

Supporting Information

The approach to 4d/4f-Polyphosphides

Nicholas Arleth,^a Michael T. Gamer,^a Ralf Köppe,^a Nikolay A. Pushkarevsky,^{b,c} Sergey N. Konchenko,^{a,b} Martin Fleischmann,^c Michael Bodensteiner,^c Manfred Scheer,^c and Peter W. Roesky^{*a}

^a Institute of Inorganic Chemistry, Karlsruhe Institute of Technology,
Engesserstrasse 15, 76131 Karlsruhe (Germany), E-mail: roesky@kit.edu.

^b Nikolaev Institute of Inorganic Chemistry SB RAS, Prosp. Lavrentieva 3, 630090
Novosibirsk (Russia, and Novosibirsk State University Pirogova str. 2, 630090
Novosibirsk (Russia), E-mail: konch@niic.nsc.ru

^c Institute of Inorganic Chemistry, University of Regensburg, 93040 Regensburg,
Germany, E-mail: Manfred.Scheer@chemie.uni-regensburg.de

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supplementary information available should be included here]. See
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Experimental

General Considerations

All manipulations of air-sensitive materials were performed with the rigorous exclusion of oxygen and moisture in flame-dried Schlenk-type glassware either on a dual manifold Schlenk line, interfaced to a high vacuum (10^{-3} torr) line, or in an argon-filled MBraun glove box. Elemental analyses were carried out with an Elementar vario Micro Cube. Hydrocarbon solvents were predried by using an MBraun solvent purification system (SPS-800) and degassed, dried and stored *in vacuo* over LiAlH₄. Tetrahydrofuran was distilled under nitrogen from potassium benzophenone ketyl before storage over LiAlH₄. IR spectra were obtained on a Bruker Tensor 37 FTIR spectrometer equipped with a room temperature DLaTGS detector and a diamond ATR (attenuated total reflection) unit; for the mid infrared region a KBr beamsplitter was used. [Cp*₂Ln(thf)₂] (Ln = Sm, Yb),¹⁻³ [{CpMo(CO)₂}₂(μ,η^{2:2}-P₂)],⁴ and [Cp*Mo(CO)₂(η³-P₃)]⁵ were prepared according to literature procedures.

Near infrared absorbance (NIR) measurements of **1a and **4****

NIR measurements of **1a** and **4** were performed with the help of an ATR diamond at room temperature using the FTIR spectrometer Bruker Tensor 37 by means of an NIR lamp, a CaF₂ beamsplitter and a room temperature InGaAs detector (Figure S3 and S11).

Magnetic Measurements

The magnetic measurements were carried out with the use of a Quantum Design SQUID magnetometer MPMS-XL in the temperature range 1.8 - 300 K and with dc applied fields ranging from 7 to -5 T. Measurements were performed on the polycrystalline samples with extreme caution. The sample bag was prepared in glove box, sealed under argon and transferred into the magnetometer immediately. The magnetic data were corrected for the sample holder.

General procedure for ampoule reactions

For the synthesis and recrystallization, two-section and three-section ampoules were used (see . The starting compounds were loaded into one section of the ampoule in an argon-filled glove box. The section with the starting materials was cooled by immersion in a liquid nitrogen bath, and the required solvent (typically 10 mL of solvent) was condensed *in vacuo* onto the starting materials. The ampoule was then flame-sealed. The reaction mixture was slowly warmed up to room temperature and then heated to 60 °C until the color had definitely changed from purple to red-brown. If a precipitate formed, the product was separated by decantation of the solution to another section of the ampoule. A concentrated solution was obtained by slow evaporation of

the solvent to the empty section of the ampoule. Crystals were obtained at room temperature and isolated by decantation of the solution to the other section of ampoule followed by drying by means of cooling the section with the mother liquor. The section with crystals was flame-sealed and opened in a glovebox.

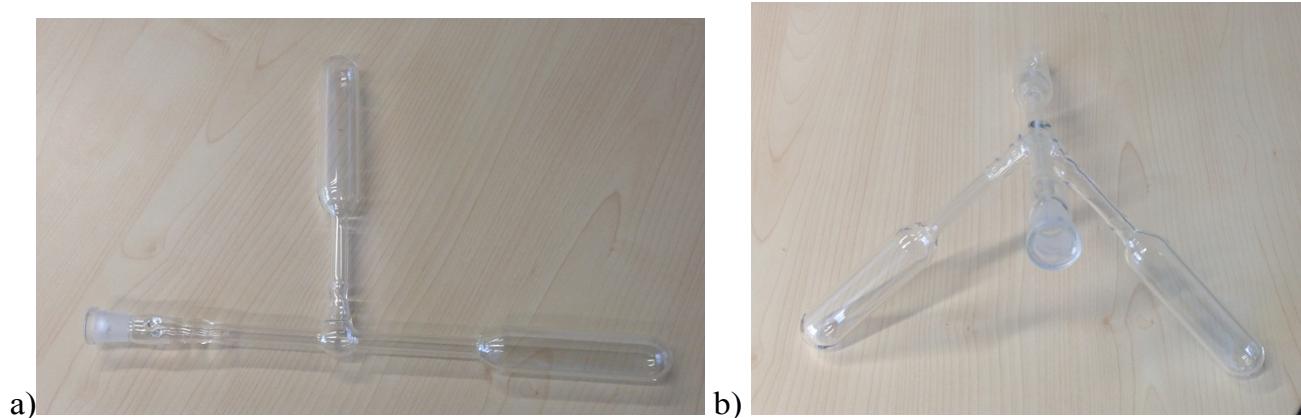


Figure S1: (a) Two section and (b) three section ampoule.

$[(\text{Cp}^*_2\text{Sm})_2\text{P}_2(\text{CpMo}(\text{CO})_2)_4]$ (1a)

In a three-section-ampoule, toluene (15 mL) was condensed at -78 °C onto a mixture of $[\text{Cp}^*_2\text{Sm}(\text{thf})_2]$ (80 mg, 0.14 mmol) and $[\{\text{CpMo}(\text{CO})_2\}_2(\mu,\eta^{2:2}\text{-P}_2)]$ (70 mg, 0.141 mmol). The resulting dark red reaction solution was heated for one week at 60° C. After two weeks at room temperature red crystals of $[(\text{Cp}^*_2\text{Sm})_2\text{P}_2(\text{CpMo}(\text{CO})_2)_4]$ were obtained. The supernatant solution was decanted and the section with $[(\text{Cp}^*_2\text{Sm})_2\text{P}_2(\text{CpMo}(\text{CO})_2)_4]$ was flamed-sealed. Yield: 10 mg, 7 % (single crystals).

IR (ATR, $\tilde{\nu}/\text{cm}^{-1}$): 2906 (vw), 2855 (w), 1945 (s), 1905 (vs), 1871 (vs), 1683 (vs), 1636 (s), 1423 (w), 1382 (w), 1058 (w), 1007 (w), 811 (vs), 789 (vs), 727 (vw), 693 (vw), 661 (vw), 557 (s), 531 (m).NIR (ATR, $\tilde{\nu}/\text{cm}^{-1}$): 9319 (m), 9217 (vw), 9151 (m), 9058 (w), 8968 (w), 8217 (s), 8158 (vw), 8048 (vs), 7960 (w), 7791 (w), 7338 (m), 7236 (vw), 7171 (vw), 6388 (w), 6126 (vw). Anal. Calc. for $\text{C}_{68}\text{H}_{80}\text{Mo}_4\text{O}_8\text{P}_2\text{Sm}_2$ ($1771.89 \text{ g}\cdot\text{mol}^{-1}$) (**1** – 1 Tol): C, 46.09; H, 4.55. Found: C, 46.01; H, 4.32.

$[(\text{Cp}^*_2\text{Sm})_2\text{P}_4(\text{CpMo}(\text{CO})_2)_2]$ (2) and $[(\text{Cp}^*_2\text{Sm})_3\text{P}_5(\text{CpMo}(\text{CO})_2)_3]$ (3)

From the remaining reaction mixture from **1a**, small amounts of two different kinds of orange crystals were obtained by slow evaporation. Due to the similar solubility and the low yields of the two products, no further analytical data could be collected.

[(Cp^{*})₂Yb)₂P₂(CpMo(CO)₂)₄] (1b)

In a two-section ampoule, toluene (15 mL) was condensed at -78 °C onto a mixture of [Cp^{*}₂Yb(thf)₂] (82 mg, 0.14 mmol) and [{CpMo(CO)₂}₂($\mu,\eta^{2:2}$ -P₂)] (70 mg, 0.141 mmol). The resulting dark red reaction solution was heated for one week at 60° C. After two weeks at room temperature, crystals of [(Cp^{*}₂Yb)₂(CpMo(CO)₂)₄P₂] were obtained. Yield: 8 mg, 6 % (single crystals).

