

Supporting Information

Ultra large optical modulation of electrochromic porous WO₃ film and the local monitoring of redox activity

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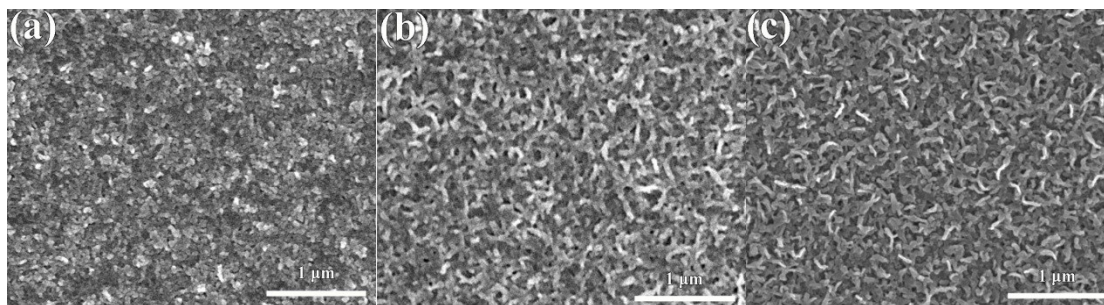


Fig. S1 SEM images of the WO_3 films prepared by pulsed deposition with different duty cycle length (a) 4 s, (b) 1 s, (c) 0.5 s.

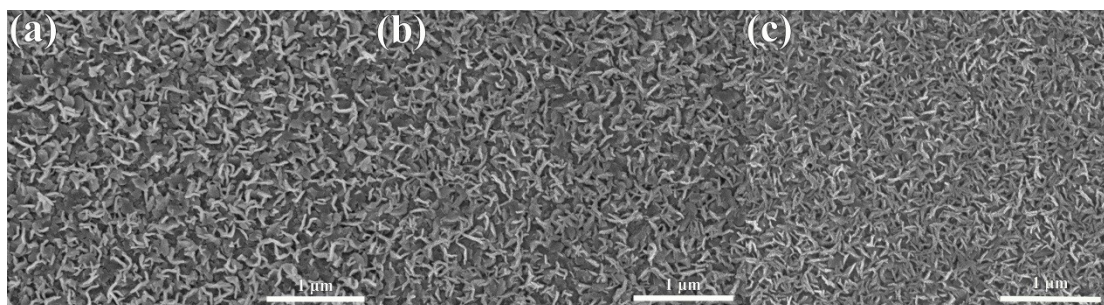


Fig. S2 SEM images of the WO_3 films prepared by pulsed deposition with different interval time (a) 0.4 s, (b) 0.6 s, (c) 0.8 s.

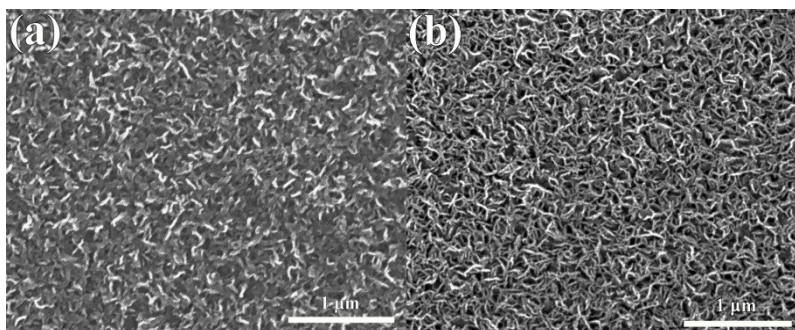


Fig. S3 SEM images of the porous WO_3 films deposited on different substrates (a) FTO-coated glass, (b) ITO-coated PET.

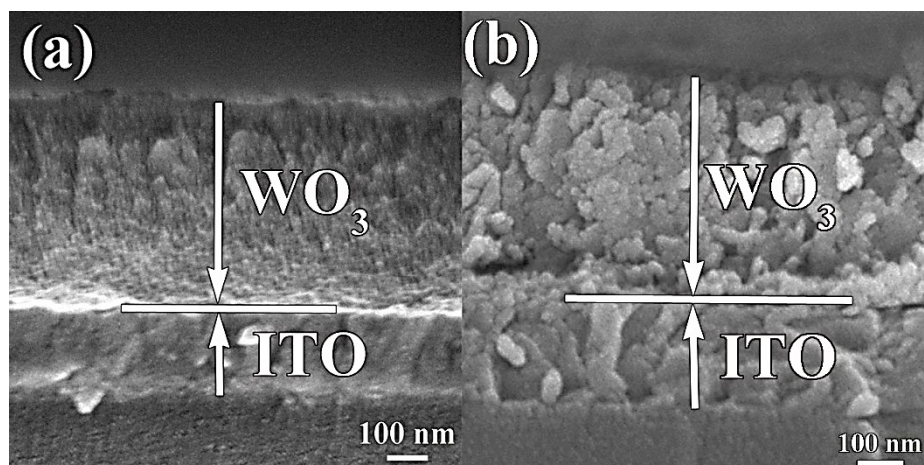


Fig. S4 The cross-section morphologies of (a) compact and (b) porous WO₃ film.

