 1.21469 |
| H | 4.18424 | -1.30726 | 1.34360 | Co | 8.39625 | 1.65493 | 3.36324 |
| C | 10.29462 | -1.90765 | 1.06655 | N | 9.58364 | 3.22892 | 4.26053 |
| H | 9.96495 | -1.98219 | 0.02402 | C | 9.97973 | 4.22334 | 3.43006 |
| H | 10.11442 | -2.87845 | 1.54927 | C | 10.80819 | 5.25210 | 3.91259 |
| H | 11.37589 | -1.72603 | 1.04855 | C | 11.21814 | 5.26730 | 5.24520 |
| C | 9.92578 | 4.11040 | 3.50943 | C | 10.78719 | 4.24388 | 6.08916 |
| C | 10.50733 | 3.97584 | 6.21902 | C | 9.97443 | 3.25560 | 5.54514 |
| H | 10.67800 | 3.90736 | 7.29567 | C | 10.02050 | 2.91955 | 1.25092 |
| C | 9.97319 | -0.78542 | 3.29617 | C | 10.81766 | 2.94310 | 0.11345 |
| C | 9.51239 | 4.21617 | 2.01904 | N | 11.30173 | 1.82961 | -0.47879 |
| C | 9.90280 | 2.92037 | 1.27040 | C | 10.92955 | 0.66100 | 0.08506 |
| C | 11.36975 | -1.57977 | 5.11777 | C | 10.11763 | 0.57864 | 1.20969 |
| H | 12.21696 | -2.17030 | 5.47573 | N | 9.67610 | 1.72231 | 1.81053 |
| C | 9.47680 | 3.25731 | 5.62032 | C | 9.62211 | -0.76039 | 1.80859 |
| H | 8.82327 | 2.61889 | 6.22013 | C | 9.90949 | -0.86350 | 3.33382 |
| C | 6.10315 | -2.24066 | 0.88472 | C | 10.74917 | -1.85773 | 3.86090 |
| H | 5.68637 | -3.11569 | 0.37953 | C | 10.90974 | -2.00321 | 5.24024 |
| C | 10.57042 | 0.67999 | -0.05550 | C | 10.19661 | -1.15176 | 6.08642 |
| H | 10.85496 | -0.21849 | -0.60640 | C | 9.40547 | -0.17081 | 5.50407 |
| C | 9.55904 | -0.77021 | 1.80280 | N | 9.28826 | 0.00993 | 4.17111 |
| C | 10.56044 | -0.87120 | 6.00630 | C | 8.08738 | -0.83214 | 1.57856 |
| H | 10.73341 | -0.89099 | 7.08458 | C | 7.50886 | -1.80252 | 0.74171 |
| C | 10.22702 | 5.42661 | 1.38582 | C | 6.12969 | -1.83204 | 0.54055 |
| H | 11.31151 | 5.26718 | 1.35447 | C | 5.33915 | -0.87274 | 1.18304 |
| H | 10.02903 | 6.34831 | 1.95085 | C | 5.97965 | 0.05898 | 1.99016 |
| H | 9.89696 | 5.58609 | 0.35294 | N | 7.30752 | 0.09203 | 2.19260 |
| C | 5.21685 | 4.65028 | 1.67718 | C | 10.31249 | -1.93032 | 1.08462 |
| H | 4.12762 | 4.69611 | 1.61501 | H | 8.12620 | -2.54556 | 0.24251 |
| C | 6.02857 | 5.69956 | 1.23613 | H | 11.10341 | 3.88385 | -0.36085 |
| H | 5.59508 | 6.60813 | 0.81052 | H | 8.82385 | 0.51064 | 6.13030 |
| C | 7.41021 | 5.57044 | 1.35321 | H | 5.39559 | 0.82578 | 2.50229 |
| H | 8.04161 | 6.38890 | 1.01651 | H | 5.47278 | 2.66475 | 3.50107 |
| C | 11.30479 | 4.77214 | 5.39641 | H | 4.25392 | -0.84609 | 1.06568 |
| H | 12.14039 | 5.34547 | 5.80590 | H | 10.10182 | -1.90469 | 0.00969 |
| C | 11.00546 | 4.84271 | 4.03504 | H | 9.97199 | -2.89962 | 1.47289 |
| H | 11.61423 | 5.47621 | 3.39452 | H | 11.40198 | -1.87251 | 1.20182 |
| C | 11.06761 | -1.53969 | 3.75580 | H | 11.07122 | 4.20525 | 7.14283 |
| H | 11.68563 | -2.10513 | 3.06259 | H | 11.56338 | -2.78166 | 5.64048 |
| N | 9.24405 | -0.05392 | 4.16124 | H | 9.62212 | 2.43462 | 6.17241 |
| N | 7.20706 | -0.02544 | 2.17651 | H | 5.68305 | -2.59241 | -0.10460 |
| N | 9.56204 | 1.72560 | 1.85663 | H | 11.31428 | -0.23575 | -0.40446 |
| N | 10.89969 | 1.84669 | -0.65199 | H | 10.24987 | -1.24034 | 7.17354 |
| N | 9.20968 | 3.29393 | 4.30701 | H | 11.12078 | 5.45438 | 1.21820 |
| N | 7.17477 | 3.39903 | 2.32668 | H | 9.66093 | 6.36613 | 1.67615 |
| Co | 8.03265 | 1.65132 | 3.33984 | H | 9.70502 | 5.43458 | 0.16589 |
| C | 5.83661 | 1.50204 | 5.55952 | H | 4.08439 | 4.34273 | 2.30539 |
| N | 6.66710 | 1.56564 | 4.74040 | H | 5.25698 | 5.97678 | 0.74354 |
| C | 4.79914 | 1.41877 | 6.57994 | H | 7.72103 | 5.89241 | 0.58505 |
| H | 5.23719 | 1.25193 | 7.57818 | H | 11.86128 | 6.07015 | 5.61399 |
| H | 4.21239 | 2.35112 | 6.61614 | H | 11.14061 | 6.05078 | 3.25485 |
| H | 4.10891 | 0.58590 | 6.36761 | H | 11.28025 | -2.53846 | 3.20049 |

