## Supporting Information

## Recognition of extended linear and cyclised polyketide mimics by a Type II acyl carrier protein.

Xu Dong, Christopher D. Bailey, Christopher Williams, John Crosby, Thomas J. Simpson, Christine L. Willis* and Matthew P. Crump*

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## Supporting Spectra

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## Experimental Section

## General

Manipulations involving air-sensitive materials were carried out on a vacuum line under $\mathrm{N}_{2}(\mathrm{~g})$, employing standard Schlenk techniques. All glassware was flame-dried before use as standard. Dry solvents were purchased (Aldrich, Fluka) or obtained by passage through an Anhydrous Engineering drying column. Other reagents requiring purification where indicated were done so according to Purification of Laboratory Chemicals (D. D. Perrin \& W.L.F Armarego, 3rd Edition, Butterworth Heinemann, 1988). Solvents for extraction and chromatography were technical grade.

Purchased chemicals (Aldrich, Acros, Fluka) were used as received (unless otherwise stated). Flash chromatography was conducted using Fluorochem silica gel 60 ( $0.040-0.063$ ). Eluting solvent used as indicated in the text. TLC was conducted with 0.25 mm Merck silica gel 60 F254 on aluminium plates, using solvent systems indicated in the text, visualising at 254 nm and developed using standard $\mathrm{KMnO}_{4}$ dip with gentle heating.

All IR spectra were obtained as thin film using Perkin-Elmer Spectrum One apparatus; peaks are reported in $\mathrm{cm}^{-1}$ with the following intensities: s (strong, $70-100 \%$ ), m (medium, $30-70$ \%), w (weak, $1-30 \%$ ). NMR spectra were obtained from Jeol Eclipse ( 400 MHz ), Delta ( 270 MHz), Lambda 300 (MHz), Delta 400, Varian VNMRS 400 or Varian 500 instruments. ${ }^{1} \mathrm{H}$ NMR chemical shifts $\delta(\mathrm{ppm})$ are reported relative to residual solvent. ${ }^{13} \mathrm{C}$ NMR shifts $\delta(\mathrm{ppm})$ are reported relative to deuterated solvent. Multiplicities are indicated as $s$ (singlet), $d$ (doublet), $t$ (triplet), $q$ (quartet), quin (quintet), $b r$ (broadened), app. (apparent) or $m$ (multiplet, when multiplicity is complex). Coupling constants, $J$, are reported in Hz. MS (EI, CI, HRMS) were conducted by the University of Bristol Mass Spectrometry Service using Fisons Autospec instruments. HPLC was performed using a Gemini C18 reverse phase column (100x4.6mm, 5 micron, Phenomenex) using an ÄKTA Purifier (GE Healthcare) at a flow rate of $1 \mathrm{ml} / \mathrm{min}$.

For protein accurate mass measurement, samples were denatured prior to analysis by nano-ESI MS. Solutions were prepared using an existing protocol for simultaneous desalting and denaturation by interaction with C4 chromatographic resin ${ }^{1}$. Nano-ESI MS analyses were performed using a QSTAR XL (ABSciex) equipped with a NanoMate ${ }^{\mathrm{TM}}$ (Advion Biosciences) chip-based nano-ESI source.

5-Phenyl-3-hydroxy-pent-1-ene (4) 3-Phenylpropionaldehyde ( 0.67 ml ) was cooled to $0{ }^{\circ} \mathrm{C}$ under nitrogen then vinyl magnesium bromide ( 1.6 M in THF, 3.13 ml ) was added drop-wise. The mixture was stirred in an ice bath for ten minutes. Then, the reaction warmed to room temperature and stirred for two hours. The reaction was quenched by the addition of saturated ammonium chloride solution ( 10 ml ) and extracted with ethyl acetate. The organic layer was combined, dried over magnesium sulfate the solvent was removed in vacuo to give a yellow oil. The crude extract was purified by silica chromatography (EtOAc: Petrol 1:3) giving the product 4 as an oil $(0.46 \mathrm{~g}, 56 \%$ yield $) . \delta_{\mathrm{H}}\left(270 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.75-1.90\left(2 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}_{2}\right) 2.71(2 \mathrm{H}, \mathrm{t}, J 7.3$ $\left.\mathrm{Hz}, 5-\mathrm{H}_{2}\right) ; 4.12(1 \mathrm{H}, \mathrm{q}, J 6.5,3-\mathrm{H}) 5.14(1 \mathrm{H}, \mathrm{dd}, J 10.6,1,1-\mathrm{HH}) ; 5.25(1 \mathrm{H}, \mathrm{d}, J 17.0,1,1-\mathrm{H} H)$; $5.92(1 \mathrm{H}, \mathrm{ddd}, J 17.0,10.6,6.5 \mathrm{~Hz}, 2-\mathrm{H}) ; 7.19-7.26(5 \mathrm{H}, \mathrm{m}$, aromatic). Spectral data are in accord with the literature. ${ }^{2,3}$

5-Phenyl-3-oxopent-1-ene (5). Alcohol $4(0.23 \mathrm{~g})$ was dissolved in in DCM ( 5 ml ) under nitrogen at room temperature and Dess Martin periodinane ( $15 \% \mathrm{w} / \mathrm{w}$ in DCM, 4.7 ml ) was added. The solution slowly became cloudy as it was stirred at room temperature overnight. The solution was filtered through filter paper and the solvent removed in vacuo. The crude product was purified by silica chromatography (using ether: petrol) giving enone 5 as an oil ( $0.14 \mathrm{~g}, 61 \%$ yield). $\delta_{\mathrm{H}}\left(270 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.80-3.0\left(4 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}_{2}\right.$ and $\left.5-\mathrm{H}_{2}\right), 5.82(1 \mathrm{H}, \mathrm{dd}, J 17.5,10.2,2-\mathrm{H})$, $6.21(1 \mathrm{H}, \mathrm{dd}, J 17.5,1.5,1-H \mathrm{H}), 6.37(1 \mathrm{H}, \mathrm{dd}, J 17.5,1.5,1-\mathrm{H} H) 7.19-7.26\left(5 \mathrm{H}, \mathrm{m}\right.$, aromatic); $\delta_{\mathrm{C}}$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 29.9 (C-5), 41.3 (C-4), 126.2 (aromatic), 128.3 (aromatic), 128.4 (aromatic),
128.6 (aromatic), 136.6 (C-2), 140.8 (C-6), 199.6 (C-3). The spectroscopic data are in accord with the literature ${ }^{4}$.

