

## Supplementary Information

### Double helix quinine-based supergelator

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### **General procedure for gel preparation**

In small screw-caped vial 7.5 mg of finely powderized **DDQn** (prepared according to the procedure reported by K. M. Kacprzak<sup>1</sup>) was placed and dissolved in 0.25 mL of AcOEt (or other solvent) on heating. After dissolution of the gelator, resultant solution was cooled down to almost room temperature followed by the quick addition of 1 mL of hydrocarbon (hexane, pentane or heptane) or other solvent. Such prepared gels were found to be stable in closed vials for months.

For larger scale preparation 150 mg of **DDQn**, 5 mL of AcOEt and 20 mL hydrocarbon solvent were routinely used for the gelation with the same efficiency.

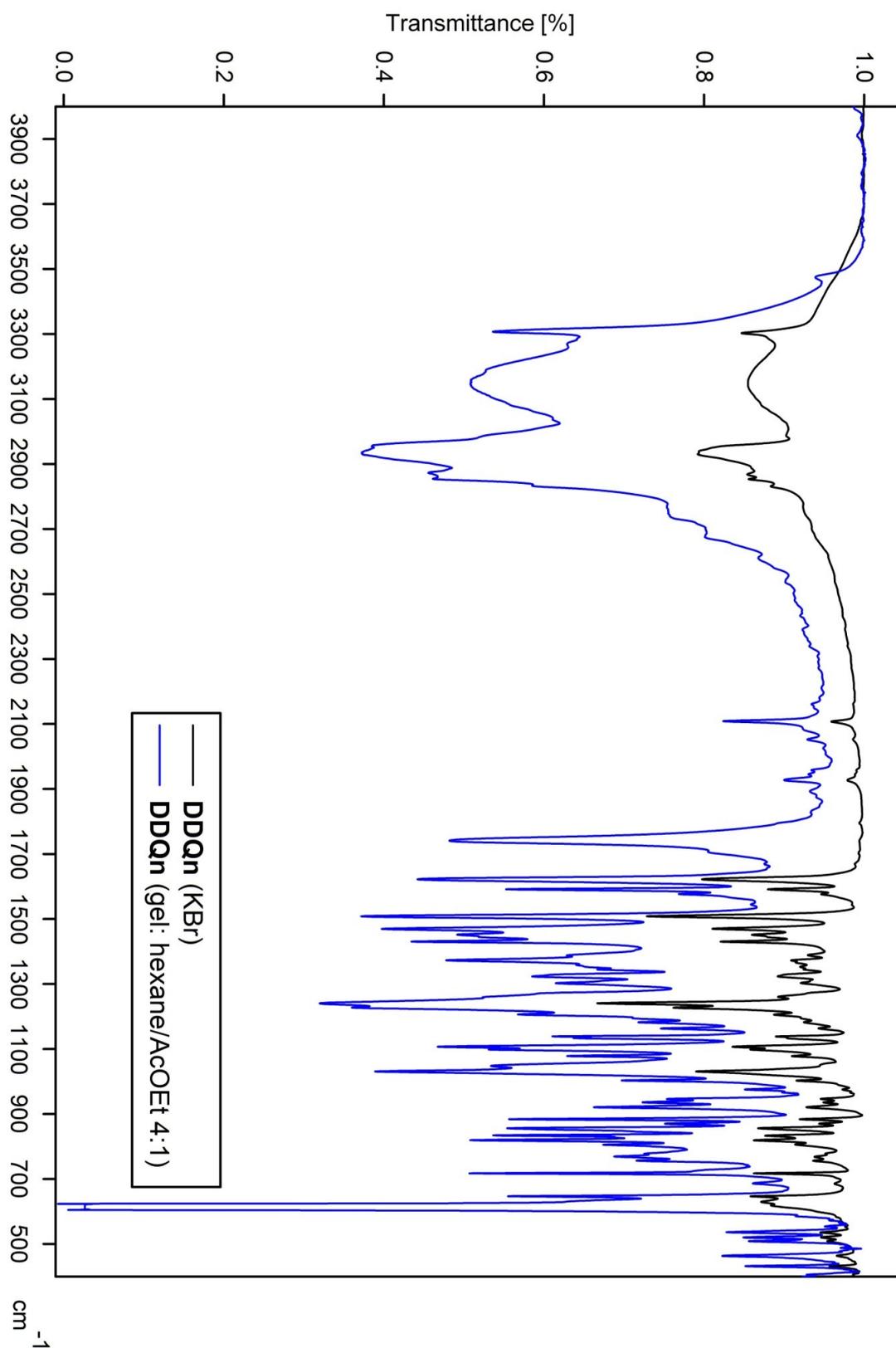


Figure S1. FT-IR spectra of **DDQn** (KBr disc) and **DDQn** in gel (AcOEt:He 1:4 v/v, film in KBr plates).

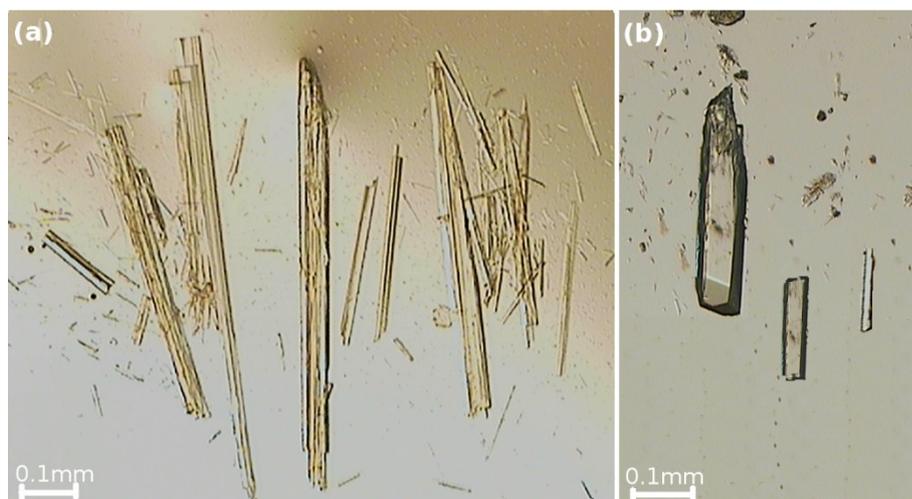


Figure S2. The crystals of (a) **DDQn** and (b) **DDQd** immersed in silicon oil.

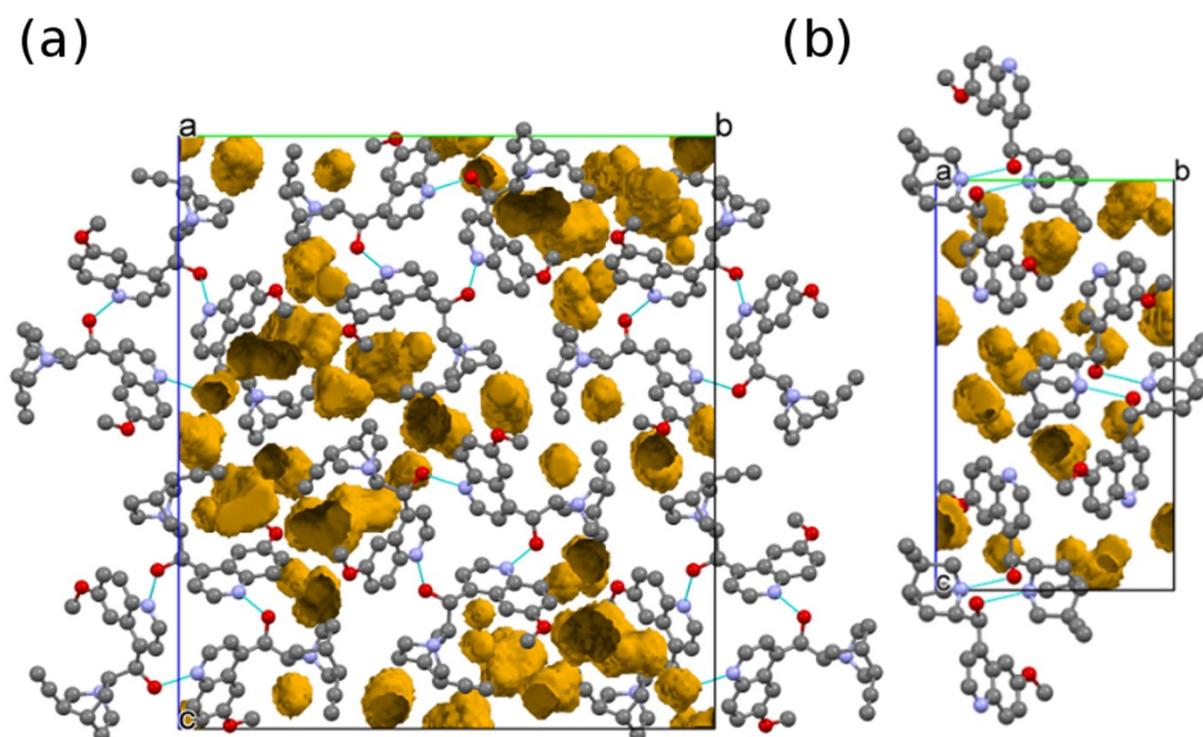


Figure S3. Voids in the crystal structure of (a) **DDQn** and (b) **DDQd**, as calculated by program Mercury<sup>2</sup> assuming the probing sphere of 0.8 Å and 0.2 Å step.

