

Supporting information

Table S1: Parameters used in the force field for describing the aldehyde and peroxide groups in the lipid bilayer, taken from [1]. The functional form for the dihedral angles is: $E = a(1+\cos(x-b)) + c(1+\cos(2x-d)) + e(1-\cos(3x-f))$. Columns indicated by * are in kJ.mol^{-1} , columns indicated by † are in degrees.

Bonds						
Bond	r_{ij} (nm)			K_{ij}^b (kJ.mol ⁻¹ .nm ⁻²)		
C-O (peroxide)	0,14180			225670		
O-O (peroxide)	0,14430			269580		
O-H (peroxide)	0,09810			44130		
Angles						
Angle	α_0 (graden)			K_{ijk}^θ (kJ.mol ⁻¹ .rad ⁻²)		
=C-C-O (peroxide)	104,00			418,40		
C-C-O (peroxide)	109,50			418,40		
C-O-O (peroxide)	105,90			598,37		
O-O-H (peroxide)	100,00			506,92		
Dihedral angles						
Dihedral angle	a*	b [†]	c*	d [†]	e*	f [†]
C=C-C-O (peroxide)	2,12	223,90	0	0	3,62	180,50
C-C-O-O (peroxide)	2,13	334,25	0	0	7,04	8,10
C-O-O-H (peroxide)	8,46	23,30	6,51	18,40	0	0
C-C-C=O (aldehyde)	0,47	180,00	1,58	180,00	2,67	180,00
Partial charges						
Functional group	CH	O		O	H	
Hydroperoxide	0,30	-0,30		-0,45	0,45	
Aldehyde	0,53	-0,53		-	-	

The parameters were derived using quantum chemical calculations, with the B3LYP method of density functional theory [2,3] and the LACV3P**++ basis set [4]. Partial atomic charges were estimated using natural population analysis [5] and the electrostatic surface potential fitting method with Merz-Kollman atomic radii [6] after the geometry optimization. For the calculation of bond and angle force constants, we restrained the bond lengths and angles at seven different values, then fitted a harmonic potential function to the energy profile. For dihedral parameters, dihedral angles were restrained at

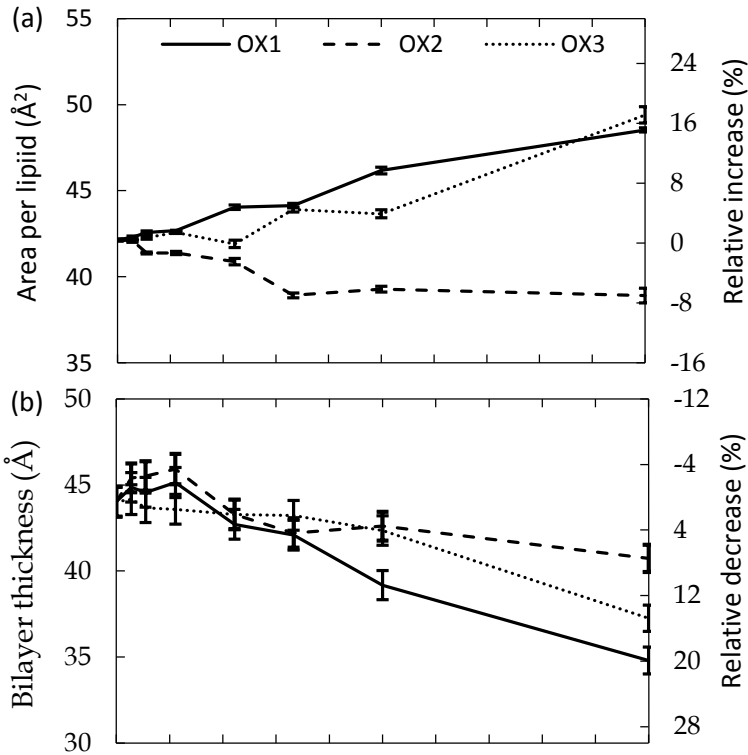
36 different values from 0 to 360°, and the standard proper dihedral function was fitted to the potential energy. For all bonded parameters, the Lennard-Jones and electrostatic energy were calculated for different geometries and subtracted from the total energy before fitting.