Preparation of 5-phenyl-3-oxo-pentyl Coenzyme A (6). An aqueous solution of Coenzyme A (CoASH) tri-lithium salt ( $100 \mathrm{mM}, 50 \mu \mathrm{l}$ ) was mixed with a solution of enone $\mathbf{5}$ in acetone (100 $\mathrm{mM}, 75 \mu \mathrm{l})$ and vortexed for five minutes, followed by the addition of distilled water to make up the solution up to 1 ml . The solution was vortexed for a further five minutes and centrifuged for five minutes at 13000 rpm before purification by HPLC using conditions described in the general experimental details. This procedure was repeated until approximately 13.8 mg of the conjugate addition product 6 was obtained ( $60-70 \%$ yield). $\delta_{\mathrm{H}}\left(600 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right) 0.81\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 0.94$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}$ ), $2.45\left(2 \mathrm{H}, \mathrm{t}, J 6.5,6^{\prime}{ }^{\prime}-\mathrm{H}_{2}\right), 2.58\left(2 \mathrm{H}, \mathrm{t}, J 6.5,9^{\prime}{ }^{\prime}-\mathrm{H}_{2}\right), 2.67\left(2 \mathrm{H}, \mathrm{t}, J 6.5, \mathrm{SCH}_{2}\right), 2.78$ ( $2 \mathrm{H}, \mathrm{t}, J 6.5, \mathrm{SCH}_{2} \mathrm{CH}_{2}$ ), 2.84-2.85 (4H, m, $\left.\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{Ph}\right), 3.30\left(2 \mathrm{H}, \mathrm{t}, J 6.5,8^{\prime}-\mathrm{H}_{2}\right) 3.46(2 \mathrm{H}, \mathrm{t}, J$ $\left.6.5,5^{\prime}{ }^{\prime}-\mathrm{H}_{2}\right), 3.61\left(1 \mathrm{H}, \mathrm{d}, J 9.4,1^{\prime}{ }^{\prime}-H \mathrm{H}\right), 3.85\left(1 \mathrm{H}, \mathrm{d}, J 9.4,1^{\prime \prime}-\mathrm{H} H\right), 4.02\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime \prime}-\mathrm{H}\right), 4.26$ $\left(2 \mathrm{H}, \mathrm{m} 5^{\prime}-\mathrm{H}_{2}\right), 4.61\left(1 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}\right), 4.86-4.88\left(2 \mathrm{H}, \mathrm{m}, 2^{\prime}-\mathrm{H}, 3^{\prime}-\mathrm{H}\right), 6.20\left(1 \mathrm{H}, \mathrm{d}, J 5.5,1^{\prime}-\mathrm{H}\right) 7.2-$ $7.31(5 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 8.41(1 \mathrm{H}, \mathrm{s}, 2-\mathrm{H}), 8.67(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H}) ; \delta \mathrm{c}\left(150 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right) 20.9\left(\mathrm{CH}_{3}\right), 23.59$ $\left(\mathrm{CH}_{3}\right)$, $27.5\left(\mathrm{SCH}_{2}\right), 31.9\left(\mathrm{CH}_{2}\right), 33.2(\mathrm{C}-9$ '' $), 38.1(\mathrm{C}-6$ '’), $38.17(\mathrm{C}-5$ '’), $41.27(\mathrm{C}-8$ '’), 45.0 $\left(\mathrm{CH}_{2}\right), 46.2\left(\mathrm{CH}_{2}\right), 67.81(\mathrm{C}-5 '), 74.7(\mathrm{C}-1$ '’), $76.9(\mathrm{C}-3$ '’), $77.1(\mathrm{C}-2$ ', $\mathrm{C}-3$ '), $86.5(\mathrm{C}-4$ '), 90.20 (C-1'), 121.8 (C5), 128.9-131.3 (ArC), 145.2 (C-8), 147.40 (C-2), 147.8 (C-6), 151.4 (C-4), 176.7 (C-7'’), 178.1 (C4’’), 217.5 (CO). ESMS m/z observed, 928.10, calculated 927.75. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ assignments of the CoASH portion were checked for consistency with that of Pal and Bearne ${ }^{5}$.

3-(tert-Butyldimethylsilyloxy)bromobenzene (8). Commercially available 3-bromophenol (3.46 $\mathrm{g}, 20 \mathrm{mmol}, 1 \mathrm{eq}$.), imidazole ( $1.5 \mathrm{~g}, 22 \mathrm{mmol}, 1.1 \mathrm{eq}$.) and DMAP ( 5 mg , cat.) were dissolved in dry $\mathrm{DCM}(80 \mathrm{ml})$ under $\mathrm{N}_{2}$ and cooled to $0^{\circ} \mathrm{C}$. A solution of tert-butyldimethylchlorosilane (3.3 $\mathrm{g}, 22 \mathrm{mmol}, 1.1 \mathrm{eq}$. .) in dry $\mathrm{DCM}(20 \mathrm{ml})$ was added slowly. The reaction was warmed to RT and stirred for 2 h and progress followed by TLC. Water ( 100 ml ) was added and the phases
separated. The organic phase was washed with sat $\mathrm{NH}_{4} \mathrm{Cl}(100 \mathrm{ml})$, dried over magnesium sulfate and the solvent removed in vacuo. The crude product was purified by column chromatography $(100 \% \mathrm{PE})$ to yield the title compound $\mathbf{8}$ as a colourless oil ( $5.12 \mathrm{~g}, 89 \%) . R_{\mathrm{f}} 0.45(100 \% \mathrm{PE}) ; \delta_{\mathrm{H}}$ ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $0.21\left(6 \mathrm{H}, \mathrm{s}, \mathrm{SiMe}_{2}\right), 0.99\left(9 \mathrm{H}, \mathrm{s}, \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 6.77(1 \mathrm{H}, \mathrm{m}, \mathrm{ArCH}), 7.02(1 \mathrm{H}$, $\mathrm{m}, \mathrm{ArCH}), 7.09(2 \mathrm{H}, \mathrm{m}, \mathrm{ArCH})$. Spectroscopic data are in accord with that published in literature ${ }^{6,7}$.

