# Wide Bandgap OPV Polymers Based on Pyridinonedithiophene Unit with Efficiency >5\% 

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Materials characterization and spectroscopy. ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectra were recorded on a Bruker DRX-500 spectrometer in $\mathrm{CDCl}_{3}$. Mass spectra were obtained on a Bruker Ultraflextreme MALDI-TOF/TOF mass spectrometer using dithranol as a matrix. UV-vis spectra were taken on a Shimadzu UV-3600 spectrofluorometer in chloroform or drop-cast on glass slides. Photoluminescence spectra were taken on a Horiba FL-1039/40. Cyclic Voltammetry data was taken with an AUTOLAB/PG-STAT12 system using a three electrode system with a polymer-coated Pt working electrode, a Pt wire auxiliary electrode, and a $\mathrm{Ag} / \mathrm{AgNO}_{3}$ reference electrode in $0.1 \mathrm{M}\left[{ }^{\left[B_{4}\right.} \mathrm{B}_{4} \mathrm{~N}^{+}\left[\mathrm{PF}_{6}\right]\right.$ - solution in degassed acetonitrile. TGA data was measured on a TA Q600 instrument using ca. 2 mg of polymer in an Al pan heated to $550^{\circ} \mathrm{C}$. Polymer Molecular weights were measured using Gel Permeation Chromatography (GPC) on a Waters Associates Liquid Chromatography instrument equipped with a Waters 510 HPLC pump, a Waters 410 differential refractometer, and a Waters 486 tunable absorbance detector using chloroform as the eluent and polystyrene as the standard. Elemental Analysis was performed by Atlantic MicroLab, Inc. in Norcross, GA.

Solar cell fabrication and testing. Polymers ( $10 \mathrm{mg} / \mathrm{mL}$ ) and $\mathrm{PC}_{71} \mathrm{BM}(15 \mathrm{mg} / \mathrm{mL})$ were dissolved in chlorobenzene/chloroform and 1,8 -diodooctane co-solvent ( $97: 3, \mathrm{v} / \mathrm{v}$ ) and stirred on hot plate at $100^{\circ} \mathrm{C}$ overnight. ITO coated glass substrates (Thin Film Devices) were cleaned with water, acetone and isopropyl alcohol under sonication for 15 min . They were then exposed to ultraviolet ozone irradiation for 1 hour. PEDOT:PSS (Clevios ${ }^{\text {TM }}$ P VP AI 4083) layer ( $\sim 40 \mathrm{~nm}$ ) was then spin-coated and dried at 80 ${ }^{\circ} \mathrm{C}$ for 30 mins. Polymer: $\mathrm{PC}_{71} \mathrm{BM}$ solutions were then spin coated in glove box. After that, metal cathodes were thermal evaporated. The as prepared solar cells were measured under 1-sun, AM 1.5G irradiation.

GIWAXS Experimental Setup. Grazing incidence wide-angle x-ray scattering measurements were performed at Beamline 8-ID-E of the Advanced Photon Source at Argonne National Laboratory. The photon energy is $7.35 \mathrm{keV}(\lambda=1.6868 \AA$ ) and data were collected on a Pilatus 1 M pixel array detector at a sample-detector distance of 204 mm . All films were prepared with $10 \mathrm{mg} / \mathrm{ml}$ of polymer in $97: 3 \mathrm{v} / \mathrm{v}$ CF/DIO solution. Blend films had a 1:1.5 Polymer to $\mathrm{PC}_{71} \mathrm{BM}$ ratio. All films were spun on clean Si at 1500 rpm for 1 minute. Spectra were collected at an incidence angle of $0.2^{\circ}$; the films were exposure for 25 seconds. To account for the gaps in the detector array, two images were taken per sample, one with the detector in the standard position and the other translated 23 mm down to fill the gap, the two images are then merged. Some of the gap fills were unsatisfactory and that section of those line cuts have been omitted for clarity.

1D line cuts were taken of the 2D scattering spectra in the in-plane and out-of-plane direction using the gixsgui software package developed by the beamline scientists at APS 8-ID-E. To normalize for the differing exposure times and air scatter, the line cuts were background subtracted utilizing an exponential fit. The background-subtracted peaks were fit using the multipeak fit function in igor pro. Scherrer analysis was performed utilizing the previously mentioned method by Smiglies to account for instrumental broadening and detection limits. The values presented represent a lower limit for correlation length, as the Scherrer analysis does not account for broadening due to defects in the crystallites.