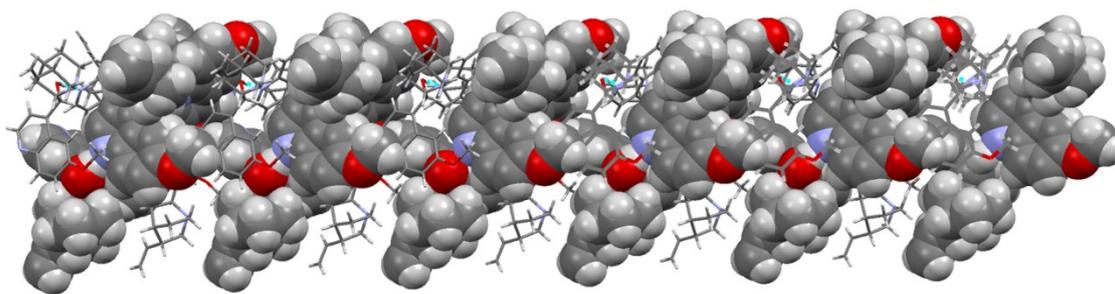


Figure S4. Projection of the OH $\cdots$ N bonded double helix of **Qn**: the molecular of one helix are represented as the space-filling models, and the other are the caped sticks.

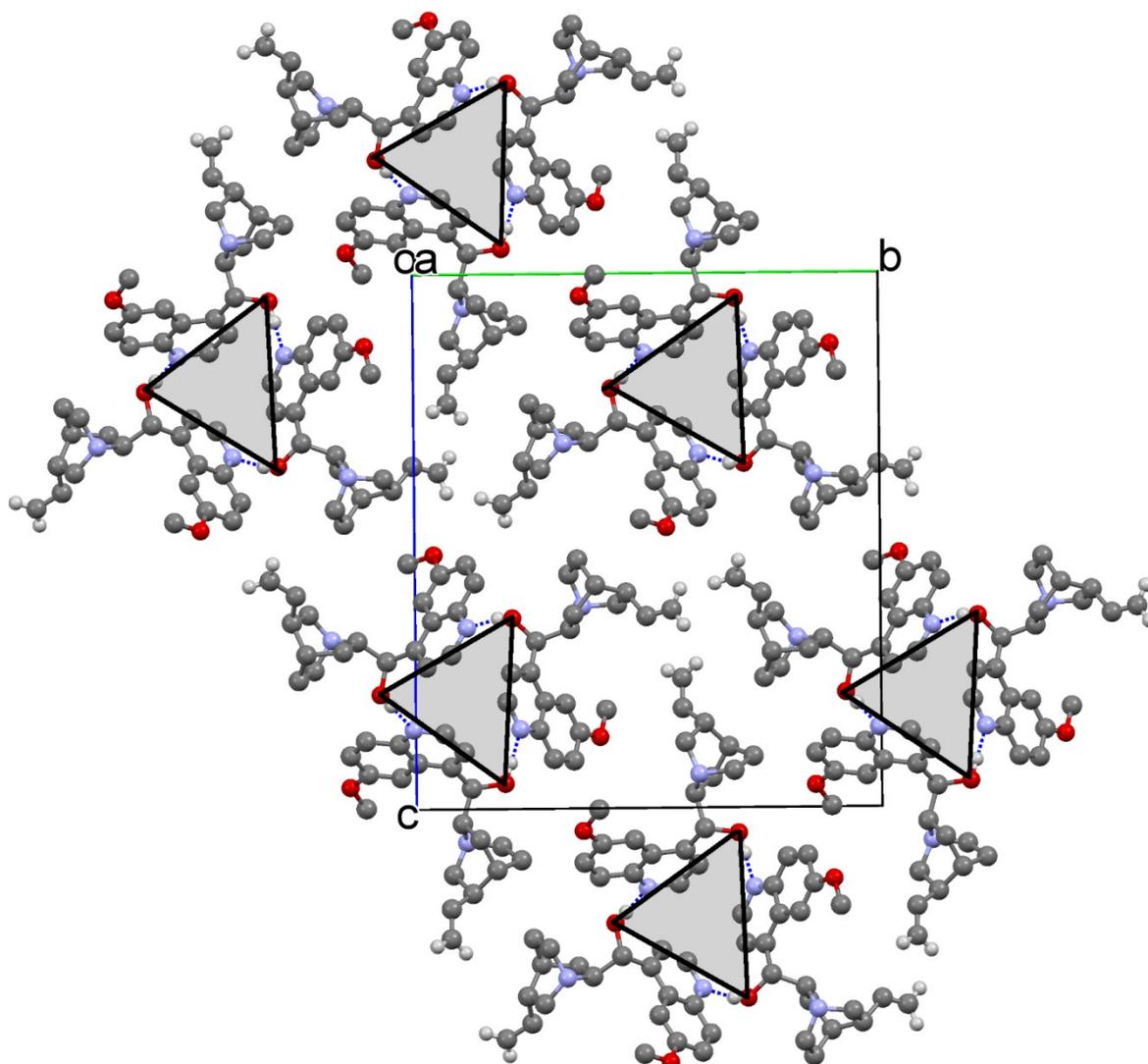


Figure S5. The crystal structure of quinine projected along the OH $\cdots$ N bonded chains down the crystal [x] axis. The H-atoms have been omitted for clarity.

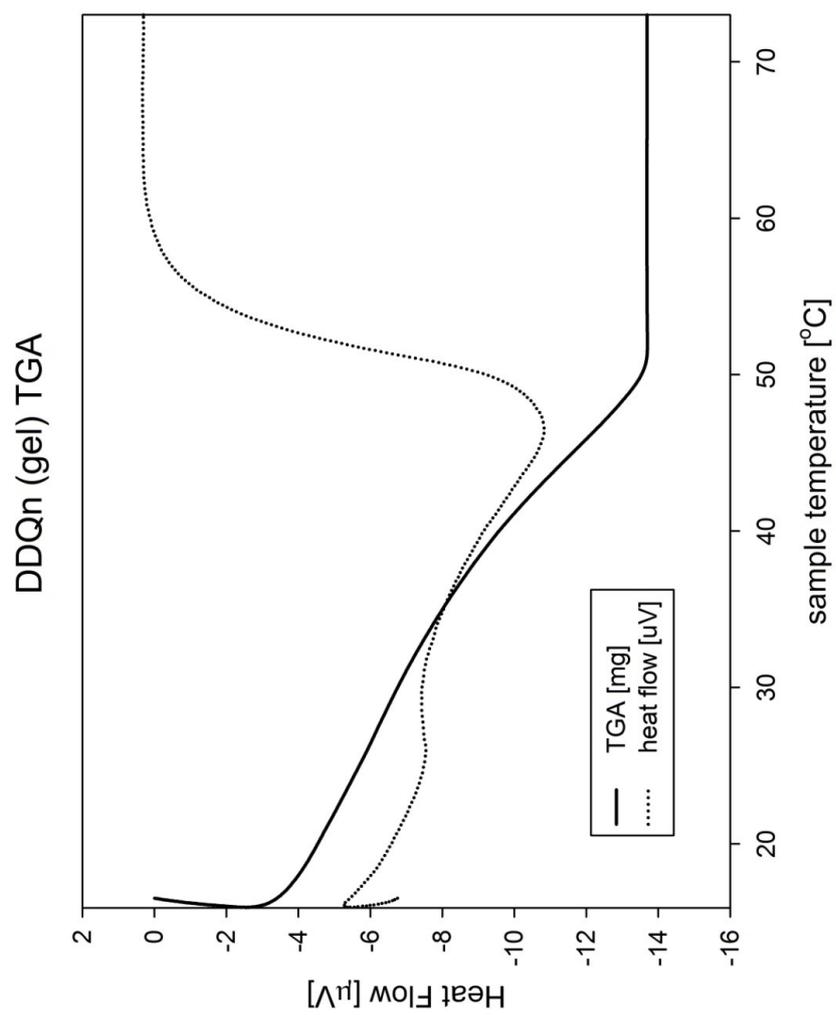


Figure S6. TGA analysis of **DDQn** gel. Sample (gel) 23 mg, heat gradient 10  $^{\circ}\text{C}/\text{min}$  (Instrument Setsys 1200 Setaram).

