

Supplementary information

The Flat-on Ambipolar Triphenylamine/C₆₀ Nano-Stacks Formed from the Self-Organization of a Pyramid-Sphere Shape Amphiphile

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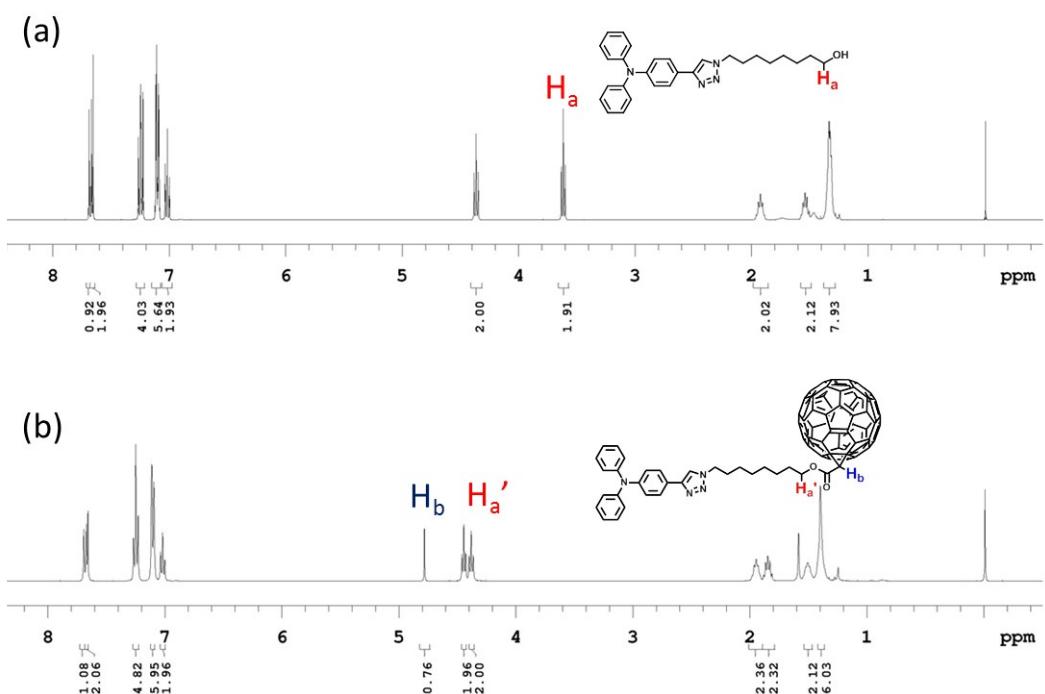


Fig. S1 ^1H NMR spectra of (a) TPA-OH and (b) TPA-C₆₀.

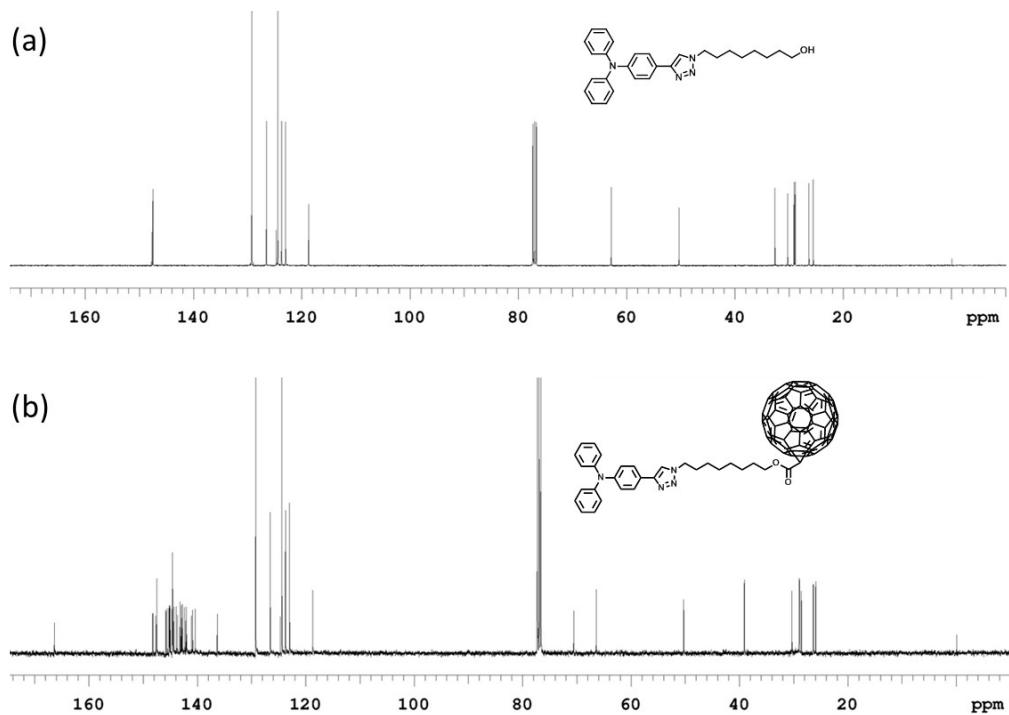


Fig. S2 ^{13}C NMR spectra of (a) TPA-OH and (b) TPA- C_{60} .