Synthesis of materials. Detailed synthetic schemes of polymers and monomer precursors can be below. All reagents were purchased through commercial suppliers and used directly unless otherwise specified. Toluene and THF were dried over sodium and distilled prior to use.

## tert-butyl (2-butyloctyl)(thiophen-3-yl)carbamate (2)


tert-Butyl (3-thienyl)carbamate (1,5.0g, 25.1mmol) was dissolved in 250 mL DMF and cooled to $0^{\circ} \mathrm{C}$. $\mathrm{NaH}(2.0 \mathrm{~g} 60 \%$ dispersion in mineral oil, 50 mmol ) was added and the solution was allowed to rise to room temperature. 5 -(bromomethyl)-undecane $(9 \mathrm{~mL}, 37.5 \mathrm{mmol})$ was added and the solution was allowed to stir overnight. The mixture was concentrated and then poured into water, followed by extraction with ethyl acetate $3 x$, the organic layers were combined, washed with water $3 x$, washed with brine, then dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The resulting yellow oil was purified via column chromatography starting with hexane as eluent then changing to 9:1 Hexane:EtOAc resulting in a clear oil in $89 \%$ yield. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 500 \mathrm{hz}\right): \delta 7.18(1 \mathrm{H}, \mathrm{dd}, \mathrm{j}=10,7) 7.03,6.98$ $\left(2 \mathrm{H}, \mathrm{br}\right.$ d) $3.57(2 \mathrm{H}, \mathrm{d}, \mathrm{j}=15) 1.60(1 \mathrm{H}, \mathrm{m}) 1.47(9 \mathrm{H}, \mathrm{s}) 1.22(16 \mathrm{H}, \mathrm{m}) 8.86(6 \mathrm{H}, \mathrm{m}){ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 500 \mathrm{hz}\right): \delta 154.6,141.1,125.5$, 123.7, 115.5, $80.3,53.4,36.6,31.9,31.4,31.1,29.8,28.6,28.4,26.3,23.2,22.7,14.2,14.1$ MALDI: $312 \mathrm{~m} / \mathrm{z}(\mathbf{2}$ - isobutylene +H$)$

( $8.2 \mathrm{~g}, 22.3 \mathrm{mmol}$ ) was dissolved in 220 mL DCM and cooled to $0^{\circ} \mathrm{C} .15 \mathrm{~mL}$ TFA was added and the solution was allowed to warm to room temperature and stir for 24 hours. Reaction turned pink upon stirring. Poured into water, washed with sodium bicarbonate, washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ to afford a reddish-purple oil. Used without further purification or characterization.

## 2-Bromo-3-Thiophenecarbonyl chloride (4)



2-bromo-3-thiophenecarboxylic acid ( $7.0 \mathrm{~g}, 33.6 \mathrm{mmol}$ ) was added to a flame-dried flask equipped with a condenser. 140 mL distilled toluene, 15 mL SOCl 2 and $\sim 1 \mathrm{~mL}$ DMF were added and the solution was heated to reflux, after which it dissolved. This was allowed to reflux overnight. The solution was then cooled and the solvents were removed via rotovap. More toluene was added and this was also removed. The light brown solid was dried and then directly used in the next reaction with no further purification or characterization.