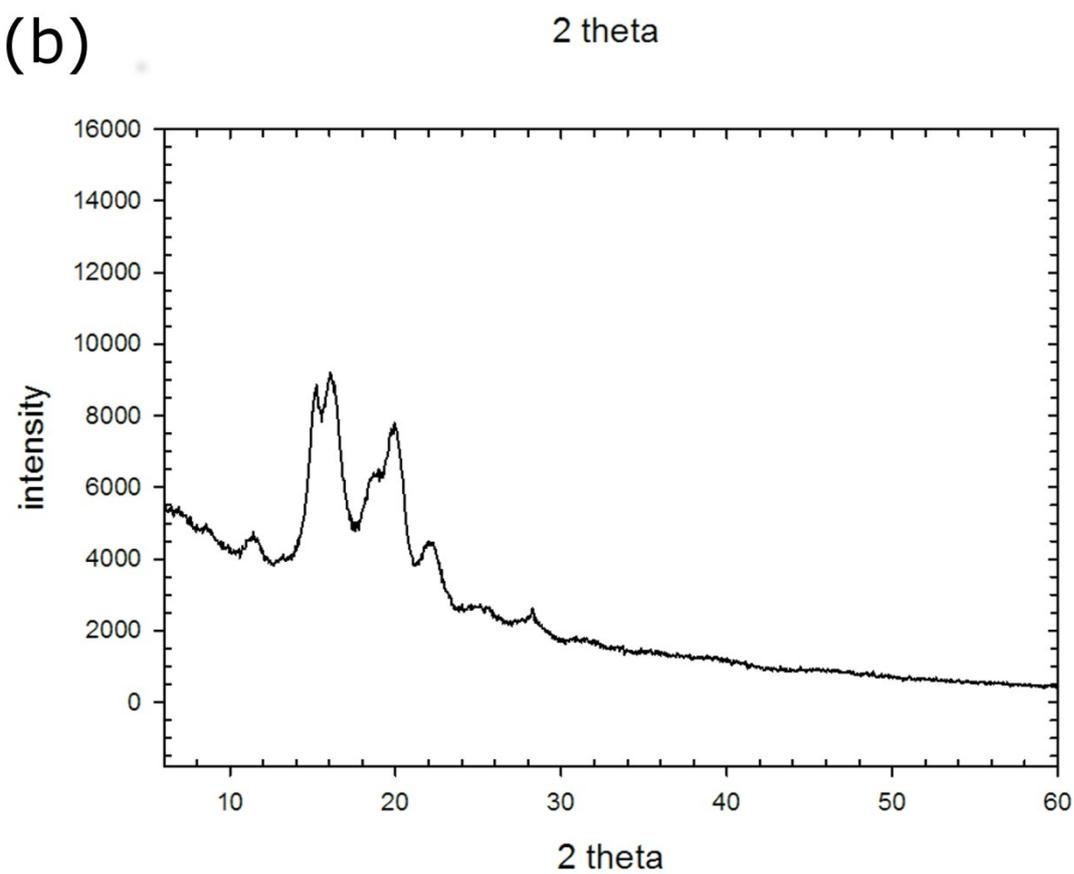
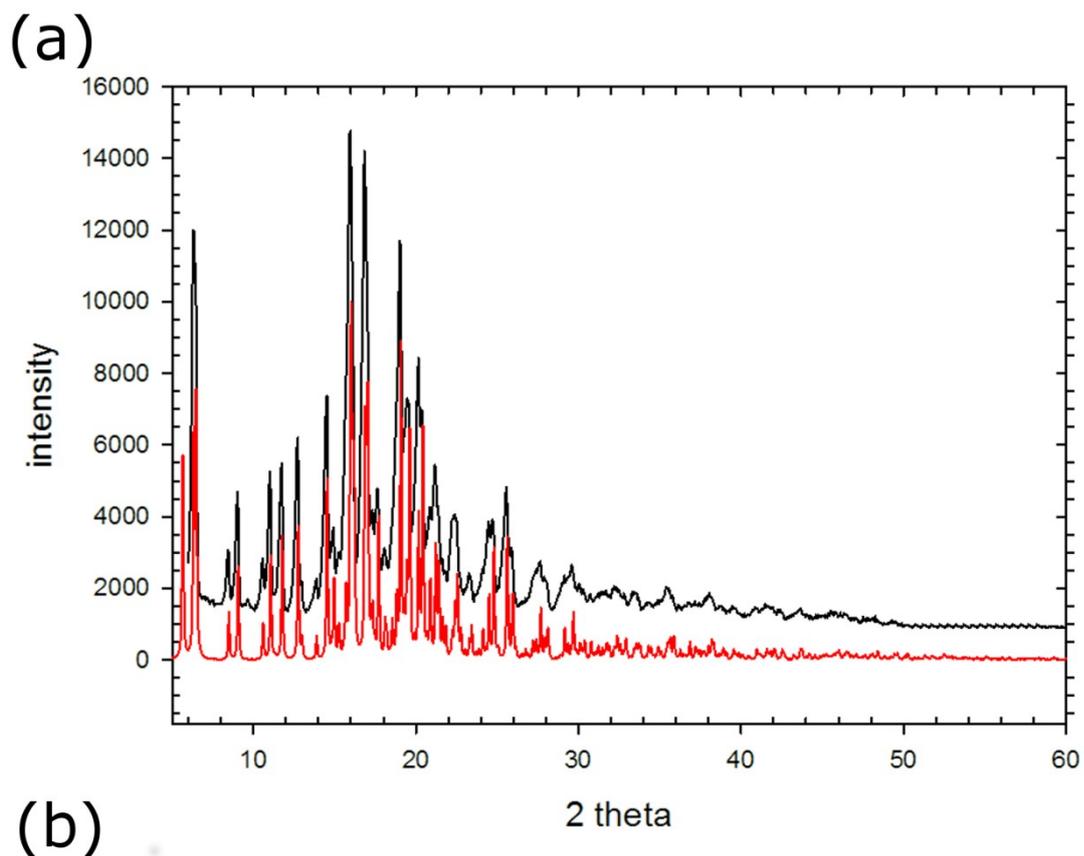


Figure S7. The XRPD pattern of **DDQn** shown in Figures S8 (a) and S9 (b) shown in their full  $2\theta$  range.

Table S1. Crystal data and structure refinement for 10,11-didehydroquinine and 10,11-didehydroquinidine.