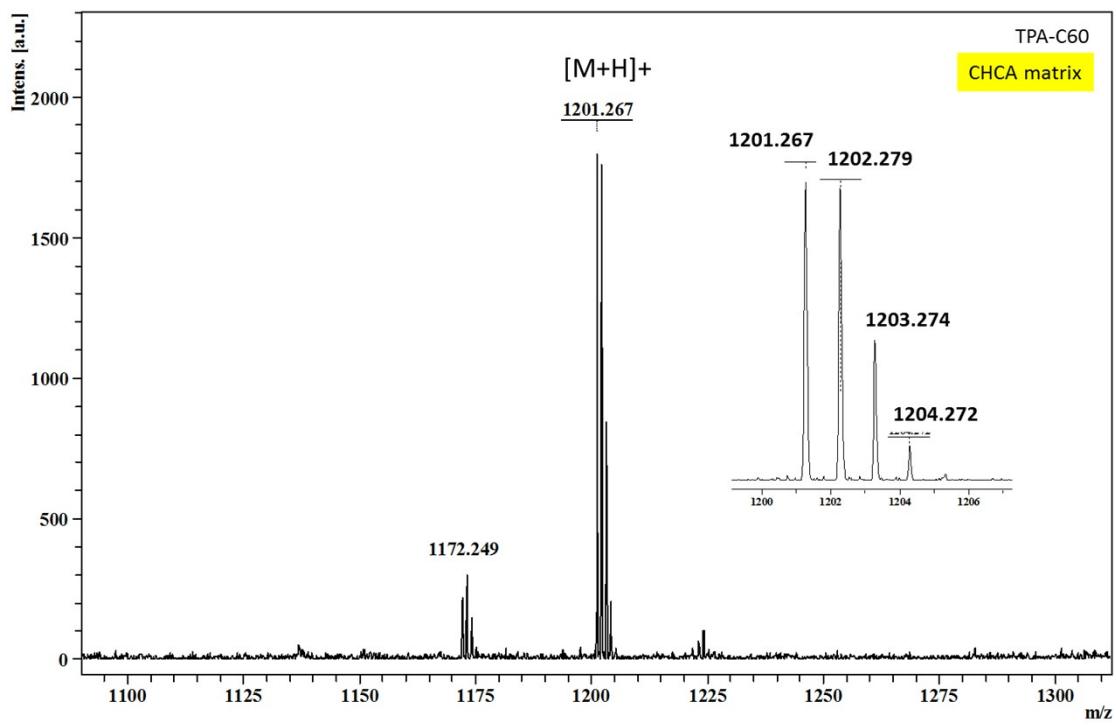


Fig. S3 MALDI TOF mass spectra of **TPA-C₆₀**.

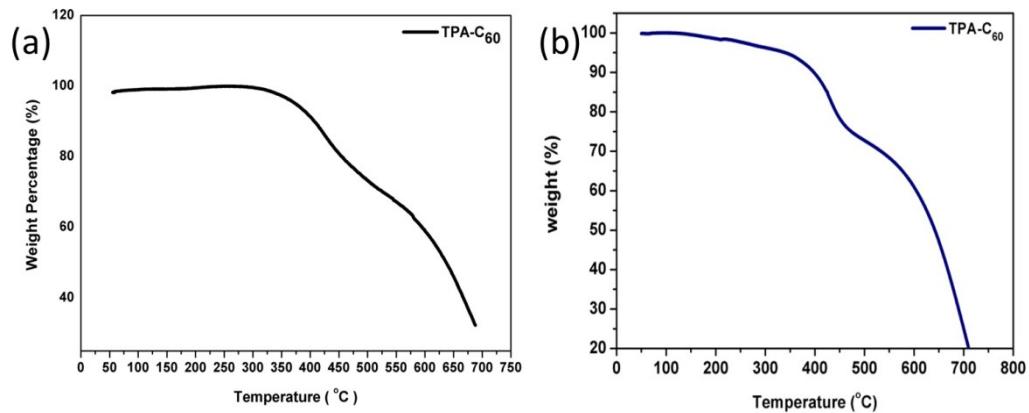


Fig. S4 Thermogravimetry analysis of the **TPA-C₆₀** samples dried by (a) a cryo pump ($\sim 6 \times 10^{-7}$ torr) and (b) a mechanical pump ($\sim 10^{-2}$ torr).

