Supporting information for

Controlled growth of porous δ -MnO₂ nanosheets on carbon fibers as a bi-functional catalyst for rechargeable

lithium-oxygen batteries

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Figure S1 SEM image of δ -MnO₂/CCFs-K80 sample showing a typical core-shell

structure.



Figure S2 SEM images of pure δ -MnO₂ sample at high (a, b) and low (c) magnification.



Figure S3 SEM images of the acid-treated CFs before loading the MnO_2 (a) and the SEM images of δ -MnO₂/CCFs-K40 (b), δ -MnO₂/CCFs-K80 (c) and δ -MnO₂/CCFs-

K120 (d) after etching the MnO_2 phase.



Figure S4 Line-scanning (marked by the red line in the insert image) elemental profiles of C, Mn and O elements.



Figure S5 Nitrogen adsorption-desorption isotherms and the pore size distribution (insert) of CFs (a), δ -MnO₂/CCFs-K80 (b) and δ -MnO₂/CCFs-K120 (c) samples.



Figure S6 Charge/discharge curves during of the cycled Li-O₂ battery based on the δ -





Figure S7 SEM images of the δ -MnO₂/CCFs-K80-XC-72 (a, b, c) and XC-72 (d, e, f) electrodes at before discharge, fully discharged and charged state.



Figure S8 SEM images of the fully discharged electrode with $\delta\text{-MnO}_2/\text{CCFs-K80}$



catalyst at high (a) and low (b) magnifications.

Figure S9 Raman spectra of δ -MnO₂/CCFs-K80 electrode after discharge at the 3rd



Figure S10 Nitrogen adsorption-desorption isotherms of XC-72 carbon (a) and the FITR spectrum of XC-72 carbon (b).



Figure S11 SEM images of the discharge products of the electrode with δ -MnO₂/CCFs-K80 catalyst after discharged at 350 mAh g⁻¹.