4-(3-tert-Butyldimethylsilyloxyphenyl)but-3-yn-1-ol (9). Bromide $\mathbf{8}$ (1 g, $3.5 \mathrm{mmol}, 1 \mathrm{eq}$.$) and$ 3-butyn-1-ol ( $0.33 \mathrm{ml}, 4.4 \mathrm{mmol}, 1.25$ eq.) were dissolved in $\mathrm{Et}_{3} \mathrm{~N}(10 \mathrm{ml})$ and cooled to $0{ }^{\circ} \mathrm{C}$ under nitrogen. $\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{PdCl}_{2}$ ( $74 \mathrm{mg}, 0.1 \mathrm{mmol}, 3 \mathrm{~mol} \%$ ) and $\mathrm{CuI}(13.5 \mathrm{mg}, 0.07 \mathrm{mmol}, 2 \mathrm{~mol} \%$ ) were added and the solution stirred for 5 mins at $0^{\circ} \mathrm{C}$. After this time the reaction was heated 70 ${ }^{\circ} \mathrm{C}$ for 16 h . The reaction was cooled to RT , $\mathrm{Et}_{2} \mathrm{O}(100 \mathrm{ml})$ was added and the organic phase washed with water $(100 \mathrm{ml})$. The aqueous phase was further extracted with $\mathrm{Et}_{2} \mathrm{O}(100 \mathrm{ml})$ and the combined organic extracts washed with brine $(100 \mathrm{ml})$, dried over magnesium sulfate, filtered and the solvent removed in vacuo to yield a brown oil which was purified by column chromatography (1:1 PE:EtOAc) to yield alcohol 9 as an orange oil ( $0.833 \mathrm{~g}, 86 \%) . \mathrm{R}_{\mathrm{f}} 0.5(1: 1$ PE:EtOAc $) ; ~ \delta_{H}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.15(1 \mathrm{H}, \mathrm{t}, J 8.0, \mathrm{ArCH}), 7.02(1 \mathrm{H}$, app. dt, $J 8.0,1, \mathrm{ArH})$, $6.90(1 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 6.79(1 \mathrm{H}, \mathrm{ddd}, J 8.0,2.5,1, \mathrm{ArH}), 3.82\left(2 \mathrm{H}\right.$, app. q, $\left.J 6,1-\mathrm{H}_{2}\right), 2.70(2 \mathrm{H}, \mathrm{t}, J$ $\left.6,2-\mathrm{H}_{2}\right), 1.82(1 \mathrm{H}, \mathrm{t}, J 6, \mathrm{OH}), 0.99(9 \mathrm{H}, \mathrm{s}, \mathrm{tBu}), 0.20\left(6 \mathrm{H}, \mathrm{s}, \mathrm{SiMe}_{2}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$, 155.4 (C), 129.3 (CH), 124.9 (CH), 124.3 (CH), 123.2 (C), 120.3 (CH), 86.0 (C), 82.4 (C), 61.1 $\left(\mathrm{CH}_{2}\right), 25.6\left(\mathrm{CH}_{3}\right), 23.8\left(\mathrm{CH}_{2}\right), 18.2(\mathrm{C}),-4.4\left(\mathrm{CH}_{3}\right)$; IR $\left(\mathrm{cm}^{-1}\right), 3345(\mathrm{br}, \mathrm{w}), 2955(\mathrm{w}), 2930(\mathrm{w})$, 2859 (w), 1596 (m), 1574 (m), 1478 (m), 1291 (m), 1253 (m), 1192 (m), 876 (m), 828 (s), 779 (s); HRMS: $m / z(\mathrm{CI})$, calculated for $\mathrm{C}_{16} \mathrm{H}_{25} \mathrm{O}_{2} \mathrm{Si} 277.1624[\mathrm{M}+\mathrm{H}]^{+}$, found $277.1618[\mathrm{M}+\mathrm{H}]^{+}$

4-(3-tert-Butyldimethylsilyloxyphenyl)butan-1-ol (10). Alcohol 9 (554 mg, $2 \mathrm{mmol}, 1 \mathrm{eq}$.$) and$ $10 \% \mathrm{Pd} / \mathrm{C}(55 \mathrm{mg}, 10 \mathrm{wt} \%)$ were added to ethanol $(20 \mathrm{ml})$ and stirred. Dissolved gases were removed by vacuum and $\mathrm{H}_{2}$ introduced by balloon through a septum. The reaction was stirred
under a balloon pressure of $\mathrm{H}_{2}$ for 2 h . Note: The reaction progression could not be followed by TLC as no $\mathrm{R}_{\mathrm{f}}$ difference between SM and product so a small aliquot can be removed and analysed by ${ }^{1} \mathrm{H}$ NMR. On completion the flask was placed under vacuum and nitrogen introduced before being filtered carefully through Celite, ensuring that the $\mathrm{Pd} / \mathrm{C}$ was not allowed to dry out. The filter cake was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and the filtrate concentrated in vacuo to yield alcohol $\mathbf{1 0}$ as a yellow oil ( $540 \mathrm{mg}, 96 \%$ ). $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.15-7.11(1 \mathrm{H}, \mathrm{m}, \mathrm{ArCH}), 6.78(1 \mathrm{H}$, br. d, $J 8$, $\mathrm{ArCH}), 6.68-6.66(2 \mathrm{H}, \mathrm{m}, \mathrm{ArCH}), 3.67\left(2 \mathrm{H}, \mathrm{t}, J 6.5,1-\mathrm{H}_{2}\right), 2.60\left(2 \mathrm{H}, \mathrm{t}, J 7.5,4-\mathrm{H}_{2}\right), 1.73-1.57$ $\left(4 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}, 3-\mathrm{H}_{2}\right), 1.23(1 \mathrm{H}$, br. s, OH$), 0.99(9 \mathrm{H}, \mathrm{s}, \mathrm{tBu}), 0.20\left(6 \mathrm{H}, \mathrm{s}, \mathrm{SiMe}_{2}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right), 155.6(\mathrm{C}), 143.9(\mathrm{C}), 129.1(\mathrm{CH}), 121.4(\mathrm{CH}), 120.2(\mathrm{CH}), 117.4(\mathrm{CH}), 62.8\left(\mathrm{CH}_{2}\right), 35.5$ $\left(\mathrm{CH}_{2}\right), 32.3\left(\mathrm{CH}_{2}\right), 27.4\left(\mathrm{CH}_{2}\right), 25.7\left(\mathrm{CH}_{3}\right), 18.2(\mathrm{C}),-4.4\left(\mathrm{CH}_{3}\right)$; IR $\left(\mathrm{cm}^{-1}\right), 3331(\mathrm{br}, \mathrm{w}), 2930$ (w), 2858 (w), 1603 (w), 1584 (w), 1484 (w), 1441 (w), 1272 (m), 1156 (m), 836 (m); HRMS: $m / z(\mathrm{CI})$, calculated for $\mathrm{C}_{16} \mathrm{H}_{29} \mathrm{O}_{2} \mathrm{Si} 281.1937[\mathrm{M}+\mathrm{H}]^{+}$, found $281.1946[\mathrm{M}+\mathrm{H}]^{+}$