1' -Co+2e- (S = 1/2, 5-coordinates)

Energy = -2791.754910
Enthalpy 0K = -2791.289685

1'-Co+2e⁻ (S = 3/2)

Energy = -2924.426051
 Enthalpy 0K = -2923.914045
 Free Energy 298K = -2923.992088

C 7.39307 -1.78633 0.71178
 H 8.00572 -2.47120 0.13158
 C 7.98258 -0.86217 1.59526
 C 10.90840 2.94318 0.31380
 H 11.28282 3.88748 -0.08442
 C 7.92912 4.29437 1.82464
 C 10.02017 0.58729 1.25683
 C 9.62186 -0.19091 5.47165
 H 9.09505 0.49276 6.14113
 C 5.87121 -0.08328 2.16949
 H 5.29627 0.60650 2.79147
 C 5.83335 3.42770 2.32690
 H 5.27213 2.67378 2.88358
 C 5.21890 -0.97864 1.33033
 H 4.12809 -0.99385 1.27773
 C 10.18387 -1.92519 0.99497
 H 9.95097 -1.85341 -0.07343
 H 9.84211 -2.90489 1.35708
 H 11.27527 -1.88686 1.08627
 C 9.85056 4.23365 3.50960
 C 10.45975 4.21654 6.22057
 H 10.64982 4.17676 7.29534
 C 9.90332 -0.91405 3.28257
 C 9.46936 4.22514 2.00540
 C 9.99591 2.91955 1.35947
 C 11.14686 -1.98091 5.08092
 H 11.86440 -2.73178 5.42109
 C 9.58375 3.31579 5.62639
 H 9.06976 2.56680 6.23302
 C 6.00736 -1.84713 0.57162
 H 5.55457 -2.57057 -0.11099
 C 10.93110 0.67480 0.21315
 H 11.32309 -0.22289 -0.26749
 C 9.52091 -0.78047 1.78458
 C 10.51599 -1.12315 5.98488
 H 10.70704 -1.17413 7.05900
 C 10.10840 5.44776 1.32004
 H 11.20047 5.42258 1.40761
 H 9.74854 6.38504 1.76690
 H 9.87455 5.46569 0.24960
 C 5.16298 4.38410 1.57350
 H 4.07196 4.38411 1.52322
 C 5.93369 5.33285 0.89754
 H 5.46633 6.10766 0.28467
 C 7.32060 5.28481 1.03045
 H 7.91893 6.03159 0.51509
 C 11.07388 5.16228 5.39634
 H 11.77691 5.89396 5.80223
 C 10.76112 5.16826 4.03771
 H 11.22886 5.91330 3.39929
 C 10.83263 -1.87356 3.72691
 H 11.31425 -2.54999 3.02562
 N 9.33291 -0.06773 4.16689
 N 7.20546 -0.00464 2.29156
 N 9.52782 1.72434 1.85722
 N 11.39934 1.83808 -0.28612
 N 9.29714 3.30152 4.31525
 N 7.16910 3.36290 2.44047
 Co 8.14019 1.64314 3.42633

C 6.06359 1.50206 5.78884
 N 6.83197 1.55802 4.91239
 C 5.10494 1.43038 6.88464
 H 5.61926 1.27934 7.84800
 H 4.51967 2.36176 6.95111
 H 4.40506 0.59212 6.73592

1'-Co+2e⁻ (S = 3/2, 5-coordinates)

Energy = -2791.757717
 Enthalpy 0K = -2791.291894
 Free Energy 298K = -2791.357768

N 7.20676 3.27949 2.59367
 C 7.88565 4.22379 1.89251
 C 7.17824 5.20144 1.17329
 C 5.78428 5.25401 1.22046
 C 5.10774 4.32398 2.01325
 C 5.86355 3.36460 2.67360
 C 9.43914 4.20529 1.93888
 C 9.98060 5.45324 1.21800
 Co 8.27467 1.64768 3.28260
 N 9.46019 3.23354 4.24257
 C 9.91352 4.21086 3.41920
 C 10.81281 5.17830 3.90117
 C 11.23125 5.15845 5.23090
 C 10.74069 4.15486 6.07054
 C 9.86456 3.22525 5.52456
 C 10.00423 2.93232 1.26528
 C 10.90981 2.95921 0.21406
 N 11.43768 1.84949 -0.34758
 C 11.01985 0.67939 0.18395
 C 10.11386 0.59090 1.23142
 N 9.59529 1.73214 1.78540
 C 9.59735 -0.75093 1.80303
 C 9.84510 -0.84767 3.33458
 C 10.70310 -1.81396 3.88566
 C 10.83581 -1.94578 5.26888
 C 10.08267 -1.10815 6.09432
 C 9.27810 -0.15318 5.48699
 N 9.18305 0.01093 4.15206
 C 8.07134 -0.82833 1.52054
 C 7.53219 -1.78261 0.63972
 C 6.16112 -1.82447 0.39133
 C 5.33768 -0.89483 1.03297
 C 5.93908 0.02525 1.88223
 N 7.26050 0.07526 2.12515
 C 10.31345 -1.91913 1.10205
 H 8.17501 -2.50562 0.14370
 H 11.24587 3.90388 -0.21848
 H 8.66433 0.51646 6.09524
 H 5.32665 0.76636 2.39883
 H 5.37531 2.62405 3.31269
 H 4.25592 -0.88187 0.88452
 H 10.14714 -1.88704 0.01942
 H 9.95752 -2.88952 1.47442
 H 11.39727 -1.86279 1.26067
 H 11.02486 4.09189 7.12301
 H 11.50015 -2.70458 5.68897
 H 9.46317 2.42619 6.15036
 H 5.74631 -2.57431 -0.28659
 H 11.44862 -0.21631 -0.27029
 H 10.11466 -1.18880 7.18287
 H 11.07604 5.47372 1.24071
 H 9.61069 6.37793 1.68196