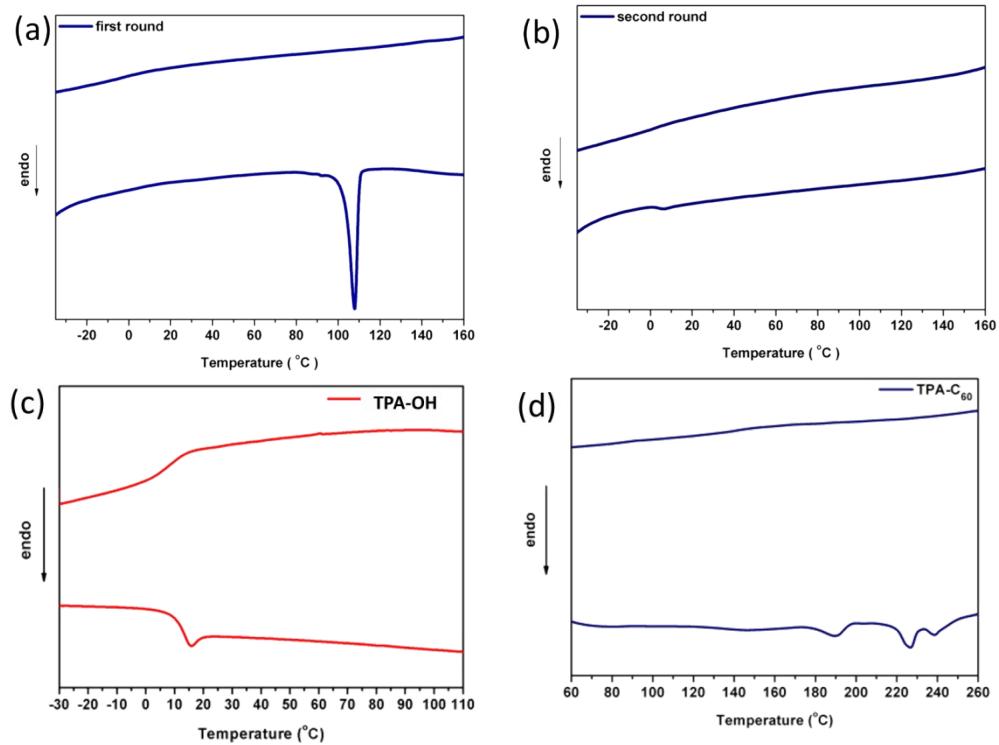


Fig. S5 DSC curves of (a) (b) 4-ethynyl-N,N-diphenylaniline, (c)TPA-OH, (d) TPA-C₆₀