4-(3-tert-Butyldimethylsilyloxyphenyl)butanal (11). Alcohol 10 ( $534 \mathrm{mg}, 1.9 \mathrm{mmol}, 1 \mathrm{eq}$. ) and DMSO ( $162 \mu \mathrm{~L}, 2.3 \mathrm{mmol}, 1.2$ eq.) were dissolved in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{ml})$ and cooled to $-78{ }^{\circ} \mathrm{C}$ under nitrogen. Oxalyl chloride ( $178 \mu \mathrm{~L}, 2.1 \mathrm{mmol}, 1.1 \mathrm{eq}$.) was added slowly and stirred for 20 mins at $-78{ }^{\circ} \mathrm{C}$. Triethylamine ( $1.32 \mathrm{ml}, 9.5 \mathrm{mmol}, 5$ eq.) was added and the reaction warmed to RT and stirred for 1 h and progress followed by TLC. On completion the reaction mixture was diluted with water ( 20 ml ) and extracted with $\mathrm{Et}_{2} \mathrm{O}(2 \times 50 \mathrm{ml})$. The combined organic phases were washed with brine ( 20 ml ), dried over magnesium sulfate, filtered and the solvent removed in vacuo to yield aldehyde $\mathbf{1 1}$ which was used without further purification ( 530 mg , quant). $\mathrm{R}_{\mathrm{f}}$ 0.63 (2:1 PE:EtOAc); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 9.77(1 \mathrm{H}, \mathrm{t}, J 1.5,1-\mathrm{H}), 7.15(1 \mathrm{H}$, app. t, $J 8$, ArCH), $6.77(1 \mathrm{H}, \mathrm{br} . \mathrm{d}, J 8, \mathrm{ArCH}), 6.70-6.66(2 \mathrm{H}, \mathrm{m}, \mathrm{ArCH}), 2.61\left(2 \mathrm{H}, \mathrm{t}, J 7.5,4-\mathrm{H}_{2}\right), 2.45(1 \mathrm{H}$, $\left.\mathrm{dt}, J 7.5,1.5,2-\mathrm{H}_{2}\right), 1.95\left(2 \mathrm{H}\right.$, app. quin, $\left.J 7.5 \mathrm{~Hz}, 3-\mathrm{H}_{2}\right), 0.99(9 \mathrm{H}, \mathrm{s}, \mathrm{tBu}), 0.20\left(6 \mathrm{H}, \mathrm{s}, \mathrm{SiMe}_{2}\right)$; $\delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right), 202.3(\mathrm{CHO}), 155.7(\mathrm{C}), 142.7(\mathrm{C}), 129.3(\mathrm{CH}), 121.5(\mathrm{CH}), 120.2(\mathrm{CH})$,
$117.7(\mathrm{CH}), 43.1\left(\mathrm{CH}_{2}\right), 34.8\left(\mathrm{CH}_{2}\right), 25.7\left(\mathrm{CH}_{3}\right), 23.5\left(\mathrm{CH}_{2}\right), 18.2(\mathrm{C}),-4.4\left(\mathrm{CH}_{3}\right)$; IR $\left(\mathrm{cm}^{-1}\right)$, 2930 (w), 2859 (w), 1725 (w), 1602 (w), 1584 (w), 1484 (w), 1273 (m), 1253 (m), 1157 (m), 836 (s), 779 (s); HRMS: $m / z(\mathrm{CI})$, calculated for $\mathrm{C}_{16} \mathrm{H}_{27} \mathrm{O}_{2} \mathrm{Si} 279.1780[\mathrm{M}+\mathrm{H}]^{+}$, found 279.1788 $[\mathrm{M}+\mathrm{H}]^{+}$

6-(3-tert-Butyldimethylsilyloxyphenyl)hex-1-en-3-ol (12). Aldehyde 11 ( $525 \mathrm{mg}, 1.9 \mathrm{mmol}, 1$ eq.) in $\mathrm{Et}_{2} \mathrm{O}(2.5 \mathrm{ml})$ was added slowly to stirring vinylmagnesium bromide solution ( 1 M in THF, $2.1 \mathrm{ml}, 2.1 \mathrm{mmol}, 1.1 \mathrm{eq}$.) at $-7{ }^{\circ} \mathrm{C}$ under nitrogen. On complete addition the reaction was warmed to RT and stirred for 1 h and progress followed by TLC. On completion the reaction mixture was quenched with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}$ solution ( 20 ml ) and water added to dissolve any precipitated salts. The aqueous solution was extracted with $\mathrm{Et}_{2} \mathrm{O}(2 \times 50 \mathrm{ml})$. The combined organics were washed brine ( 20 ml ), dried over magnesium sulfate, filtered and the solvent removed in vacuo to yield allylic alcohol $\mathbf{1 2}$ as a colourless oil, which was used without further purification ( $540 \mathrm{mg}, 94 \%$ ). $\mathrm{R}_{\mathrm{f}} 0.5$ (2:1 PE:EtOAc); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.13(1 \mathrm{H}, \mathrm{m}, \mathrm{ArCH})$, $6.78(1 \mathrm{H}, \mathrm{br} . \mathrm{d}, J 8, \mathrm{ArCH}), 6.67-6.66(2 \mathrm{H}, \mathrm{m}, \mathrm{ArCH}), 5.86(1 \mathrm{H}, \mathrm{ddd}, J 17,10.5,6.5,2-\mathrm{H}), 5.22$ $(1 \mathrm{H}, \mathrm{dt}, J 17,1.5,1-\mathrm{H} H), 5.11(1 \mathrm{H}, \mathrm{dt}, J 10.5,1.5,1-H \mathrm{H}), 4.11(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 2.59(2 \mathrm{H}, \mathrm{t}, J 7.5$, $\left.6-\mathrm{H}_{2}\right), 1.77-1.54\left(4 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}_{2}, 5-\mathrm{H}_{2}\right), 1.46(1 \mathrm{H}, \mathrm{br} . \mathrm{d}, J 4.5, \mathrm{OH}), 0.99(9 \mathrm{H}, \mathrm{s}, \mathrm{tBu}), 0.20(6 \mathrm{H}, \mathrm{s}$, $\mathrm{SiMe}_{2}$ ); $\delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 155.6(\mathrm{C}), 143.8(\mathrm{C}), 141.1(\mathrm{CH}), 129.1(\mathrm{CH}), 121.4(\mathrm{CH}), 120.2$ $(\mathrm{CH}), 117.4(\mathrm{CH}), 114.7\left(\mathrm{CH}_{2}\right), 73.1(\mathrm{CH}), 36.5\left(\mathrm{CH}_{2}\right), 35.6\left(\mathrm{CH}_{2}\right), 27.0\left(\mathrm{CH}_{2}\right), 25.7\left(\mathrm{CH}_{3}\right), 18.2$ (C), -4.4 (CH3); IR (cm-1), 3371 (br, w) 2930 (w), 2858 (w), 1603 (w), 1585 (w), 1485 (w), 1274 (w), 1157 (w), $840(\mathrm{w})$; HRMS: $m / z(E S I)$, calculated for $\mathrm{C}_{18} \mathrm{H}_{30} \mathrm{O}_{2} \mathrm{NaSi} 329.1907$ [M+Na] ${ }^{+}$, found $329.1917[\mathrm{M}+\mathrm{Na}]^{+}$