H 9.68335 5.45260 0.16228
 H 4.02112 4.33799 2.12043
 H 5.24261 6.02207 0.66345
 H 7.70766 5.94074 0.57797
 H 11.92273 5.91859 5.60235
 H 11.19097 5.95947 3.24662
 H 11.26767 -2.48338 3.24176

2'-Co+2e⁻ (S = 1/2)

Energy = -2924.418746
 Enthalpy 0K = -2923.908671
 Free Energy 298K = -2923.986856

C 9.26288 3.01334 6.82118
 H 10.09468 2.40916 7.19001
 C 9.28924 3.52003 5.52667
 H 10.13467 3.31241 4.86756
 C 8.19216 4.28124 5.10829
 H 8.13787 4.68362 4.09360
 C 7.16830 4.53260 6.01764
 H 6.32693 5.14253 5.69806
 C 7.22340 4.00820 7.32478
 C 6.17787 4.39706 8.40454
 C 5.22191 5.45528 7.81822
 H 4.54860 5.85045 8.58714
 H 4.60794 5.03166 7.01163
 H 5.78079 6.30706 7.41310
 C 6.94896 4.99015 9.60467
 C 6.87056 6.34167 9.95798
 H 6.22262 7.02208 9.40069
 C 8.36886 6.05369 11.63800
 H 8.95335 6.46630 12.46884
 C 8.47467 4.71830 11.31212
 H 9.15280 4.06843 11.87233
 C 5.35147 3.18120 8.89868
 C 3.96369 3.27975 9.06428
 H 3.43261 4.20121 8.84346
 C 3.24685 2.17265 9.51726
 H 2.16300 2.22715 9.64401
 C 3.93288 0.99440 9.79903
 H 3.37544 0.12802 10.14531
 C 5.32381 0.95422 9.64366
 C 8.37534 1.16906 12.52662
 H 8.95806 2.09140 12.56642
 C 8.37986 0.28400 13.59867
 H 8.96050 0.51075 14.49555
 C 6.89473 -1.09950 12.31564
 H 6.30468 -2.00821 12.22216
 C 6.92401 -0.14908 11.28349
 C 6.12397 -0.33886 9.96701
 C 5.14500 -1.51913 10.12970
 H 4.49411 -1.61697 9.25145
 H 4.50251 -1.38683 11.00919
 H 5.68545 -2.46604 10.25204
 C 7.13224 -0.66081 8.83679
 C 6.96822 -1.75934 7.97843
 H 6.07357 -2.37659 8.02898
 C 7.96006 -2.08593 7.05208
 H 7.83318 -2.94038 6.38344
 C 9.12712 -1.30806 7.02616
 H 9.95656 -1.53836 6.35402
 C 9.20055 -0.21543 7.87807
 H 10.08851 0.42061 7.88548
 N 8.25743 3.20960 7.68570

N 7.76731 4.14065 10.30462
 N 6.00404 2.04229 9.21195
 N 7.67799 0.96550 11.39403
 N 8.22182 0.14159 8.73138
 Co 8.11937 2.12410 9.59094
 C 7.62385 -0.88498 13.48891
 H 7.59244 -1.61665 14.29978
 N 7.55102 6.90474 10.96844
 C 11.24827 2.41377 10.18353
 N 10.11904 2.24172 9.96823
 C 12.65767 2.65955 10.46222
 H 13.25737 2.57086 9.54257
 H 12.79296 3.67481 10.86765
 H 13.03939 1.93425 11.19779

2'-Co+2e⁻ (S = 1/2, 5-coordinates)

Energy = -2791.756947
 Enthalpy 0K = -2791.291447
 Free Energy 298K = -2791.356060

N 7.81458 6.40534 11.38747
 C 6.94870 5.97938 10.45031
 C 7.07059 4.78631 9.73731
 N 8.10946 3.92131 10.02424
 C 9.04329 4.40012 10.90373
 C 8.89329 5.59913 11.56307
 C 6.14738 4.42363 8.54339
 C 5.29637 3.14540 8.80830
 C 3.90449 3.13357 8.63548
 C 3.19665 1.95746 8.86586
 C 3.87706 0.81332 9.27543
 C 5.26607 0.87404 9.45126
 N 5.92717 2.02101 9.20609
 C 6.09144 -0.35317 9.94276
 C 7.20664 -0.75332 8.93625
 C 7.26596 -2.02391 8.35257
 C 8.35617 -2.40114 7.56062
 C 9.40759 -1.48010 7.40457
 C 9.28199 -0.23232 7.98642
 N 8.19373 0.16546 8.69906
 Co 7.94140 2.02594 9.38103
 N 7.65064 1.04491 11.28882
 C 8.30208 1.36000 12.42110
 C 8.15307 0.65803 13.61134
 C 7.27547 -0.43006 13.62519
 C 6.59777 -0.76038 12.45216
 C 6.79871 -0.00602 11.28358
 C 7.08349 4.18900 7.33026
 C 7.16025 5.09174 6.25595
 C 8.04669 4.85630 5.20535
 C 8.85280 3.71425 5.24753
 C 8.72772 2.87268 6.34707
 N 7.87813 3.09626 7.36172
 C 5.19741 5.59915 8.24562
 C 5.13803 -1.54252 10.15603
 H 9.33071 1.96380 6.41702
 H 9.55707 3.47661 4.44762
 H 8.10110 5.55312 4.36536
 H 6.53300 5.97985 6.22625
 H 4.50012 5.76083 9.07753
 H 4.60911 5.41839 7.33595
 H 5.75755 6.53018 8.10442
 H 6.10507 6.64839 10.26381
 H 9.66461 5.93749 12.26482