## 2-bromo-N-(2-butyloctyl)-N-(thiophen-3-yl)thiophene-3-carboxamide (5)



600 mg NaH ( $60 \%$ in mineral oil, 15.0 mmol ) was dispersed in 60 mL distilled toluene at $0^{\circ} \mathrm{C} .3(1.3 \mathrm{~g}, 5 \mathrm{mmol}$ ) was added and the solution was allowed to rise to room temperature and stirred for 30 minutes. The solution was cooled back down to $0^{\circ} \mathrm{C}$ and a solution of $4(1.1 \mathrm{~g}, 5 \mathrm{mmol})$ dissolved in toluene was added dropwise. The mixture was allowed to stir overnight. Solution was allowed to cool, then poured into water. The organic layer was washed with water twice, then brine, then dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Toluene was removed, and the product was purified by column chromatography ( $30 \% \mathrm{EtOAc}: 70 \% \mathrm{Hexane}$ ) 1.5 g produced ( $65 \%$ yield) ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 500 \mathrm{hz}\right): \delta 77.13(1 \mathrm{H}, \mathrm{br}$ s) $7.00(1 \mathrm{H}, \mathrm{br}$ s) $6.86(1 \mathrm{H}, \mathrm{br}$ s) $6.78(1 \mathrm{H}, \mathrm{br} \mathrm{s}) 3.81(2 \mathrm{H}, \mathrm{d} \mathrm{j}=12) 1.62(1 \mathrm{H}, \mathrm{m}) 1.2-$ $1.4(16 \mathrm{H}, \mathrm{m}) 0.87(6 \mathrm{H}, \mathrm{m}){ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 500 \mathrm{hz}\right): \delta 165.3,140.6,138.3,127.0,126.2,125.6,125.1,119.8,111.8,52.5,36.2$, $31.9,31.3,31.0,29.7,28.6,26.3,23.1,22.7,14.2,14.1$ MALDI: 456 (90\%) 458 (100\%) m/z (5 + H)

4-(2-butyloctyl)dithieno[3,2-b:2',3'-d]pyridin-5(4H)-one (6)


Combine 5 ( $1.5 \mathrm{~g}, 3.3 \mathrm{mmol}$ ) with $\mathrm{Pd}(\mathrm{OAc})_{2}(74 \mathrm{mg}, 0.33 \mathrm{mmol}), \mathrm{K}_{2} \mathrm{CO}_{3}$ ( $450 \mathrm{mg}, 3.3 \mathrm{mmol}$ ) pivalic acid ( $102 \mathrm{mg}, 1 \mathrm{mmol}$ ) and 33 mL degassed DMA in a round bottom flask equipped with a condenser. The setup was subjected to 3 cycles of pump/backfill with Ar and then heated to $110^{\circ} \mathrm{C}$ overnight. Cool, pour into water, extract with EtOAc, wash with water, brine, dry over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Produced $1.1 \mathrm{~g}\left(90 \%\right.$ yield) of red oil with blue fluorescence. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 500 \mathrm{hz}\right): \delta 7.70(1 \mathrm{H}, \mathrm{d}, \mathrm{j}=11) 7.44(1 \mathrm{H}, \mathrm{d}, \mathrm{j}=11) 7.23$ $(1 \mathrm{H}, \mathrm{d}, \mathrm{j}=11) 7.08(1 \mathrm{H}, \mathrm{d}, \mathrm{j}=11) 4.22(2 \mathrm{H}, \mathrm{br} \mathrm{s}) 1.99(1 \mathrm{H}, \mathrm{m}) 1.2-1.4(16 \mathrm{H}, \mathrm{m}) 0.86(6 \mathrm{H}, \mathrm{m}){ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 500 \mathrm{hz}\right): \delta 158.7,141.3$,
$140.4,128.6,126.4,125.5,122.5,117.4,114.1,49.0,37.1,31.7,31.4,31.2,29.5,28.7,26.5,23.0,22.5,14.0,14.0 \mathrm{MALDI}: 376$ $\mathrm{m} / \mathrm{z}(6+\mathrm{H})$