6-(3-Hydroxyphenyl)hex-1-en-3-ol (13) Allylic alcohol 12 ( $530 \mathrm{mg}, 1.7 \mathrm{mmol}, 1 \mathrm{eq}$.$) was$ dissolved in EtOH $(20 \mathrm{ml})$ and conc. aq. $\mathrm{HCl}(0.4 \mathrm{ml})$ added and the reaction stirred for 16 h at RT and progress followed by TLC. On completion the solvent was removed in vacuo and the
resultant residue purified by column chromatography ( $2: 1 \mathrm{PE}: E t O A c$ ) to yield allylic alcohol 13 as a brown oil ( $246 \mathrm{mg}, 75 \%$ ). $\mathrm{R}_{\mathrm{f}} 0.175$ (2:1 PE:EtOAc); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) 7.17-7.13 ( 1 H , m, ArCH), $6.76(1 \mathrm{H}, \mathrm{br} . \mathrm{d}, J 8, \mathrm{ArCH}), 6.67-6.65(2 \mathrm{H}, \mathrm{m}, \mathrm{ArCH}), 5.87(1 \mathrm{H}, \mathrm{ddd}, J 17,10.5,6.5$, $2-\mathrm{H}), 5.23(1 \mathrm{H}, \mathrm{dt}, J 17,1.5,1-H \mathrm{H}), 5.11(1 \mathrm{H}, \mathrm{dt}, J 10.5,1.5,1-\mathrm{H} H), 4.91(1 \mathrm{H}, \mathrm{br} . \mathrm{s}, \mathrm{ArOH})$, $4.13(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 2.60\left(2 \mathrm{H}, \mathrm{t}, J 7.5,6-\mathrm{H}_{2}\right), 1.80-1.50\left(5 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}_{2}, 5-\mathrm{H}_{2} \& \mathrm{OH}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 155.6(\mathrm{C}), 144.2(\mathrm{C}), 141.0(\mathrm{CH}), 129.5(\mathrm{CH}), 120.9(\mathrm{CH}), 115.3(\mathrm{CH}), 114.9\left(\mathrm{CH}_{2}\right)$, $112.7(\mathrm{CH}), 73.2(\mathrm{CH}), 36.4\left(\mathrm{CH}_{2}\right), 35.6\left(\mathrm{CH}_{2}\right), 26.9\left(\mathrm{CH}_{2}\right)$; $\mathrm{IR}\left(\mathrm{cm}^{-1}\right), 3305(\mathrm{br}, \mathrm{w}), 2939(\mathrm{w})$, 1588 (m), 1456 (m), 1267 (m), 1155 (m), 908 (s), 730 (s); HRMS: m/z (ESI), calculated for $\mathrm{C}_{12} \mathrm{H}_{16} \mathrm{O}_{2} \mathrm{Na} 215.1042[\mathrm{M}+\mathrm{Na}]^{+}$, found $215.1051[\mathrm{M}+\mathrm{Na}]^{+}$