| | | | |
|---|----------|----------|----------|
| H | 9.93216 | 3.78173 | 11.06511 |
| H | 3.36595 | 4.02325 | 8.32083 |
| H | 2.11290 | 1.93224 | 8.72740 |
| H | 3.31808 | -0.10151 | 9.45223 |
| H | 8.95952 | 2.23064 | 12.36860 |
| H | 8.70372 | 0.96519 | 14.50261 |
| H | 5.91307 | -1.60507 | 12.46231 |
| H | 4.62904 | -1.81146 | 9.22161 |
| H | 4.36938 | -1.30715 | 10.90340 |
| H | 5.68253 | -2.42849 | 10.50342 |
| H | 6.46714 | -2.74495 | 8.51299 |
| H | 8.39254 | -3.39004 | 7.09970 |
| H | 10.30684 | -1.72724 | 6.83615 |
| H | 10.08114 | 0.50772 | 7.88727 |
| H | 7.11440 | -1.01248 | 14.53558 |

2'-Co+2e⁻ (S = 3/2, 5-coordinates)

| | |
|------------------|----------------|
| Energy | = -2791.755476 |
| Enthalpy 0K | = -2791.290020 |
| Free Energy 298K | = -2791.354878 |

| | | | |
|----|---------|----------|----------|
| N | 7.72456 | 6.56695 | 11.23977 |
| C | 6.92045 | 6.09865 | 10.27192 |
| C | 7.02833 | 4.83476 | 9.68546 |
| N | 7.98988 | 3.95116 | 10.13605 |
| C | 8.85667 | 4.45797 | 11.06424 |
| C | 8.72595 | 5.71952 | 11.59762 |
| C | 6.16093 | 4.39125 | 8.48231 |
| C | 5.30929 | 3.14672 | 8.85845 |
| C | 3.91159 | 3.20013 | 8.93132 |
| C | 3.19165 | 2.05637 | 9.27342 |
| C | 3.88282 | 0.87944 | 9.57100 |
| C | 5.27890 | 0.87891 | 9.51492 |
| N | 5.95526 | 1.99165 | 9.15363 |
| C | 6.11555 | -0.37147 | 9.91443 |
| C | 7.20379 | -0.72694 | 8.86289 |
| C | 7.18974 | -1.94205 | 8.16252 |
| C | 8.24793 | -2.29340 | 7.32114 |
| C | 9.33480 | -1.41955 | 7.22262 |
| C | 9.27551 | -0.22126 | 7.91869 |
| N | 8.22358 | 0.15563 | 8.68093 |
| Co | 7.96953 | 2.03132 | 9.39138 |
| N | 7.66600 | 1.00334 | 11.28777 |
| C | 8.28057 | 1.32236 | 12.43959 |
| C | 8.11849 | 0.60685 | 13.62068 |
| C | 7.26991 | -0.50255 | 13.60473 |
| C | 6.61976 | -0.83081 | 12.41498 |
| C | 6.82896 | -0.05826 | 11.26063 |
| C | 7.14662 | 4.10091 | 7.31673 |
| C | 7.14740 | 4.87070 | 6.13900 |
| C | 8.10194 | 4.65323 | 5.14744 |
| C | 9.07047 | 3.66637 | 5.35591 |
| C | 9.00751 | 2.94054 | 6.53863 |
| N | 8.07549 | 3.12699 | 7.48770 |
| C | 5.21350 | 5.53467 | 8.07308 |
| C | 5.17397 | -1.57636 | 10.09622 |
| H | 9.74130 | 2.15543 | 6.73147 |
| H | 9.85023 | 3.45643 | 4.62099 |
| H | 8.08896 | 5.24751 | 4.23056 |
| H | 6.40175 | 5.64578 | 5.98159 |
| H | 4.54325 | 5.80264 | 8.89795 |
| H | 4.59549 | 5.24944 | 7.21039 |
| H | 5.77621 | 6.43763 | 7.81036 |
| H | 6.13871 | 6.78994 | 9.94804 |

| | | | |
|---|----------|----------|----------|
| H | 9.44524 | 6.07690 | 12.34358 |
| H | 9.68905 | 3.81473 | 11.36684 |
| H | 3.37104 | 4.11711 | 8.71340 |
| H | 2.10001 | 2.07978 | 9.30690 |
| H | 3.32168 | -0.01141 | 9.84116 |
| H | 8.92236 | 2.20432 | 12.41248 |
| H | 8.64427 | 0.91677 | 14.52595 |
| H | 5.95317 | -1.68992 | 12.40064 |
| H | 4.61396 | -1.77959 | 9.17419 |
| H | 4.44350 | -1.39335 | 10.89304 |
| H | 5.73368 | -2.48220 | 10.36157 |
| H | 6.35856 | -2.63468 | 8.27022 |
| H | 8.22989 | -3.23954 | 6.77583 |
| H | 10.21171 | -1.65720 | 6.61679 |
| H | 10.10805 | 0.48480 | 7.87390 |
| H | 7.10965 | -1.10321 | 14.50330 |