Table S2: Comparison between our calculated data and experimental data from literature, for a pure POPC bilayer.

	Calculated value	Experimental data from literature [7-9]
Surface area per lipid (Å ²)	63.5 ± 0.4	63.0 – 64.3
Thickness of the bilayer (Å)	38.6 ± 0.6	37.5 – 39.1

Table S3: Calculated values for the POPC bilayer, without and with cholesterol (concentration of 50%).

	POPC	POPC/Cholesterol
Surface area per lipid (Å ²)	63,5 ± 0,4	42,1 ± 0,1
Bilayer thickness (Å)	38,6 ± 0,6	44,0 ± 0,6
Average S _{CD}	0,176	0,347



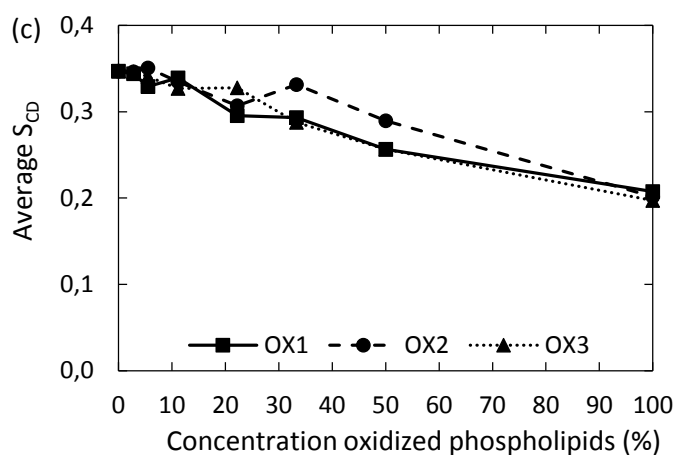


Figure S1: Surface area per lipid (a), thickness of the bilayer (b) and average deuterium order parameter (c), as a function of the concentration of the oxidized phospholipids, for three types of oxidation products, for the model systems with 50% cholesterol.

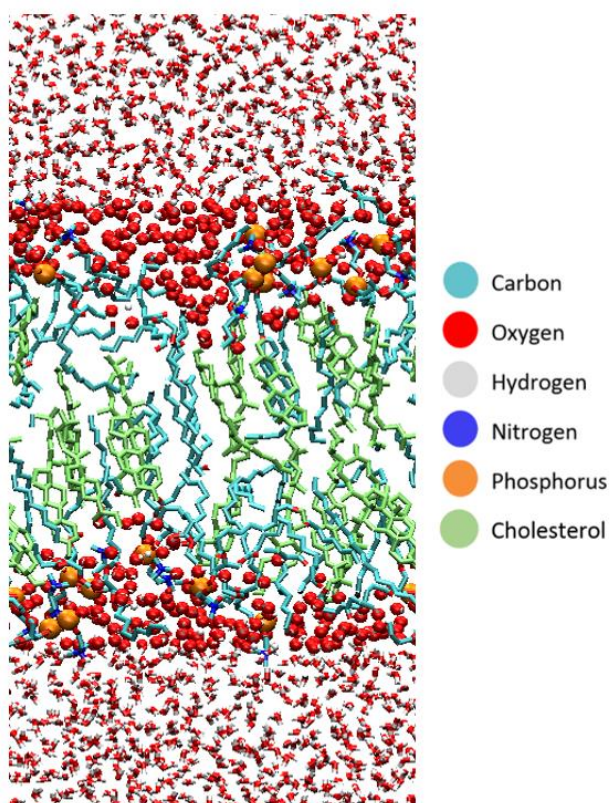


Figure S2: Structure of the bilayer with 50% POPC and 50% cholesterol, and 100% oxidation of the phospholipids, showing that no pore formation occurs.

References

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