6-(3-Hydroxyphenyl)hex-1-en-3-one (14) Alcohol $\mathbf{1 3}$ ( $100 \mathrm{mg}, 0.5 \mathrm{mmol}, 1 \mathrm{eq}$.) was dissolved in THF ( 10 ml ) and $\mathrm{MnO}_{2}(1.8 \mathrm{~g}, 21 \mathrm{mmol}, 40 \mathrm{eq}$.) added and the reaction stirred for 4 h at RT and progress followed by TLC. On completion the slurry was filtered through celite and the filter cake washed with DCM. The solvent was removed in vacuo yielding enone 14 as an orange oil ( $25 \mathrm{mg}, 25 \%$ ). $\mathrm{R}_{\mathrm{f}} 0.325$ (2:1 PE:EtOAc); $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.15(1 \mathrm{H}, \mathrm{m}, \mathrm{ArCH}), 6.76(1 \mathrm{H}$, br. d, $J 8, \mathrm{ArCH}), 6.69-6.67(2 \mathrm{H}, \mathrm{m}, \mathrm{ArCH}), 6.35(1 \mathrm{H}, \mathrm{dd}, J 17.5,10.5,2-\mathrm{H}), 6.19(1 \mathrm{H}, \mathrm{dd}, J$ $17.5,1,1-H \mathrm{H}), 5.82(1 \mathrm{H}, \mathrm{dd}, J 10.5,1,1-\mathrm{H} H), 4.82(1 \mathrm{H}, \mathrm{br} . \mathrm{s}, \mathrm{ArOH}), 2.62\left(2 \mathrm{H}, \mathrm{t}, J 7.5,4-\mathrm{H}_{2}\right.$ or $\left.6-\mathrm{H}_{2}\right), 2.60\left(2 \mathrm{H}, \mathrm{t}, J 7.5 \mathrm{~Hz}, 4-\mathrm{H}_{2}\right.$ or $\left.6-\mathrm{H}_{2}\right), 1.96\left(2 \mathrm{H}\right.$, quin, $\left.J 7.5,5-\mathrm{H}_{2}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$, $200.7(\mathrm{CO}), 155.6(\mathrm{CH}), 143.5(\mathrm{C}), 136.5(\mathrm{CH}), 129.6(\mathrm{CH}), 128.1\left(\mathrm{CH}_{2}\right), 121.0\left(\mathrm{CH}_{2}\right), 115.4$ (CH), $112.9(\mathrm{C}), 38.7\left(\mathrm{CH}_{2}\right), 34.9\left(\mathrm{CH}_{2}\right), 25.1\left(\mathrm{CH}_{2}\right)$; IR $\left(\mathrm{cm}^{-1}\right), 3357(\mathrm{br}, \mathrm{w}), 2933(\mathrm{w}), 1667(\mathrm{~s})$, 1613 (s), 1598 (s), 1587 ( s$), 1455$ ( s$), 1403$ (m), 1273 (m), 1225 (m), 1155 (s), 968 (m), 781 (s), 695 (s); HRMS: $m / z(\mathrm{CI})$, calculated for $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{O}_{2} 191.1072[\mathrm{M}+\mathrm{H}]^{+}$, found $191.1076[\mathrm{M}+\mathrm{H}]^{+}$

6-(3-Hydroxyphenyl)hexyl-3-one CoASH (15) CoASH ( $20 \mu \mathrm{~L}, 100 \mathrm{mM}$ ) was mixed with enone 14 in acetone ( $20 \mu \mathrm{~L}, 100 \mathrm{mM}$ ) and $2 \mu \mathrm{~L}$ of potassium carbonate solution in $\mathrm{H}_{2} \mathrm{O}(1 \mathrm{M})$.

The solution was initially cloudy and then vortexed for 5 minutes until clear to yield 15, checked by ESMS mass (theoretical mass): $958.24 \mathrm{Da}(957.78 \mathrm{Da})$ and was used directly in the coupling to ${ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$ labelled-ACP.

7-Hydroxyoct-1-en-3-one (17) Dry THF ( 20 ml ) was cooled to - $78{ }^{\circ} \mathrm{C}$, followed by the addition of $\delta$-hexalactone ( $0.62 \mathrm{ml}, 5.5 \mathrm{mmol}$ ) under nitrogen. Magnesium vinyl bromide ( $3.2 \mathrm{ml}, 1.6 \mathrm{M}$ in THF) was then added drop-wise and reaction stirred at $-78^{\circ} \mathrm{C}$ for 2 h under nitrogen. The reaction was washed with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ aqueous solution $(10 \mathrm{ml})$ and extracted with EtOAc ( $3 \times 20 \mathrm{ml}$ ) and the combined organic phase dried over magnesium sulfate and concentrated in vacuo. The crude product was purified through a silica column (EtOAc:petrol) to yield 7-hydroxy-oct-en-3one $\mathbf{1 7}$ as a colourless oil $(0.55 \mathrm{~g}, 3.9 \mathrm{mmol}, 71 \%) . \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.24(3 \mathrm{H}, \mathrm{d}, J 6.2,8-$ $\left.\mathrm{H}_{3}\right), 1.56-1.62\left(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}_{2}\right), 1.80-1.86\left(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}_{2}\right), 2.59\left(2 \mathrm{H}, \mathrm{t}, J 6.8,4-\mathrm{H}_{2}\right), 3.68(1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H})$, $5.80(1 \mathrm{H}, \mathrm{dd}, J 10.2,1.7,1-H \mathrm{H}), 6.19(1 \mathrm{H}, \mathrm{dd}, J 17.0,1.7, \mathrm{~Hz}, 1-\mathrm{H} H), 6.32(1 \mathrm{H}, \mathrm{dd}, J 17.0,10.2$, $2-\mathrm{H}) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 19.7(\mathrm{C}-5), 23.1$ (C-8), 38.2 (C-6), 39.1 (C-4), $67.0(\mathrm{C}-7), 127.9(\mathrm{C}-1)$, 136.1 (C-2), 171.8 (C-3). Spectroscopic data are in accord with the literature. ${ }^{8}$

Oct-1-ene-3,7-dione (18). Dess Martin periodinane ( 0.58 g ) was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{ml})$ under nitrogen, and this suspension was added drop-wise to a solution of alcohol $\mathbf{1 7}(0.14 \mathrm{~g}$, $1 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{ml})$ and stirred at room temperature for two hours. A cloudy solution formed during the addition. The reaction filtered through a silica column (EtOAc:petrol) and the solvent removed in vacuo to yield enone 18 as a colourless oil ( $0.11 \mathrm{~g}, 0.79 \mathrm{mmol}, 79 \%$ ). $\delta_{\mathrm{H}}(300$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.94\left(3 \mathrm{H}, \mathrm{s}, 8-\mathrm{H}_{3}\right), 1.75\left(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}_{2}\right), 2.47\left(2 \mathrm{H}, \mathrm{t}, J 7.2,6-\mathrm{H}_{2}\right), 2.60(2 \mathrm{H}, \mathrm{t}, J 7.2,4-$ $\mathrm{H}_{2}$ ), $5.75(1 \mathrm{H}, \mathrm{dd}, J 10.2,1.5,1-H \mathrm{H}), 6.11-6.17(1 \mathrm{H}, \mathrm{dd} J 17.6,1.5,1-\mathrm{H} H), 6.28$ ( 1 H , dd $J 17.6$ $10.2,2-\mathrm{H}) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 18.9$ (C-5), 22.2 (C-8), 39.3 (C-4), 43.1 (C-6), 128.5 (C-1), 137.8 (C-2), 172.5 (C-3), 208.3 (C-7). Spectroscopic data are in accord with the literature. ${ }^{9}$