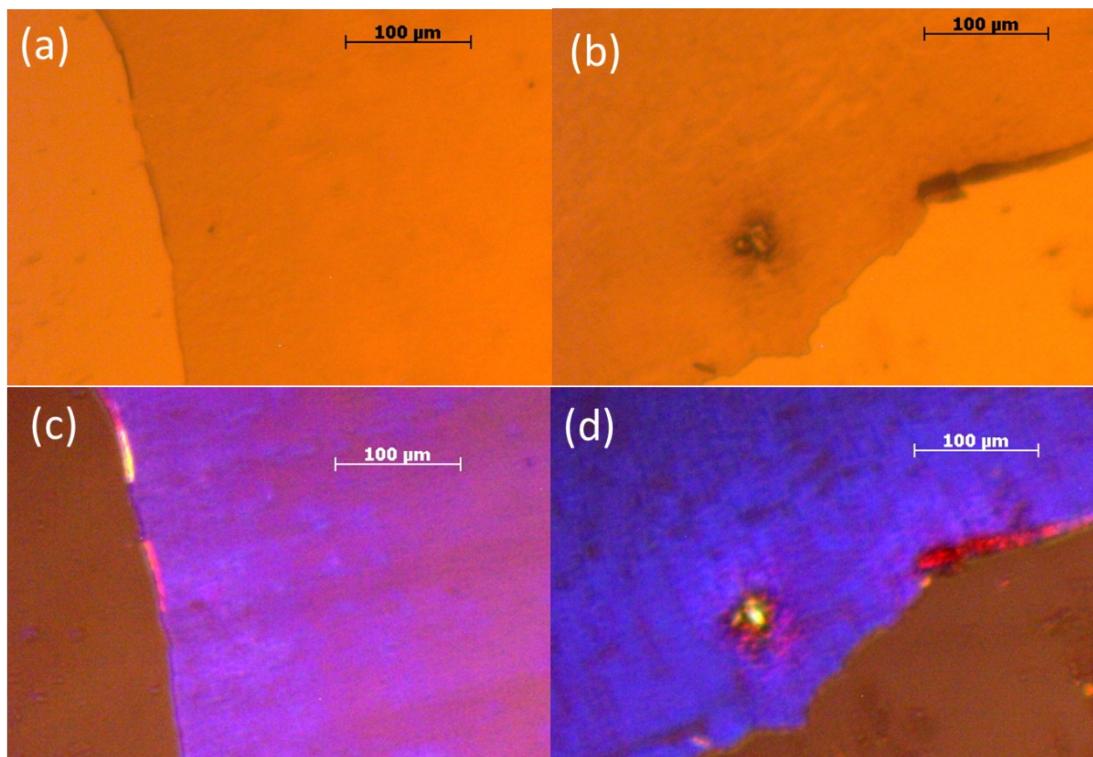


Fig. S6 Optical microscope (OM) images (a) (b), and POM images (c) (d) of TPA-C₆₀ processed via the PAC method with ODCB.

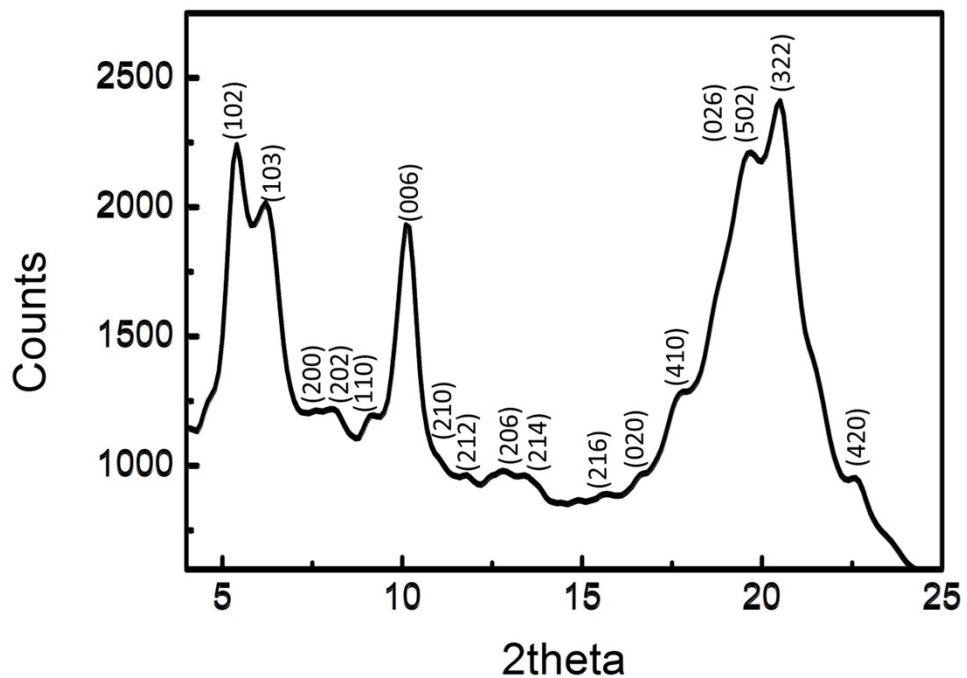


Fig. S7 The powder X-ray diffraction pattern of TPA-C₆₀.