Table S1 Chemical state and composition of W and O obtained by deconvolution of the corresponding region spectra of porous WO₃ film.

Elements	Peak Binding Energy (eV)	Composition (At%)
W4f WO ₃	35.9	25%
	38.0	
O1s O-W	531.0	75%
O/W	—	3.0

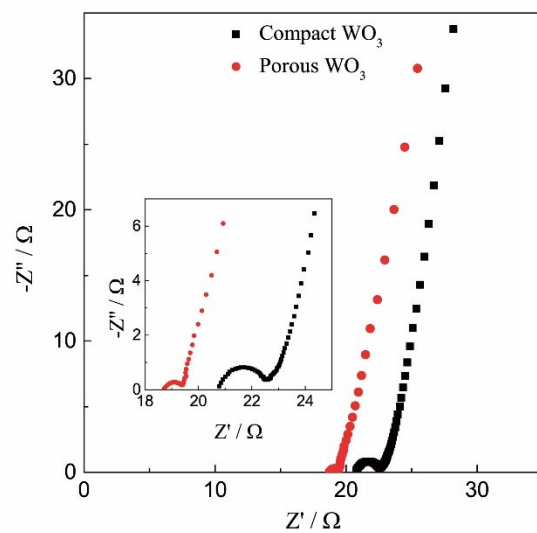


Fig. S5 Nyquist plots of compact and porous WO_3 film, the inset shows the magnified high-frequency region.

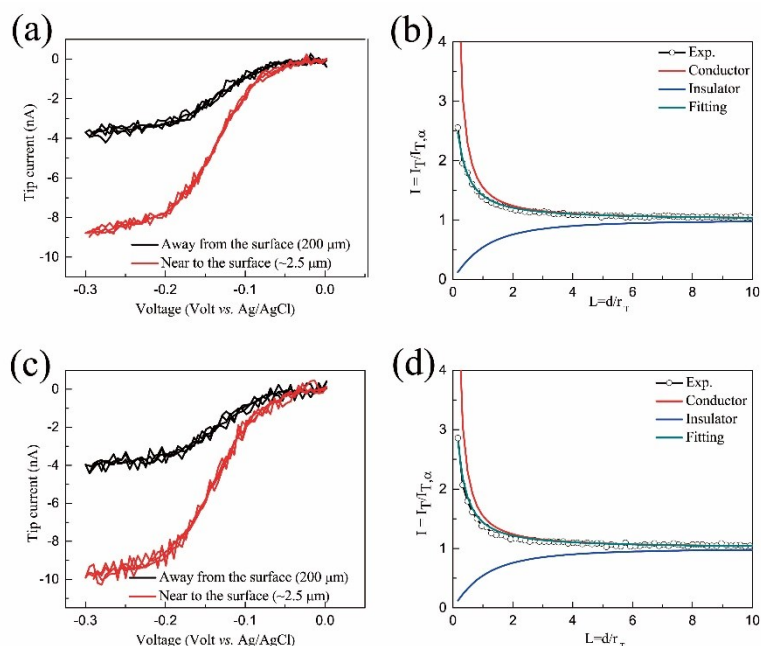


Fig. S6 The localized CVs and approach curves of Pt microelectrodes recorded on the surface of WO_3 electrodes $[\text{Ru}(\text{NH}_3)_6]^{3+}/[\text{Ru}(\text{NH}_3)_6]^{2+}$ redox mediator: (a-b) compact WO_3 film electrode, (c-d) porous WO_3 film electrode.

Effective heterogeneous charge transfer rate constants k of WO_3 films were calculated using the relation $k_{\text{eff}} = \kappa D / r_T$, where κ is the normalized heterogeneous rate constant which is obtained from the normalized approach curves I_T vs. L were fitted to the analytical approximation of Cornut and Lefrou.¹ The κ value of compact and porous WO_3 films are 3.91 and 5.19 obtained from Figure S5b and d, respectively. r_T is the radius of microelectrode (12.5 μm). D is the diffusion coefficient for redox species ($D = 8.4 \times 10^{-6} \text{ cm}^2/\text{s}$). The k_{eff} of porous WO_3 film is $3.49 \times 10^{-2} \text{ cm s}^{-1}$, which is higher than that of the compact WO_3 film ($2.63 \times 10^{-2} \text{ cm s}^{-1}$).

1. R. Cornut and C. Lefrou, *J. Electroanal. Chem.*, 2008, 621, 178-184.