IR (ATR, $\tilde{\nu}/\text{cm}^{-1}$): 2900 (w), 2856 (w), 1906 (vs), 1874 (vs), 1687 (vs), 1638 (s), 1486 (vw), 1432 (m), 1382 (w), 1107 (vw), 1059 (w), 1007 (m), 790 (s), 727 (w), 696 (vw), 585 (m), 553 (s), 504 (m). Anal. Calc. for C₆₈H₈₀Mo₄O₈P₂Yb₂ (1817.10 g·mol⁻¹): C, 44.49; H, 4.44. Found: C, 44.76; H, 4.27.

[(Cp^{*}₂Sm)₂P₆(Cp^{*}Mo(CO)₂)₂] (4)

Toluene (15 mL) was condensed at -78 °C onto a mixture of [Cp^{*}₂Sm(thf)₂] (178 mg, 0.316 mmol) and [Cp^{*}Mo(CO)₂(η^3 -P₃)] (120 mg, 0.316 mmol). The resulting reaction mixture was heated for one week at 60 °C. After two weeks at room temperature, crystals of [(Cp^{*}₂Sm)₂P₆(Cp^{*}Mo(CO)₂)₂] were obtained. Yield: 20 mg, 14 % (single crystals).

IR (ATR, $\tilde{\nu}/\text{cm}^{-1}$): 2958 (m), 2901 (s), 2853 (s), 2722 (vw), 1983 (m), 1917 (vs), 1701 (vs), 1646 (m), 1477 (w), 1442 (m), 1377 (m), 1150 (vw), 1104 (w), 1067 (w), 1026 (m), 799 (vw), 728 (m), 693 (w), 607 (vw), 555 (m), 515 (w). NIR (ATR, $\tilde{\nu}/\text{cm}^{-1}$): 9489 (vs), 9224 (vw), 9109 (vs), 8719 (w), 8107 (s), 7967 (vs), 7781 (w), 7338 (w), 7236 (w), 7152 (w), 6702 (m), 6566 (w), 6395 (w), 6325 (w). Anal. Calc. for C₇₁H₉₈Mo₂O₄P₆Sm₂ (1694.04 g·mol⁻¹) (**4 – 1Tol**): C, 50.34; H, 5.83. Found: C, 50.46; H, 6.02.

[(Cp^{*}₂Yb)₂P₆(Cp^{*}Mo(CO)₂)₂] (5)

Toluene (15 mL) was condensed at -78 °C onto a mixture of [Cp^{*}₂Yb(thf)₂] (157 mg, 0.273 mmol) and [Cp^{*}Mo(CO)₂(η^3 -P₃)] (102 mg, 0.273 mmol). The resulting reaction mixture was heated for one week at 60 °C. After two weeks at room temperature, black needles of [(Cp^{*}₂Yb)₂P₆(Cp^{*}Mo(CO)₂)₂] were obtained. Yield: 14 mg, 18 % (single crystals). IR (ATR, $\tilde{\nu}/\text{cm}^{-1}$): 3023 (vw), 2899 (s), 2854 (s), 2721 (vw), 1985 (w), 1916 (vs), 1737 (w), 1696 (vs), 1669 (s), 1493 (w), 1477 (w), 1445 (m), 1376 (m), 1310 (vw), 1240 (w), 1155 (vw), 1066 (m), 1025 (m), 798 (w), 728 (m), 693 (w), 609 (w), 555 (m). Anal. Calc. for C₆₇H₉₄Mo₂O₄P₆Yb₂ (1687.36 g·mol⁻¹) (**5 – 0.5 Tol**): C, 47.69; H, 5.62. Found: C, 48.14; H, 5.03.

X-ray Crystallographic Studies of 1-5

A suitable crystal was covered in mineral oil (Aldrich) and mounted on a glass fiber. The crystal was transferred directly to a cold stream of a STOE IPDS 2, STOE StadiVari or Xcalibur diffractometer.

All structures were solved using SHELXS-2013.⁶ The remaining non-hydrogen atoms were located from successive difference Fourier map calculations. The refinements were carried out by using full-matrix least-squares techniques on F , minimizing the function $(F_o - F_c)^2$, where the weight is defined as $4F_o^2/2(F_o^2)$ and F_o and F_c are the observed and calculated structure factor amplitudes using the program SHELXL-2013.⁶ Hydrogen atom positions were calculated. The locations of the largest peaks in the final difference Fourier map calculation as well as the magnitude of the residual electron densities in each case were of no chemical significance. Positional parameters, hydrogen atom parameters, thermal parameters, bond distances and angles have been deposited as supporting information.

Crystallographic data (excluding structure factors) for the structures reported in this paper have been deposited with the Cambridge Crystallographic Data Centre as a supplementary publication no. CCDC 1402049-1402054. Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge CB21EZ, UK (fax: +(44)1223-336-033; email: deposit@ccdc.cam.ac.uk)

Crystal data for **1a**: $C_{68}H_{80}Mo_4O_8P_2Sm_2 \bullet C_7H_8$, $M = 1863.85$, $a = 9.5023(3)$ Å, $b = 12.6103(4)$ Å, $c = 15.1418(5)$ Å, $\alpha = 97.784(3)^\circ$, $\beta = 90.790(3)^\circ$, $\gamma = 100.069(3)^\circ$, $V = 1768.68(10)$ Å³, $T = 100$ K, space group $P-1$, $Z = 1$, 17149 reflections measured, 6969 independent reflections ($R_{int} = 0.0684$). The final R_I values were 0.0516 ($I > 2\sigma(I)$). The final $wR(F^2)$ values were 0.1266 ($I > 2\sigma(I)$). The final R_I values were 0.0715 (all data). The final $wR(F^2)$ values were 0.1389 (all data). The goodness of fit on F^2 was 0.997.

Crystal data for **1b**: $C_{68}H_{80}Mo_4O_8P_2Yb_2$, $M = 1817.10$, $a = 9.4563(7)$ Å, $b = 12.5189(13)$ Å, $c = 14.9787(13)$ Å, $\alpha = 84.903(8)^\circ$, $\beta = 71.669(7)^\circ$, $\gamma = 76.285(7)^\circ$, $V = 1635.0(3)$ Å³, $T = 150$ K, space group $P-1$, $Z = 1$, 14805 reflections measured, 5918 independent reflections ($R_{int} = 0.0698$). The final R_I values were 0.0392 ($I > 2\sigma(I)$). The final $wR(F^2)$ values were 0.0708 ($I > 2\sigma(I)$). The final R_I values were 0.0654 (all data). The final $wR(F^2)$ values were 0.0772 (all data). The goodness of fit on F^2 was 0.945.

Crystal data for **2**: $C_{54}H_{70}Mo_2O_4P_4Sm_2 \bullet 2(C_7H_8)$, $M = 1583.82$, $a = 9.6185(7)$ Å, $b = 10.2667(7)$ Å, $c = 17.5096(13)$ Å, $\alpha = 83.519(6)^\circ$, $\beta = 77.307(6)^\circ$, $\gamma = 78.461(6)^\circ$, $V = 1648.44(21)$ Å³, $T = 100$ K, space group $P-1$, $Z = 1$, 12256 reflections measured, 5964 independent reflections ($R_{int} = 0.1155$). The final R_I values were 0.0708 ($I > 2\sigma(I)$). The final $wR(F^2)$ values were 0.1908 ($I >$

$2\sigma(I)$). The final R_I values were 0.0860 (all data). The final $wR(F^2)$ values were 0.2034 (all data). The goodness of fit on F^2 was 1.062.

Crystal data for **3**: $C_{81}H_{105}Mo_3O_6P_5Sm_3$, $M = 2068.36$, $a = 10.130(2)$ Å, $b = 25.686(5)$ Å, $c = 15.724(3)$ Å, $\alpha = 90^\circ$, $\beta = 96.36(3)^\circ$, $\gamma = 90^\circ$, $V = 4066.3(14)$ Å³, $T = 100$ K, space group $P121/m1$, $Z = 2$, 33852 reflections measured, 7394 independent reflections ($R_{int} = 0.2911$). The final R_I values were 0.1092 ($I > 2\sigma(I)$). The final $wR(F^2)$ values were 0.2500 ($I > 2\sigma(I)$). The final R_I values were 0.1889 (all data). The final $wR(F^2)$ values were 0.2946 (all data). The goodness of fit on F^2 was 0.980.

Crystal data for **4**: $C_{64}H_{90}Mo_2O_4P_6Sm_2 \bullet 2(C_7H_8)$, $M = 1786.02$, $a = 11.1671(8)$ Å, $b = 12.7931(10)$ Å, $c = 14.7276(9)$ Å, $\alpha = 98.952(6)^\circ$, $\beta = 110.227(6)^\circ$, $\gamma = 95.319(6)^\circ$, $V = 1925.7(2)$ Å³, $T = 123$ K, space group $P-1$, $Z = 1$, 15246 reflections measured, 6698 independent reflections ($R_{int} = 0.0368$). The final R_I values were 0.0343 ($I > 2\sigma(I)$). The final $wR(F^2)$ values were 0.0799 ($I > 2\sigma(I)$). The final R_I values were 0.0368 (all data). The final $wR(F^2)$ values were 0.0815 (all data). The goodness of fit on F^2 was 1.066.