	10,11-didehydroquinine	10,11-didehydroquinine	10,11-didehydroquinidine
Empirical formula	C <sub>20</sub> H <sub>22</sub> N <sub>2</sub> O <sub>2</sub>	C <sub>20</sub> H <sub>22</sub> N <sub>2</sub> O <sub>2</sub>	C <sub>20</sub> H <sub>22</sub> N <sub>2</sub> O <sub>2</sub>
Formula weight	322.40	322.40	322.40
Temperature	120(2) K	296(2) K	296(2) K
Wavelength	1.54184 Å	1.54184 Å	1.54184 Å
Crystal system	Orthorhombic	Orthorhombic	Orthorhombic
Space group	<i>P</i> 2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	<i>P</i> 2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	<i>P</i> 2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>
Unit cell dimensions:			
a	5.98370(10) Å	6.0461(4) Å	7.13240(10) Å
b	27.8283(6) Å	27.8882(19) Å	11.67130(10) Å
c	30.8628(5) Å	31.178(2) Å	20.2977(2) Å
Volume	5139.16(16) Å <sup>3</sup>	5257.0(6) Å <sup>3</sup>	1689.67(3) Å <sup>3</sup>
Z	12	12	4
Calculated density	1.250 gcm <sup>-3</sup>	1.222 gcm <sup>-3</sup>	1.267 gcm <sup>-3</sup>
Absorption coefficient	0.646 mm <sup>-1</sup>	0.631 mm <sup>-1</sup>	0.655 mm <sup>-1</sup>
F(000)	2064	2064	688
Crystal size	0.25 x 0.15 x 0.10 mm	0.15 x 0.10 x 0.08 mm	0.20 x 0.20 x 0.05 mm
Theta range for data collection	2.14 to 73.76°	2.12 to 71.10°	4.36 to 75.59°
Limiting indices h,k,l	-6/7, -34/33, -36/38	-7/7, -34/34, -37/37	-8/6, -13/14, -25/25
Reflections collected / unique	39117/10270 R <sub>int</sub> =0.0550	82985/10026 R <sub>int</sub> =0.0896	9289/3212 R <sub>int</sub> =0.0187
Completeness	to $\theta = 73.76$ 99.1 %	to $\theta = 71.10$ 98.7 %	to $\theta = 75.59$ 97.5 %
Refinement method	Full-matrix least-squares on F <sup>2</sup>	Full-matrix least-squares on F <sup>2</sup>	Full-matrix least-squares on F <sup>2</sup>
Data/restraints/parameters	10270/3/659	10026/3/659	3212/0/222
Goodness-of-fit on F <sup>2</sup>	1.122	1.040	1.050
Final R indices [I>2 $\sigma$ (I)]	R <sub>1</sub> = 0.0743 wR <sub>2</sub> = 0.2239	R <sub>1</sub> = 0.0573 wR <sub>2</sub> = 0.1169	R <sub>1</sub> = 0.0292 wR <sub>2</sub> = 0.0774
R indices (all data)	R <sub>1</sub> = 0.0777 wR <sub>2</sub> = 0.2251	R <sub>1</sub> = 0.0994 wR <sub>2</sub> = 0.1386	R <sub>1</sub> = 0.0307 wR <sub>2</sub> = 0.0789
Largest diff. peak and hole	0.433 and -0.291 e.Å <sup>-3</sup>	0.154 and -0.144 e.Å <sup>-3</sup>	0.132 and -0.102 e.Å <sup>-3</sup>

Table S2. Bond lengths [ $\text{\AA}$ ] for 10,11-didehydroquinine in 296 K.

O(1) – C(9)	1.421(4)	O(1B) – C(9B)	1.426(4)	O(1C) – C(9C)	1.418(3)
N(2) – C(20)	1.369(4)	C(20B) – N(2B)	1.370(4)	C(20C) – N(2C)	1.370(4)
C(20) – C(19)	1.412(4)	C(19B) – C(20B)	1.418(5)	C(20C) – C(19C)	1.414(4)
C(20) – C(18)	1.415(4)	C(20B) – C(18B)	1.418(5)	C(20C) – C(18C)	1.421(4)
N(2) – C(12)	1.317(4)	N(2B) – C(12B)	1.307(4)	N(2C) – C(12C)	1.297(4)
O(2) – C(16)	1.376(4)	O(2B) – C(16B)	1.375(4)	O(2C) – C(16C)	1.364(4)
O(2) – C(21)	1.419(4)	O(2B) – C(21B)	1.427(5)	O(2C) – C(21C)	1.401(5)
C(13) – C(14)	1.360(5)	C(13B) – C(14B)	1.374(5)	C(13C) – C(14C)	1.370(4)
C(13) – C(12)	1.394(5)	C(13B) – C(12B)	1.394(5)	C(13C) – C(12C)	1.392(5)
C(15) – C(16)	1.362(4)	C(15B) – C(16B)	1.370(5)	C(15C) – C(16C)	1.365(5)
C(15) – C(19)	1.415(4)	C(15B) – C(19B)	1.420(4)	C(15C) – C(19C)	1.413(4)
C(17) – C(18)	1.342(5)	C(17B) – C(18B)	1.355(5)	C(17C) – C(18C)	1.342(5)
N(1) – C(8)	1.479(4)	C(8B) – N(1B)	1.473(4)	N(1C) – C(8C)	1.478(4)
C(7) – C(8)	1.549(4)	C(7B) – C(8B)	1.532(4)	C(7C) – C(8C)	1.525(4)
C(8) – C(9)	1.532(4)	C(8B) – C(9B)	1.533(4)	C(8C) – C(9C)	1.536(4)
C(14) – C(19)	1.430(4)	C(14B) – C(19B)	1.419(4)	C(14C) – C(19C)	1.422(4)
C(14) – C(9)	1.516(4)	C(14B) – C(9B)	1.519(5)	C(14C) – C(9C)	1.514(4)
C(4) – C(7)	1.521(4)	C(4B) – C(7B)	1.522(5)	C(4C) – C(7C)	1.525(5)
C(16) – C(17)	1.411(5)	C(16B) – C(17B)	1.393(5)	C(16C) – C(17C)	1.404(5)
N(1) – C(2)	1.459(4)	N(1B) – C(21B)	1.451(5)	N(1C) – C(2C)	1.464(5)
N(1) – C(6)	1.476(4)	N(1B) – C(6B)	1.492(6)	N(1C) – C(6C)	1.470(5)
C(4) – C(5)	1.512(5)	C(4B) – C(5B)	1.515(6)	C(4C) – C(5C)	1.531(6)
C(4) – C(3)	1.533(5)	C(4B) – C(3B)	1.535(6)	C(3C) – C(4C)	1.528(5)
C(10) – C(3)	1.473(6)	C(10B) – C(3B)	1.447(7)	C(10C) – C(3C)	1.445(7)
C(2) – C(3)	1.552(5)	C(2B) – C(3B)	1.534(6)	C(2C) – C(3C)	1.560(5)
C(6) – C(5)	1.530(5)	C(6B) – C(5B)	1.527(7)	C(6C) – C(5C)	1.536(6)
C(10) – C(11)	1.163(6)	C(10B) – C(11B)	1.161(7)	C(10C) – C(11C)	1.188(7)

Table S3. Bond lengths [ $\text{\AA}$ ] for 10,11-didehydroquinine in 120 K.

O(1) – C(9)	1.417(5)	O(1B) – C(9B)	1.418(5)	O(1C) – C(9C)	1.414(5)
N(2) – C(20)	1.367(6)	C(20B) – N(2B)	1.362(6)	C(20C) – N(2C)	1.366(6)
C(20) – C(19)	1.413(6)	C(19B) – C(20B)	1.434(6)	C(20C) – C(19C)	1.418(6)
C(20) – C(18)	1.420(6)	C(20B) – C(18B)	1.411(6)	C(20C) – C(18C)	1.423(6)
N(2) – C(12)	1.319(6)	N(2B) – C(12B)	1.325(6)	N(2C) – C(12C)	1.317(6)
O(2) – C(16)	1.370(6)	O(2B) – C(16B)	1.373(6)	O(2C) – C(16C)	1.351(5)
O(2) – C(21)	1.428(6)	O(2B) – C(21B)	1.427(7)	O(2C) – C(21C)	1.428(7)
C(13) – C(14)	1.382(7)	C(13B) – C(14B)	1.372(6)	C(13C) – C(14C)	1.370(6)
C(13) – C(12)	1.401(7)	C(13B) – C(12B)	1.404(7)	C(13C) – C(12C)	1.405(6)
C(15) – C(16)	1.373(6)	C(15B) – C(16B)	1.371(6)	C(15C) – C(16C)	1.373(6)
C(15) – C(19)	1.433(6)	C(15B) – C(19B)	1.418(6)	C(15C) – C(19C)	1.415(6)
C(17) – C(18)	1.359(7)	C(17B) – C(18B)	1.375(8)	C(17C) – C(18C)	1.360(7)
N(1) – C(8)	1.489(6)	C(8B) – N(1B)	1.477(6)	N(1C) – C(8C)	1.476(5)
C(7) – C(8)	1.555(6)	C(7B) – C(8B)	1.541(6)	C(7C) – C(8C)	1.543(6)
C(8) – C(9)	1.533(6)	C(8B) – C(9B)	1.534(6)	C(8C) – C(9C)	1.540(6)
C(14) – C(19)	1.436(6)	C(14B) – C(19B)	1.425(6)	C(14C) – C(19C)	1.431(5)
C(14) – C(9)	1.513(6)	C(14B) – C(9B)	1.522(6)	C(14C) – C(9C)	1.514(6)
C(4) – C(7)	1.537(6)	C(4B) – C(7B)	1.534(6)	C(4C) – C(7C)	1.531(6)
C(16) – C(17)	1.408(7)	C(16B) – C(17B)	1.412(7)	C(16C) – C(17C)	1.414(7)
N(1) – C(2)	1.458(6)	N(1B) – C(2B)	1.472(7)	N(1C) – C(2C)	1.465(6)
N(1) – C(6)	1.485(6)	N(1B) – C(6B)	1.485(7)	N(1C) – C(6C)	1.476(6)
C(4) – C(5)	1.513(6)	C(4B) – C(5B)	1.522(7)	C(4C) – C(5C)	1.537(7)
C(4) – C(3)	1.536(6)	C(4B) – C(3B)	1.537(7)	C(3C) – C(4C)	1.545(7)
C(10) – C(3)	1.471(7)	C(10B) – C(3B)	1.455(8)	C(10C) – C(3C)	1.463(8)
C(2) – C(3)	1.573(6)	C(2B) – C(3B)	1.569(7)	C(2C) – C(3C)	1.564(7)
C(6) – C(5)	1.549(6)	C(6B) – C(5B)	1.545(8)	C(6C) – C(5C)	1.539(7)
C(10) – C(11)	1.184(8)	C(10B) – C(11B)	1.162(9)	C(10C) – C(11C)	1.180(8)