2'-Co+2e⁻ (S = 3/2)

| | |
|------------------|----------------|
| Energy | = -2924.420980 |
| Enthalpy 0K | = -2923.909953 |
| Free Energy 298K | = -2923.989904 |

| | | | |
|---|----------|----------|----------|
| C | 9.28716 | 3.01420 | 6.77435 |
| H | 10.12520 | 2.41406 | 7.13602 |
| C | 9.29987 | 3.52289 | 5.48018 |
| H | 10.13973 | 3.31882 | 4.81264 |
| C | 8.20118 | 4.28608 | 5.07634 |
| H | 8.13679 | 4.69192 | 4.06355 |
| C | 7.18722 | 4.53683 | 5.99805 |
| H | 6.34475 | 5.15027 | 5.68900 |
| C | 7.25282 | 4.00654 | 7.30212 |
| C | 6.21018 | 4.38559 | 8.38909 |
| C | 5.25777 | 5.45069 | 7.80982 |
| H | 4.57242 | 5.83205 | 8.57379 |
| H | 4.65195 | 5.03513 | 6.99295 |
| H | 5.81917 | 6.31128 | 7.42503 |
| C | 6.95502 | 4.95255 | 9.61626 |
| C | 6.77176 | 6.26529 | 10.07480 |
| H | 6.09037 | 6.94162 | 9.55340 |
| C | 8.24454 | 5.94822 | 11.77176 |
| H | 8.77987 | 6.33608 | 12.64611 |
| C | 8.45387 | 4.65365 | 11.34436 |
| H | 9.16883 | 4.01222 | 11.86683 |
| C | 5.38468 | 3.17392 | 8.89027 |
| C | 4.01009 | 3.28962 | 9.11961 |
| H | 3.48243 | 4.21698 | 8.91364 |
| C | 3.29315 | 2.19715 | 9.60976 |
| H | 2.21512 | 2.26396 | 9.77166 |
| C | 3.98248 | 1.00906 | 9.87181 |
| H | 3.43258 | 0.15112 | 10.25050 |
| C | 5.35618 | 0.94676 | 9.63625 |
| C | 8.40718 | 1.17807 | 12.49067 |
| H | 9.06504 | 2.04930 | 12.48961 |
| C | 8.28869 | 0.38966 | 13.62906 |
| H | 8.84593 | 0.64122 | 14.53379 |
| C | 6.76086 | -0.97829 | 12.37949 |
| H | 6.10437 | -1.84394 | 12.32708 |
| C | 6.92295 | -0.12683 | 11.27185 |
| C | 6.15667 | -0.34836 | 9.94189 |
| C | 5.18300 | -1.53358 | 10.09795 |
| H | 4.54604 | -1.63905 | 9.21034 |
| H | 4.52195 | -1.39729 | 10.96119 |
| H | 5.72696 | -2.47611 | 10.24490 |
| C | 7.16690 | -0.67674 | 8.80952 |

| | | | | | | | |
|----|----------|----------|----------|----|----------|---------|----------|
| C | 7.00120 | -1.79224 | 7.96900 | C | 9.04015 | 3.29593 | 0.48717 |
| H | 6.11530 | -2.41845 | 8.04074 | H | 9.77417 | 2.65595 | 0.98064 |
| C | 7.97832 | -2.11665 | 7.02777 | C | 8.23463 | 4.16828 | 1.23834 |
| H | 7.84569 | -2.98094 | 6.37227 | C | 8.22168 | 4.15705 | 2.78452 |
| C | 9.12879 | -1.32946 | 6.95967 | C | 9.21445 | 3.09605 | 3.30083 |
| H | 9.94242 | -1.55613 | 6.26739 | H | 9.00301 | 2.11573 | 2.86128 |
| C | 9.21046 | -0.22768 | 7.80468 | H | 9.14811 | 2.98584 | 4.39049 |
| H | 10.09145 | 0.41726 | 7.79083 | H | 10.24792 | 3.36089 | 3.04076 |
| N | 8.28973 | 3.20838 | 7.64874 | C | 6.78237 | 3.76253 | 3.18252 |
| N | 7.82335 | 4.11552 | 10.27054 | C | 6.47215 | 2.51423 | 3.74093 |
| N | 6.04446 | 2.01963 | 9.16799 | H | 7.26316 | 1.80021 | 3.97724 |
| N | 7.75594 | 0.93745 | 11.34165 | C | 4.25253 | 2.98353 | 3.71291 |
| N | 8.24856 | 0.11935 | 8.67404 | H | 3.21854 | 2.68745 | 3.92523 |
| Co | 8.17690 | 2.10384 | 9.53314 | C | 4.52397 | 4.21541 | 3.15033 |
| C | 7.44457 | -0.72510 | 13.56738 | H | 3.70751 | 4.89673 | 2.89836 |
| H | 7.31820 | -1.38653 | 14.42795 | N | 5.30356 | 7.93292 | 2.98215 |
| N | 7.38997 | 6.79093 | 11.13787 | N | 6.91331 | 8.19919 | 0.68487 |
| C | 11.28944 | 2.41388 | 10.11017 | N | 7.95535 | 6.64842 | 2.97084 |
| N | 10.15933 | 2.23326 | 9.89776 | N | 7.35560 | 4.99259 | 0.59288 |
| C | 12.69717 | 2.67071 | 10.38818 | N | 8.99244 | 3.18162 | -0.84342 |
| H | 13.30279 | 2.56014 | 9.47451 | N | 5.78329 | 4.64491 | 2.89103 |
| H | 12.83036 | 3.69615 | 10.76842 | N | 5.22980 | 2.10729 | 4.03322 |
| H | 13.07883 | 1.96602 | 11.14394 | Co | 6.23349 | 6.47454 | 1.73115 |