Crystal data for **5**: $C_{64}H_{90}Mo_2O_4P_6Yb_2 \bullet 2(C_7H_8)$, $M = 1831.40$, $a = 11.1924(3)$ Å, $b = 12.6285(3)$ Å, $c = 14.6371(4)$ Å, $\alpha = 99.581(2)^\circ$, $\beta = 110.455(2)^\circ$, $\gamma = 94.847(2)^\circ$, $V = 1888.44(9)$ Å³, $T = 100$ K, space group $P-1$, $Z = 1$, 17122 reflections measured, 7340 independent reflections ($R_{int} = 0.0260$). The final R_I values were 0.0410 ($I > 2\sigma(I)$). The final $wR(F^2)$ values were 0.1014 ($I > 2\sigma(I)$). The final R_I values were 0.0447 (all data). The final $wR(F^2)$ values were 0.1033 (all data). The goodness of fit on F^2 was 1.084.

Figure S2. IR-spectrum of **1a**.

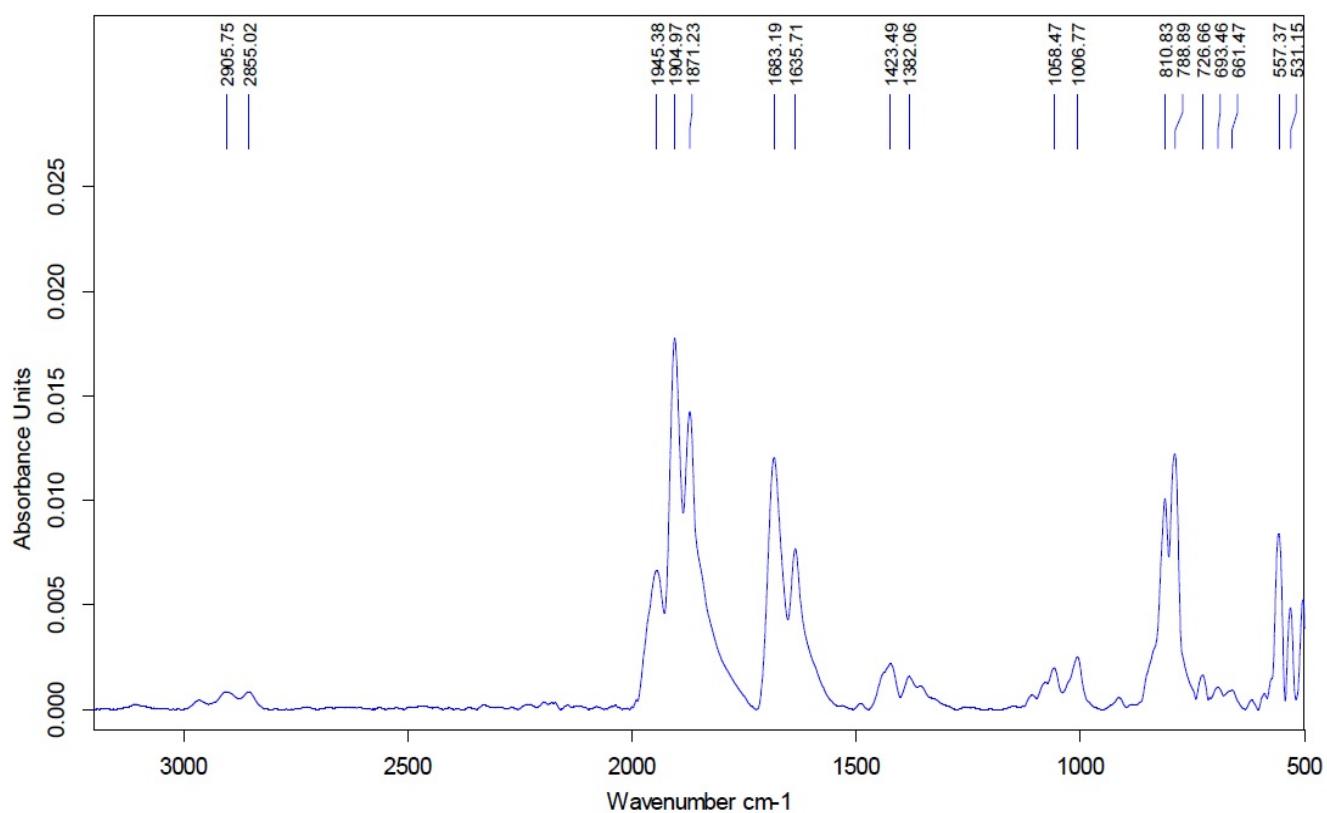


Figure S3. NIR-spectrum of **1a**.

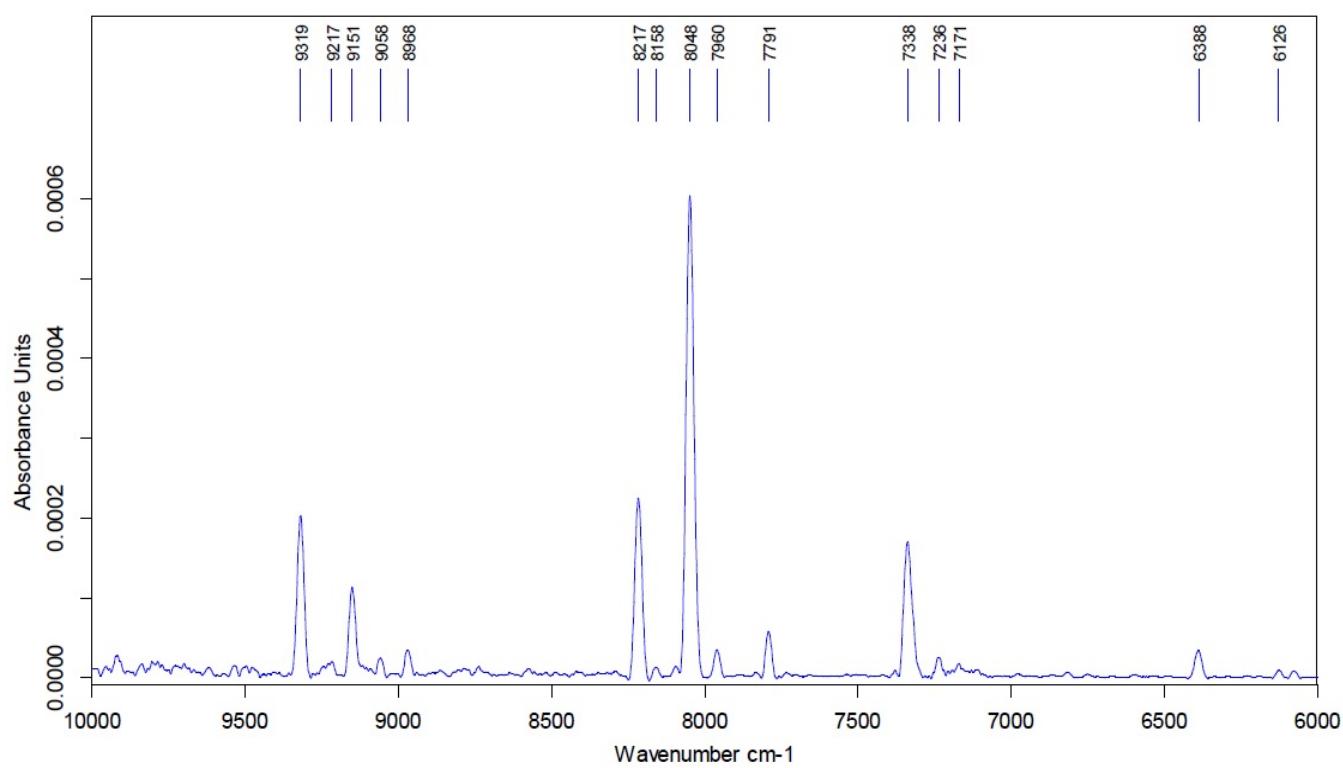


Figure S4. IR-spectrum of **1b**.