Table S4. Bond lengths [Å] 10,11-didehydroquinidine.

O(1) – C(9)	1.4177(15)
N(2) – C(20)	1.3708(18)
C(20) – C(19)	1.4236(18)
C(20) – C(18)	1.405(2)
N(2) – C(12)	1.305(2)
O(2) – C(16)	1.3630(18)
O(2) – C(21)	1.4061(18)
C(13) – C(14)	1.363(2)
C(13) – C(12)	1.405(2)
C(15) – C(16)	1.3723(18)
C(15) – C(19)	1.4104(19)
C(17) – C(18)	1.364(2)
N(1) – C(8)	1.4869(16)
C(7) – C(8)	1.5423(17)
C(8) – C(9)	1.5392(18)
C(14) – C(19)	1.4303(18)
C(14) – C(9)	1.5207(18)
C(4) – C(7)	1.534(2)
C(16) – C(17)	1.4031(19)
N(1) – C(2)	1.4754(16)
N(1) – C(6)	1.4797(17)
C(4) – C(5)	1.521(2)
C(4) – C(3)	1.5397(19)
C(10) – C(3)	1.467(2)
C(2) – C(3)	1.5569(18)
C(6) – C(5)	1.541(2)
C(10) – C(11)	1.170(2)

Table S5. Angles [°] for 10,11-didehydroquinine in 296 K.

N(2) – C(20) – C(18)	117.6(3)	N(2B) – C(20B) – C(18B)	117.6(3)	N(2C) – C(20C) – C(18C)	117.5(3)
C(19) – C(20) – C(18)	119.2(3)	C(19B) – C(20B) – C(18B)	119.6(3)	C(19C) – C(20C) – C(18C)	119.4(3)
C(12) – N(2) – C(20)	116.9(3)	C(12B) – N(2B) – C(20B)	117.1(3)	C(12C) – N(2C) – C(20C)	116.7(3)
C(16) – O(2) – C(21)	117.6(3)	C(16B) – O(2B) – C(21B)	117.5(3)	C(16C) – O(2C) – C(21C)	117.4(3)
C(14) – C(13) – C(12)	120.8(3)	C(14B) – C(13B) – C(12B)	120.2(3)	C(14C) – C(13C) – C(12C)	120.0(3)
C(16) – C(15) – C(19)	119.9(3)	C(16B) – C(15B) – C(19B)	119.5(3)	C(16C) – C(15C) – C(19C)	119.7(3)
C(17) – C(18) – C(20)	121.3(3)	C(17B) – C(18B) – C(20B)	119.9(4)	C(17C) – C(18C) – C(20C)	119.9(3)
N(1) – C(8) – C(7)	111.1(2)	N(1B) – C(8B) – C(7B)	110.2(3)	N(1C) – C(8C) – C(7C)	111.1(2)
N(1) – C(8) – C(9)	112.6(3)	N(1B) – C(8B) – C(9B)	112.8(3)	N(1C) – C(8C) – C(9C)	111.5(3)
C(7) – C(8) – C(9)	113.4(3)	C(7B) – C(8B) – C(9B)	114.8(3)	C(7C) – C(8C) – C(9C)	114.2(3)
C(13) – C(14) – C(19)	117.7(3)	C(13B) – C(14B) – C(19B)	117.6(3)	C(13C) – C(14C) – C(19C)	117.5(3)
C(13) – C(14) – C(9)	120.8(3)	C(13B) – C(14B) – C(9B)	120.9(3)	C(13C) – C(14C) – C(9C)	120.7(3)
C(19) – C(14) – C(9)	121.4(3)	C(19B) – C(14B) – C(9B)	121.5(3)	C(19C) – C(14C) – C(9C)	121.7(3)
O(1) – C(9) – C(14)	110.8(3)	O(1B) – C(9B) – C(14B)	110.0(3)	O(1C) – C(9C) – C(14C)	112.1(3)
O(1) – C(9) – C(8)	109.8(2)	O(1B) – C(9B) – C(8B)	109.3(3)	O(1C) – C(9C) – C(8C)	109.3(2)
C(14) – C(9) – C(8)	109.4(2)	C(14B) – C(9B) – C(8B)	109.6(2)	C(14C) – C(9C) – C(8C)	110.3(2)
C(20) – C(19) – C(14)	117.4(3)	C(20B) – C(19B) – C(14B)	118.0(3)	C(20C) – C(19C) – C(14C)	117.7(3)
C(20) – C(19) – C(15)	118.7(3)	C(20B) – C(19B) – C(15B)	118.8(3)	C(20C) – C(19C) – C(15C)	119.0(3)
C(15) – C(19) – C(14)	123.9(3)	C(15B) – C(19B) – C(14B)	123.2(3)	C(15C) – C(19C) – C(14C)	123.3(3)
C(4) – C(7) – C(8)	108.3(3)	C(4B) – C(7B) – C(8B)	107.8(3)	C(4C) – C(7C) – C(8C)	108.2(3)
N(2) – C(12) – C(13)	123.8(3)	N(2B) – C(12B) – C(13B)	124.4(4)	N(2C) – C(12C) – C(13C)	125.1(3)
C(15) – C(16) – O(2)	125.4(3)	C(15B) – C(16B) – O(2B)	124.7(4)	C(15C) – C(16C) – O(2C)	124.7(4)
C(15) – C(16) – C(17)	121.3(3)	C(15B) – C(16B) – C(17B)	121.3(4)	C(15C) – C(16C) – C(17C)	120.9(3)
O(2) – C(16) – C(17)	113.3(3)	O(2B) – C(16B) – C(17B)	114.0(4)	O(2C) – C(16C) – C(17C)	114.4(3)
C(2) – N(1) – C(8)	108.0(3)	C(2B) – N(1B) – C(8B)	107.1(3)	C(2C) – N(1C) – C(8C)	107.6(3)
C(2) – N(1) – C(6)	107.9(3)	C(2B) – N(1B) – C(6B)	108.6(3)	C(2C) – N(1C) – C(6C)	108.2(3)
C(8) – N(1) – C(6)	111.3(3)	C(8B) – N(1B) – C(6B)	110.2(3)	C(8C) – N(1C) – C(6C)	111.2(3)
C(5) – C(4) – C(7)	108.6(3)	C(5B) – C(4B) – C(7B)	108.2(3)	C(5C) – C(4C) – C(7C)	107.6(3)
C(5) – C(4) – C(3)	108.1(3)	C(5B) – C(4B) – C(3B)	107.1(4)	C(5C) – C(4C) – C(3C)	107.1(3)
C(7) – C(4) – C(3)	110.1(3)	C(7B) – C(4B) – C(3B)	109.5(3)	C(7C) – C(4C) – C(3C)	110.6(3)
C(10) – C(3) – C(2)	112.9(3)	C(10B) – C(3B) – C(2B)	113.2(4)	C(10C) – C(3C) – C(2C)	110.0(4)
C(10) – C(3) – C(4)	112.0(3)	C(10B) – C(3B) – C(4B)	111.8(4)	C(10C) – C(3C) – C(4C)	113.5(3)
C(4) – C(3) – C(2)	107.3(3)	C(4B) – C(3B) – C(2B)	108.0(3)	C(4C) – C(3C) – C(2C)	107.4(3)
C(4) – C(5) – C(6)	109.2(3)	C(4B) – C(5B) – C(6B)	107.8(4)	C(4C) – C(5C) – C(6C)	108.7(3)
N(1) – C(2) – C(3)	112.2(3)	N(1B) – C(2B) – C(3B)	111.3(4)	N(1C) – C(2C) – C(3C)	111.4(3)
C(11) – C(10) – C(3)	179.0(5)	C(11B) – C(10B) – C(3B)	176.0(7)	C(11C) – C(10C) – C(3C)	177.5(5)
N(1) – C(6) – C(5)	111.3(3)	N(1B) – C(6B) – C(5B)	111.1(4)	N(1C) – C(6C) – C(5C)	110.8(3)
C(18) – C(17) – C(16)	119.5(3)	C(18B) – C(17B) – C(16B)	120.9(4)	C(18C) – C(17C) – C(16C)	121.1(3)
N(2) – C(20) – C(19)	123.3(3)	N(2B) – C(20B) – C(19B)	122.8(3)	N(2C) – C(20C) – C(19C)	123.0(3)