3'-Co+2e⁻ (S = 1/2)

| | |
|------------------|----------------|
| Energy | = -2940.442323 |
| Enthalpy 0K | = -2939.942839 |
| Free Energy 298K | = -2940.021143 |

| | | | |
|----|----------|---------|----------|
| C | 9.04015 | 3.29593 | 0.48717 |
| H | 9.77417 | 2.65595 | 0.98064 |
| C | 8.23463 | 4.16828 | 1.23834 |
| C | 8.22168 | 4.15705 | 2.78452 |
| C | 9.21445 | 3.09605 | 3.30083 |
| H | 9.00301 | 2.11573 | 2.86128 |
| H | 9.14811 | 2.98584 | 4.39049 |
| H | 10.24792 | 3.36089 | 3.04076 |
| C | 6.78237 | 3.76253 | 3.18252 |
| C | 6.47215 | 2.51423 | 3.74093 |
| H | 7.26316 | 1.80021 | 3.97724 |
| C | 4.25253 | 2.98353 | 3.71291 |
| H | 3.21854 | 2.68745 | 3.92523 |
| C | 4.52397 | 4.21541 | 3.15033 |
| H | 3.70751 | 4.89673 | 2.89836 |
| N | 5.30356 | 7.93292 | 2.98215 |
| N | 6.91331 | 8.19919 | 0.68487 |
| N | 7.95535 | 6.64842 | 2.97084 |
| N | 7.35560 | 4.99259 | 0.59288 |
| N | 8.99244 | 3.18162 | -0.84342 |
| N | 5.78329 | 4.64491 | 2.89103 |
| N | 5.22980 | 2.10729 | 4.03322 |
| Co | 6.23349 | 6.47454 | 1.73115 |
| C | 3.64245 | 5.85492 | -0.11713 |
| N | 4.54268 | 6.17813 | 0.54016 |
| C | 2.51865 | 5.41868 | -0.93642 |
| H | 1.74051 | 6.19663 | -0.97270 |
| H | 2.85261 | 5.20744 | -1.96422 |
| H | 2.08012 | 4.49935 | -0.51787 |

| | | | |
|---|----------|----------|----------|
| C | 3.99668 | 7.85961 | 3.30437 |
| H | 3.47568 | 6.96155 | 2.96817 |
| C | 3.31839 | 8.83986 | 4.01445 |
| H | 2.25965 | 8.71261 | 4.25004 |
| C | 4.03117 | 9.97522 | 4.41551 |
| H | 3.55038 | 10.77579 | 4.98225 |
| C | 5.38156 | 10.06490 | 4.07340 |
| H | 5.93742 | 10.94936 | 4.37528 |
| C | 6.00370 | 9.03049 | 3.35641 |
| C | 7.49424 | 9.10699 | 2.92891 |
| C | 8.12677 | 10.38561 | 3.51134 |
| H | 8.03495 | 10.41368 | 4.60396 |
| H | 7.63601 | 11.28269 | 3.11463 |
| H | 9.19364 | 10.44997 | 3.26265 |
| C | 7.54799 | 9.17384 | 1.37846 |
| C | 8.18947 | 10.22558 | 0.70163 |
| H | 8.72583 | 10.99775 | 1.24802 |
| C | 8.14762 | 10.29884 | -0.69024 |
| H | 8.65348 | 11.11225 | -1.21554 |
| C | 7.43466 | 9.31760 | -1.38828 |
| H | 7.34726 | 9.33431 | -2.47657 |
| C | 6.84145 | 8.29907 | -0.65539 |
| H | 6.27942 | 7.51089 | -1.15906 |
| C | 8.26713 | 7.86408 | 3.46220 |
| C | 9.25892 | 8.00797 | 4.44280 |
| H | 9.51738 | 8.97862 | 4.85580 |
| C | 9.92611 | 6.87812 | 4.90160 |
| H | 10.69993 | 6.96613 | 5.66834 |
| C | 9.60026 | 5.63593 | 4.37150 |
| H | 10.12560 | 4.75616 | 4.73016 |
| C | 8.60168 | 5.53869 | 3.38863 |
| C | 7.26195 | 4.82707 | -0.74622 |
| H | 6.51988 | 5.43653 | -1.26804 |
| C | 8.05874 | 3.94650 | -1.45216 |
| H | 7.96071 | 3.85721 | -2.54004 |

| | | | |
|----|----------|----------|----------|
| N | 5.27329 | 2.24280 | 4.24039 |
| C | 6.50758 | 2.59603 | 3.87458 |
| C | 6.82279 | 3.81985 | 3.26058 |
| N | 5.83058 | 4.71617 | 3.00214 |
| C | 4.57979 | 4.34853 | 3.35247 |
| C | 4.30381 | 3.13664 | 3.96184 |
| C | 8.24759 | 4.16889 | 2.77849 |
| C | 9.24151 | 3.08831 | 3.24606 |
| Co | 6.42404 | 6.48158 | 1.77074 |
| N | 7.22756 | 4.98372 | 0.63799 |
| C | 6.99480 | 4.76606 | -0.68549 |
| C | 7.72098 | 3.85736 | -1.42649 |
| N | 8.71048 | 3.10922 | -0.88441 |
| C | 8.87689 | 3.25976 | 0.43578 |
| C | 8.15222 | 4.15737 | 1.23059 |
| C | 8.69364 | 5.54344 | 3.35962 |
| C | 9.77124 | 5.65297 | 4.25116 |
| C | 10.11362 | 6.90086 | 4.76339 |
| C | 9.38686 | 8.02471 | 4.37991 |
| C | 8.32218 | 7.87138 | 3.48226 |
| N | 8.00680 | 6.65077 | 3.00827 |
| C | 7.48382 | 9.08660 | 2.98374 |
| C | 7.62291 | 9.21506 | 1.44034 |
| C | 8.12545 | 10.37653 | 0.83042 |
| C | 8.20614 | 10.46073 | -0.55934 |
| C | 7.77675 | 9.37218 | -1.32337 |
| C | 7.29522 | 8.25435 | -0.65304 |
| N | 7.21746 | 8.16722 | 0.68655 |
| C | 5.96781 | 8.92253 | 3.29114 |