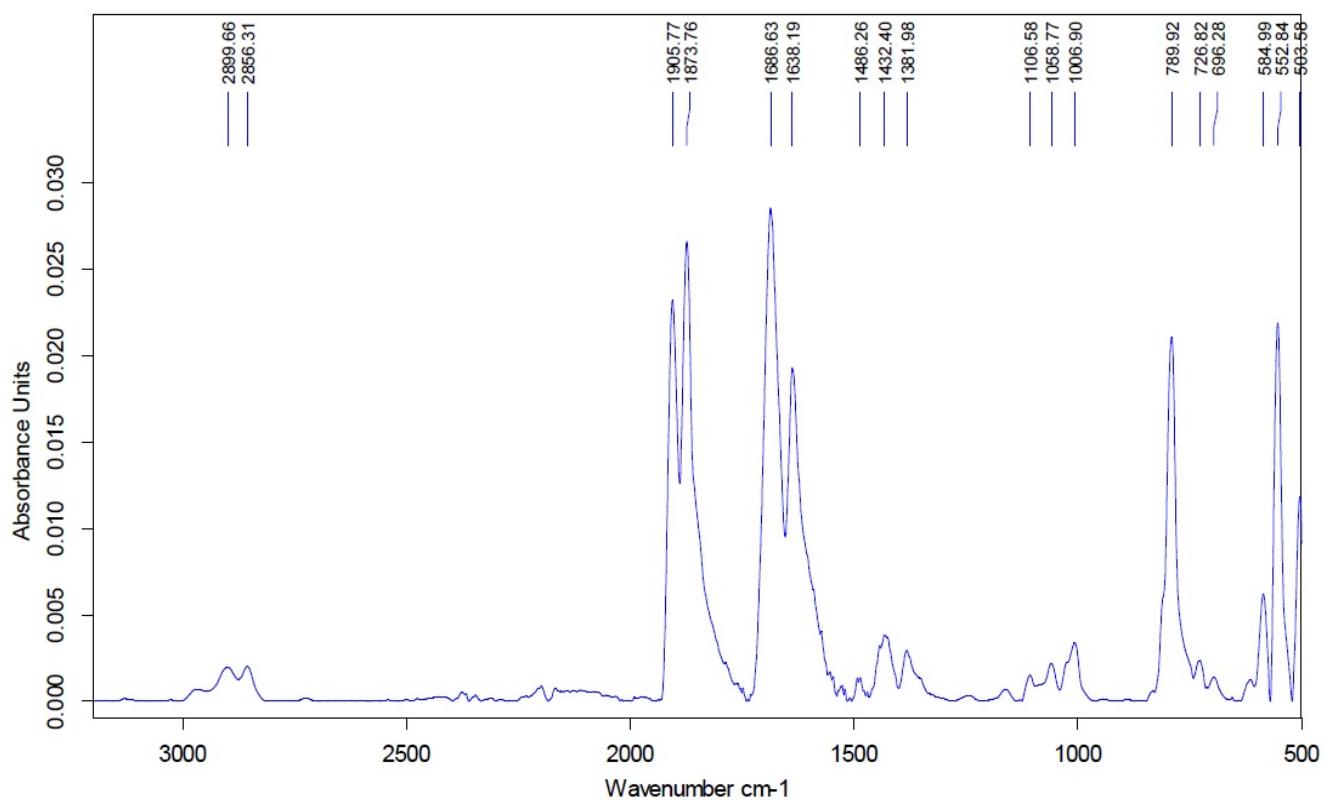


Figure S5. Solid-state structure of **3** showing the full disordered of the central P_5 -core. Hydrogen atoms are omitted for clarity.

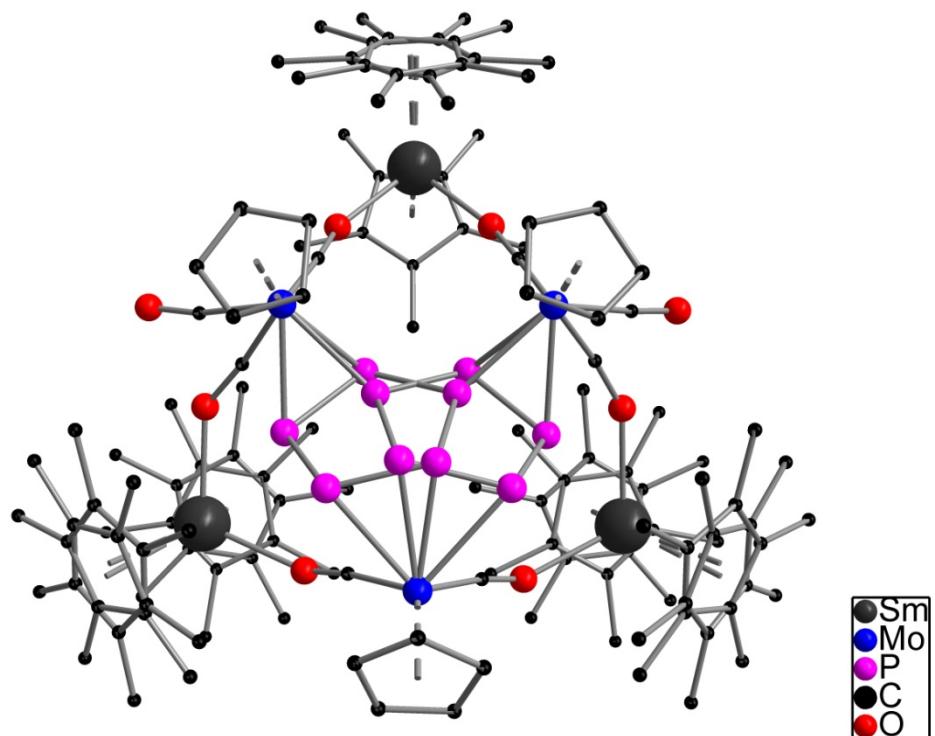


Figure S6. Solid-state structure of **3** showing each part of the disordered structure separated.

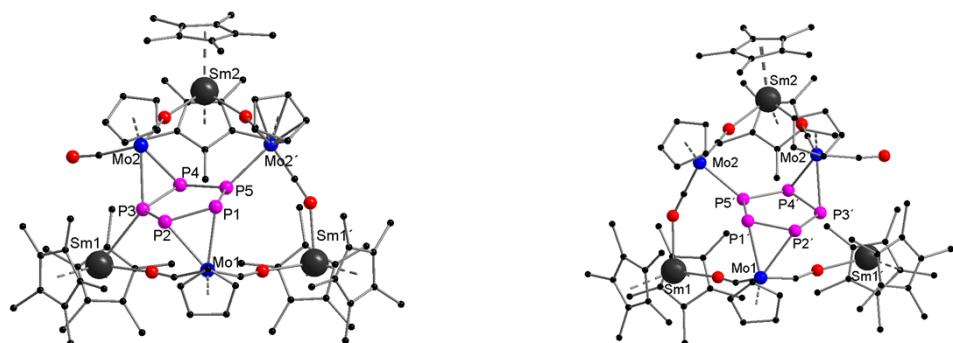


Figure S7. Solid-state structure of **4** showing the full disordered of the central P₆-core. Hydrogen atoms are omitted for clarity.

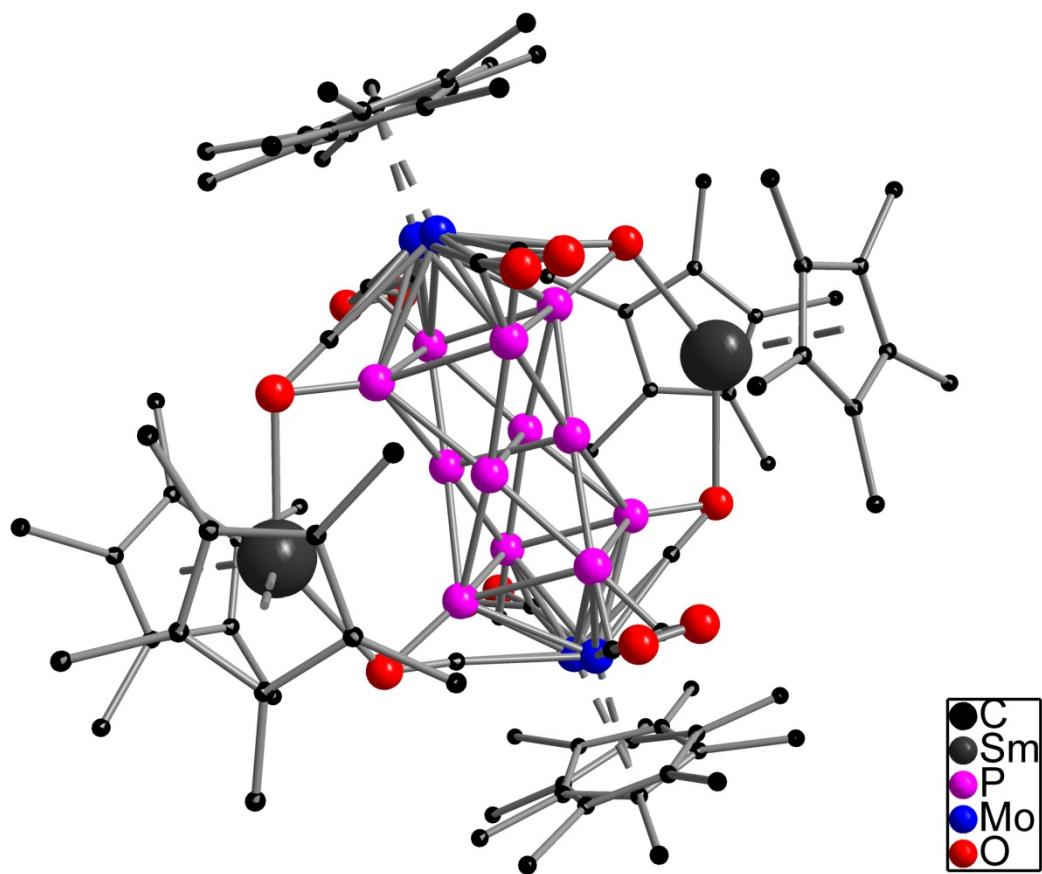


Figure S8. Solid-state structure of **4** showing the four-fold disordering of the central P₆-core only.