Synthesis and purification of 3,7-dioxo-octyl CoASH (19) An aqueous solution of CoASH
( $100 \mathrm{mM}, 50 \mu \mathrm{l}$ ) was added to enone $\mathbf{1 8}$ in acetone ( $100 \mathrm{mM}, 65 \mu \mathrm{l}$ ) and mixed gently at room temperature for 2 hours, centrifuged and purified by HPLC using the conditions described in the general experimental details. ESMS mass (theoretical mass): $908.1 \mathrm{Da}(907.71 \mathrm{Da})$ for $19 . \delta_{\mathrm{H}}$ $\left(600 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right) 0.81\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 0.93\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.74\left(2 \mathrm{H}\right.$, pent, $\left.J 7.3, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right), 2.16$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{COCH}_{3}$ ), $2.46\left(2 \mathrm{H}, \mathrm{t}, J 6.5,6\right.$ ' $\left.-\mathrm{H}_{2}\right), 2.52-2.56\left(4 \mathrm{H}, \mathrm{m}, \mathrm{H}-4 / \mathrm{H}-6, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right), 2.64(2 \mathrm{H}$, t, $\left.J 6.5,9^{\prime}{ }^{\prime}-\mathrm{H}_{2}\right), 2.73\left(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 6.5, \mathrm{CH}_{2} \mathrm{~S}\right), 2.81\left(2 \mathrm{H}, \mathrm{t}, J 6.5, \mathrm{COCH}_{2}\right), 3.34\left(2 \mathrm{H}, \mathrm{t}, J 6.6,8^{\prime}{ }^{-}-\mathrm{H}_{2}\right)$,
 $\left.3^{\prime}-\mathrm{H}\right), 4.26\left(2 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}_{2}\right), 4.62\left(1 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}\right), 4.86-4.88\left(2 \mathrm{H}, \mathrm{m}, 2^{\prime}-\mathrm{H} / 3^{\prime}-\mathrm{H}\right), 6.20(1 \mathrm{H}, \mathrm{d}, J 5.5$, 1'-H), $8.44(1 \mathrm{H}, \mathrm{s}, 2-\mathrm{H}), 8.67(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H}) ; \delta \mathrm{c}\left(150 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right) 19.0\left(\mathrm{CH}_{2}\right), 20.9\left(\mathrm{CH}_{3}\right), 23.3$
 (C4/C6), 44.8 (C2), 67.8 (C-5'), 74.6 (C-1''), 76.8 (C-2'/C-3' overlapped), 77.1 (C-3'"), 86.3 (C$\left.4^{\prime}\right), 90.4$ (C-1'), 121.22 (C-5), 145.2 (C-8), 147.1 (C-6), 147.4 (C-2), 151.40 (C-4), 176.7 (C-7’’), 178.1 C-4", 217.9 (CO), $221.0(\mathrm{CO}) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ assignments of the CoASH portion were checked for consistency with that of Pal and Bearne ${ }^{5}$.

Preparation act ACP derivatives 7, $\mathbf{1 6}$ and $\mathbf{2 0}$ for NMR. 100 mM CoASH derivative solution $\left(\mathrm{H}_{2} \mathrm{O}, 2 \mu \mathrm{~L}\right)$ was added to a solution of unlabelled, ${ }^{15} \mathrm{~N}$ or ${ }^{13} \mathrm{C} /{ }^{15} \mathrm{~N}$ ACP solution $\left(1.0 \mathrm{mM}, \mathrm{H}_{2} \mathrm{O}\right.$, $10 \mu \mathrm{~L}$ ) with $40 \mu \mathrm{M}$ ACPS solution ( $5 \mu \mathrm{~L}$ in assay buffer) in a 1.5 mL Eppendorf tube. The protein solution was made up to $100 \mu \mathrm{~L}$ with 50 mM Tris, $10 \mathrm{mM} \mathrm{MgCl}_{2}(\mathrm{pH} 8.8)$ and incubated at $37^{\circ} \mathrm{C}$ for 24 hours. The progress of the assay was checked by mass spectrometry for 7 and 20 using unlabelled ACP ( 7 [observed mass 9600 Da , expected 9600 Da ], $\mathbf{2 0}$ [observed mass 9579.0, expected mass 9581.0 Da . Preparation of ${ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$ labelled ACP derivatives were subsequently prepared $\left({ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}\right.$ labelled 5-phenyl-3-oxo-pentyl ACP 7 (expected mass for $100 \%{ }^{13} \mathrm{C} /{ }^{15} \mathrm{~N}$ incorporation 10101 Da , observed mass 10082 ( $\sim 97 \%$ incorporation)) and ${ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$ labelled 3,7dioxooctyl ACP 20 (expected mass for $100 \%{ }^{13} \mathrm{C} /{ }^{15} \mathrm{~N}$ incorporation 10081 Da , observed mass $10063 \mathrm{Da}) .{ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$ labelled $\mathbf{1 6}$ [observed mass 10118 Da , expected 10129.6 for $100 \%{ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$
labelling]. Twenty reactions were performed simultaneously to obtain sufficient amounts of labelled ACP. After the reaction went to $>95 \%$ completion by mass spectrometry, the derivatised ACP was concentrated and buffer exchanged 5 times into 20 mM potassium phosphate ( pH 5.5 ) using Centricon ultrafiltration unit ( 4 ml , Millipore, MWCO 3000 Da ). The final samples ( 500 $\mu \mathrm{L}$ ) were $1.7 \mathrm{mM}, \mathrm{pH} 5.4 . \mathrm{NaN}_{3}(0.1 \mathrm{mM})$ in $92 \% \mathrm{H}_{2} \mathrm{O} / 8 \% \mathrm{D}_{2} \mathrm{O}$ respectively.

Biological NMR data collection for acylated act ACP. All protein NMR experiments for act ACP were acquired at $25{ }^{\circ} \mathrm{C}$ on a Varian INOVA 600 MHz spectrometer and a cryoprobe equipped Varian VNMRS 600 MHz spectrometer. HNCACB, CBCACONH, CCONH, HCCONH, HNHA, HCCH-TOCSY, HNCO experiments were used to assign the backbone and side chain atoms. ${ }^{15} \mathrm{~N}$ - edited and ${ }^{13} \mathrm{C}$-edited NOESY datasets were acquired with 100 ms mixing times to generate distance restraints for structure calculation. For determining the interactions between isotopically labelled protein and unlabelled $4{ }^{\prime}$-PP cofactor and derivatives, $F_{2} \mathrm{f}$ (i.e. ${ }^{13} \mathrm{C}$ signals removed in $F_{2}$ ) and $F_{1} \mathrm{f} F_{2} \mathrm{f}$-filtered TOCSY and NOESYs ( 150 ms mixing time) were acquired to detect the chemical shifts of the ligands and the protein-ligand contacts ${ }^{10}$.