Table S6. Angles [°] for 10,11-didehydroquinine in 120 K.

N(2) – C(20) – C(18)	117.6(4)	N(2B) – C(20B) – C(18B)	118.2(4)	N(2C) – C(20C) – C(18C)	117.1(4)
C(19) – C(20) – C(18)	119.1(4)	C(19B) – C(20B) – C(18B)	119.3(4)	C(19C) – C(20C) – C(18C)	119.3(4)
C(12) – N(2) – C(20)	117.1(4)	C(12B) – N(2B) – C(20B)	117.9(4)	C(12C) – N(2C) – C(20C)	117.2(4)
C(16) – O(2) – C(21)	117.6(4)	C(16B) – O(2B) – C(21B)	117.1(4)	C(16C) – O(2C) – C(21C)	117.7(4)
C(14) – C(13) – C(12)	120.1(4)	C(14B) – C(13B) – C(12B)	120.7(4)	C(14C) – C(13C) – C(12C)	120.3(4)
C(16) – C(15) – C(19)	119.4(4)	C(16B) – C(15B) – C(19B)	120.6(5)	C(16C) – C(15C) – C(19C)	120.0(4)
C(17) – C(18) – C(20)	120.8(4)	C(17B) – C(18B) – C(20B)	120.7(5)	C(17C) – C(18C) – C(20C)	120.0(4)
N(1) – C(8) – C(7)	111.5(3)	N(1B) – C(8B) – C(7B)	110.3(4)	N(1C) – C(8C) – C(7C)	111.6(3)
N(1) – C(8) – C(9)	112.4(4)	N(1B) – C(8B) – C(9B)	112.9(4)	N(1C) – C(8C) – C(9C)	111.5(3)
C(7) – C(8) – C(9)	112.6(4)	C(7B) – C(8B) – C(9B)	113.8(4)	C(7C) – C(8C) – C(9C)	113.8(4)
C(13) – C(14) – C(19)	117.3(4)	C(13B) – C(14B) – C(19B)	117.8(4)	C(13C) – C(14C) – C(19C)	118.0(4)
C(13) – C(14) – C(9)	120.1(4)	C(13B) – C(14B) – C(9B)	119.7(4)	C(13C) – C(14C) – C(9C)	120.6(4)
C(19) – C(14) – C(9)	122.5(4)	C(19B) – C(14B) – C(9B)	122.4(4)	C(19C) – C(14C) – C(9C)	121.5(4)
O(1) – C(9) – C(14)	111.5(4)	O(1B) – C(9B) – C(14B)	110.5(4)	O(1C) – C(9C) – C(14C)	111.6(4)
O(1) – C(9) – C(8)	109.8(3)	O(1B) – C(9B) – C(8B)	110.0(3)	O(1C) – C(9C) – C(8C)	109.4(3)
C(14) – C(9) – C(8)	108.8(3)	C(14B) – C(9B) – C(8B)	109.2(3)	C(14C) – C(9C) – C(8C)	110.7(3)
C(20) – C(19) – C(14)	117.9(4)	C(20B) – C(19B) – C(14B)	117.8(4)	C(20C) – C(19C) – C(14C)	117.2(4)
C(20) – C(19) – C(15)	119.2(4)	C(20B) – C(19B) – C(15B)	118.5(4)	C(20C) – C(19C) – C(15C)	119.3(4)
C(15) – C(19) – C(14)	122.8(4)	C(15B) – C(19B) – C(14B)	123.7(4)	C(15C) – C(19C) – C(14C)	123.5(4)
C(4) – C(7) – C(8)	107.8(4)	C(4B) – C(7B) – C(8B)	107.6(4)	C(4C) – C(7C) – C(8C)	107.8(4)
N(2) – C(12) – C(13)	124.3(4)	N(2B) – C(12B) – C(13B)	123.3(4)	N(2C) – C(12C) – C(13C)	123.8(4)
C(15) – C(16) – O(2)	124.6(4)	C(15B) – C(16B) – O(2B)	125.1(5)	C(15C) – C(16C) – O(2C)	125.3(4)
C(15) – C(16) – C(17)	121.1(4)	C(15B) – C(16B) – C(17B)	120.9(5)	C(15C) – C(16C) – C(17C)	120.5(4)
O(2) – C(16) – C(17)	114.3(4)	O(2B) – C(16B) – C(17B)	114.0(4)	O(2C) – C(16C) – C(17C)	114.2(4)
C(2) – N(1) – C(8)	107.9(3)	C(2B) – N(1B) – C(8B)	106.8(4)	C(2C) – N(1C) – C(8C)	107.6(4)
C(2) – N(1) – C(6)	107.9(3)	C(2B) – N(1B) – C(6B)	107.4(4)	C(2C) – N(1C) – C(6C)	107.7(4)
C(8) – N(1) – C(6)	111.1(3)	C(8B) – N(1B) – C(6B)	111.6(4)	C(8C) – N(1C) – C(6C)	111.2(4)
C(5) – C(4) – C(7)	109.1(4)	C(5B) – C(4B) – C(7B)	107.9(4)	C(5C) – C(4C) – C(7C)	107.9(4)
C(5) – C(4) – C(3)	109.0(4)	C(5B) – C(4B) – C(3B)	107.7(5)	C(5C) – C(4C) – C(3C)	107.2(4)
C(7) – C(4) – C(3)	109.7(4)	C(7B) – C(4B) – C(3B)	110.2(4)	C(7C) – C(4C) – C(3C)	109.7(4)
C(10) – C(3) – C(2)	112.9(4)	C(10B) – C(3B) – C(2B)	113.0(5)	C(10C) – C(3C) – C(2C)	110.9(4)
C(10) – C(3) – C(4)	112.7(4)	C(10B) – C(3B) – C(4B)	111.9(4)	C(10C) – C(3C) – C(4C)	112.5(4)
C(4) – C(3) – C(2)	106.9(4)	C(4B) – C(3B) – C(2B)	106.9(4)	C(4C) – C(3C) – C(2C)	107.9(4)
C(4) – C(5) – C(6)	108.5(4)	C(4B) – C(5B) – C(6B)	107.7(4)	C(4C) – C(5C) – C(6C)	108.5(4)
N(1) – C(2) – C(3)	112.3(4)	N(1B) – C(2B) – C(3B)	111.2(4)	N(1C) – C(2C) – C(3C)	111.4(4)
C(11) – C(10) – C(3)	178.7(6)	C(11B) – C(10B) – C(3B)	178.4(7)	C(11C) – C(10C) – C(3C)	176.5(6)
N(1) – C(6) – C(5)	111.6(4)	N(1B) – C(6B) – C(5B)	111.4(4)	N(1C) – C(6C) – C(5C)	111.8(4)
C(18) – C(17) – C(16)	120.3(4)	C(18B) – C(17B) – C(16B)	119.9(5)	C(18C) – C(17C) – C(16C)	120.9(4)
N(2) – C(20) – C(19)	123.3(4)	N(2B) – C(20B) – C(19B)	122.5(4)	N(2C) – C(20C) – C(19C)	123.6(4)

Table S7. Angles[°] for 10,11-didehydroquinine.