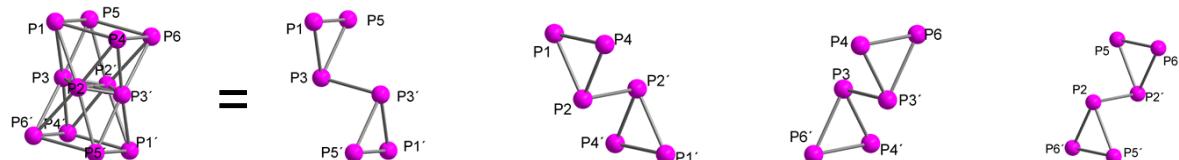


Figure S9. Temperature-dependent SQUID magnetization data for **4**. $\chi_M T$ versus T plot at 1000 Oe and M versus T plot.

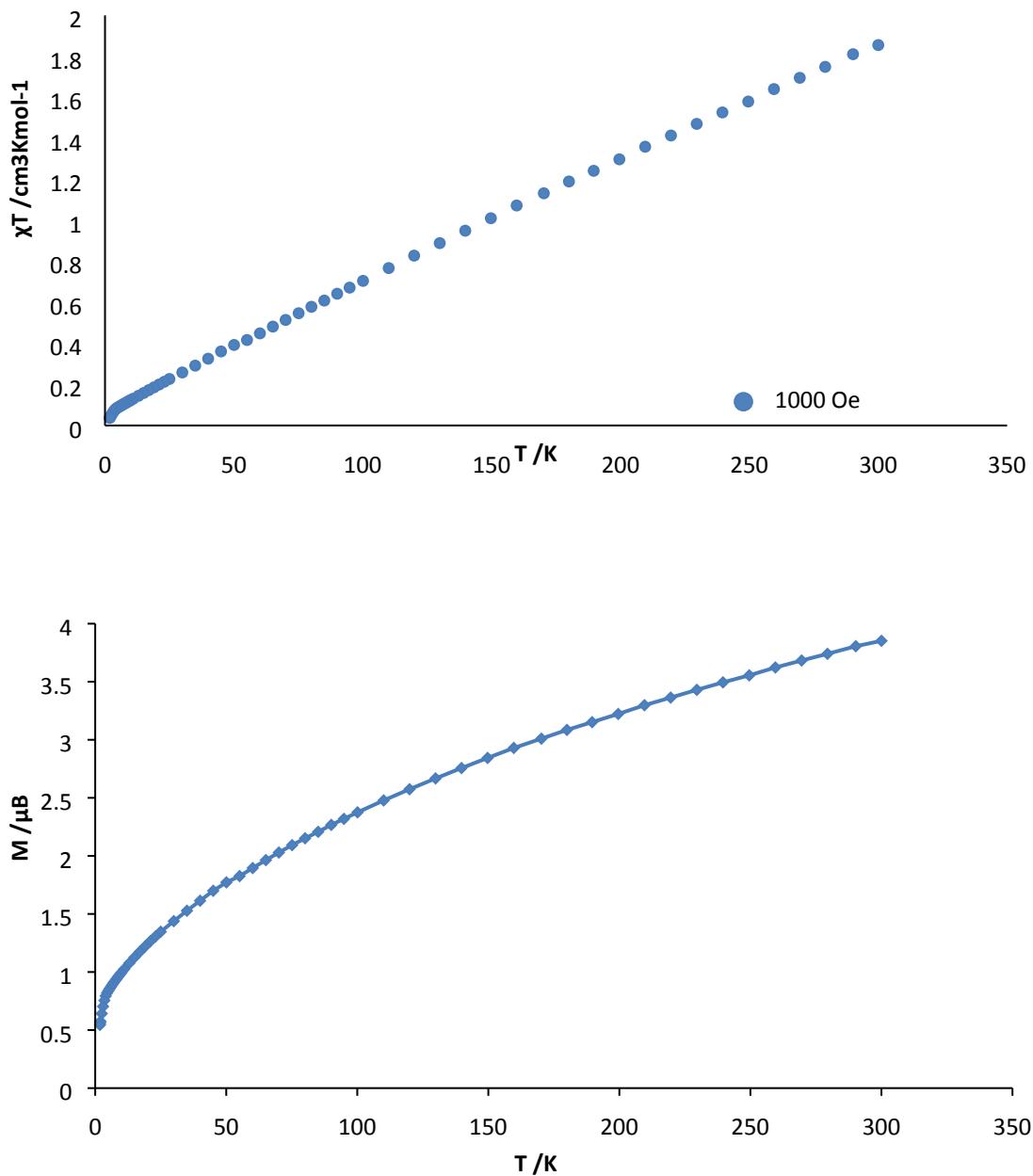


Figure S10. IR-spectrum of 4.

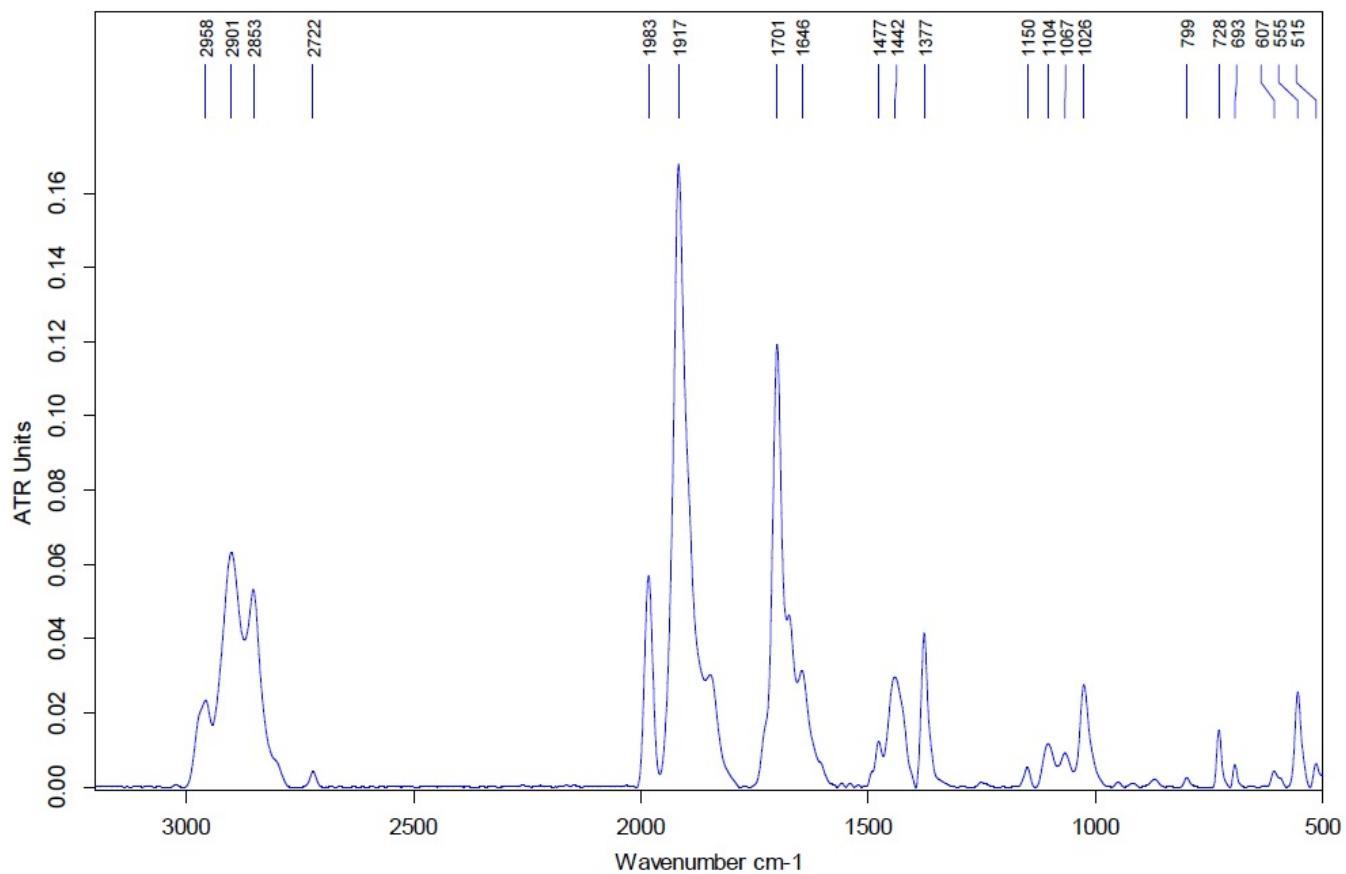


Figure S11. NIR-spectrum of 4.