| | | | | | | | |
|---|----------|----------|----------|----|----------|---------|----------|
| C | 5.26186 | 9.84644 | 4.07325 | H | 9.95412 | 4.74500 | 4.86348 |
| C | 3.87667 | 9.74088 | 4.22665 | C | 8.57984 | 5.55432 | 3.38614 |
| C | 3.21194 | 8.70766 | 3.55087 | C | 7.20233 | 4.87365 | -0.78386 |
| C | 3.96825 | 7.81374 | 2.81169 | H | 6.42714 | 5.46191 | -1.28125 |
| N | 5.31820 | 7.87680 | 2.70548 | C | 8.00284 | 4.01795 | -1.51561 |
| C | 8.00758 | 10.36708 | 3.65834 | H | 7.87782 | 3.93098 | -2.60108 |
| H | 3.48621 | 6.99426 | 2.27265 | C | 9.04797 | 3.38175 | 0.39348 |
| H | 2.12732 | 8.59230 | 3.60230 | H | 9.81020 | 2.75708 | 0.86258 |
| H | 3.33156 | 10.45766 | 4.84416 | C | 8.24387 | 4.23134 | 1.17381 |
| H | 5.77896 | 10.66473 | 4.56914 | C | 8.26143 | 4.18864 | 2.72042 |
| H | 7.91456 | 10.30338 | 4.75005 | C | 9.29640 | 3.14811 | 3.19382 |
| H | 7.44443 | 11.24780 | 3.32707 | H | 9.11942 | 2.17598 | 2.72135 |
| H | 9.06506 | 10.53823 | 3.42047 | H | 9.23902 | 2.99073 | 4.27731 |
| H | 8.45983 | 11.22225 | 1.42647 | H | 10.31782 | 3.46646 | 2.94326 |
| H | 8.60110 | 11.36209 | -1.03394 | C | 6.84191 | 3.75593 | 3.14016 |
| H | 7.82019 | 9.38140 | -2.41422 | C | 6.56892 | 2.50910 | 3.72063 |
| H | 6.96800 | 7.37461 | -1.21041 | H | 7.38128 | 1.81210 | 3.93765 |
| H | 9.65955 | 8.99619 | 4.78309 | C | 4.34280 | 2.94338 | 3.75348 |
| H | 10.94871 | 6.99806 | 5.46148 | H | 3.32058 | 2.63162 | 3.99768 |
| H | 10.34538 | 4.78188 | 4.55345 | C | 4.57503 | 4.17136 | 3.16839 |
| H | 6.19420 | 5.34957 | -1.15041 | H | 3.73780 | 4.83102 | 2.92525 |
| H | 7.51214 | 3.72312 | -2.49374 | N | 5.30278 | 7.92428 | 3.03150 |
| H | 9.64507 | 2.62406 | 0.88113 | N | 6.86828 | 8.21576 | 0.67363 |
| H | 8.97534 | 2.10630 | 2.84088 | N | 7.96121 | 6.66724 | 2.93362 |
| H | 9.25036 | 3.00259 | 4.34031 | N | 7.33145 | 5.04020 | 0.55249 |
| H | 10.26088 | 3.31928 | 2.90961 | N | 8.97264 | 3.27763 | -0.93623 |
| H | 7.28870 | 1.86537 | 4.09278 | N | 5.81867 | 4.62083 | 2.86156 |
| H | 3.27790 | 2.87792 | 4.24610 | N | 5.34491 | 2.08287 | 4.05187 |
| H | 3.77606 | 5.06027 | 3.14954 | Co | 6.19948 | 6.44872 | 1.72631 |

3'-Co+2e⁻ (S = 3/2)

| | |
|------------------|----------------|
| Energy | = -2940.444296 |
| Enthalpy 0K | = -2939.944425 |
| Free Energy 298K | = -2940.021618 |

| | | | |
|---|----------|----------|----------|
| C | 4.01508 | 7.82693 | 3.40291 |
| H | 3.51651 | 6.89059 | 3.14658 |
| C | 3.32479 | 8.83525 | 4.06584 |
| H | 2.27872 | 8.69326 | 4.34499 |
| C | 4.00940 | 10.01804 | 4.35349 |
| H | 3.51484 | 10.84702 | 4.86546 |
| C | 5.34847 | 10.12227 | 3.97727 |
| H | 5.88512 | 11.04146 | 4.20035 |
| C | 5.98389 | 9.05329 | 3.32309 |
| C | 7.47853 | 9.11387 | 2.90801 |
| C | 8.11754 | 10.39844 | 3.47194 |
| H | 7.96978 | 10.47606 | 4.55577 |
| H | 7.67817 | 11.29357 | 3.01304 |
| H | 9.19842 | 10.41827 | 3.28326 |
| C | 7.57566 | 9.13680 | 1.35878 |
| C | 8.34516 | 10.09478 | 0.67673 |
| H | 8.94583 | 10.82147 | 1.21859 |
| C | 8.34810 | 10.12695 | -0.71796 |
| H | 8.95125 | 10.86660 | -1.25003 |
| C | 7.55598 | 9.20993 | -1.41032 |
| H | 7.49846 | 9.20665 | -2.50069 |
| C | 6.83734 | 8.28000 | -0.66692 |
| H | 6.20549 | 7.54139 | -1.16286 |
| C | 8.20956 | 7.87098 | 3.49133 |
| C | 9.08368 | 7.99880 | 4.57689 |
| H | 9.28498 | 8.96234 | 5.03762 |
| C | 9.71138 | 6.86095 | 5.07670 |
| H | 10.40051 | 6.93448 | 5.92155 |
| C | 9.45918 | 5.63115 | 4.47732 |