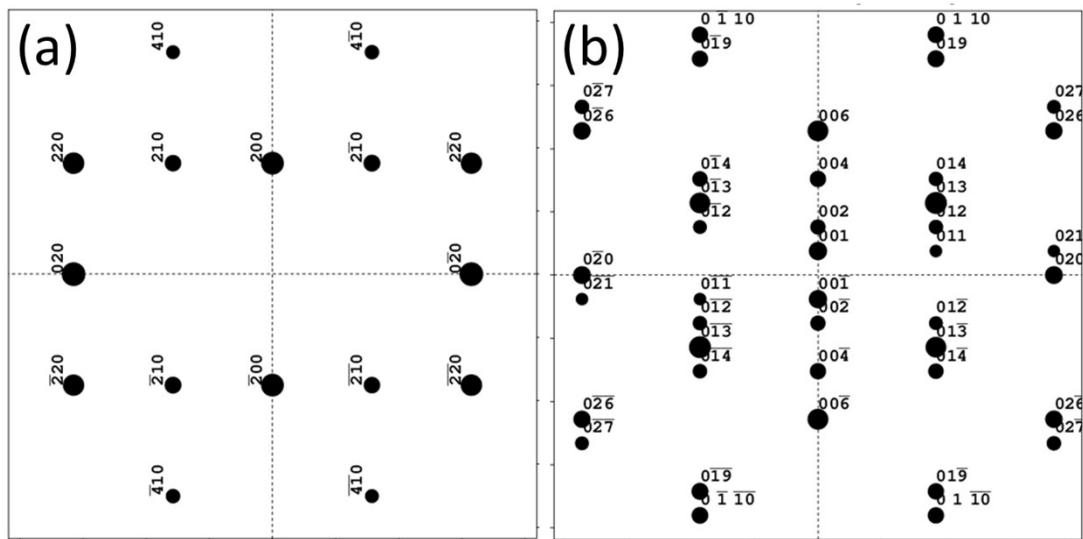


Fig. S8 Simulated ED patterns of **TPA-C₆₀** along (a) the [001] zone and (b) the [100] zone.

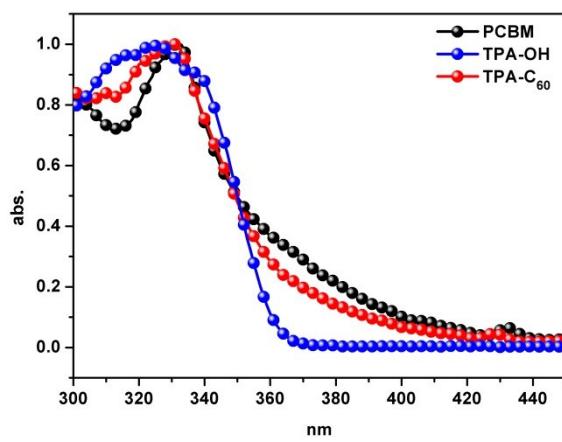


Fig. S9 UV-Vis spectra of TPA-C₆₀, TPA-OH, and PCBM.

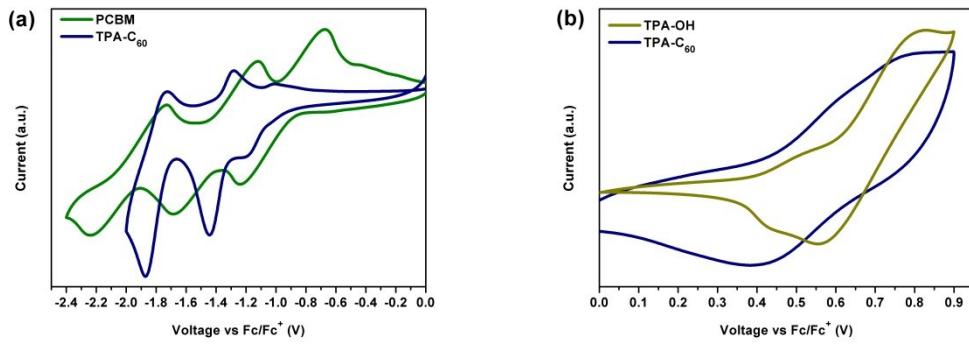


Fig. S10 (a) Reduction and (b) oxidation cyclic voltammetry curves of TPA-C₆₀ and comparisons.

Table S1. The d-spacings of the (hkl) planes of the crystalline phase of **TPA-C₆₀**.

(hkl) plane	Theoretical d-spacing (Å)	Experimental d-spacing (Å)
(102)	16.4	16.3
(103)	14.2	14.2
(200)	11.4	11.6
(202)	10.4	10.9
(110)	9.7	9.7
(006)	8.5	8.7
(210)	7.8	8.0
(210)	7.5	7.5
(206)	6.8	6.9
(214)	6.7	6.6
(216)	5.7	5.7
(020)	5.4	5.3
(410)	5.0	5.0
(026)	4.5	4.5
(502)	4.5	4.5
(322)	4.3	4.3
(420)	3.9	3.9

Table S2. Reduction potential, oxidized potential, HOMO and LUMO energy of PC₆₁BM, **TPA-C₆₀** and TPA-OH.

	E _{ox} ^{onset} [V]	E _{re} ^{onset} [V]	HOMO [eV]	LUMO [eV]
PC₆₁BM		-0.88		-3.91
TPA-C₆₀	0.41	-0.85	-5.20	-3.94
TPA-OH	0.40		-5.19	