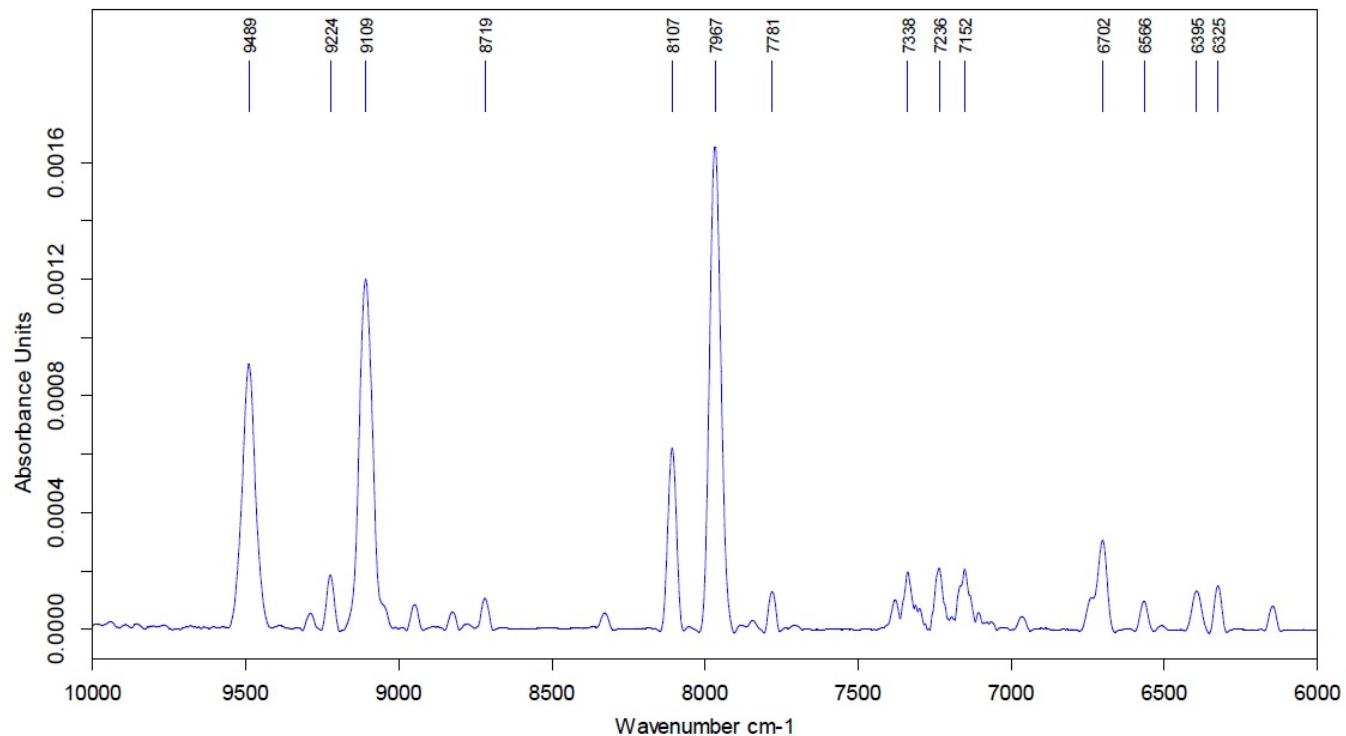
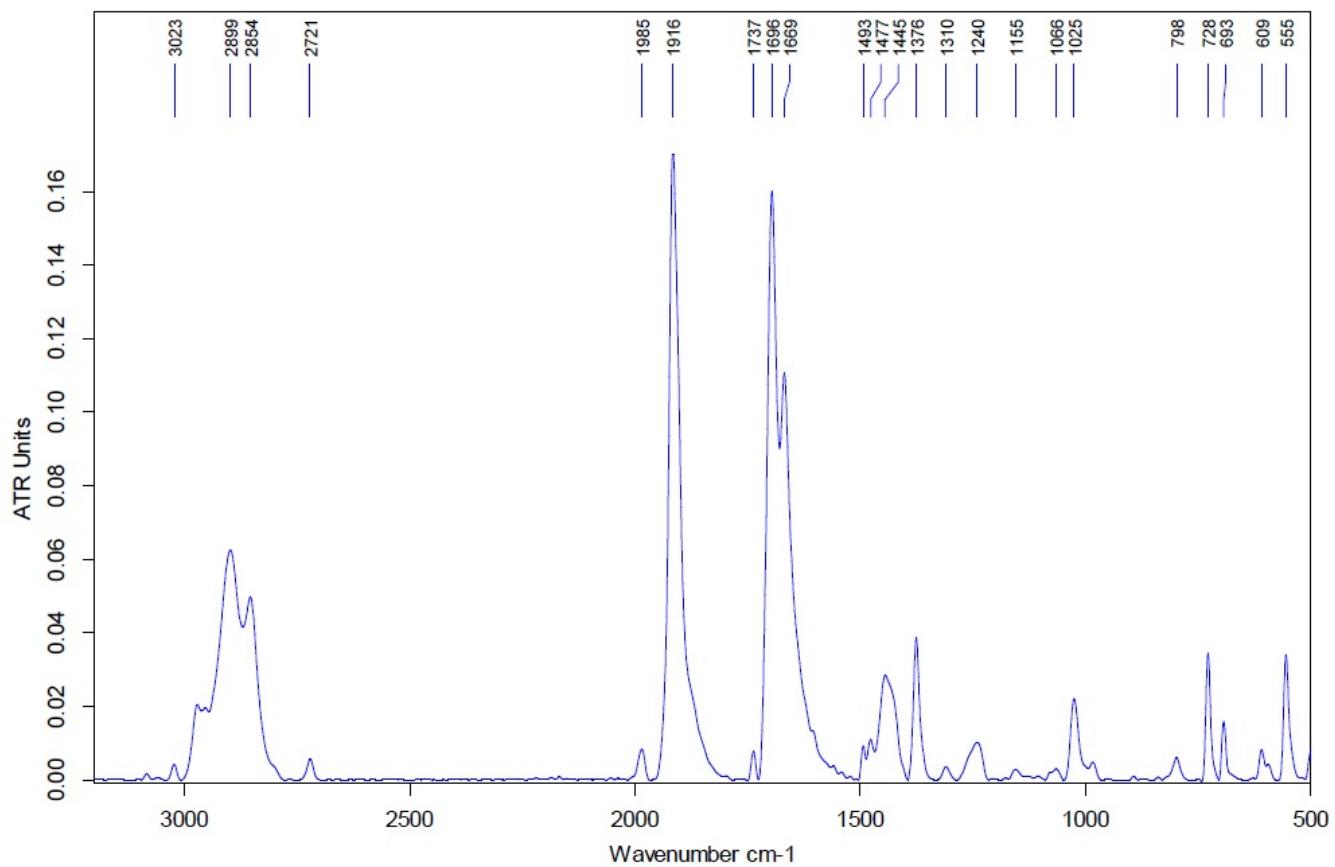


Figure S12. IR-spectrum of **5**.



Supplemental contribution/quantum chemical calculations

Table S1: Results of the quantum chemical calculations (distances given in Å; shared electron numbers SEN and partial charges Q determined by Ahlrichs-Heinzmann population analysis, each 10 MAOs chosen for Sm and Mo) on **1a**, **2**, **3**, and $[\{CpMo(CO)_2\}_2(\mu,\eta^{2:2}-P_2)]$.