All the NMR data were processed using NMRPipe ${ }^{11}$ spectral processing and analysis system, and the assignment and NOE data collection were analyzed using CCPN Analysis Version 2.1.3 ${ }^{12,13}$. Chemical shift differences were calculated using the following function ${ }^{14}$ : $\Delta \delta_{\text {avg }}=\left[0.5 \times\left(\Delta \delta \mathrm{H}^{2}+0.2 \times \Delta \delta \mathrm{N}^{2}\right)\right]^{0.5}$, where $\Delta \delta \mathrm{H}$ and $\Delta \delta \mathrm{N}$ are the values of chemical shift changes in H and N dimensions, respectively.

The NOE peak list and assignments of protein atoms were exported using CCPNMR Format Converter ${ }^{15}$ in NmrView format, which is accepted by the structure calculation programme. The $\varphi$ and $\psi$ dihedral angle restraints were analyzed using Torsion Angle Likelihood Obtained from

Shift and sequence similarity (TALOS). The chemical shift input for TALOS was generated using CCPNMR Format Converter ${ }^{15}$. Initially, classification of predictions was automatically determined by TALOS, and the borderline predictions were then manually assessed. Violations flagged by the structure calculation programme were either relaxed or removed from the TALOS restraints.

All structure calculations were performed using Ambiguous Restraints for Iterative Assignment (ARIA) version 1.2 or 2.2 coupled to CNS version 1.2. Topology and parameter files for the modified Ser 42 were generated using the Dundee PRODRG server and added to the existing CNS files. The number of dynamics steps was increased over default values to 20000 and 16000 for the first and second cooling stages, respectively ${ }^{16}$. Initially, the chemical shift table, TALOS restraints and NOE peak lists were used as input data for structure calculation. After the first run, the problematic NOE restraints were flagged and checked manually in the spectra. Usually, the errors arose from noise, unaliased peaks and assignments falling out of the defined tolerances. The refined input was then used in the next run of structure calculations. This cycle was repeated until low energy and non-violated structures were obtained. In the last iteration of the calculation 100 structures were generated, and the 20 lowest energy structures were further refined in explicit water using the RECOORD protocol ${ }^{17}$. RMSD values were calculated using Molmol, and the structure qualities were assessed using the iCing web-server (https://nmr.cmbi.ru.nl/icing/iCing.html\#file), which provided the assessments of the structure by WHAT IF and PROCHECK. Three-dimensional structures were analyzed and drawn using Pymol ${ }^{18}$ Version 0.99 and 1.3. Internal hydrophobic cavities were predicted using the CastP web server ${ }^{19}$

Figure S1. Strips extracted from a ${ }^{13} \mathrm{C}$-edited NOESY-HSQC three-dimensional spectrum of 5-phenyl-3-oxo-pentyl ACP. Representative strips from the ${ }^{13} \mathrm{C}$ methyl shifts for the groups of A, Ala49 H $\beta^{*}$, B, Ala65 H $\beta^{*}$ and C, Val68 H $\gamma^{*}$. The protein-ligand NOEs are labeled with their assignments and chemical shifts of the corresponding ligand atoms. The numbering of atoms in the aromatic moiety of the 4'-PP derivative is shown below.


Figure S2. F1f F2f NOEs of a) 5-phenyl-3-oxo-pentyl and b) 3,7-dioxo-octyl act ACPs. In a) NOEs from $4^{\prime}$-PP methyl group protons ( $\mathrm{H} 30^{*} / 31^{*}$ ) to the atoms partway along the $4^{\prime}$ '-PP chain (shown with arrows), indicative that the 4 '-PP side chain is bent back on itself and the aromatic group may be protected by the protein. Long-range NOEs are also observed in the F1fF2f spectrum of b) with the addition of weak NOEs from $\mathrm{H} 30^{*} / 31^{*}$ protons to H 6 " and H 9 ".



| Table S2 Structural statistics and quality indicators for 3,7-dioxo-octyl act ACP |  |
| :---: | :---: |
| Number of restraints | [ |
| total | 2308 |
| unambiguous | 1548 |
| ambiguous | 760 |
| intra residue | 883 |
| sequential | 515 |
| medium range | 334 |
| long range | 576 |
| Violation per structure |  |
| NOE $>0.5 \AA$ | 0 |
| NOE $>0.3 \AA$ | 0 |
| NOE $>0.1 \AA$ | 7 |
| TALOS $\varphi / \psi$ | 0 |
| RMSD ( ${ }_{\text {( }}$ ) |  |
| well-ordered residues | $0.49 \pm 0.06$ |
| all residues | $1.00 \pm 0.12$ |
| Ramachandran plot |  |
| most favoured | 86.5\% |
| additionally allowed | 12.5\% |
| generously allowed | 0.7\% |
| disallowed | 0.3\% |
| Z-scores |  |
| 2 nd generation packing quality | -1.035 |
| Ramachandran plot appearance | -2.449 |
| $\chi 1 / \chi 2$ rotamer normality | -1.366 |
| Backbone conformation | 0.189 |

Mass Spectra of 5-phenyl-3-oxo-pentyl CoASH 6 and 3,7-dioxo-octyl CoASH 20.


ESMS of 6-(3-hydroxyphenyl)hexyl CoASH 15.


ESMS of 3,7-dioxo-octyl ACP.



Time course showing production of 5-phenyl-3-oxo-pentyl ACP with time points taken at A) 6 hours, B) 23 hours and C) 31 hours


ESMS of (A) ${ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$ labelled 5-phenyl-3-oxo-pentyl ACP 7 (expected mass for $100 \% 13 \mathrm{C} / 15 \mathrm{~N}$ incorporation 10101 Da ) and (B) ${ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$ labelled 3,7-dioxooctyl ACP 20 (expected mass for 100 $\%{ }^{13} \mathrm{C} /{ }^{15} \mathrm{~N}$ incorporation 10081 Da ). The observed masses are fractionally lower indicating $\sim 97 \%$ isotopic incorporation.












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