| | | | |
|----|----------|---------|----------|
| H | 9.95412 | 4.74500 | 4.86348 |
| C | 8.57984 | 5.55432 | 3.38614 |
| C | 7.20233 | 4.87365 | -0.78386 |
| H | 6.42714 | 5.46191 | -1.28125 |
| C | 8.00284 | 4.01795 | -1.51561 |
| H | 7.87782 | 3.93098 | -2.60108 |
| C | 9.04797 | 3.38175 | 0.39348 |
| H | 9.81020 | 2.75708 | 0.86258 |
| C | 8.24387 | 4.23134 | 1.17381 |
| H | 8.26143 | 4.18864 | 2.72042 |
| C | 9.29640 | 3.14811 | 3.19382 |
| H | 9.11942 | 2.17598 | 2.72135 |
| H | 9.23902 | 2.99073 | 4.27731 |
| H | 10.31782 | 3.46646 | 2.94326 |
| C | 6.84191 | 3.75593 | 3.14016 |
| C | 6.56892 | 2.50910 | 3.72063 |
| H | 7.38128 | 1.81210 | 3.93765 |
| C | 4.34280 | 2.94338 | 3.75348 |
| H | 3.32058 | 2.63162 | 3.99768 |
| C | 4.57503 | 4.17136 | 3.16839 |
| H | 3.73780 | 4.83102 | 2.92525 |
| N | 5.30278 | 7.92428 | 3.03150 |
| N | 6.86828 | 8.21576 | 0.67363 |
| N | 7.96121 | 6.66724 | 2.93362 |
| N | 7.33145 | 5.04020 | 0.55249 |
| N | 8.97264 | 3.27763 | -0.93623 |
| N | 5.81867 | 4.62083 | 2.86156 |
| N | 5.34491 | 2.08287 | 4.05187 |
| Co | 6.19948 | 6.44872 | 1.72631 |
| C | 3.58748 | 5.87890 | -0.04213 |
| N | 4.50302 | 6.18284 | 0.60612 |
| C | 2.44618 | 5.46322 | -0.84790 |
| H | 1.64512 | 6.21777 | -0.80724 |
| H | 2.74489 | 5.32708 | -1.89924 |
| H | 2.04747 | 4.50655 | -0.47515 |

3'-Co+2e⁻ (S = 3/2, 5-coordinates)

| | | | |
|------------------|----------------|---------|----------|
| Energy | = -2807.778830 | | |
| Enthalpy 0K | = -2807.324358 | | |
| Free Energy 298K | = -2807.389113 | | |
| N | 5.24335 | 2.27795 | 4.17803 |
| C | 6.48807 | 2.65727 | 3.87655 |
| C | 6.81447 | 3.85593 | 3.22355 |
| N | 5.81760 | 4.72584 | 2.86576 |
| C | 4.55229 | 4.31279 | 3.12796 |
| C | 4.27044 | 3.12304 | 3.76909 |
| C | 8.25591 | 4.19739 | 2.78319 |
| C | 9.23321 | 3.11887 | 3.28865 |
| Co | 6.42023 | 6.48593 | 1.77130 |
| N | 7.32840 | 5.02571 | 0.60438 |
| C | 7.08053 | 4.76704 | -0.70862 |
| C | 7.73741 | 3.77800 | -1.40989 |
| N | 8.67139 | 2.98486 | -0.83471 |
| C | 8.84114 | 3.17276 | 0.47931 |
| C | 8.18486 | 4.15500 | 1.23499 |
| C | 8.69613 | 5.57286 | 3.35800 |
| C | 9.77030 | 5.69089 | 4.25323 |
| C | 10.10972 | 6.94040 | 4.76347 |
| C | 9.37974 | 8.06124 | 4.37601 |
| C | 8.31944 | 7.90001 | 3.47594 |
| N | 8.00519 | 6.67873 | 3.00176 |
| C | 7.47787 | 9.11205 | 2.97311 |
| C | 7.60185 | 9.22568 | 1.42661 |

C 8.14001 10.36519 0.80506
C 8.21904 10.43334 -0.58564
C 7.75450 9.35334 -1.33731
C 7.23771 8.25782 -0.65480
N 7.16111 8.18632 0.68501
C 5.97191 8.93958 3.31916
C 5.28736 9.85648 4.13163
C 3.91618 9.72029 4.35361
C 3.24282 8.66607 3.73089
C 3.98188 7.78599 2.95333
N 5.31559 7.88764 2.76774
C 8.00781 10.40097 3.62659
H 3.49173 6.95129 2.44890
H 2.16672 8.52211 3.84622
H 3.38657 10.43263 4.99008
H 5.81080 10.68832 4.59657
H 7.92681 10.35178 4.71971
H 7.44267 11.27838 3.28835
H 9.06352 10.56657 3.37835
H 8.50271 11.20586 1.39134
H 8.63933 11.31810 -1.06977
H 7.79378 9.35118 -2.42833
H 6.87885 7.38590 -1.20345
H 9.64936 9.03500 4.77606
H 10.94496 7.04167 5.46078
H 10.34710 4.82248 4.55772
H 6.32663 5.38706 -1.20352
H 7.52325 3.61917 -2.47279
H 9.55935 2.50105 0.95374
H 8.95601 2.12906 2.91245
H 9.23184 3.06408 4.38481
H 10.25767 3.32778 2.95221
H 7.27423 1.96466 4.18433
H 3.23258 2.83793 3.97430
H 3.73789 4.97221 2.81954

Reference:

1. S. Grimme, *Chem.-Eur. J.*, 2012, **18**, 9955.