	$[(Cp_2^*Ln)_2P_2]$ $(CpMo(CO)_2)_4$	$[(Cp^*_2Sm)_2P_4]$ $(CpMo(CO)_2)_2$	$[(Cp^*_2Sm)_3P_5]$ $(CpMo(CO)_2)_3$	$[\{CpMo(CO)_2\}_2(\mu,\eta^{2:2}-P_2)]$
	1a	2	3	
r(P-P)	2.078	2.181 (P1-P2, P1'-P2'), 2.328 (P1-P2', P1'-P2)	2.205 (P1-P5, P4-P5), 2.151 (P3-P4), 2.174 (P1-P2), 2.278 (P2-P3)	2.121
r(Mo-P)	2.406	2.568 (Mo'-P1', Mo- P1) 2.588 (Mo-P1', Mo'- P1)	2.394 (Mo2'-P5) 2.663 (Mo2-P4) 2.630 (Mo1-P1) 2.587 (Mo2-P3) 2.606 (Mo1-P2)	2.505
r(Mo-Mo)	3.247			3.087
SEN(P-P)	1.38	1.24 (P1-P2, P1'-P2'), 0.98 (P1-P2', P1'-P2)	1.12 (P1-P5, P4-P5), 1.24 (P3-P4), 1.30 (P1-P2), 1.04 (P2-P3)	1.41
SEN(Mo-P)	0.77	0.59	0.66 (Mo2'-P5) 0.53 (Mo2-P4) 0.43 (Mo1-P1) 0.61 (Mo2-P3) 0.54 (Mo1-P2)	0.55
SEN(Mo-Mo)	0.23			0.30
SEN(Sm-P)			0.74 (Sm1-P3)	
Q(Sm)	1.04	0.38	Sm1: 0.35 Sm1': 1.21 Sm2: 1.10	
Q(Mo)	-0.52	-0.52	Mo1: -0.45 Mo2: -0.55 Mo2': -0.41	-0.54
Q(P)	0.02	0.03	P1: -0.20 P2: -0.08 P3: 0.00 P4: -0.01 P5: -0.08	0.07

Table S2: Results of the quantum chemical calculations (distances given in Å; shared electron numbers SEN and partial charges Q determined by Ahlrichs-Heinzmann population analysis) on the model compounds $\text{Na}_4\text{P}_2^{2+}$, Na_2P_4 , NaP_5 and $\text{Na}_3\text{P}_5^{2+}$ as well as T_{d} - and $\text{D}_{2\text{h}}\text{-P}_4$.

	$\text{Na}_4\text{P}_2^{2+}$	Na_2P_4	NaP_5	$\text{Na}_3\text{P}_5^{2+}$	P_4	P_4
symmetry	$\text{D}_{2\text{h}}$	$\text{D}_{4\text{h}}$	$\text{C}_{5\text{v}}$	C_s	T_{d}	$\text{D}_{2\text{h}}$
$r(\text{P-P})$	2.063	2.168	2.147	2.112/2.122/213.0	2.233	207.9/235.2
SEN(P-P)	1.74	1.44	1.38	1.43/1.38/1.33	1.08	1.66/0.96
Q(Na)	0.56	0.40	0.54	0.69		
Q(P)	-0.12	-0.20	-0.11	-0.01	0.00	0.00

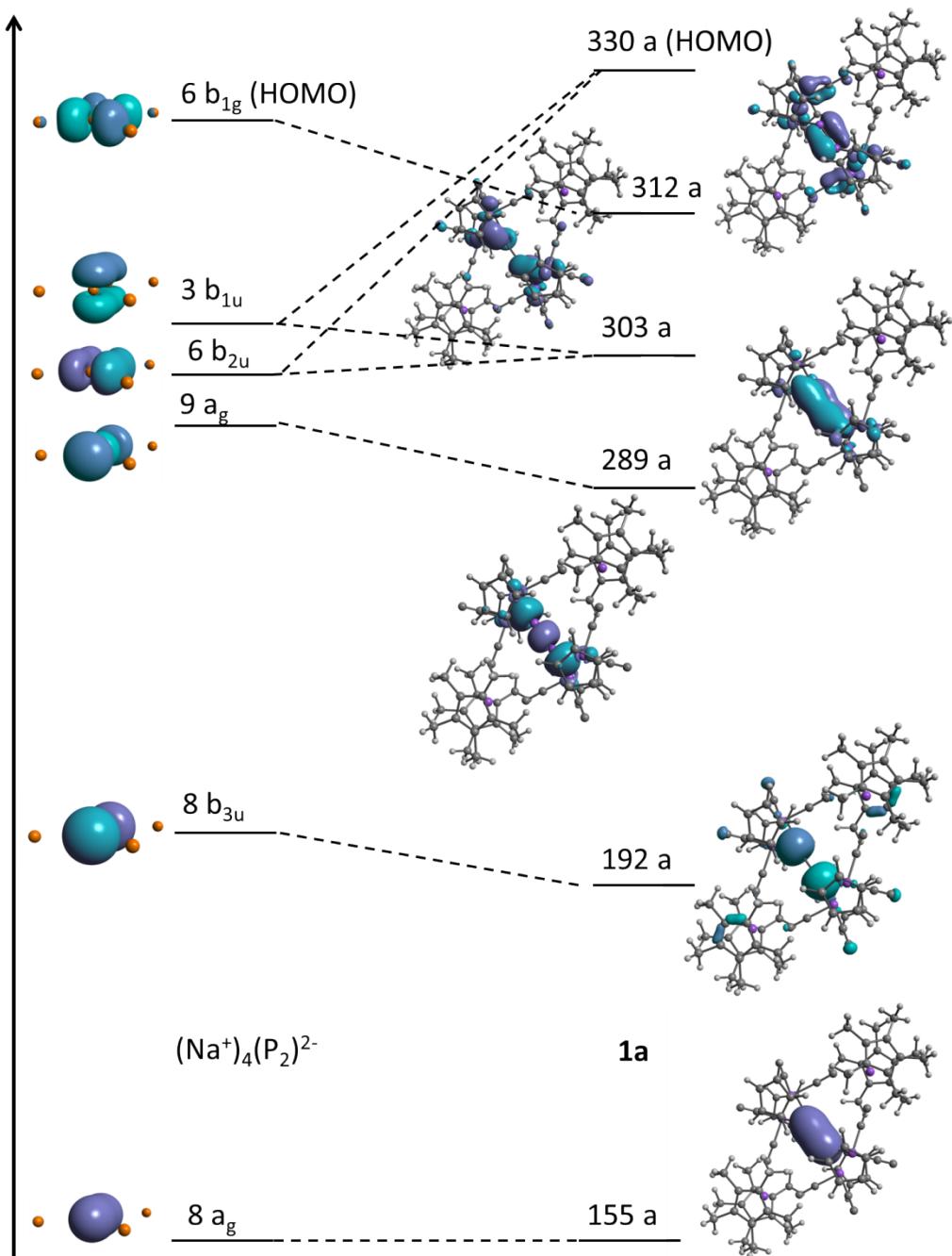


Figure S13: Comparison of the MO diagrams of $\text{Na}_4\text{P}_2^{2+}$ and $[(\text{Cp}_2^*\text{Ln})_2\text{P}_2(\text{CpMo}(\text{CO})_2)_4]$. Only the MOs with strong phosphorus contribution are shown.

Cartesian Coordinates of the calculated Molecules (given in a.u.)

$[(\text{Cp}_2^*\text{Ln})_2\text{P}_2(\text{CpMo}(\text{CO})_2)_4]$ (**1a**)