## 2,7-dibromo-4-(2-butyloctyl)dithieno[3,2-b:2',3'-d]pyridin-5(4H)-one (7)



Dissolve $6(0.43 \mathrm{~g}, 1.14 \mathrm{mmol})$ in DMF $(25 \mathrm{~mL})$ and add NBS $(2 \mathrm{~g}, 4.6 \mathrm{mmol})$. Cover reaction vessel with foil to protect from ambient light and allow to stir overnight. Dissolve in EtOAc, wash with $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ solution, wash with water, wash with brine. Dry over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Dry thoroughly under high vacuum and then recrystallize from Hexane. 590mg dark yellow powder produced ( $97 \%$ yield). ${ }^{1} \mathrm{H}-\mathrm{NMR}^{\left(\mathrm{CDCl}_{3}, 500 \mathrm{hz}\right): ~} 7.63(1 \mathrm{H}, \mathrm{s}) 7.05(1 \mathrm{H}, \mathrm{s}) 4.11(2 \mathrm{H}, \mathrm{br} \mathrm{s}) 1.92(1 \mathrm{H}, \mathrm{m}) 1.2-1.4(16 \mathrm{H}, \mathrm{m}) 0.86(6 \mathrm{H}, \mathrm{m}){ }^{13} \mathrm{C}-\mathrm{NMR}(\mathrm{CDCl} 3$, 500hz): $\delta 157.2,141.6,140.0,128.9,128.8,120.6,115.1,114.3,111.0,49.3,37.2,31.8,31.4,31.1,29.6,28.7,26.5,23.1,22.6$, 14.1, 14.1 MALDI: 532 (50\%) 534 (100\%) 536 (50\%) m/z (7 + H)

General synthetic scheme for polymers. 0.1 mmol 7 was combined with 0.1 mmol of desired distannyl coupling partner and 0.005 mmol tetrakis(triphenylphosphino)palladium(0) in a flame-dried 10 mL round-bottom flask equipped with a condenser. The reaction vessel was evacuated and backfilled with Ar 3 times, and then 3.3 mL of anhydrous toluene and 0.8 mL of anhydrous DMF were added. The resulting solution was heated to $120^{\circ} \mathrm{C}$ for 24 hours. After cooling, the polymer was precipitated from methanol, after which it was washed in a Soxhlet extractor successively with methanol, acetone, and hexanes, then extracted with chloroform. The chloroform portion was filtered through celite, then concentrated and once more precipiated from methanol, after which it was centrifuged and dried to afford the final polymer.



Figure S1. Monomer CV (left) and energy level comparison to TPTI (right).


Figure S2. Polymer TGA Traces.


Figure S3. Polymer CV Curves.


Figure $S 4 . J_{s c}$ dependence on light Intensity measurements for PPDT3 and PPDT4 devices.


Figure S5. a, GIWAXS 2-D Scattering Images of Pure Polymers; b, GIWAXS 2-D Scattering Images of Polymer:PCBM blends.

Polymer ${ }^{1} \mathrm{H}-\mathrm{NMR}$




Table S1. Polymer Elemental Analysis.

| Polymer | Calc C | Found C | Calc H | Found H | Calc S | Found S |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| PPDT1 | 68.98 | 65.60 | 7.77 | 7.78 | 15.67 | 13.33 |
| PPDT2 | 72.27 | 71.72 | 8.29 | 8.23 | 15.75 | 15.23 |
| PPDT3 | 69.50 | 68.37 | 7.11 | 7.17 | 20.24 | 19.55 |
| PPDT4 | 69.74 | 69.44 | 7.85 | 7.83 | 18.16 | 18.05 |
| PPDT5 | 68.39 | 67.09 | 8.03 | 7.98 | 16.23 | 15.81 |

Table S2. Summary of average $J_{s c}, V_{o c,}, F F$ and PCE values from six devices for each PPDT:PCBM system.

| Polymer | $V_{\text {oc }}(\mathrm{V})$ | $J_{s c}\left(\mathrm{~mA} / \mathrm{cm}^{2}\right)$ | $F F(\%)$ | $P C E(\%)$ |
| :---: | :---: | :---: | :---: | :---: |
| PPDT1 | $0.87 \pm 0.01$ | $8.22 \pm 0.08$ | $44.9 \pm 0.40$ | $3.20 \pm 0.08$ |
| PPDT2 | $0.85 \pm 0.01$ | $9.10 \pm 0.15$ | $52.5 \pm 0.31$ | $4.04 \pm 0.12$ |
| PPDT3 | $0.89 \pm 0.01$ | $8.33 \pm 0.12$ | $70.3 \pm 0.26$ | $5.19 \pm 0.14$ |
| PPDT4 | $0.86 \pm 0.01$ | $3.27 \pm 0.21$ | $31.0 \pm 0.36$ | $0.86 \pm 0.08$ |
| PPDT5 | $0.75 \pm 0.01$ | $8.05 \pm 0.24$ | $40.5 \pm 0.23$ | $2.43 \pm 0.09$ |