N(2) – C(20) – C(18)	117.90(12)
C(19) – C(20) – C(18)	119.11(13)
C(12) – N(2) – C(20)	116.66(13)
C(16) – O(2) – C(21)	119.38(13)
C(14) – C(13) – C(12)	120.16 (14)
C(16) – C(15) – C(19)	121.02(12)
C(17) – C(18) – C(20)	121.31(13)
N(1) – C(8) – C(7)	109.86(10)
N(1) – C(8) – C(9)	111.76(9)
C(7) – C(8) – C(9)	114.54(11)
C(13) – C(14) – C(19)	117.54(13)
C(13) – C(14) – C(9)	121.30(12)
C(19) – C(14) – C(9)	121.16(11)
O(1) – C(9) – C(14)	111.13(11)
O(1) – C(9) – C(8)	110.44(10)
C(14) – C(9) – C(8)	110.98(10)
C(20) – C(19) – C(14)	117.76(12)
C(20) – C(19) – C(15)	118.31(12)
C(15) – C(19) – C(14)	123.92(12)
C(4) – C(7) – C(8)	107.73(11)
N(2) – C(12) – C(13)	124.87(15)
C(15) – C(16) – O(2)	116.20(12)
C(15) – C(16) – C(17)	120.32(13)
O(2) – C(16) – C(17)	123.48(13)
C(2) – N(1) – C(8)	110.76(10)
C(2) – N(1) – C(6)	107.96(10)
C(8) – N(1) – C(6)	107.83(10)
C(5) – C(4) – C(7)	109.76(12)
C(5) – C(4) – C(3)	107.39(13)
C(7) – C(4) – C(3)	108.65(11)
C(10) – C(3) – C(2)	112.27(13)
C(10) – C(3) – C(4)	113.69(13)
C(4) – C(3) – C(2)	106.46(11)
C(4) – C(5) – C(6)	108.34(11)
N(1) – C(2) – C(3)	111.40(11)
C(11) – C(10) – C(3)	178.04(17)
N(1) – C(6) – C(5)	110.46(11)
C(18) – C(17) – C(16)	119.91(13)
N(2) – C(20) – C(19)	122.99(13)

Table S8. Torsion angles [°] for 10,11-didehydroquinine and 10,11-didehydroquinidine.

10,11-ddQn	Mol A	Mol B	Mol C	Mol A	Mol B	Mol C
Temperature	296 K			120 K		
C(12) – C(13) – C(14) – C(19)	1.2(5)	0.9(5)	0.2(4)	1.0(6)	1.2(6)	-0.2(6)
C(13) – C(14) – C(19) – C1(5)	-179.1(3)	-180.0(3)	-179.6(3)	-179.2(4)	179.9(4)	-179.7(4)
C(14) – C(19) – C(15) – C(16)	-178.9(3)	177.2(3)	178.5(3)	-178.5(4)	177.7(4)	179.6(4)
C(19) – C(15) – C(16) – C(17)	-0.2(5)	0.7(5)	1.4(5)	0.5(6)	-0.3(7)	0.5(7)
C(20) – C(19) – C(14) – C(13)	0.6(4)	-1.2(4)	0.0(4)	1.0(6)	-1.7(6)	0.3(6)
C(19) – C(20) – C(18) – C(17)	0.8(5)	0.4(5)	0.6(5)	0.4(6)	0.9(7)	1.3(7)
C(15) – C(16) – C(17) – C(18)	-0.8(5)	0.8(6)	-0.8(6)	-1.9(7)	1.7(8)	0.4(7)
C(16) – C(17) – C(18) – C(20)	0.5(5)	-1.3(6)	-0.2(6)	1.4(7)	-2.0(8)	-1.2(7)
C(17) – C(18) – C(20) – N(2)	-178.9(3)	-178.4(3)	-178.8(3)	-179.2(4)	-178.7(5)	-179.0(4)
C(14) – C(19) – C(20) – N(2)	-1.7(4)	1.0(5)	-0.2(4)	-2.3(6)	1.5(6)	-0.1(6)
C(15) – C(19) – C(20) – N(2)	178.0(3)	179.8(3)	179.4(3)	177.9(4)	-180.0(4)	179.9(4)
N(2) – C(12) – C(13) – C(14)	-2.1(5)	-0.5(6)	-0.2(5)	-2.1(7)	-0.5(7)	0.0(7)
C(14) – C(19) – C(20) – C(18)	178.5(3)	-177.8(3)	-179.5(3)	178.1(4)	-178.1(4)	179.6(4)
C(15) – C(19) – C(20) – C(18)	-1.8(4)	1.0(5)	0.1(4)	-1.7(6)	0.4(6)	-0.4(6)
C(19) – C(20) – N(2) – C(12)	0.9(4)	-0.4(5)	0.3(4)	1.4(6)	-0.8(6)	-0.1(6)
C(20) – N(2) – C(12) – C(13)	1.0(5)	0.2(5)	-0.1(5)	0.9(6)	0.3(7)	0.2(6)
C(18) – C(20) – N(2) – C(12)	-179.3(3)	178.4(3)	179.5(3)	-179.0(4)	178.8(4)	-179.8(4)
C(20) – C(19) – C(15) – C(16)	1.5(4)	-1.5(5)	-1.1(4)	1.2(6)	-0.7(6)	-0.4(6)
C(20) – C(19) – C(14) – C(9)	178.6(3)	175.8(3)	179.6(3)	178.2(4)	174.2(4)	179.1(4)
C(21) – O(2) – C(16) – C(15)	0.6(5)	5.8(6)	10.2(6)	0.7(6)	6.5(7)	9.3(7)
C(21) – O(2) – C(16) – C(17)	-179.4(3)	-173.6(4)	-169.7(4)	-179.7(4)	-172.7(5)	-169.5(5)
C(12) – C(13) – C(14) – C(9)	-176.9(3)	-176.1(3)	-179.4(3)	-176.2(4)	-174.8(4)	-179.1(4)
C(13) – C(14) – C(9) – O(1)	-23.2(4)	-15.9(4)	-13.6(4)	-23.3(5)	-16.8(5)	-13.1(5)
C(19) – C(14) – C(9) – O(1)	158.8(3)	167.2(3)	166.8(3)	159.7(4)	167.3(4)	168.0(4)
C(15) – C(19) – C(14) – C(9)	-1.0(4)	-3.0(5)	0.0(4)	-2.0(6)	-4.2(6)	-0.9(6)
C(19) – C(15) – C(16) – O(2)	179.8(3)	-178.7(3)	-178.5(3)	-179.8(4)	-179.5(4)	-178.2(4)
O(2) – C(16) – C(17) – C(18)	179.2(3)	-179.8(4)	179.1(3)	178.4(4)	-179.1(5)	179.2(4)
C(13) – C(14) – C(9) – C(8)	97.9(3)	104.3(3)	108.3(3)	98.0(5)	104.3(5)	109.0(4)
C(19) – C(14) – C(9) – C(8)	-80.0(3)	-72.6(4)	-71.3(4)	-79.1(5)	-71.6(5)	-69.9(5)
C(2) – N(1) – C(8) – C(7)	64.0(3)	73.1(4)	70.1(4)	64.3(4)	73.6(5)	70.0(5)
C(6) – N(1) – C(8) – C(7)	-54.3(4)	-44.9(4)	-48.3(4)	-53.9(5)	-43.6(5)	-47.7(5)
C(2) – N(1) – C(8) – C(9)	-167.6(3)	-157.2(3)	-161.3(3)	-168.2(3)	-157.9(4)	-161.5(4)
C(6) – N(1) – C(8) – C(9)	74.2(3)	84.8(4)	80.3(3)	73.7(4)	85.0(5)	80.8(4)
C(4) – C(7) – C(8) – N(1)	-6.0(4)	-20.6(4)	-16.3(4)	-6.2(5)	-21.7(5)	-15.8(5)
C(9) – C(8) – C(7) – C(4)	-134.0(3)	-149.2(3)	-143.5(3)	-133.6(4)	-149.7(4)	-143.0(4)
C(14) – C(9) – C(8) – N(1)	158.3(3)	167.4(3)	156.8(3)	158.2(3)	169.5(4)	156.0(3)
O(1) – C(9) – C(8) – N(1)	-79.9(3)	-72.0(4)	-79.6(3)	-79.5(4)	-69.1(5)	-80.5(4)
C(7) – C(8) – C(9) – C(14)	-74.5(3)	-65.2(4)	-76.2(3)	-74.9(4)	-63.9(5)	-76.6(4)
O(1) – C(9) – C(8) – C(7)	47.3(3)	55.4(4)	47.3(4)	47.4(5)	57.5(5)	46.8(5)
C(8) – C(7) – C(4) – C(5)	62.7(4)	71.1(4)	68.1(4)	63.3(5)	71.7(5)	67.6(5)
C(8) – C(7) – C(4) – C(3)	-55.5(3)	-45.3(4)	-48.6(4)	-56.0(5)	-45.7(5)	-48.8(5)
C(11) – C(10) – C(3) – C(4)	-91(25)	98(8)	96(15)	14(27)	46(28)	97(11)
C(11) – C(10) – C(3) – C(2)	148(10)	-24(9)	-24(15)	-108(27)	-75(28)	-24(11)
C(5) – C(4) – C(3) – C(10)	176.9(3)	-177.6(4)	-176.1(3)	177.1(4)	-176.7(4)	-176.2(4)
C(7) – C(4) – C(3) – C(10)	-64.7(4)	-60.5(4)	-59.1(4)	-63.5(5)	-59.2(6)	-59.3(5)
C(5) – C(4) – C(3) – C(2)	-58.7(3)	-52.4(4)	-54.2(4)	-58.3(4)	-52.4(5)	-53.5(5)
C(7) – C(4) – C(3) – C(2)	59.7(3)	64.6(4)	62.8(4)	61.1(4)	65.1(5)	63.4(5)
C(6) – N(1) – C(2) – C(3)	60.9(3)	66.8(5)	66.3(4)	61.4(4)	68.0(6)	66.3(5)
C(8) – N(1) – C(2) – C(3)	-59.5(3)	-52.2(5)	-54.0(4)	-58.7(4)	-51.8(6)	-53.7(5)

