

Transformation Process and Mechanism between the α - Conformation and β -Conformation of Conjugated Polymer PFO in Precursor Solution

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Supporting Information

1. The calculation method of the proportion of β -conformation in PFO solution¹

The proportion of β -conformation in PFO solution can be calculated according to Lambert-Beer's law (Eq. S2), where c_α and c_β are the concentrations of α -conformation and β -conformation, respectively, A (obtained from experiments) is the absorbency and ε is the absorption coefficient. In UV-vis absorption spectra of PFO solutions, it is difficult to obtain the ε for α -conformation and β -conformation. Thus we used the method of molecular simulation to the absorption spectra to calculate the proportion of $\varepsilon_\beta/\varepsilon_\alpha$.²

$$\alpha\% + \beta\% = 1 \quad (\text{S1})$$

$$\beta\% = \frac{c_\beta}{c_\beta + c_\alpha} = \frac{A_\beta \times \varepsilon_\alpha}{A_\beta \times \varepsilon_\alpha + A_\alpha \times \varepsilon_\beta} \quad (\text{S2})$$

By defining six monomer units as the conjugation length,³ we optimized the symmetry geometries of PFO chains for the approximate calculation of ε using Density Functional Theory (DFT).⁴ To calculate the oscillator strengths (f), Time-Dependent Density Functional Theory (TD-DFT) was used,⁵ as this method has been found to give reliable results.⁶ DFT and TD-DFT are both with a B3LYP hybrid functional basis set level of 6-31G*.⁷ All calculations were performed using the Gaussian 03 package.⁶ The oscillator strengths (f) of the two conformations are 4.46 for α -conformation at 386 nm and 4.83 for β -conformation at 437 nm. Thus, the proportion of $\varepsilon_\beta/\varepsilon_\alpha$ can be obtained from Eq. S3 and Eq. S4, where k is a constant and ν is the vibration frequency of two conformations. We can calculate the proportion of β -conformation in each UV-vis absorption spectra of PFO solutions using Eq. S2 and Eq. S4.

$$f = k \int \varepsilon d\nu, \quad f \propto \varepsilon \quad (\text{S3})$$

$$\frac{\varepsilon_{\beta}}{\varepsilon_{\alpha}} = \frac{f_{\beta}}{f_{\alpha}} = \frac{4.83}{4.46} = 1.08 \quad (\text{S4})$$

REFERENCES

- (1) Huang, L.; Huang, X. N.; Sun, G. N.; Gu, C.; Lu D.; Ma, Y. G. *J. Phys. Chem. C* **2012**, *116*, 7993-7999.
- (2) Gu, C.; Tang, S.; Yang, B.; Liu, S. J.; Lv, Y.; Wang, H.; Yang, S. M.; Hanif, M.; Lu, D.; Shen, F. Z.; Ma, Y. G. *Electrochim. Acta* **2009**, *54*, 7006-7011.
- (3) Ng, M. F.; Sun, S. L.; Zhang, R. Q. *J. Appl. Phys.* **2005**, *97*, 103513.
- (4) Hohenberg, P.; Kohn, W. *Phys. Rev. B* **1964**, *136*, 864-871.
- (5) Runge, E.; Gross, E. K. U. *Phys. Rev. Lett.* **1984**, *52*, 997-1000.
- (6) Hirata, S.; Lee, T. J.; Gordon, M. H. *J. Chem. Phys.* **1999**, *111*, 8904-8912.
- (7) Lee, C.; Yang, W.; Parr, R. G. *Phys. Rev. B* **1988**, *37*, 785-789.

2. Light emission spectra of the transformation process from α -conformation to β -conformation

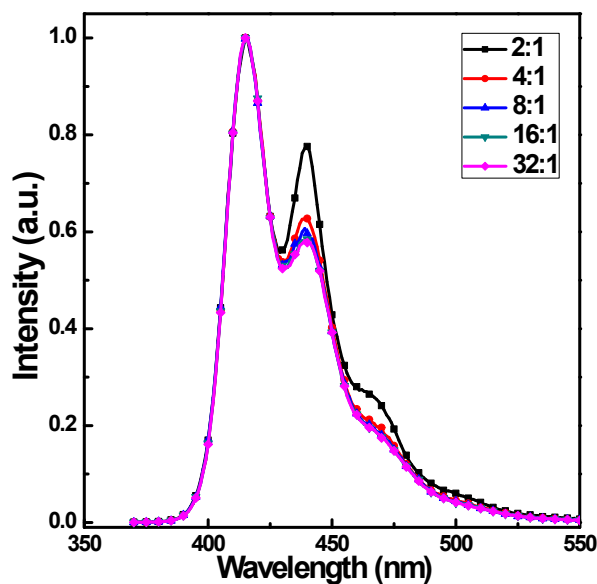


Fig.S1 Normalized light emission spectra of five PFO samples with different toluene/ethanol ratios: 32:1, 16:1, 8:1, 4:1, 2:1. All the five curves were normalized at the 415 nm.

From Fig.S1, it can be seen that the corresponding light emission spectra is well consistent with the UV-vis absorption spectra Fig.2 (a) in the transformation process from α -conformation to β -conformation.