1.90296250200252	-13.82810402990115	8.09534720152958 h
0.08122360005252	-14.84040850928407	10.74781306916709 h
1.37981219598989	-13.32253036468599	10.06893260714207 c
0.14023700628127	-10.76210982108238	10.25462714082543 c
3.12034011868955	-13.44567223958481	11.22950563951258 h
2.10081361552728	-6.38499874179512	8.10803247395473 sm
0.36366432516577	-9.00393562278579	12.32149153544817 c
-1.79623458133009	-9.84804709039902	8.58549018910282 c
0.74308860372980	-6.52505761835600	3.74200396442685 o
0.55006996143110	-2.08250390095479	7.85773233485902 o
-2.76637565189906	-7.53050742569598	9.60853311748393 c
-1.42366485650434	-6.99774888538406	11.90959369437644 c
6.59214457568642	-4.82088920278127	6.05034533375946 c
6.84055093745446	-7.51591578600646	6.30229819199671 c
7.00357581757021	-8.09883668028852	8.95233193438237 c
6.82253254791578	-5.76266175750081	10.33650701072655 c
6.58097664707015	-3.74159490644463	8.53832573999748 c
1.83565304463779	-9.42455544331244	14.72432686216367 c
-2.81897132871851	-11.23170169726158	6.31416826412209 c
-0.25732801015706	-6.08500904503992	1.71864829632883 c
-0.29226672059125	-0.25195397447684	6.75040026413424 c
-4.96436167458927	-6.05287487093342	8.56357474596062 c
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7.12839272145268	-9.37755587432994	4.16086286352399 c
7.73553347801920	-10.62939919800644	10.03750387187459 c
7.25195465521446	-5.41403176674349	13.13335852581672 c
6.58115131985332	-0.96191815085404	9.15814758903328 c
3.55942517508755	-10.58232404058777	14.41696181079272 h
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-1.34427270025411	-12.36323561746168	5.33458984730631 h
-1.91568940289937	-5.60200474798470	-1.42106077117737 mo
-1.86859755728729	2.54827595541933	5.16189297209614 mo
-5.08695350119064	-4.14163997599047	9.41961247870102 h
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6.55298694373891	-11.32028788671585	4.71748877880993 h
9.13242693655519	-9.50384049182191	3.51487395309791 h
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9.24342660176124	-4.83775556290041	13.52605171047707	h
6.90507271374498	-7.17504396761900	14.21603744047348	h
5.59332651534314	0.16736174142674	7.69033787254221	h
8.54919290548709	-0.21507984731932	9.29863488568231	h
5.64336123749621	-0.54738684213067	10.99147788253720	h
-0.37336578120060	-1.32658613395400	-1.39778186147951	p
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-5.70292440594707	-5.32559501850061	-4.08933284740734	c
-5.06651994666469	-7.92298396423230	-3.68808065003744	c
-5.29697994873109	-8.45177058391237	-1.03453276906739	c
-6.05410016832481	-6.15324573046095	0.20090116071225	c
-6.30232940601505	-4.23070614260374	-1.68957415766638	c
0.37390963257701	1.32717440862281	1.39718821307288	p
0.40248772618451	4.58498746178593	7.30311445093421	c
-5.93474374175377	0.78423625099102	5.64824892736753	c
-5.98873817891798	2.04326321832465	3.25311600627792	c
-5.68901988945465	4.69548968816598	3.69117336133280	c
-5.41236288820401	5.08229006229943	6.35371190113451	c
-5.58209716311298	2.66522472421912	7.57857496158817	c
1.86926100357004	-2.54829277181908	-5.16242235255147	mo
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-5.72360801700738	-4.35436413048413	-5.91987541741338	h
-4.54310671283884	-9.28151231298895	-5.16670795392077	h
-5.02629672794088	-10.29846838145748	-0.12931314263630	h
-6.45821912767175	-5.93180734092210	2.22340945235708	h
-6.89276290639923	-2.27003143806304	-1.36757853678393	h
1.91625972857264	5.60240500095467	1.42106562103362	mo
1.57064127063006	5.76092163738548	8.78295281240271	o
-6.17626447278180	-1.25107508569177	5.95957150201203	h
-6.22614812457104	1.13745580473912	1.40487390580314	h
-5.65967454996688	6.18122038928901	2.24703281888747	h
-5.16645012994183	6.91652767021470	7.29269202236378	h
-5.53066948249937	2.32734779808104	9.62490994822418	h
0.29311449176665	0.25215415116823	-6.75104963254254	c
-0.40309564391092	-4.58468839903970	-7.30278689052908	c
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5.68890244982821	-4.69644897923053	-3.69127527129291	c
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0.25799073919080	6.08463281805843	-1.71911322866396	c
-0.66957176789995	7.98204277945500	2.64576721194848	c
5.70282270314661	5.32620953865269	4.09006376245296	c
5.06628270969133	7.92357447591039	3.68876176637606	c
5.29716035931690	8.45251994175472	1.03530758528642	c
6.05468989662013	6.15416252871580	-0.20013604499202	c
6.30276029782197	4.23147474341001	1.69029690046745	c
-0.54957223413490	2.08274826458711	-7.85807372966993	o
-1.57179220114653	-5.76056838430821	-8.78212473148298	o

6.17793788316183	1.24993936631710	-5.96033706527028 h
6.22677671457305	-1.13835310237382	-1.40514387517116 h
5.65915712896344	-6.18192817510887	-2.24682670013210 h
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5.53193540433425	-2.32894590646508	-9.62549420250546 h
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4.54240698751334	9.28205915242516	5.16744961324922 h
5.02638666915973	10.29938084733994	0.13005096798519 h
6.45906147050896	5.93277215460938	-2.22280747662307 h
6.89341860672274	2.27070079230659	1.36830460245046 h
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2.76614144127141	7.53098368713462	-9.60889743278575 c
1.42320903211511	6.99851105332494	-11.90993324383672 c
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1.79607139793305	9.84841659492585	-8.58550841047729 c
-6.59191174566873	4.82033112148266	-6.05040421581223 c
-6.84094148195079	7.51531918153094	-6.30205804266289 c
-7.00408859159478	8.09852704029329	-8.95202257205999 c
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4.96423776096487	6.05320550978775	-8.56425568069928 c
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-1.83632423329785	9.42562018246386	-14.72411479486987 c
-1.38032300596018	13.32295438767629	-10.06814359031350 c
2.81895189256206	11.23177961133498	-6.31403955328169 c
-6.53781176038071	3.36407253980913	-3.60658552388110 c
-7.12902831879112	9.37663393739735	-4.16038642741704 c
-7.73629760057505	10.62915815289970	-10.03688652471535 c
-7.25183876636626	5.41418014284663	-13.13337027640777 c
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5.08684497076526	4.14200725394686	-9.42061692496857 h
6.79593262234667	7.02280535511729	-8.95606115924951 h
4.84022020214490	5.80210051710574	-6.47930397750890 h
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3.47994366237748	5.35055980641543	-15.07935754625170 h
2.58587367304038	3.11069183881199	-12.70691582278593 h
-3.56001766453966	10.58346413920892	-14.41653718081293 h
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-0.08187863753905	14.84113998613990	-10.74665533331277 h
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3.63402295700814	9.91940128095827	-4.89280173735528 h
4.34940103982467	12.57475912703138	-6.86635164355566 h
1.34429562017933	12.36340188977283	-5.33437437734869 h
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-8.45800300991292	2.65162556186378	-3.09798048981657 h
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-6.55401324123169	11.31955529324756	-4.71683204732224 h

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-7.07401260858620	10.87830032423972	-12.01307262599672 h
-9.83489137131778	10.84642139808900	-10.07147179913171 h
-6.98515904246185	12.23576170423269	-8.91313371933781 h
-6.02732439039822	3.91672634489148	-13.95905188231414 h
-9.24315266968684	4.83749700284766	-13.52623217032422 h
-6.90530816509257	7.17541954784825	-14.21580023216561 h
-5.59197571049172	-0.16747210802977	-7.69097181745625 h
-8.54787551148369	0.21441058845315	-9.29925843037465 h
-5.64211267290813	0.54765252760091	-10.99200879312267 h

[(Cp^{*})₂Sm)₂P₄(CpMo(CO)₂)₂] (2)

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9.81688776807678	19.77196364085793	1.54388458722821 c
8.00823216405194	23.88351186865954	2.13875419911116 mo
9.87633329128720	20.41903412356979	4.18100153761666 c
11.46419411373965	21.45831249219117	0.22342883433466 c
8.73346967779394	28.56355955552543	0.91914409625797 p
6.64047391697562	26.31768012858825	-1.83028146273638 p
12.54141891386481	23.16580299137826	2.02865863471603 c
11.57065258791290	22.52498659228827	4.47390202461685 c
4.64066896627686	22.40228518809662	1.61978178874903 c
6.52968710989899	25.53319691493463	4.94649241275463 c
8.88408077270788	19.42342792120186	5.70601574068083 h
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5.00216037799461	30.71008440202762	1.83069428221617 p
2.90942264624504	28.46333594150157	-0.91898555158862 p
13.90030611545894	24.67777684771742	1.61801582121453 h
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3.71053307478081	30.68204083581789	7.47785347340809 sm
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0.09967110742449	27.77718347827180	9.88794909898787 c
8.14240458450512	33.58272686780669	7.39591458643397 c
6.89758279764969	32.08082822941581	11.33673367131459 c
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6.19048413358778	35.22060467918927	8.34403426882082 c
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0.90166795088517	25.68157745109736	11.64738398702556 c
9.76414965025097	34.08834291217484	5.11192159243319 c
6.93434088349721	30.63634161273187	13.79644083563478 c
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3.28434108760399	34.64886927373280	14.29529357361863 h
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1.03745899896233	27.37322950224246	-9.11336781251168 c
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7.74509879999501	21.21299571319001	-12.66680897597638 c
13.30662849302046	21.52189049098325	-9.88349733198750 h
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6.29684955618637	19.50752651925423	-5.02082011256975 h
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[(Cp₂Sm)₃P₅(CpMo(CO)₂)₃] (3)

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3.25571825909646	22.57886551395781	-0.64844842778809 o
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[{CpMo(CO)₂}₂(μ,η^{2:2}-P₂)]

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-3.68984332704307	-2.51685562417895	0.16206375272972 c
1.83521143869115	6.63732163707562	5.39352607806425 o
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5.79697181476800	2.43138898495034	-0.51452909616275 o
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Na₄P₂²⁺

1.94929529295146	0.00000000000000	0.00000000000000 p
-1.94929529295146	0.00000000000000	0.00000000000000 p
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