C(10) – C(3) – C(2) – N(1)	122.4(4)	110.9(4)	114.5(4)	121.7(4)	109.7(5)	113.5(5)
C(4) – C(3) – C(2) – N(1)	-1.5(4)	-13.4(5)	-9.5(4)	-2.8(5)	-13.9(6)	-10.1(5)
C(2) – N(1) – C(6) – C(5)	-59.9(4)	-50.0(5)	-54.4(4)	-59.5(5)	-51.6(6)	-55.1(5)
C(8) – N(1) – C(6) – C(5)	58.5(4)	67.1(5)	63.6(4)	58.6(5)	65.1(6)	62.5(5)
C(7) – C(4) – C(5) – C(6)	-59.1(4)	-49.5(5)	-53.0(4)	-59.1(5)	-50.6(6)	-53.5(5)
C(3) – C(4) – C(5) – C(6)	60.3(4)	68.4(5)	66.0(4)	60.6(5)	68.3(5)	64.6(5)
N(1) – C(6) – C(5) – C(4)	-0.7(5)	-16.0(6)	-10.2(5)	-1.2(6)	-14.2(7)	-9.4(6)

Table S9. Torsion angles[°] for 10,11-didehydroquinidine.

C(12) – C(13) – C(14) – C(19)	-0.2(2)	C(19) – C(14) – C(9) – C(8)	82.21(14)
C(13) – C(14) – C(19) – C(15)	179.21(13)	C(2) – N(1) – C(8) – C(7)	45.97(13)
C(14) – C(19) – C(15) – C(16)	179.31(13)	C(6) – N(1) – C(8) – C(7)	-71.98(12)
C(19) – C(15) – C(16) – C(17)	-0.7(2)	C(2) – N(1) – C(8) – C(9)	-82.32(12)
C(20) – C(19) – C(14) – C(13)	-1.12(18)	C(6) – N(1) – C(8) – C(9)	159.73(10)
C(19) – C(20) – C(18) – C(17)	-1.3(2)	C(4) – C(7) – C(8) – N(1)	20.13(14)
C(15) – C(16) – C(17) – C(18)	0.8(2)	C(9) – C(8) – C(7) – C(4)	146.87(11)
C(16) – C(17) – C(18) – C(20)	0.3(2)	C(14) – C(9) – C(8) – N(1)	-157.62(10)
C(17) – C(18) – C(20) – N(2)	178.64(14)	O(1) – C(9) – C(8) – N(1)	78.68(13)
C(14) – C(19) – C(20) – N(2)	1.70(19)	C(7) – C(8) – C(9) – C(14)	76.63(13)
C(15) – C(19) – C(20) – N(2)	-178.60(12)	O(1) – C(9) – C(8) – C(7)	-47.08(14)
N(2) – C(12) – C(13) – C(14)	1.0(3)	C(8) – C(7) – C(4) – C(5)	46.19(14)
C(14) – C(19) – C(20) – C(18)	-178.36(12)	C(8) – C(7) – C(4) – C(3)	-70.97(14)
C(15) – C(19) – C(20) – C(18)	1.33(19)	C(11) – C(10) – C(3) – C(4)	-105(6)
C(19) – C(20) – N(2) – C(12)	-0.9(2)	C(11) – C(10) – C(3) – C(2)	134(6)
C(20) – N(2) – C(12) – C(13)	-0.6(3)	C(5) – C(4) – C(3) – C(10)	166.98(12)
C(18) – C(20) – N(2) – C(12)	179.20(15)	C(7) – C(4) – C(3) – C(10)	-74.36(14)
C(20) – C(19) – C(15) – C(16)	-0.36(19)	C(5) – C(4) – C(3) – C(2)	-68.88(14)
C(20) – C(19) – C(14) – C(9)	178.59(11)	C(7) – C(4) – C(3) – C(2)	49.77(15)
C(21) – O(2) – C(16) – C(15)	172.83(15)	C(6) – N(1) – C(2) – C(3)	50.07(15)
C(21) – O(2) – C(16) – C(17)	-7.7(2)	C(8) – N(1) – C(2) – C(3)	-67.78(14)
C(12) – C(13) – C(14) – C(9)	-179.82(14)	C(10) – C(3) – C(2) – N(1)	141.03(12)
C(13) – C(14) – C(9) – O(1)	25.22(17)	C(4) – C(3) – C(2) – N(1)	16.02(16)
C(19) – C(14) – C(9) – O(1)	-154.48(11)	C(2) – N(1) – C(6) – C(5)	-68.76(15)
C(15) – C(19) – C(14) – C(9)	-1.09(19)	C(8) – N(1) – C(6) – C(5)	50.97(15)
C(19) – C(15) – C(16) – O(2)	178.76(12)	C(7) – C(4) – C(5) – C(6)	-66.27(15)
O(2) – C(16) – C(17) – C(18)	-178.65(14)	C(3) – C(4) – C(5) – C(6)	51.68(15)
C(13) – C(14) – C(9) – C(8)	-98.09(15)	N(1) – C(6) – C(5) – C(4)	15.07(17)

Table S10. Hydrogen bonds for 10,11-didehydroquinine in 120 K [ $\text{\AA}$  and  $^\circ$ ].

D-H $\cdots$ A	H $\cdots$ A	D $\cdots$ A	D-H $\cdots$ A
O(1) – H(1) $\cdots$ N(2B)	1.978	2.786(5)	168.05
O(1B) – H(1B) $\cdots$ N(2C) <sup>i</sup>	1.939	2.758(5)	177.03
O(1C) – H(1C) $\cdots$ N(2) <sup>i</sup>	1.956	2.775(5)	173.31

Symmetry codes: (i)=-1+x,y,z; (ii)= -x,1/2+y,1.5-z

## References

- 1 K. M. Kacprzak, W. Linder, N. M. Maier, *Chirality*, 2008, **20**, 441-445
- 2 C. F. Macrae, P. R. Edgington, P. McCabe, E. Pidcock, G. P. Shields, R. Taylor, M. Towler, J. van de Streek, *J. Appl. Crystallogr.* 2006, **39**, 453–457.